[1+1]-CONDENSATION OF 12-OXO-DERIVATIVES OF RICINOLEIC ACID ESTERS WITH HYDRAZINE HYDRATE ON THE ROUTE TO MACROCYCLES

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The reactivities of the keto analogs of methyl ricinolate and its triglyceride with hydrazine hydrate were studied. The ester was found to be inert, which made it possible to synthesize a macrocyclic azine with a side chain hydrazone moiety from a 12-oxo-derivative.

Keywords: methyl ricinoleate, castor oil, *O*,*N*-containing macroheterocycles, esters, hydrazone and azine moieties, synthesis.

Enantiomerically pure 34-membered macrolide **4** with a diacylhydrazine moiety was prepared quickly and efficiently by us earlier [1] from methyl ricinoleate (**2**), which is available via transesterification of castor oil (**1**) isolated from *Ricinus communis* seeds [2]. It consisted of sequential [2 + 1]-condensation of **2** with adipylchloride and [1 + 1]-condensation of the obtained *bis*[(9*Z*,12*R*)-1-methoxy-1-oxooctadec-9-en-12-yl]-1,4-hexanedicarboxylate (**3**) with hydrazine hydrate.



a. MeOH, H⁺; b. ClOC(CH₂)₄COCl, Py; c. N₂H₄·H₂O, 1,4-dioxane

In continuation of these studies, possible incorporation of 12-oxo-derivatives **5** and **6** of **2** and its triglyceride (castor oil, **1**) in the [1 + 1]-condensation with hydrazine hydrate was studied.

Corey oxidation of **2** to keto analog **5** followed by reaction with hydrazine hydrate in dioxane that occurred only at the ketones formed azine **7** and not macrocycle **8** with azine and diacylhydrazine moieties even if a large excess of hydrazine hydrate was used.

In a similar manner, the keto analog of castor oil 6 reacted to give macrocycle 9 and/or 10 in which one of the carbonyl groups was converted to a hydrazone whereas the other two were linked by an azine group. Isomers 9 and 10, which differed in the size of the rings, could not be differentiated using NMR spectroscopy and mass spectrometry.

Disappearance in the IR spectra of bands at 1710 and 1715 cm⁻¹ that were characteristic of the keto groups of starting **5** and **6** proved that hydrazine hydrate reacted with them. The appearance of bands at 1668 and 1663 cm⁻¹ (C=N) proved that azine groups formed in **7** and **9** and/or **10**.

Retention in ¹³C NMR spectra of ester resonances [173.95 ppm in 7 and 172.82 and 173.34 in 9 and/or 10] and in PMR spectra of the resonance at 3.65 ppm corresponding to the OCH₃ group of 7 and two multiplets at 4.20–4.28 and 5.26–5.31 for CH₂ and CH groups of the glyceride moiety in 9 and/or 10 proved that the ester did not react with hydrazine hydrate. This indicated that the carboxylates were less reactive than the carbonyls.

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0009-3130/17/5302-0231 [©]2017 Springer Science+Business Media New York



a. PCC, CH₂Cl₂; b. NH₂NH₂·H₂O, 1,4-dioxane

 13 C NMR spectra were missing resonances for the carbonyls in **5** and **6** [208.81 and 209.01 ppm] and showed resonances at 154.00 that indicated azine groups had formed. Furthermore, PMR spectra of **9** and/or **10** contained a broad singlet at 5.56–5.61 ppm for an =N–NH₂ group.

All these results indicated that azine diester 7 and macrocyclic azine hydrazones 9 and/or 10 had formed. This was confirmed in mass spectra by the detection of the corresponding molecular masses, i.e., 616 $[M]^+$, 639 $[M + Na]^+$, and 655 $[M + K]^+$ for 7 and 936 $[M]^+$, 959 $[M + Na]^+$, and 975 $[M + K]^+$ for 9 and/or 10.

Thus, the study of reactions of hydrazine hydrate with keto-derivatives of methyl ricinoleate and its triglyceride found that the esters were inert. This enabled a macrocyclic azine containing a side-chain hydrazone to be synthesized from the 12-oxo-derivative of castor oil.

