

## PHENOLIC CONSTITUENTS FROM SEEDS OF *COPTIS JAPONICA* VAR. *DISSECTA*

MIZUO MIZUNO, HIROYUKI KOJIMA, TOSHIYUKI TANAKA, MUNEKAZU IINUMA, RIE KIMURA, MIN ZHI-DA\* and HIROKO MURATA†

Department of Pharmacognosy, Gifu Pharmaceutical University, 6-1 Mitahora-higashi 5 chome, Gifu 502, Japan; \*Department of Phytochemistry, Nanjing College of Pharmacy, Nanjing, China; †Faculty of Pharmaceutical Science, Setsunan University, 45-1 Nagaotoge-cho, Hirakata, Osaka 573-01, Japan

(Received 2 December 1986)

**Key Word Index**—*Coptis japonica* var. *dissecta*; Ranunculaceae, cleomiscosin A; aquillochin; 2',4,4'-trihydroxy-6'-methoxydihydrochalcone; 7,4'-dihydroxy-5-methoxyflavanone.

**Abstract**—In addition to coumarinolignans (cleomiscosin A and aquillochin) and 7,4'-dihydroxy-5-methoxyflavanone, a new dihydrochalcone, 2',4,4'-trihydroxy-6'-methoxydihydrochalcone, was isolated from the seeds of *Coptis japonica* var. *dissecta*. The structure of the dihydrochalcone was confirmed by comparison with relevant synthetic samples.

### INTRODUCTION

The rhizomes of *Coptis* species are used as a Chinese crude drug for digestive problems. Previous phytochemical studies showed the occurrence of alkaloids (berberine, coptisine and magnoflorine) [1] and of flavonoids (coptisine I and II) [2]. Our preceding paper dealt with the isolation of the benzophenanthridine alkaloids, sanguinarine, norsanguinarine, oxysanguinarine and 6-acetonyl-5,6-dihydrosanguinarine [3]. In this paper, the isolation of four phenolic constituents from seeds of *C. japonica* var. *dissecta* and the confirmation by syntheses of the dihydrochalcones and the flavanones related with the natural products are discussed.

### RESULTS AND DISCUSSION

Compound 1 gave a base peak in its mass spectrum of 386.0999 which agreed with a molecular formula of  $C_{20}H_{18}O_8$  (386.1001). In the  $^1H$  NMR spectrum, after the addition of  $D_2O$ , two peaks at 5.06 (t) and 9.19 (s) disappeared and another two at 3.39 (ddd) and 3.67 (ddd) changed to double doublets, which showed 1 contained a  $-CH-CH-CH_2OH$  moiety. The ABX pattern in the region of 6.8–7.1 ppm suggested 1 also contained a trisubstituted benzene. On irradiation of the peak at 3.78 ppm of a methoxyl group, the NOE was observed at the 7.03 ppm peak (d,  $J = 1.7$  Hz), indicating that 1 contained a coniferyl alcohol moiety. This was further supported by a fragment of  $m/z$  180 in the mass spectrum. A coumarin moiety in 1 was revealed from the signals of the *cis*-olefinic protons at 6.34 and 7.96 ppm ( $J = 8.2$  Hz) in the  $^1H$  NMR, the absorption at  $1700\text{ cm}^{-1}$  in the IR, and the absorption at 325 nm in the UV spectrum. The fragment of a coumarin moiety,  $m/z$  208, revealed it had one methoxyl and two hydroxyl groups, and the NOE was observed at the peak (6.91 ppm s) on irradiation of the methoxyl at 3.78 ppm. Thus, the coumarin was identified as fraxetin. From the above spectral data, 1 was deduced to be either cleomiscosin A or B but after comparison with the authentic samples, 1 was identified as cleomiscosin A [4–8].

The  $^1H$  NMR spectrum of compound 2 was similar to that of 1 except for an ABX pattern in the aromatic region. Thus, in 2, instead of coniferyl alcohol, a sinapyl alcohol moiety was confirmed by the  $^1H$  NMR data [6.75 (2H, s and 3.77 ppm (6H, s)] and supported by a fragment at  $m/z$  210 in the mass spectrum. Compound 2 was therefore deduced to be aquillochin and its identity confirmed by comparison with an authentic sample [9].

