

Note

## Enantioselective Synthesis of Chiral #-Thio-Quaternary Stereogenic Centers via Phase-Transfer-Catalyzed #-Alkylation of #-Acylthiomalonates

Min Woo Ha, Jun Young Lee, Doyoung Kim, Geumwoo Lee,  
Jae Kyun Lee, Suckchang Hong, and Hyeung-Geun Park

*J. Org. Chem.*, **Just Accepted Manuscript** • DOI: 10.1021/acs.joc.7b02605 • Publication Date (Web): 20 Dec 2017

Downloaded from <http://pubs.acs.org> on December 21, 2017

### Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

1  
2  
3 **Enantioselective Synthesis of Chiral  $\alpha$ -Thio-Quaternary**  
4  
5  
6 **Stereogenic Centers via Phase-Transfer-Catalyzed  $\alpha$ -**  
7  
8  
9  
10 **Alkylation of  $\alpha$ -Acylthiomalonates**  
11  
12  
13

14 Min Woo Ha,<sup>†</sup> Jun Young Lee,<sup>†</sup> Doyoung Kim,<sup>†</sup> Geumwoo Lee,<sup>†</sup> Jae Kyun Lee,<sup>‡</sup> Suckchang Hong,<sup>†</sup>  
15 and Hyeung-geun Park<sup>†,\*</sup>  
16  
17  
18

19  
20 <sup>†</sup>*Research Institute of Pharmaceutical Sciences and College of Pharmacy, Seoul National University,*  
21 *Seoul 151-742, Korea.*  
22  
23

24 <sup>‡</sup>*Neuro-Medicine Center, Korea Institute of Science and Technology, PO Box 131, Cheongyang,*  
25 *Seoul 130-650, Republic of Korea.*  
26  
27  
28  
29

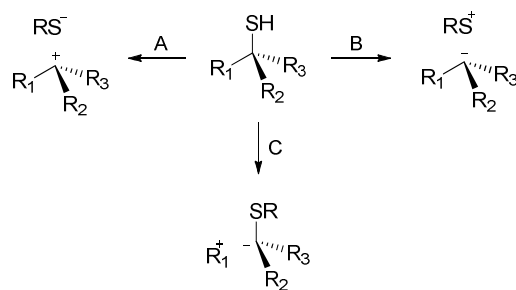
30  
31 **Corresponding author**  
32

33  
34 Hyeung-geun Park, Tel.: +82-2-880-7821, 8264, Fax: +82-2-872-9129, E-mail: hgpark@snu.ac.kr  
35  
36  
37  
38  
39

40 **Abstract**  
41

42 An efficient synthetic method for establishing chiral  $\alpha$ -thio- $\alpha$ -quaternary stereogenic center was  
43 successfully developed. The enantioselective  $\alpha$ -alkylation of  $\alpha$ -acylthiomalonates under phase-  
44 transfer catalytic conditions [50% aq. KOH, toluene,  $-20$  °C and (*S,S*)-3,4,5-trifluorophenyl-NAS  
45 bromide] provided the corresponding  $\alpha$ -acylthio- $\alpha$ -alkylmalonates in high chemical yields (up to  
46 99%) and high optical yields (up to 98% ee).  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

Organosulfur compounds play important roles in biological systems. Two natural amino acids contain sulfur residues (cysteine and methionine), and their functional groups mediate various biological processes in the enzymes of living organisms. Additionally, many commercially available organosulfur compounds are best-selling drugs.<sup>1</sup> As the need for variable sulfur-containing compounds for the development of new drugs increases, the development of asymmetric synthetic methods for organosulfur compounds having chirality at the sulfur carbon has become increasingly important. The asymmetric preparation of secondary chiral sulfur (II) compounds via stereospecific substitution is well developed.<sup>2</sup> However, few methods are available for the asymmetric synthesis of tertiary thiols. Although chiral tertiary alcohols are generally obtained by the enantioselective addition of a nucleophile to a prochiral ketone<sup>3</sup>, the analogous approach to the preparation of chiral tertiary thiols is difficult due to the instability of thioketones and the possibility of nucleophilic attack occurring at the sulfur rather than at the carbon.<sup>4</sup> Therefore, the synthesis of enantiomerically pure tertiary thiols is very challenging despite the simplicity of their structures.

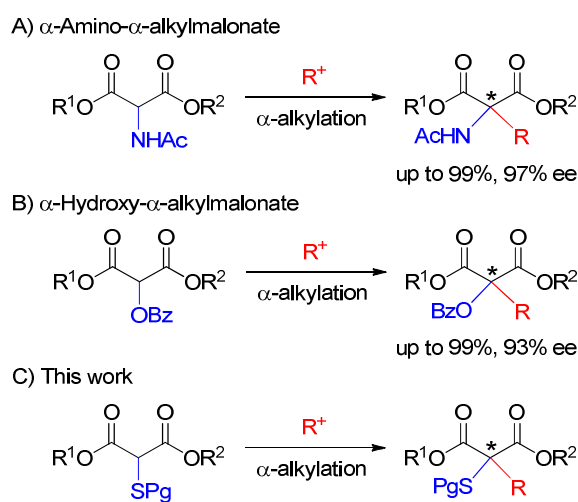


**Scheme 1.** Synthetic methods for tertiary thiol or thioether

Three general synthetic approaches are available for the preparation of a chiral tertiary thiol, and they are classified according to the disconnection of either a C-S (A and B) or a C-C bond (C) (Scheme 1).<sup>5</sup> The thiol can be obtained by stereoselective attack of either a sulfur nucleophile on an electrophilic carbon (A)<sup>6</sup> or a nucleophilic carbon on an electrophilic sulfur (B)<sup>7</sup>. Additionally, the stereoselective  $\alpha$ -alkylation of a secondary organosulfur substrate can generate a quaternary

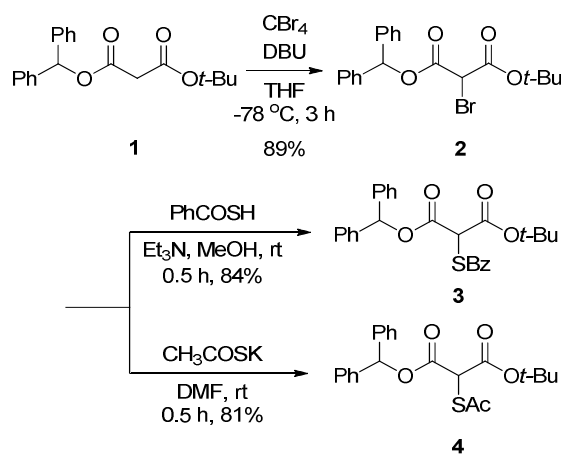
1  
2  
3 stereogenic center (C).<sup>8</sup> Generally, the C-C bond formation methods (C) use a chiral auxiliary, a  
4 stoichiometric chiral base, a rearrangement reaction or self-regeneration of a chiral secondary sulfide  
5 to impart tertiary chirality. However, a catalytic synthesis of a chiral tertiary sulfur system via C-C  
6 bond formation has not yet been reported.  
7  
8  
9

10  
11 In 2011, we reported the asymmetric synthesis of quaternary chiral  $\alpha,\alpha$ -dialkylmalonates as  
12 versatile chiral starting materials in both high chemical and optical yields via enantioselective phase-  
13 transfer catalytic (PTC)<sup>9</sup>  $\alpha$ -alkylation of malonates with chiral quaternary ammonium salts.<sup>10</sup> The  
14 established synthetic method was applied to the preparation of chiral quaternary  $\alpha$ -aminomalonates<sup>11</sup>  
15 (Scheme 2, A) and  $\alpha$ -hydroxymalonates<sup>12</sup> (Scheme 2, B), which were successfully converted to  
16 useful chiral molecules bearing quaternary stereogenic centers. To expand our enantioselective  
17 PTC  $\alpha$ -alkylation methodology, we planned to synthesize chiral quaternary  $\alpha$ -thiomalonates that  
18 have not yet been reported. As shown in the synthetic strategy illustrated in Scheme 2, a sulfur group  
19 is incorporated into the  $\alpha$ -position of malonates, and the subsequent enantioselective PTC  $\alpha$ -  
20 alkylations with various electrophiles afford a versatile library of quaternary chiral  $\alpha$ -thiomalonates  
21 (Scheme 2, C). Herein, we report efficient enantioselective synthetic method for the preparation of  
22 quaternary  $\alpha$ -thio- $\alpha$ -alkylmalonates.  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37



55 **Scheme 2.** Synthetic strategy for chiral quaternary  $\alpha$ -thiomalonates.  
56  
57  
58  
59  
60

