

Available online at www.sciencedirect.com





European Journal of Medicinal Chemistry 43 (2008) 2699-2716

Original article

http://www.elsevier.com/locate/ejmech

Tryptophan-containing dipeptide derivatives as potent PPAR γ antagonists: Design, synthesis, biological evaluation, and molecular modeling

Guanghui Deng^a, Zhiguo Liu^a, Fei Ye^a, Xiaomin Luo^a, Weiliang Zhu^a, Xu Shen^{a,b}, Hong Liu^{a,*}, Hualiang Jiang^{a,b}

^a Drug Discovery and Design Centre, State Key Laboratory of Drug Research, Shanghai Institute of Materia Medica, Graduate School of the Chinese Academy of Sciences, Shanghai Institutes for Biological Sciences,

Chinese Academy of Sciences, 555 Zuchongzhi Road, Zhangjiang Hi-Tech Park, Shanghai 201203, China

^b School of Pharmacy, East China University of Science and Technology, Shanghai 200237, China

Received 20 August 2007; received in revised form 15 January 2008; accepted 18 January 2008 Available online 3 February 2008

Abstract

The discovery of peroxisome proliferator-activated receptor γ (PPAR γ) antagonists (also termed "selective PPAR γ modulators, SPPAR γ M") is now of a great interest in the treatment of diabetes and obesity. The structure of compound **1a** (G3335, Fig. 1), a novel class of PPAR γ antagonist, is entirely different from that of other reported PPAR γ antagonists. A series of 35 novel analogues (**1b–l**, **9a–d**, **13a–t**) were designed, synthesized and evaluated against the agonistic effects exerted by rosiglitazone. These results indicated that most functional groups of **1a** were conserved, and six new compounds (**1b**, **1c**, and **9a–d**) exhibited strong PPAR γ antagonistic activities (IC₅₀ values of 5.2–25.8 μ M) against 10 μ M rosiglitazone in the promotion of the PPAR γ –LBD–CBP (ligand-binding domain and cAMP-response-element binding protein) interaction as investigated by yeast two-hybrid technology based assay. Molecular modeling studies for compounds **1a–d**, **1h**, **9c–d**, and **13a** were also presented.

© 2008 Elsevier Masson SAS. All rights reserved.

Keywords: Difficult sequence; Dipeptide; PPAR; SAR; Tryptophan

1. Introduction

Peroxisome proliferator-activated receptors (PPARs) are transcription factors belonging to the family of ligand-inducible nuclear receptors [1,2] and have been determined as attractive targets for pharmaceutical intervention of metabolic syndrome [3,4]. PPARs function by forming heterodimers with the 9*cis*-retinoic acid receptor (RXR) and subsequently binding to specific peroxisome proliferator response elements (PPREs/ DR1), thus regulating transcription of their target genes involved in a variety of physiological functions [5]. PPAR γ is the most extensively studied among the PPAR subtypes [6], and is believed to be a key regulator of adipocyte differentiation and carbohydrate metabolism [7]. PPAR γ plays dual roles in the regulation of insulin sensitivity, one in adipocyte differentiation that contributes to insulin sensitization and the other in adipocyte hypertrophy that leads to insulin resistance, which indicates that both PPARy agonists and antagonists (also termed "selective PPARy modulators, SPPARyM") are effective in ameliorating insulin sensitivity [8]. Clinical benefits have been achieved through the use of PPAR γ agonists – thiazolidinediones (TZDs), such as pioglitazone, rosiglitazone and troglitazone [9]. However, TZDs enhance adipocyte differentiation to improve insulin sensitization, inducing weight gain and edema (fluid retention) [10–12]. PPAR γ antagonists are able to reduce triglyceride (TG) content in white adipose tissue, skeletal muscle, and liver [13]. Furthermore, they can also enhance leptin's effects, and increase fatty acid combustion and energy dissipation, thereby ameliorating HF diet-induced obesity and insulin resistance [14–16]. For example, halofenate

^{*} Corresponding author. Tel.: +86 21 50807042; fax: +86 21 50807088. *E-mail address:* hliu@mail.shcnc.ac.cn (H. Liu).

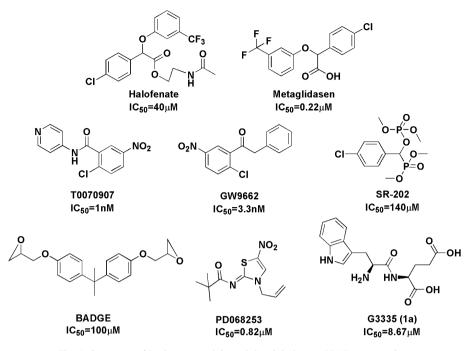


Fig. 1. Structures of lead compound 1a and the eight known PPARy antagonists.

 $(IC_{50} = 40 \ \mu M \text{ against } 10 \ \mu M \text{ rosiglitazone, Fig. 1})$ acts as a PPAR γ antagonist in vitro, showing different interactions with cofactors and leading to differential effects on gene expression in a distinct mechanism of action from that of an agonist (rosiglitazone). In the diabetic ob/ob model, halofenate acutely lowered glucose in a dose-dependent manner [17]. Another PPAR γ antagonist metaglidasen [18,19] (also termed MBX-102, Metabolex Inc.) is now in phase III clinical trials, whose profile of lowering glucose, lipids, and uric acid while avoiding weight gain and edema, clearly differentiates metaglidasen from TZDs. PPAR γ antagonist T0070907 (IC₅₀ = 1 nM against 5 nM rosiglitazone) [20] (Fig. 1) blocks fat accumulation in isolated human adipocytes, and is now being developed for the potential treatment of obesity in preclinical trials. Some other PPAR γ antagonists have been discovered, such as GW9662 ($IC_{50} = 3.3$ nM against 100 nM rosiglitazone) [21], SR-202 (IC₅₀ = 220 μ M against 5 μ M troglitazone) [22], BADGE (IC₅₀ = 100 μ M against 100 nM rosiglitazone) [23], and PD068235 (IC₅₀ = 0.82 μ M against 8 μ M rosiglitazone) [24] (Fig. 1). In fact, the discovery of PPAR γ antagonists (also termed "selective PPAR γ modulators, SPPAR γ M") is now of special importance in the treatment of diabetes and obesity [25-27].

In our published paper [28], we successfully developed a computational virtual screening approach to search for new PPAR γ antagonists from large database, MDL Available Chemicals Directory ACD (www.mdl.com) and SPECS (http://www.specs.net). Twenty-two compounds were found through the virtual screening approach to show binding affinities to PPAR γ at submicromolar or micromolar level according to Biacore 3000 results based on the surface plasmon resonance (SPR) technique. According to binding potency, structural similarity, and drug-like analysis, compound **1a** (G3335, Fig. 1) was selected from these twenty-two compounds by virtual screening as the starting point for further structural optimization [28]. Herein, we report the synthesis, biological evaluation, and molecular modeling of a series of novel dipeptides and their analogues. Thirty-five analogues (1b-l, 9a-d, 13a-t) were synthesized and evaluated against the agonistic effects exerted by rosiglitazone. Six compounds (1b-c, 9a-d) showed strong PPARy antagonistic activities (IC₅₀ values of 5.2–25.8 µM) in antagonizing 10 µM rosiglitazone in the promotion of the PPARy-LBD-CBP interaction by yeast two-hybrid assay. Moreover, all the synthetic compounds were evaluated by FlexX docking and investigated by molecular modeling docking with PPAR γ -LBD, providing possible binding modes at atomic level. Based on these findings, the structure-activity relationships (SARs) of 1a and its analogues were explored for further development of these novel dipeptide class PPARy antagonists.

2. Chemistry

2.1. Design of analogues of compound 1a

A series of 35 novel analogues (1b-l, 9a-d, 13a-t) were designed and synthesized on the basis of the structural feature of the screening hit **1a**, which exhibited potent PPAR γ antagonist activities recently published by Ye et al. [28]. Ye presented that residues Cys285, Arg288, Ser289, and His449 in the active site of PPAR γ play important roles in PPAR γ -LBD-**1a** binding models. Shortening/lengthening the side chain of the C-terminus of dipeptide **1a**, and/or replacing the indole group of the N-terminus with phenyl or naphthyl ring, we obtained compounds **1b-f** (Table 1) to detect and optimize the ligand-binding modes with residues Arg288,

Table 1 Chemical structures together with the FlexX scores of compounds (1a–l, 9a–9d, 13a–13t) and their activities

Compound	Structure	FlexX score	% Inhibition
		(kcal/mol)	at 10 mM ^a
1a	H-L-Trp-L-Glu-OH	-23.3	43.2
1b	H-L-Trp-L-Asp-OH	-21.0	57.8
1c	H-L-Trp-L-Aad-OHb	-19.9	70.0
1d	H-L-Phe-L-Asp-OH	-20.9	1.12
1e	H-L-Phe-L-Glu-OH	-22.7	4.3
1f	H-L-2-Nal-L-Glu-OH ^c	-23.7	2.1
1g	H-D-Trp-L-Asp-OH	-22.6	0.68
1h	H-D-Trp-D-Asp-OH	-22.7	0.42
1i	H-L-Trp-D-Asp-OH	-21.6	15.5
1j	H-L-Trp-D-Glu-OH	-24.8	0.40
1k	H-D-Trp-L-Glu-OH	-23.5	6.60
11	H-D-Trp-D-Glu-OH	-23.5	26.7
9a	H-L-Trp-L-Gln-OH	-26.7	38.6
9b	H-L-Trp-L-Asn-OH	-26.2	36.3
9c	H-L-Trp-L-Arg-OH	-18.7	45.0
9d	H-L-Trp-L-Leu-OH	-21.6	58.2
13a	H-L-Trp-L-Asp(OBzl)-OBzl	-18.4	5.33
13b	H-L-Trp-L-Glu(OMe)-OMe	-16.7	4.05
13c	H-L-Trp-L-Asp(OMe)-OMe	-14.9	NI ^d
13d	H-L-Trp-L-Aad(OMe)-OMe	-15.1	8.74
13e	H-L-Trp-L-Leu-OBzl	-17.7	NI ^d
13f	H-L-Trp-L-Leu-OMe	-20.0	5.29
13g	H-L-2-Nal-L-Glu(OBzl)-OBzl	-12.3	27.7
13h	H-L-Trp-D-Glu(OMe)-OMe	-17.6	NI ^d
13i	H-L-Phe-L-Asp(OBzl)-OBzl	-16.3	NI ^d
13j	H-L-Phe-L-Glu(OMe)-OMe	-10.8	3.99
13k	H-L-2-Nal-L-Glu(OMe)-OMe	-13.2	14.7
13l	H-L-Trp-D-Glu(OBzl)-OBzl	-15.3	NI ^d
13m	H-L-Phe-L-Glu(OBzl)-OBzl	-15.3	NI ^d
13n	H-L-2-Nal-L-Leu-OMe	-14.2	4.34
130	H-D-Phe-L-Aad(OMe)-OMe	-12.9	NI ^d
13p	H-D-Phe-L-Leu-OMe	-14.4	NI ^d
13q	H-D-Phe-L-Asp(OMe)-OMe	-15.0	NI ^d
13r	H-D-Phe-L-Glu(OMe)-OMe	-11.2	NI ^d
13s	H-L-Tyr(O-methyl)-L-	-13.2	5.26
	Glu(OMe)-OMe		
13t	H-L-Phe-L-Ala-OMe	-14.0	NI ^d

^a Data are means of three independent experiments.

^b Aad, L-2-aminoadipic acid.

^c Nal, L-2-naphthylalanine.

^d NI, no inhibition.

Ser289, and His449 in PPAR γ . According to the results of molecular modeling, the two nitrogen atoms on the asymmetric centers of **1a** forming hydrogen bonds with Cys285 [28], the isomers **1g**–**1** (Table 1) of **1a** and **1b** were designed to explore the configuration required by the residue Cys285 in PPAR γ . On replacing the carboxylic acid in the side chain of the C-terminus with amide, guanidyl or isobutyl group, compounds **9a**–**d** (Table 1) were obtained to explore the hydrogen bond formation properties. To improve the lipophilicity and stability, 20 corresponding esterified derivatives (**13a**–**t**) were designed and synthesized.

2.2. Synthetic methods

Scheme 1 depicts the sequence of reactions that led to the preparation of compounds **1a–1** via traditional strategy, using

corresponding L- or D-amino acids as the starting materials. In general, treatment of commercially available L- or D-amino acids (1 equiv) with benzyl alcohol (6.5 equiv) catalyzed by toluene-4-sulfonic acid monohydrate (1.25 equiv) afforded dibenzylated compound **3**. Then compound **3** reacted with **4** to form compounds **5a**–**1**, and *N*-hydroxylbenzotriazole (HOBt) was employed as an anti-racemization reagent. The deprotection of **5a**–**1** was easily conducted in one-pot to afford dipeptides **1a**–**1** through catalytic hydrogenation on Pd/C with bubbling H₂ in quantitative yield.

