

A Simple Synthesis of 2-Substituted-4*H*-3,1-benzoxazin-4-ones by Palladium-Catalyzed Cyclocarbonylation of *o*-Iodoanilines with Acid Chlorides

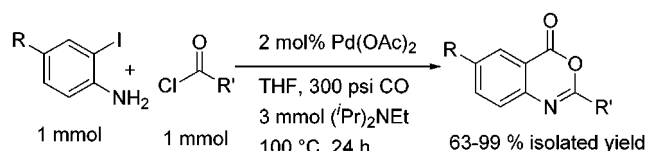
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ABSTRACT



One-pot reaction of *o*-iodoanilines with acid chlorides and carbon monoxide, in the presence of a palladium catalyst and diisopropylethylamine, regioselectively affords 2-substituted-4*H*-3,1-benzoxazin-4-ones in good to excellent yields. The reaction is believed to proceed via *in situ* amide formation from an *o*-iodoaniline and an acid chloride, followed by oxidative addition to Pd(0), CO insertion, and intramolecular cyclization to form the 2-substituted-4*H*-3,1-benzoxazin-4-one derivatives.

4*H*-3,1-Benzoxazin-4-ones (acylanthranils) are a class of fused heterocycles of considerable interest owing to their biological activity.¹ Indeed some of these compounds act as chymotrypsin inactivators,^{1a} inhibitors of human leukocyte elastase,^{1b,c} serine protease,^{1d} and 2-aryl derivatives have the ability to lower the concentration of plasma cholesterol and triglyceride.^{1e} Moreover, 2-substituted-4*H*-3,1-benzoxazin-4-ones were reported to be used as precursors for the preparation of pharmaceutically active compounds such as antimicrobial agents (*N*-substituted-quinazolin-4-one derivatives)² and analgesics (4-hydroxy-3-quinoline-carboxamides).³ Several methods have been reported for the

preparation of 2-substituted-4*H*-3,1-benzoxazin-4-ones.⁴ The most popular synthetic pathways involve the use of anthranilic acid or its derivatives,⁵ *N*-acylanthranilic acids,⁶ or isatonic anhydride.⁷ Other synthetic methods such as oxidation of indoles,⁸ [4 + 2] cycloaddition of 1,2,3-benzotriazin-

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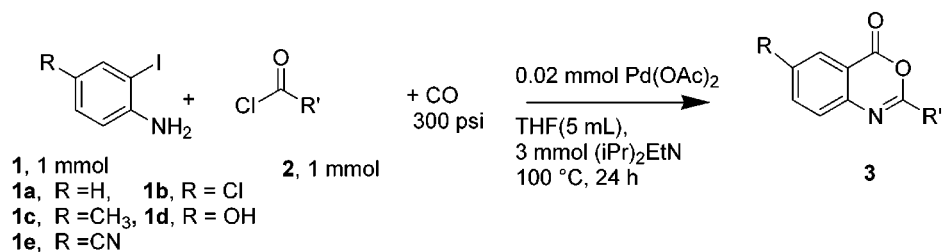
(4) For a recent review in the chemistry of 4*H*-3,1-benzoxazin-4-ones, see: Coppola, G. M. *J. Heterocycl. Chem.* **1999**, *36*, 563 and references therein.

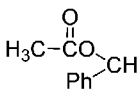
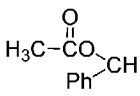
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Table 1. Cyclocarbonylation of *o*-Iodoanilines (**1**) with Acid Chlorides (**2**) Catalyzed by Palladium Acetate^a

entry	1	R'COCl, 2 , R' =	product	isolated yield ^b (%)
1	1a	CH ₃ 2a	3a	99 ^c
2	1a	(Ph) ₂ CH 2b	3b	90
3	1b	2b	3c	96
4	1c	2b	3d	94
5	1d	2b	3e	94
6	1e	2b	3f	80
7	1a	PhCH ₂ 2c	3g	63
8	1c	2c	3h	81
9	1b	<i>p</i> -ClC ₆ H ₄ CH ₂ 2d	3i	67
10	1c	2d	3j	71
11	1a	Ph(CH ₂ CH ₂)CH 2e	3k	95
12	1b	2e	3l	94
13	1e	 2e	3m	86
14	1b	 2f	3n	95
15	1c	2f	3o	91
16	1a	PhSCH ₂ 2g	3p	80
17	1c	2g	3q	91
18	1a	Ph 2h	3r	98
19	1c	<i>o</i> -CH ₃ OC ₆ H ₄ 2i	3s	89
20	1a	PhCH=CH 2j	3t	98
21	1a	<i>t</i> -Bu 2k	3u	98

