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Overcoming Mutagenicity and Ion Channel Activity: Optimization of Selective Spleen Tyrosine Kinase Inhibitors

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Spleen Tyrosine Kinase. Ames. Mutagenicity. Ion Channels.

Abstract

Development of a series of highly kinome selective Spleen Tyrosine Kinase (Syk) inhibitors with favorable drug-like properties is described. Early leads were discovered through X-ray crystallographic analysis, and a systematic survey of cores within a selected chemical space focused on ligand binding efficiency. Attenuation of hERG ion channel activity inherent within the initial chemotype was guided through modulation of physicochemical properties including logD, PSA, and pKa. PSA proved most effective for prospective compound design. Further profiling of an advanced compound revealed bacterial mutagenicity in the Ames test using TA97a *Salmonella* strain, and subsequent study demonstrated that this mutagenicity was pervasive throughout the series. Identification of intercalation as a likely mechanism for the mutagenicity enabled modification of the core scaffold. Implementation of a DNA binding assay as a pre-screen and models in DNA allowed resolution of the mutagenicity risk, affording molecules with favorable potency, selectivity, pharmacokinetic, and off-target profiles.

Introduction

The utility of kinase inhibitors in treating oncological disorders was affirmed by the approval of imatinib in 2001.¹ In recent years, small molecule inhibitors of protein kinases have been examined for use in various autoimmune disorders, including rheumatoid arthritis (RA).² RA is a systemic inflammatory disorder that affects ~1% of the worldwide population with women three times more likely than men to develop the disease.³ Disease progression is characterized by

synovial hyperplasia, inflammation, and irreversible joint destruction. This commonly leads to disability and consequent reduction in the quality of life and, if untreated, reduces life expectancy in patients. The pathogenesis of RA consists of a cascade of pro-inflammatory cytokines and chemokines leading to the recruitment of inflammatory cells. Traditionally, T cells, macrophages, neutrophils, and synovial fibroblasts were thought to be associated with RA;⁴ however, current evidence suggests that autoreactive B cells, immune complexes, osteoclasts, mast cells, and basophils also contribute to the disease.⁵

Current therapies used to treat RA generally fall into two broad categories: palliative therapies and disease-modifying agents. Palliative therapies are prescribed to provide symptomatic relief from the pain and inflammation characteristic of RA. Disease-modifying therapies including conventional disease-modifying antirheumatic drugs (DMARDs) and biologic agents are prescribed to halt the destructive course of RA and prevent debilitating joint damage. Despite their success, the use of conventional DMARDs and biologics is limited by non-response in a subset of patients, safety concerns, and the high cost and discomfort of continuous injections with biologics. The 2012 approval of tofacitinib, an oral pan-Janus kinase (JAK) inhibitor, for treatment of RA cases that are inadequately responsive to methotrexate provides proof of concept for the therapeutic approach of kinase inhibition for the treatment of RA.⁶

Spleen tyrosine kinase (Syk) is a key enzyme involved in Fc and B cell receptor (BCR) signaling in multiple cell types involved in the immune response.⁷ Inhibition of Syk activity is expected to slow the destructive course of RA by affecting the initiation phase via modulation of signaling downstream of the BCR, the effector phase via blockade of the Fc receptors (Fc ϵ R, FC γ R) in various immune cells in the synovium, and the ensuing tissue damage via inhibition of syk or osteoclast maturation.⁸ Multiple preclinical studies have shown that genetic depletion of Syk or

