



Antitumor agents. 271: Total synthesis and evaluation of brazilein and analogs as anti-inflammatory and cytotoxic agents

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ABSTRACT

The first total synthesis of the naturally occurring tetracyclic homoisoflavonoid brazilein (**1**) and 14 new analogs (**1a–n**) is reported. Target compounds and intermediates were assayed for anti-inflammatory effects on superoxide anion generation and elastase release by human neutrophils in response to fMLP/CB, and for cytotoxic activity against nasopharyngeal (KB), vincristine-resistant nasopharyngeal (KBvin), lung (A549) and prostate (DU-145) human cancer cell lines. The most active compound **1b** showed potent effects on superoxide anion generation and elastase release with IC₅₀ values of 1.2 and 1.9 μM, respectively, and was 65 times more potent than phenylmethylsulfonyl fluoride (PMSF), the positive control, in the latter assay. Additionally, **1b** exhibited broad spectrum in vitro anticancer activity with IC₅₀ values of 6–11 μM against the four tested cancer cell lines.

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Naturally occurring homoisoflavonoids (3-benzylidene-4-chromanones) are related structurally to flavonoids, and exhibit various biological activities.¹ Synthesis of the 3-benzylidene-4-chromanone skeleton is usually based on the condensation of 4-chromanones with aromatic aldehydes in the presence of an acidic or basic catalyst.^{2,3} Although members of this family have been the subjects of several syntheses,^{4,5} there are no reports on the total synthesis of brazilein (**1**) nor the preparation of **1**-analogs.

In our prior study, we isolated the unique tetracyclic homoisoflavonoid brazilein (**1**), whose stereochemistry was obtained as dextrorotatory,⁶ from an ethyl acetate extract of *Caesalpinia sappan* L. (Legminosae). Brazilein ((6aR)-6a,7-dihydro-3,6a,10-trihydrobenz[b]indeno[1,2-d]pyran-9-(6H)-one) is a natural red pigment used in traditional Chinese medicine. It also exhibits significant cytotoxic activity against HepG2 and Hep3B (liver), MDA-MB-231 and MCF-7 (breast), A549 (lung), and Ca9-22 (gingival) human cancer cell lines (Table 1).^{6b} Previous literatures showed that brazilein (**1**) exhibited immunosuppressive activity in mice lymphocytes,⁷ cardiotoxic effects in isolated rat hearts,⁸ and anti-oxidant properties.^{9,10} Brazilin (**2**) is the hydrogenation product of brazilein (**1**) in the proposed biosynthetic route,¹¹ and can be oxidized to brazilein

(**1**) when the extract has been exposed to air and light. Beyond this redox chemistry, our interest in devising a general synthesis for this skeleton was based on literature reports that brazilin (**2**) has cytotoxicity (Table 1)⁶ and anti-inflammatory properties,^{12,13} acts as a micromolar telomerase inhibitor,¹⁴ and produces DNA nicks.¹⁵ These properties are often associated with useful anticancer agents. Additionally, to the best of our knowledge, brazilein (**1**) and its derivatives have not been studied for anti-inflammatory and cytotoxic effects. Considering the structural similarities between brazilin (**2**) and brazilein (**1**), we wanted to determine whether the latter exhibits a similar spectrum of in vitro anti-inflammatory and anticancer effects.

We report herein a strategy that enables the preparation of **1** by construction of the benzylidene moiety via acid- or base-catalyzed

Table 1
Cytotoxic effects of brazilein (**1**) and brazilin (**2**)

Compd	Cancer cell lines (IC ₅₀ value ^a , μg/mL)					
	HepG2	Hep3B	MDA-MB-231	MCF7	A549	Ca9-22
Brazilein (1)	3.15	3.42	2.36	3.96	9.68	8.63
Brazilin (2)	3.47	4.54	3.09	6.29	15.56	9.78
Doxorubicin ^b	0.34	0.56	1.05	0.42	0.32	0.14

^a IC₅₀ represents the 50% inhibitory concentration.

^b Positive control.

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aldol condensation with aryl aldehydes. All newly synthesized compounds, including structurally related intermediates, were assayed for in vitro cytotoxicity against four human cancer cell lines, [KB (nasopharyngeal), KBvin (multidrug-resistant nasopharyngeal over-expressing P-gp), A549 (lung), and DU-145 (prostate)] and for anti-inflammatory action in terms of superoxide anion generation and elastase release by human neutrophils in response to fMLP/CB.