Compounds 9a-d were synthesized through the route outlined in Scheme 2. Compound 6 was prepared according to Wilchek's method [29] using DCC to couple *N*-carbobenzoxyl-L-tryptophan and *p*-nitrophenol (HONP). Compound 6 of 1 equiv in THF treated with 1.3 equiv of corresponding amino acids afforded compounds 8a-d, which could be purified by silica gel chromatography, and then followed by hydrogenation to yield 9a-d quantitatively.

Scheme 3 depicts the sequence of reactions that led to the preparation of compounds 13a-t using Boc protected amino acids (10a-e) and amino acid esters (11a-j) as starting materials. Compounds 12a-t were synthesized by coupling 10a-e with 11a-j using EDCI/HOBt in CH₂Cl₂/DMF at room temperature, and then were deprotected by TFA/triisopropylsilane (TIPS) to afford compounds 13a-t.

2.3. Molecular docking

To investigate PPAR γ antagonists—PPAR γ —LBD interaction, homology modeling and molecular docking were employed. The static conformation of PPAR γ —LBD binding to **1a** was constructed by homology modeling with the homology module contained in the Insight II software package (based on templates 1KKQ [30] and 2PRG [31]) [28,32]. All compounds (**1a**–l, **9a**–d, **13a**–t) were calculated using FlexX docking to evaluate their binding potency.

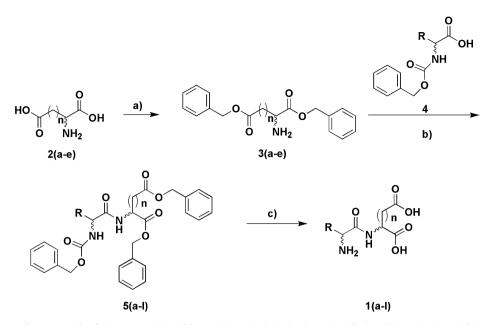
3. Results and discussions

3.1. Analogues' design and synthesis

Thirty-five compounds (1b–l, 9a–d, 13a–t) were designed and synthesized, and their chemical structures are shown in Table 1. These compounds were synthesized through the routes outlined in Schemes 1–3 via classical strategies, and the details of synthetic procedures and structural characterization are described in Section 5.

3.2. Biological activities

We have exploited an efficient approach in the discovery of PPAR γ agonists and antagonists by the yeast two-hybrid system, based on the fact that PPAR γ interacts with the coactivator CBP in ligand-dependent fashion [34]. We successfully employed the MEL1 reporter gene to evaluate the protein—protein interactions by conducting a convenient α -galactosidase assay in the AH109 yeast strain with genes for PPAR γ –LBD and

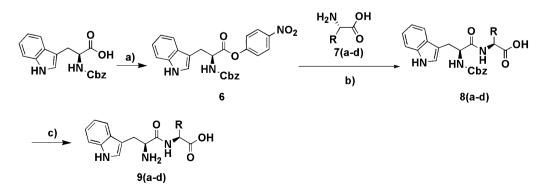


Scheme 1. Synthetic routes of compound 1a-l. Reagents and conditions: (a) benzyl alcohol, toluene-4-sulfonic acid monohydrate (1.25 equiv), benzene, reflux; (b) EDCI (1.05 equiv), HOBt (1 equiv), TEA (10 equiv), CH₂Cl₂/DMF (10/1, v/v), rt, overnight; (c) H₂, Pd/C, MeOH/H₂O/TIPS (80/20/1, v/v), rt, 8 h.

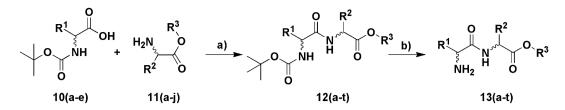
CBP introduced at the N-terminus. The activated MEL1 reporter gene (initiated by ligand and PPAR γ activation) expresses a-galactosidase which would hydrolyze PNP-a-Gal into p-nitrophenol and D-galactose for detection. For the primary assay, the percent antagonistic activities of compounds 1a-l, 9a-d, 13a-t at the concentration of 10 µM blocking rosiglitazone (10 μ M) in the stimulation of the PPAR γ -LBD-CBP interaction were measured. The results are summarized in Table 1. The details for bioassay procedures are described in Section 5. Six new compounds (1b-c, 9a-d) can remarkably block rosiglitazone in the stimulation of the PPAR γ -LBD-CBP interaction (percent inhibition at $10 \,\mu\text{M} > 35\%$), indicating that these six compounds are potent PPARy antagonists. To determine the exact potency of the compounds that exhibited significant antagonistic activities, six compounds (1b-c, 9a-d) were further investigated in concentrationresponse studies, and the results are summarized in Fig. 2. Compounds 1b-c and 9a-d showed antagonistic activities with IC₅₀ values of 5.2, 15.1, 18.7, 18.4, 15.4, and 25.8 µM against rosiglitazone (10 µM), respectively.

3.3. Structure-activity relationship (SAR)

The 3D structure of PPAR γ -LBD itself indicates that the antagonistic binding site of PPARy-LBD consists of three subsites (A-C, Fig. 3a-h), which form a T-shaped groove: site A is a small light hydrophobic binding pocket formed by Ser289, Val290, His323, Ile326, Leu330, Val339, His449, and Tyr473. Right to site A, there is a shallow larger hydrophilic pocket, which is called site B, formed by Phe282, Gln286, Arg288, and Met364. On the opposite side to site B, a deep hydrophobic cavity called site C connects sites A and B, which is composed of Cys285, Tyr327, Ile445, Val446, His449, Leu453, and Tyr473. The binding models indicate that active compounds (1a-c, 9c-d, Figs. 3a-c, f, g and 4a-c, f, g) interact with PPAR γ -LBD in a similar way obviously differing from that of inactive compounds (1d, 1h) and 13a, Figs. 3d, e, h and 4d, e, h). The indole group, side chain of C-terminus, and 1-free amino acid group (in the Cterminus) of this dipeptide class antagonists occupy subpockets B, C and A, respectively, contributing a lot of features for



Scheme 2. Synthetic routes of compounds **9a-d**. Reagents and conditions: (a) DCC (1 equiv), HONP (1.02 equiv), THF; (b) TEA (2 equiv), H₂O/THF (1/1, v/v), rt, overnight; (c) H₂, Pd/C, MeOH/H₂O/TIPS (80/20/1, v/v), rt, 8 h.



Scheme 3. Synthetic routes of compounds 13a-t. Reagents and conditions: (a) EDCI (1 equiv), HOBt (1 equiv), TEA (10 equiv), CH₂Cl₂/DMF (10/1, v/v), rt, overnight; (b) TFA/CH₂Cl₂/TIPS (1/40/0.4, v/v), 0 °C to rt, 12 h.

the binding interaction on the basis of molecular docking characteristics and biological activities.

As shown in Table 1, the initial compound **1a** exhibited 43.2% inhibition rate at the concentration of 10 μ M against rosiglitazone (10 μ M) in the yeast two-hybrid assay. Shortening/lengthening the side chain of C-terminus of **1a** (compounds **1b** and **1c**) both improved the antagonistic activities to 57.8 and 70.0%, respectively (Table 1). In Figs. 3 and 4, compound **1b** (Figs. 3b and 4b) shows more hydrogen bonds than those of **1a** (Figs. 3a and 4a), and compound **1c** (Figs. 3c and 4c) deserves rich hydrophobic contacts with Phe282, Arg288, Leu453, and Leu465. Thus, the antagonistic activities of compounds have the order of **1c** > **1b** > **1a** (Table 1).

The indole group of compound 1a forms favorite hydrophobic interaction in subpocket B. Replacing the indole group with phenyl or naphthyl group resulted in a great loss of antagonistic activities (1d-f, Table 1), which indicates that the indole group of compound 1a is preferred at the N-terminus for the antagonistic activity. Actually, the docked conformations of 1d indicate that the hydrophobic subpocket B can be perfectly occupied by the indole group, but not by the phenyl group (Figs. 3d and 4d).

The diastereomeric configuration of the dipeptide scaffold is essential for its activity. Compounds 1g-1, the isomers of 1a or 1b, presented less to little antagonistic effects than the (S,S) isomer of 1a or 1b (shown in Table 1). These data

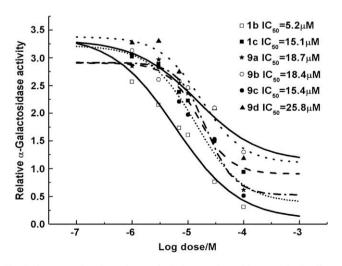


Fig. 2. Concentration dependence of relative α -galactosidase activity by **1b–c** and **9a–d** as determined by yeast two-hybrid assay. The concentration of rosiglitazone was kept constant at 10 μ M, while the concentration of compounds ranged from 1 to 100 μ M.

demonstrate that the stereo requirement for the interaction is strict. The binding mode of (R,R)-1h, an isomer of (S,S)-1b, suggests that the binding features are highly similar to 1b, except that the C-terminal side chain of dipeptide 1h stretches into the subpocket A rather than the subpocket C (Figs. 3e and 4e). Such stereo difference results in the loss of inhibitory activity. Moreover, the C-terminal side chain of these dipeptides could play another important role in hydrophobic patch to subpocket C (9c-d, Figs. 3c, d and 4c, d). In the side chain of C-terminus, polar groups (such as carboxylic acid, amide, guanidyl groups, 1a-b, 9a-c) or non-polar groups (such as isopropyl group, 9d) retained antagonistic activities. The isopropyl group with higher inhibition rate (58.2%) (Table 1) suggests that smaller hydrophobic group easily enters the deep hydrophobic subpocket C of PPARy-LBD to exert stronger antagonistic activity (9d, Fig. 3d).

Compounds 13a-t own significantly higher binding energies as shown in Table 1. This is mainly because the blocking of the benzyl or methylester group (13a-t) will impair the capability of carboxyl group to form hydrogen bond. Compounds 13a-t, in which 1-carboxyl acid protected by benzyl or methylesters, showed decreased antagonistic activities in the assay, and many of them were proved to completely lose the antagonistic activity (shown in Table 1). 1-Free carboxylic acid of the C-terminus of dipeptide 1a forms hydrogen bonds with residues Ser289, His323, and His449 in subpocket A. In the binding modeling using compound 13a (Fig. 3h) as an example, the ester groups on the C-terminal of these dipeptides are too large to be inserted into the subpockets A and C. Moreover, the ester group on the C-terminal cannot form hydrogen bonds with Ser289 and His323 any more. As a result, the binding conformation of 13a is far away from the subpocket A and subpocket C, and therefore losing antagonistic activity. These demonstrate that 1-free carboxylic acid of the C-terminus is preferred in the further antagonist optimization.

4. Conclusions

In summary, a series of novel PPAR γ antagonists (also termed "selective PPAR γ modulators, SPPAR γ M") were designed and synthesized based on the initial hit **1a**. All compounds were evaluated by yeast two-hybrid assay for their PPAR γ antagonistic activities. Selected compounds for concentration—response studies showed prominent antagonistic activities with IC₅₀ values ranging from 5.2 to 25.8 μ M. Basing on successful homology modeling of PPAR γ , binding

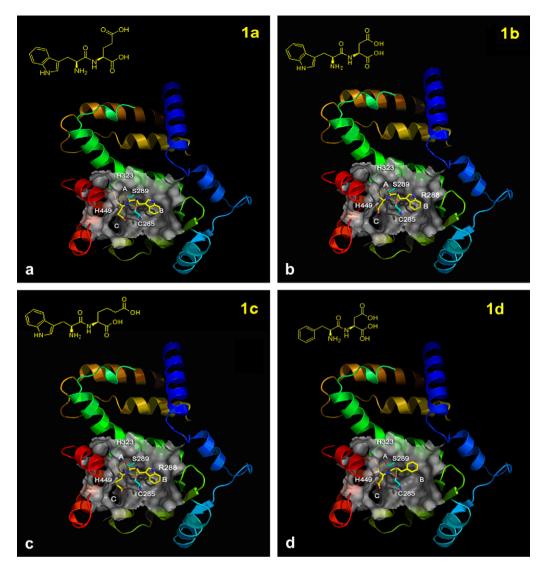


Fig. 3. Three-dimensional binding modes of 1a (a), 1b (b), 1c (c), 1d (d), 1h (e), 9c (f), 9d (g) and 13a (h) to PPAR γ -LBD derived from the docking simulations. Images were generated using Pymol program (http://www.pymol.org).

affinity was evaluated by FlexX, and SARs were explored. Further studies are in progress.