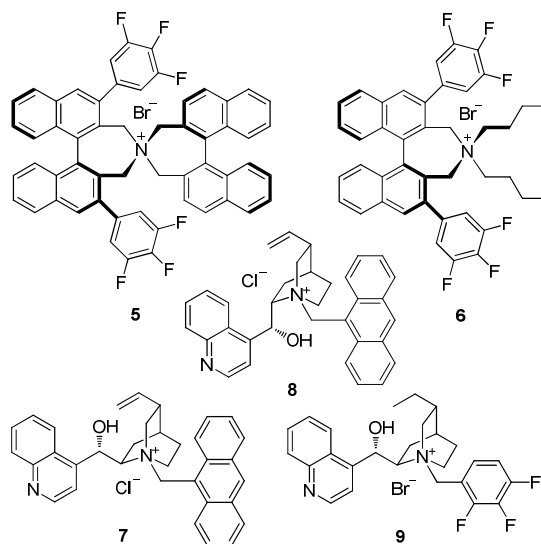
1  
2  
3  
4  
5 First, we designed enantiotopic unsymmetrical  $\alpha$ -thiomalonates, bearing non nucleophilic  
6 sulfur as substrate for the PTC  $\alpha$ -alkylation. Since both diphenylmethyl and tert-butyl ester groups  
7 were shown to be necessary for high enantioselectivity in previous reports<sup>11,12</sup>, we chose  
8 diphenylmethyl tert-butyl malonate (**1**) as a template. Two  $\alpha$ -acylthiomalonates were prepared by  $\alpha$ -  
9 bromination using  $\text{CBr}_4$  in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) followed by  
10 substitution with benzoethiolic acid or potassium thioacetate to afford  $\alpha$ -benzoylthiomalonate (**3**) or  
11  $\alpha$ -acetylthiomalonate (**4**), respectively (Scheme 3).  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22



**Scheme 3.** Preparation of  $\alpha$ -acylthiomalonates.

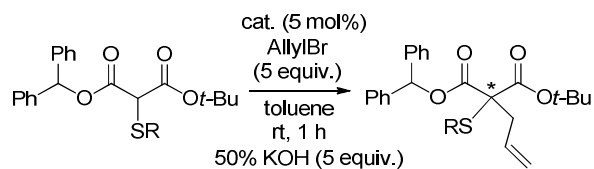
43 The phase-transfer catalytic  $\alpha$ -allylation of the prepared  $\alpha$ -acylthiomalonates were  
44 performed to examine their efficiency as substrates. The enantioselective PTC allylations of **3** and **4**  
45 were carried out under typical PTC conditions<sup>12</sup> [allyl bromide (5.0 equiv.), 50% KOH (aq., 5.0  
46 equiv.), toluene, room temperature] in the presence of chiral quaternary ammonium salts (**5** – **9**)<sup>13</sup>  
47 (Figure 1). As shown in Table 1, both  $\alpha$ -acylthiomalonates successfully afforded their corresponding  
48  $\alpha$ -allylated products in the presence of the Maruoka's catalyst, (*S,S*)-3,4,5-trifluorophenyl-NAS  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

bromide (**5**) (entries 1 – 2). The acetylthio group (**4**, 78% ee) gave a higher enantioselectivity than that of the benzoylthio group (**3**, 72% ee). This result is opposite of that obtained for the  $\alpha$ -hydroxymalonates system<sup>12</sup>, which may be due to the difference in size between the two atoms (oxygen and sulfur).



**Figure 1.** Representative chiral phase-transfer catalysts.

**Table 1.** Enantioselective PTC  $\alpha$ -alkylation of  $\alpha$ -acylthiomalonates.<sup>a</sup>

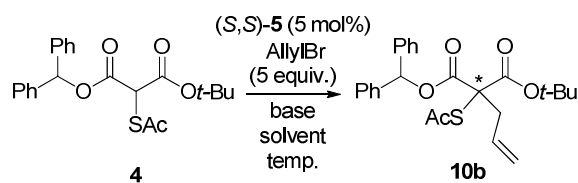


entry	substrate	cat.	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	<b>3</b>	<b>5</b>	97	72
-----				
2	<b>4</b>	<b>5</b>	99	78
3	<b>4</b>	<b>6</b>	91	11
4	<b>4</b>	<b>7</b>	77	15
5	<b>4</b>	<b>8</b>	40	-14
6	<b>4</b>	<b>9</b>	96	24

<sup>a</sup>Reactions were performed with 5.0 equiv. of allyl bromide and 5.0 equiv. of 50% KOH(*aq.*) under the given conditions. <sup>b</sup>Isolated yields. <sup>c</sup>Enantiopurity was determined by HPLC analysis using a chiral column (DAICEL Chiralpak AD-H).

Among the examined chiral PTC catalysts, the Maruoka's catalyst (**5**) gave the highest enantioselectivity (entries 1 and 2). The lower enantioselectivities obtained for the other catalysts (**6** – **9**), including Lygo's catalysts (**7**, **8**), were consistent with the trends observed in the previous reports.<sup>11,12</sup>

**Table 2.** Optimization of the reaction conditions for the enantioselective PTC  $\alpha$ -allylation of  $\alpha$ -acetylthiomalonate.<sup>a</sup>



entry	base	solvent	<i>T</i> (°C)	time (h)	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	50% KOH	toluene	rt	1	99	78
2	50% CsOH	toluene	rt	1	87	71
3	s-KOH	toluene	rt	1	96	72
4	s-CsOH	toluene	rt	1	36	76
5	50% KOH	CH <sub>2</sub> Cl <sub>2</sub>	rt	1	75	65
6	50% KOH	THF	rt	1	85	45
7	50% KOH	toluene	0	20	95	84
8	50% KOH	toluene	-20	40	95	89
9	50% KOH	toluene	-40	144	33	90

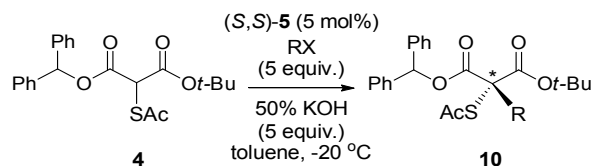
<sup>a</sup>Reactions were performed with 5.0 equiv. of allyl bromide and 5.0 equiv. of base under the given conditions. <sup>b</sup>Isolated yields. <sup>c</sup>Enantiopurity was determined by HPLC analysis using a chiral column (DAICEL Chiralpak AD-H).

1  
2  
3  
4  
5 Next, the optimization of the base and temperature conditions were performed using  
6  
7 substrate **4** and the best catalyst (*S,S*)-**5** (Table 2). Chemical yield and enantioselectivity were  
8  
9 dependent on the base and solvent conditions at room temperature (entries 1 – 6). The highest  
10  
11 enantioselectivity was observed with 50% KOH base in toluene (entries 1). In case of temperature,  
12  
13 higher enantioselectivities were observed at lower reaction temperatures (entry 1, entries 7 – 9).  
14  
15 However, a longer reaction time resulted in a lower chemical yield at –40 °C (entry 9). When the  
16  
17 PTC reaction was kept for 1 week, the alkylated product was partially hydrolyzed and immediately  
18  
19 desulfurized to mono-alkylated malonate that was further alkylated to  $\alpha,\alpha$ -dialkylated malonate as a  
20  
21 side product. Consequentially, 50% KOH base in toluene at –20 °C was selected as the optimal  
22  
23 reaction conditions based on the chemical yield, optical yield, and reaction time (entry 8; 95% yield,  
24  
25 89% ee).  
26  
27  
28  
29  
30

31 We then turned our attention to determine the scope and limitations of electrophiles in the  
32  
33 optimized enantioselective PTC alkylation of **4**. As shown in Table 3, most of the alkylating agents  
34  
35 afforded very high enantioselectivities (**10c–p**, 90–98% ee). However, an unactivated alkylating  
36  
37 reagent gave no alkylated product (data not shown), and propargyl bromide (**10a**) resulted in a  
38  
39 relatively poor enantioselectivity, which may be due to geometrical constraints. The achievement of  
40  
41 very high enantioselectivities (up to 98% ee) shows that this methodology is very efficient for the  
42  
43 enantioselective synthesis of quaternary  $\alpha$ -thio- $\alpha$ -alkylmalonates. Note that the enantioselectivities  
44  
45 were higher than those of the  $\alpha$ -acyloxymalonate system (up to 93% ee).<sup>12</sup> To the best of our  
46  
47 knowledge, this is the first report to accomplish the enantioselective catalytic synthesis of quaternary  
48  
49  $\alpha$ -thio- $\alpha$ -alkylmalonates.  
50  
51  
52  
53  
54

55 **Table 3.** Enantioselective synthesis of  $\alpha$ -acetylthio- $\alpha$ -alkylmalonates via PTC  $\alpha$ -alkylation<sup>a</sup>  
56



