# Design, synthesis, and biological evaluation of novel sulindac derivatives as partial agonists of $\operatorname{PPAR} \gamma$ with potential anti-diabetic efficacy 

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#### Abstract

Peroxisome proliferator-activated receptor gamma (PPAR $\gamma$ ) is a valuable drug target for diabetic treatment and ligands of PPAR $\gamma$ have shown potent anti-diabetic efficacy. However, to overcome the severe side effects of current PPAR $\gamma$-targeted drugs, novel PPAR $\gamma$ ligands need to be developed. Sulindac, an identified ligand of PPAR $\gamma$, is widely used in clinic as a non-steroidal anti-inflammatory drug. To explore its potential application for diabetes, we designed and synthesized a series of sulindac derivatives to investigate their structure-activity relationship as PPAR $\gamma$ ligand and potential anti-diabetic effect. We found that meta-substitution in sulindac's benzylidene moiety was beneficial to PPAR $\gamma$ binding and transactivation. $Z$ rather than $E$ configuration of the benzylidene double bond endowed derivatives with the selectivity of PPAR $\gamma$ activation. The indene fluorine is essential for binding and regulating PPAR $\gamma$. Compared with rosiglitazone, compound 6b with benzyloxyl meta-substitution and $Z$ benzylidene double bond weakly induced adipogenesis and PPAR $\gamma$-targeted gene expression. However, $\mathbf{6 b}$ potently improved glucose tolerance in a diabetic mice model. Unlike rosiglitazone, $\mathbf{6 b}$ was devoid of apparent toxicity to osteoblastic formation. Thus, we provided some useful guidelines for PPAR $\gamma$-based optimization of sulindac and an anti-diabetic lead compound with less side effects.


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## 1. Introduction

Peroxisome proliferator-activated receptor gamma (PPAR $\gamma$ ) belonging to nuclear receptor superfamily is a major regulator in adipocyte differentiation, lipid metabolism, glucose homeostasis and insulin sensitivity [1-3]. The physiopathological activities of $\operatorname{PPAR} \gamma$ are tightly controlled by its endogenous and exogenous ligands including 15-deoxy-delta 12,14-prostaglandin $\mathrm{J}_{2}$, eicosapentaenoic acid, 9-HODE, and 13-HODE [4-8], while its pharmacological modulation by ligands has been applied to the

[^0]treatment of diabetes [9]. Actos (pioglitazone) and Avandia (rosiglitazone), the two potent agonists of PPAR $\gamma$, effectively increase insulin sensitization and improve glycemic control in patients with type 2 diabetes [10-12]. However, strong PPAR $\gamma$-activating drugs were once withdrawn from the market or had restricted prescription due to their severe adverse effects such as weight gain, edema, liver injury and heart failure [13,14]. In addition, PPAR $\gamma$ and its potent ligands also negatively regulate osteoblastogenesis, increasing the risk of osteoporosis. Thus, novel PPAR $\gamma$-targeting drugs with less above side effects should be devised for diabetes treatment [15].

Repositioning of clinical drugs is a promising strategy for drug development [16-18]. Sulindac, a non-steroidal anti-inflammatory drug (NSAID), binds to PPAR $\gamma$ and regulates PPAR $\gamma$ activity, as well as exerts PPAR $\gamma$-dependent physiological activities [19-21]. However, sulindac has its well-known adverse effects from long-term
clinical application, including potential gastrointestinal and cardiovascular side effects mostly owning to its potent inhibition of cyclooxygenases (COX-1 and COX-2) [22]. We and others have shown that the methythiol group of sulindac sulfide metabolized from methyl sulfoxide of sulindac is essential for COX inhibition [23,24]. However, the PPAR $\gamma$-dependent structure-activity relationship (SAR) of sulindac derivatives needs to be comprehensively defined.

Felts et al. has reported that the optimized derivative of sulindac (compound 24, Scheme 1) binds to PPAR $\gamma$ and regulates PPAR $\gamma$ activities with slightly improved $\mathrm{EC}_{50}$ value compared with that of the hit $Z$-sulindac sulfide [25]. A disadvantage of compound $\mathbf{2 4}$ is its less selectivity in PPAR $\gamma$ regulation because it activates both PPAR $\alpha$ and PPAR $\gamma$. Furthermore, the sulindac derivatives reported in Felts's work are of the $E$ benzylidene double bond, while the geometric isomerism of the benzylidene double bond is $Z$ in sulindac and sulindac sulfide (Scheme 1). Also, all the derivatives they reported are $E$ - $2^{\prime}$-des-methyl (removal of the indenyl methyl group). These altered geometric isomerism may lead to distinct metabolic modes and side effects between sulindac and these derivatives. In addition, according to the crystal structure of sulindac-PPAR $\gamma$ complex, the ligand binding pocket (LBP) of PPAR $\gamma$ is large enough to adopt two sulindac molecule [26], which seems to complicate the design of its further optimization.

In this work, we designed and synthesized a series of sulindac derivatives with the core scaffold unchanged and the methyl sulfoxide group replaced, and evaluated their PPAR $\gamma$-based SAR and anti-diabetic activity.

## 2. Results and discussion

### 2.1. Chemistry

Synthesis method of indenone (Rec1 and Rec2, Scheme 2) from starting materials (various benzaldehydes) by Perkin reaction, hydrogenation and Friedel-Crafts reaction was described in our previous papers (see Scheme 2), and the characterization of compounds $\mathbf{1 , 2}, \mathbf{4 a}$ and $\mathbf{1 0}$ were also shown in our published papers [23,27,28]. The general synthesis process for intermediates Rec1 and Rec2 is shown in method section. The methyl group (purple) in indene gives predominantly $Z$-isomers. Our target molecules were produced by the condensation reaction between indene and different substituted aromatic aldehydes using standard method described also in our previous papers (see Scheme 2).

### 2.2. SAR study of sulindac derivatives

We found that sulindac only slightly activated Gal4/DBD-PPAR $\gamma /$ LBD chimera transactivation in our mammalian-one-hybrid assay ( $\mathrm{EC}_{50}$ : $22.65 \pm 2.94 \mu \mathrm{M}$, max fold induction: $1.34 \pm 0.10$ ) (Table 1, compound 1). In order to remove its COX inhibition activity, the methyl sulfoxide was substituted to hydrogen to obtain compound 2. Interestingly, 2 had higher potency ( $\mathrm{EC}_{50}: 2.78 \pm 0.70 \mu \mathrm{M}$ ) and efficacy (max fold induction: $3.62 \pm 1.47$ ) than sulindac for inducing PPAR $\gamma$ transactivation (Table 1 and Supplementary Fig. 1). Consistently, 2 had higher PPAR $\gamma$ binding affinity ( $\mathrm{K}_{\mathrm{D}}=7.94 \pm 1.83 \mu \mathrm{M}$ ) than sulindac ( $\mathrm{K}_{\mathrm{D}}>100 \mu \mathrm{M}$ ) according to our fluorescence titration


Z-Sulindac Sulfide a): activating PPAR $\gamma$ $E C_{50}=0.8 \pm 0.2 \mu \mathrm{M}$ Max Fold Induction $=3.4$


Compound 6b
a): activating PPAR $\gamma$ but not PPARa
$E C_{50}=0.93 \pm 0.44 \mu \mathrm{M}$
Max Fold Induction $=10.26$

## Our work

Scheme 1. Design strategy for novel sulindac analogues targeting to PPAR $\gamma$.
a) propionic anhydride,
$\mathrm{K}_{2} \mathrm{CO}_{3}$, reflux,
b) $\mathrm{H}_{2} / \mathrm{Pd}-\mathrm{C}, \mathrm{MeOH}$, r. t.
c) polyphosphoric acid, $80{ }^{\circ} \mathrm{C}$

d) ethyl bromoacetate, zinc powder, iodine, THF, reflux;
$\mathrm{NaOH}, \mathrm{MeOH}$, r. t., overnight;


| $1 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=p-\mathrm{SOCH}_{3}$ | 6a $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=p-\mathrm{OCH}_{2} \mathrm{Ph}$ |
| :---: | :---: |
| $2 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{H}$ | 6b $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{OCH}_{2} \mathrm{Ph}$ |
| 3a $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=p-\mathrm{F}$ | $7 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{Br}$ |
| 3b $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{F}$ | $8 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{Cl}$ |
| 4a $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=p-\mathrm{C}_{3} \mathrm{H}_{7}$ | $9 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m-\mathrm{OH}$ |
| 4b $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=\mathrm{m}-\mathrm{C}_{3} \mathrm{H}_{7}$ | $10 \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=\mathrm{m}-\mathrm{OCH}_{3}$ |
| 5a $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=p$-OPh | $11_{z} \mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=\mathrm{m}-\mathrm{CH}_{3}$ |
| 5b $\mathrm{R}_{1}=\mathrm{F}, \mathrm{R}=m$-OPh | 12z $\mathrm{R}_{1}=\mathrm{H}, \mathrm{R}=m-\mathrm{CH}_{3}$ |



Scheme 2. General synthesis method for the sulindac analogues 1-12 $\mathbf{E}_{\text {. a) }}$ : Propionic or acetic anhydride, $\mathrm{K}_{2} \mathrm{CO}_{3}$, reflux. b): $\mathrm{H}_{2} / \mathrm{Pd}-\mathrm{C}$, MeOH, r. t. c): Polyphosphoric acid (PPA), $80^{\circ} \mathrm{C} \mathrm{d}$ ): Reformatsky reaction, ethyl bromoacetate, zinc powder, iodine, THF, reflux; $\mathrm{NaOH}, \mathrm{MeOH}, \mathrm{r} . \mathrm{t}$., overnight; e): $\mathrm{MeONa} / \mathrm{MeOH}, 80{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}$.

