

Iridium Complex-Catalyzed Highly Selective Cross [2 + 2 + 2] Cycloaddition of Two Different Monoynes: 2:1 Coupling versus 1:2 Coupling

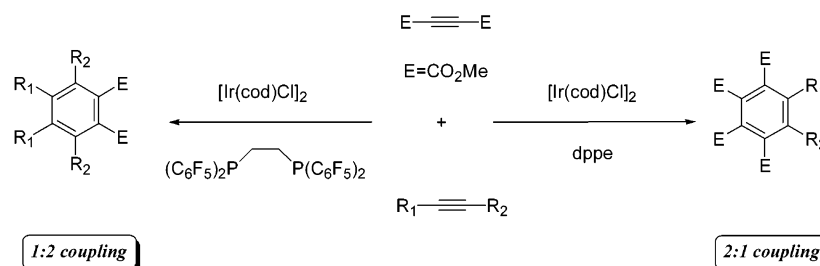
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ABSTRACT



Highly selective cross [2 + 2 + 2] cycloaddition of two different monoynes is achieved by using a catalytic amount of $[\text{Ir(cod)Cl}]_2$ and ligand. The ligand had a considerable effect on the reaction. When 1,2-bis(diphenylphosphino)ethane was used, two molecules of dimethyl acetylenedicarboxylate (DMAD) reacted with one molecule of a monoyne to give the 2:1 coupling product. When 1,2-bis(dipentafluorophenylphosphino)ethane was used instead of dppe, one molecule of DMAD reacted with two molecules of a monoyne to give the 1:2 coupling product.

In the course of our study on iridium complex-catalyzed organic synthesis,¹ we previously found that $[\text{Ir(cod)Cl}]_2/\text{dppe}$ is an efficient catalyst for [2 + 2 + 2]cycloaddition of α,ω -diynes with monoynes.² We report here the extension of this chemistry to the highly selective cross [2 + 2 + 2] cycloaddition of two different monoynes (Scheme 1). The

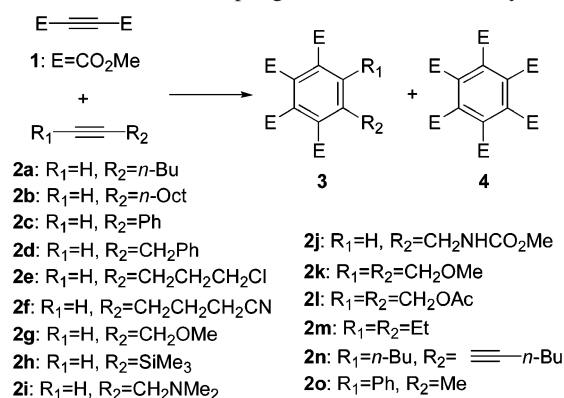
[2 + 2 + 2] cycloaddition of monoynes is one of the most straightforward routes to polysubstituted benzenes, but the drawback of this route is selectivity. The highly selective cross [2 + 2 + 2] cycloaddition of two or three different monoynes is a challenging problem.³

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Scheme 1. 2:1 Coupling of DMAD with Monoynes

The rich chemistry of metallacycles gives various useful transformations in selective organic synthesis.⁴ Both late and early transition metals can be used for this chemistry. Catalytic reactions via nickelacycles,⁵ palladacycles,⁶ cobaltacycles,⁷ rhodacycles⁸ and ruthenacycles⁹ have been extensively studied and used for the synthesis of cyclic compounds from an acyclic substrate. Since Collman et al. first reported the synthesis of an iridacyclopentadiene from the reaction of IrCl(N₂)(PPh₃)₂ with DMAD (**1**) in 1968,¹⁰ the reactivity and structure of iridacyclopentadiene have been studied.¹¹ However, there have been few reports on catalytic organic synthesis via iridacyclopentadiene.¹²

The reaction of **1** with 1-hexyne (**2a**) in the presence of a catalytic amount of [Ir(cod)Cl]₂/dppe gave **3a** and **4**.^{13,14} Product **3a** resulted from the 2:1 coupling of two molecules of **1** with one molecule of **2a**. Product **4** resulted from

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Table 1. 2:1 Coupling of DMAD with 1-Hexyne (**2a**)^a

entry	1 (mmol)	2a (mmol)	time (h)	yield of 3a (%) ^b	yield of 4 (%) ^b
1	2	1	0.5	89	11
2	2	1.2	1	98	2
3	2	2	1	98	2
4	2	5	24	97	3