In the  $^1H$  NMR spectrum of compound 3, two peaks at 2.76 and 3.16 ppm in the  $A_2B_2$  pattern showed the presence of a moiety of  $-C=O-CH_2-CH_2-$ . Two peaks at 5.88 and 5.97 ppm (1H, each d,  $J = 2.0$  Hz) and two other peaks at 6.66 and 7.28 ppm (2H, each d,  $J = 8.5$  Hz) also showed *meta* coupled and *ortho* coupled benzene moieties. From the above spectral evidence, 3 was thought to be either 2',4,4'-hydroxy-6'-methoxy- (3), 2',4,6'-trihydroxy-4'-methoxy- (4) or 2',4',6'-trihydroxy-4-methoxydihydrochalcone (5). To clarify the structure of the naturally occurring compound and to compare the spectroscopic differences between 3, 4 and 5, the three dihydrochalcones were synthesized (see Experimental). Tables 1 and 2 show the physical and spectral data for synthetic 3, 4 and 5. The  $R_f$  values on thin-layer chromatography were quite similar to each other in the two solvent systems. It is also difficult to distinguish between 3 and 4, from their UV and mass spectra, which are very similar. However,  $^1H$  NMR spectra revealed sufficient differences among 3, 4 and 5 to distinguish between them. Because of the symmetrical arrangement of substituents in both 4 and 5, the protons at C-3' and 5' were observed as a singlet, but these protons in 3 were observed as a doublet. By comprehensive spectral comparisons, dihydrochalcone (3) was characterized as the new compound 2',4,4'-trihydroxy-6'-methoxydihydrochalcone.

Compound 6 from its positive Mg–HCl reaction was considered to be a flavonoid. In the  $^1H$  NMR, the signals at 2.72 (dd), 2.97 (dd) and 5.73 (dd) ppm were attributable to the protons at C-3 and C-2 of a flavanone with two hydroxyl and one methoxyl groups. An  $A_2B_2$  pattern at 6.88 and 7.33 and a singlet at 6.08 ppm suggested 6 was trioxxygenated at C-4', 5 and 7. No bathochromic shift was observed after addition of  $AlCl_3$  in the UV, indicating that

Table 1. Comparison of the physical and spectroscopic properties of the prepared flavonoids (3-7)

	3	4	5	6	7
mp (°)	193	160	192	229 (dec.)	125
TLC					
I*	0.45	0.44	0.45	0.42	0.64
II	0.56	0.56	0.53	0.29	0.74
UV $\lambda_{\text{max}}^{\text{MeOH}}$	289(4.0)†	286(4.1)	286(4.2)	283(4.2), 310sh(3.8)	285(4.0), 325sh(3.8)
+ NaOMe	—	—	—	318	288, 365
+ AlCl <sub>3</sub>	—	—	—	283, 310sh	312, 380
+ AlCl <sub>3</sub> /HCl	—	—	—	283, 310sh	309, 370
+ NaOAc	—	—	—	285sh, 318	286, 320sh
NaOAc/H <sub>3</sub> BO <sub>3</sub>	—	—	—	283sh, 305sh	286, 320sh
EIMS (m/z)					
(rel. int.)	282(22), 167(100), 120(42)	288(29), 167(100), 120(45)	288(21), 153(46), 133(44), 121(100)	286(78), 167(100), 120(49)	286(81), 167(100), 120(43)

\*Solvent systems; I CHCl<sub>3</sub>-MeOH = 8:1, II: C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO = 2:1.

†Log  $\epsilon$ .