The general synthesis of brazilein (**1**) was achieved through the route outlined in Scheme 1. 7-Hydroxy-4-chromanone (**5**), a key intermediate, was readily obtained by reaction of resorcinol (**3**) and 3-chloropropionic acid promoted by trifluoromethanesulfonic acid to give 2',4'-dihydroxy-3-chloropropiophenone (**4**), which was further cyclized using 2 M NaOH¹⁶ to give **5** in 78% yield. Treatment of **5** with iodomethane in the presence of K₂CO₃/acetone gave **6** (60%). Acid catalyzed¹⁷ condensation of **6** with 3,4-dimethoxybenzaldehyde afforded **7** (78%). Epoxidation of **7** with alkaline hydrogen peroxide gave epoxy ketone **8** (90%), which was reduced to the diol **9** (89%) with LiAlH₄.³ Cyclization to the desired product was affected by HClO₄³ to yield trimethylbrazilein **10** in 31% yield. Compound **10** was demethylated using pyridine hydrochloride¹⁸ to give **2** in 36% yield. Brazilein (**2**) was oxidized with a catalytic amount of iodine solution to give brazilein (**1**) in 78% yield.

In addition, the brazilein derivatives **1a**, **1b/c**, and **1d** were obtained by esterification of **1** with acetic anhydride, propionyl chloride, and benzoic chloride, respectively, in pyridine. For **1e** and **1f**, **1** was reacted in strong base (NaH) with dimethylcarbamyl chloride and diethylcarbamyl chloride respectively. Compounds **1g–n** were synthesized by the general Williamson method as described in previous reports.^{19,20} Specifically, **1g–n** were prepared by alkylation of **1** with iodomethane (**1g** and **h**), iodoethane (**1i** and **j**), benzyl bromide (**1k**), allylbromide (**1l** and **m**), and prenyl bromide (**1n**), in the presence of K₂CO₃ in acetone. Accordingly, the first total syntheses of **1** and 14 analogs (**1a–n**) (Fig. 1) were accomplished.

Anti-inflammatory activity: Synthesized **1**, **2**, and intermediates **7–10** together with analogs **1a–n**, which are divided into three classes based on the substitution pattern at C-3, C-6a, and C-10 and functionalized into ester, carbamate, and ether, were evaluated for anti-inflammatory action based on effects against superoxide anion generation and elastase release by human neutrophils in response to fMLP/CB. The assays were performed using established protocols,^{21,22} which are widely used to identify potential anti-inflammatory compounds. Table 2 lists the results for the test compounds, as well as diphenyleioidonium (DPI) and phenylmethylsulfonyl fluoride (PMSF), included as positive

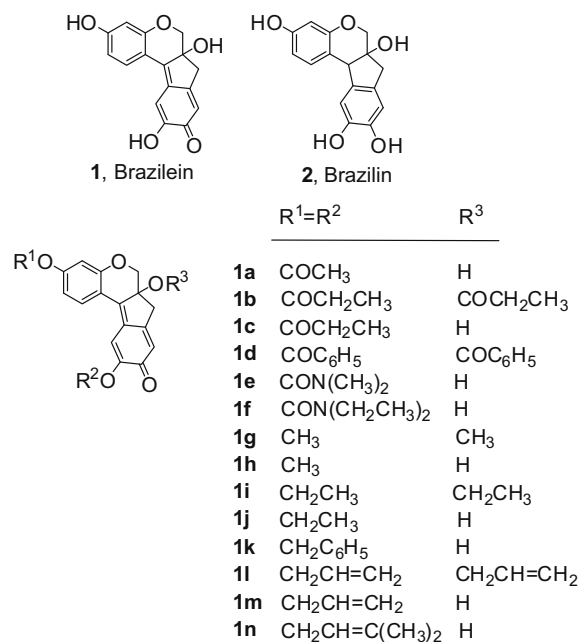
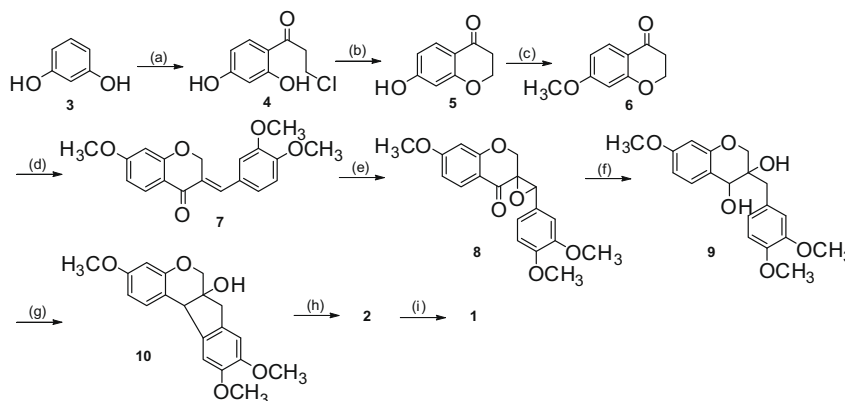