^a A mixture of **1**, **2a**, [Ir(cod)Cl]₂ (0.02 mmol), dppe (0.04 mmol), and toluene (5 mL) was stirred under refluxing toluene. ^b Isolated yield based on **1**.

cyclotrimerization of **1**. The molar ratio of **1** to **2a** affected the yield of **3a**. The results are summarized in Table 1. The reaction using 0.6 equiv of **2a** gave **3a** in 98% yield (entry 2). Cyclotrimerization of **1** was not suppressed completely. Product **4** was obtained in 2% yield.

On the basis of these results, we examined the scope of the 2:1 coupling. The results are summarized in Table 2.

Table 2. 2:1 Coupling of DMAD with **2**^a

entry	2	conditions	product	yield 3 (%) ^b	yield 4 (%) ^b
1	2b	toluene reflux 1 h	3b	96	2
2	2c	xylene reflux 20 h	3c	38	50
3	2d	toluene reflux 5 h	3d	86	3
4	2e	toluene reflux 5 h	3e	83	2
5 ^c	2f	THF reflux 12 h	3f	91	0
6 ^c	2g	THF reflux 0.5 h	3g	89	0
7 ^c	2h	THF reflux 20 h	3h	52	16
8 ^c	2i	dioxane reflux 20 h	3i	41	0
9 ^c	2j	THF reflux 1 h	3j	90	0
10	2k	THF 50 °C 2 h	3k	98	0
11	2l	THF 50 °C 1 h	3l	93	0
12 ^d	2m	THF 50 °C 1 h	3m	89	10
13 ^{d,e}	2n	THF 50 °C 2 h	3n	73	11
14 ^{d,f}	2o	THF 50 °C 3.5 h	3o	78	15

^a A mixture of **1** (2 mmol), **2** (1.2 mmol), [Ir(cod)Cl]₂ (0.02 mmol), dppe (0.04 mmol), and solvent (5 mL) was stirred. ^b Isolated yield based on **1**. ^c [Ir(cod)Cl]₂ (0.03 mmol), dppe (0.06 mmol). ^d **2** (2 mmol). ^e Slow addition of **1** for 0.5 h. ^f Slow addition of **1** for 2 h.

DMAD smoothly reacted with 0.6 equiv of various monoynes (**2**) under mild conditions. The reaction with 1-decyne (**2b**) proceeded at room temperature to give a product in high yield (entry 1). The reaction with phenylacetylene (**2c**)

(13) **General Procedure.** A typical procedure is described (Table 1, entry 2). To a toluene solution (5 mL) of [Ir(cod)Cl]₂ (13.4 mg, 0.02 mmol) and dppe (15.9 mg, 0.04 mmol) was added 1-hexyne (0.099 g, 1.2 mmol) via a syringe. Dimethyl acetylenedicarboxylate (0.284 g, 2 mmol) was then added to the solution by a syringe. The reaction mixture was stirred under reflux for 1 h. The progress of the reaction was monitored by GLC. After dimethyl acetylenedicarboxylate was consumed, toluene was evaporated in vacuo. Column chromatography of the residue gave **3a** as a colorless oil (*n*-hexane/AcOEt = 80/20, 0.359 g, yield 98%) and **4a** as a colorless oil (*n*-hexane/AcOEt = 60/40, 0.003 g, yield 2%).

(14) Preparation of [Ir(cod)Cl]₂, see: Crabtree, R. H.; Quirk, J. M.; Felkin, H.; Fillebeen-Khan, T. *Synth. React. Inorg. Met.-Org. Chem.* **1982**, *12*, 407.

In conclusion, we have found a highly selective cross [2 + 2 + 2] cycloaddition of two different monoynes catalyzed by $[\text{Ir}(\text{cod})\text{Cl}]_2/\text{ligand}$ under mild conditions. The reaction is quite useful and practical for the synthesis of polysubstituted benzene derivatives. The experimental procedure for the reaction is quite simple. Monoynes are added to the solution of a catalyst, and then DMAD (**1**) is added. Slow addition of the substrate using a syringe pump is unnecessary except when **2n** or **2o** is used. Extension of the scope of this reaction and mechanistic studies are underway.

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Supporting Information Available: Experimental procedures and compound characterization data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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