Table 2. <sup>1</sup>H NMR spectral data of the prepared dihydrochalcones (3-5) and flavanones (6 and 7)

	3	4	5	6	7
2	7.01 d (8.3)	7.20 d (8.4)	7.14 d (8.6)	5.30 dd (1.20, 4.8)	5.34 dd (12.0, 4.5)
3	(6.66 d (8.3))	6.66 d (8.4)	6.83 d (8.6)	2.65 dd (18.0, 4.8)	2.68 dd (17.0, 4.5)
4	—	—	—	3.01 dd (18.0, 12.0)	3.15 dd (17.0, 12.0)
5	6.66 d (8.3)	6.66 d (8.4)	6.83 d (8.6)	—	—
6	7.01 d (8.3)	7.02 d (8.4)	7.14 d (8.6)	5.98 d (2.0)	6.06 br s
8	—	—	—	6.05 d (2.0)	6.06 br s
2'	—	—	—	7.30 d (9.0)	7.30 d (7.0)
3'	5.88 d (2.1)	5.94 s	5.82 s	6.78 d (9.0)	6.88 d (7.0)
5'	5.97 d (2.1)	5.94 s	5.80 s	6.78 d (9.0)	6.88 d (7.0)
6'	—	—	—	7.30 d (9.0)	7.30 d (7.0)
$\alpha$	3.15 t (7.6)	3.25 t (7.7)	3.24 t (7.6)	—	—
$\beta$	2.75 t (7.6)	2.80 t (7.7)	2.82 t (7.6)	—	—
OMe	3.80 s	3.74 s	3.71 s	3.78 s	3.78 s
OH	9.50 s	9.14 s	10.33 s	9.40 s	12.05 s
OH	13.75 s	12.32 s	12.21 s	10.30 br s	—

All spectra were measured in DMSO-*d*<sub>6</sub>. In parentheses *J* are shown in Hz.

the 5-position was methoxylated and thus suggesting that **6** was 7,4'-dihydroxy-5-methoxyflavanone. To confirm the structure, 7,4'-dihydroxy-5-methoxyflavone (**6**) and its isomer, 5,4'-dihydroxy-7-methoxyflavanone (**7**) were prepared. The syntheses are described in Experimental and the physical and spectral data are shown in Tables 1 and 2. By comparison of the synthetic flavanones, the naturally occurring flavanone was confirmed to have the structure **6**, which was previously isolated from *Achyrocline flaccida* [10].

#### EXPERIMENTAL

Mps are uncorr. MS were obtained on a JEOL JMS-D300 operating at 70 eV. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were taken on a Hitachi Perkin-Elmer R-20B 60 MHz instrument, a JEOL JNM-GX 270 instrument at 270 MHz, and a Varian XL-300 spectrometer at 300 MHz, chemical shifts were given in  $\delta$  (ppm) with tetramethylsilane as an int. standard. TLC was

carried out with Kiesel-gel 60F-254 (Merck). Fujigel BW820-MH and Sephadex LH-60 were used for CC.

**Extraction and isolation** Fresh seeds (7 kg) were percolated successively with *n*-hexane and MeOH. The MeOH solution was concd. and chromatographed on silica gel using C<sub>6</sub>H<sub>6</sub>-EtOAc (30:1) as solvent system. The eluate was rechromatographed on silica gel and Sephadex LH-60 (MeOH) for further purification. As the result of recryst., **1** (5 mg), **2** (10 mg), **3** (5 mg) and **6** (20 mg) were obtained.

**Cleomiscosin A 1.** Colourless needles, mp 224° (MeOH), C<sub>20</sub>H<sub>18</sub>O<sub>8</sub> (*M*, 386). EIMS *m/z* (rel. int.): 386 (71.6), 368 (35.9), 208 (8.9), 180 (97.4), 131 (100). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 233 (4.45), 288 (3.84), 325 (4.09). IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3300, 1700, 1610, 1570, 1420, 1290, 1140. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  3.39 (1H, *ddd*, *J* = 3.5, 4.3, 11.3 Hz, H-9'a), 3.67 (1H, *ddd*, *J* = 2.2, 4.3, 11.3 Hz, H-9'b), 3.78 (3H, *s*, C3'-OCH<sub>3</sub>), 3.79 (3H, *s*, C6-OCH<sub>3</sub>), 4.32 (1H, *ddd*, *J* = 2.2, 3.5, 7.0 Hz, H-8'), 4.99 (1H, *d*, *J* = 7.0 Hz, H-7'), 5.06 (1H, *t*, *J* = 4.3 Hz, C9'-OH), 6.34 (1H, *d*, *J* = 8.1 Hz, H-3), 6.82 (1H, *d*, *J* = 7.2 Hz, H-5'), 6.88 (1H, *dd*, *J* = 1.7, 7.2 Hz, H-6'), 6.91

