SYNTHESIS OF THE Rhodiola rosea GLYCOSIDE ROSAVIN

S. A. Patov, V. V. Punegov, and A. V. Kuchin

A synthetic scheme was proposed for the glycoside rosavin that includes a one-step glycosylation of cinnamyl alcohol with a disaccharide. The structure of the product was confirmed by PMR and ^{13}C NMR spectroscopy.

Key words: Rhodiola rosea, rosavin, glycoside synthesis.

The principal biologically active compounds in the extract of *Rhodiola rosea* are glycosides of *trans*-cinnamyl alcohol (rosin, rosavin, rosavin, and salidroside) [1-4].

We proposed a synthetic scheme for rosavin (1) because the glycoside is difficult to isolate from the plant material. The scheme included preparation of the disaccharide 2,3,4-tri-*O*-acetyl- α -L-arabinopyranosyl- $(1\rightarrow 6')$ -(1',2',3',4'-diisopropylidene-D-glucopyranose) (2) and its subsequent addition to the aglycon [5].

Disaccharide **2** was prepared by first synthesizing 1,2,3,4-diisopropylidene-D-glucopyranose (**3**) with a free hydroxyl on C₆. Glycosylation of **3** with 2,3,4-tri-*O*-acetyl- β -L-arabinopryanosylbromide (**4**) in the presence of a mixture of silver perchlorate and calcium carbonate (AgClO₄, CaCO₃) produced **2**. The IR spectrum of **2** contained bands characteristic of v C–O–C at 1670 cm⁻¹ and δ C–O–C (α -L-glycoside bond) at 824. The ¹³C NMR spectrum exhibited signals for C₇ and C₁₀ of the isopropylidene groups at 110.3 and 111.7 ppm, respectively; for methyls C_{8,9} and C_{11,12} at 26-28 ppm; for the α -L-glycoside bond between C₆ of glucose and C₁' of arabinose at 102.3 ppm; and for C₁ of glucose at 92.7 ppm. Signals for C atoms of arabinose and glucose were found at 62.4-72.2 ppm; of acetates, at 169.7-170.8 ppm (O=C–O) and 20.5-20.7 ppm (OCH₃).



The glycosylating agent was isolated from 2 by removing the protecting isopropylidene groups by heating to 50°C in dilute acetic acid. This approach did not affect the acetates and did not decompose the disacccharide at the glycoside bond.

The resulting 2,3,4-tri-*O*-acetyl- α -L-arabinopyranosyl- $(1 \rightarrow 6)$ - $(\beta$ -D-glucopyranose) was acylated with acetic anhydride in the presence of a catalytic amount of HClO₄ and then dissolved in a small amount of CHCl₃ with added methylmercaptan and boron trifluoride etherate (BF₃·Et₂O) to produce glycosylating agent **5**.

Institute of Chemistry, Komi Scientific Center, Urals Division, Russian Academy of Sciences, 167982, Syktyvkar, ul. Pervomaiskaya, 48, fax (88212) 43 66 77, e-mail: patov-sa@chemi.komisc.ru. Translated from Khimiya Prirodnykh Soedinenii, No. 4, pp. 323-324, July-August, 2006. Original article submitted April 24, 2006.

Mercaptan derivative **5** was purified and added to cinnamyl alcohol (6). The synthesis was performed in anhydrous CH_2Cl_2 using a solution of I_2 in CH_3CN as a catalyst. Acylated rosavin was hydrolyzed using the Zemplen reaction.

The product was purified in a concentrating cartridge filled with reversed-phase sorbent (Diasorb 130-C16T) to remove residual cinnamyl alcohol and carbohydrates. The effluent from the cartridge contained chromatographically pure (HPLC monitoring) rosavin. The structure of the synthetic rosavin was confirmed by NMR spectroscopy.

The ¹³C NMR and PMR spectra of rosavin exhibited chemical shifts that were consistent with the formation of a β -D-glycoside bond between glucose and the aglycon (102.3, dd, and 4.5 ppm) and an α -L-glycoside bond between arabinose and glucose (103.5 and 4.5 ppm, dd).

The angle of rotation $[\alpha]_D^{20}$ -56.3° (CHCl₃:MeOH) of the synthesized rosavin also confirmed that it had been prepared {lit. $[\alpha]_D^{20}$ -56.5° (CHCl₃:MeOH) [6]}.

EXPERIMENTAL

Materials and Methods. PMR and ¹³C NMR spectra were recorded on a Bruker DRX 400 spectrometer at working frequency 400 and 100 MHz, respectively, in DMSO-d₆ with TMS.

HPLC was performed on a Millichrom-2 chromatograph with a 2/60-mm column packed with Silasorb C18 and elution by $CH_3OH:C_4H_9OH:H_2O$ (35:1:14) with UV detection at 252 nm and flow rate 50 µL/min; TLC, on Silufol plates (Czech Rep.) using toluene:ethylacetate (6:1 and 4:1). The developer was phosphotungstic acid (20%) in ethanol with heating to 120°C for 1 min. Column chromatography was carried out over silica gel L (100/160 µm) using toluene:ethylacetate (10:1-4:1) as eluent.