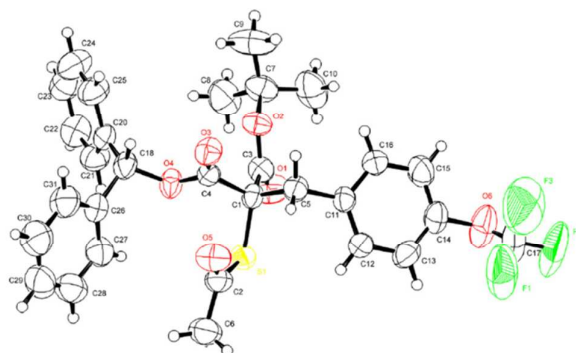


entry	RX	time (h)	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	CH≡CBr ( <b>a</b> )	23	81	61
2	CH <sub>2</sub> =CHCH <sub>2</sub> Br ( <b>b</b> )	40	95	89
3	CH <sub>2</sub> =C(CH <sub>3</sub> )CH <sub>2</sub> Br ( <b>c</b> )	23	91	94
4	CH <sub>2</sub> =C(Br)CH <sub>2</sub> Br ( <b>d</b> )	23	86	92
5	PhCH <sub>2</sub> Br ( <b>e</b> )	8	99	95
6	4-Me-PhCH <sub>2</sub> Br ( <b>f</b> )	4	96	96
7	4-F-PhCH <sub>2</sub> Br ( <b>g</b> )	4	99	97
8	4-Cl-PhCH <sub>2</sub> Br ( <b>h</b> )	7	99	98
9	4-Br-PhCH <sub>2</sub> Br ( <b>i</b> )	5	99	97
10	4-NO <sub>2</sub> -PhCH <sub>2</sub> Br ( <b>j</b> )	4	98	95
11	4-t-Bu-PhCH <sub>2</sub> Br ( <b>k</b> )	7	99	96
12	4-CF <sub>3</sub> -PhCH <sub>2</sub> Br ( <b>l</b> )	6	99	96
13	4-CF <sub>3</sub> O-PhCH <sub>2</sub> Br ( <b>m</b> )	5	86	96( <i>R</i> ) <sup>d</sup>
14	4-CO <sub>2</sub> Me-PhCH <sub>2</sub> Br ( <b>n</b> )	6	99	96
15	3-CH <sub>3</sub> O-PhCH <sub>2</sub> Br ( <b>o</b> )	9	96	96
16	β-naphthyl-CH <sub>2</sub> Br ( <b>p</b> )	30	91	90

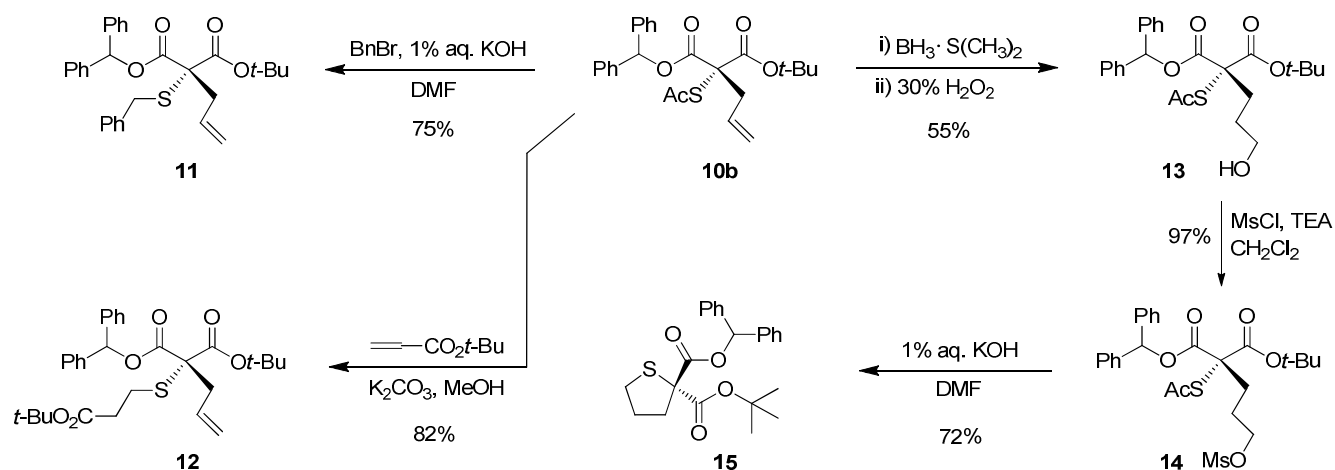
<sup>a</sup>Reactions were performed with 5.0 equiv. of alkyl bromide and 5.0 equiv. of 50% KOH(*aq.*) under the given conditions.

<sup>b</sup>Isolated yields. <sup>c</sup>Enantiopurity was determined by HPLC analysis using a chiral column (DAICEL Chiralpak AD-H).

<sup>d</sup>Absolute configuration of **10m** was determined as *R* by X-ray crystallography (Figure 2)<sup>14</sup> and the absolute configuration of other compounds were tentatively assigned as *R* based on the X-ray crystallography of **10m**.



**Figure 2.** X-ray Crystallographic Structure of **10m**.<sup>14</sup>



**Scheme 4** Derivatizations of chiral acetylthiomalonate **10b**

As a demonstration of the synthetic potential of this methodology, derivatizations of **10b** were performed, as illustrated in Scheme 4. The hydrolysis of chiral acetylthiomalonate **10b** under 1% aq-KOH or aq-K<sub>2</sub>CO<sub>3</sub> conditions in the presence of benzyl bromide or tert-butyl acrylate directly provided the corresponding *S*-alkylated products **11** and **12**, respectively.<sup>15</sup> Additionally, the hydroboration of **10b** by borane and hydrogenperoxide afforded the corresponding alcohol **13**. The mesylation of **13** followed by hydrolysis using 1% aq-KOH in dimethylformamide (DMF) successively afforded the cyclized tetrahydrothiofuran **15**.<sup>16</sup>

In conclusion, a new asymmetric synthetic method was successfully developed for the preparation of  $\alpha$ -acylthio- $\alpha$ -alkylmalonates via PTC  $\alpha$ -alkylation. The enantioselective PTC  $\alpha$ -alkylations of diphenylmethyl-*tert*-butyl  $\alpha$ -acetylthiomalonate provided the corresponding  $\alpha$ -acetylthio- $\alpha$ -alkylmalonates in high chemical yields (up to 99%) and optical yields (up to 98% ee). Our newly established PTC reaction provides an efficient method to prepare useful versatile chiral molecules containing  $\alpha$ -sulfur quaternary stereogenic centers.

## Experimental Section

### *General Methods*

All reagents purchased from commercial sources were used without further purification. Commercially available KOH pellet (99%) was grinded to prepare Solid KOH as powder form. 50% w/v aqueous KOH was used as stock solution. Phase-transfer catalysts (**8** and **9**) were prepared according to the reported procedure. Phase-transfer catalysts (**5**, **6**, and **7**) were purchased from the commercial sources. TLC analyses were performed using pre-coated TLC plate (silica gel 60 GF<sub>254</sub>, 0.25 mm). Flash column chromatography was performed on flash silica gel 230~400 mesh size. The values of enantiomeric excess (ee) of chiral products were determined by HPLC using 4.6 mm × 250 mm Daicel Chiralpak AD-H. Infrared analyses (KBr pellet) were performed by FT-IR. <sup>1</sup>H-NMR spectra was recorded at 300MHz, 400MHz, or 500MHz with reference to CHCl<sub>3</sub> (δ 7.24) or CH<sub>3</sub>OH (δ 3.31). <sup>13</sup>C-NMR spectra was obtained by 100 MHz, 125 MHz, or 150MHz spectrometer relative to the central CDCl<sub>3</sub> (δ 77.0) or CD<sub>3</sub>OD (δ 49.0) resonance. Coupling constants (*J*) in <sup>1</sup>H-NMR are in Hz. Low-resolution mass spectra (LRMS) and high-resolution mass spectra (HRMS) were measured on positive-ion FAB, CI, or Q-TOF (ESI) spectrometer. Melting points were measured on melting point apparatus and were uncorrected. Optical rotations were measured on polarimeter and calibrated with pure solvent as blank.

### *(A) Procedure for preparation of PTC substrates*

To a stirred solution of benzhydryl *tert*-butyl malonate (**1**, 1.63 g, 5.0 mmol) prepared according to the already reported procedure in dry THF (100 mL) was added DBU (822 μL, 5.5 mmol) at -78 °C under argon atmosphere. After stirring for 1 hour at the designated temperature, carbon tetrabromide (2 g, 6.0 mmol) was added little by little, and the reaction mixture was stirred for 3 hours at the same temperature. When the starting material was consumed, a saturated ammonium chloride solution was

1  
2  
3 added. Extraction using dichloromethane (150 mL x 2), drying over anhydrous MgSO<sub>4</sub>, filtration,  
4  
5 and concentration were performed to yield the residue which can be purified by flash column  
6  
7 chromatography (silica gel, Hexane : EtOAc = 30:1 ) to afford the  $\alpha$ -mono-brominated malonate, **2**  
8  
9 (1.8 g, 89%) as a pale yellow oil. Potassium thioacetate (423 mg, 3.7 mmol) was added to a stirred  
10  
11 solution of  $\alpha$ -bromomalonate, **2** (1.0 g, 2.47 mmol) in dry dimethylformamide (10 mL) at room  
12  
13 temperature under argon atmosphere. The reaction was stirred until the TLC analysis showed that the  
14  
15 reaction was complete. The reaction solvent was evaporated and diluted with EtOAc (100 ml),  
16  
17 extracted with brine (100 ml x 2), dried over anhydrous MgSO<sub>4</sub>, filtered, and concentrated in *vacuo*.  
18  
19 The residue was purified by column chromatography (silica gel, Hexane : EtOAc = 10:1) to afford 1-  
20  
21 benzhydryl 3-(*tert*-butyl) 2-(acetylthio)malonate (**4**, 801 mg, 81% yield) as a white solid. 1-  
22  
23 benzhydryl 3-(*tert*-butyl) 2-(benzoylthio)malonate (**3**) was synthesized in the same manner using 1.5  
24  
25 equivalent of benzothioic acid and triethylamine in anhydrous methanol instead of potassium  
26  
27 thioacetate in DMF.  
28  
29  
30  
31  
32