Figure 1. Molecular structures of brazilein (**1**), brazilein (**2**), and synthetic **1**-analogs.

controls for superoxide anion generation and elastase release, respectively. Results are expressed as the mean \pm S.E.M., and comparisons were made using Student's *t*-test. A probability of 0.05 or less was considered significant.

Among the esterified compounds **1a–d**, analogs **1a** (di-acetate) and **1d** (tri-benzoate) showed selective anti-inflammatory activity (**1a** vs superoxide anion generation, 18.7 μ M; **1d** vs elastase release, 13.3 μ M). Analog **1b** (tri-propionate) showed the highest potency against both superoxide anion generation and elastase release with significant IC₅₀ values of 1.2 and 1.9 μ M, respectively. Analog **1c** (di-propionate) was slightly less potent with IC₅₀ values of 1.65 and 3.28 μ M, but still was more active than the parent compound **1**. Therefore, replacing the hydroxy groups with propionate esters increased potency in both assays, particularly when all three hydroxyls were esterified in **1b**.

N-Diethyl carbamate **1f** showed a weak effect (IC₅₀ 15.5 μ M) in response to fMLP/CB-induced elastase release, but neither **1f** nor the *N*-dimethyl carbamate **1e** showed a significant effect against superoxide anion generation. Although the ester and carbamate functional group present parallel physical and chemical properties,



Scheme 1. Total synthesis of brazilein (**1**). Reagents and conditions: (a) 3-Chloropropionic acid, CF₃SO₃H, 80 °C, 1 h, 92%; (b) 2 M NaOH, 0 °C, 2 h, 78%; (c) iodomethane, K₂CO₃, acetone, 60 °C, 5 h, 60%; (d) 3,4-dimethoxybenzaldehyde, EtOH, HCl gas, rt, overnight, 78%; (e) H₂O₂, dioxane, rt, 24 h, 90%; (f) LiAlH₄, THF, rt, overnight, 89%; (g) HClO₄, acetic acid, rt, overnight, 31%; (h) pyridine–HCl, 190–200 °C, 3 h, 36%; (i) 1 N I₂(aq), rt, overnight, 78%.

Table 2

Inhibitory effects of compounds on superoxide anion generation and elastase release by human neutrophils in response to FMLP/CB

Compd	Superoxide anion IC ₅₀ (μM) ^a or (Inh%)	Elastase release IC ₅₀ (μM) ^a or (Inh%)
Brazilin (1)	4.0 ± 0.42	14.2 ± 1.32
Brazilin (2)	4.6 ± 0.63	16.7 ± 0.41
1a	18.71 ± 2.82	(40.87 ± 3.49)***
1b	1.2 ± 0.02	1.9 ± 0.31
1c	1.65 ± 0.09	3.28 ± 0.51
1d	(14.14 ± 5.70)*	13.25 ± 1.14
1e	(16.37 ± 3.71)	(11.39 ± 4.76)
1f	(−7.00 ± 4.90)	15.50 ± 2.40
1g	4.65 ± 1.31	13.6 ± 1.09
1h	(18.61 ± 6.29)*	(26.76 ± 6.60)*
1i	23.94 ± 5.48	(35.54 ± 1.71)***
1j	4.35 ± 0.91	8.37 ± 0.41
1k	(33.38 ± 7.75)**	8.76 ± 1.51
1l	16.16 ± 0.40	11.83 ± 2.00
1m	19.72 ± 4.44	16.54 ± 1.70
1n	19.69 ± 5.82	10.06 ± 0.60
7	NT ^b	NT ^b
8	1.90 ± 0.32	(43.71 ± 2.00)***
9	8.38 ± 0.02	10.25 ± 0.14
10	4.52 ± 1.21	(35.76 ± 2.60)*
DPI ^c	0.7 ± 0.4	
PMSF ^c		130.9 ± 2.91

^a IC₅₀ represents the 50% inhibitory concentration of the compound. If 50% inhibition was not reached at any test dose, the percentage of inhibition obtained at a test dose of 10 μg/mL is given in parentheses (Inh%). Results are presented as mean ± S.E.M. (n = 3–5). *p < 0.05, **p < 0.01, and ***p < 0.001 compared with the control value.

