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# Total Synthesis of ( $\pm$ )-Armepavines and ( $\pm$ )-Nuciferines From (2Nitroethenyl)benzene Derivatives 

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# TOTAL SYNTHESIS OF ( $\pm$ )-ARMEPAVINES AND ( $\pm$ )-NUCIFERINES FROM (2-NITROETHENYL)BENZENE DERIVATIVES 

Chia-Fu Chang, ${ }^{1}$ Chu-Yun Huang, ${ }^{1}$ Yu-Chao Huang, ${ }^{1}$ Kuan-Yu Lin, ${ }^{1}$ Yean-Jang Lee, ${ }^{1}$ and Chau-Jong Wang ${ }^{2}$<br>${ }^{1}$ Department of Chemistry, National Changhua University of Education, Changhua, Taiwan<br>${ }^{2}$ Institute of Biochemistry, Chung Shan Medical University, Taichung, Taiwan

A concise route to armepavine 1 and nuciferine 2 and 3, which can be isolated from the leaves of Nelumbo nucifera (Nymphaceae), has been achieved in which the longest linear sequence is only six steps from commercially available benzaldehyde in $28 \%, 21 \%$, and $20 \%$ overall yield, respectively. The key transformations in the synthesis are the radical cyclization of aryl bromide with $\mathrm{Bu}_{3} \mathrm{SnH}$ and the Pictet-Spengler reaction of N -substituted amine with aldehyde.

Keywords: Armepavine; Nef; nuciferines; Pictet-Spengler

## INTRODUCTION

As part of our research aiming to uncover new natural products with improved biological activities, including antioxidant, anti-human immunodeficiency virus (HIV), and tumor growth inhibition activities, we attended to the flavonoid, coumestan, and maleimide family. ${ }^{[1]}$ Recently, several structurally interesting alkaloids with substituted isoquinoline ${ }^{[2]}$ have been isolated from the leaves of Nelumbo nucifera (Nymphaceae) and show significant anti-HIV activity ( $\mathrm{EC}_{50}$ value of $0.8 \mu \mathrm{~g} / \mathrm{mL}$, $\mathrm{TI}>125)^{[3]}$ and also inhibit platelet aggregation ${ }^{[3]}$ (Fig. 1). Although armepavine (1) and nuciferine (2) have been synthesized by several groups, ${ }^{[2 c, 4]}$ they were achieved with very time-consuming and complicated synthetic approaches. To overcome these technical difficulties, we report our studies on the synthesis of nuciferine analogs from the readily available (2-nitroethenyl)benzene derivatives by sequential Pictet-Spengler and radical cyclization.

Retrosynthetic analysis (Scheme 1) suggested that $\mathbf{2}$ and $\mathbf{3}$ can be secured by Heck coupling or radical cyclization ${ }^{[4]]}$ of 4. The latter should be accessible from 5 and 6 with Pictet-Spengler ${ }^{[5]}$ or Bischler-Napieralski ${ }^{[6]}$ reactions. In addition, commercially

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1 R=Me Armepavine


2 R=Me Nuciferine $3 \mathrm{R}=\mathrm{H} \mathrm{N}$-Nornuciferine

Figure 1. Armepavine and nuciferines from the leaves of Nelumbo nucifera (Nymphaceae).


Scheme 1. Retrosynthesis of ( $\pm$ )-nuciferines.
available ketones and aldehydes would allow facile introduction of a wide range of different substituents of $\mathbf{5}$ and $\mathbf{6}$ by $\mathrm{Nef}^{[7]}$ or Willgerodt-Kindler ${ }^{[8]}$ reactions.

## RESULTS AND DISCUSSION

With the aim of developing a concise route to synthesize aryl acetaldehyde and amine with diverse substituents, the commercially available $7 \mathbf{a}-\mathbf{c}$ were used to prepare the corresponding aldehydes 10a-c and amines 11a-c from this compound (Scheme 2). In the beginning of our synthesis, aldehydes $7 \mathbf{a}-\mathbf{c}$ were transformed by aldol condensation with $\mathrm{CH}_{3} \mathrm{NO}_{2} / \mathrm{NH}_{4} \mathrm{OAc}$ under acetic acid to obtain the corresponding nitro compounds $\mathbf{8 a - c}$ in excellent yield. Subsequently, selective 1,4-reduction of conjugated $\mathbf{8 a - c}$ proceeded with $\mathrm{NaBH}_{4}$ to provide the saturated nitro product $\mathbf{9 a - c}$. According to the Nef-type reaction, the nitro group of $\mathbf{9 a - c}$ was readily changed to the carbonyl group under acidic conditions to give the desired 10a and 10b in quantitative yield, respectively, as well as 10c in $96 \%$ yield. At the same time, compounds 11a-c were prepared by direct reduction of $\mathbf{8 a - c}$ with $\mathrm{LiAlH}_{4}$ in tetrahydrofuran (THF) at $0^{\circ} \mathrm{C}$. With 10a and $\mathbf{1 1 b}$ in hand, we tried to synthesize tetrahydroisoquinoline $\mathbf{1 3}$ through the coupling of $\mathbf{1 0 a}$ and $\mathbf{1 1 b}$ using the PictetSpengler reaction. However, the acid-catalyzed reaction using different Lewis acids, such as trifluoroacetic acid (TFA), $\mathrm{HCl}, p$-toluenesulfonic acid ( $p$-TSA), and $\mathrm{BF}_{3}-$ $\mathrm{OEt}_{2}$, to give $\mathbf{1 3}$ was unsuccessful. The major product was the recovered starting material or amine hydrochloride salt. This problem could be due to the need for anhydrous conditions, because the iminium salt is readily hydrolyzed under