33 ***(B) Typical experimental procedure for enantioselective phase-transfer catalytic alkylation.***

34  
35 To a solution of 1-benzhydryl 3-(*tert*-butyl) 2-(acetylthio)malonate (**4**, 30 mg, 0.075 mmol) and  
36  
37 (*S,S*)-3,4,5-trifluorophenyl-NAS bromide (**5**, 3.4 mg, 0.0037 mmol) in toluene (250  $\mu$ L) was added  
38  
39 allyl bromide (32.5  $\mu$ L, 0.38 mmol) at room temperature. At the designated low temperature,  
40  
41 aqueous 50% w/v aqueous KOH (32.6  $\mu$ L, 0.38 mmol) was added to the reaction mixture and stirred  
42  
43 until the starting material disappeared. After completion of the reaction, the reaction mixture was  
44  
45 diluted with ethyl acetate (20 mL), washed with brine (10 mL x 2), dried over anhydrous MgSO<sub>4</sub>,  
46  
47 filtered, and concentrated in *vacuo*. The residue was purified by flash column chromatography on  
48  
49 silica gel eluting with Hexane-EtOAc solution (15:1) to afford **10b** (32.7 mg, 99% yield) as a  
50  
51 colorless oil.  
52  
53  
54  
55  
56  
57  
58  
59  
60

**(C) Analytical data****1-Benzhydryl 3-(*tert*-butyl) 2-(benzoylthio)malonate (3)**

Following the procedure (A) from the compound **2** (1.0 g, 2.47 mmol), the title molecule **3** was obtained as a colorless oil (960 mg, 84% yield). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>) δ 7.95 (d, *J* = 7.85 Hz, 2H), 7.59 (t, *J* = 7.35 Hz, 1H), 7.45 (t, *J* = 7.73 Hz, 2H), 7.36 ~ 7.26 (m, 10H), 6.96 (s, 1H), 5.31 (s, 1H), 1.38 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 188.6, 165.3, 164.3, 139.20, 139.18, 135.7, 134.0, 128.7, 128.5, 128.4, 128.08, 128.06, 127.5, 127.2, 127.1, 83.9, 78.9, 51.6, 27.6 ppm; IR (KBr) 3033, 2980, 1738, 1675, 1449, 1288, 1144, 906, 772 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>27</sub>H<sub>26</sub>O<sub>5</sub>SNa 485.1399; found 485.1404.

**1-Benzhydryl 3-(*tert*-butyl) 2-(acetylthio)malonate (4)**

Following the procedure (A) from the compound **2** (1.0 g, 2.47 mmol), the title molecule **4** was obtained as a white solid (801mg, 81% yield). mp 70 °C; <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ 7.32 ~ 7.24 (m, 10H), 6.90 (s, 1H), 5.07 (s, 1H), 2.37 (s, 3H), 1.33 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 192.2, 165.2, 164.3, 139.2, 139.1, 128.48, 128.45, 128.13, 128.09, 127.2, 127.1, 83.9, 78.8, 51.6, 29.9, 27.6 ppm; IR (KBr) 3033, 2981, 1738, 1545, 1446, 1220, 1139, 948, 767 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>22</sub>H<sub>24</sub>O<sub>5</sub>SNa 423.1242; found 423.1247.

**1-Benzhydryl 3-(*tert*-butyl) 2-(benzoylthio)-2-allylmalonate (3')**

Following the procedure (B) from the substrate **3** (30 mg, 0.065 mmol), the title molecule **3'** (not numbered in the manuscript) was obtained as a colorless oil (31.7 mg, 97% yield). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>) δ 7.84 (d, *J* = 7.40 Hz, 2H), 7.56 (t, *J* = 7.40 Hz, 1H) 7.41 (t, *J* = 7.78 Hz, 2H), 7.35 ~ 7.24 (m, 8H), 7.14 ~ 7.12 (m, 3H), 6.98 (s, 1H), 5.84 ~ 5.75 (m, 1H), 5.06 ~ 5.03 (m, 2H), 3.23 ~ 3.15 (m, 2H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 190.0, 166.4, 165.5, 139.5, 139.3, 136.5, 133.6, 132.1, 128.6, 128.4, 128.2, 128.1, 127.8, 127.6, 127.20, 127.16, 119.7, 84.0, 78.6, 66.0,

39.1, 27.6 ppm; IR (KBr) 3033, 2979, 1732, 1665, 1449, 1370, 1219, 911, 772  $\text{cm}^{-1}$ ; HRMS (FAB)  $m/z$ :  $[\text{M}+\text{Na}]^+$  Calcd for  $\text{C}_{30}\text{H}_{30}\text{O}_5\text{SNa}$  525.1712; found 525.1707. The enantioselectivity was determined by chiral HPLC analysis (DAICEL Chiralpak AD-H, hexane : 2-propanol = 80 : 20, flow rate = 1.0 ml/min, 23 °C,  $\lambda$  = 254 nm) retention time; major isomer 15.34 min, minor isomer 19.55 min, 72% ee,  $[\alpha]_{\text{D}}^{20} = -16.21$  ( $c$  1.0,  $\text{CHCl}_3$ ).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(prop-2-yn-1-yl)malonate (10a)**

Following the procedure (B) from the substrate 4 (30 mg, 0.075 mmol) using propargyl bromide, the title molecule 10a was obtained as a yellow oil (26.6 mg, 81% yield).  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.32 ~ 7.26 (m, 10H), 6.94 (s, 1H), 3.32 (ddd,  $J_1 = 23.66$  Hz,  $J_2 = 17.45$  Hz,  $J_3 = 2.60$  Hz, 2H), 2.23 (s, 3H), 2.01 ~ 2.00 (m, 1H), 1.30 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\delta$  194.2, 165.2, 164.1, 139.2, 139.1, 128.39, 128.35, 128.2, 128.09, 128.08, 127.4, 127.3, 84.5, 79.0, 78.7, 71.8, 65.3, 29.9, 27.4, 26.6 ppm; IR (KBr) 2979, 1738, 1705, 1370, 1258, 954, 756, 700  $\text{cm}^{-1}$ ; HRMS (FAB)  $m/z$ :  $[\text{M}+\text{Na}]^+$  Calcd for  $\text{C}_{25}\text{H}_{26}\text{O}_5\text{SNa}$  461.1399; found 461.1395. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80 : 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: major isomer 12.06 min, minor isomer 18.71 min, 61% ee,  $[\alpha]_{\text{D}}^{20} = -3.47$  ( $c$  1.0,  $\text{CHCl}_3$ ).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-allylmalonate (10b)**

Following the procedure (B) from the substrate 4 (30 mg, 0.075 mmol) using allyl bromide, the title molecule 10b was obtained as a colorless oil (31.4 mg, 95% yield).  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.34 ~ 7.25 (m, 10H), 6.92 (s, 1H), 5.76 ~ 5.68 (m, 1H), 5.04 (s, 1H) 5.02 (d,  $J = 5.20$  Hz, 1H), 3.10 ~ 3.02 (m, 2H), 2.18 (s, 3H), 1.26 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  193.6, 166.2, 165.3, 139.4, 139.3, 132.0, 128.4, 128.3, 128.1, 128.0, 127.5, 127.2, 119.6, 83.9, 78.6, 65.7, 39.0, 30.0, 27.5 ppm; IR (KBr) 2927, 1732, 1695, 1451, 1370, 1220, 1153, 953, 768  $\text{cm}^{-1}$ ; HRMS (FAB)  $m/z$ :

1  
2  
3 [M+Na]<sup>+</sup> Calcd for C<sub>25</sub>H<sub>28</sub>O<sub>5</sub>SNa 463.1555; found 463.1557. The enantioselectivity was determined  
4 by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0  
5 mL/min, 23 °C, λ = 254 nm) retention time: minor isomer 10.51 min, major isomer 15.96 min, 89%  
6 ee, [α]<sub>D</sub><sup>20</sup> = -14.47 (c 1.0, CHCl<sub>3</sub>).  
7  
8  
9

10  
11  
12  
13 **(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(2-methylallyl)malonate (10c)**