EXPERIMENTAL

General comments were published before [3].

(Z)-Methyl 12-oxooctadec-9-enoate (5) was prepared from castor oil (1) by the literature method [4]. Yield 1.20 g (94%). IR spectrum (v, cm⁻¹): 1710 (C=O), 1740 (CO₂), 3010 (CH=). ¹H NMR spectrum (300 MHz, CDCl₃, δ , ppm, J/Hz): 0.89 (3H, t, J = 6.7, H-18), 1.21–1.38 (16H, m, H-4+H-7, H-14+H-17), 1.43–1.48 (2H, m, H-13), 1.57–1.61 (2H, m, H-3), 2.30 (2H, t, J = 7.0, H-2), 2.38–2.43 (2H, m, H-8), 3.13 (2H, d, J = 6.0, H-11), 3.65 (3H, s, OCH₃), 5.38–5.58 (2H, m, H-9, 10). ¹³C NMR spectrum (75 MHz, CDCl₃, δ , ppm): 13.96 (q, C-18), 22.54, 24.82, 25.63, 28.99, 29.0, 29.31, 29.48, 31.78 (all t, C-3+7, C-14+C-17), 27.27 (t, C-8), 33.93 (t, C-13), 35.28 (t, C-2), 36.76 (t, C-11), 51.25 (q, OCH₃), 120.89 (d, C-10), 133.19 (d, C-9), 174.09 (s, C-1), 208.81 (s, C-12). Mass spectrum (ESI): CH₃CN–H₂O, 95:5, (Scan+) *m/z*: 310 (M⁺), 279 (M – OCH₃)⁺.

Glyceryl tri-(12-oxo-9Z-octadecanoate) (6) was prepared from castor oil (1) by the literature method [5]. Yield 3.65 g (74%). IR spectrum (v, cm⁻¹): 1715 (C=O), 1740 (CO₂), 3010 (CH=). ¹H NMR spectrum (300 MHz, CDCl₃, δ , ppm, J/Hz): 0.84 (9H, t, J = 6.7, 3(H-18), 1.20–1.50 (48H, m, 3(H-4+H-7, H-14+H-17)), 1.50–1.90 (6H, m, 3(H-3)), 1.90–2.00 (6H, m, 3(H-8)), 2.30 (6H, t, J = 6.5, 3(H-2)), 2.44 (6H, t, J = 7.0, 3(H-13)), 3.16 (6H, d, J = 6.0, 3(H-11)), 4.20–4.28 (4H, m, H-1', 3'), 5.26–5.31 (1H, m, H-2'), 5.56–5.61 (6H, m, 3(H-9), 3(H-10)). ¹³C NMR spectrum (75 MHz, CDCl₃, δ , ppm): 13.98 (q, 3(C-18)), 27.52 (t, 3(C-8)), 22.50, 23.84, 24.87, 28.93, 29.09, 29.31, 31.64 (all t, 3(C-3+7, C-14+C-17)), 34.02 (CH₂CO₂C-1', CH₂CO₂C-3'), 34.18 (t, CH₂CO₂C-2'), 41.71 (t, 3(C-11)), 42.36 (t, 3(C-13)), 62.14 (t, C-1', 3'), 69.02 (d, C-2'), 121.13 (d, C-10), 133.49 (d, C-9), 172.76 and 173.14 (d, CO₂C-2'; t, CO₂C-1', CO₂C-3'), 209.01 (s, C-12). Mass spectrum (ESI, *m/z*): CH₃CN–H₂O, 95:5, (Scan+): 926 [M]⁺, 949 [M + Na]⁺.

General Method for Preparing Azines. Diketone 5 or 6 (1.0 mmol) in distilled dioxane (8.5 mL) was stirred vigorously, treated slowly dropwise with hydrazine hydrate (0.1 mL, 0.1 g, 2.0 mmol, 98%), and stirred for 24 h (TLC monitoring). The dioxane was evaporated at reduced pressure. The residue was dissolved in CH_2Cl_2 (10 mL), washed with H_2O (3 × 3 mL), dried over MgSO₄ and evaporated. The resulting residue was rinsed by decantation with hexane (100 mL) and dried *in vacuo*.