Scheme 2. Synthesis of aryl amines and aryl aldehydes.
acid-catalyzed reactions. Therefore, we sought to overcome this problem by converting the primary amine into the secondary amine. Amine 11b was protected with the electron-withdrawing groups $\mathrm{CO}_{2} \mathrm{Me}$ and Boc to provide 12a, b, respectively (Scheme 2). Subsequently, the $N$-methoxycarbonyl hydroisoquinoline obtained under $\mathrm{BF}_{3}-\mathrm{OEt}_{2}$ catalysis by condensation-cyclization of $N$-methoxycarbonyl amine 12a with aldehyde 10a may be used without further purification, and then the process proceeded smoothly using the reductive reaction with $\mathrm{LiAlH}_{4}$ to provide the desired product $\mathbf{1 3}(67 \%)$ in three steps (Scheme 3). Finally, the isopropyl group in $\mathbf{1 3}$ was removed with $\mathrm{BCl}_{3}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0^{\circ} \mathrm{C}$ to afford armepavine 1 in $69 \%$ yield. The structure of 1 was confirmed by x-ray crystallography. ${ }^{[9]}$ With the success of achieving 12a, b, we employed the same Lewis acid $\mathrm{BF}_{3}-\mathrm{OEt}_{2}$-catalyzed Pictet-Spengler reaction in nuciferines $\mathbf{2}$ and $\mathbf{3}$ syntheses. Coupling of 12a, $\mathbf{b}$ respectively with aldehyde 10c proceeded readily in mixed solvent $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CHCl}_{3} 1: 1\right)$ under $\mathrm{BF}_{3}-\mathrm{OEt}_{2}$ catalysis to give the bromoarmepavine $\mathbf{1 4 a}, \mathbf{b}$, which can then undergo the radical cyclization with azobisisobutyronitrile (AIBN) $/ \mathrm{Bu}_{3} \mathrm{SnH}$ to accomplish 15a, b in $52 \%$ and $56 \%$ yield, respectively. Trying to obtain 15a, b via palladium-catalyzed coupling ${ }^{[2 c, 4 \mathrm{~g}]}$ proved unsuccessful. With precursor 15a, $\mathbf{b}$ in hand, completion of the final steps in the nuciferine syntheses was straightforward, requiring either reduction or deprotection. Previous reduction of the methoxycarbonyl group in compound 15a with $\mathrm{LiAlH}_{4}$ provided the $N$-methyl nuciferine 2 in $66 \%$ yield. Also, direct acid-catalyzed deprotection of the $t$-butoxycarbonyl group in compound $\mathbf{1 5 b}$ with $\mathrm{AcCl} / \mathrm{MeOH}$ furnished the expected product $\mathbf{3}$ in $59 \%$ yield. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of the synthetic products are in agreement with those of reported natural product derivatives.

It is worth noting that an alternative construction ${ }^{[4 \mathrm{~g}]}$ of armepavine $\mathbf{1}$ was created by Willgerodt-Kindler and Bischler-Napieralski reactions (Scheme 4). Using the Willgerodt-Kindler and hydrolysis procedures, the phenylacetic acid $\mathbf{1 7}$ can be readily prepared from the commercially available hydroxyacetophenone in excellent




Scheme 3. Synthesis of $(+/-)$-armepavines and $(+/-)$-nuciferines.
yield (three steps, $67 \%$ ). First, the protection of hydroxyacetophenone was employed with $i-\mathrm{PrBr}$ and $\mathrm{K}_{2} \mathrm{CO}_{3}$ to obtain $\mathbf{1 6}$ in $95 \%$ yield, and then the acetophenone $\mathbf{1 6}$ was readily transformed by sulfur and morpholine with catalyst $p$-TSA under refluxing conditions to afford the expected thioacetomorpholide, which could directly be used by hydrolysis with $20 \% \mathrm{NaOH}$ and catalyst tetrabutylammonium bromide (TBAB) to provide the desired acid 17 . The acetic acid $\mathbf{1 7}$ was treated with oxalyl chloride, followed by reaction with the readily available phenylethamine 11b to obtain the


Scheme 4. Synthesis of isoquinoline 13.
desired amide $\mathbf{1 8}$ in $82 \%$ yield. With the required amide $\mathbf{1 8}$ in hand, the BischlerNapieralski cyclization with $\mathrm{POCl}_{3}$ in acetonitrile gave the isoquinimium ion, which can undergo the reduction with $\mathrm{NaBH}_{4}$ to afford isoquinoline 19 in $60 \%$ yield (two steps). Sequential condensation and reduction of 19 with formaldehyde and $\mathrm{NaBH}_{4}$, respectively, provided the isopropyl armepavine $\mathbf{1 3}$ in $83 \%$ yield.

## CONCLUSION

In summary, a concise route to armepavine $\mathbf{1}$ and nuciferine $\mathbf{2}$ and $\mathbf{3}$ has been achieved in which the longest linear sequence is only six steps from commercially available benzaldehyde in $28 \%$, $21 \%$, and $20 \%$ overall yield, respectively. Alternatively, $\mathbf{1}$ was also accomplished in $19 \%$ overall yield from commercially available 4-hydroxyacetophenone by a route in which the longest linear sequence is only eight steps. Thus, it is demonstrated that using the Lewis acid $\mathrm{BF}_{3}-\mathrm{OEt}_{2}$-catalyzed PictetSpengler reaction gives a good yield and is easily modified to give access to a variety of different armepavine and nuciferine analogs. The preparation of these compounds is currently under way, and their biological activities will be investigated to evaluate the efficacy of these compounds as antitumor agents.

## EXPERIMENTAL

Melting points were determined on a Mel Temp II melting-point apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were measured with a Bruker 300 spectrometer. Chemical shifts are reported in parts per million ( $\delta, \mathrm{ppm}$ ) using $\mathrm{CHCl}_{3}$ ( $\delta_{\mathrm{H}} 7.26$ ) as an internal standard. Low-resolution mass spectra (MS) and high-resolution mass spectra (HRMS) were determined on a Jeol JMS-HX 110 mass spectrometer from National Chung-Tsing University, Taichung. Elemental analyses were performed on a Heracus CHN-OS Rapid spectrometer in the Taichung Instrumentation Center, National Science Council, Taiwan. Solvents were freshly distilled prior to use from phosphorus pentoxide or $\mathrm{CaH}_{2}$. THF was distilled from sodium diphenyl ketyl. All reactions were carried out under a nitrogen atmosphere unless otherwise stated. Silica gel (silica gel 60, 230-400 mesh, Merck) was used for chromatography. Organic extracts were dried over anhydrous $\mathrm{MgSO}_{4}$.

## 1-Isopropoxy-4-(2-nitrovinyl)benzene (8a) ${ }^{[10]}$

A solution of 4-isopropoxy benzaldehyde (7a) $(1.51 \mathrm{~g}, 9.2 \mathrm{mmol})$ in $\mathrm{NH}_{4} \mathrm{OAC}$ $(1.40 \mathrm{~g}) / \mathrm{AcOH}(20.0 \mathrm{~mL})$ was stirred for 10 min at $25^{\circ} \mathrm{C}$, followed by slow addition of $\mathrm{CH}_{3} \mathrm{NO}_{2}(2.0 \mathrm{~mL}, 36.9 \mathrm{mmol})$ with stirring for 5 min . The resulting mixture was heated to reflux at $120^{\circ} \mathrm{C}$ for 4 h , then quenched by addition of ice water $(10.0 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane- $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} 2: 1\right)$ to obtain the desired $\mathbf{8 a}(1.62 \mathrm{~g}, 85 \%)$ as an orange solid: $\mathrm{mp} 50-51^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR $\delta: 1.21(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 6 \mathrm{H}), 4.48$ (hept, $\left.J=6.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 6.77$ (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.32(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.80$ (d, $J=13.5 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR : $\delta 21.7,70.1,116.1,121.9,131.1,134.6,139.0,161.4$.