5. Experimental section

5.1. Yeast two-hybrid based assay

The yeast two-hybrid system used for PPAR γ antagonist screening was constructed by the method published by Chen et al. [34]. The MEL1 reporter gene (the GAL gene family member) was used in the yeast two-hybrid system for detection of protein—protein interactions through conducting α -galactosidase assays [28,35]. By the PCR technique, mCBP (aa 1–464) and hPPAR γ –LBD (aa 204–477) were inserted into the vectors pGADT7 and pGBKT7 (Clontech), respectively, to provide pGADT7–CBP and pGBKT7–PPAR γ plasmids, which were then co-transfected into yeast strain AH109 by the lithium acetate method [35]. The corresponding yeast cells were cultured in SD minimal medium (3 mL) without leucine

and tryptophan $(T^{-}L^{-})$ (yeast nitrogen base without amino acids 6.7 g L^{-1} , D-(+)-glucose 20 g L^{-1} , yeast synthetic drop-out medium supplement without leucine and tryptophan 1.54 g L⁻¹) at 30 °C with shaking (250 rpm). When the OD_{600} of the cells had reached about 1.0, the cultures were diluted 20 times with fresh T⁻L⁻ medium, and different concentrations of the 35 compounds synthesized in this work (10 μ L) with rosiglitazone (10 mM, 1 µL) or DMSO (11 µL, as a blank) were then added to diluted cultures (989 µL). After further incubation at 30 °C for 16 h, each sample (200 µL) was transferred to a 96-well plate (Corning Costar 96-well flat-bottomed plate), and the OD₆₀₀ value was measured with a Benchmark PlusTM microplate spectrophotometer (Bio-Rad). The cells were then centrifuged and the supernatant (16 µL) was transferred to another 96-well plate with assay buffer (48 μ L; 16 μ L of 100 mM PNP- α -Gal + 32 μ L of 0.5 M sodium acetate, pH 4.5 per well). After a further 60 min incubation at 30 °C, the reaction was terminated by the addition of stop solution (136 µL, 1 M Na₂CO₃) in each

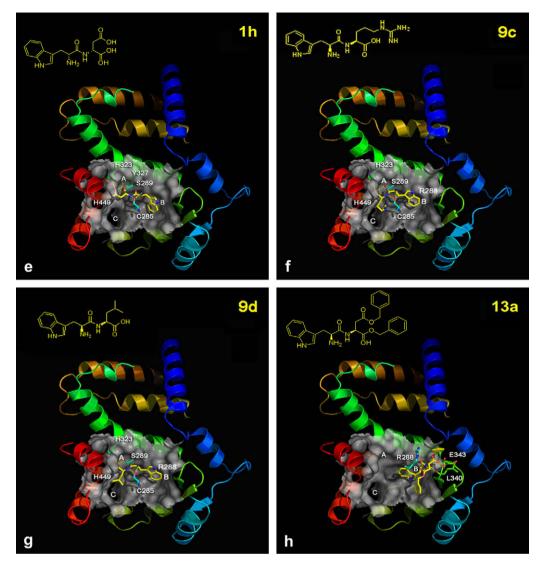


Fig. 3. (Continued).

well, and the OD₄₁₀ was measured. One unit of α -galactosidase is defined as the amount of enzyme that hydrolyzes 1 mmol PNP- α -Gal to *p*-nitrophenol and D-galactose in 1 min at 30 °C in acetate buffer (pH 4.5) [36]. The α -galactosidase activity in milliunits per well was calculated according to the following formula:

 α -Galactosidase activity (milliunits)

$$= \mathrm{OD}_{410} \times V_{\mathrm{f}} \times 1000 / [(\varepsilon \times b) \times t \times V_{\mathrm{i}} \times \mathrm{OD}_{600}]$$

In the formula, *t* is the elapsed time of incubation (min), V_f is the final volume of assay (200 µL), V_i is the volume of culture medium supernatant added (16 µL), OD_{600} is the optical density of the cells at the start of the assay, and $\varepsilon \times b$ is *p*-nitrophenol molar absorptivity at 410 nm × the light path (cm) = 10.5 (mL µmol⁻¹; Yeast Protocols Handbook PT3024-1, Clontech).

The inhibition assays of compounds (**1b–c**, **9a–d**) were performed in the same manner as mentioned above. The overnight cultures of yeast cells containing pGADT7–CBP and pGBKT7–PPAR γ were diluted with fresh media to an initial OD₆₀₀ of 0.05 and treated with rosiglitazone (10 µM) and increasing concentrations of the six compounds or DMSO (as vehicle control) for 16 h at 30 °C. Relative α -galactosidase activity was measured on the cells treated with 1% DMSO. To calculate the half-maximal antagonistic concentration (IC₅₀), the relative α -galactosidase activity is plotted against the common logarithm of the compound concentration, and the data were fitted using the sigmoidal fitting model by the Origin7.0 software.

5.2. Molecular modeling

The 3D structures of compounds were constructed with the standard geometric parameters of the SYBYL 6.8 molecular modeling software package (http://www.tripos.com). The geometries of these antagonists were subsequently optimized by the Power method encoded in SYBYL 6.8. The Tripos force field and Gasteiger–Hückel charges were employed during the antagonist minimization [37]. As described in our previous report [28], 3D model of PPAR γ –LBD was generated by homology modeling, based on 1KKQ [30] and 2PRG [31]. The structure-conserved regions of the two template

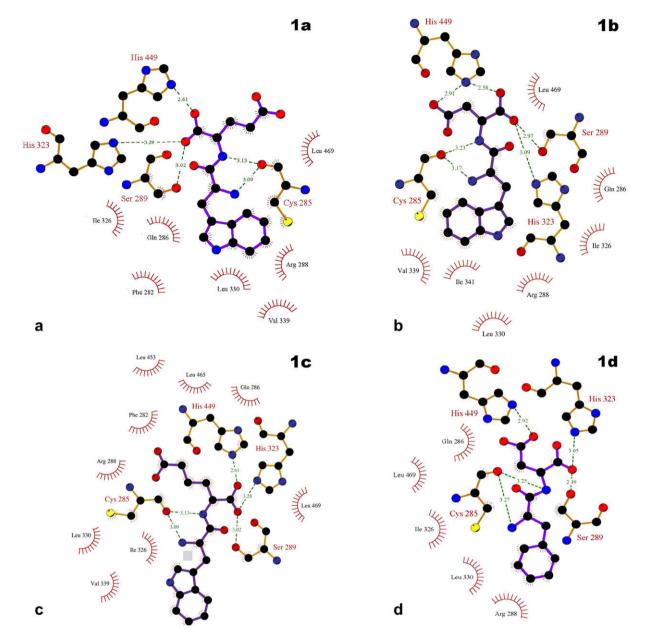


Fig. 4. Two-dimensional representation for the interacting mode of compounds 1a (a), 1b (b), 1c (c), 1d (d), 1h (e), 9c (f), 9d (g) and 13a (h) with PPAR γ -LBD as plotted with LIGPLOT program [33]. Dashed lines represent H-bonds and spiked residues represent hydrophobic contacts with the ligands.

proteins were determined by structure alignment of the homology module in Insight II [32]. The final model protein was validated both by Procheck and by Profiles-3D [38,39]. All compounds (1a–1, 9a–d, 13a–t) were docked into the constructed PPAR γ –LBD to obtain the corresponding binding energies by FlexX (incorporated in SYBYL 6.8).

All the molecular modeling and docking simulations were performed on a Silicon Graphics Origin3200 workstation (with four CPUs).

5.3. Chemistry

The reagents were purchased from Alfa Aesar, Acros, Sigma–Aldrich, Shanghai Chemical Reagent Company, and GL Biochem and were used without further purification. Analytical thin-layer chromatography (TLC) was HSGF 254 silica gel (0.15–0.2 mm thickness, Yantai Huiyou Company, China). CH₂Cl₂ was distilled from CaH₂ and DMF was purchased in HPLC grade. Solvents were evaporated under reduced pressure and below 50 °C if no description was noted. Melting points were measured in capillary tube with an SGW X-4 melting point apparatus without correction. Rotary nuclear magnetic resonance (NMR) spectra were obtained on a Brucker AMX-300 NMR (TMS as IS). Chemical shifts were reported in parts per million (ppm, δ) downfield from tetramethylsilane (TMS). Proton coupling patterns were described as singlet (s), doublet (d), triplet (t), quartet (q), multiplet (m), and broad (br). Low- and high-resolution mass spectra (LRMS and HRMS) were given with electric and electrospray ionization (EI, ESI) produced, respectively, by Finnigan

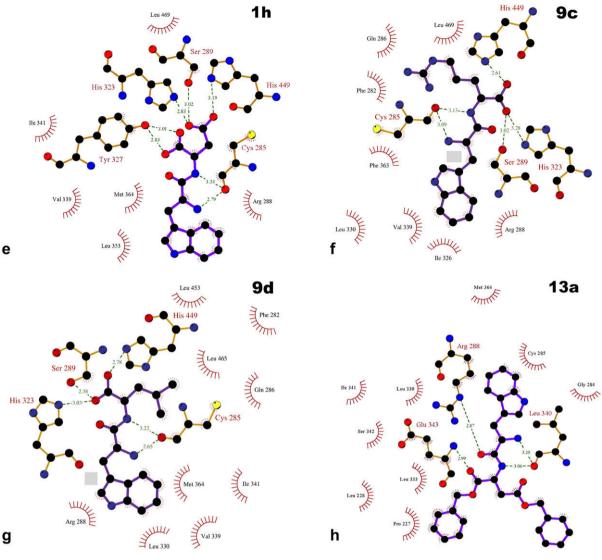


Fig. 4. (Continued).

MAT-95, LCQ-DECA spectrometer, and IonSpec 4.7 T. Degrees of optical activity were measured with PE-341 polarimeter.

5.4. General procedures for preparation of 3a-e using 3a as an example

5.4.1. L-Dibenzylester glutamate p-toluenesulfonate (3a)

A mixture of L-glutamic acid (4 g, 27 mmol), *p*-toluenesulfonic acid (6.445 g, 34 mmol), benzyl alcohol (18 mL) and benzene (27 mL) was heated to reflux for 4 h with vigorous stirring, then benzene was removed under reduced pressure and fresh benzene was added to reflux for another 1 h. This procedure was repeated three times. Then ethyl ether (50 mL) was added to form a white precipitate and filtered. The filter cake was washed with ethyl ether and dried under high vacuum to give 12.5 g (92%) of **3a** as a white solid. ¹H NMR (CD₃OD): $\delta = 7.69$ (d, 2H), 7.36 (m, 10H), 7.22 (d, 2H), 5.24 (s, 2H), 5.10 (s, 2H), 4.15 (d, 1H), 2.52 (m, 2H), 2.35 (s, 3H), 2.17 (m, 2H). 5.4.2. L-Dibenzylester aspartate p-toluenesulfonate (3b)

Compound **3b** was prepared from L-aspartic acid using a procedure similar to that described for the preparation of **3a** as a white solid, yield 90%. ¹H NMR (DMSO- d_6): $\delta = 7.49$ (d, 2H), 7.37 (m, 10H), 7.12 (d, 2H), 5.19 (s, 2H), 5.11 (s, 2H), 4.49 (t, 1H), 3.02 (m, 2H), 2.29 (s, 3H).

5.4.3. L-2-Aminoadipic acid dibenzylester p-toluenesulfonate (**3c**)

Compound **3c** was prepared from L-2-aminoadipic acid using a procedure similar to that described for the preparation of **3a** as a white solid, yield 87%. ¹H NMR (DMSO- d_6): $\delta = 7.55$ (d, 2H), 7.38 (m, 10H), 7.15 (d, 2H), 5.18 (s, 2H), 4.97 (s, 2H), 4.49 (t, 1H), 2.42 (m, 2H), 2.29 (s, 3H), 1.92 (m, 2H), 1.71 (m, 2H).

5.4.4. D-Dibenzylester glutamate p-toluenesulfonate (3d)

Compound **3d** was prepared from D-glutamic acid using a procedure similar to that described for the preparation of **3a** as a white solid, yield 83%. ¹H NMR (CD₃OD): $\delta = 7.70$

(d, 2H), 7.36 (m, 10H), 7.22 (d, 2H), 5.24 (s, 2H), 5.09 (s, 2H), 4.14 (d, 1H), 2.52 (m, 2H), 2.35 (s, 3H), 2.17 (m, 2H).

5.4.5. *D*-Dibenzylester aspartate *p*-toluenesulfonate (3e)

Compound **3e** was prepared from D-aspartic acid using a procedure similar to that described for the preparation of **3a** as a white solid, yield 94%. ¹H NMR (DMSO- d_6): $\delta = 7.43$ (d, 2H), 7.37 (m, 10H), 7.15 (d, 2H), 5.19 (s, 2H), 5.11 (s, 2H), 4.46 (t, 1H), 3.01 (m, 2H), 2.28 (s, 3H).