^b Abnormal absorption at OD 405 and 550 nm.

^c DPI and PMSF were used as positive controls.

the data showed that the former did provide better activity against either superoxide anion generation or elastase release.

Among the eight compounds (**1g–n**) containing ether substituents, **1g** (tri-methoxy) and **1j** (di-ethoxy) displayed potent effects in the superoxide anion assay (IC₅₀ 4.7 and 4.4 μM, respectively). Analog **1j** (di-ethoxy) also showed a moderate effect (IC₅₀ 8.4 μM) against elastase release, as did **1k** (di-benzoyl) (IC₅₀ 8.8 μM). Compounds **1l–n** (allyl and prenyl ethers) were nonselective, but with only weak inhibitory effects in both assays (IC₅₀ 10–20 μM). These findings suggest that **1g** and **1j** merit further investigation as potential anti-inflammatory compounds.

Furthermore, among the synthetic intermediates **7–10** (Table 1), the epoxy ketone **8** was significantly potent and selective (IC₅₀ 1.9 μM) against superoxide anion generation. The diol **9** was moderately active in both anti-inflammatory assays (IC₅₀ 8.4 and 10.3 μM), while the tetracyclic **10** was potent only against superoxide anion generation (IC₅₀ 4.5 μM).

In conclusion, among all screened compounds, including ester (**1a–d**), carbamate (**1e** and **1f**), and ether (**1g–n**) derivatives, the preliminary structure–activity relationship were addressed in the order: a) ester > ether > carbamate; b) ethyl = methyl > allyl = prenyl > benzyl. In particular, the most potent analog **1b** was 65-fold more active than PMSF, the positive control, in the elastase release assay.

Anti-cancer activity: The homoisoflavonoids **1a–n** as well as the synthesized **1**, **2** and intermediates **7–10** were examined for in vitro cytotoxic activity against four human cancer lines. Table 3 lists the IC₅₀ values obtained with test compounds compared to the anticancer drug paclitaxel as a positive control. Among the 14 derivatives (**1a–n**), seven compounds (**1a–c**, **1g**, **1i**, **1j**, and **1m**) showed moderate potency against the tested human cancer cell lines with IC₅₀ values of 5–18 μM; however, they all were much less active than paclitaxel.

In our previous study, the parent compound **1** showed moderate cytotoxic activity against six cancer cell lines (IC₅₀ 8–34 μM,

Table 3Cytotoxic effects of **1**-analogs

Compd	Cancer cell line (IC ₅₀ value ^a , μM)			
	KB	KBvin	A549	DU-145
1a	10.0 ± 1.3	ND ^b	8.1 ± 1.2	8.2 ± 0.72
1b	6.6 ± 2.21	11.3 ± 1.53	13.5 ± 1.12	11.0 ± 0.47
1c	7.7 ± 3.5	ND ^b	7.8 ± 0.65	7.1 ± 0.44
1g	15.2 ± 1.41	16.9 ± 0.42	>20	17.5 ± 0.67
1i	18.9 ± 1.11	17.9 ± 0.9	>20	18.1 ± 1.54
1j	11.3 ± 0.78	16.5 ± 1.56	15.3 ± 1.04	11.0 ± 0.67
1m	5.4 ± 0.29	ND ^b	9.4 ± 0.49	9.1 ± 0.75
7	>20	18.3 ± 0.26	>20	>20
8	9.8 ± 1.28	6.4 ± 0.67	>20	10.6 ± 0.42
9	>20	>20	>20	>20
10	10.5 ± 0.48	7.6 ± 0.73	>20	13.8 ± 1.26
Paclitaxel ^c	1.55 × 10 ^{−3}	1.09 × 10 ^{−3}	1.93 × 10 ^{−3}	2.35 × 10 ^{−3}

^a Data are expressed as mean ± SD (n = 3).