## 1,2-Dimethoxy-4-(2-nitrovinyl)benzene (8b)

According to previous procedures, 3,4-dimethoxy benzaldehyde (7b) ( 1.73 g , $10.4 \mathrm{mmol})$ in $\mathrm{NH}_{4} \mathrm{OAC}(1.58 \mathrm{~g}) / \mathrm{AcOH}(20.0 \mathrm{~mL})$ was stirred for 10 min at $25^{\circ} \mathrm{C}$, followed by slow addition of $\mathrm{CH}_{3} \mathrm{NO}_{2}(2.3 \mathrm{~mL}, 41.7 \mathrm{mmol})$ with stirring for 5 min . The resulting mixture was heated to reflux at $120^{\circ} \mathrm{C}$ for 4 h , then quenched by addition of ice water $(10.0 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane $\left.-\mathrm{CH}_{2} \mathrm{Cl}_{2} 2: 1\right)$ to provide the expected $\mathbf{8 b}$ $(1.70 \mathrm{~g}, 78 \%)$ as an orange solid: $\mathrm{mp} 131-132{ }^{\circ} \mathrm{C}$ (lit. ${ }^{[11]} \mathrm{mp} 140-142{ }^{\circ} \mathrm{C}$ ), ${ }^{1} \mathrm{H}$ NMR $\delta: 3.84(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 6.83(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.09$ (dd, $J=8.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.87(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR : $\delta 55.8,55.9,110.2,111.2,122.7,124.5,135.0,139.2,149.4,152.7$.

## 1-Bromo-2-(2-nitrovinyl)benzene (8c)

Compound $8 \mathbf{c}$ was prepared, using the previous procedure, from 2-bromo benzaldehyde ( 7 c ) $\left(1.00 \mathrm{~g}, 5.4 \mathrm{mmol}\right.$ ) in $\mathrm{NH}_{4} \mathrm{OAC}(0.73 \mathrm{~g}) / \mathrm{AcOH}(15.0 \mathrm{~mL})$, followed by slow addition of $\mathrm{CH}_{3} \mathrm{NO}_{2}(1.0 \mathrm{~mL}, 19.2 \mathrm{mmol})$ with stirring for 5 min . The resulting mixture was heated to reflux at $120^{\circ} \mathrm{C}$ for 4 h , then quenched by addition of ice water $(10.0 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane- $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} 2: 1\right)$ to provide the expected 8c $(0.94 \mathrm{~g}, 76 \%)$ as a light yellow solid: $\mathrm{mp} 87-88^{\circ} \mathrm{C}$ (lit. ${ }^{[11]} \mathrm{mp} 86-87{ }^{\circ} \mathrm{C}$ ); ${ }^{1} \mathrm{H}$ NMR $\delta 7.21-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.44(\mathrm{~d}, J=13.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{dd}, J=7.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.58$ (dd, $J=7.5,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=13.8 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta 126.2,128.0$, 128.4, 130.1, 132.9, 133.8, 137.4, 138.7.

## 1-Isopropoxy-4-(2-nitroethyl)benzene (9a)

To a solution of $8 \mathbf{8 a}(0.71 \mathrm{~g}, 3.4 \mathrm{mmol})$ in $\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}(15.0 / 15.0 \mathrm{~mL})$ was cooled for 10 min at $0^{\circ} \mathrm{C}$. One portion of $\mathrm{NaBH}_{4}(0.19 \mathrm{~g}, 5.1 \mathrm{mmol})$ the solution was added to and stirred for 20 min , followed by the other portion of $\mathrm{NaBH}_{4}$ $(0.19 \mathrm{~g}, 5.1 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The mixture was allowed to stir for 30 min and then quenched with distilled $\mathrm{H}_{2} \mathrm{O}$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20.0 \mathrm{~mL})$. After removal of solvent, the residue was subjected to column chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate $8: 1$ ) to provide $9 \mathrm{a}(0.54 \mathrm{~g}, 75 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta: 1.24$ (d, $J=6.0 \mathrm{~Hz}, 6 \mathrm{H}$ ), $3.14(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.42$ (hept, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.46 (t, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.74(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.00(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta$ 21.9, 32.6, 69.8, 76.5, 116.1, 127.3, 129.5, 157.1; HRMS (EI) calcd. for $\mathrm{C}_{11} \mathrm{H}_{15} \mathrm{NO}_{3}$ $\left(\mathrm{M}^{+}\right)$209.1052; found 209.1048.

## 1,2-Dimethoxy-4-(2-nitroethyl)benzene (9b)

Compound $9 \mathbf{b}$ was prepared, using the previous procedure, from $\mathbf{8 b}(0.59 \mathrm{~g}$, $2.9 \mathrm{mmol})$ in $\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}(10.0 / 10.0 \mathrm{~mL})$, cooled for 10 min at $0^{\circ} \mathrm{C}$. One portion of $\mathrm{NaBH}_{4}(0.16 \mathrm{~g}, 4.3 \mathrm{mmol})$ was added, followed by the other portion of $\mathrm{NaBH}_{4}$
$(0.16 \mathrm{~g}, 4.3 \mathrm{mmol})$ with stirring for 30 min at $0^{\circ} \mathrm{C}$. The resulting mixture was then quenched by addition of distilled $\mathrm{H}_{2} \mathrm{O}(5.0 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 20.0 \mathrm{~mL})$. The combined organic layers were concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate 8:1) to give the expected $9 \mathrm{~b}(0.42 \mathrm{~g}, 70 \%)$ as a yellow oil (lit. $\left.{ }^{[12]} \mathrm{mp} 53-54{ }^{\circ} \mathrm{C}\right) ;{ }^{1} \mathrm{H}$ NMR $\delta: 3.19(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 4.51(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.63(\mathrm{~d}, J=1.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.67(\mathrm{dd}, J=8.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta 33.1,55.9$, 55.9, 76.5, 111.5, 111.7, 120.6, 128.0, 148.3, 149.2 .