inhibition of Syk with small molecule inhibitors (including published R788 (1), PRT-062607 (2), RO9021 (3), and GS-9973 (4)) attenuates inflammation, tissue damage and bone destruction in preclinical RA models (Figure 1).^{9,10,11,12} Human proof-of-concept has been achieved in RA clinical trials with the poorly selective oral Syk inhibitor 1 (52% of 100 kinases tested >100X Syk IC₅₀).¹³ Additional human proof-of-biology has been established with the B-cell depletion therapy Rituximab, approved for the treatment of RA.¹⁴ Dosing with 1 in humans was limited by hypertension and diarrhea as the most common adverse events. We hypothesized that the hypertension observed was driven by off-target activity¹⁵ and sought to develop a selective Syk inhibitor for the treatment of various autoimmune and hematological diseases, including RA, oncology,^{16,17} and asthma,^{18,19} devoid of this undesired pharmacology. This manuscript describes the preparation of diaminocarboxamides that are highly selective, potent Syk inhibitors as well as details regarding the attenuation of inherent hERG ion channel activity and mutagenicity risk within the chemotype to ultimately afford novel Syk inhibitors with favorable overall profiles. **Figure 1.** Diverse Syk Inhibitor Motifs from the Literature.



Chemistry

The main two points of structure activity relationship diversity for the aminodipyridylcarboxamides were in the basic amine region and the aminoheterocycle (solvent front) region. We developed modular approaches allowing late stage introduction of each of these functionalities (Scheme 1). Namely, the basic amine portion of the molecule could be introduced first starting from a dihalopyridyl nitrile under S_NAr or palladium-mediated

conditions. As an example of the former method, the synthesis of picolinamide **5** commenced via reaction of 3-bromo-5-fluoropyridine-2-carbonitrile (**6**) and *tert*-butyl [(1S,2R)-2-aminocyclohexyl]carbamate mediated by base to afford bromide **7** in 82% yield. Introduction of 2-amino-4,6-dimethylpyridine was accomplished under Buchwald–Hartwig conditions in high yield.²⁰ Alkaline peroxide promoted the hydrolysis of nitrile **8** to the corresponding amide, which was subsequently deprotected under acidic conditions and free-based with ammonia in methanol to provide carboxamide **5** in 84% yield over three steps.

Alternatively, picolinamide **9** was prepared using the latter method of early-stage aminoheterocycle introduction. Anionic addition of 4,6-dimethylpyridin-2-amine to 5-bromo-3-fluoropicolinonitrile (**10**) gave bromopyridine **11**. Palladium-mediated coupling of *tert*-butyl (1-(aminomethyl)cyclopropyl)carbamate gave diaminopyridine **12**, which was further functionalized via acidic deprotection and hydrolysis to afford carboxamide **9**.

Scheme 1. Synthesis of N-linked heterocycles.



(a) *tert*-butyl [(1*S*,2*R*)-2-aminocyclohexyl]carbamate, DIEA, 110 °C; (b) 2-amino-4,6-dimethylpyridine, $Pd_2(dba)_3$, Xantphos, Cs_2CO_3 , dioxane, 80 °C; (c) NaOH, 35% aq. H_2O_2 , DMSO, rt; (d) HCl, dioxane; (e) NH₃, MeOH, rt.

Scheme 2. Alternative approach to the synthesis of *N*-linked heterocycles.



(a) 2-amino-4,6-dimethylpyridine, KO-*t*-Bu, THF, 0 °C; (b) *tert*-butyl (1-(aminomethyl)cyclopropyl)carbamate, $Pd_2(dba)_3$, Xantphos, Cs_2CO_3 , dioxane, 80 °C; (c) TFA, CH₂Cl₂, rt (d) NaOH, 35% aq. H₂O₂, DMSO, rt.

For the later generation *C*-linked carboxamides, a modified synthetic approach was employed. As an exemplar, preparation of **13** commenced via S_NAr addition of *tert*-butyl ((1*S*,2*R*)-2-aminocyclohexyl)carbamate and 2,6-dichloropyrazine (**14**) to give chloropyrazine **15** in 88% yield. Iodination *para* to the newly installed amine was accomplished using *N*-iodosuccinimide to afford dihalopyrazine **16** in 91% yield. Carefully monitored cyanation under palladium-mediated conditions provided nitrile **17** in 83% yield, which was subsequently functionalized via a Suzuki–Miyaura coupling to install the indole moiety.²¹ Deprotection and hydrolysis of the nitrile **18** delivered carboxamide **13** in 48% yield over three steps.

Scheme 3. Synthesis of *C*-linked heterocycles.