14  
15 Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 3-bromo-2-  
16 methylpropene, the title molecule **10c** was obtained as a pale yellow oil (31.0 mg, 91% yield). <sup>1</sup>H-  
17 NMR (300 MHz, CDCl<sub>3</sub>) δ 7.37 ~ 7.26 (m, 10H), 6.94 (s, 1H), 4.81 (s, 1H), 4.67 (s, 1H), 3.14 (s,  
18 2H), 2.20 (s, 3H), 1.64 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 193.9, 166.6, 165.7,  
19 2H), 2.20 (s, 3H), 1.64 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 193.9, 166.6, 165.7,  
20 140.4, 139.32, 139.30, 128.4, 128.3, 128.1, 128.0, 127.6, 127.3, 116.1, 83.9, 78.7, 65.7, 41.4, 29.9,  
21 27.4, 23.4 ppm; IR (KBr) 2853, 1732, 1693, 1455, 1370, 1255, 1200, 950, 761 cm<sup>-1</sup>; HRMS (FAB)  
22 m/z: [M+Na]<sup>+</sup> Calcd for C<sub>26</sub>H<sub>30</sub>O<sub>5</sub>SNa 477.1712; found 477.1726. The enantioselectivity was  
23 determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow  
24 rate = 1.0 mL/min, 23 °C, λ = 254 nm) retention time: minor isomer 8.44 min, major isomer 12.60  
25 min, 94% ee, [α]<sub>D</sub><sup>20</sup> = -23.78 (c 1.0, CHCl<sub>3</sub>).  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38

39 **(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(2-bromoallyl)malonate (10d)**

40  
41 Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 2,3-bromopropene, the  
42 title molecule **10d** was obtained as a yellow oil (33.5 mg, 86% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ  
43 7.37 ~ 7.27 (s, 10H), 6.94 (s, 1H), 5.53 (s, 1H), 5.50 (d, J = 1.83 Hz, 1H), 3.63 (dd, J<sub>1</sub> = 22.40 Hz, J<sub>2</sub>  
44 = 15.83 Hz, 2H), 2.22 (s, 3H), 1.28 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) δ 194.3, 165.7, 164.6,  
45 139.1, 128.44, 128.36, 128.2, 128.1, 127.5, 127.3, 127.2, 121.9, 84.5, 79.1, 65.5, 44.0, 29.9, 27.4  
46 ppm; IR (KBr) 2928, 1735, 1692, 1370, 1245, 1146, 1003, 953, 761 cm<sup>-1</sup>; HRMS (FAB) m/z:  
47 [M+Na]<sup>+</sup> Calcd for C<sub>25</sub>H<sub>27</sub>O<sub>5</sub>BrSNa 541.0660; found 541.0655. The enantioselectivity was  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 10.58 min, major isomer 15.07 min, 92% ee,  $[\alpha]_D^{20} = -1.29$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-benzylmalonate (10e)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using benzyl bromide, the title molecule **10e** was obtained as a colorless oil (36.4 mg, 99% yield). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.36 ~ 7.31 (m, 4H), 7.30 ~ 7.25 (m, 6H), 7.20 ~ 7.13 (m, 3H), 7.02 (d, *J* = 6.90 Hz, 2H), 6.94 (s, 1H), 3.67 (dd, *J*<sub>1</sub> = 25.78 Hz, *J*<sub>2</sub> = 14.43 Hz, 2H), 2.17 (s, 3H), 1.25 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  194.0, 166.3, 165.2, 139.24, 139.23, 135.4, 130.6, 128.5, 128.3, 128.2, 127.95, 127.91, 127.7, 127.2, 127.1, 84.0, 78.8, 67.3, 39.5, 30.0, 27.5 ppm; IR (KBr) 3033, 2979, 1732, 1691, 1455, 1220, 1147, 951, 773 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>29</sub>H<sub>30</sub>O<sub>5</sub>SNa 513.1712; found 513.1722. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 11.94 min, major isomer 16.44 min, 95% ee,  $[\alpha]_D^{20} = +74.53$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-methylbenzyl)malonate (10f)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-methylbenzyl bromide, the title molecule **10f** was obtained as a pale yellow oil (36.4 mg, 96% yield). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 ~ 7.31 (m, 4H), 7.30 ~ 7.28 (m, 6H), 6.96 ~ 6.94 (m, 3H), 6.89 (d, *J* = 7.80 Hz, 2H), 3.62 (d, *J* = 3.70 Hz, 2H), 2.26 (s, 3H), 2.21 (s, 3H), 1.26 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  194.0, 166.4, 165.3, 139.3, 136.7, 132.2, 130.4, 128.6, 128.4, 128.3, 128.1, 127.9, 127.7, 127.2, 83.9, 78.7, 67.5, 39.1, 30.0, 27.5, 21.1 ppm; IR (KBr) 2927, 1733, 1692, 1455, 1370, 1257, 1182, 952, 746 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>30</sub>H<sub>32</sub>O<sub>5</sub>SNa 527.1868; found



527.1861. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 13.99 min, major isomer 16.66 min, 96% ee,  $[\alpha]_D^{20} = +93.26$  ( $c$  1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-fluorobenzyl)malonate (10g)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-fluorobenzyl bromide, the title molecule **10g** was obtained as a pale yellow oil (37.8 mg, 99% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.38 ~ 7.26 (m, 10H), 7.00 ~ 6.95 (m, 3H), 6.86 ~ 6.80 (m, 2H), 3.64 (dd,  $J_1 = 16.11$  Hz,  $J_2 = 14.67$  Hz, 2H), 2.23 (s, 3H), 1.26 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.18, 165.17, 162.1 (d,  $J = 244.0$  Hz), 139.14, 139.1, 132.1 (d,  $J = 7.80$  Hz), 131.0 (d,  $J = 3.20$  Hz), 128.5, 128.3, 128.2, 128.0, 127.6, 127.2, 114.7 (d,  $J = 21.10$  Hz), 84.1, 78.8, 67.2, 38.8, 30.0, 27.5 ppm; IR (KBr) 2930, 1732, 1692, 1509, 1257, 1221, 1148, 772, 700 cm<sup>-1</sup>; HRMS (FAB)  $m/z$ : [M+Na]<sup>+</sup> Calcd for C<sub>29</sub>H<sub>29</sub>O<sub>5</sub>FSNa 531.1617; found 531.1599. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 12.45 min, major isomer 14.55 min, 97% ee,  $[\alpha]_D^{20} = +56.50$  ( $c$  1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-chlorobenzyl)malonate (10h)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol), using 4-chlorobenzyl bromide, the title molecule **10h** was obtained as a pale yellow oil (39.0 mg, 99% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.30 ( $J = 8.22$  Hz, 2H), 6.94 ~ 6.92 (m, 3H), 3.63 (s, 2H), 2.23 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.1, 165.1, 139.1, 139.0, 133.8, 133.1, 131.9, 128.5, 128.3, 128.2, 128.1, 127.6, 127.2, 84.2, 78.9, 67.0, 38.9, 30.0, 27.5 ppm; IR (KBr) 2926, 1733, 1692, 1493, 1256, 1181, 951, 772 cm<sup>-1</sup>; HRMS (FAB)  $m/z$ : [M+Na]<sup>+</sup> Calcd for C<sub>29</sub>H<sub>29</sub>O<sub>5</sub>ClSNa 547.1322; found 547.1331. The enantioselectivity was determined by chiral HPLC analysis

(DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 15.51 min, major isomer 18.61 min, 98% ee,  $[\alpha]_D^{20} = +96.64$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-bromobenzyl)malonate (10i)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-bromobenzyl bromide, the title molecule **10i** was obtained as a pale yellow oil (42.3 mg, 99% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.36 ~ 7.25 (m, 12H), 6.94 (s, 1H), 6.87 (d, *J* = 8.43 Hz, 2H), 3.62 (s, 2H), 2.23 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.1, 165.1, 139.1, 139.0, 134.4, 132.3, 131.0, 128.5, 128.4, 128.3, 128.1, 127.6, 127.2, 121.3, 84.2, 78.9, 67.0, 38.9, 30.0, 27.5 ppm; IR (KBr) 2979, 1733, 1693, 1489, 1256, 1148, 951, 759, 699 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>29</sub>H<sub>29</sub>O<sub>5</sub>BrSNa 591.0817; found 591.0828. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 18.44 min, major isomer 21.40 min, 97% ee,  $[\alpha]_D^{20} = +111.17$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-nitrobenzyl)malonate (10j)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-nitrobenzyl bromide, the title molecule **10j** was obtained as a pale yellow oil (39.4 mg, 98% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.96 (d, *J* = 8.43 Hz, 2H), 7.34 ~ 7.25 (m, 10H), 7.15 (d, *J* = 8.43 Hz, 2H), 6.94 (s, 1H), 3.76 (dd, *J*<sub>1</sub> = 16.65 Hz, *J*<sub>2</sub> = 14.28 Hz, 2H), 2.25 (s, 3H), 1.29 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.8, 165.8, 164.9, 147.1, 143.1, 138.9, 138.8, 131.4, 128.5, 128.4, 128.2, 127.5, 127.2, 123.0, 84.5, 79.1, 66.5, 39.3, 30.1, 27.5 ppm; IR (KBr) 2926, 1732, 1694, 1522, 1348, 1147, 951, 764, 699 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>29</sub>H<sub>29</sub>O<sub>7</sub>NSNa 558.1562; found 558.1591. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane :

2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 24.01 min, major isomer 26.66 min, 95% ee,  $[\alpha]_D^{20} = +79.89$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-(tert-butyl)benzyl)malonate (10k)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-(tert-butyl)benzyl bromide, the title molecule **10k** was obtained as a pale yellow oil (40.6 mg, 99% yield). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.36 ~ 7.25 (m, 10H), 7.15 (d, *J* = 8.20 Hz, 2H), 6.97 (d, *J* = 8.15 Hz, 2H), 6.94 (s, 1H), 3.63 (dd, *J*<sub>1</sub> = 31.40 Hz, *J*<sub>2</sub> = 14.45 Hz, 2H), 2.21 (s, 3H), 1.25 (s, 9H), 1.23 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  194.0, 166.4, 165.3, 149.9, 139.3, 132.2, 130.3, 128.4, 128.3, 128.1, 127.9, 127.6, 127.3, 124.8, 83.9, 78.7, 67.3, 39.1, 34.4, 31.3, 30.0, 27.5 ppm; IR (KBr) 2963, 1733, 1692, 1456, 1257, 1182, 952, 759, 699 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>33</sub>H<sub>38</sub>O<sub>5</sub>SNa 569.2338; found 569.2325. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 90: 10, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 14.78 min, major isomer 16.53 min, 96% ee,  $[\alpha]_D^{20} = +53.64$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-(trifluoromethyl)benzyl)malonate (10l)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-(trifluoromethyl)benzyl bromide, the title molecule **10l** was obtained as a pale yellow oil (41.5 mg, 99% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.40 ~ 7.25 (m, 12H), 7.11 (d, *J* = 7.86 Hz, 2H), 6.95 (s, 1H), 3.72 (s, 2H), 2.24 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.0, 165.0, 139.5, 139.04, 138.99, 130.9, 129.4 (d, *J* = 32.10 Hz), 128.5, 128.3, 128.2, 128.1, 127.6, 127.2, 124.8 (d, *J* = 3.00 Hz), 124.1 (d, *J* = 270.50 Hz), 84.3, 78.9, 66.8, 39.3, 30.0, 27.5 ppm; IR (KBr) 2926, 1733, 1694, 1371, 1148, 1067, 761, 699 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>30</sub>H<sub>29</sub>O<sub>5</sub>F<sub>3</sub>SNa 581.1586; found 581.1575. The enantioselectivity was determined by chiral HPLC

analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 99: 1, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 78.77 min, major isomer 87.07 min, 96% ee,  $[\alpha]_D^{20} = +36.16$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-(trifluoromethoxy)benzyl)malonate (10m)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 4-(trifluoromethoxy)benzyl bromide, the title molecule **10m** was obtained as a white solid (37.0 mg, 86% yield). mp 68 °C; <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.37 ~ 7.26 (m, 9H), 7.06 ~ 6.95 (m, 6H), 3.67 (dd,  $J_1 = 20.16$  Hz,  $J_2 = 14.46$  Hz, 2H), 2.23 (s, 3H), 1.25 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.1, 165.1, 148.4, 139.10, 139.06, 134.1, 132.0, 128.5, 128.3, 128.2, 128.1, 127.6, 127.2, 120.3, 84.3, 78.9, 66.9, 38.9, 30.0, 27.4 ppm; IR (KBr) 2981, 1733, 1693, 1508, 1371, 1149, 951, 743, 699 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>30</sub>H<sub>29</sub>O<sub>6</sub>F<sub>3</sub>SNa 597.1535; found 597.1545.

The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 99: 1, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 54.51 min, major isomer 56.66 min, 96% ee,  $[\alpha]_D^{20} = +52.02$  (*c* 1.0, CHCl<sub>3</sub>).

**(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(4-(methoxycarbonyl)benzyl)malonate (10n)**

Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using methyl 4-(bromomethyl)benzoate, the title molecule **10n** was obtained as a pale yellow oil (40.7 mg, 99% yield). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.82 (d,  $J = 8.22$  Hz, 2H), 7.36 ~ 7.26 (m, 10H), 7.09 (d,  $J = 8.25$  Hz, 2H), 6.95 (s, 1H), 3.88 (s, 3H), 3.71 (d,  $J_1 = 16.02$  Hz,  $J_2 = 14.37$  Hz, 2H), 2.23 (s, 3H), 1.27 (s, 9H) ppm; <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  193.9, 166.8, 166.0, 165.0, 140.8, 139.1, 139.0, 130.6, 129.2, 129.0, 128.5, 128.3, 128.2, 128.0, 127.6, 127.2, 84.3, 78.9, 66.9, 52.0, 39.4, 30.0, 27.5 ppm; IR (KBr) 2925, 1725, 1693, 1370, 1147, 1021, 952, 760, 700 cm<sup>-1</sup>; HRMS (FAB) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>31</sub>H<sub>32</sub>O<sub>7</sub>SNa 571.1766; found 571.1755. The enantioselectivity was determined

1  
2  
3 by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0  
4 mL/min, 23 °C,  $\lambda = 254$  nm) retention time: minor isomer 16.98 min, major isomer 21.67 min, 96%  
5 ee,  $[\alpha]_D^{20} = +110.02$  ( $c$  1.0,  $\text{CHCl}_3$ ).  
6  
7  
8  
9

10  
11 **(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(3-methoxybenzyl)malonate (10o)**

12  
13 Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 3-methoxybenzyl  
14 bromide, the title molecule **10o** was obtained as a pale yellow oil (37.5 mg, 96% yield).  $^1\text{H-NMR}$   
15 (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.39 ~ 7.23 (m, 10H), 7.11 ~ 7.05 (m, 1H), 6.96 (s, 1H), 6.76 ~ 6.72 (m, 1H),  
16 6.64 ~ 6.61 (m, 2H), 3.66 (dd,  $J_1 = 18.03$  Hz,  $J_2 = 14.58$  Hz, 2H), 3.66 (s, 3H), 2.22 (s, 3H), 1.25 (s,  
17 9H) ppm;  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  194.0, 166.3, 165.2, 159.1, 139.2, 136.9, 128.8, 128.5,  
18 128.3, 128.1, 128.0, 127.5, 127.3, 122.9, 116.3, 112.6, 84.0, 78.7, 67.2, 55.0, 39.5, 30.0, 27.4 ppm;  
19 IR (KBr) 2931, 1733, 1692, 1455, 1370, 1148, 1052, 953, 771  $\text{cm}^{-1}$ ; HRMS (FAB)  $m/z$ :  $[\text{M}+\text{Na}]^+$   
20 Calcd for  $\text{C}_{30}\text{H}_{32}\text{O}_6\text{SNa}$  543.1817; found 543.1808. The enantioselectivity was determined by chiral  
21 HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min,  
22 23 °C,  $\lambda = 254$  nm) retention time: minor isomer 11.29 min, major isomer 14.08 min, 96% ee,  $[\alpha]_D^{20}$   
23 = +35.86 ( $c$  1.0,  $\text{CHCl}_3$ ).  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39

40 **(R)-1-Benzhydryl 3-(tert-butyl)-2-(acetylthio)-2-(naphthalen-2-ylmethyl)malonate (10p)**

41 Following the procedure **(B)** from the substrate **4** (30 mg, 0.075 mmol) using 2-(bromomethyl)  
42 naphthalene, the title molecule **10p** was obtained as a pale yellow oil (37.0 mg, 91% yield).  $^1\text{H-NMR}$   
43 (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.78 ~ 7.75 (m, 1H), 7.66 ~ 7.59 (m, 2H), 7.49 (s, 1H), 7.44 ~ 7.26 (m, 12H),  
44 7.18 (d,  $J = 8.40$  Hz, 1H), 6.97 (s, 1H), 3.90 ~ 3.80 (dd,  $J_1 = 17.10$  Hz,  $J_2 = 14.37$  Hz, 2H), 2.24 (s,  
45 3H), 1.26 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  194.1, 166.3, 165.3, 139.22, 139.18, 133.04,  
46 133.00, 132.5, 129.4, 128.6, 128.5, 128.3, 128.2, 128.0, 127.7, 127.6, 127.5, 127.4, 127.2, 125.8,  
47 125.7, 84.1, 78.8, 67.5, 39.6, 30.0, 27.5 ppm; IR (KBr) 2928, 1732, 1691, 1370, 1216, 1147, 771,  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

700  $\text{cm}^{-1}$ ; HRMS (FAB)  $m/z$ :  $[\text{M}+\text{Na}]^+$  Calcd for  $\text{C}_{33}\text{H}_{32}\text{O}_5\text{SNa}$  563.1868; found 563.1887. The enantioselectivity was determined by chiral HPLC analysis (DIACEL Chiralpak AD-H, hexane : 2-propanol = 80: 20, flow rate = 1.0 mL/min, 23 °C,  $\lambda$  = 254 nm) retention time: minor isomer 13.96 min, major isomer 19.34 min, 90% ee,  $[\alpha]_{\text{D}}^{20} = +188.19$  ( $c$  1.0,  $\text{CHCl}_3$ ).