(1H, s, H-5), 7.03 (1H, d,  $J = 1.7$  Hz, H-2'), 7.96 (1H, d,  $J = 8.1$  Hz, H-4), 9.19 (1H, s, C4'-OH).  $^{13}\text{C}$  NMR (pyridine- $d_5$ ):  $\delta$  55.9, 56.3 (each q,  $2 \times \text{OCH}_3$ ), 60.8 (t, C9'), 77.6 (d, C8'), 80.0 (d, C7'), 101.2 (d, C5), 112.0 (s, C10), 112.4 (d, C2'), 113.9 (d, C3), 116.7 (d, C5'), 121.8 (d, C6'), 127.7 (s, C1'), 134.9 (s, C8), 136.3 (s, C7), 138.5 (s, C9), 144.6 (d, C4), 146.5 (s, C6), 149.0 (s, C4'), 160.9 (s, C3), 171.5 (s, C2).

**Aquillochin 2.** Colourless needles, mp 195° (decomp.) (MeOH),  $\text{C}_{21}\text{H}_{20}\text{O}_9$  ( $M_r$ , 416). EIMS  $m/z$  (rel. int.): 416 (31.7), 398 (19.5), 249 (41.5), 210 (68.3), 208 (100). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 225 sh (3.49), 315 (3.25), 383 (2.39). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1690, 1440.  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  3.39 (1H, *ddd*,  $J = 2.2, 4.3, 11.3$  Hz, H-9'a) 3.66 (1H, *ddd*,  $J = 3.5, 4.3, 11.3$  Hz, H-9'b), 3.77 (6H, s, C3', 5'-OCH<sub>3</sub>), 3.78 (3H, s, C6-OCH<sub>3</sub>), 4.36 (1H, *ddd*,  $J = 2.2, 3.5, 7.0$  Hz, H-8'), 4.99 (1H, d,  $J = 7.0$  Hz, H-7'), 5.06 (1H, t,  $J = 4.3$  Hz, C9'-OH), 6.34 (1H, d,  $J = 8.1$  Hz, H-3), 6.75 (2H, s, H-2', 6'), 6.91 (1H, s, H-5), 7.92 (1H, d,  $J = 8.1$  Hz, H-4), 8.53 (1H, s, C4'-OH).

**2',4,4'-Trihydroxy-6'-methoxydihydrochalcone 3.** Colourless needles, mp 195° (EtOH),  $\text{C}_{16}\text{H}_{16}\text{O}_5$  ( $M_r$ , 288). For MS (see Table 1).  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  2.76 (2H, t,  $J = 7.5$  Hz, H<sub>2</sub>- $\beta$ ), 3.16 (2H, t,  $J = 7.5$  Hz, H<sub>2</sub>- $\alpha$ ), 3.81 (3H, s, OCH<sub>3</sub>), 5.88, 5.97 (1H, each d,  $J = 2.0$  Hz, H-3' and 5'), 6.66 (2H, d,  $J = 8.5$  Hz, H-3, 5), 7.28 (2H, d,  $J = 8.5$  Hz, H-2, 6), 9.12, 10.62, 13.75 (1H, each s, 3  $\times$  OH).