Diisopropylideneglucose (6) was prepared by the literature method [7]; 2,3,4-tri-O-acetyl- β -L-arabinopyranosylbromide (5), by the literature method [7], silver carbonate, by reacting silver nitrate with saturated Na₂CO₃ solution in the dark and drying the resulting yellow powder with anhydrous acetone and then in vacuo. Boron trifluoride etherate (BF₃·OEt₂) was vacuum distilled. Crystalline iodine was purified by sublimation. Acetic anhydride (chemically pure, Russia), silver perchlorate (Sigma, USA) and methylmercaptan (Sigma, USA) were purchased commercially.

2,3,4-Tri-*O*-acetyl- α -L-arabinopyranosyl-(1 \rightarrow 6')-(1',2',3',4'-diisopropylidene-D-glucopyranose) (2). A solution of 1,2,3,4-diisopropylidene- α -D-glucopyranose (100 mg, 0.42 mmol) in anhydrous acetone (25 mL) was stirred, treated with an equimolar amount of glycosylating agent 2,3,4-tri-*O*-acetyl- β -L-arabinopyranosylbromide (172.6 mg, 0.42 mmol) and a two-fold excess of silver prechlorate (176 mg, 0.84 mmol) with CaCO₃ (84 mg, 0.84 mmol), irradiated for 15 min in an ultrasonic bath, and stirred for 1 d at room temperature. After the reaction was finished, the mixture was filtered. The solvent was evaporated. The solid was chromatographed over a silica-gel column using toluene:ethylacetate (9:1-4:1). Yield of **2**, 44.5 mg (18%). The IR and NMR spectra are described in the text.

2,3,4-Tri-*O*-acetyl- α -L-arabinopyranosyl-(1 \rightarrow 6')-(2',3',4'-tri-*O*-acetyl- β -methyl-1'-thio-D-glucopyranoside) (5). 2,3,4-Tri-*O*-acetyl- α -L-arabinopyranosyl-(1 \rightarrow 6')-(1',2',3',4'-tetra-*O*-acetyl- β -D-glucopyranoside) (1 g, 1.6 mmol) was dissolved in CHCl₃ (10 mL) and treated with methylmercaptan (92 mg, 1.9 mmol) and then boron trifluoride etherate (542 mg, 8 mmol). The mixture was stirred for 6 h, washed with saturated Na₂CO₃ solution and water, and dried over anhydrous Na₂SO₄ to afford **5** in 64% yield.

Total Acetate of Rosavin. 2,3,4-Tri-*O*-acetyl- α -L-arabinopyranosyl-(1-6')-(2',3',4'-tri-*O*-acetyl- β -methyl-1'-thio-Dglucopyranoside) was dissolved in dry CH₃CN (5 mL), stirred, and treated with cinnamic alcohol (33.5 mg, 0.25 mmol) and iodine (64.5 mg, 0.25 mmol). The mixture was stirred and cooled with ice for 1 h and then stirred at room temperature for 7 h. After the reaction was finished the reaction mixture was dissolved in CHCl₃ (20 mL) and washed with water (3 × 100 mL) and saturated Na₂S₂O₃ solution (50 mL) and again with water (100 mL). The organic layer was dried over anhydrous Na₂SO₄. The solvent was evaporated. The resulting product was separated over a column of silica gel using toluene:ethylacetate (9:1-4:1), yield 34.8 mg (30% of theoretical).

Rosavin (1). A solution of protected glycoside (186 mg, 0.4 mmol) in anhydrous methanol (25 mL) was treated with clean Na (10 mg, 0.4 mmol) dissolved in methanol (25 mL), stirred at room temperature for 5 h, and neutralized by adding cation-exchanger KU-2 (1 g) in the H-form [8]. Yield of **1**, 113 mg (95%).

¹³C NMR spectrum (ppm): 136.7, 126.3, 127.4, 128.5, 131.5, 127.5, 70.5, 102.3, 103.5, 64.8-74.5.

PMR spectrum (ppm, J/Hz): 7.45 d, 7.34 t, 6.35 m, 3.17 d, 6 (d, J = 16), 7.25 t, 4.3 (dd, J = 7), 3.3 (t, C-2'), 3.15 (t, C-3'), 3.05 (t, C-4'), 3.37 (q, C-5'), 3.05 (t, C-6'), 3.35 (t, C-2''), 3.15 (t, C-3''), 3.95 (dd, C-4''), 3.35 (m, C-5'').

REFERENCES

- 1. V. A. Kurkin and G. G. Zapesochnaya, *Khim.-Farm. Zh.*, **20**, No. 10, 1231 (1986).
- 2. *State Pharmacopeia* [in Russian], Meditsina, Moscow (1990).
- 3. V. A. Kurkin and L. G. Bondarenko, *Khim.-Farm. Zh.*, No. 4, 47 (1995).
- G. G. Zapesochnaya, V. A. Kurkin, V. A. Kir'yanov, and L. T. Bondarenko, in: *Chemical and Medical—Biological Evaluation of New Phytopreparations: A Collection of Scientific Works* [in Russian], Moscow (1989), p. 3.
- 5. A. F. Bochkov, S. A. Afanas'ev, and V. V. Zaikov, *Formation and Breaking of Glycoside Bonds* [in Russian], Nauka, Moscow (1978), p. 24.
- 6. V. A. Kurkin, *Phenylpropanoids, Promising Natural Biologically Active Compounds* [in Russian], Samara (1996).
- 7. Yu. A. Zhdanov, Practicum on Carbohydrate Chemistry [in Russian], Vysshaya Shkola, Moscow (1973).
- 8. L. F. Tietze and T. Eicher, *Preparative Organic Chemistry* [translated from Ger.], Mir, Moscow (1999).