## 1-Bromo-2-(2-nitroethyl)benzene (9c)

According to previous procedures, $8 \mathrm{c}(0.45 \mathrm{~g}, 2.0 \mathrm{mmol})$ in $\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(10.0 / 10.0 \mathrm{~mL})$ was cooled down to $0^{\circ} \mathrm{C}$, and one portion of $\mathrm{NaBH}_{4}(0.19 \mathrm{~g}$, 5.1 mmol ) was added with stirring for 10 min , followed by the other portion of $\mathrm{NaBH}_{4}(0.19 \mathrm{~g}, 5.1 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The mixture was allowed to stir for 20 min and then quenched with distilled $\mathrm{H}_{2} \mathrm{O}$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15.0 \mathrm{~mL})$. After removal of solvent, the residue was subjected to column chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate $\left.8: 1\right)$ to obtain $9 \mathrm{c}(0.30 \mathrm{~g}, 67 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta$ $3.34(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.54(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.04-7.20(\mathrm{~m}, 3 \mathrm{H}), 7.46$ (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 33.7,74.2,124.2,127.9,129.2,131.0,133.1,134.9$.

## 4-Isopropoxyphenylacetaldehyde (10a)

A mixture of $9 \mathbf{a}(0.51 \mathrm{~g}, 2.4 \mathrm{mmol})$ and $\mathrm{NaOH}(0.39 \mathrm{~g}, 9.8 \mathrm{mmol})$ in ethanol $(5.0 \mathrm{~mL})$ was stirred for 10 min at $25^{\circ} \mathrm{C}$, and then the solvent was evaporated in vacuo to give the desired sodium nitronate, which was dissolved in distilled $\mathrm{H}_{2} \mathrm{O}$ $(5.0 \mathrm{~mL})$. The solution was added into the two-layer mixture of concentrated $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}(2.5 / 5.0 \mathrm{~mL})$ and pentane $(20.0 \mathrm{~mL})$ for 1 h at $0^{\circ} \mathrm{C}$ and then warmed to $25^{\circ} \mathrm{C}$. After stirring for 4 h , the resulting suspension was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 20.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacou, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate 5:1) to provide the expected $10 \mathrm{a}(0.43 \mathrm{~g}, 100 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta: 1.31(\mathrm{~d}, J=6.3 \mathrm{~Hz}$, 6 H ), 3.57 (d, $J=2.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 4.51 (hept, $J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.08(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 9.67(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta 22.0,49.7,69.9$, 116.3, 123.4, 130.6, 157.2, 199.8, HRMS (EI) calcd. for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right)$178.0994; found 178.0988.

## 3,4-Dimethoxyphenylacetaldehyde (10b) ${ }^{[13]}$

A solution of $9 \mathbf{b}(0.49 \mathrm{~g}, 2.3 \mathrm{mmol})$ and $\mathrm{NaOH}(0.37 \mathrm{~g}, 9.2 \mathrm{mmol})$ in ethanol $(5.0 \mathrm{~mL})$ was stirred for 10 min at $25^{\circ} \mathrm{C}$, and then the removal of solvent in vacuo gave the intermediate sodium nitronate, which was dissolved in distilled $\mathrm{H}_{2} \mathrm{O}$ $(5.0 \mathrm{~mL})$. The solution was introduced into the two-layer mixture of concentrated $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}(2.5 / 5.0 \mathrm{~mL})$ and pentane $(20.0 \mathrm{~mL})$ for 1 h at $0^{\circ} \mathrm{C}$ and then warmed to $25^{\circ} \mathrm{C}$. After stirring for 5 h , the resulting suspension was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 20.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ 1:1) to provide
the expected $\mathbf{1 0 b}(0.42 \mathrm{~g}, 100 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta 3.56(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.81(\mathrm{~s}$, $6 \mathrm{H}), 6.64(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{dd}, J=8.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H})$, $9.66(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 50.1,55.8,55.9,111.6,112.6,121.8,124.1,148.4$, 149.3, 199.5.

## 2-Bromophenylacetaldehyde (10c) ${ }^{[14]}$

Compound 10c was prepared, using the previous procedure, from $9 \mathrm{c}(0.30 \mathrm{~g}$, $1.3 \mathrm{mmol})$ and $\mathrm{NaOH}(0.21 \mathrm{~g}, 5.2 \mathrm{mmol})$ in ethanol $(5.0 \mathrm{~mL})$ and stirred for 10 min at $25^{\circ} \mathrm{C}$, and then the solvent was evaporated in vacuo to give the desired sodium nitronate, which was dissolved in distilled $\mathrm{H}_{2} \mathrm{O}(3.0 \mathrm{~mL})$. The solution was added into the two-layer mixture of concentrated $\mathrm{H}_{2} \mathrm{SO}_{4} / \mathrm{H}_{2} \mathrm{O}(2.0 / 4.0 \mathrm{~mL})$ and pentane $(15.0 \mathrm{~mL})$ for 1 h at $0^{\circ} \mathrm{C}$ and then warmed to $25^{\circ} \mathrm{C}$. After stirring for 3 h , the resulting suspension was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15.0 \mathrm{~mL})$. The combined organic layers were concentrated in vacuo, and the residue was subjected to flash chromatography ( $\mathrm{SiO}_{2}$, hexane-ethyl acetate $10: 1$ ) to obtain the aldehyde $\mathbf{1 0 c}(0.25 \mathrm{~g}, 96 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] \delta: 3.92(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.19-7.37(\mathrm{~m}, 3 \mathrm{H}), 7.62$ $(\mathrm{d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 9.74(\mathrm{t}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] \delta 50.0,124.6$, 127.8, 129.1, 132.2, 132.6, 133.6, 197.6.

## 2-(4-Isopropoxyphenyl)ethylamine (11a) ${ }^{[15]}$

$\mathrm{LiAlH}_{4}(0.22 \mathrm{~g}, 5.8 \mathrm{mmol})$ was added to a solution of $\mathbf{8 a}(0.40 \mathrm{~g}, 1.9 \mathrm{mmol})$ in dried THF $(10.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and then stirred under $\mathrm{N}_{2}$ overnight at room temperature. The reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(5.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl-acetate- $\left.\mathrm{MeOH} 8: 2: 1\right)$ to provide the desired 11a ( $0.25 \mathrm{~g}, 73 \%$ ) as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta: 1.25(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 6 \mathrm{H}), 1.66$ (brs, 2H), 2.61 (t, $J=6.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.85 (t, $J=6.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 4.43 (hept, $J=6.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.75(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta 22.0,38.9$, 43.5, 69.9, 115.9, 129.7, 131.5, 156.3.