(a) *tert*-butyl [(1*S*,2*R*)-2-aminocyclohexyl]carbamate, DIEA, 115 °C; (b) NIS, DMF, 65 °C; (c) $Zn(CN)_2$, Pd(PPh₃)₄, DMF, 120 °C; (d) 1*H*-indole-2-boronic acid pinacol ester, Pd(OAc)₂, PCy₃, K₂CO₃, THF, 65 °C; (e) TFA, CH₂Cl₂, rt; (f) KOH, 35% aq. H₂O₂, DMSO, rt.

Results and Discussion

Our primary medicinal chemistry goal was to identify a highly kinome selective molecule that demonstrated favorable properties for further development activities. Carboxamide scaffolds (e.g. PRT-062607) were known in the literature at the outset of our efforts as potent Syk inhibitors. Examining the available structural and crystallographic data for Syk revealed Asp-512

as a key residue to target for selectivity enhancement.²² Namely, an appropriately positioned array of hydrogen bond donors, such as the diamine present in YM-70220, had been demonstrated to recruit Asp-512 toward the ATP site, affording improved kinome selectivity as this motion of the catalytic Asp is unfavorable for a number of kinases (Figure 3).²³ Having settled upon Asp-512 as a key point of interaction, we next undertook a systematic core scan of diaminocarboxamides (Table 1). Our objective was to ensure a high ligand efficiency for Syk, while obviating the presence of a potentially reactive aniline substructure, which has been reported to increase risk for toxicity.²⁴ Generally, ligand efficiencies were in an acceptable range for all permutations tested (0.42–0.56); structure activity relationships revealed that aryl carboxamides possessing nitrogen *ortho* to the carboxamide were beneficial to potency (picolinamide 5 vs. benzamide 19), while a nitrogen at the *meta* position was deleterious to potency (benzamide 19 vs. nicotinamide 20).²⁵ We hypothesized that an intramolecular hydrogen bond between the amide NH and the heteroaryl nitrogen served to pre-organize the ground state conformation in a planar geometry stabilizing the bioactive conformation as well as lower the penalty for desolvation (23 versus 24, Figure 2). In contrast, an antagonistic steric interaction between the NH of the amide and the CH ortho to the amide is present in chemotypes lacking this heteroaryl nitrogen. Further, we proposed that the meta-nitrogen present in nicotinamide 20 and pyridazinecarboxamide 21 favors a ground state diamine conformation that requires a 180° rotation for recruitment of Asp-512 in the bioactive conformation (25 versus 26). Picolinamide 22 demonstrated an antagonistic interaction between the aniline solvent front and the picolinamide core resulting in a 100-fold intrinsic potency loss versus picolinamide 5 (27 versus 28). The potency advantage of the aminodipyridine was aligned with our strategy to avoid

anilines. With this data set, we chose to focus our efforts on the 3,5-diaminopicolinamides represented by compound **5**.

Table 1. Carboxamide Core Scan: Enzyme Potency and Ligand Binding Efficiency (LBE).

Analog	Structure	Syk IC ₅₀ (nM)	LBE
19		2.8	0.47
20		19	0.42
5		0.06	0.56
21		4.7 ^b	0.45 ^b
22		8.8	0.43

^aSee supporting information for details. ^bSee reference 25.

Figure 2. Rationale for Intrinsic Potency Differences Across Cores.



Figure 3. Interaction of Basic Amine with Asp-512 for Syk Potency and Selectivity. X-ray Cocrystal Structure of YM-70220 in Syk (Yellow) Overlayed with AMP-PMP Structure with Syk (Cyan).