### **(R)-1-Benzhydryl 3-tert-butyl 2-allyl-2-(benzylthio)malonate (11)**

To a solution of acetylthio-allylmalonate, **10b** (27 mg, 62  $\mu\text{mol}$ ) and benzyl bromide (11  $\mu\text{L}$ , 93  $\mu\text{mol}$ ) in dimethylformamide (1.5 mL) were added 1% aq. KOH (0.52 mL, 93  $\mu\text{mol}$ ) at room temperature. After completion of the reaction, the mixture was diluted with ethyl acetate (10 mL) and the organic layer was separated and collected. Drying over anhydrous  $\text{MgSO}_4$ , filtration, and concentration were performed, and the residue was purified by flash column chromatography on silica gel eluting with Hexane-EtOAc solution (24:1) to afford the desired product, **11** (22.6 mg, 75% yield) as a colorless oil.  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 ~ 7.15 (m, 15H), 6.97 (s, 1H), 3.69 (s, 2H), 5.87 ~ 5.73 (m, 1H), 5.10 ~ 5.04 (m, 2H), 2.88 (d,  $J = 7.2$  Hz, 2H), 1.35 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  167.9, 167.0, 139.7, 139.6, 136.3, 131.7, 129.6, 128.72, 128.69, 128.66, 128.4, 128.3, 127.8, 127.5, 127.4, 119.6, 83.5, 78.6, 62.4, 38.5, 34.6, 27.9 ppm; IR (KBr) 2979, 1726, 1496, 1369, 1232, 1130, 1027, 841, 772  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{Na}]^+$  Calcd for  $\text{C}_{30}\text{H}_{32}\text{O}_4\text{SNa}$  511.1914; found 511.1917.  $[\alpha]_{\text{D}}^{20} = -1.27$  ( $c$  1.0,  $\text{CHCl}_3$ ).

### **(R)-1-Benzhydryl 3-tert-butyl 2-allyl-2-((3-(tert-butoxy)-3-oxopropyl)thio)malonate (12)**

To a solution of acetylthio-allylmalonate, **10b** (44 mg, 0.10 mmol) in methanol (2.0 mL) were added  $\text{K}_2\text{CO}_3$  (21 mg, 0.15 mmol) and *tert*-butyl acrylate (22  $\mu\text{L}$ , 0.15 mmol) at room temperature. After completion of the reaction, the methanol was removed *in vacuo* and the residue was diluted with ethyl acetate (10 mL) and the organic layer was washed with brine and separated. Drying over anhydrous  $\text{MgSO}_4$ , filtration, and concentration were performed, and the residue was purified by

1  
2  
3 flash column chromatography on silica gel eluting with Hexane-EtOAc solution (20:1) to afford the  
4  
5 desired product, **12** (43 mg, 82% yield) as a colorless oil. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ 7.40 ~ 7.25  
6  
7 (m, 10H), 6.92 (s, 1H), 5.83 ~ 5.69 (m, 1H), 5.08 ~ 5.03 (m, 2H), 2.83 (d, *J* = 6.9 Hz, 2H), 2.75 ~  
8  
9 2.60 (m, 2H), 2.35 (t, *J* = 7.5 Hz, 2H) 1.42 (s, 9H), 1.32 (s, 9H) ppm; <sup>13</sup>C-NMR (200 MHz, CDCl<sub>3</sub>) δ  
10  
11 170.7, 167.7, 166.9, 139.40, 139.36, 131.4, 128.4, 128.1, 128.0, 127.5, 127.1, 119.3, 83.2, 80.8, 78.4,  
12  
13 77.2, 77.0, 76.8, 61.8, 38.2, 34.9, 28.0, 27.6, 24.5 ppm; IR (KBr) 2932, 1729, 1641, 1496, 1369,  
14  
15 1253, 1080, 843, 772 cm<sup>-1</sup>; HRMS (ESI) *m/z*: [M+Na]<sup>+</sup> Calcd for C<sub>30</sub>H<sub>38</sub>O<sub>6</sub>SNa 549.2281; found  
16  
17 549.2285. [α]<sub>D</sub><sup>20</sup> = -1.63 (*c* 1.0, CHCl<sub>3</sub>).

18  
19  
20  
21  
22 **(*R*)-1-Benzhydryl 3-*tert*-butyl 2-(acetylthio)-2-(3-hydroxypropyl)malonate (13)**

23  
24 To a solution of the olefin, **10b** (45.5 mg, 0.103 mmol) in anhydrous THF (2 mL) was added borane  
25  
26 dimethyl sulfide complex solution (1.0 M in DCM, 230 μL, 0.227 mmol) dropwise under argon at 0  
27  
28 °C, and stirred for 3 hours. After adding THF:MeOH (1:1) (1 ml) for quenching reaction, Aqueous  
29  
30 phosphate buffer solution (1 mL) and aqueous H<sub>2</sub>O<sub>2</sub> (30 % w/w, 25 μL) were added dropwise and the  
31  
32 mixture was stirred for overnight at room temperature. After evaporating the mixture, the organic  
33  
34 phase was extracted for two times (5 mL x 2), combined, dried over MgSO<sub>4</sub>, filtered, and then  
35  
36 concentrated. The residue was purified by column chromatography (silica gel, Hexane : EtOAc = 2:1)  
37  
38 to afford the desired primary alcohol, **13** (26.0 mg, 55% yield) as a colorless oil. <sup>1</sup>H-NMR (300 MHz,  
39  
40 CDCl<sub>3</sub>) δ 7.37 ~ 7.26 (m, 10H), 6.94 (s, 1H), 3.57 (t, *J* = 6.45 Hz, 2H), 2.41 ~ 2.35 (m, 2H), 2.20 (s,  
41  
42 3H), 1.60 ~ 1.49 (m, 2H), 1.29 (s, 9H) ppm; <sup>13</sup>C-NMR (200 MHz, CDCl<sub>3</sub>) δ 193.6, 166.6, 165.7,  
43  
44 139.4, 139.3, 128.4, 128.3, 128.1, 128.0, 127.5, 127.2, 83.9, 78.6, 66.1, 62.4, 31.0, 30.0, 28.3, 27.5  
45  
46 ppm; IR (KBr) 3566, 2934, 1729, 1497, 1370, 1150, 951, 839, 772 cm<sup>-1</sup>; HRMS (ESI) *m/z*: [M+Na]<sup>+</sup>  
47  
48 Calcd for C<sub>25</sub>H<sub>30</sub>O<sub>6</sub>SNa 481.1655; found 481.1651. [α]<sub>D</sub><sup>20</sup> = -5.83 (*c* 1.0, CHCl<sub>3</sub>).

49  
50  
51  
52  
53  
54 **(*R*)-1-Benzhydryl 3-*tert*-butyl 2-(acetylthio)-2-(3-((methylsulfonyl)oxy)propyl)malonate (14)**

The primary alcohol, **13** (13.5 mg, 29.4  $\mu\text{mol}$ ) in dichloromethane (2 mL) was treated with methanesulfonyl chloride (3  $\mu\text{L}$ , 44.2  $\mu\text{mol}$ ) in the presence of trimethylamine (8  $\mu\text{L}$ , 58.9  $\mu\text{mol}$ ) at room temperature under Ar atmosphere and stirred for 1 hour. After completion of the reaction, the reaction mixture was added with saturated sodium bicarbonate solution (5 mL), diluted with dichloromethane (10 mL), washed with brine (5 mL), dried over anhydrous  $\text{MgSO}_4$ , filtered, and concentrated in *vacuo*. The residue was purified by flash column chromatography on silica gel eluting with Hexane-EtOAc solution (3:1) to afford the desired mesylate, **14** (15.3 mg, 97% yield) as a colorless oil.  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 ~ 7.25 (m, 10H), 6.93 (s, 1H), 4.16 (td,  $J_1 = 6.3$  Hz,  $J_2 = 1.8$  Hz, 2H), 2.96 (s, 3H), 2.46 ~ 2.39 (m, 2H), 2.21 (s, 3H), 1.80 ~ 1.70 (m, 2H), 1.29 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  193.9, 166.4, 165.6, 139.5, 139.3, 128.7, 128.6, 128.4, 128.3, 127.7, 127.4, 84.5, 79.0, 69.4, 65.9, 37.6, 31.0, 30.2, 27.7, 25.2 ppm; IR (KBr) 2978, 1730, 1696, 1496, 1356, 1217, 1112, 949, 771  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{Na}]^+$  Calcd for  $\text{C}_{26}\text{H}_{32}\text{O}_8\text{S}_2\text{Na}$  559.1431; found 559.1440.  $[\alpha]_D^{20} = -11.51$  ( $c$  1.0,  $\text{CHCl}_3$ ).