## 2-(3,4-Dimethoxyphenyl)ethylamine (11b) ${ }^{[16]}$

A mixture of $\mathbf{8 b}(0.30 \mathrm{~g}, 1.4 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.16 \mathrm{~g}, 4.3 \mathrm{mmol})$ in dried THF ( 10.0 mL ) was stirred at $25^{\circ} \mathrm{C}$ and then heated to reflux under $\mathrm{N}_{2}$ for 3 h . The reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(5.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, ethyl acetate- $\left.\mathrm{MeOH} 19: 1\right)$ to provide the desired 11b $(0.21 \mathrm{~g}, 83 \%)$ as an oil (lit. $\left.{ }^{[17]} \mathrm{mp} 155-156^{\circ} \mathrm{C}\right),{ }^{1} \mathrm{H}$ NMR $\delta: 1.30$ (brs, 2 H ), 2.58 $(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.81(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.87(\mathrm{~s}, 3 \mathrm{H}), 6.60-6.66$ $(\mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 39.2,43.2,55.4,55.5,111.0,111.7,120.3,132.0,147.1,148.5$.

## 2-(2-Bromophenyl)ethylamine (11c) ${ }^{[18]}$

$\mathrm{LiAlH}_{4}(0.22 \mathrm{~g}, 5.8 \mathrm{mmol})$ was added to a solution of $\mathbf{8 c}(0.44 \mathrm{~g}, 1.9 \mathrm{mmol})$ in dried THF $(10.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$, and then stirred under $\mathrm{N}_{2}$ overnight at room
temperature. The reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(5.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10.0 \mathrm{~mL})$. The combined organic layers was concentrated in vacuo, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl-acetate- MeOH 8:2:1) to provide the expected $11 \mathrm{c}(0.20 \mathrm{~g}, 56 \%)$ as a oil. ${ }^{1} \mathrm{H}$ NMR $\delta: 1.55$ (brs, 2H), 2.80-2.93 (m, 4H), 6.99-7.03 (m, 1H), 7.16-7.20 (m, 2H), 7.47 (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 40.1,42.0,124.6,127.4,128.0,130.9,132.9$, 139.0.

## Methyl N-[2-(3,4-Dimethoxyphenyl)ethyl]carbamate (12a) ${ }^{[19]}$

A solution of $\mathbf{1 1 b}(1.00 \mathrm{~g}, 5.5 \mathrm{mmol})$ in dried THF ( 30.0 mL ) was dissolved, followed by dropwise addition of methyl chloroformate ( $0.5 \mathrm{~mL}, 6.6 \mathrm{mmol}$ ) under $\mathrm{N}_{2}$ at $0^{\circ} \mathrm{C}$. The resulting solution was stirred at room temperature, and the reaction was monitored by thin-layer chromatography (TLC). The reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(20.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl-acetate $1: 1$ ) to provide the desired 12a ( $1.25 \mathrm{~g}, 95 \%$ ) as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta 2.69(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.35(\mathrm{q}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.60(\mathrm{~s}, 3 \mathrm{H}), 3.80(\mathrm{~s}$, $3 \mathrm{H}), 3.87(\mathrm{~s}, 3 \mathrm{H}), 4.80($ brs, 1 H$), 6.65-6.68(\mathrm{~m}, 2 \mathrm{H}), 6.75(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 35.6,42.3,52.0,55.7,55.8,111.3,111.9,120.6,131.2,147.6,148.9,157.0$.

## tert-Butyl N-[2-(3,4-Dimethoxyphenyl)ethyl]carbamate (12b)

According to the procedure of Toste and Still, ${ }^{[20]}$ a mixture of $\mathbf{1 1 b}(1.40 \mathrm{~g}$, 7.7 mmol ), dimethylaminopyradine (DMAP; $0.05 \mathrm{~g}, 0.4 \mathrm{mmol}$ ), and $\mathrm{N}, \mathrm{N}$-diisopropylethylamine (DIEA; $2.00 \mathrm{~g}, 15.5 \mathrm{mmol}$ ) in dried $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30.0 \mathrm{~mL})$ was added to $\mathrm{Boc}_{2} \mathrm{O}(2.00 \mathrm{~g}, 9.2 \mathrm{mmol})$ with dropwise at $0^{\circ} \mathrm{C}$. Stirring continued at $0^{\circ} \mathrm{C}$ for 30 min and at room temperature for 2 h . The reaction was quenched with saturated aqueous solution of $\mathrm{NH}_{4} \mathrm{Cl}(10.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 30.0 \mathrm{~mL})$. Evaporation and chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl-acetate$\mathrm{CH}_{2} \mathrm{Cl}_{2}$ 1:1:1) gave 12b ( $2.02 \mathrm{~g}, 93 \%$ ) as a white solid: mp $61-62^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\delta$ $1.21(\mathrm{~s}, 9 \mathrm{H}), 2.50(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.10(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.61(\mathrm{~s}, 3 \mathrm{H}), 3.63(\mathrm{~s}$, $3 \mathrm{H}), 4.60($ brs, 1 H$), 6.49-6.60(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 28.3,35.7,42.0,55.6,55.7$, $78.9,111.3,111.9,120.6,131.5,147.5,148.8,155.8$.