Further profiling of representative 3,5-diaminopicolinamide **5** revealed a number of favorable aspects; high enzymatic (Syk IC₅₀ = 60 pM) and human whole blood potency (hWB CysLT IC₅₀ = 58 nM) had been achieved, along with exquisite kinome selectivity (100% of 265 kinases tested >100X Syk IC₅₀). Carboxamide **5** also demonstrated high enzymatic selectivity versus the homologous enzyme Zeta-chain-associated protein kinase 70 (Zap70) with an IC₅₀ of 5.3 nM (Zap70 IC₅₀/Syk IC₅₀ = 90X).²⁶ Further, the physicochemical properties (HPLC logD = 2.1, PSA = 118) and oral pharmacokinetic profile (Rat Plasma Cl_p (Cl_u) = 33 (710) mL/min/kg, Rat T_{1/2} = 2.7 h, F = 19%) of this molecule were in a desirable range.²⁷ Picolinamide **5** was profiled in the rat collagen induced arthritis model and demonstrated therapeutic efficacy at doses as low as 3 mg/kg QD PO (C_{max} = 0.2 μ M, C_{min} = 0.007 μ M, Figure 4). However, a key off-target activity of compound **5** was inhibition of the human Ether-a-go-go Related Gene (hERG, *I*_{Kr} IC₅₀ = 1.6 μ M). Several strategies have proven effective in the medicinal chemistry literature for attenuating delayed rectifying potassium channel activity including modulation of basicity and polarity.²⁸ We targeted the diamine and aminoheterocycle directed toward the solvent front for

modification in an attempt to merge the favorable properties of carboxamide **5** with weaker ion channel activity.

Figure 4. Rat Collagen Induced Arthritis Efficacy of 3,5-diamino picolinamide 5.



Table 2 details a systematic survey of the diamine portion of the molecules while holding the 2-amino-4,6-dimethyl pyridine constant. A number of calculated and measured properties were then leveraged to examine correlations between each category of properties and the corresponding hERG ion channel activities. A good correlation has been established in the literature, and was recapitulated with these chemotypes, between an automated patch clamp system, PatchXpress, and a higher throughput binding assay for measuring or estimating I_{Kr} IC₅₀ values.^{29,30} It is well-precedented that attenuation of amine basicity can result in weaker delayed rectifying potassium channel binding.³¹ To maintain good Syk potency, it was important to avoid rendering the diamine non-basic at physiological pH (e.g. **29** and **33**). Calculated and/or measured pKa's above 7.0 showed little to no correlation with ion channel activity as measured

via PatchXpress (Figure 5). There was also a lack of correlation between measured HPLC logD and I_{Kr} activity within a logD range of 0.7 and 2.3, though it did trend slightly in the expected direction with more polar molecules showing lower ion channel binding (Figure 5). The most robust correlation was found between polar surface area (PSA) and hERG ion channel activity, as a higher PSA biased molecules toward weaker activity, while maintaining Syk potency (**32**). The diamino-tetrahydrothiopyran dioxide identified from this survey and exemplified in picolinamide **32** emerged as a preferred moiety for interacting with Asp-512 while maintaining potency (Syk IC₅₀ = 0.1 nM), reasonable kinome selectivity (100% of kinases >100X, 265 kinases tested), and an improved off-target profile (hERG, I_{Kr} IC₅₀ = 29 µM).

 Table 2. Amine Region hERG Ion Channel SAR.



Analog	Amine	Syk IC ₅₀ (nM) ^a	Measured pKa	HPLC LogD	PSA ^b	<i>I</i> _{Kr} IC ₅₀ (μM)
5	NH ₂	0.06	9.0	2.1	118	1.6
29	F NH2 F NH	7	7.2 ^c	2.1	122	5.2
30		371	5.1	2.6	123	42 ^d
9	NH ₂ NH	0.5	7.9	1.9	124	1.7
31	NH2 NH	0.8	7.8	1.4	135	1.6
32		0.1	7.0	1.5	157	29



^aSee supporting information for details. ^bPolar surface area. ^cCalculated pKa. ^dMK-499 displacement IC₅₀ (μ M).



Figure 5. Physicochemical Properties versus hERG Ion Channel Activity.