5.5. General procedures for preparation of 5a-l

5.5.1. N-Carbobenzoxyl-L-tryptophan-

L-glutamate dibenzylester (5a)

To a mixture of L-dibenzylester glutamate p-toluenesulfonate (3a) (0.5 g, 1 mmol), N-carbobenzoxyl-L-tryptophan (0.34 g, 1 mmol), 1-hydroxybenzotriazole (HOBt, 0.14 g, 1 mmol) and triethylamine (1.4 mL, 10 mmol) in 5% DMF/ CH₂Cl₂ (20 mL), 1-ethyl-(3-dimethylaminopropyl)-carbodiimide hydrochloride (EDCI, 0.2 g, 1.05 mmol) was added in portions for 5 min. The solution was stirred overnight at room temperature (rt) followed by TLC detection. The solution was washed, dried, filtered, and condensed to form a yellow oil. The oil can be purified by flash chromatography with petroleum ether/ethyl acetate (4/1, v/v) to afford a white powder (0.59 g, 91%). Mp 98–99 °C; ¹H NMR (CDCl₃): $\delta = 7.82$ (s, 1H), 7.62 (d, 1H), 7.21-7.40 (m, 16H), 6.95 (s, 1H), 6.38 (d, 1H), 5.43 (d, 1H), 5.15 (m, 2H), 5.06 (s, 2H), 5.03 (s, 2H), 4.52 (m, 2H), 3.39 (dd, 1H), 3.10 (q, 1H), 2.14 (br, 3H), 1.83 (m, 1H).

5.5.2. N-Carbobenzoxyl-L-tryptophan-

L-aspartate dibenzylester (5b)

Compound **5b** was prepared from *N*-carbobenzoxyl-L-tryptophan and L-dibenzylester aspartate *p*-toluenesulfonate (**3b**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 87%. Mp 151–153 °C; ¹H NMR (CDCl₃): $\delta = 7.84$ (s, 1H), 7.62 (d, 1H), 6.95–7.41 (br, 16H), 6.85 (d, 1H), 5.12 (m, 1H), 5.08 (s, 2H), 4.98 (d, 2H), 4.80 (m, 1H), 3.25 (m, 1H), 3.20 (m, 1H), 3.01 (dd, 1H), 2.8 (dd, 1H).

5.5.3. N-Carbobenzoxyl-L-tryptophan-

L-2-aminoadipate dibenzylester (5c)

Compound **5c** was prepared from *N*-carbobenzoxyl-L-tryptophan and L-2-aminoadipic acid dibenzylester *p*-toluenesulfonate (**3c**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 85%. Mp 99– 100 °C; ¹H NMR (CDCl₃): $\delta = 8.39$ (s, 1H), 7.65 (d, 1H), 7.10–7.36 (m, 15H), 7.13 (t, 1H), 7.08 (t, 1H), 6.31 (d, 1H), 5.47 (d, 1H), 5.13 (s, 2H), 4.99 (s, 2H), 4.97 (s, 2H), 4.55 (m, 1H), 4.48 (q, 1H), 3.34 (dd, 1H), 3.14 (q, 1H), 2.20 (t, 1H), 1.64 (m, 1H), 1.51 (m, 1H).

5.5.4. N-Carbobenzoxyl-L-phenylalanine-L-aspartate dibenzylester (5d)

Compound **5d** was prepared from *N*-carbobenzoxyl-L-phenylalanine and L-dibenzylester aspartate *p*-toluenesulfonate (3b) using a procedure similar to that described for the preparation of 5a as a white solid, yield 86%. ¹H NMR (CDCl₃): $\delta = 7.10-7.41$ (m, 20H), 5.12 (s, 2H), 4.99 (s, 2H), 4.97 (s, 2H), 4.22 (t, 1H), 3.98 (m, 1H), 3.23 (m, 1H), 2.91 (q, 1H), 2.28 (m, 2H).

5.5.5. N-Carbobenzoxyl-L-phenylalanine-

L-glutamate dibenzylester (5*e*)

Compound **5e** was prepared from *N*-carbobenzoxyl-L-phenylalanine and L-dibenzylester glutamate *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 88%. Mp 100–101 °C; ¹H NMR (CDCl₃): $\delta = 7.08-7.39$ (m, 20H), 6.56 (d, 1H), 5.11 (s, 2H), 5.18 (d, 2H), 5.10 (s, 2H), 4.81 (m, 1H), 4.36 (q, 1H), 3.05 (dd, 2H), 2.22 (m, 3H), 1.98 (m, 1H).

5.5.6. N-Carbobenzoxyl-L-2-naphthylalanine-L-glutamate dibenzylester (**5f**)

Compound **5f** was prepared from *N*-carbobenzoxyl-L-2naphthylalanine and L-dibenzylester glutamate *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 80%. Mp 110– 111 °C; ¹H NMR (CDCl₃): $\delta = 7.73$ (m, 3H), 7.65 (s, 1H),

7.22-7.43 (m, 18H), 5.12 (d, 2H), 5.02 (s, 2H), 4.89 (s, 2H), 4.42 (m, 2H), 3.18 (q, 1H), 2.98 (q, 1H), 2.25 (s, 3H), 1.8 (m, 1H).

5.5.7. N-Carbobenzoxyl-D-tryptophan-

L-aspartate dibenzylester (5g)

Compound **5g** was prepared from *N*-carbobenzoxyl-D-tryptophan and L-dibenzylester aspartate *p*-toluenesulfonate (**3b**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 85%. Mp 116–118 °C; ¹H NMR (CDCl₃): $\delta = 7.63$ (s, 1H), 7.57 (d, 1H), 7.06–7.36 (br, 17H), 6.90 (s, 1H), 6.60 (d, 1H), 5.44 (m, 1H), 5.08 (s, 2H), 5.02 (s, 2H), 4.85 (q, 2H), 4.79 (m, 1H), 4.49 (br, 1H), 3.08–3.27 (m, 2H), 2.83 (dd, 1H), 2.30 (dd, 1H), 1.61 (br, 1H).

5.5.8. N-Carbobenzoxyl-D-tryptophan-

D-aspartate dibenzylester (5h)

Compound **5h** was prepared from *N*-carbobenzoxyl-D-tryptophan and D-dibenzylester aspartate *p*-toluenesulfonate (**3e**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 80%. Mp 155–157 °C; ¹H NMR (CDCl₃): $\delta = 7.79$ (s, 1H), 7.65 (d, 1H), 7.05–7.38 (br, 17H), 6.96 (d, 1H), 6.65 (d, 1H), 5.38 (br, 1H), 5.08 (m, 4H), 4.93 (d, 2H), 4.78 (m, 1H), 4.50 (br, 1H), 3.19 (m, 2H), 2.94 (m, 2H).

5.5.9. N-Carbobenzoxyl-L-tryptophan-

D-aspartate dibenzylester (5i)

Compound **5i** was prepared from *N*-carbobenzoxyl-L-tryptophan and D-dibenzylester aspartate *p*-toluenesulfonate (**3e**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 88%. Mp 115–117 °C; ¹H NMR (CDCl₃): $\delta = 7.64$ (s, 1H), 7.60 (d, 1H), 7.05–7.41 (br, 16H), 6.91 (s, 1H), 6.61 (d, 1H), 5.44 (m, 1H), 5.09 (s, 2H), 5.02 (s, 2H), 4.79 (m, 1H), 4.49 (br, 1H), 3.08–3.16 (m, 2H), 2.83 (dd, 1H), 2.27 (dd, 1H), 1.60 (br, 1H).

5.5.10. N-Carbobenzoxyl-L-tryptophan-D-glutamate dibenzylester (**5j**)

Compound **5j** was prepared from *N*-carbobenzoxyl-Ltryptophan and D-dibenzylester glutamate *p*-toluenesulfonate (**3d**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 88%. Mp 54–55 °C; ¹H NMR (CD₃OD): $\delta = 7.54$ (d, 1H), 6.90–7.41 (br, 19H), 5.08 (s, 2H), 5.06 (s, 2H), 4.97 (s, 2H), 4.46 (br, 2H), 3.17 (m, 1H), 3.04 (m, 1H), 2.32 (m, 2H), 2.12 (br, 1H), 1.78 (m, 1H).

5.5.11. N-Carbobenzoxyl-D-tryptophan-L-glutamate dibenzylester (5k)

Compound **5k** was prepared from *N*-carbobenzoxyl-Dtryptophan and L-dibenzylester glutamate *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 85%. Mp 115–117 °C; ¹H NMR (CDCl₃): $\delta = 7.77$ (s, 1H), 7.65 (d, 1H), 7.05–7.42 (br, 17H), 6.93 (d, 1H), 6.27 (d, 1H), 5.43 (d, 1H), 5.10 (s, 2H), 5.07 (s, 2H), 5.06 (s, 2H), 4.54 (m, 2H), 3.40 (m, 1H), 3.15 (q, 1H), 1.93 (m, 2H).

5.5.12. N-Carbobenzoxyl-D-tryptophan-D-glutamate dibenzylester (5l)

Compound **51** was prepared from *N*-carbobenzoxyl-Dtryptophan and D-dibenzylester glutamate *p*-toluenesulfonate (**3d**) using a procedure similar to that described for the preparation of **5a** as a white solid, yield 92%. Mp 116–117 °C; ¹H NMR (CDCl₃): $\delta = 7.86$ (s, 1H), 7.62 (d, 1H), 7.08–7.39 (br, 15H), 7.15 (t, 1H), 7.10 (t, 1H), 6.95 (s, 1H), 6.40 (d, 1H), 5.43 (d, 1H), 5.10 (d, 2H), 5.06 (s, 2H), 5.03 (s, 2H), 4.52 (m, 2H), 3.38 (m, 1H), 3.11 (q, 1H), 2.13 (m, 3H), 1.80 (m, 1H).

5.6. General procedures for preparation of *la–l* using compound *la* as an example

5.6.1. L-Tryptophan-L-glutamic acid (1a)

A solution of **5a** (0.59 g, 0.91 mmol) and triisopropylsilane (100 µL) in 80% methanol—water (10 mL) was hydrogenated in the presence of 0.1 g of 10% palladium on charcoal (Pd/C) with stirring for 8 h. The solution was filtered and the filtrate was evaporated to form a light oil. The oil was dissolved in water and washed with ethyl acetate. The water portion was dried by lyophilization to form **1a** (0.3 g, 95%). Mp 116–118 °C; $[\alpha]_D^{20}$ +11.2 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 7.66$ (d, 1H), 7.33 (d, 1H), 7.21 (s, 1H), 7.06 (t, 1H), 6.95 (t, 1H), 4.21 (m, 1H), 3.93 (m, 1H), 3.21 (dd, 1H), 2.97 (m, 1H), 2.28 (m, 2H), 1.87 (m, 2H); ESI-MS *m*/*z* 334, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₆H₂₀N₃O₅ $[M + H]^+$ 334.1403, found 334.1416.

5.6.2. L-Tryptophan-L-aspartic acid (1b)

Compound **1b** was prepared from *N*-carbobenzoxyl-Ltryptophan-L-aspartate dibenzylester (**5b**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 98%. Mp 167–168 °C; $[\alpha]_D^{20}$ +20.3 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 7.46$ (d, 1H), 7.34 (d, 1H), 7.12 (t, 1H), 7.03 (d, 1H), 4.50 (t, 1H), 4.38 (q, 2H), 4.10 (dd, 1H), 3.32 (dd, 1H), 3.01 (q, 1H), 2.74 (m, 2H); ESI-MS *m/z* 320, $[M + H]^+$; HRMS (ESI) *m/z* calcd C₁₅H₁₈N₃O₅ $[M + H]^+$ 320.1232, found 320.1246.

5.6.3. L-Tryptophan-L-2-aminoadipic acid (1c)

Compound **1c** was prepared from *N*-carbobenzoxyl-Ltryptophan-L-2-aminoadipate dibenzylester (**5c**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 98%. Mp 99–100 °C; $[\alpha]_D^{20}$ +18.9 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 10.98$ (s, 1H), 8.48 (d, 1H), 7.63 (d, 1H), 7.32 (d, 1H), 7.20 (s, 1H), 7.05 (t, 1H), 6.95 (t, 1H), 4.16 (m, 3H), 3.87 (q, 1H), 3.16 (dd, 1H), 2.96 (q, 1H), 2.23 (m, 2H), 1.83 (m, 2H); ESI-MS *m*/*z* 348, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₇H₂₂N₃O₅ [M + H]⁺ 348.1559, found 348.1549.

5.6.4. L-Phenylalanine-L-aspartic acid (1d)

Compound **1d** was prepared from *N*-carbobenzoxyl-L-phenylalanine-L-aspartate dibenzylester (**5d**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 95%. Mp 57–58 °C; $[\alpha]_D^{20}$ +25.1 (*c* 0.5, water); ¹H NMR (CD₃OD): $\delta = 8.89$ (d, 1H), 7.28 (m, 5H), 4.50 (t, 1H), 4.08 (q, 1H), 3.14 (dd, 1H), 2.94 (q, 1H), 2.66 (q, 2H); ESI-MS *m*/*z* 281, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₃H₁₇N₂O₅ $[M + H]^+$ 281.1137, found 281.1146.