## 1-(4-Isopropoxybenzyl)-6,7-dimethoxy-2-methyl-1,2,3,4-tetrahydroisoquinoline (13)

A mixture of $\mathbf{1 2 a}(0.51 \mathrm{~g}, 2.1 \mathrm{mmol})$ and $\mathbf{1 0 a}(0.75 \mathrm{~g}, 4.2 \mathrm{mmol})$ in mixed solvents of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and $\mathrm{CHCl}_{3}(20.0 / 20.0 \mathrm{~mL})$ was introduced by dropwise addition of $\mathrm{BF}_{3}-\mathrm{OEt}_{2}(0.5 \mathrm{~mL}, 3.9 \mathrm{mmol})$ at $-78^{\circ} \mathrm{C}$ for 1 h . The resulting solution was stirred at room temperature, and the reaction was monitored by TLC. The reaction was quenched with saturated aqueous solution of $\mathrm{NaHCO}_{3}(10.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the crude product was directly used with no further purification. To a solution of the crude product in dried THF $(40.0 \mathrm{~mL}), \mathrm{LiAlH}_{4}(0.24 \mathrm{~g}, 6.3 \mathrm{mmol})$ was added at $0^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. The reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(10.0 \mathrm{~mL})$, which was extracted with
$\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$-ethyl acetate- MeOH 10:2:1) to provide the desired $13(0.50 \mathrm{~g}, 67 \%)$ as an oil. ${ }^{1} \mathrm{H}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] \delta: 1.45(\mathrm{~d}, J=6.0 \mathrm{~Hz}$, $6 \mathrm{H}), 2.42(\mathrm{~s}, 3 \mathrm{H}), 2.44-2.81(\mathrm{~m}, 4 \mathrm{H}), 2.99-3.15(\mathrm{~m}, 2 \mathrm{H}), 3.56(\mathrm{~s}, 3 \mathrm{H}), 3.63(\mathrm{t}$, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 3 \mathrm{H}), 4.52$ (hept, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.32(\mathrm{~s}, 1 \mathrm{H}), 6.57$ (s, $1 \mathrm{H}), 6.74(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.01(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right]: \delta$ $21.4,25.3,39.9,42.1,47.0,54.9,55.0,64.7,69.1,111.5,111.7,115.1,126.4,129.6$, 130.7, 132.0, 147.0, 147.6, 156.1; HRMS (EI) calcd. for $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{NO}_{3}\left(\mathrm{M}^{+}\right)$ 355.2147 ; found 355.2142 .

## 1,2-Dimethoxy-4,5,6a,7-tetrahydrodibenzo[de,g] quinoline-6-carboxylic Acid Methyl Ester (15a) ${ }^{[2 c]}$

Compound 15a was prepared, using the previous procedure, from 12a ( 0.30 g , $1.3 \mathrm{mmol})$ and $\mathbf{1 0 c}(0.50 \mathrm{~g}, 2.5 \mathrm{mmol})$ in mixed solvents of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and $\mathrm{CHCl}_{3}(20.0 /$ 20.0 mL ), followed by dropwise addition of $\mathrm{BF}_{3}-\mathrm{OEt}_{2}(0.3 \mathrm{~mL}, 2.5 \mathrm{mmol})$ at $-78^{\circ} \mathrm{C}$ for 1 h by TLC monitoring. The reaction was quenched with saturated aqueous solution of $\mathrm{NaHCO}_{3}(10.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the crude 14a was directly used no further purification. A mixture of $14 \mathrm{a}(0.53 \mathrm{~g}, 1.3 \mathrm{mmol})$, $\operatorname{AIBN}(0.01 \mathrm{~g}, 63.2 \mu \mathrm{~mol})$, and $\mathrm{Bu}_{3} \mathrm{SnH}(0.37 \mathrm{~g}$, 1.3 mmol ) in dried toluene ( 35.0 mL ) was heated to reflux under $\mathrm{N}_{2}$. The reaction was monitored by TLC and quenched with addition of cyclohexane ( 10.0 mL ), which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 35.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate 5:1) to obtain the desired 15a ( $0.22 \mathrm{~g}, 52 \%$ ) as an oil. ${ }^{1} \mathrm{H}$ NMR $\delta 2.51(\mathrm{~d}, J=15.0 \mathrm{~Hz}, 1 \mathrm{H})$, $2.66-2.89(\mathrm{~m}, 4 \mathrm{H}), 3.52(\mathrm{~s}, 3 \mathrm{H}), 3.62(\mathrm{~s}, 3 \mathrm{H}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 4.31(\mathrm{~d}, J=10.5 \mathrm{~Hz}$, $1 \mathrm{H}), 4.61$ (d, $J=12.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{~s}, 1 \mathrm{H}), 7.10-7.20(\mathrm{~m}, 3 \mathrm{H}), 8.31$ (d, $J=7.8 \mathrm{~Hz}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 30.2,35.2,38.9,51.6,52.6,55.9,59.9,111.5,126.0,127.0,127.5$, 127.6, 128.3, 128.4, 129.6, 131.7, 136.7, 145.6, 152.0, 155.9.

## 1,2-Dimethoxy-4,5,6a,7-tetrahydrodibenzo[de,g] quinoline-6-carboxylic Acid tert-Butyl Ester (15b) ${ }^{[2 c]}$

A mixture of $\mathbf{1 2 b}(0.28 \mathrm{~g}, 1.0 \mathrm{mmol})$ and $\mathbf{1 0 c}(0.39 \mathrm{~g}, 2.0 \mathrm{mmol})$ in mixed solvents of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and $\mathrm{CHCl}_{3}(20.0 / 20.0 \mathrm{~mL})$ was stirred and followed by dropwise addition of $\mathrm{BF}_{3}-\mathrm{OEt}_{2}(0.3 \mathrm{~mL}, 2.5 \mathrm{mmol})$ at $-78^{\circ} \mathrm{C}$ for 1 h by TLC monitoring. The reaction was quenched with saturated aqueous solution of $\mathrm{NaHCO}_{3}$ $(10.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the crude $\mathbf{1 4 b}$ was directly used no further purification. A mixture of $\mathbf{1 4 b}$ $(0.46 \mathrm{~g}, 1.0 \mathrm{mmol}), \operatorname{AIBN}(0.01 \mathrm{~g}, 63.2 \mu \mathrm{~mol})$, and $\mathrm{Bu}_{3} \operatorname{SnH}(0.29 \mathrm{~g}, 1.0 \mathrm{mmol})$ in dried toluene ( 30.0 mL ) was heated to reflux under $\mathrm{N}_{2}$. The reaction was monitored by TLC and quenched with addition of cyclohexane ( 10.0 mL ), which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate $\left.4: 1\right)$ to obtain the desired $\mathbf{1 5 b}$ $(0.21 \mathrm{~g}, 56 \%)$ as a white solid: $\mathrm{mp} 184-185^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR $\delta 1.41(\mathrm{~s}, 9 \mathrm{H}),: 2.56(\mathrm{~d}$, $J=14.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.70-2.90(\mathrm{~m}, 4 \mathrm{H}), 3.64(\mathrm{~s}, 3 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 4.34(\mathrm{~d}, J=9.9 \mathrm{~Hz}$, $1 \mathrm{H}), 4.58(\mathrm{~d}, J=12.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{~s}, 1 \mathrm{H}), 7.12-7.25(\mathrm{~m}, 3 \mathrm{H}), 8.35(\mathrm{~d}, J=7.8 \mathrm{~Hz}$,
$1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 28.5,30.4,35.4,38.4,51.6,55.9,60.0,79.8,111.4,126.5,126.9$, 127.5, 127.6, 128.1, 128.4, 129.8, 131.7, 137.0, 145.5, 151.9, 154.6; HRMS (EI) calcd. for $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{4}\left(\mathrm{M}^{+}\right)$381.1940; found 381.1943.