Efforts were next directed toward examining the impact of the aminoheterocycle directed toward the solvent front on hERG ion channel activity. In this case, the cyclohexyldiamino group was held constant to ensure a sufficient dynamic range for assessing the impact of the heterocycle. Again, polarity emerged as the key driver, as both measured HPLC logD and calculated PSA demonstrated strong correlations with hERG interaction. Namely lower HPLC logD (<1.0) and higher PSA (>145) effectively ablated potent I_{Kr} activity in the absence of attenuated diamine basicity. Having demonstrated the impacts of PSA across two distinct portions of the molecules on ion hERG activity, planning of future molecules was guided by calculation of this physicochemical property. This prospective analysis allowed rapid triage and prioritization of novel inhibitor designs. In short order, picolinamide **42** was synthesized and demonstrated a favorable overall profile. Carboxamide **42** was a potent and selective Syk

inhibitor (Syk IC₅₀ = 0.6 nM, 99% of kinases >100X, 265 kinases tested, BLK IC₅₀ = 32 nM, ROS IC₅₀ = 43 nM) that demonstrated good cell and functional activity (hWB CysLT IC₅₀ = 56 nM) lacking the ion channel activity of previous prototypes (hERG, I_{Kr} IC₅₀ >30 μ M). Though **42** has a higher calculated PSA (PSA = 167) than many drug-like molecules (typically PSA ≤140 for oral bioavailability), a favorable pharmacokinetic profile in preclinical species was obtained, including oral exposure (Rat Plasma Cl_p (Cl_u) = 19 (270) mL/min/kg, Rat T_{1/2} = 2.8 h, 30 %F). The calculated PSA is believed to be an overestimate due to the extensive intramolecular hydrogen bonding network within the ground state of the molecule that serves to mask some of the inherent polarity from a topological assessment.

 Table 3. Solvent Front hERG Ion Channel SAR.



Analog	Solvent Group	Syk IC ₅₀ (nM) ^a	HPLC LogD	PSA ^b	I _{Kr} IC ₅₀ (μΜ)
5	N 	0.06	2.1	118	1.6
35		0.5	2.2	118	0.6
36	N v v v v v	0.8	1.7	130	1.7
37	N N Z	0.1	1.4	138	1
38	N to HO	0.9	1.3	138	5.9



^aSee supporting information for details. ^bPolar surface area.

Figure 6. Physicochemical Properties versus hERG Ion Channel Activity



Figure 7. Profile of Picolinamide 42.



While preparing carboxamide **42** to move toward development, a report appeared in the literature of mutagenicity in the bacterial reverse mutation (Ames) test for a molecule of a similar chemotype, GSK-143 (**43**).³² The report described a positive Ames test result for pyrimidinecarboxamide **43** in *Salmonella typhimurium* strain TA1537 without metabolic activation (Figure 8). Mutation in this strain often reflects intercalation of the test molecule into

DNA, and such activity can also be detected in Salmonella strains TA97 and TA97a which are considered equally acceptable with TA1537 under international test guidelines.³³ Since TA97a is the strain we use routinely, we selected this strain when carrying out a minimal screen, and performed the mutation testing in the presence of rat S-9 because the mutagenicity of 43 and some other candidates was enhanced by S-9 in our hands, and potential mutagenicity of aromatic amine metabolites (see below) would require metabolic activation. Our lead compound 42, induced mutation in strain TA97a in the presence of rat S9.. Similarly, picolinamide 5 was tested, and showed a positive result at $\geq 300 \,\mu\text{g/plate}$. Speculating that a metabolite could be the culprit in these results, we prepared and tested two potential degradative products. Diaminopyridine 45 had been found as a *N*-dealkylative metabolite in rat and human hepatocyte MetID studies. This molecule was predicted by DEREK for Windows (version 12.5.4, Lhasa Ltd., Leeds, UK) and Multi-CASE (MC4PC version 2.3.0.15, Multi-CASE Inc., Ohio, USA) to be positive in the Ames test due to the known mutagenic substructure of the primary aromatic amine in 45 (anilines 46 and 47 are known Ames positive compounds that feature this motif). However, the assay in strain TA97a was negative for 45 up to 3000 µg/plate with rat S9. Tricycle 44 had been observed to form upon standing with acid and could have been a prime suspect for intercalation. However, this compound was also negative in in TA97a with S9 up to 3000 μ g/plate. With these results in hand, we began to formulate hypotheses to overcome the putative intercalation liability within this series.