Table 1
The study of compounds on PPAR $\gamma$ binding and transcriptional activation.

|  |
| :--- | :--- | :--- | :--- | :--- | :--- |

[^1]assay (Table 1 and Supplementary Fig. 2), suggesting that the stronger ability of $\mathbf{2}$ in activating PPAR $\gamma$ came from its higher PPAR $\gamma$ binding affinity. These results indicated that the methyl sulfoxide group was not required or even detrimental for sulindac to transactivate and bind to PPAR $\gamma$.

The methyl sulfoxide group is at the para-position of sulindac's benzylidene moiety. We then investigated whether chemical modification of this para-position with other functional groups could obtain optimization as to PPAR $\gamma$ modulation. Derivatives with fluorine (3a), isopropyl (4a), phenoxyl (5a) and benzyloxyl (6a) substitutions at the para-position were synthesized. We found that only the fluorine substituted derivative 3a had comparable capability as 2 in activating (max fold induction: $3.18 \pm 0.47$ ) and binding to $\left(K_{D}=6.67 \pm 0.65 \mu \mathrm{M}\right)$ PPAR $\gamma$. However, the other derivatives exhibited weaker capability $\left(\mathrm{EC}_{50}>50 \mu \mathrm{M}\right.$, max fold induction $<1.1$ and $K_{D}>100 \mu \mathrm{M}$ ) (Table 1 and Supplementary Figs. 1 and 2), suggesting that chemical modifications at the para-position was indeed not appropriate for PPAR $\gamma$-based optimization.

We then synthesized meta-substituted isomers ( $\mathbf{3 b}, \mathbf{4 b}, \mathbf{5 b}, \mathbf{6 b}$ ) of above compounds. Surprisingly, all the four meta-substituted derivatives showed higher potency and efficacy ( $\mathrm{EC}_{50} \leq 6.41 \mu \mathrm{M}$, max fold induction $\geq 4.39$ ) than their corresponding parasubstituted isomers on transactivating Gal4/DBD-PPAR $\gamma / \mathrm{LBD}$ (Table 1 and Supplementary Fig. 1). The meta- but not the parasubstituted derivatives also substantially stimulated the transactivation of RXR $\alpha / \operatorname{PPAR} \gamma$ heterodimers (Fig. 1A). Consistently, the meta-substituted derivatives had higher PPAR $\gamma$ binding affinity $\left(K_{D} \leq 2.47 \mu \mathrm{M}\right)$ (Table 1 and Supplementary Fig. 2). Hence, substitutions at the meta-rather than the para-position was favorable to PPAR $\gamma$ regulation.

To obtain SAR of meta-substitutions, we synthesized more metaderivatives 7-10 with various substitutions (Table 1). However, all the derivatives had no obvious improvement in regulating and binding to PPAR $\gamma$. We then explored the effect of meta-substitutions with large steric groups such as phenoxyl (5b) and benzyloxyl (6b). Surprisingly, 5b and 6b exhibited significantly higher potency and efficacy than 2 on activating and binding PPAR $\gamma(5 b$, $\mathrm{EC}_{50}$ : $1.178 \pm 0.27 \mu \mathrm{M}$, max fold induction: $9.03 \pm 0.40$, and $K_{D}=1.73 \pm 0.65 \mu \mathrm{M} ; \mathbf{6 b}, \mathrm{EC}_{50}: 0.93 \pm 0.44 \mu \mathrm{M}$, max fold induction: $10.26 \pm 0.94$, and $K_{D}=0.96 \pm 0.35 \mu \mathrm{M}$ ) (Table 1 and Supplementary

Figs. 1 and 2). Also, they had the strongest capability in activating PPAR $\gamma$ among all the synthesized meta-derivatives (Table 1 and Supplementary Fig. 1). The aromatic ring of phenoxyl and benzyloxyl moiety introduced in $\mathbf{5 b}$ and $\mathbf{6 b}$ formed $\pi-\pi$ interaction with aromatic amino acid residues (His449) in the ligand binding pocket of PPAR $\gamma$ (Supplementary Fig. 4B and 4C), leading to enhanced binding energy. Hence, the derivatives with phenoxyl and benzyloxyl substitutions bound to PPAR $\gamma$ and likely induced a transactivating conformation of PPAR $\gamma$.

It has been shown that sulindac derivatives with $E$ benzylidene double bond are PPAR $\gamma$ ligands [25]. We then compared the PPAR $\gamma$ based activity of the $E$ and $Z$ derivatives with methyl group at the meta-position of benzylidene moiety ( $\mathbf{1 1}_{Z}$ and $\left.\mathbf{1 1}_{\boldsymbol{E}}\right)$. To obtain $\mathbf{1 1}_{\boldsymbol{E}}$, the methyl group in the indene moiety need to be removed to make the $E$ configuration available. Interestingly, both $\mathbf{1 1}_{Z}$ and $\mathbf{1 1}_{E}$ activated and bound to PPAR $\gamma$ (Table 1 and Fig. 1B and Supplementary Figs. 1 and 2). Felts et al. reported that the $E$ derivatives are mixed agonists of PPAR $\gamma$ and $\operatorname{PPAR} \alpha$ [25]. Consistently, we found that $\mathbf{1 1}_{\boldsymbol{E}}$ activated both PPAR $\alpha$ and PPAR $\gamma$ but not PPAR $\beta$ (Fig. 2A). However, $\mathbf{1 1}_{Z}$ only activated PPAR $\gamma$, but not $\operatorname{PPAR} \alpha$ or PPAR $\beta$ (Fig. 2A). Moreover, all the $Z$ derivatives that activated PPAR $\gamma$ did not activate $\operatorname{PPAR} \alpha$ or PPAR $\beta$ (Fig. 2B), indicating the PPAR $\gamma$ selectivity of the $Z$ derivatives. Thus, $Z$ rather than $E$ configuration rendered the derivatives with PPAR $\gamma$ selectivity. The advantage of the strict PPAR $\gamma$ selectivity of $Z$ configuration is to avoid the PPAR $\alpha$-targeted side effects in the future clinical application.

We explored whether fluorine group in the indene moiety was essential for the derivatives to activate PPAR $\gamma$. Mammalian-onehybrid assay showed that $\mathbf{1 1}_{Z}$ and $\mathbf{1 1}_{E}$ could strongly induce PPAR $\gamma$ transactivation. However, their corresponding compounds $\mathbf{1 2}_{Z}$ and $\mathbf{1 2}_{E}$ without fluorine substitution failed to either bind to PPAR $\gamma$ or activate PPAR $\gamma$ (Table 1 and Fig. 1B and Supplementary Figs. 1 and 2). Thus, fluorine group in the indene moiety was vital for both the $Z$ and $E$ derivatives to activate and bind to PPAR $\gamma$.

Moreover, we investigated the effect of $\mathbf{6 b}$ on COX-2 activity in vitro. Sulindac sulfide exhibited strong potency on inhibiting COX-2 activity with an $\mathrm{IC}_{50}$ at $1.396 \mu \mathrm{M}$, whereas the potency of $\mathbf{6 b}$ was much weaker ( $\mathrm{IC}_{50}, 286.1 \mu \mathrm{M}$ ) (Supplementary Fig. 5), verifying previous reports that the methythiol group of sulindac sulfide is essential for COX inhibition.

A


B


Fig. 1. Derivatives with meta-substitutions and fluorine activate the transcriptional activity of RXR $\alpha / \mathrm{PPAR} \gamma$ heterodimer. Cos-7 cells were co-transfected with PPRE-luciferase, renilla-luciferase, RXR $\alpha$ and PPAR $\gamma$ expression plasmids, and then treated with the indicated compounds for 12 h . Cells were then harvested, and firefly and renilla luciferase activities were measured. Renilla luciferase values were normalized to firefly luciferase activity to obtain the relative luciferase activity. DM, DMSO; RGZ, rosiglitazone ( $0.1 \mu \mathrm{M}$ ); RLA, relative luciferase activity. Data are presented as means $\pm$ SD. Statistical significance of differences of each condition vs. DMSO control was calculated by student's $t$-test. *p < 0.05 , ${ }^{* *}$ p < 0.01, ***p < 0.001 .




 student's $t$-test. *p $<0.05,{ }^{* *} \mathrm{p}<0.01,{ }^{* * *} \mathrm{p}<0.001$.