## 4-(6,7-Dimethoxy-2-methyl-1,2,3,4-tetrahydro-isoquinolin-1-ylmethyl)phenol (1)

$\mathrm{BCl}_{3}(1.6 \mathrm{~mL}, 1 \mathrm{M})$ was added dropwise to a solution of $\mathbf{1 3}(0.58 \mathrm{~g}, 1.6 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The reaction was stirred at room temperature and monitored by TLC. The resulting reaction was quenched by treatment with saturated aqueous solution of $\mathrm{NaHCO}_{3}(5.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate $\left.7: 1\right)$ to afford $1(0.35 \mathrm{~g}, 69 \%)$ as a white solid: mp $155-156{ }^{\circ} \mathrm{C}$ (lit. ${ }^{[4 \mathrm{~b}]} \mathrm{mp} 166^{\circ} \mathrm{C}$ ); ${ }^{1} \mathrm{H}$ NMR $\delta: 2.54(\mathrm{~s}, 3 \mathrm{H}), 2.59-2.96(\mathrm{~m}$, 4H), 3.13-3.28 (m, 2H), 3.53 (s, 3H), 3.71 (s, 1H), 3.74 (t, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.83 (s, $3 \mathrm{H}), 5.97(\mathrm{~s}, 1 \mathrm{H}), 6.56(\mathrm{~s}, 1 \mathrm{H}), 6.63(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\delta 24.3,40.3,41.9,45.8,55.4,55.7,64.8,111.1,111.2,115.5,124.9$, 128.3, 130.4, 130.7, 146.2, 147.3, 155.0; HRMS (FAB+H) calcd. for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NO}_{3}$ $\left[(\mathrm{M}+\mathrm{H})^{+}\right]$314.1756; found 314.1766.

## 1,2-Dimethoxy-6-methyl-5,6,6a,7-tetrahydro-4Hdibenzo[de,g]quinoline (2)

A solution of $\mathbf{1 5 a}(0.14 \mathrm{~g}, 0.4 \mathrm{mmol})$ in THF $(10.0 \mathrm{~mL})$ was treated by dropwise addition of $\mathrm{LiAlH}_{4}(1.6 \mathrm{~mL}$ of 1 M in THF, 1.6 mmol$)$ at $0^{\circ} \mathrm{C}$ for 1 h . The resulting reaction was quenched with distilled $\mathrm{H}_{2} \mathrm{O}(5.0 \mathrm{~mL})$ and allowed to warm to room temperature. When a gelatinous mixture formed, it was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(20.0 \mathrm{~mL})$. The mixture was filtered, washing with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and MeOH . The solvent was evaporated, and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hex-ane-ethyl acetate $1: 6$ ) to afford $2(80 \mathrm{mg}, 66 \%)$ as a colorless oil (lit. ${ }^{[21]} \mathrm{mp}$ $\left.165^{\circ} \mathrm{C}\right) .{ }^{1} \mathrm{H}$ NMR $\delta: 2.38-2.62(\mathrm{~m}, 3 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}), 2.93-3.15(\mathrm{~m}, 4 \mathrm{H}), 3.76(\mathrm{~s}$, $3 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 6.55(\mathrm{~s}, 1 \mathrm{H}), 7.12-7.25(\mathrm{~m}, 3 \mathrm{H}), 8.29(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 29.1,35.0,43.9,53.2,55.8,60.2,62.3,111.2,126.8,127.0,127.3,127.7$, $127.8,128.3,128.6,132.1,136.4,145.1,152.0$; HRMS (EI) calcd. for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{NO}_{2}$ $\left(\mathrm{M}^{+}\right)$295.1572; found 295.1568 .

## 1,2-Dimethoxy-5,6,6a,7-tetrahydro-4H-dibenzo[de,g]quinoline (3)

Compound 15b ( $0.10 \mathrm{~g}, 0.3 \mathrm{mmol}$ ) was dissolved in $\mathrm{MeOH}(5.0 \mathrm{~mL})$ and followed by dropwise addition of $\mathrm{AcCl}(30.0 \mu \mathrm{~L}, 0.4 \mathrm{mmol})$ at room temperature. The reaction was stirred at room temperature and monitored by TLC for 3 h . The resulting reaction was quenched by treatment with saturated aqueous solution of $\mathrm{NaHCO}_{3}(3.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-ethyl acetate- $\mathrm{MeOH} 8: 4: 1$ ) to provide the desired $\mathbf{3}(44 \mathrm{mg}, 59 \%)$ as a yellow solid: mp 208-209 ${ }^{\circ} \mathrm{C}$ (lit. ${ }^{[4 \mathrm{a}]} \mathrm{mp} 136-137^{\circ} \mathrm{C}$ ); ${ }^{1} \mathrm{H}$ NMR $\delta: 2.56-2.79$ (m, 3H), 2.89-2.96 (m, 2H), 3.28-3.30 (m, 1H), 3.54 (s, 3H), 3.72-3.78 (m, 1H), $3.76(\mathrm{~s}, 3 \mathrm{H})$,
$6.52(\mathrm{~s}, 1 \mathrm{H}), 7.09-7.19(\mathrm{~m}, 3 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H}), 8.26(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta$ $28.6,37.0,42.9,53.4,55.9,60.2,111.7,126.6,127.1,127.4,127.7,127.8,128.3$, 128.4, 132.0, 135.7, 145.3, 152.3; HRMS (EI) calcd. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{2}\left(\mathrm{M}^{+}\right)$ 281.1416; found 281.1410.