Figure 8. Initial Compounds for Mutagenicity Assessment.



Leveraging DNA co-crystals with acridine, a docking model was developed for our Syk inhibitors (represented by picolinamide **5**) in DNA (Figure 9).³⁴ This model revealed three distinct opportunities to disrupt DNA intercalation with our molecules. Namely, reducing the basicity of the amine functionality should disfavor the donation of hydrogen bonds to the DNA carbonyl. Second, growing toward, or shrinking from, the DNA backbone was predicted to disrupt the fit of our small molecules in DNA, leading to a less thermodynamically favorable interaction. Finally, introducing three-dimensionality could serve to disfavor the stabilizing pistack with DNA, mitigating the intercalation liability. We set out to test each of these hypotheses. Cognizant of the possibility that general cytotoxicity of molecules could confound comparisons across results from the mutagenicity assay, we also sought to develop an orthogonal higher throughput in vitro assay specific for DNA interaction.

We observed that the UV-Vis spectrum of some carboxamide compounds showed a saturable perturbation upon addition of duplex DNA, as reported for other organic ligands.³⁵ This facile assay allowed inhibitors to be tested prior to, or concurrent with, the TA97a mutagenicity assay, although the metabolic activation system (S9) could not be incorporated into the DNA interaction assay. To validate this approach, picolinamides **5** and **42** were profiled, and each showed positive data for DNA interaction, consistent with the mutagenicity results.³⁶ Further, metabolite **45** was negative in the UV-Vis assay, again matching the mutagenicity results for this

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compound.³⁷ Of note, tricycle **44**, which was negative in the TA97a mutagenicity, was positive for DNA interaction in the UV-Vis assay. Thus the UV-Vis assay can give false positive results compared with the bacterial mutagenicity assay in TA97a (though the results may be more robust, since cytotoxicity is not a confounder in this model), but we never observed a false negative in this assay,³⁸ supporting its use as an orthogonal assessment of intercalation risk and serving as a stage gate for the lower throughput bacterial mutagenicity assay.

Figure 9. Docking Model of Picolinamide 5 in DNA (from Acridine/DNA Co-Crystals).



As shown in table 4, our first hypothesis regarding the attenuation of basicity still afforded mutagenic/UV-Vis positive compounds (e.g. compound **48**). Ablating the basicity completely, as in carboxamide **49**, afforded compounds predicted to lack intercalation, but at the expense of Syk activity (Syk $IC_{50} = 60$ nM). As previously detailed the basic amine interaction of our carboxamide inhibitors with Asp-512 is a critical element for potency and selectivity on the enzyme.

We next systematically surveyed substitution at the 4-, 5-, and 6-positions of the aminopyridine projected toward solvent. Examples **50**, **51**, and **52** were selected to evaluate the effectiveness of growing toward the DNA backbone to disrupt the fit of our inhibitors in DNA.

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4-Substituted aliphatic alcohol **50** was negative in the TA97a mutagenicity assay, but positive in the UV-Vis assay. Of note, the compound could only be tested up to 1000 µg/plate in the bacterial assay, as higher concentration testing was precluded by general cytotoxicity. Pyrazole **51**, featuring a large substituent at the 5-position of the aminopyridine, which should be pointed directly at the DNA backbone, was negative in both the TA97a mutagenicity assay and UV-Vis assays while maintaining Syk potency (Syk IC₅₀ = 1 nM), representing a potential advance in our efforts. Growing from the 6-position of the aminopyridine, as in triazole **52**, gave very strong mutagenicity and UV-Vis positive results. Though we had gained some favorable data with pyrazole **51**, its physicochemical properties led to a particularly large cell shift (hWB CysLT IC₅₀/Syk IC₅₀ = 2800X for **51** versus 98X for **42**). Overall, the strategy of growing the molecule to mitigate mutagenicity did not seem the best path from an efficiency or physicochemical properties perspective, so we chose to prioritize alternative tactics.