### 2.3. 11 $\mathbf{Z}_{\text {Z }}$ and $\mathbf{6 b}$ weakly induced RXR $\alpha / P P A R \gamma$ targeted gene expression and adipogenesis

To further characterize the properties of sulindac derivatives as PPAR $\gamma$ ligands, we evaluated their adipogenic activity, one of the major physiological functions of PPAR $\gamma$ agonists [29-31]. To this end, we selected two derivatives $\mathbf{6 b}$ and $\mathbf{1 1}_{\boldsymbol{z}}$ showing PPAR $\gamma$ agonistic activity, as well as derivative $\mathbf{6 a}$ without apparent agonistic activity as a negative control. PPAR $\gamma$ agonist rosiglitazone $(1 \mu \mathrm{M})$ induced substantial adipogenesis in mouse 3T3-L1 preadipocytes as indicated by the significant increase of Oil red 0 staining (Fig. 3A and 3B). As expected, $\mathbf{6 b}$ and $\mathbf{1 1}_{\mathbf{Z}}$ also induced adipocyte differentiation. However, they were much less potent than rosiglitazone in adipogenesis induction (Fig. 3A and 3B), correlating with their relatively low induction of PPAR $\gamma$ transactivation (Fig. 2). Consistently, 6a without apparent PPAR $\gamma$ agonistic activity did not show pro-adipogenic effect (Fig. 3A and 3B). $\mathbf{6 b}$ was also less potent than rosiglitazone in upregulating the mRNA expressions of adipocyte lipid binding protein 2 (aP2) and PPAR $\gamma$, two PPAR $\gamma$ targeted genes (Fig. 3C).

## 2.4. $\boldsymbol{6 b}$ improved hyperglycaemia in murine diabetes model

The binding and activation of PPAR $\gamma$ prompted us to investigate the therapeutic effect of the derivatives on diabetes [10]. To this
end, we selected $\mathbf{6 b}$ because of its relatively high ability in PPAR $\gamma$ modulation. Compared with the control mice fed with normal diet, the high-fat-diet (HFD) model mice exhibited serious glucose intolerance, indicated by the continuous high level of glucose after glucose spike (Fig. 4A). Both rosiglitazone and $\mathbf{6 b}$ significantly improved glucose tolerance, indicated by the reduced glucose level compared with the vehicle treatment group (Fig. 4A). The antidiabetes effect of $\mathbf{6 b}$ was further verified by its improvement of insulin resistance in our cell-based assay (Fig. 4B and 4C). High concentration of palmitic acid (PA) treatment impaired insulin signal in HepG2 cells, showing from reduced insulin-induced AKT phosphorylation [32-34]. 6b and rosiglitazone potently enhanced insulin-induced AKT activation in this insulin-resistance cell model (Fig. 4B and 4C). Thus, 6b might increase insulin sensitivity to improve glucose tolerance.

### 2.5. 6b and $\mathbf{1 1}_{Z}$ had no significant effect on osteoblastic differentiation

One of the adverse effects of rosiglitazone and other thiazolidinediones is the inhibition of osteoblastogenesis, resulting in osteoporosis from long-term usage [35-39]. As shown in Fig. 5A and 5B, the calcification, a marker of osteoblast, was strongly induced by inducing agents in rat osteosarcoma UMR106 cells as stained by Alizarin Red S. Rosiglitazone strongly reduced




 ${ }^{* * *}$ p $<0.001$.


Fig. 4. 6b Improves glucose tolerance. (A) The HFD-induced diabetes mice were treated with the vehicle or the compounds ( $4 \mathrm{mg} / \mathrm{kg}$ of rosiglitazone or $40 \mathrm{mg} / \mathrm{kg}$ of $\mathbf{6 b}$ ) for 1 month . The glucose tolerance test was performed. Statistical significance was calculated by Two-way ANOVA. (B, C) HepG2 cells treated with or without 200 mM palmitic acid for 24 h , and then co-treated with the indicated compounds ( $1 \mu \mathrm{M}$ of rosiglitazone or $5 \mu \mathrm{M}$ of $\mathbf{6 b}$ ) for 24 h . Cells were then treated with 10 nM of insulin for 15 min and harvested. Protein expression was detected by western blotting using anti-p-AKTser473, p-AKTthr308, AKT and $\beta$-actin antibodies. The expression of $\beta$-actin served as a loading control (B). The relative expression ratios of p-AKTthr308/AKT and p-AKTser473/AKT were measured by the Image J software (C). PA, palmitic acid; RGZ, rosiglitazone; ND, normal diet; HFD, high-fat-diet. Data are presented as means $\pm$ SD. ${ }^{*} \mathrm{p}<0.05,{ }^{* * *} \mathrm{p}<0.001$, Student's $t$-test.
calcification in UMR106 cells (Fig. 5A and 5B). Also, the activity of alkaline phosphatase (ALP), a marker enzyme of osteoblast [40-42], was inhibited by rosiglitazone (Fig. 5C). These results
verified that rosiglitazone inhibits osteoblastogenesis. In contrast to rosiglitazone, $\mathbf{6 b}$ and $\mathbf{1 1} \boldsymbol{Z}$ did not show inhibitory effect on mineralization and ALP activity at $5 \mu \mathrm{M}$ concentration (Fig. 5), suggesting
their less adverse effect on osteoblastogenesis than rosiglitazone. Similar results were obtained by using mouse primary mesenchymal stem cells (Supplementary Fig. 6). These data were also in agreement with the lower agonistic activity of $\mathbf{6 b}$ and $\mathbf{1 1}_{\mathbf{Z}}$ than rosiglitazone (Fig. 2B). Hence, the derivatives of sulindac may avoid the adverse effect of potent PPAR $\gamma$ agonists on osteoblastogenesis. Moreover, we did not observe apparent effect of $\mathbf{6 b}$ on HFD mouse weight (Supplementary Fig. 7).

### 2.6. Molecular docking study of $\mathbf{6 b}$ to the ligand-binding pocket (LBP) of PPAR $\gamma$

Molecular docking was utilized to uncover the binding modes of PPAR $\gamma$ with $\mathbf{6 b}$ ( $\mathbf{5 b}$ for comparison). Since the crystal structures of sulindac sulfide/PPAR $\gamma$-LBD (PDB ID: 4XUH) and rosiglitazone/ PPAR $\gamma$-LBD (PDB ID: 5YCP) have been reported [19,43], sulindac sulfide and rosiglitazone were selected as the reference binding modes. After molecular redocking to their corresponding binding pocket, the binding energies of sulindac sulfide and rosiglitazone were $-10.353 \mathrm{kcal} \mathrm{mol}^{-1}$ and $-9.422 \mathrm{kcal} \mathrm{mol}^{-1}$, respectively (Fig. 6A).

The 3 D structure of $\mathbf{5 b}$ and $\mathbf{6 b}$ were established based upon the chemical structure of sulindac sulfide and then docked into the binding pocket of PPAR $\gamma$ randomly according to the Glide algorithm and optimized parameters. To our surprised, both $\mathbf{5 b}$ and $\mathbf{6 b}$ took a similar binding pattern as rosiglitazone but not as sulindac sulfide (Fig. 6 and Supplementary Figs. 3 and 4) with a reasonable binding energy ( $-8.448 \mathrm{kcal} \mathrm{mol}^{-1}$ for $\mathbf{5 b}$ and $-8.724 \mathrm{kcal} \mathrm{mol}^{-1}$ for $\mathbf{6 b}$ ). The binding pocket analysis showed that PPAR $\gamma$-LBP could recruit two sulindac sulfides in the crystal complexes (PDB ID: 4XUH) (Fig. 6A). The introduction of the phenoxyl or benzyloxyl steric groups in the meta-position of sulindac scaffold would force $\mathbf{5 b}$ and

6b to assume different positions in the LBP in order to avoid steric clashes with protein residues and, in this case, unlike sulindac, only one molecule could occupy the LBP.

The carboxylic acid of sulindac sulfide (molecule 502) and thiazolidinedione moiety of rosiglitazone could form four hydrogen bonds with His449, Tyr473, His323 and Ser289 in the Y-shape pocket of PPAR $\gamma$-LBP (Fig. 6B). Interestingly, the carboxylic acid group of $\mathbf{5 b}$ and $\mathbf{6 b}$ could not afford the same interaction with the four corresponding amino acid residues. Instead, they formed two hydrogen bonds with amides of Ser342 and Glu343. Meanwhile, their phenoxyl or benzyloxyl groups could stay close to the aromatic residues Tyr473, His323 and Tyr327 to form only additional hydrophobic interaction and construct a $\pi-\pi$ stacking interaction towards His 449 with a reasonable distance ( $4.61 \AA$ for $\mathbf{6 b}$ and 4.65 Å for 5b, Supplementary Fig. 4B and 4C), without any classical hydrogen bonds in this pocket (Fig. 6B). 5b and $\mathbf{6 b}$ did not form any hydrogen bond with Tyr473, and thereby they could not stabilize the H12 of PPAR $\gamma$ effectively. This was in accordance with their property of weak agonists of PPAR $\gamma$. Furthermore, the benzene ring of benzylidene moiety of $\mathbf{5 b}$ and $\mathbf{6 b}$ overlapped nicely with the benzene ring of rosiglitazone that would have a contact with Cys285, Met364 and Leu330. Thus, 5b and 6b represented an interesting binding mode with PPAR $\gamma$-LBD, which was much different from sulindac sulfide.