^b ND: not determined.

^c Positive control.

Table 1).^{6b} Comparison of **1** to its synthetic intermediates **2** and **7–10** showed that the epoxy ketone **8** had comparable or enhanced activity, while the reduced diol **9** was inactive (IC₅₀ > 20 μM). In addition, trimethylbrazilin **10** was more active than **2** with IC₅₀ values of 10–15 μM against all four human cancer cell lines. The enhanced cytotoxicity of epoxide analog **8** could account for its twofold greater potency than **1** against superoxide anion generation.

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References and notes

- Lockhart, I. M. In Ellis, G. P., Ed.; *The Chemistry of Heterocyclic Compounds; Chromenes, Chromanones and Chromones*; John Wiley & Sons: New York, 1977.
- Malhotra, S.; Sharma, V. K.; Parmar, V. S. *J. Chem. Res.* **1988**, 179.
- Farkas, L.; Gottsegen, A.; Norgradi, M. *Tetrahedron* **1970**, *26*, 2787.
- Huang, Y. D.; Zhang, J.; Pettus, T. *Org. Lett.* **2005**, *26*, 5841.
- Davis, F.; Chen, B. C. *J. Org. Chem.* **1993**, *58*, 1751.
- (a) Kim, D. S.; Baek, N. I.; Oh, S. R.; Jung, K. Y.; Lee, I. S.; Lee, H. K. *Phytochemistry* **1997**, *46*, 177; (b) Lai, W. C. Master Thesis, Kaohsiung Medical University, July 2007.
- Ye, M.; Xie, W. D.; Lei, F.; Meng, Z.; Zhao, Y. N.; Su, H.; Du, L. J. *Int. Immunopharmacol.* **2006**, *6*, 426.
- Zhao, Y. N.; Pan, Y.; Tao, J. L.; Xing, D. M.; Du, L. J. *Pharmacology* **2006**, *76*, 76.
- Kabbash, A.; Yagi, A.; Ishizu, T.; Haraguchi, H.; Fujioka, T.; Moustafa, S. M.; El-Bassuony, A. A. *Saudi Pharm. J.* **2008**, *16*, 25.
- Hu, J.; Yan, X.; Wang, W.; Wu, H.; Hua, L.; Du Lijun. *Tsinghua Sci. Technol.* **2008**, *13*, 474.
- Hikino, H.; Taguchi, T.; Fujimura, H.; Hiramoto, Y. *Planta Med.* **1977**, *31*, 214.
- Bae, I. K.; Min, H. Y.; Han, A. R.; Seo, E. K.; Lee, S. K. *Eur. J. Pharmacol.* **2005**, *513*, 237.
- Nagai, M.; Nagumo, S.; Lee, S. M.; Eguchi, I.; Kawai, K. I. *Chem. Pharm. Bull.* **1986**, *34*, 1.
- Tolman, R. L.; Chin, A. C. WO Patent 0,193,864, Dec 13, 2001.
- Mar, W.; Lee, H. T.; Je, K. H.; Choi, H. Y.; Seo, E. K. *Arch. Pharmacol. Res.* **2003**, *26*, 147.
- Namikoshi, M.; Nakata, H.; Yamada, H.; Nagai, M.; Saitoh, T. *Chem. Pharm. Bull.* **1987**, *35*, 2761.
- Venkateswarlu, S.; Panchanula, G. P.; Guraiah, M. B.; Subbaraju, G. V. *Tetrahedron* **2005**, *61*, 3013.
- Siddaiah, V.; Rao, C. V.; Venkateswarlu, S.; Krishnaraju, A. V.; Subbaraju, G. V. *Bioorg. Med. Chem.* **2006**, *14*, 2545.
- Gao, G. Y.; Li, D. J.; Keung, W. M. *Bioorg. Med. Chem.* **2003**, *11*, 4069.
- Hosoda, S.; Hashimoto, Y. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 5414.
- Chang, H. L.; Chang, F. R.; Chen, J. S.; Wang, H. P.; Wu, Y. H.; Wang, C. C.; Wu, Y. C.; Hwang, T. L. *Eur. J. Pharmacol.* **2008**, *586*, 332.
- Hwang, T. L.; Yeh, S. H.; Leu, Y. L.; Chern, C. Y.; Hsu, H. C. *Br. J. Pharmacol.* **2006**, *148*, 78.