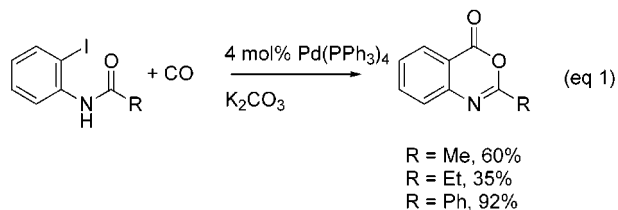
^a All reactions were conducted in THF (5 mL) using **1** (1 mmol), **2** (1 mmol), (*i*-Pr)₂NEt (3 mmol), Pd(OAc)₂ (0.02 mmol), and 300 psi of CO, at 100 °C for 24 h. ^b Isolated yield following silica gel column chromatography. ^c 0.02 mmol of dppf was used in this reaction.

4-ones with benzaldehydes,⁹ electrochemical cyclization of *o*-trichloroacetylanilides,¹⁰ and solid-phase synthesis¹¹ were described.

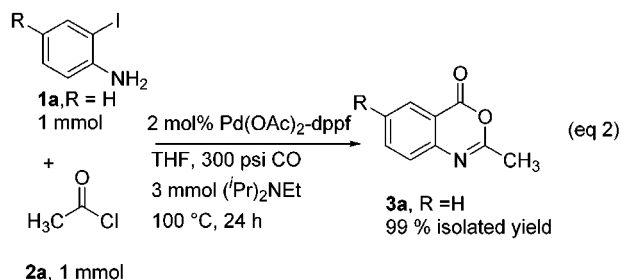
Two papers have reported the use of palladium-catalyzed

carbonylation methodology for the preparation of 2-substituted-4*H*-3,1-benzoxazin-4-ones. For example, thallation and subsequent palladium-catalyzed carbonylation of *N*-acetylaniline¹² and Pd(0)-catalyzed carbonylation of *o*-iodoanilines

with unsaturated halides or triflates have been reported for the synthesis of 2-substituted-4*H*-3,1-benzoxazin-4-ones.¹³ In the latter publication, the authors observed that *o*-acylamidoiodobenzene can undergo annulation to give 2-substituted-4*H*-3,1-benzoxazin-4-ones in the presence of K₂CO₃ and 4 mol % of Pd(PPh₃)₄ under an atmosphere of carbon monoxide (eq 1).



We envisioned that the cyclocarbonylation would be accessible by directly using *o*-iodoaniline and acid chlorides in the reaction under carbon monoxide pressure in the presence of base and palladium catalyst. Therefore, we initiated our investigation by using a mixture of *o*-iodoaniline (**1a**, R = H), 1 equiv of acetyl chloride (**2a**), 300 psi of CO, 3 equiv of (iPr)₂NEt, and 2 mol % of Pd(OAc)₂-1,1'-bis-(diphenylphosphino)ferrocene (dppf) in THF, and after 24 h at 100 °C **3a** was isolated in 99% yield (eq 2). This encouraging result led us to examine the one-pot three-component reaction for the formation of the pharmaceutically interesting 2-substituted-4*H*-3,1-benzoxazin-4-ones (**3**) by palladium-catalyzed cyclocarbonylation reactions of *o*-iodoanilines (**1**) with a variety of acid chlorides (**2**). Herein we wish to report the results from this investigation.



Treatment of *o*-iodoanilines (**1**) with carbon monoxide and a variety of acid chlorides (**2**) in the presence of 2 mol % of Pd(OAc)₂ in THF at 100 °C afforded 2-substituted-4*H*-3,1-benzoxazin-4-ones (**3**) in good to excellent yields (Table 1).¹⁴ A phosphine ligand is not necessary for this reaction, e.g., reactions of *o*-iodoanilines (**1a–c**) with diphenylacetyl chloride using 2 mol % of Pd(OAc)₂ and dppf in THF at 100 °C under 300 psi of CO for 24 h resulted in the formation