5.6.5. L-Phenylalanine-L-glutamic acid (1e)

Compound **1e** was prepared from *N*-carbobenzoxyl-L-phenylalanine-L-glutamate dibenzylester (**5e**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 97%. Mp 226–227 °C; $[\alpha]_D^{20}$ +12.1 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): δ = 7.30 (m, 3H), 7.17 (q, 2H), 4.25 (q, 1H), 3.68 (t, 1H), 3.06 (m, 2H), 2.31 (t, 2H), 1.98 (m, 1H), 1.80 (m, 1H); ESI-MS *m*/*z* 295, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₄H₁₉N₂O₅ [M + H]⁺ 295.1294, found 295.1304.

5.6.6. L-2-Naphthylalanine-L-glutamic acid (1f)

Compound **1f** was prepared from *N*-carbobenzoxyl-L-2naphthylalanine-L-glutamate dibenzylester (**5f**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 99%. Mp 57–58 °C; $[\alpha]_D^{20}$ +21.1 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 8.72$ (d, 1H), 7.61 (m, 4H), 7.43 (m, 3H), 4.16 (m, 2H), 3.24 (dd, 1H), 3.20 (q, 1H), 2.19 (t, 2H), 1.8 (m, 2H); ESI-MS *m*/*z* 345, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₈H₂₁N₂O₅ [M + H]⁺ 345.1450, found 345.1455.

5.6.7. D-Tryptophan-L-aspartic acid (1g)

Compound **1g** was prepared from *N*-carbobenzoxyl-Dtryptophan-L-aspartate dibenzylester (**5g**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 94%. Mp 161–163 °C, $[\alpha]_D^{20}$ –39.2 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): δ = 10.99 (d, 1H), 8.56 (d, 1H), 7.67 (d, 1H), 7.33 (d, 1H), 7.19 (s, 1H), 7.08 (t, 1H), 6.98 (t, 1H), 4.23 (m, 1H), 4.06 (m, 1H), 2.91–3.21 (m, 2H), 2.40 (m, 1H), 2.24 (m, 1H); ESI-MS *m*/*z* 320, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₅H₁₈N₃O₅ [M + H]⁺ 320.1246, found 320.1240.

5.6.8. D-Tryptophan-D-aspartic acid (1h)

Compound **1h** was prepared from *N*-carbobenzoxyl-Dtryptophan-D-aspartate dibenzylester (**5h**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 94%. Mp 190–191 °C; $[\alpha]_D^{20}$ –37.8 (*c* 0.5, water); ¹H NMR (CD₃OD): δ = 7.69 (d, 1H), 7.36 (d, 1H), 7.25 (s, 1H), 7.04–7.14 (m, 2H), 4.56 (m, 1H), 4.16 (m, 1H), 3.45 (dd, 1H), 3.22 (q, 1H), 2.80 (m, 2H); ESI-MS *m*/*z* 320, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₅H₁₈N₃O₅ [M + H]⁺ 320.1246, found 320.1222.

5.6.9. L-Tryptophan-D-aspartic acid (1i)

Compound **1i** was prepared from *N*-carbobenzoxyl-Ltryptophan-D-aspartate dibenzylester (**5i**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 99%. Mp 163–164 °C; $[\alpha]_D^{20}$ +25.8 (*c* 0.5, water); ¹H NMR (CD₃OD): δ = 7.64 (d, 1H), 7.35 (d, 1H), 7.19 (s, 1H), 7.10 (m, 2H), 4.58 (t, 1H), 4.10 (t, 1H), 3.33 (m, 1H), 3.22 (m, 1H), 2.54 (m, 2H); ESI-MS *m*/*z* 320, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₅H₁₈N₃O₅ [M + H]⁺ 320.1246, found 320.1240.

5.6.10. L-Tryptophan-D-glutamic acid (1j)

Compound **1j** was prepared from *N*-carbobenzoxyl-Ltryptophan-D-glutamate dibenzylester (**5j**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 99%. Mp 49–50 °C; $[\alpha]_D^{20}$ –45.2 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): δ = 7.70 (d, 1H), 7.34 (d, 1H), 7.20 (s, 1H), 7.07 (t, 1H), 7.00 (t, 1H), 4.20 (m, 1H), 3.99 (m, 1H), 3.28 (m, 1H), 3.08 (m, 1H), 2.14 (t, 2H), 1.80 (m, 2H); ESI-MS *m*/*z* 334, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₆H₂₀N₃O₅ [M + H]⁺ 334.1403, found 334.1393.

5.6.11. D-Tryptophan-L-glutamic acid (1k)

Compound **1k** was prepared from *N*-carbobenzoxyl-Dtryptophan-L-glutamate dibenzylester (**5k**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 91%. Mp 169–170 °C; $[\alpha]_D^{20}$ –85.2 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 10.94$ (d, 1H), 8.30 (d, 1H), 7.63 (d, 1H), 7.32 (d, 1H), 7.17 (s, 1H), 7.02 (t, 1H), 6.98 (t, 1H), 4.08 (m, 2H), 3.13 (dd, 1H), 2.93 (q, 1H), 2.11 (m, 2H), 1.71 (m, 2H); ESI-MS *m*/*z* 334, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₆H₂₀N₃O₅ $[M + H]^+$ 334.1403, found 334.1424.

5.6.12. D-Tryptophan-D-glutamic acid (11)

Compound **11** was prepared from *N*-carbobenzoxyl-D-tryptophan-D-glutamate dibenzylester (**51**) using a procedure similar to that described for the preparation of **1a** as a white solid, yield 100%. Mp 170–172 °C; $[\alpha]_D^{20}$ –30.4 (*c* 0.5, water); ¹H NMR (CD₃OD): $\delta = 7.69$ (d, 1H), 7.38 (d, 1H),

7.22 (s, 1H), 7.11 (t, 1H), 4.33 (m, 1H), 4.19 (q, 1H), 3.47 (dd, 1H), 3.20 (q, 1H), 2.34 (t, 2H), 2.18 (m, 1H), 2.00 (m, 1H); ESI-MS *m*/*z* 334, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd $C_{16}H_{20}N_3O_5 [M + H]^+$ 334.1403, found 334.1419.

5.7. N-Carbobenzoxyl-L-tryptophan p-nitrophenyl ester ($\mathbf{6}$)

To a stirred solution of *N*-carbobenzoxyl-L-tryptophan (3 g, 8.9 mmol) and *p*-nitrophenol (1.26 g, 9.1 mmol) in THF (30 mL) was added *N*,*N'*-dicyclohexylcarbodiimide (DCC, 1.8 g, 8.9 mmol). The solution was kept at ambient temperature for 4 h and then filtered. The filtrate was condensed under reduced pressure below 45 °C and purified by flash chromatography with ethyl acetate/petrol ether (1/2 v/v), and 3.5 g of compound **6** was obtained (85%). Mp 105–107 °C. ¹H NMR (CDCl₃): $\delta = 8.19$ (s, 1H), 8.16 (br, 2H), 7.56 (d, 1H), 7.40 (d, 1H), 7.34 (br, 5H), 7.23 (t, 1H), 7.11 (m, 2H), 6.97 (d, 2H), 5.14 (s, 2H), 3.45 (m, 2H).

5.8. General procedures for preparation of **8a-d** using *N*-carbobenzoxyl-L-tryptophan-L-arginine (**8c**) as an example

5.8.1. N-Carbobenzoxyl-L-tryptophan-L-arginine (8c)

Unprotected arginine (0.23 g, 1.3 mmol) was dissolved in water (5 mL) at 45 °C, then the solution was cooled to room temperature and triethylamine (0.26 g, 2.6 mmol) was added. Then compound **6** (0.46 g, 1 mmol) in THF (5 mL) was added dropwise and stirred overnight. After evaporation, the oil was purified by flash chromatography (ethyl acetate/petrol ether = 1/2, v/v, then CH₂Cl₂/methane = 5/1, v/v) yielding a white solid **8c** (0.8 g, yield 94%). Mp 89–90 °C. ¹H NMR (CD₃OD): δ = 7.60 (d, 1H), 7.28 (m, 5H), 7.07 (m, 2H), 6.97 (t, 1H), 5.00 (d, 2H), 4.43 (q, 1H), 4.23 (t, 1H), 3.30 (m, 3H), 3.10 (m, 3H), 1.82 (m, 1H), 1.68 (m, 1H), 1.50 (m, 2H).

5.8.2. N-Carbobenzoxyl-L-tryptophan-L-glutamine (8a)

Compound **8a** was prepared from *N*-carbobenzoxyl-Ltryptophan *p*-nitrophenyl ester (**6**) and L-glutamine using a procedure similar to that described for the preparation of **8c** as a white solid, yield 99%. Mp 148–149 °C; ¹H NMR (CD₃OD): $\delta = 7.59$ (d, 1H), 7.2–7.35 (m, 6H), 7.09 (s, 1H), 7.06 (t, 1H), 6.98 (t, 1H), 4.97 (s, 2H), 4.46 (q, 1H), 4.25 (m, 1H), 3.34 (s, 1H), 3.10 (q, 1H), 3.15 (m, 3H), 1.92 (m, 1H).

5.8.3. N-Carbobenzoxyl-L-tryptophan-L-asparagine (8b)

Compound **8b** was prepared from *N*-carbobenzoxyl-Ltryptophan *p*-nitrophenyl ester (**6**) and L-asparagine using a procedure similar to that described for the preparation of **8c** as a white solid, yield 98%. Mp 140–141 °C; ¹H NMR (CD₃OD): $\delta = 7.56$ (d, 2H), 7.25 (m, 7H), 7.07 (s, 1H), 7.05 (t, 1H), 6.95 (t, 1H), 4.58 (t, 1H), 4.48 (t, 1H), 3.12 (m, 2H), 2.70 (m, 2H).

5.8.4. N-Carbobenzoxyl-L-tryptophan-L-leucine (8d)

Compound **8d** was prepared from *N*-carbobenzoxyl-L-tryptophan *p*-nitrophenyl ester (**6**) and L-leucine using a procedure similar to that described for the preparation of **8c** as a white solid, yield 90%. Mp 120–121 °C; ¹H NMR (CD₃OD): $\delta = 8.167$ (s, 1H), 7.69 (d, 1H), 7.34 (m, 6H), 7.13 (t, 1H), 7.11 (t, 1H), 7.08 (s, 1H), 6.08 (d, 1H), 5.52 (d, 1H), 5.11 (s, 2H), 4.51 (m, 2H), 3.31 (dd, 1H), 3.15 (q, 1H), 1.40 (m, 3H), 0.82 (t, 6H).

5.9. General procedures for preparation of **9a-d** using *L*-tryptophan-*L*-arginine (**9c**) as an example

5.9.1. L-Tryptophan-L-arginine (9c)

A solution of **8c** (0.59 g, 1.2 mmol) and TIPS (150 µL) in 80% methanol—water (15 mL) was hydrogenated in the presence of 0.05 g of 10% palladium on charcoal for 8 h. After filtration, the solution was evaporated and dissolved in water (20 mL), and washed with ethyl acetate (50 mL × 3). The water portion was followed by lyophilization to form a white solid **9c** (0.4 g, 100%). Mp 94–95 °C; $[\alpha]_D^{20}$ +22.0 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): δ = 10.93 (s, 1H), 9.16 (d, 1H), 8.13 (d, 1H), 7.54 (d, 1H), 7.33 (d, 1H), 7.20 (s, 1H), 7.05 (t, 1H), 6.96 (t, 1H), 3.91 (t, 1H), 3.43 (m, 2H), 3.01 (s, 2H), 2.72 (q, 1H), 1.62 (m, 2H), 1.55 (d, 2H); ESI-MS *m*/*z* 361, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₇H₂₅N₆O₃ $[M + H]^+$ 361.1988, found 361.1978.

5.9.2. L-Tryptophan-L-glutamine (9a)

Compound **9a** was prepared from *N*-carbobenzoxyl-L-tryptophan-L-glutamine (**8a**) using a procedure similar to that described for the preparation of **9c** as a white solid, yield 99%. Mp 165–167 °C; $[\alpha]_D^{20}$ +30.1 (*c* 0.5, water); ¹H NMR (CD₃OD): $\delta = 7.58$ (d, 1H), 7.35 (d, 1H), 7.22 (t, 1H), 6.99 (t, 1H), 4.46 (t, 1H), 4.26 (m, 1H), 3.20 (t, 1H), 2.23 (m, 3H), 1.90 (m, 1H); ESI-MS *m*/*z* 333, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₆H₂₁N₄O₄ $[M + H]^+$ 333.1563, found 333.1577.

5.9.3. L-Tryptophan-L-asparagine (9b)

Compound **9b** was prepared from *N*-carbobenzoxyl-L-tryptophan-L-asparagine (**8b**) using a procedure similar to that described for the preparation of **9c** as a white solid, 97%. Mp 143–144 °C; $[\alpha]_D^{20}$ –2.1 (*c* 0.5, water); ¹H NMR (DMSO*d*₆): δ = 7.61 (d, 1H), 7.34 (d, 1H), 7.22 (s, 1H), 6.99 (m, 3H), 4.38 (s, 1H), 3.63 (m, 2H), 3.20 (dd, 2H), 2.81 (q, 2H); ESI-MS *m*/*z* 319, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₅H₁₉N₃O₅ [M + H]⁺ 319.1416, found 319.1406.