Disruption of the pi stacking interaction with DNA was pursued via incorporation of sp³ groups on the aminopyridine heterocycle (e.g. picolinamide **53**). Again, this strategy was not successful at overcoming intercalation in our hands, as carboxamide **53** was positive in both the mutagenicity and UV-Vis studies. Having pursued a number of paths with limited success, we decided to look at alternative substructures to mitigate the intercalation liability.

Further pursuing a change in the shape of the carboxamide chemotype to grow toward the DNA backbone and disfavor intercalation, we envisioned a carbon-link of the solvent group to the core carboxamide.³⁹ This design was applied to the synthesis of benzimidazole **54**, which demonstrated proof of concept for generation of a modestly potent/selective carbon-linked Syk inhibitor (Syk IC₅₀ = 7 nM, 97% of kinases >100X, 101 kinases tested, LRRK2 IC₅₀ = 130 nM, CHK2 IC₅₀ = 160 nM, TSSK3 IC₅₀ = 420 nM) that filled a different chemical space than the

previous iteration of nitrogen-linked analogs (Figure 10).⁴⁰ We also pivoted this motif to pyrazinecarboxamide 55, which delivered a highly favorable profile (Syk $IC_{50} = 0.5 \text{ nM}$, 99% of kinases >100X, 101 kinases tested, LRRK2 IC₅₀ = 30 nM, hWB CysLT IC₅₀ = 130 nM, Rat Plasma $Cl_p(Cl_u) = 44$ (255) mL/min/kg, Rat $T_{1/2} = 2.8$ h). Further, carboxamides 54 and 55 were negative in the UV-Vis assay for intercalation, and pyrazine 55 was also negative in the mutagenicity assay in strain TA97a with rat S9. Cyclohexyldiamine 55 was cytotoxic at concentrations greater than 850 µg/plate, so to mitigate cytotoxicity, the sulfonyldiamine analog 56, was prepared. Pyrazine 56 was also negative in the UV-Vis assay, as well as the TA97a mutagenicity assay with rat S9, this time tested up to 5000 μ g/plate. Delighted by these results, the team set out to further interrogate these findings and compare them in a third assay format. A DNA unwinding assay was utilized to study carboxamides 42, 55, and 56; the results were in full agreement with the TA97a mutagenicity and UV-Vis assays, further supporting the viability of the carbon-linked motif as a method for obviating intercalation. Application of the design principles that had been used to overcome ion channel activity and mutagenicity afforded tetrahydropyranyldiamine 57 (Figure 11). Pyrazine 57 achieved the desired profile we had targeted from the outset with good intrinsic potency (Syk $IC_{50} = 0.7$ nM), high selectivity (99%) of kinases >100X, 101 kinases tested, LRRK2 IC₅₀ = 33 nM), potent cell functional activity (hWB CysLT IC₅₀ = 62 nM) and a favorable pharmacokinetic profile in preclinical species (Rat Plasma Cl_p (Cl_u) = 25 (100) mL/min/kg, Rat $T_{1/2}$ = 3.5 h, 29 %F). Further, this molecule 57 posed low risk for ion channel activity and DNA interaction as assessed by our ion channel and UV-Vis displacement assays (hERG IC₅₀ = 22 μ M and UV-Vis = Negative).⁴¹

A recent report from the literature correlated Ames mutagenicity to a metric defined as "ovality" for a series of pyridazinecarboxamide Syk inhibitors.²⁵ Ovality was defined as the ratio

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of the molecular surface area to the minimum surface area.⁴² We attempted to retrospectively apply a similar calculation to test its predictivity for our studies. We found this metric to be highly sensitive to subtle changes in the geometry of the solvent front heteroaryl substituent. Across our data set, we observed no correlation between ovality and bacterial mutagenicity. As detailed, the key to circumvention of the mutagenicity issue in our case was a mechanistic understanding of the source of the liability (intercalation) along with testable design hypotheses to alter the shape of the carboxamide Syk inhibitors. We pursued both increasing the spherical nature of our molecules as well as two-dimensional perturbation of the spatial fit, ultimately affording the *C*-linked carboxamides that have shown no mutagenicity risk to date.