Dimethyl 12,15-Dihexyl-13,14-diazahexacosa-9,12,14,17-tetraenoate (7). Yield 0.45 g (22%). IR spectrum (v, cm⁻¹): 1738 (C=O), 1668 (C=N). ¹H NMR spectrum (δ , ppm, J/Hz): 0.87 (3H, t, J = 6.7, H-18), 1.18–1.30 (16H, m, H-4+H-7, H-14+H-17), 1.55–1.61 (2H, m, H-3), 2.29 (2H, t, J = 7, H-2), 2.00–2.15 (2H, m, H-8), 2.90 (2H, t, J = 7.0, H-13), 3.05 (2H, d, J = 6.0, H-11), 3.69 (3H, s, OCH₃), 5.22–5.61 (2H, m, H-9, 10). ¹³C NMR spectrum (δ , ppm): 13.77 (q, C-18), 22.29, 26.02, 26.31, 27.01, 28.83, 29.19, 31.27, 31.42, 35.99 (all t, C-3+7, C-14+C-17), 27.33 (t, C-8), 29.39 (t, C-11), 33.75 (t, C-2), 37.21 (t, C-13), 51.11 (q, OCH₃), 121.97 (d, C-10), 131.90 (d, C-9), 154.00 (s, C-12), 173.95 (s, C-1). Mass spectrum (ESI, *m/z*): CH₃CN–H₂O 95:5, (Scan+): 616 [M]⁺, 639 [M + Na]⁺, 655 [M + K]⁺.

13,16-Dihexyl-1,28-dioxa-14,15-diaza-29-(2'-oxa-14'-hydrazono-11'Z-eicosen-3'-on-1'-yl)-cyclotriaconta-10Z,13,15,18Z-tetraen-2,27-dione (9) and/or **13,16-dihexyl-1,28-dioxa-14,15-diaza-30-(1'-oxa-13'-hydrazono-10'Z-eicosen-2'-on-1'-yl)-cyclohentriaconta-10Z,13,15,18Z-tetraen-2,27-dione (10).** Yield 0.19 g (37%). IR spectrum (v, cm⁻¹): 1739 (C=O), 1663 (C=N). ¹H NMR spectrum (δ , ppm, J/Hz): 0.81 (9H, t, J = 6.7, H-20', 2(H-6'')), 1.10–1.23 (48H, m, H-5+H-8, H-21+H-24, H-6'+H-9', H-16'+H-19', 2(H-2''+H-5'')), 1.48–1.51 (6H, m, H-4, 25, 5'), 1.95–2.06 (6H, m, 3(H-8)), 2.23 (6H, t, J = 7, H-9, 20, 10'), 2.35 (6H, t, J = 7.0, H-15, 2(H-1'')), 3.05 (6H, d, J = 6.0, H-12, 18, 13'), 4.15–4.25 (4H, m, H-30, 1'), 5.25–5.29 (1H, m, H-29), 5.51–5.57 (6H, m, H-10, 11, 18, 19, 11', 12'), 5.58–6.01 (2H, br.s, NH₂). ¹³C NMR spectrum (δ , ppm): 14.03 (q, C-20', 2(C-6'')), 27.52 (t, C-9, 20, 10'), 22.46, 23.84, 24.84, 28.02, 28.81, 29.09, 29.31, 31.52, 31.64 (all t, C-4+C-8, C-21+C-25, C-5'+C-9', C-16'+C-19', 2(C-2''+C-5''), 34.00 (C-3, 4'), 34.07 (t, C-26), 41.71 (t, C-12, 18, 13'), 42.36 (t, C-15', 2(C-1'')), 63.69 and 65.08 (t, C-30, 1'; d, C-29), 120.13 (d, C-11, 18, 12'), 131.49 (d, C-10, 19, 11'), 153.03 (s, C-14'), 154.00 (s, C-13, 16), 172.76 (s, C-27), 173.14 (s, C-2, 3'). Mass spectrum (ESI, *m/z*): CH₃CN–H₂O 95:5, (Scan+): 936 [M]⁺, 959 [M + Na]⁺, 975 [M + K]⁺.

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