5.9.4. L-Tryptophan-L-leucine (9d)

Compound **9d** was prepared from *N*-carbobenzoxyl-L-tryptophan-L-leucine (**8d**) using a procedure similar to that described for the preparation of **9c** as a white solid, yield 95%. Mp 122–124 °C; $[\alpha]_D^{20}$ +15.1 (*c* 0.5, water); ¹H NMR (DMSO-*d*₆): $\delta = 10.95$ (d, 1H), 8.73 (m, 1H), 7.60 (d, 1H), 7.31 (d, 1H), 7.19 (s, 1H), 7.03 (t, 1H), 6.94 (t, 1H), 4.16 (q, 1H), 3.67 (q, 1H), 3.15 (dd, 1H), 2.85 (q, 1H), 1.48 (m, 3H), 0.82 (t, 6H); ESI-MS m/z 318, $[M + H]^+$; HRMS (ESI) m/z calcd $C_{17}H_{24}N_3O_3$ $[M + H]^+$ 318.1818, found 318.1807.

5.10. General procedures for preparation of **12a-t** using **12a** as an example

$5.10.1.\ N-tert-Butoxy carbonyl-{\it L-tryptophan-}$

L-aspartate dibenzylester (12a)

To a mixture of *N-tert*-butoxycarbonyl-L-tryptophan (0.46 g, 1.5 mmol), L-dibenzylester aspartate *p*-toluenesulfonate (**3b**, 0.73 g, 1.5 mmol), 1-hydroxybenzotriazole (0.2 g, 1.5 mmol) and triethylamine (1.4 mL, 10 mmol) in 5% DMF in CH₂Cl₂ (20 mL), 1-ethyl-(3-dimethylaminopropyl)-carbodiimide hydrochloride (EDCI, 0.4 g, 2 mmol) was added in portions for 5 min. The solution was stirred overnight at rt. The solution was washed, dried, filtered and condensed *in vacuo* to form a yellow oil. The oil can be purified by flash chromatography with petroleum ether/ethyl acetate (2/1, v/v) to afford a white powder **12a** (0.89 g, 99%). Mp 60–61 °C; ¹H NMR (CDCl₃): δ = 7.55 (d, 1H), 7.20–7.39 (br s, 11H), 7.07 (m, 2H), 7.00 (t, 1H), 5.08 (s, 2H), 5.03 (s, 2H), 4.83 (m, 1H), 4.36 (m, 1H), 3.18 (dd, 1H), 3.00 (m, 1H), 2.80 (t, 2H), 1.34 (s, 9H).

5.10.2. N-tert-Butoxycarbonyl-L-tryptophan-

L-glutamate dimethylester (12b)

Compound **12b** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 92%. Mp 105–107 °C; ¹H NMR (CDCl₃): $\delta = 8.19$ (s, 1H), 7.63 (d, 1H), 7.35 (d, 1H), 7.17 (t, 1H), 7.11 (t, 1H), 7.09 (s, 1H), 6.44 (br, 1H), 5.12 (br, 1H), 4.51 (m, 1H), 4.44 (m, 1H), 3.64 (s, 3H), 3.62 (s, 3H), 3.33 (dd, 1H), 3.15 (q, 1H), 2.12 (m, 3H), 1.87 (m, 1H).

5.10.3. N-tert-Butoxycarbonyl-L-tryptophan-

L-aspartate dimethylester (12c)

Compound **12c** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-aspartate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 80%. Mp 76–78 °C; ¹H NMR (CDCl₃): $\delta = 8.21$ (br, 1H), 7.63 (d, 1H), 7.34 (d, 1H), 7.16 (t, 1H), 7.10 (t, 1H), 7.08 (s, 1H), 6.78 (d, 1H), 5.12 (br, 1H), 4.72 (s, 1H), 4.45 (br, 1H), 3.66 (s, 3H), 3.57 (s, 3H), 3.29 (dd, 1H), 3.18 (q, 1H), 2.88 (m, 2H), 2.75 (dd, 1H), 1.42 (s, 9H).

5.10.4. N-tert-Butoxycarbonyl-L-tryptophan-

L-2-aminoadipate dimethylester (12d)

Compound **12d** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-2-aminoadipate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 90%. Mp 130– 131 °C; ¹H NMR (CDCl₃): $\delta = 7.61$ (d, 1H), 7.33 (d, 2H), 7.08 (m, 3H), 4.45 (m, 1H), 4.07 (m, 1H), 3.65 (s, 3H), 3.62 (s, 3H), 3.33 (dd, 1H), 3.21 (q, 1H), 2.21 (m, 2H), 1.62 (m, 2H), 1.52 (m, 2H), 1.48 (s, 9H).

5.10.5. N-tert-Butoxycarbonyl-L-tryptophan-L-leucine benzylester (12e)

Compound **12e** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-leucine benzylester using a procedure similar to that described for the preparation of **12a** as a white solid, yield 95%. Mp 90–91 °C; $[\alpha]_D^{20}$ –1.2 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.65 (d, 1H), 7.34 (d, 1H), 7.16 (m, 6H), 7.07 (m, 2H), 5.12 (s, 2H), 4.58 (m, 1H), 3.84 (dd, 1H), 3.40 (dd, 1H), 2.95 (q, 1H), 1.55 (m, 3H), 1.48 (s, 9H), 0.90 (s, 3H), 0.88 (s, 3H).

5.10.6. N-tert-Butoxycarbonyl-L-tryptophan-L-leucine methylester (**12f**)

Compound **12f** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-leucine methylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 90%. Mp 68–70 °C; ¹H NMR (CDCl₃): $\delta = 8.35$ (d, 1H), 7.65 (d, 1H), 7.33 (d, 1H), 7.16 (t, 1H), 7.07 (m, 2H), 4.45 (m, 1H), 3.75 (dd, 1H), 3.34 (dd, 1H), 2.95 (q, 1H), 1.55 (m, 3H), 1.48 (s, 9H), 1.20 (s, 3H), 1.17 (s, 3H).

5.10.7. N-tert-Butoxycarbonyl-L-2-naphthylalanine-L-glutamate dibenzylester (**12g**)

Compound **12g** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine and L-glutamate dibenzylester *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **12a** as a white solid, yield 98%. Mp 140–141 °C; ¹H NMR (CD₃OD): $\delta = 7.73$ (m, 3H), 7.65 (s, 1H), 7.27–7.40 (m, 13H), 5.11 (d, 2H), 5.02 (s, 2H), 4.41 (m, 1H), 3.17 (dd, 1H), 2.97 (q, 1H), 2.05 (m, 3H), 1.80 (m, 1H), 1.28 (s, 9H).

5.10.8. N-tert-Butoxycarbonyl-L-tryptophan-D-glutamate dimethylester (12h)

Compound **12h** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and D-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 90%. ¹H NMR (CDCl₃): $\delta = 7.62$ (d, 1H), 7.35 (d, 1H), 7.16 (t, 1H), 7.08 (t, 1H), 7.05 (s, 1H), 4.43 (m, 1H), 4.22 (m, 1H), 3.23 (s, 3H), 3.18 (s, 3H), 3.08 (dd, 1H), 3.01 (q, 1H), 2.11 (m, 3H), 1.90 (m, 1H).

5.10.9. N-tert-Butoxycarbonyl-L-phenylalanine-L-aspartate dibenzylester (**12i**)

Compound **12i** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine and L-aspartate dibenzylester *p*-toluenesulfonate (**3b**) using a procedure similar to that described for the preparation of **12a** as a white solid, yield 98%. ¹H NMR (CDCl₃): $\delta = 7.84$ (s, 1H), 7.60 (d, 1H), 6.77–7.36 (m, 13H), 6.96 (s, 1H), 6.75 (d, 1H), 5.06 (s, 2H), 4.95 (d, 2H), 4.76 (m, 1H), 4.43 (br, 1H), 3.48 (dd, 1H), 3.17 (q, 1H), 2.99 (dd, 1H), 2.81 (m, 1H), 1.41 (s, 9H).

5.10.10. N-tert-Butoxycarbonyl-L-phenylalanine-Lglutamate dimethylester (12j)

Compound **12j** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine and L-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 99%. ¹H NMR (CDCl₃): $\delta = 7.23$ (m, 5H), 4.68 (t, 1H), 3.64 (s, 3H), 3.58 (s, 3H), 3.52 (m, 1H), 2.90 (m, 2H), 2.66 (q, 2H), 1.42 (s, 9H).

5.10.11. N-tert-Butoxycarbonyl-L-2-naphthylalanine-L-glutamate dimethylester (**12k**)

Compound **12k** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine and L-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 99%. Mp 136– 137 °C; ¹H NMR (CD₃OD): $\delta = 7.76-7.81$ (m, 3H), 7.64 (m, 1H), 7.45 (m, 2H), 7.35 (dd, 1H), 6.66 (br, 1H), 5.02 (br, 1H), 4.50 (m, 2H), 3.67 (s, 3H), 3.60 (s, 3H), 3.22 (d, 2H), 2.08 (m, 3H), 1.81 (m, 1H), 1.38 (s, 9H).

5.10.12. N-tert-Butoxycarbonyl-L-tryptophan-D-glutamate dibenzylester (12l)

Compound **121** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan and L-glutamate dibenzylester *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **12a** as a white solid, yield 85%. Mp 123–124 °C; ¹H NMR (CD₃OD): $\delta = 7.54$ (d, 1H), 7.24–7.34 (m, 11H), 7.04 (m, 3H), 5.10 (d, 2H), 5.06 (s, 2H), 4.35 (m, 2H), 3.14 (dd, 1H), 3.02 (q, 1H), 2.01 (m, 3H), 1.78 (m, 1H).

5.10.13. N-tert-Butoxycarbonyl-L-phenylalanine-L-glutamate dibenzylester (**12m**)

Compound **12m** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine and L-glutamate dibenzylester *p*-toluenesulfonate (**3a**) using a procedure similar to that described for the preparation of **12a** as a white solid, yield 95%. Mp 100– 101 °C; ¹H NMR (CD₃OD): $\delta = 7.35$ (m, 10H), 7.23 (m, 5H), 6.55 (d, 1H), 5.15 (d, 2H), 5.10 (s, 2H), 4.62 (m, 1H), 4.34 (d, 1H), 3.05 (m, 2H), 2.30 (m, 3H), 1.98 (m, 1H), 1.63 (s, 2H), 1.40 (s, 9H).

5.10.14. N-tert-Butoxycarbonyl-L-2-naphthylalanine-L-leucine methylester (**12n**)

Compound **12n** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine and L-leucine methylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 99%. Mp 41–42 °C; ¹H NMR (CDCl₃): $\delta = 7.79$ (m, 3H), 7.64 (s, 1H), 7.47 (m, 2H), 7.37 (d, 1H), 6.12 (d, 1H), 5.07 (br, 1H), 4.48 (br, 2H), 3.67 (s, 3H), 3.23 (m, 2H), 1.74 (s, 1H), 1.39 (s, 9H), 0.70 (t, 6H).

5.10.15. N-tert-Butoxycarbonyl-D-phenylalanine-L-2-aminoadipate dimethylester (**120**)

Compound **120** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine and L-2-aminoadipate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 99%. Mp 85–86 °C; ¹H NMR (CDCl₃): δ = 7.19–7.31 (m, 5H), 6.38 (d, 1H), 5.00 (br, 1H), 4.51 (q, 1H), 4.37 (q, 1H), 3.72 (s, 3H), 3.67 (s, 3H), 3.05 (d, 2H), 2.25 (t, 2H), 1.71 (m, 1H), 1.58 (m, 1H), 1.41 (s, 9H).

5.10.16. N-tert-Butoxycarbonyl-D-phenylalanine-L-leucine methylester (**12p**)

Compound **12p** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine and L-leucine methylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 98%. Mp 128–129 °C; ¹H NMR (CDCl₃): $\delta = 7.19-7.33$ (m, 5H), 6.15 (d, 1H), 4.99 (br, 1H), 4.52 (m, 1H), 4.37 (m, 1H), 3.70 (s, 3H), 3.06 (d, 1H), 1.67 (m, 1H), 1.39 (s, 9H), 0.85 (t, 6H).

5.10.17. N-tert-Butoxycarbonyl-D-phenylalanine-L-aspartate dimethylester (**12q**)

Compound **12q** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine and L-aspartate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 96%. Mp 100–101 °C; ¹H NMR (CDCl₃): $\delta = 7.19-7.32$ (m, 5H), 4.88 (m, 1H), 3.65 (s, 3H), 3.62 (s, 3H), 3.59 (m, 1H), 2.92 (m, 2H), 2.68 (m, 2H), 1.45 (s, 9H).