### 2.7. Metabolic stability and pharmacokinetic study of $\mathbf{6 b}$

Metabolic stability of $\mathbf{6 b}$ was determined in rat liver microsome. As shown in Fig. 7A, 6b was stable in buffer, but decreased gradually over the microsome-incubation time period. It decreased by $40 \%$ after incubation with rat liver microsome for 90 min . The half-life ( $\mathrm{T}_{1 / 2}$ ) of $\mathbf{6 b}$ and the clearance rate in rat liver microsome are
A

| DMSO | Mineralized medium |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | DMSO | RGZ | $\mathbf{6 b}(\mu \mathrm{M})$ |  |  | $11_{Z}(\mu \mathrm{M})$ |  |  |
|  |  |  | 0.1 | 1 | 5 | 0.1 | 1 | 5 |



B


C


Mineralized medium

Fig. 5. Less osteoblastogenesis inhibition of $\mathbf{6 b}$ and $\mathbf{1 1}_{z}$. UMR- 106 cells were seeded into 96 -well plates. Cells were then incubated with the mineralized medium containing 10 mM of $\beta$-glycerophosphate disodium salt hydrate and $50 \mu \mathrm{M}$ of L -ascorbic acid with or without the compounds for 48 h . (A) Cells were fixed with $4 \%$ paraformaldehyde and stained with Alizarin Red S for 30 min . (B) Image J software was used to analyze and quantify the calcium deposition. (C) Activity of alkaline phosphatase was measure, and the values were normalized to total protein concentrations. RGZ, rosiglitazone; DM, DMSO; ALP, alkaline phosphatase. Data are presented as means $\pm$ SD. Statistical significance of differences of each condition vs. DMSO control was calculated by student's $t$-test. ${ }^{*} \mathrm{p}<0.05,{ }^{* *} \mathrm{p}<0.01,{ }^{* * *} \mathrm{p}<0.001, \mathrm{~ns}$ : p $>0.05$.


Fig. 6. Proposed binding mode of $\mathbf{6 b}$ in comparison with sulindac sulfide and rosiglitazone in the crystal structure of ligand binding pocket (LBP) of PPAR $\gamma$. (A) Overlapping of binding pose of sulindac sulfide (red for 502 and blue for 501 in $4 x u h$ ), rosiglitazone (yellow) and $\mathbf{6 b}$ (green) in PPAR $\gamma$-LBP. (B) Binding pose of $\mathbf{6 b}$ (green) in PPAR $\gamma$-LBP predicted by Glide SP.
117.45 min and $14.75 \mu \mathrm{~L} \mathrm{~min}{ }^{-1} \cdot \mathrm{mg}^{-1}$, respectively. We further performed pharmacokinetic study of $\mathbf{6 b}$ in rats. $\mathbf{6 b}$ was administered by oral absorption ( $30 \mathrm{mg} / \mathrm{kg}$ ) or intravenous injection ( 3 mg / kg ), and the plasma concentrations of $\mathbf{6 b}$ at different time points was measured (Fig. 7B and 7C). As shown in Table 2, $\mathbf{6 b}$ exhibited a good oral bioavailability in rats ( $\mathrm{F}=97 \%$ ). The maximal plasma concentration of $\mathbf{6 b}$ reached $80.25 \pm 8.66 \mathrm{mg} / \mathrm{L}$ at 1.5 h after oral administration and the elimination half-life in female rats was $4.75 \pm 0.84 \mathrm{~h}$.

## 3. Conclusion

Sulindac derivatives showed interesting PPAR $\gamma$-based structural-activity relationship. As to PPAR $\gamma$ binding and activation, the meta-position substitution in the benzylidene moiety was much more effective than the para-position substitution. Unlike E derivatives, $Z$ derivatives showed the PPAR $\gamma$ selectivity. Phenoxyl and benzyloxyl meta-substitutions endowed the derivatives with strong potency on regulation and binding of PPAR $\gamma$. Indene fluorine


Fig. 7. Pharmacokinetic profiles of $\mathbf{6 b}$. (A) Stability of $\mathbf{6 b}$ was examined using rat liver microsome. $\mathbf{6 b}(40 \mu \mathrm{M})$ was incubated with microsome for different time, and the percentage of $\mathbf{6 b}$ remaining amount to $\mathbf{6 b}$ amount at 0 min was determined. Data are presented as means $\pm$ SEM, $n=3$. (B, C) Plasma concentration-time profiles of $\mathbf{6 b}$ in Sprague-Dawley (SD) rats after intravenous administration at $3 \mathrm{mg} / \mathrm{kg}$ (B) or oral administration at $30 \mathrm{mg} / \mathrm{kg}$ (C). Also see Table 2.

Table 2
Pharmacokinetic parameters of $\mathbf{6 b}$ in rats.

| Parameters | Unit | Oral $(30 \mathrm{mg} / \mathrm{kg})$ | IV $(3 \mathrm{mg} / \mathrm{kg})$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{AUC}_{0-\mathrm{t}}$ | $\mathrm{mg} / \mathrm{L} * \mathrm{~h}$ | $379.69 \pm 98.32$ | $39.30 \pm 5.45$ |
| $\mathrm{AUC}_{0-\infty}$ | $\mathrm{mg} / \mathrm{L} * \mathrm{~h}$ | $444.53 \pm 136.60$ | $51.58 \pm 11.31$ |
| $\mathrm{~T}_{1 / 2 \mathrm{z}}$ | h | $4.75 \pm 0.84$ | $7.06 \pm 1.41$ |
| $\mathrm{~T}_{\max }$ | h | $1.5 \pm 0.71$ | - |
| $\mathrm{V}_{\mathrm{z}} / \mathrm{F} / \mathrm{L}$ | L kg | $0.47 \pm 0.06$ | $0.59 \pm 0.01$ |
| $\mathrm{CL}_{z} / \mathrm{F}$ | $\mathrm{L} / \mathrm{h} / \mathrm{kg}$ | $0.07 \pm 0.02$ | $0.06 \pm 0.01$ |
| $\mathrm{C}_{\max }$ | $\mathrm{mg} / \mathrm{L}$ | $80.25 \pm 8.66$ | $14.29 \pm 2.67$ |

$\mathrm{AUC}_{0-\mathrm{t}}$, area under the plasma concentration-time curve values of $\mathbf{6 b}$ during 12 h ; $A U C_{0-\infty}$, able to cover the total AUC; $T_{1 / 2 z}$, terminal elimination half-life; $T_{\text {max }}$, the time to reach a peak concentration; $\mathrm{V}_{\mathrm{z}} / \mathrm{F}$, distribution volume; $\mathrm{CL}_{z} / \mathrm{F}$, clearance; $\mathrm{C}_{\text {max }}$, peak plasma concentration. Data are presented as mean $\pm \mathrm{SD}, \mathrm{n}=2$. Data was calculated by software DAS 3.2.8. Also see Fig. 7B and 7C.
was vital for the derivatives binding to PPAR $\gamma$ and regulating PPAR $\gamma$ transactivation. Hence, our results provide some useful guidance for designing sulindac derivatives targeting PPAR $\gamma$.

6b exhibited strong anti-diabetic effect through improving insulin resistance. Compared with clinical drug rosiglitazone, the derivatives had less PPAR $\gamma$ transactivating activity, which may reduce the severe side effects from the potent induction of PPAR $\gamma$ targeted genes. Indeed, $\mathbf{6 b}$ and $\mathbf{1 1}_{\boldsymbol{Z}}$ had no obvious effect on inhibiting osteoblastic differentiation and had relatively low effect on adipogenic induction. Thus, the meta-substitution and $Z$ configuration derivatives of sulindac may be the next generation of PPAR $\gamma$-targeting drugs for diabetes.

## 4. Experimental

### 4.1. Chemistry

General synthesis of indene Rec1 and Rec2.
Zn ( $320 \mathrm{mg}, 5 \mathrm{mmol}$ ) was placed in a 50 mL two-necked round bottom flask which was evacuated and flushed with nitrogen and then 12 mL THF was added as solvent. Ethyl bromoacetate ( 350 mg , 2.1 mmol ) was added dropwise, followed by a catalytic amount of $\mathrm{I}_{2}$ to initiate the reaction. Different indenone ( $164 \mathrm{mg}, 1 \mathrm{mmol}$ ) was added dropwise, and after 30 min , the reaction was refluxed for 4 h . After quenching reaction by $\mathrm{HCl}(10 \%, 10 \mathrm{~mL})$, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$ and washed with water $(3 \times 10 \mathrm{~mL})$. The combined organic layers were dried and concentrated under vacuum. To the residue, $\mathrm{NaOH}(1 \mathrm{~N}, 10 \mathrm{~mL})$ and $\mathrm{MeOH}(6.7 \mathrm{~mL})$ was added. After stirring overnight at room temperature, the mixture was quenched by $\mathrm{HCl}(20 \%, 2.5 \mathrm{~mL})$. The aqueous solution was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ). The combined extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (ethyl acetate: $\mathrm{PE}=1: 10$ ) to afford compound Rec1 and Rec2.