**(R)-2-Benzhydryl 2-tert-butyl dihydrothiophene-2,2(3H)-dicarboxylate (15)**

To a solution of the mesylate, **14** (16 mg, 0.03 mmol) in dimethylformamide (0.1 mL) was added 1% aq. KOH (2  $\mu\text{L}$ , 35.3  $\mu\text{mol}$ ) at room temperature. After complete conversion of the starting material to the thiocycle compound, the mixture was diluted with Ethyl acetate (5 mL x 2) and the organic layer was separated and collected. Drying over anhydrous  $\text{MgSO}_4$ , filtration, and concentration were performed, and the residue was purified by flash column chromatography on silica gel eluting with Hexane-EtOAc solution (24:1) to afford the desired ring product, **15** (8.5 mg, 72% yield) as a colorless oil.  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40 ~ 7.26 (m, 10H), 6.92 (s, 1H), 3.04 ~ 2.92 (m, 2H), 2.43 ~ 2.32 (m, 2H), 2.22 ~ 2.08 (m, 2H), 1.26 (s, 9H) ppm;  $^{13}\text{C-NMR}$  (150 MHz,  $\text{CDCl}_3$ )  $\delta$  169.4, 168.8, 139.6, 139.5, 128.5, 128.4, 128.1, 127.9, 127.4, 127.1, 82.7, 78.1, 67.2, 38.4, 34.0, 30.8, 27.5 ppm; IR (KBr) 2977, 1734, 1496, 1369, 1256, 1133, 1047, 993, 745  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$ :



[M+Na]<sup>+</sup> Calcd for C<sub>23</sub>H<sub>26</sub>O<sub>4</sub>SNa 421.1444; found 421.1439. [ $\alpha$ ]<sub>D</sub><sup>20</sup> = -3.33 (*c* 1.0, CHCl<sub>3</sub>).

## Acknowledgements

This work was supported by grants from the National Research Foundation of Korea (No. 2016R1A2B2008109) and BK21 Plus Program in 2016.

## Supporting Information

Spectral data of all new compounds as well as an X-ray crystallographic analysis of (*R*)-**10m**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References

- (1) 2009 Top 200 Branded Drugs by Retail Dollars. Drug Topics, 2010; [modernmedicine.com/drugtopics/data/articlestandard//drugtopics/252010/674961/article.pdf](http://modernmedicine.com/drugtopics/data/articlestandard//drugtopics/252010/674961/article.pdf).
- (2) Procter, D. J. *J. Chem. Soc., Perkin Trans. 1* **2001**, 335.
- (3) Shibasaki, M.; Kanai, M. *Chem. Rev.* **2008**, *108*, 2853.
- (4) Voss, J. J. *Sulfur Chem.* **2009**, *30*, 167.
- (5) For recent reviews on the chiral C-S bond formations, see: a) Clayden, J.; MacLellan, P. *Beilstein J. Org. Chem.* **2011**, *7*, 582.; b) Chauhan, P.; Mahajan, S.; Enders, D. *Chem. Rev.* **2014**, *114*, 8807.
- (6) a) Palomo, C.; Oiarbide, M.; Dias, F.; López, R.; Linden, A. *Angew. Chem., Int. Ed.* **2004**, *43*, 3307.; b) Palomo, C.; Oiarbide, M.; López, R.; González, P. B.; Gómez-Bengoa, E.; Saá, J. M.; Linden, A. *J. Am. Chem. Soc.* **2006**, *128*, 15236.; c) Avenoza, A.; Busto, J. H.; Jiménez-Osés, G.; Peregrina, J. M. *J. Org. Chem.* **2006**, *71*, 1692.; d) Avenoza, A.; Busto, J. H.; Jiménez-Osés, G.; Peregrina, J. M. *Org. Lett.* **2006**, *8*, 2855.; e) La Clair, J. J. *Angew. Chem., Int. Ed.* **2006**, *45*, 2769.; f) Lu, H.-H.; Zhang, F.-G.; Meng, X.-G.; Duan, S.-W.; Xiao, W.-J. *Org. Lett.* **2009**, *11*, 3946.; g) Weaver, J. D.; Morris, D. K.; Tunge, J. A. *Synlett* **2010**, 470.

1  
2  
3 (7) a) Marigo, M.; Wabnitz, T. C.; Fielenbach, D.; Jørgensen, K. A. *Angew. Chem. Int. Ed.* **2005**, *44*,  
4 794.; b) Zhao, G.-L.; Rios, R.; Vesely, J.; Eriksson, L.; Córdova, A. *Angew. Chem. Int. Ed.* **2008**, *47*,  
5 8468.; c) Lin, A.; Fang, L.; Zhu, X.; Zhu, C.; Cheng, Y. *Adv. Synth. Catal.* **2011**, *353*, 545.; d) Wang,  
6 C.; Yang, X.; Loh, C. C. J.; Raabe, G.; Enders, E. *Chem. Eur. J.* **2012**, *18*, 11531.; e) Cai, Y.; Li, J.;  
7 Chen, W.; Xie, M.; Liu, X.; Lin, L.; Feng, X. *Org. Lett.* **2012**, *14*, 2726.; f) Han, Z.; Chen, W.; Dong,  
8 S.; Yang, C.; Liu, H.; Pan, Y.; Yan, L.; Jiang, Z. *Org. Lett.* **2012**, *14*, 4670.; g) Shirakawa, S.;  
9 Tokuda, T.; Kasai, A.; Maruoka, K. *Org. Lett.* **2013**, *15*, 3350.

10  
11 (8) a) Sonawane, R. P.; Fröhlich, R.; Hoppe, D. *Chem. Commun.* **2006**, 3101.; b) MacLellan, P.;  
12 Clayden, J. *Chem. Commun.* **2011**, *47*, 3395.; c) Clayden, J.; Dufour, J.; Grainger, D. M.; Helliwell,  
13 M. J. *Am. Chem. Soc.* **2007**, *129*, 7488.; d) Clayden, J.; Donnard, M.; Lefranc, J.; Minassi, A.;  
14 Tetlow, D. J. *J. Am. Chem. Soc.* **2010**, *132*, 6624.; e) Tetlow, D. J.; Hennecke, U.; Raftery, J.;  
15 Waring, M. J.; Clarke, D. S.; Clayden, J. *Org. Lett.* **2010**, *12*, 5442.; f) Clayden, J.; Farnaby, W.;  
16 Grainger, D. M.; Hennecke, U.; Mancinelli, M.; Tetlow, D. J.; Hillier, I. H.; Vincent, M. A. *J. Am.*  
17 *Chem. Soc.* **2009**, *131*, 3410.; g) Fournier, A. M.; Brown, R. A.; Farnaby, W.; Miyatake-Onozabal,  
18 H.; Clayden, J. *Org. Lett.* **2010**, *12*, 2222.

19  
20 (9) For recent reviews on the phase-transfer catalysis, see: a) Maruoka, K.; Ooi, T. *Chem. Rev.* **2003**,  
21 *103*, 3013; b) O'Donnell, M. J. *Acc. Chem. Res.* **2004**, *37*, 506; c) Lygo, B.; Andrews, B. I. *Acc.*  
22 *Chem. Res.* **2004**, *37*, 518; d) Ooi, T.; Maruoka, K. *Angew. Chem., Int. Ed.* **2007**, *46*, 4222; e)  
23 Hashimoto, T.; Maruoka, K. *Chem. Rev.* **2007**, *107*, 5656; f) Jew, S.-s.; Park, H.-g. *Chem. Commun.*  
24 **2009**, 7090; g) Shirakawa, S.; Maruoka, K. *Angew. Chem. Int. Ed.* **2013**, *52*, 4312.

25  
26 (10) Hong, S.; Lee, J.; Kim, M.; Park, Y.; Park, C.; Kim, M.-h.; Jew, S.-s.; Park, H.-g. *J. Am. Chem.*  
27 *Soc.* **2011**, *133*, 4924.

28  
29 (11) Ha, M. W.; Lee, M.; Choi, S.; Kim, S.; Hong, S.; Park, H.-g. *J. Org. Chem.* **2015**, *80*, 3270.

30  
31 (12) Ha, M. W.; Choi, S.; Lee, J. Y.; Lee, J. K.; Lee, J.; Hong, S.; Park, H.-g. *RSC Adv.* **2016**, *6*,  
32 77427.

(13) a) Ooi, T.; Kameda, M.; Maruoka, K. *J. Am. Chem. Soc.* **2003**, *125*, 5139; b) Kitamura, M.; Shirakawa, S.; Maruoka, K. *Angew. Chem. Int. Ed.* **2005**, *44*, 1549; c) Lygo, B.; Wainwright, P. G. *Tetrahedron Lett.* **1997**, *38*, 8595; d) Corey, E. J.; Xu, F.; Noe, M. C. *J. Am. Chem. Soc.* **1997**, *119*, 12414; e) Jew, S.-s.; Yoo, M.-S.; Jeong, B.-S.; Park, H.-g. *Org. Lett.* **2002**, *4*, 4245.

(14) Bijvoet method of anomalous X-ray scattering was used for the enantiomer assignment of **10m** and the molecular structure show the atomic numbering and 50% probability displacement ellipsoid. CCDC 1533604 contains the supplementary crystallographic data for compound **10m**. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

(15) The corresponding free thiol from the hydrolysis of **10b** in alkali basic condition was unstable and desulfurized to the corresponding mono-allyl-malonate.

(16) Hakimelahi, G. H.; Just, G. *Tetrahedron Lett.* **1980**, *21*, 2119.

## TOC