5.10.18. N-tert-Butoxycarbonyl-D-phenylalanine-Lglutamate dimethylester (12r)

Compound **12r** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine and L-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 95%. Mp 95–96 °C; ¹H NMR (CDCl₃): $\delta = 7.18-7.31$ (m, 5H), 6.53 (d, 1H), 4.96 (br, 1H), 4.55 (m, 1H), 4.33 (q, 1H), 3.70 (s, 3H), 3.06 (t, 2H), 2.22 (m, 2H), 2.15 (m, 1H), 1.90 (m, 1H), 1.41 (s, 9H).

5.10.19. O-Methyl-N-tert-butoxycarbonyl-L-tyrosine-L-glutamate dimethylester (**12s**)

Compound **12s** was prepared from *O*-methyl-*N*-tert-butoxycarbonyl-L-tyrosine and L-glutamate dimethylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 98%. Mp 99– 100 °C; ¹H NMR (CDCl₃): $\delta = 7.10$ (d, 2H), 6.83 (d, 2H), 6.51 (d, 1H), 4.94 (br, 1H), 4.56 (m, 1H), 4.29 (q, 1H), 3.78 (s, 3H), 3.71 (s, 3H), 3.66 (s, 3H), 2.97 (m, 2H), 2.13–2.34 (m, 3H), 1.94 (m, 1H), 1.37 (s, 9H).

5.10.20. N-tert-Butoxycarbonyl-L-phenylalanine-L-alanine methylester (12t)

Compound **12t** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine and L-alanine methylester hydrochloride using a procedure similar to that described for the preparation of **12a** as a white solid, yield 95%. Mp 99–101 °C; ¹H NMR (CDCl₃): $\delta = 7.18-7.31$ (m, 5H), 6.47 (d, 1H), 5.01 (br, 1H), 4.51 (q, 1H), 4.36 (br, 1H), 3.70 (s, 3H), 3.06 (t, 2H), 1.42 (s, 9H), 1.33 (d, 3H).

5.11. General procedures for preparation of **13a-t** using **13a** as an example

5.11.1. L-Tryptophan-L-aspartate dibenzylester (13a)

To a solution of **12a** (0.5 g, 0.8 mmol) in anhydrous CH₂Cl₂ (10 mL) and TIPS (0.1 mL) in ice bath, TFA (0.25 mL) was added. After 4 h the solution was allowed to gradually reach ambient temperature and stirred for another 8 h. The solvent and TFA were removed by reduced pressure below 30 °C using additional amounts of CH₂Cl₂ to help complete evaporation of TFA. The dark oil was purified by chromatography with CH₂Cl₂/MeOH (20/1, v/v) to form a white solid **13a** (0.3 g, 82%). Mp 88–89 °C; $[\alpha]_{D}^{20}$ –21.2 (*c* 0.5, MeOH); ¹H NMR (CDCl₃) δ = 7.59 (t, 2H), 7.29 (m, 8H), 7.18 (m, 4H), 6.44 (s, 1H), 6.22 (s, 1H), 5.05 (s, 2H), 4.97 (dd, H), 4.70 (s, 2H), 3.33 (m, 2H), 2.78 (m, 2H); ESI-MS *m*/*z* 500, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₂₉H₃₀N₃O₅ [M + H]⁺ 500.2182, found 500.2185.

5.11.2. L-Tryptophan-L-glutamate dimethylester (13b)

Compound **13b** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-L-glutamate dimethylester (**12b**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 90%. Mp 110–111 °C; $[\alpha]_D^{20}$ –12.7 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.33$ (br, 1H), 7.90 (d, 1H), 7.67 (d, 1H), 7.38 (d, 1H), 7.20 (t, 1H), 7.12 (m, 2H), 4.60 (m, 1H), 3.83 (q, 1H), 3.77 (s, 3H), 3.67 (s, 3H), 3.35 (dd, 1H), 3.06 (q, 1H), 2.30 (m, 4H); ESI-MS *m*/*z* 362, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₈H₂₄N₃O₅ [M + H]⁺ 362.1700, found 362.1716.

5.11.3. L-Tryptophan-L-aspartate dimethylester (13c)

Compound **13c** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-L-aspartate dimethylester (**12c**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 65%. Mp 231–233 °C; $[\alpha]_D^{20}$ –42.8 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.15$ (1H), 7.64 (d, 1H), 7.34 (d, 1H), 7.16 (t, 1H), 7.11 (t, 1H), 7.09 (s, 1H), 6.75 (d, 1H), 5.18 (m, 1H), 4.73 (m, 1H), 4.45 (m, 1H), 3.67 (s, 3H), 3.58 (s, 3H), 3.29 (dd, 1H), 3.20 (q, 1H), 2.93 (m, 1H), 2.76 (m, 1H), 2.68 (dd, 1H); ESI-MS *m*/*z* 348, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₇H₂₂N₃O₅ [M + H]⁺ 348.1559, found 348.1548.

5.11.4. L-Tryptophan-L-2-aminoadipate dimethylester (13d)

Compound **13d** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-L-2-aminoadipate dimethylester (**12d**) using a procedure similar to that described for the preparation of **13a** as a yellow oil, yield 88%. Oil; $[\alpha]_D^{20}$ –4.6 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.78$ (br, 1H), 7.95 (m, 1H), 7.60 (d, 1H), 7.38 (d, 1H), 7.15 (m, 2H), 7.06 (t, 1H), 4.44 (m, 1H), 4.05 (m, 1H), 3.67 (s, 3H), 3.61 (s, 3H), 3.26 (dd, 1H), 3.21 (q, 1H), 2.22 (m, 2H), 1.74 (m, 1H), 1.62 (m, 1H), 1.51 (m, 2H), 1.24 (s, 1H); ESI-MS *m*/*z* 376, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₉H₂₆N₃O₅ [M + H]⁺ 376.1872, found 376.1878.

5.11.5. L-Tryptophan-L-leucine benzylester (13e)

Compound **13e** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-L-leucine benzylester (**12e**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 93%. Mp 78–80 °C; $[\alpha]_D^{20}$ +5.4 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.66 (d, 1H), 7.34 (d, 1H), 7.16 (m, 6H), 7.10 (s, 1H), 7.07 (t, 1H), 5.12 (s, 2H), 4.58 (m, 1H), 3.83 (m, 1H), 3.27 (dd, 1H), 2.95 (q, 1H), 1.56 (m, 1H), 1.50 (m, 2H), 0.87 (s, 3H), 0.85 (s, 3H); ESI-MS *m*/*z* 408, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₂₄H₃₀N₃O₃ [M + H]⁺ 408.2287, found 408.2301.

5.11.6. L-Tryptophan-L-leucine methylester (13f)

Compound **13f** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-L-leucine methylester (**12f**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 76%. Mp > 250 °C; $[\alpha]_D^{20} - 1.2$ (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.40$ (br, 1H), 7.66 (d, 1H), 7.43 (d, 1H), 7.16 (t, 1H), 7.07 (t, 1H), 7.03 (s, 1H), 6.08 (d, 1H), 4.40 (m, 1H), 3.7 (dd, 1H), 3.38 (dd, 1H), 2.95 (q, 1H), 1.55 (m, 1H), 1.52 (m, 2H), 1.20 (s, 3H), 1.17 (s, 3H); ESI-MS *m*/*z* 332, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₈H₂₆N₃O₃ $[M + H]^+$ 332.1974, found 332.1980.

5.11.7. L-2-Naphthylalanine-L-glutamate dibenzylester (13g)

Compound **13g** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine-L-glutamate dibenzylester (**12g**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 92%. Mp 100–102 °C; $[\alpha]_D^{20}$ -36.4 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): δ = 7.99 (s, 1H), 7.69–7.82 (m, 4H), 7.27–7.44 (m, 13H), 5.48 (s, 1H), 5.13 (s, 2H), 5.01 (s, 2H), 4.43 (dd, 1H), 4.10 (m, 1H), 3.21 (m, 2H), 1.90 (m, 3H), 1.72 (m, 1H); ESI-MS *m*/*z* 525, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₃₂H₃₃N₂O₅ [M + H]⁺ 525.2390, found 525.2377.

5.11.8. L-Tryptophan-D-glutamate dimethylester (13h)

Compound **13h** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-D-glutamate dimethylester (**12h**) using a procedure similar to that described for the preparation of **13a** as a white solid. Mp 120–121 °C; $[\alpha]_D^{20}$ –2.5 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.32$ (br, 1H), 7.70 (br, 1H), 7.67 (d, 1H), 7.40 (d, 1H), 7.25 (t, 1H), 7.18 (t, 1H), 7.10 (s, 1H), 4.44 (m, 1H), 3.78 (q, 1H), 3.65 (s, 3H), 3.57 (s, 3H), 3.35 (dd, 1H), 3.21 (q, 1H), 2.30 (m, 4H); ESI-MS *m*/*z* 362, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₈H₂₄N₃O₅ [M + H]⁺ 362.1700, found 362.1706.

5.11.9. L-Phenylalanine-L-aspartate dibenzylester (13i)

Compound **13i** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine-L-aspartate dibenzylester (**12i**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 92%. Mp 136–137 °C; $[\alpha]_D^{20}$ +1.7 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 7.63$ (d, 1H), 7.12–7.32 (m, 15H), 5.03 (s, 2H), 4.92 (d, 2H), 4.74 (m, 1H), 4.30 (m, 1H), 3.12 (d, 2H), 2.84 (m, 2H); ESI-MS *m*/z 461, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₂₇H₂₉N₂O₅ $[M + H]^+$ 461.2077, found 461.2089.

5.11.10. L-Phenylalanine-L-glutamate dimethylester (13j)

Compound **13j** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine-L-glutamate dimethylester (**12j**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 94%. Mp 225–226 °C; $[\alpha]_D^{20}$ +3.4 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.24 (m, 5H), 4.65 (t, 1H), 3.66 (s, 3H), 3.63 (s, 3H), 3.52 (m, 1H), 2.88 (m, 2H), 2.60 (m, 2H); ESI-MS *m/z* 309, $[M + H]^+$; HRMS (ESI) *m/z* calcd C₁₅H₂₁N₂O₅ $[M + H]^+$ 309.1450, found 309.1438.

5.11.11. L-2-Naphthylalanine-L-glutamate dimethylester (13k)

Compound **13k** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine-L-glutamate dimethylester (**12k**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 97%. Mp 89–90 °C; $[\alpha]_D^{20}$ –45.9 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): $\delta = 7.77$ (m, 3H), 7.66 (m, 1H), 7.34 (m, 2H), 7.23 (dd, 1H), 6.50 (d, 1H), 5.02 (m, 1H), 4.52 (m, 2H), 3.57 (s, 3H), 3.50 (s, 3H), 3.12 (d, 2H), 2.18 (m, 3H), 1.90 (m, 1H); ESI-MS *m*/*z* 373, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₂₀H₂₅N₂O₅ [M + H]⁺ 373.1763, found 373.1763.

5.11.12. L-Tryptophan-D-glutamate dibenzylester (131)

Compound **13I** was prepared from *N*-tert-butoxycarbonyl-L-tryptophan-D-glutamate dibenzylester (**12I**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 87%. Mp 54–55 °C; $[\alpha]_D^{20}$ +44.3 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.54 (br, 1H), 6.89–7.31 (m, 15H), 4.94 (s, 2H), 4.71 (br, 2H), 4.31 (m, 1H), 3.35 (br, 2H), 2.89 (m, 1H), 1.82 (m, 4H); ESI-MS *m*/*z* 514, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₃₀H₃₂N₃O₅ [M + H]⁺ 514.2342, found 514.2356.

5.11.13. L-Phenylalanine-L-glutamate dibenzylester (13m)

Compound **13m** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine-L-glutamate dibenzylester (**12m**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 89%. Mp 190–192 °C; $[\alpha]_D^{20}$ –35.7 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): δ = 7.66 (d, 1H), 7.09– 7.41 (m, 15H), 5.13 (s, 2H), 5.01 (d, 2H), 4.76 (m, 1H), 4.44 (m, 1H), 3.12 (m, 2H), 2.03 (m, 3H), 1.78 (m, 1H); ESI-MS *m*/*z* 475, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₂₈H₃₁N₂O₅ [M + H]⁺ 475.2234, found 475.2248.

5.11.14. L-2-Naphthylalanine-L-leucine methylester (13n)

Compound **13n** was prepared from *N*-tert-butoxycarbonyl-L-2-naphthylalanine-L-leucine methylester (**12n**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 94%. Mp 94–96 °C; $[\alpha]_D^{20}$ –70.0 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): $\delta = 8.21$ (br, 1H), 7.77 (m, 4H), 7.52 (d, 1H), 7.42 (q, 2H), 4.70 (br, 1H), 4.38 (m, 1H), 3.43 (s, 3H), 1.35 (m, 2H), 1.29 (s, 1H), 1.12 (m, 1H), 0.85 (m, 1H), 0.65 (q, 6H); ESI-MS *m*/*z* 343, $[M + H]^+$; HRMS

(ESI) m/z calcd $C_{20}H_{27}N_2O_3$ $[M + H]^+$ 343.2022, found 343.2015.