**Synthesis of 2',4,4'-trihydroxy-6'-methoxydihydrochalcone 3.** 4-Benzyloxy-2-hydroxy-6-methoxyacetophenone (2.2 g, 8.1 mmol) and *p*-benzyloxybenzaldehyde (8) (1.7 g, 8.0 mmol) were added to methylcellosolve soln. (200 ml) containing KOH (30 g), and the mixture was stirred at room temp. overnight. The reaction mixture was acidified with 20% HCl and extracted with H<sub>2</sub>O and EtOAc. The EtOAc extract was purified by CC on silica gel to give 4,4'-dibenzoyloxy-2'-hydroxy-6'-methoxychalcone (9) (3.0 g, 6.4 mmol) as yellow needles, mp 133° (EtOH). EIMS  $m/z$  (rel. int.): 466 (10.0), 375 (5.0), 191 (4.5), 91 (100).  $^1\text{H}$  NMR (60 MHz, CDCl<sub>3</sub>):  $\delta$  3.88 (3H, s, OCH<sub>3</sub>), 5.04, 5.08 (2H, each s, CH<sub>2</sub>Ph), 6.01, 6.16 (1H, each d,  $J = 2.4$  Hz, H-3', 5'), 6.94, 7.53 (2H, each d,  $J = 9.0$  Hz, H-2, 3, 5, 6), 7.37 (10H, br s, 2  $\times$  Ph), 7.45 (2H, s, H- $\alpha$  and  $\beta$ ), 14.31 (1H, s, OH). The chalcone 9 (1.5 g, 3.2 mmol) was debenzylated and reduced with Pd-C/H<sub>2</sub> (atmospheric pressure) in EtOAc (100 ml) for 6 hr and purified by silica gel chromatography to give 2',4,4'-trihydroxy-6'-methoxydihydrochalcone (3) (0.7 g, 2.4 mmol) as colourless needles from EtOH. Physical and spectral data are shown in Tables 1 and 2.

**Synthesis of 2',4,6-trihydroxy-4'-methoxydihydrochalcone 4.** 2,6-Diisopropoxy-4-methoxyacetophenone (0.6 g, 2.1 mmol) and *p*-isopropoxybenzaldehyde (0.3 g, 2.1 mmol) were condensed to give 2',4,6'-triisopropoxy-4'-methoxychalcone (0.7 g, 1.7 mmol) as an orange oil. EIMS  $m/z$  (rel. int.): 412 (78.9), 392 (20.0), 167 (100), 120 (48.9).  $^1\text{H}$  NMR (60 MHz, CDCl<sub>3</sub>):  $\delta$  1.20 (12H, d,  $J = 6.0$  Hz,  $2 \times \text{Me}_2$ ), 1.32 (6H, d,  $J = 6.0$  Hz, Me<sub>2</sub>), 3.78 (3H, s, OCH<sub>3</sub>), 4.43 [3H, m,  $J = 6.0$  Hz,  $3 \times \text{CH}(\text{Me})_2$ ], 6.11 (2H, s, H-3', 5'), 6.71, 7.22 (1H, each d,  $J = 15.0$  Hz, H- $\alpha$  and  $\beta$ ), 6.85, 7.45 (2H, each d,  $J = 7.5$  Hz, H-2, 3, 5, 6). The resulting chalcone (0.7 g, 1.7 mmol) was reduced with Pd-C/H<sub>2</sub> in EtOAc for 3 hr to give 2',4,6'-triisopropoxy-4'-methoxydihydrochalcone (0.5 g, 1.2 mmol) as an orange oil. EIMS  $m/z$  (rel. int.): 414 (12.9), 355 (4.9), 167 (100), 120 (48.8).  $^1\text{H}$  NMR (60 MHz, CDCl<sub>3</sub>):  $\delta$  1.26 (18H, d,  $J = 6.0$  Hz,  $3 \times \text{Me}_2$ ), 2.96 (4H, br s, H<sub>2</sub>- $\alpha$  and  $\beta$ ), 3.70 (3H, s, OCH<sub>3</sub>), 4.48 [3H, m,  $J = 6.0$  Hz,  $3 \times \text{CH}(\text{Me})_2$ ], 6.14 (2H, s, H-3', 5'), 6.80, 7.15 (2H, each d,  $J = 8.2$  Hz, H-2, 3, 5, 6). The reduced chalcone (0.7 g, 1.2 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) was cooled and treated with BCl<sub>3</sub> (2 ml) in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) at -70°C. The soln was allowed to warm room temp. poured into H<sub>2</sub>O (100 ml) and the mixture extracted with EtOAc. The conc. EtOAc extract was recryst. from EtOH to give 2',4,6'-trihydroxy-4'-

methoxydihydrochalcone (4) (0.3 g, 1.0 mmol) as colourless needles.