General procedure for the synthesis of sulindac derivatives. Using MeONa as the base catalyst and MeOH as the solvent, indene could react with different substituted aromatic aldehydes to construct structurally diverse sulindac analogues with moderate yields for the following SAR study (see Scheme 1).
4.1.1. (Z)-2-(5-fluoro-1-(4-fluorobenzylidene)-2-methyl-1H-inden-3-yl) acetic acid (3a)

Yellow solid; M.p. $184-187{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , DMSO-d $\mathrm{d}_{6}$ ) $\delta=12.42$ (br. s., 1H), 7.58 (dd, $J=5.7,8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.31-7.35$ (m, 3H), 7.17 (dd, $J=5.1,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.02$ (dd, $J=2.4,9.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.73 (dt, $J=2.4,8.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 2 \mathrm{H}), 2.15(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 151 MHz , DMSO- $d_{6}$ ) $\delta=172.06,163.45(\mathrm{~d}, J=68.2 \mathrm{~Hz}), 161.86(\mathrm{~d}, J=71.5 \mathrm{~Hz})$, 147.41 (d, $J=8.8 \mathrm{~Hz}$ ), 140.09, 138.36, 132.93, 132.91, 132.53, 131.80 (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{C}), 130.31,129.97,123.48(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 116.11(\mathrm{~d}$,
$J=22.0 \mathrm{~Hz}), 110.79(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 106.40(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 31.55$, 10.70. HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~F}_{2} \mathrm{NaO}_{2}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$: 335.0854; found: 335.0857. HPLC purity: $98.54 \%$.

### 4.1.2. (Z)-2-(5-fluoro-1-(3-fluorobenzylidene)-2-methyl-1H-inden-3-yl) acetic acid (3b)

Yellow solid; yield 74\%; M.p. $182-183{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=7.41-7.35(\mathrm{~m}, 1 \mathrm{H}), 7.26(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.17(\mathrm{~d}$, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.12(\mathrm{~s}, 1 \mathrm{H}), 7.07(\mathrm{t}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.57(\mathrm{t}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=176.64,163.28(\mathrm{~d}, J=246.5 \mathrm{~Hz}, 1 \mathrm{C}), 162.78(\mathrm{~d}$, $J=246.5 \mathrm{~Hz}, 1 \mathrm{C}), 146.35(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 141.07,138.73(\mathrm{~d}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{C}), 138.67,130.74,130.13(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{C}), 129.50(\mathrm{~d}$, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 128.84,125.00(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 123.87(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, 1C), 116.03 ( $\mathrm{d}, J=22.0 \mathrm{~Hz}, 1 \mathrm{C}$ ), 115.11 (d, $J=19.8 \mathrm{~Hz}, 1 \mathrm{C}), 110.87$ (d, $J=23.1 \mathrm{~Hz}, 1 \mathrm{C}), 105.96(\mathrm{~d}, J=23.1 \mathrm{~Hz}, 1 \mathrm{C}), 31.42,10.52$; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~F}_{2} \mathrm{NaO}_{2}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$: 335.0854; found: 335.0849. HPLC purity: $98.41 \%$.
4.1.3. (Z)-2-(5-fluoro-1-(3-isopropylbenzylidene)-2-methyl-1H-inden-3-yl) acetic acid (4b)

Yellow solid; yield $77 \%$; M.p. $130-131{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=7.40(\mathrm{~s}, 1 \mathrm{H}), 7.35(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.28(\mathrm{~m}, 2 \mathrm{H})$, $7.24-7.21(\mathrm{~m}, 2 \mathrm{H}), 6.88(\mathrm{dd}, J=2.4,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dt}, J=2.4$, $8.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.59 (s, 2H), 2.93 (spt, J=6.9 Hz, 1H), 2.21 (s, 3H), 1.27 $(\mathrm{d}, J=7.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=176.23,163.11(\mathrm{~d}$, $J=245.4 \mathrm{~Hz}, 1 \mathrm{C}), 149.01,146.22(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 140.02,138.88$, 136.36, 131.22, 129.97 (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), 129.85 (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), $128.49,127.57,126.70,126.59,123.88$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 110.56$ (d, $J=23.1 \mathrm{~Hz}, 1 \mathrm{C}), 105.68(\mathrm{~d}, J=24.2 \mathrm{~Hz}, 1 \mathrm{C}), 34.16,31.41,23.94(\mathrm{~s}, 2 \mathrm{C})$, 10.59. HRMS (ESI) calcd for $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{FNaO}_{2}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]$: 359.1417; found: 359.1418. HPLC purity: 99.56\%.
4.1.4. (Z)-2-(5-fluoro-2-methyl-1-(4-phenoxybenzylidene)-1H-inden-3-yl) acetic acid (5a)

Yellow solid; yield $65 \%$; M.p. $154-155{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=7.47(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7.41(\mathrm{~m}, 3 \mathrm{H}), 7.14-7.18(\mathrm{~m}$, 2 H ), 7.08-7.12 (m, 2H), 7.04 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.88(\mathrm{dd}, J=8.9$, $2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{td}, J=8.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.58(\mathrm{~s}, 2 \mathrm{H}), 2.20(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ $\operatorname{NMR}\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=176.3,163.1(\mathrm{~d}, J=247.6 \mathrm{~Hz}), 157.7,156.5$, 146.2 (d, $J=8.8 \mathrm{~Hz}$ ), 139.8, 138.9, 131.1 (2C), 130.2, 129.9 (2C), 129.9 $(\mathrm{d}, J=2.2 \mathrm{~Hz}), 129.7(\mathrm{~d}, J=2.2 \mathrm{~Hz}), 123.9,123.7(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 119.5$ (2C), 118.3 (2C), 110.7 ( $\mathrm{d}, J=23.1 \mathrm{~Hz}$ ), $105.8(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 31.4$, 29.7, 10.6; HRMS (ESI) calcd for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{FNaO}_{3}^{+}$[ $\mathrm{M}+\mathrm{Na}^{+}$]: 409.1210; found: 409.1203.
4.1.5. (Z)-2-(5-fluoro-2-methyl-1-(3-phenoxybenzylidene)-1H-inden-3-yl) acetic acid (5b)

Yellow solid; yield 53\%; M.p. $110-112{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=7.38(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.26(\mathrm{dd}$, $J=5.2,8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.22(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H}), 7.12-7.08(\mathrm{~m}$, $2 \mathrm{H}), 7.06-7.02(\mathrm{~m}, 3 \mathrm{H}), 6.85(\mathrm{dd}, J=2.4,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{dt}, J=2.5$, $8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.55(\mathrm{~s}, 2 \mathrm{H}), 2.16(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=176.62,163.23(\mathrm{~d}, J=246.5 \mathrm{~Hz}, 1 \mathrm{C}), 157.67,156.92,146.30(\mathrm{~d}$, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 140.60,138.82,138.28,130.43,129.94,129.92(\mathrm{~s}, 2 \mathrm{C})$, $129.82,129.64(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 124.15,123.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{C})$, 123.70, 119.28 ( $\mathrm{s}, 2 \mathrm{C}$ ), $119.24,118.72,110.76$ ( $\mathrm{d}, J=22.0 \mathrm{~Hz}, 1 \mathrm{C}$ ), 105.87 (d, $J=24.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), 31.45, 10.59. HRMS (ESI)calcd for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{FNaO}_{3}^{+}$[ $\mathrm{M}+\mathrm{Na}^{+}$]: 409.1210; found: 409.1205. HPLC purity: 98.47\%.
4.1.6. (Z)-2-(1-(4-(benzyloxy)benzylidene)-5-fluoro-2-methyl-1H-inden-3-yl) acetic acid (6a)

Yellow solid; yield $35 \%$; M.p. $135-136{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , DMSO- $\mathrm{d}_{6}$ ) $\delta=7.51(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.49(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.44-$
$7.40(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{~s}, 1 \mathrm{H}), 7.14(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H})$, 7.01 (dd, $J=2.6,9.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{dt}, J=2.4,8.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{~s}, 2 \mathrm{H})$, 3.57 (s, 2H), $2.15(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=175.51$, $163.02(\mathrm{~d}, J=246.5 \mathrm{~Hz}, 1 \mathrm{C}), 158.93,146.16(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 139.30$, 138.95, 136.70, 130.98 (s, 2C), 130.71, 129.84 (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), 129.61 ( $\mathrm{d}, \mathrm{J}=2.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), 128.98, 128.67 ( $\mathrm{s}, 2 \mathrm{C}$ ), 128.13, 127.56 ( $\mathrm{s}, 2 \mathrm{C}$ ), $123.61(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 114.84(\mathrm{~s}, 2 \mathrm{C}), 110.56(\mathrm{~d}, J=23.1 \mathrm{~Hz}, 1 \mathrm{C})$, $105.62(\mathrm{~d}, J=23.1 \mathrm{~Hz}, 1 \mathrm{C}), 70.12,31.30,10.61$. HRMS (ESI) calcd for $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{FNaO}_{3}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]: 423.1367$; found: 423.1359.
4.1.7. (Z)-2-(1-(3-(benzyloxy)benzylidene)-5-fluoro-2-methyl-1H-inden-3-yl) acetic acid (6b)