Table 4. Mutagenicity SAR.

Analog	Structure	Syk IC ₅₀ (nM) ^a	Mutagenicity (µg/plate) ^b	UV-Vis ^c	DNA Unwinding ^d	Ovality ^e
48		51	Positive (1000)	Positive	ND	1.42
49		63	Negative (3000)	Negative	ND	1.42
42	O=S	0.6	Positive (1000)	Positive	Positive	1.39
50	$\begin{array}{c} & & \\$	0.4	Negative (1000)	Positive	ND	1.42

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^a ^aSee supporting information for details. ^bNumber represents highest scorable concentration for bacterial mutagenicity negative compounds and lowest concentration a positive score was achieved for positive compounds. ^cUV-Vis positive defined as $K_d \leq 500 \mu M$. ^dSee supporting information for details and primary data. ^eOvality calculated using a method adapted from the report by Lucas, M. C. et al.²⁵

Figure 10. Overlay of *N*-linked Carboxamide (42, Red) with *C*-linked Carboxamide (56, Blue).



Figure 11. Profile of Pyrazine 57.



Syk IC₅₀: 0.7 nM hWB CysLT IC₅₀: 62 nM Kinases >100X: 100% (100) Rat Plasma Cl_p (Cl_u): 25 (100) mL/min/kg Rat T_{1/2}: 3.5 h, 29% F hERG IC₅₀: 22 μM UV-Vis Displacement: Negative

Conclusions

A novel series of orally efficacious, selective carboxamide Syk inhibitors were discovered via systematic survey of chemotypes prioritized by ligand efficiencies and structural alerts. The critical basic amine functionality of the 3,5-diamino picolinamides afforded undesired hERG ion channel activity, which was mitigated via modulation of physicochemical properties. PSA emerged as the most effective calculated property in circumventing hERG activity and was used to prospectively design Syk inhibitors devoid of this liability. A mutagenicity risk was subsequently discovered within these chemotypes. The mechanism for mutagenesis was identified as involving intercalation. Several structure-based design hypotheses were interrogated to obviate this issue; modulation of the two-dimensional shape of the molecules led to the identification of an alternative series of *C*-linked heterocyclic carboxamides, delivering drug-like molecules devoid of hERG ion channel activity, mutagenicity, and intercalation risk, while maintaining Syk potency and selectivity.

Supporting Information.

Experimental procedures for synthesis of final compounds and intermediates, and descriptions of assays performed. This material is available free of charge via the Internet at http://pubs.acs.org.

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ABBREVIATIONS

Syk, Spleen tyrosine kinase; RA, rheumatoid arthritis; DMARD, disease-modifying antirheumatic drug; Fc, fragment, crystallizable; BCR, B cell receptor; Syk IC₅₀, recombinant human Syk IC₅₀; LBE, ligand binding efficiency; hWB, human whole blood; MetID, metabolite

 identification; S_NAr, nucleophilic aromatic substitution; MeOH, methanol; DIEA, *N*,*N*-diisopropylethylamine; NIS, *N*-iodosuccinimide.

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 37 All compounds tested to a top concentration of 406 μM DNA. See supporting information for further details.

³⁸ 10 of 30 compounds tested were negative in the UV-Vis and bacterial mutagenicity assays.

³⁹ A detailed account of the design principles applied to this scaffold re-design will be the subject of a future communication.

⁴⁰ X-ray crystallograppic data for the *C*-linked carboxamides will be detailed in a future communication.

⁴¹ We note that pyrazine **57** has not been tested in TA97a or the other strains used in the Ames bacterial mutagenicity assay.

⁴² The minimum surface area is the surface area of a sphere having a volume equal to the solvent-excluded volume of the molecule (computed from the Connolly molecular surface area and solvent-excluded volume properties).