**Synthesis of 2',4,6'-trihydroxy-4-methoxydihydrochalcone 5.** 4,6-Dibenzoyloxy-2-hydroxyacetophenone (0.8 g, 2.4 mmol) and *p*-methoxybenzaldehyde (0.3 g, 2.2 mmol) were condensed to give 4',6'-dibenzoyloxy-2'-hydroxy-4-methoxychalcone (0.8 g, 1.7 mmol) as yellow needles, mp 138° from EtOH. EIMS  $m/z$  (rel. int.): 466 (86.6), 375 (100), 161 (97.7), 121 (72.2).  $^1\text{H}$  NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$  3.83 (3H, s, OCH<sub>3</sub>), 5.06, 5.10 (2H, each s, CH<sub>2</sub>Ph), 6.17, 6.22 (1H, each d,  $J = 2.3$  Hz, H-3', 5'), 6.71, 7.02 (2H, each d,  $J = 8.8$  Hz, H-2, 3, 5, 6), 7.35-7.67 (10H, m, 2  $\times$  Ph), 7.70, 7.80 (1H, each d,  $J = 15.4$  Hz, H- $\alpha$  and  $\beta$ ), 14.70 (1H, s, OH). The resulting dihydrochalcone (0.8 g, 1.4 mmol) was debenzylated and reduced with Pd-C/H<sub>2</sub> in EtOAc for 6 hr to give 2',4,6'-trihydroxy-4-methoxydihydrochalcone (5) (0.4 g, 1.4 mmol) as colourless needles from EtOH.

**7,4'-Dihydroxy-5-methoxyflavanone 6.** Colourless needles, mp 232° (decomp.) (MeOH),  $\text{C}_{16}\text{H}_{14}\text{O}_5$  ( $M_r$ , 286). For MS and UV data see Table 1. IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1600, 1590, 1450, 1210, 1100.  $^1\text{H}$  NMR (60 MHz, DMSO- $d_6$ ):  $\delta$  2.72 (1H, *dd*,  $J = 4.5, 18.0$  Hz, H-3a), 2.97 (1H, *dd*,  $J = 12.0, 18.0$  Hz, H-3b), 3.84 (3H, s, OCH<sub>3</sub>), 5.73 (1H, *dd*,  $J = 4.5, 12.0$  Hz, H-2), 6.08 (2H, br s, H-6, 8), 6.88 (2H, d,  $J = 8.3$  Hz, H-3', 5'), 7.33 (2H, d,  $J = 8.3$  Hz, H-2', 6'), 9.30, 10.24 (1H, each br s, OH).