Yellow solid; yield $73 \%$; M.p. $169-171{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $(600 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta=7.40-7.44(\mathrm{~m}, 2 \mathrm{H}), 7.37(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.30-7.35(\mathrm{~m}$, 2H), 7.22-7.28 (m, 1H), 7.17 (s, 1H), 7.06-7.12 (m, 2H), 6.99 (dd, $J=8.3,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{dd}, J=8.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{td}, J=8.8$, $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.06(\mathrm{~s}, 2 \mathrm{H}) 3.57$ (s, 2H), 2.19 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( 151 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=176.6,163.1(\mathrm{~d}, J=246.5 \mathrm{~Hz}), 158.8,146.2(\mathrm{~d}, J=8.8 \mathrm{~Hz})$, $140.4,138.8,137.9,136.8,130.4,130.2$ (d, $J=2.2 \mathrm{~Hz}), 129.7$ (d, $J=3.3 \mathrm{~Hz}), 129.7,128.6(2 \mathrm{C}), 128.1,127.5(2 \mathrm{C}), 124.0(\mathrm{~d}, J=8.8 \mathrm{~Hz})$, $122.0,115.2,115.1,110.8(\mathrm{~d}, J=22.0 \mathrm{~Hz}), 105.8(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 70.0$, 31.4, 10.6; HPLC purity: 99.81\%.
4.1.8. (Z)-2-(1-(3-bromobenzylidene)-5-fluoro-2-methyl-1H-inden-3-yl) acetic acid (7)

Yellow solid; yield $70 \%$; M.p. $163-163{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $(600 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta=7.64(\mathrm{~s}, 1 \mathrm{H}), 7.51(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.42(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.28-7.33(\mathrm{~m}, 1 \mathrm{H}), 7.17$ (dd, $J=8.4,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~s}, 1 \mathrm{H}), 6.88$ (dd, $J=8.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{td}, J=8.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{~s}, 2 \mathrm{H}), 2.19(\mathrm{~s}$, $3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=175.6,163.3(\mathrm{~d}, J=246.5 \mathrm{~Hz}$ ), $146.4(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 141.2,138.7,138.6,132.0,131.1,130.9,130.1$, $129.5,128.5,127.8,123.8(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 122.6,110.9(\mathrm{~d}, J=22.0 \mathrm{~Hz})$, $106.0(\mathrm{~d}, J=24.2 \mathrm{~Hz}), 31.3,10.5$; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{BrFNaO}_{2}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]: 395.0053$; found: 395.0047. HPLC purity: 98.72\%.
4.1.9. (Z)-2-(1-(3-chlorobenzylidene)-5-fluoro-2-methyl-1H-inden-3-yl) acetic acid (8)

Yellow solid; yield $75 \%$; M.p. $162-163{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $(600 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta=7.47(\mathrm{~s}, 1 \mathrm{H}), 7.33-7.39(\mathrm{~m}, 3 \mathrm{H}), 7.16(\mathrm{dd}, J=8.3,5.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.11(\mathrm{~s}, 1 \mathrm{H}), 6.88(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.58(\mathrm{td}, J=8.8,2.1 \mathrm{~Hz}$, $1 \mathrm{H}), 3.58(\mathrm{~s}, 2 \mathrm{H}), 2.19(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=175.8$, $163.3(\mathrm{~d}, J=244.3 \mathrm{~Hz}), 146.4(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 141.2,138.6,138.4,134.5$, $130.8,129.8,129.5(\mathrm{~d}, J=2.2 \mathrm{~Hz}), 129.1,128.6,128.2,127.4$, 123.8 ( d , $J=8.8 \mathrm{~Hz}), 110.9(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 106.0(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 31.3,10.5$; HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{ClFNaO}_{2}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]: 351.0559$; found: 351.0553. HPLC purity: $99.32 \%$.
4.1.10. (Z)-2-(5-fluoro-1-(3-hydroxybenzylidene)-2-methyl-1H-inden-3-yl) acetic acid (9)

Yellow solid; yield $45 \%$; M.p. $172-177{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , DMSO- $d_{6}$ ) $\delta=9.64$ (br. s., 1H), 7.30-7.27 (m, 3H), 7.01 (dd, $J=2.5$, $9.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~s}, 1 \mathrm{H}), 6.82(\mathrm{dd}, J=2.3$, $8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{dt}, J=2.4,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 2 \mathrm{H}), 2.14(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( 151 MHz, DMSO- $d_{6}$ ) $\delta=172.11,162.80(\mathrm{~d}, J=243.2 \mathrm{~Hz}, 1 \mathrm{C})$, 157.87, 147.29 (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 139.75,138.39,137.81,132.27$ (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 131.60,130.15,130.08$ (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 123.80$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 120.22,116.04,115.87,110.66(\mathrm{~d}, J=23.1 \mathrm{~Hz}, 1 \mathrm{C})$, 106.24 (d, $J=23.1 \mathrm{~Hz}, 1 \mathrm{C}$ ), $31.55,10.71$. HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{16} \mathrm{FO}_{3}^{+}\left[\mathrm{M}+\mathrm{H}^{+}\right]$: 311.1078 ; found: 311.1078 . HPLC purity: $97.34 \%$.

### 4.1.11. (Z)-2-(5-fluoro-1-(3-methoxybenzylidene)-2-methyl-1H-

 inden-3-yl) acetic acid (10)Yellow solid; yield $43 \%$; M.p. $131-133{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , DMSO- $d_{6}$ ) $\delta=7.41(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.34(\mathrm{~s}, 1 \mathrm{H}), 7.25(\mathrm{dd}, J=5.3$,
$8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~d}, \mathrm{~J}=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.08(\mathrm{~s}, 1 \mathrm{H}), 7.03-6.97(\mathrm{~m}, 2 \mathrm{H})$, 6.73 (dd, $J=2.4,8.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.78 (s, 3H), 3.58 (s, 2H), 2.15 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( 151 MHz, DMSO- $d_{6}$ ) $\delta=172.10,162.85(\mathrm{~d}, J=243.2 \mathrm{~Hz}$, 1C), $159.70,147.37$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 140.02,138.35,137.93,132.48$ $(\mathrm{d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 131.23,130.20,130.04(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 123.74(\mathrm{~d}$, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}), 121.80,114.70,114.59,110.71$ (d, $J=23.1 \mathrm{~Hz}, 1 \mathrm{C})$, 106.34 (d, $J=24.2 \mathrm{~Hz}, 1 \mathrm{C}$ ), $55.60,31.56,10.70$. HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{FNaO}_{3}^{+}\left[\mathrm{M}+\mathrm{Na}^{+}\right]:$347.1054; found: 347.1056.
4.1.12. (Z)-2-(5-fluoro-2-methyl-1-(3-methylbenzylidene)-1H-inden-3-yl) acetic acid (11z)

Yellow solid; yield $63 \%$; M.p. $142-144{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=7.29(\mathrm{~m}, 3 \mathrm{H}), 7.24-7.27(\mathrm{~m}, 1 \mathrm{H}), 7.15-7.19(\mathrm{~m}, 2 \mathrm{H}), 6.86$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{t}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.56(\mathrm{~s}, 2 \mathrm{H}), 2.37(\mathrm{~s}$, 3H), 2.18 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=177.0,163.1$ (d, $J=249.8 \mathrm{~Hz}), 146.2(\mathrm{~d}, J=8.8 \mathrm{~Hz}), 140.1,138.9,138.2,136.5,131.0$, $130.0,129.9,129.0,128.5(\mathrm{~d}, J=4.4 \mathrm{~Hz}), 128.4,126.4,123.9$ (d, $J=8.8 \mathrm{~Hz}), 110.7(\mathrm{~d}, J=22.0 \mathrm{~Hz}), 105.7(\mathrm{~d}, J=23.1 \mathrm{~Hz}), 31.5,21.4$, 10.6; HRMS (ESI) calcd for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{FNaO}_{2}^{+}$[ $\mathrm{M}+\mathrm{Na}^{+}$]: 331.1105; found: 331.1100. HPLC purity: $97.86 \%$.
4.1.13. (E)-2-(5-fluoro-1-(3-methylbenzylidene)-1H-inden-3-yl) acetic acid (11E)

Yellow solid; ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.58(\mathrm{dd}, J=5.0$, $8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.38-7.34(\mathrm{~m}, 3 \mathrm{H}), 7.30(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{~d}$, $J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{~s}, 1 \mathrm{H}), 7.01(\mathrm{dd}, J=2.2,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dt}$, $J=2.2,8.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.66(\mathrm{~s}, 2 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 151 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=176.37,163.00(\mathrm{~d}, J=246.5 \mathrm{~Hz}, 1 \mathrm{C}), 143.11(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, 1C), 138.42, 137.93 (d, $J=2.2 \mathrm{~Hz}, 1 \mathrm{C}), 137.60,136.58,133.58$ (d, $J=3.3 \mathrm{~Hz}, 1 \mathrm{C}), 130.84,129.45,129.24,128.69,127.32,127.05,120.08$ (d, $J=8.8 \mathrm{~Hz}, 1 \mathrm{C}$ ), 112.06 (d, $J=23.1 \mathrm{~Hz}, 1 \mathrm{C}$ ), 106.64 (d, $J=24.2 \mathrm{~Hz}$, 1C), 33.86, 21.47. HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{FNaO}_{2}+\left[\mathrm{M}+\mathrm{Na}^{+}\right]$: 317.0948; found: 317.0945.
4.1.14. (Z)-2-(2-methyl-1-(3-methylbenzylidene)-1H-inden-3-yl) acetic acid (12z)