5.11.15. D-Phenylalanine-L-2-aminoadipate dimethylester (**130**)

Compound **130** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine-L-2-aminoadipate dimethylester (**120**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 96%. Mp 218–219 °C; $[\alpha]_D^{20}$ -26.5 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.83 (br, 1H), 7.23–7.33 (m, 5H), 4.55 (q, 1H), 3.80 (m, 1H), 3.72 (s, 3H), 3.65 (s, 3H), 3.28 (dd, 1H), 2.78 (q, 1H), 2.29 (m, 3H), 1.83 (m, 1H), 1.61 (m, 2H); ESI-MS *m*/*z* 337, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₇H₂₅N₂O₅ [M + H]⁺ 337.1763, found 337.1795.

5.11.16. D-Phenylalanine-L-leucine methylester (13p)

Compound **13p** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine-L-leucine methylester (**12p**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 95%. Mp 82–84 °C; $[\alpha]_D^{20}$ –81.0 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): $\delta = 7.22-7.34$ (m, 5H), 4.32 (q, 1H), 3.88 (t, 1H), 3.69 (s, 3H), 3.01 (dd, 2H), 1.65 (m, 1H), 1.37 (s, 9H), 0.85 (t, 6H); ESI-MS *m*/*z* 293, $[M + H]^+$; HRMS (ESI) *m*/*z* calcd C₁₆H₂₅N₂O₃ $[M + H]^+$ 293.1865, found 293.1855.

5.11.17. D-Phenylalanine-L-aspartate dimethylester (13q)

Compound **13q** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine-L-aspartate dimethylester (**12q**) using a procedure similar to that described for the preparation of **13a** as a white solid. Mp 59–60 °C; $[\alpha]_D^{20}$ –19.8 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): $\delta = 7.19-7.27$ (m, 5H), 4.75 (t, 1H), 3.70 (s, 3H), 3.64 (s, 3H), 3.59 (t, 1H), 2.92 (q, 1H), 2.81 (q, 1H), 2.68 (dd, 2H); EI-MS *m*/*z* 309 (M + 1), 120 (100%); HRMS (EI) *m*/*z* calcd C₁₅H₂₁N₂O₅ (M + 1) 309.1445, found 309.1450.

5.11.18. D-Phenylalanine-L-glutamate dimethylester (13r)

Compound **13r** was prepared from *N*-tert-butoxycarbonyl-D-phenylalanine-L-glutamate dimethylester (**12r**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 93%. Mp 121–122 °C; $[\alpha]_D^{20}$ –8.3 (*c* 0.5, MeOH); ¹H NMR (CDCl₃): δ = 7.20–7.33 (m, 5H), 6.55 (d, 1H), 5.01 (br, 1H), 4.57 (m, 1H), 4.37 (m, 1H), 3.73 (s, 3H), 3.68 (s, 3H), 3.07 (d, 2H), 2.16 (m, 2H), 1.86 (m, 1H); ESI-MS *m*/*z* 323, [M + H]⁺; HRMS (ESI) *m*/*z* calcd C₁₆H₂₃N₂O₅ [M + H]⁺ 323.1607, found 323.1601.

5.11.19. O-Methyl-L-tyrosine-L-glutamate dimethylester (13s)

Compound **13s** was prepared from *O*-methyl-*N*-tert-butoxycarbonyl-L-tyrosine-L-glutamate dimethylester using a procedure similar to that described for the preparation of **13a** as a white solid, yield 91%. Mp 253–254 °C; $[\alpha]_D^{20}$ –0.8 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): δ = 7.12 (d, 2H), 6.85 (d, 2H), 4.41 (q, 1H), 3.76 (s, 3H), 3.69 (s, 3H), 3.61 (s, 3H), 2.95 (dd, 1H), 2.84 (q, 1H), 2.34 (t, 2H), 1.15 (s, 2H), 1.91 (m, 1H); ESI-MS m/z 353, $[M + H]^+$; HRMS (ESI) m/z calcd $C_{17}H_{25}N_2O_6$ $[M + H]^+$ 353.1713, found 353.1705.

5.11.20. L-Phenylalanine-L-alanine methylester (13t)

Compound **13t** was prepared from *N*-tert-butoxycarbonyl-L-phenylalanine-L-alanine methylester (**12t**) using a procedure similar to that described for the preparation of **13a** as a white solid, yield 93%. Mp > 250 °C; $[\alpha]_D^{20}$ +13.5 (*c* 0.5, MeOH); ¹H NMR (CD₃OD): δ = 7.18–7.33 (m, 5H), 4.43 (q, 1H), 4.30 (t, 1H), 3.70 (s, 3H), 2.93 (m, 2H), 1.37 (d, 3H); EI-MS *m*/*z* 250 (M⁺), 120 (100%); HRMS (EI) *m*/*z* calcd C₁₃H₁₈N₂O₃ (M⁺) 250.1317, found 250.1325.

Acknowledgements

We gratefully acknowledge the generous support from the National Natural Science Foundation of China (Grants 20372069, 29725203 and 20472094), the Key Technologies R&D Program from CAS (Grants 2006BAI01B02 and 2005BA711A01), and the 863 Hi-Tech Program of China (Grants 2006AA020402 and 2006AA020602).

References

- [1] A.R. Miller, Drug Development Research 67 (2006) 574-578.
- [2] M.C. Carmona, K. Louche, B. Lefebvre, A. Pilon, N. Hennuyer, V. Audinot-Bouchez, C. Fievet, G. Torpier, P. Formstecher, P. Renard, P. Lefebvre, C. Dacquet, B. Staels, L. Casteilla, L. Penicaud, Diabetes 56 (2007) 2797–2808.
- [3] V.T. Chetty, A.M. Sharma, Vascular Pharmacology 45 (2006) 46-53.
- [4] S. Kersten, B. Desvergne, W. Wahli, Nature 405 (2000) 421-424.
- [5] N.B. Mettu, T.B. Stanley, M.A. Dwyer, M.S. Jansen, J.E. Allen, J.M. Hall, D.P. McDonnell, Molecular Endocrinology 21 (2007) 2361–2377.
- [6] W. Walter, Swiss Medical Weekly 132 (2002) 83-89.
- [7] Z. Jiming, M.W. Katina, D.M. Jheem, Biochemical and Biophysical Research Communications 293 (2002) 274–283.
- [8] T. Fujimura, C. Kimura, T. Oe, Y. Takata, H. Sakuma, I. Aramori, S. Mutoh, Journal of Pharmacology and Experimental Therapeutics 318 (2006) 863–871.
- [9] L. Nagy, P. Tontonoz, J.G. Alvarez, H. Chen, R.M. Evans, Cell 93 (1998) 229–240.
- [10] N. Marx, H. Duez, J.C. Fruchart, B. Staels, Circulation Research 94 (2004) 1168–1178.
- [11] H. Kitajima, M. Nakamura, H. Tamakawa, N. Goto, Bioorganic & Medicinal Chemistry Letters 10 (2000) 2453–2456.
- [12] K. Yamamoto, T. Itoh, D. Abe, M. Shimizu, T. Kanda, T. Koyama, M. Nishikawa, T. Tamai, H. Ooizumi, S. Yamada, Bioorganic & Medicinal Chemistry Letters 15 (2005) 517–522.
- [13] J. Auwerx, Diabetologia 42 (1999) 1033-1049.
- [14] T. Yamauchi, H. Waki, J. Kamon, K. Murakami, K. Motojima, K. Komeda, H. Miki, N. Kubota, Y. Terauchi, A. Tsuchida, N. Tsuboyama-Kasaoka, N. Yamauchi, T. Ide, W. Hori, S. Kato, M. Fukayama, Y. Akanuma, O. Ezaki, A. Itai, R. Nagai, S. Kimura, K. Tobe, H. Kagechika, K. Shudo, T. Kadowaki, Journal of Clinical Investigation 108 (2001) 1001–1013.
- [15] K.R. Kim, H.N. Choi, H.J. Lee, H.A. Baek, H.S. Park, K.Y. Jang, M.J. Chung, W.S. Moon, Oncology 18 (2007) 825–832.
- [16] J.P. Berger, A.E. Petro, K.L. Macnaul, L.J. Kelly, B.B. Zhang, K. Richards, A. Elbrecht, B.A. Johnson, G. Zhou, T.W. Doebber, C. Biswas, M. Parikh, N. Sharma, M.R. Tanen, G.M. Thompson, J. Ventre, A.D. Adams, R. Mosley, R.S. Surwit, D.E. Moller, Molecular Endocrinology 17 (2003) 662–676.

- [17] T. Allen, F. Zhang, S.A. Moodie, L.E. Clemens, A. Smith, F. Gregoire, A. Bell, G.E. Muscat, T.A. Gustafson, Diabetes 55 (2006) 2523–2533.
- [18] A.H. Barnett, S. Bellary, Practical Diabetes International 23 (2006) 129–134.
- [19] J. Rosenstock, F. Flores-Lozano, S. Schwartz, G. Gonzalez-Galvez, D. Karpf, Diabetes 54 (Suppl. 1) (2005) A11.
- [20] G. Lee, F. Elwood, J. McNally, J. Weiszmann, M. Lindstrom, K. Amaral, M. Nakamura, S. Miao, P. Cao, R.M. Learned, J.L. Chen, Y. Li, Journal of Biological Chemistry 31 (2002) 19649–19657.
- [21] L.M. Leesnitzer, D.J. Parks, R.K. Bledsoe, J.E. Cobb, J.L. Collins, T.G. Consler, R.G. Davis, E.A. Hull-Ryde, J.M. Lenhard, L. Patel, K.D. Plunket, J.L. Shenk, J.B. Stimmel, C. Therapontos, T.M. Willson, S.G. Blanchard, Biochemistry 41 (2002) 6640–6650.
- [22] J. Rieusset, F. Touri, L. Michalik, P. Escher, B. Desvergne, E. Niesor, W. Wahli, Molecular Endocrinology 16 (2002) 2628–2644.
- [23] H.M. Wright, C.B. Clish, T. Mikami, S. Hauser, K. Yanagi, R. Hiramatsu, C.N. Serhan, B.M. Spiegelman, Journal of Biological Chemistry 275 (2000) 1873–1877.
- [24] H.S. Camp, A. Chaudhry, T. Leff, Endocrinology 142 (2001) 3207-3213.
- [25] E. Burgermeister, A. Schnoebelen, A. Flament, J. Benz, M. Stihle, B. Gsell, A. Rufer, A. Ruf, B. Kuhn, H.P. Märki, J. Mizrahi, E. Sebokova, E. Niesor, M. Meyer, Molecular Endocrinology 20 (2006) 809-830.
- [26] J. Nishiu, M. Ito, Diabetes Obesity & Metabolism 8 (2006) 508-516.
- [27] J.N. Feige, L. Gelman, D. Rossi, V. Zoete, R. Metivier, C. Tudor, S.I. Anghel, A. Grosdidier, C. Lathion, Y. Engelborghs, O. Michielin,

W. Wahli, B. Desvergne, Journal of Biological Chemistry 282 (2007) 19152–19166.

- [28] F. Ye, Z. Zhen, H. Lou, J. Shen, K. Chen, X. Shen, H. Jiang, ChemBio-Chem. 7 (2006) 74–83.
- [29] M. Wilchek, A. Patchornik, Journal of Organic Chemistry 28 (1963) 1874–1875.
- [30] H.E. Xu, T.B. Stanley, V.G. Montana, M.H. Lambert, B.G. Shearer, J.E. Cobb, D.D. Mckee, C.M. Galardi, K.D. Plunket, R.T. Nolte, D.J. Parks, J.T. Moore, S.A. Kliewer, T.M. Willson, J.B. Stimmel, Nature 415 (2002) 813–817.
- [31] R.T. Nolte, G.B. Wisely, S. Westin, J.E. Cobb, M.H. Lambert, R. Kurokawa, M.G. Rosenfeld, T.M. Willson, C.K. Glass, M.V. Milburn, Nature 395 (1998) 137–143.
- [32] Insight II 2000, User Guide, MSI Inc., San Diego, USA, 2000.
- [33] A.C. Wallace, R.A. Laskowski, J.M. Thornton, Protein Engineering 8 (1995) 127–134.
- [34] Q. Chen, J. Chen, T. Sun, J. Shen, X. Shen, H. Jiang, Analytical Biochemistry 335 (2004) 253–259.
- [35] R.D. Gietz, R.A. Woods, Methods in Enzymology 350 (2002) 87-96.
- [36] P.S. Lazo, A.G. Ochoa, S. Gascon, European Journal of Biochemistry 77 (1977) 375–382.
- [37] J.G. Vinter, A. Davis, M.R. Saunders, Journal of Computer Aided Molecular Design 1 (1987) 31–51.
- [38] R.A. Laskowski, M.W. MacArthur, D.S. Moss, J.M. Thornton, Journal of Applied Crystallography 26 (1993) 283–291.
- [39] Profiles-3D, User Guide, MSI Inc., San Diego, USA, 1998.