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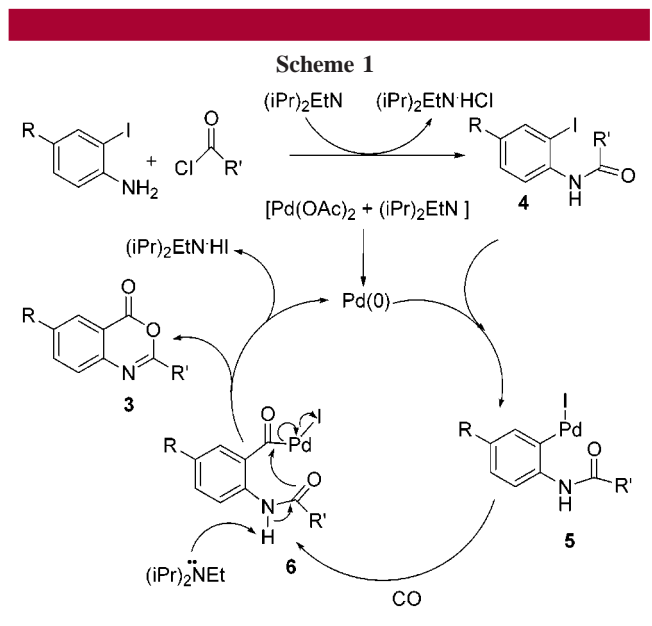
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of **3b**, **3c**, and **3d** in 88, 94, and 90% yields, respectively. Without dppf, the yields were 90, 96, and 94% (entries 2–4). The palladium catalyst is needed to form the desired product as the corresponding amide was formed in 96% yield from the reaction of **1a** with **2b** in THF under 300 psi of CO in the absence of Pd(OAc)₂. Both electron rich and electron poor *o*-iodoanilines reacted with acid chlorides to form 2-substituted-4*H*-3,1-benzoxazin-4-ones in fine yields. This annulation process tolerates nitrile as well as hydroxyl substituents on the *o*-iodoaniline (entries 5, 6, and 13). In general using α -di- or trisubstituted acid chlorides affords better product yields (entries 2–6, 11–15, 18, 19, and 21) compared to those of the α -monosubstituted analogues (entries 7–10, 16, and 17). When cinnamoyl chloride (**2j**) was used for the palladium-catalyzed cyclocarbonylation reaction with **1a**, 2-styryl-4*H*-3,1-benzoxazin-4-one **3t** having the *E* configuration was isolated in excellent yield (entry 20). These results demonstrate the superiority of this method in comparison to previously reported Pd-based syntheses of **3**, e.g., **3a** was formed in 99% while 40 and 60% yields of **3a** were obtained from the thallation–palladium carbonylation of *N*-acetylaniline¹² and Pd(0)-catalyzed carbonylation of *N*-acetyl-*o*-iodoaniline,¹³ respectively. In addition, **3s** was isolated in 89% yield while the yield of a similar compound, 2-(*o*-methoxyphenyl)-4*H*-3,1-benzoxazin-4-one, was 29% by Pd(0)-catalyzed carbonylation of *o*-methoxyiodobenzene with *o*-iodoaniline; also **3t** was obtained in 98% yield by following the reaction described herein while the yields were only 41 and 47% when β -bromostyrene and *o*-iodoaniline were subjected to carbonylation.¹³

The cyclocarbonylation reaction appears to proceed via in situ formation of amide (**4**) from the reaction of *o*-iodoanilines with acid chlorides in the presence of base. Oxidative addition of **4** to the in situ generated palladium(0) species¹⁵ leads to complex **5**. Carbon monoxide insertion into the aryl carbon–palladium bond of **5** would afford the arylpalladium iodide complex **6**. Cyclization is presumably facilitated by a deprotonation of the amide proton (or the



proton of the enol tautomer of **6**) by diisopropylethylamine to give the desired product with regeneration of palladium(0) (Scheme 1).

In conclusion, we have demonstrated that 2-substituted-4*H*-3,1-benzoxazin-4-one derivatives can be easily synthe-

sized by reaction of *o*-iodoanilines, acid chlorides, and carbon monoxide in the presence of a palladium catalyst and diisopropylethylamine. The present methodology is a versatile synthetic approach for the one-pot three-component reaction leading to 2-substituted-4*H*-3,1-benzoxazin-4-ones.

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Acknowledgment. We are grateful to the Natural Sciences and Engineering Research Council of Canada for supporting this research.

Supporting Information Available: General experimental procedures for the carbonylation reactions and characterization data for all products. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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