Yellow solid; yield $21 \%$; ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.36(\mathrm{~d}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{~s}, 1 \mathrm{H}), 7.33-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.20(\mathrm{~s}, 1 \mathrm{H}), 7.18(\mathrm{~d}$, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.16(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{dt}, J=1.3,7.5 \mathrm{~Hz}, 1 \mathrm{H})$, 3.61 (s, 2H), $2.38(\mathrm{~s}, 3 \mathrm{H}), 2.19(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=176.69,143.88,141.16,138.05,136.80,136.74,134.13,130.99$, $130.86,129.93,128.83,128.33,127.83,126.38,124.61,122.87,117.99$, 31.46, 21.40, 10.48 .
4.1.15. (E)-2-(1-(3-methylbenzylidene)-1H-inden-3-yl) acetic acid (12E)

Yellow solid, ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=77.69(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.44(\mathrm{~s}, 1 \mathrm{H}), 7.39-7.42(\mathrm{~m}, 2 \mathrm{H}), 7.31-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.29(\mathrm{~d}$, $J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.27(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.00(\mathrm{~s}$, $1 \mathrm{H}), 3.71(\mathrm{~s}, 2 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=176.14$, $141.29,139.00,138.64,138.33,137.95,136.81,130.86,129.26,128.71$, 128.64, 127.43, 127.35, 125.57, 125.25, 119.10, 119.04, 33.89, 21.47. HRMS (ESI) calcd for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{O}_{2}^{-}$[ $\mathrm{M}^{-}$]: 275.1078; found: 275.1081.

### 4.2. Materials for biological assays

Insulin (19278), dexamethasone (D2915), IBMX (I5879), $\beta$ Glycerophosphate disodium salt hydrate (G9422), l-ascorbic acid (A4544), D-(+)-Glucose (G7021), WY-14643 (C7081), GW501516 (43732), rosiglitazone (R2408), Oil Red O (O0625) Solutol HS 15 (70142-34-6), and Alizarin Red S (A5533) were purchased from Sigma-Aldrich; Alkaline phosphatase Kit (PO321) and COX-2 enzyme immune assay kit (S0168) were purchased from Beyotime Biotechnology (China). IPTG Dioxane Free (A600168) and Imidazole (A600277) were purchased from Sangon Biotech (China).

Glycerol was purchased from Sinopharm Chemical Reagent Co. Ltd. PPRE-Luc, pG5-luc, pcmv-myc-RXR $\alpha$, pcmv-myc-PPAR $\gamma$, pBindPPAR $\gamma$ LBD, pBind-PPAR $\alpha$ LBD, pBind-PPAR $\beta$ LBD and pCMV-renilla plasmids were cloned by our lab. Phospho-Akt Ser473 (\#4060) and Phospho-Akt Thr308 (\#13038) were purchased from cell Signaling Technology. AKT1/2/3 (H-136) antibody was purchased from Santa Cruz. $\beta$-actin (A5441) antibody was purchased from Sigma-Aldrich. One touch Ultra Easy and Blood glucose test strips were purchased from LifeScan Inc. TRIzol LS Reagent (10296-010) was purchased from life technology. FastStart Universal SYBR Green Master (Rox) (4913850001) was purchased from Roche. Phosphate buffer (451201) and rat liver microsome (452511) were purchased from Corning.

### 4.3. Molecular docking and structural comparison

The chemical structures of $\mathbf{5 b}$ and $\mathbf{6 b}$ were constructed based upon the 502 molecule (Sulindac sulfide) in protein complex (PDBID: 4XUH). 64 conformations of $\mathbf{5 b}$ were generated by Confgen module in Schrodinger suite and then docked into the PPAR $\gamma$ binding site by Glide-HTVS and SP method, respectively. The parameters were set as that: Gridcenter on 502 molecule and GridBox was $10 \AA \AA .128$ conformations of $\mathbf{6 b}$ were generated and similar docking was done. Two sulindac sulfide in protein complex (PDBID: 4 XUH ) named 502 and 501 were selected as reference compounds. Also, rosiglitazone in protein complex (PDBID: 5YCP) was chosen for comparison. All the figures were generated by Schrodinger suite or PyMol software.

### 4.4. Mammalian one-hybrid assay (pBind system reporter assay)

HEK293T cells were co-transfected with pG5-luc (Promega) and pBind-PPAR $\gamma /$ LBD, pBind-PPAR $\alpha /$ LBD, or pBind-PPAR $\beta /$ LBD plasmid at cell density of $60-80 \%$. After 24 h , cells were treated with the compounds or the PPAR positive agonists (rosiglitazone, WY-14643 or GW501516) for 12 h . Cells were lysed by Promega passive lysis buffer, and the fluorescence values of luciferase and Renilla were measured by Promega Microplate Reader. Renilla luciferase values were normalized to firefly luciferase activity to obtain the relative luciferase activity for plotting.

### 4.5. PPRE-luciferase reporter assay

Cos-7 cells seeded at 24-well plates were co-transfected with PPRE-Luc (addgene), pCMV-myc-RXR $\alpha$, pCMV-myc-PPAR $\gamma$ and pCMV-renilla plasmids. After 24 h , cells were treated with the compounds for 12 h . Cells were then collected and lysed with Promega passive lysis buffer. The fluorescence values of luciferase and renilla were measured by Promega Microplate Reader. Renilla luciferase values were normalized to firefly luciferase activity to obtain the relative luciferase activity for plotting.

### 4.6. PPAR $\gamma$-LBD protein purification

The DNA fragment encoding amino acid sequence (237-504) of PPAR $\gamma / \mathrm{LBD}$ was cloned into the expression vector $\mathrm{pET}-15 \mathrm{~b}$. Recombinant plasmid was transformed into E. coli (BL21 DE3 strain) and protein expression was induced by 1 mM isopropyl $\beta$-D-1thiogalactopyranoside (IPTG) at $16{ }^{\circ} \mathrm{C}$ for 8 h . PPAR $\gamma$-LBD protein was purified by nickel column using Akta avant instrument. The running buffer for purification was buffer A ( 25 mM Tris, 150 mM $\mathrm{NaCl}, 25 \mathrm{mM}$ imidazole, $10 \%$ glycerol, [ pH 7.5 ]). The elution buffer was buffer B ( 25 mM Tris, 500 mM imidazole, $10 \%$ glycerol, [ pH 7.5]).

### 4.7. Fluorescence titration assay

Fluorescence titration was implemented on CARY ECLIPS spectra-fluorophotometer (Varian, USA). Data collection was ranged from 290 to 500 nm upon excitation at 282 nm for PPAR $\gamma$. Excitation and emission bandwidths were all 5 nm . A 3 cm quartz cell was used for measurements. Reactions were all initiated by stepwise adding a little aliquot volume of ligand into 3.0 mL titrand at $25^{\circ} \mathrm{C}$. In this study, PPAR $\gamma$ was prepared in PBS buffer to give working concentrations of $3 \mu \mathrm{M}$ to provide an optimal fluorescence peak. Solution of ligand was 2000 times higher than PPAR's to ensure the final ratio of ligand to titrand as " 5 ". Ligands were injected by $1: 1000$ into solution of PPAR $\gamma$ for each titration, blended after each injection, and the three times numerical reading in succession was required to guarantee the system stabilization. Furthermore, to be a control, PBS buffer with isometric DMSO is introduced to duplicate the whole process as ligand, to exclude the nonspecific quenching. Data fitting was executed on OriginPro 2016 (Origin, Inc), and the one site binding analysis for spectrum titration experiment according to literature was employed here to determine the dissociation constant ( $\mathrm{K}_{\mathrm{D}}$ ) [44].

### 4.8. 3T3-L1 adipocyte formation assay

3T3-L1 cell seeded at 24 -well plate were cultured for 2 days after confluence. Culture medium was replaced with fresh medium containing differentiation agent (5 mM 3-Isobutyl-1methylxanthine, $1 \mu \mathrm{M}$ dexamethasone and $10 \mu \mathrm{~g} / \mathrm{ml}$ Insulin) and the testing compounds. After 48 h , cells were treated with fresh medium containing $10 \mu \mathrm{~g} / \mathrm{ml}$ insulin for another 48 h and then a fresh medium for 72 h . Lipid droplets and nuclei were stained with oil red and hematoxylin, respectively [45,46].

### 4.9. Glucose tolerance test

C57/B6 mice were fed with high-fat diet for 3 months to make the concentration of fasting blood glucose more than 11.1 mM . The mice were then fed with compounds $\mathbf{6 b}(40 \mathrm{mg} / \mathrm{kg})$ or rosiglitazone ( $4 \mathrm{mg} / \mathrm{kg}$ ) for another 1 month. The glucose tolerance test was carried out according to the reported literature. All experiments involving animals followed the approved protocols by the Laboratory Animal Center in Xiamen University.