**Synthesis of 5,4'-dihydroxy-7-methoxyflavanone 7.** Compound 7 was prepared in a similar way to 6, by condensation of 2-hydroxy-4,6-dimethoxyacetophenone (2.0 g, 9.4 mmol) and 6 (2.0 g, 9.4 mmol) to give 4-benzyloxy-2'-hydroxy-4',6'-dimethoxychalcone (3.0 g, 7.7 mmol) as yellow needles mp 130° from EtOH. EIMS  $m/z$  (rel. int.): 390 (20.0), 197 (10.0), 91 (100).  $^1\text{H}$  NMR (60 MHz, CDCl<sub>3</sub>):  $\delta$  3.82, 3.90 (3H, each s, OCH<sub>3</sub>), 5.15 (2H, s, CH<sub>2</sub>Ph), 5.97, 6.13 (1H, each d,  $J = 2.4$  Hz, H-3', 5'), 7.02, 7.58 (2H, each d,  $J = 9.0$  Hz, H-2, 3, 5, 6), 7.43 (5H, s, Ph), 7.82 (2H, br s, H- $\alpha$  and  $\beta$ ), 14.43 (1H, s, OH). On treatment with phosphoric acid, the chalcone (3.0 g, 7.7 mmol) was converted to 4'-benzyloxy-5,7-dimethoxyflavanone (1.2 g, 3.1 mmol), mp 159° (EtOH), pale yellow plates. EIMS  $m/z$  (rel. int.): 390 (15.0), 180 (8.5), 91 (100).  $^1\text{H}$  NMR (60 MHz, CDCl<sub>3</sub>):  $\delta$  2.65 (1H, *dd*,  $J = 4.8, 16.5$  Hz, H-3a), 3.02 (1H, *dd*,  $J = 12.0, 16.5$  Hz, H-3b), 3.78, 3.86 (3H, each s, OCH<sub>3</sub>), 5.07 (1H, s, CH<sub>2</sub>Ph), 5.33 (1H, *dd*,  $J = 4.8, 12.0$  Hz, H-2), 6.07, 6.13 (1H, each d,  $J = 1.8$  Hz, H-6, 8), 6.98, 7.37 (2H, each d,  $J = 9.0$  Hz, H-2', 3', 5', 6'), 7.39 (5H, s, Ph). The above flavanone (1.2 g, 3.1 mmol) was debenzylated with Pd-C/H<sub>2</sub> (atmospheric pressure) to give 5,7-dimethoxyflavanone (0.6 g, 2.0 mmol) as colourless mp 158°, from EtOH. EIMS  $m/z$  (rel. int.): 300 (70.0), 181 (100), 120 (35.8).  $^1\text{H}$  NMR (60 MHz, DMSO- $d_6$ ):  $\delta$  2.57 (1H, *dd*,  $J = 4.8, 16.5$  Hz, H-3a), 3.01 (1H, *dd*,  $J = 12.0, 16.5$  Hz, H-3b), 3.80, 3.83 (3H, each s, OCH<sub>3</sub>), 5.35 (1H, *dd*,  $J = 4.8, 12.0$  Hz, H-2), 6.12 (2H, br s, H-6, 8), 6.83, 7.30 (2H, each d,  $J = 9.0$  Hz, H-2', 3', 5', 6'), 9.22 (1H, s, OH). The debenzylated flavanone (0.6 g, 2.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) was cooled, treated with BCl<sub>3</sub> (0.5 ml) at -70° and worked up in the same manner as far 4 plus silica gel CC to give 5,4'-dihydroxy-7-methoxyflavanone (0.3 g, 1.0 mmol) as colourless needles from EtOH.

**Acknowledgement**—The authors are grateful to Prof. Dr. K. Ito (Meijyo University) for authentic samples of cleomiscosin A and aquillochin.

## REFERENCES

- Ikuta, A. and Itokawa, H. (1983) *Shoyakugaku Zasshi* 37, 195.
- Fujiwara, H., Nonaka, G., Yagai, A. and Nishioka, I. (1976) *Chem. Pharm. Bull.* 24, 407.

3. Mizuno, M., Kojima, H., Tanaka, T., Iinuma, M., Zhi-da, M. and Murata, H. (1987) *J. Nat. Prod.* (in press).
4. Ray, A. B., Chattopadhyay, S. K., Konno, C. and Hikino, H. (1980) *Tetrahedron Letters* **21**, 4477.
5. Ray, A. B., Chattopadhyay, S. K., Konno, C. and Hikino, H. (1982) *Heterocycles* **19**, 19.
6. Arisawa, M., Kinghorn, A. D., Cordell, G. A. and Farnsworth, N. R. (1983) *J. Nat. Prod.* **46**, 222.
7. Handa, S., Kinghorn, A. D., Cordell, G. A. and Farnsworth, N. R. (1983) *J. Nat. Prod.* **46**, 360.
8. Bhandari, P., Pant, P. and Prastogi, R. (1982) *Phytochemistry* **21**, 2147.
9. Ray, A. B., Chattopadhyay, S. K., Kumar, S., Konno, S., Kiso, Y. and Hikino, H. (1985) *Tetrahedron* **41**, 209.
10. Norbedo, C., Ferraro, G. and Cousio, J. D. (1982) *J. Nat. Prod.* **45**, 635.