### 4.10. Mineralization of UMR106 cells

UMR106 cells were seeded into 96 -well plates for 48 h . Cells were then induced with mineralized medium ( 10 mM of $\beta$-Glycerophosphate disodium salt hydrate and $50 \mu \mathrm{M}$ of L -ascorbic acid) in the presence or absence of the compounds. After 48 h , cells were fixed with $4 \%$ paraformaldehyde and washed with PBS twice, and then stained with Alizarin Red S for 30 min . Analysis of calcium deposition was performed by software Image J [42].

### 4.11. Osteoblastogenesis assay using mesenchymal stem cells

Mesenchymal stem cells were obtained from mouse bone marrow. Cells were cultured in $\alpha$-MEM medium supplemented with $10 \%$ FBS, 2 mM glutamine, $100 \mathrm{U} / \mathrm{ml}$ penicillin, and $100 \mathrm{mg} / \mathrm{ml}$ streptomycin. Nonadherent cells were removed after 24 h , and adherent cells were maintained with medium replenishment every 3 days. MSCs were then induced with mineralized medium ( 10 mM of $\beta$-glycerophosphate and $200 \mu \mathrm{M}$ of L -ascorbic acid). After 12 days, cells were fixed with $4 \%$ paraformaldehyde and washed with PBS twice, and then stained with Alizarin Red S for 3 min .

### 4.12. Western blotting assay

Cell lysates were boiled in SDS sample buffer. Samples were electrophoresed on $10 \%$ SDS-PAGE gels and then transferred to polyvinylidene difluoride membranes (Millipore). The membranes were blocked with $5 \%$ skimmed milk in TBST buffer ( 50 mM TrisHCl [pH7.4], 150 mM NaCl , and $0.1 \%$ Tween 20) for 1 h , then incubated with primary antibodies overnight at $4{ }^{\circ} \mathrm{C}$ and secondary antibodies for 1 h at room temperature. Proteins were detected using the ECL system (Thermo).

### 4.13. Quantitative Real-Time PCR analysis

Total RNA was isolated from 3T3-L1 adipocytes using TRIzol reagent and $3 \mu \mathrm{~g}$ of RNA was reverse transcribed to cDNA. Quantitative PCR analysis was performed using FastStart Universal SYBR Green Master. The relative amount of mRNA was calculated and normalized to GAPDH. The sequences of the primers were as follows: aP2-F: $5^{\prime-}$ AAGGTGAAGAGCATCATAACCCT-3'; aP2-R: 5'- TCACGCCTTTCATAACACATTCC -3'; PPAR $\gamma-\mathrm{F}$ : 5'- GCATGGTGCCTTCGCTGA-3'; PPAR $\gamma-\mathrm{R}$ : 5'- TGGCATCTCTGTGTCAACCATG-3'; GAPDH-F: 5'- TGCACCAC-CAACTGCTTAGC-3'; GAPDH-R: 5'- GGATGCAGGGATGATGTTCT - ${ }^{\prime}$ '.

### 4.14. Alkaline phosphatase activity assay

Cells were lysed in lysis buffer ( $150 \mathrm{mM} \mathrm{NaCl}, 1 \% \mathrm{NP}-40,50 \mathrm{mM}$ Tris [pH 8.0]). Lysates were mixed with chromogenic substrate for 10 min. ALP activity was detected by Promega Microplate Reader, and the values were normalized to protein concentrations measured by Pierce BCA Protein Assay Kit.

### 4.15. In vitro cyclooxygenase (COX) inhibition assay

The probe is catalyzed by COX-2 to the fluorescence substance (ex560/em590). There is linear correlation between the COX-2 activity and the fluorescence intensity. In brief, COX-2 cofactor, COX-2 protein and compounds or DMSO were added to the COX-2 assay buffer for incubation at $37{ }^{\circ} \mathrm{C}$ for 10 min . The probe was added to the reaction buffer and mixed well. Arachidonic acid was then added to the reaction buffer and the fluorescence intensity was measured at the Tecan Spark Microplate Reader.

### 4.16. Stability of $\mathbf{6 b}$ in rat liver microsome

6b in DMSO ( $4 \mu \mathrm{~L}, 0.4 \mathrm{mM}$ ) was added to $396 \mu \mathrm{~L}$ microsome suspension. The resulting mixture was incubated at $37^{\circ} \mathrm{C}$, and the catalytic reactions were then quenched at $0 \mathrm{~min}, 5 \mathrm{~min} 15 \mathrm{~min}, 30$ min, 45 min , or 90 min by adding $1,200 \mu \mathrm{~L}$ of cold MeOH . The mixture was centrifuged at $13,000 \mathrm{rpm}$ for 10 min and the supernatants were transferred to a glass tube and dried over nitrogen. The dried residue was reconstituted with $70 \% \mathrm{MeOH}$ and analyzed with Thermo Fisher Scientific Q Exactive Orbitrap Mass Spectrometers (LC-MS). Microsome suspension: $364 \mu \mathrm{~L}$ phosphate buffer (Corning, 451201) $+8 \mu \mathrm{~L}$ rat liver microsome (Corning, $452511)+4 \mu \mathrm{~L}$ solution $\mathrm{A}($ Corning, 451220$)+4 \mu \mathrm{~L}$ solution B (Corning, 451200).

### 4.17. In vivo pharmacokinetic study of $\mathbf{6 b}$

All animal experiments were approved by the Animal Care and Use Committee of Xiamen University, in accordance with the animal care and use guidelines. 6b was dissolved in phosphatebuffered saline (pH 7.4) containing 5\% (w/v) Solutol HS 15 (Sigma, 70142-34-6). 6b solution was administered orally at the dosage of $30 \mathrm{mg} / \mathrm{kg}$ or intravenously at the dosage of $3 \mathrm{mg} / \mathrm{kg}$ to Sprague-

Dawley (SD) Rats ( $180-220 \mathrm{~g}$ ) after overnight fasting. Blood samples ( $400 \mu \mathrm{~L}$ ) were collected at $0,0.25,0.5,1,2,3,4,6,8,10$ and 12 h after administration of $\mathbf{6 b}$. The blood samples were centrifuged to obtain plasma at $4^{\circ} \mathrm{C}$ and at $4,000 \mathrm{~g}$. These plasma samples ( $100 \mu \mathrm{~L}$ ) were used for the quantification of $\mathbf{6 b}$ concentration by Thermo Fisher Scientific Ultimate 3000HPLC and Orbitrap.

### 4.18. Statistical analysis

The data are presented as mean $\pm$ SD or mean $\pm$ SEM and represent at least two independent experiments. Two-tailed unpaired Student's $t$-test or Two-way ANOVA were used for statistical analysis using the GraphPad Prism 5.0 software. For all statistical analysis: ${ }^{*} \mathrm{p}<0.05,{ }^{* *} \mathrm{p}<0.01,{ }^{* * *} \mathrm{p}<0.001,{ }^{* * * *} \mathrm{p}<0.0001$ and ns p $>0.05$.

## Author contributions

Organic synthesis, W.Z., Z.Y., L.Y.; NMR and HPLC measurement and spectra interpretation, Z.Z., W.Z., Z.Y., Q.Y., H.Y.; molecular docking and structural comparison, Z.Z; fluorescence titration assay, F.H., J-j.C.; in vitro and in vivo evaluation of biological activities, F.H., Y.L.; metabolic stability and pharmacokinetic study, Jy.C., C.W.; original draft preparation, F.H., Z.Z., X.Z., Y.S., H.Z.; supervision, Y.S., H.Z.. All authors have given approval to the final version of the manuscript.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Abbreviations used

| DM | Dimethyl sulfoxide |
| :--- | :--- |
| DMSO | Dimethyl sulfoxide |
| RGZ | Rosiglitazone |
| RLA | relative luciferase activity |
| WY | wy14643 |
| GW | gw501516 |
| Alp | Alkaline phosphatase |
| HFD | High fat diet |
| PA | palmitic acid |
| IBMX | 3-Isobutyl-1-methylxanthine |
| RNA | Ribo nucleic Acid |
| mRNA | messenger RNA |
| RLE | relative expression |
| aP2 | Adipocyte Protein 2 |
| PPAR $\gamma$ | peroxisome proliferators-activated receptor gamma |
| PPAR $\alpha$ | peroxisome proliferators-activated receptor alpha |
| PPAR $\beta$ | peroxisome proliferators-activated receptor beta |
| RXR $\alpha$ | Retinoid X receptor alpha |
| PPRE | peroxisome proliferator response element |

Luc Luciferase
TA transcriptional activity
Max Maximum
p-AKT Phospho-Akt/protein kinase B
PH Hydrogen ion concentration
$\mu \mathrm{M} \quad$ Micromoles per liter
$\mathrm{mM} \quad$ Millimoles per liter

## Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.ejmech.2021.113542.

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[^1]:    ${ }^{\text {a }}$ The potency of the compounds on activation of PPAR $\gamma$ were measured by mammalian-one-hybrid assay.
    ${ }^{\mathrm{b}}$ The efficacy of the compounds on activation of PPAR $\gamma$ were measured by mammalian-one-hybrid assay.
    ${ }^{c}$ Compound binding to PPAR $\gamma$ was measured by fluorescence titration assay. All the values are expressed as the mean $\pm$ standard deviation (SD) from the dose-response curves of at least two independent experiments. Also see the Supplementary Figs. 1 and 2. TA, transcriptional activity.