## (4-Isopropoxyphenyl)acetic Acid (17)

$i-\operatorname{PrBr}(3.3 \mathrm{~mL}, 35.1 \mathrm{mmol})$ was added to a mixture of 4-hydroxyacetophenone $(3.31 \mathrm{~g}, 24.3 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(4.15 \mathrm{~g}, 30.1 \mathrm{mmol})$ in acetone $(50.0 \mathrm{~mL})$ at $50^{\circ} \mathrm{C}$ and then refluxed at $70^{\circ} \mathrm{C}$ for 8 h by TLC monitoring. After cooling, the solution was evaporated in vacuo, and the brown residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane- $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} 10: 1\right)$ to give $16(4.12 \mathrm{~g}, 95 \%)$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\delta$ : $1.30(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 6 \mathrm{H}), 2.48(\mathrm{~s}, 3 \mathrm{H}), 4.58$ (hept, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{~d}, J=8.7 \mathrm{~Hz}$, $\mathrm{Hz}, 2 \mathrm{H}), 7.86(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR : $\delta 21.8,26.2,70.0,115.0,129.7,130.6$, $162.0,198.0$. A mixture of $16(3.01 \mathrm{~g}, 16.9 \mathrm{mmol})$, sulfur ( $1.08 \mathrm{~g}, 33.8 \mathrm{mmol}$ ), morpholine ( $4.4 \mathrm{~mL}, 50.3 \mathrm{mmol}$ ), and $p$-toluenesulfonic acid ( $97 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) was refluxed at $120^{\circ} \mathrm{C}$ for 4 h . After completion of the reaction, the mixture was allowed to cool, and $20 \% \mathrm{NaOH}(20.0 \mathrm{~mL})$ and tetrabutylammonium bromide ( $27 \mathrm{mg}, 85.0 \mu \mathrm{~mol}$ ) were added. Hydrolysis continued for further 8 h at $100^{\circ} \mathrm{C}$. The cooled reaction mixture was filtered and the filtrate was acidified with HCl to pH 2 , which was extracted with ethyl acetate $(3 \times 50.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane-ethyl acetate $\left.10: 1\right)$ to afford $\mathbf{1 7}$ $(2.33 \mathrm{~g}, 71 \%)$ as a white solid: mp $56-57{ }^{\circ} \mathrm{C}$ (lit. ${ }^{[22]} \mathrm{mp} 57-61^{\circ} \mathrm{C}$ ), ${ }^{1} \mathrm{H}$ NMR $\delta: 1.32$ (d, $J=6.0 \mathrm{~Hz}, 6 \mathrm{H}$ ), $3.56(\mathrm{~s}, 2 \mathrm{H}), 4.51$ (hept, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=8.7 \mathrm{~Hz}$, 2 H ), 7.17 (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 11.98 (brs, 1 H ); ${ }^{13} \mathrm{C}$ NMR $\delta 22.0,40.1,69.9,115.9$, 125.1, 130.4, 157.1, 178.3.

## N-[2-(3,4-Dimethoxyphenyl)-ethyl]-2-(4-isopropoxyphenyl)acetamide (18)

Oxalyl chloride $(0.2 \mathrm{~mL}, 2.4 \mathrm{mmol})$ was added to a solution of $\mathbf{1 7}(0.39 \mathrm{~g}$, 2.0 mmol ) in dried $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15.0 \mathrm{~mL})$ dropwise at room temperature with stirring for 30 min , followed by evaporation to give the residue, which was directly used with no further purification. A solution of the crude product in dried $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20.0 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$, followed by dropwise addition of $\mathbf{1 1 b}(0.36 \mathrm{~g}, 2.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10.0 \mathrm{~mL})$ for 1 h . The resulting reaction was quenched by treatment with saturated aqueous solution of $\mathrm{NaHCO}_{3}(5.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(3 \times 30.0 \mathrm{~mL})$. After the removal of solvent, the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane- $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-ethyl acetate $\left.5: 1: 1\right)$ to provide the desired $\mathbf{1 8}$ $(0.59 \mathrm{~g}, 82 \%)$ as a white solid: mp $80-81{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] \delta 1.26$ (d, $J=6.0 \mathrm{~Hz}, 6 \mathrm{H}), 2.66(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.35(\mathrm{~s}, 2 \mathrm{H}), 3.36(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.73$ $(\mathrm{s}, 3 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 4.55(\mathrm{hept}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.62-6.65(\mathrm{~m}, 1 \mathrm{H}), 6.76-6.81(\mathrm{~m}$, $4 \mathrm{H}), 7.01($ brs, 1 H$), 7.13(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right]: \delta 21.4,35.0$, 40.7, 42.1, 55.0, 55.2, 69.2, 111.9, 112.6, 115.5, 120.6, 128.1, 130.1, 132.1, 147.9, 149.3, 156.7, 170.3; HRMS (EI) calcd. for $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{4}\left(\mathrm{M}^{+}\right) 357.1940$; found 357.1939.

## 1-(4-Isopropoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (19)

$\mathrm{POCl}_{3}(0.1 \mathrm{~mL}, 1.1 \mathrm{mmol})$ was added to a solution of $\mathbf{1 8}(0.26 \mathrm{~g}, 0.7 \mathrm{mmol})$ in dried acetonitrile $(10.0 \mathrm{~mL})$ at room temperature, and the resulting solution was refluxed for 2 h . The reaction was quenched with saturated aqueous solution of $\mathrm{NaHCO}_{3}(3.0 \mathrm{~mL})$, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10.0 \mathrm{~mL})$. After the removal of solvent, the crude product was directly used no further purification. The crude reaction mixture was dissolved in $\mathrm{MeOH}(10.0 \mathrm{~mL})$ at room temperature for $5 \mathrm{~min} . \mathrm{NaBH}_{4}(0.11 \mathrm{~g}, 2.9 \mathrm{mmol})$ was added at room temperature, and the solution was stirred and monitored by TLC for 1 h . The solvent was evaporated and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$-ethyl acetate 1:15) to obtain $19(0.15 \mathrm{~g}, 60 \%)$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\delta: 1.33(\mathrm{~d}, J=6.0 \mathrm{~Hz}$, $6 \mathrm{H}), 2.18(\mathrm{brs}, 1 \mathrm{H}), 2.70-2.95(\mathrm{~m}, 4 \mathrm{H}), 3.11-3.25(\mathrm{~m}, 2 \mathrm{H}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 3.85(\mathrm{~s}$, $3 \mathrm{H}), 4.08-4.12(\mathrm{~m}, 1 \mathrm{H}), 4.52$ (hept, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{~s}, 1 \mathrm{H}), 6.62(\mathrm{~s}, 1 \mathrm{H}), 6.85$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.14(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\delta 22.1,29.4,40.6,41.7$, $55.8,55.9,56.8,69.8,109.4,111.7,116.0,127.2,130.3,130.4,130.7,146.9,147.3$, 156.5.

## Compound 13 from 19

To a solution of $19(0.14 \mathrm{~g}, 0.4 \mathrm{mmol})$ in $\mathrm{MeOH}(5.0 \mathrm{~mL}), 37 \%$ aqueous solution of formaldehyde $(0.06 \mathrm{~mL}, 0.8 \mathrm{mmol})$ was added at room temperature with stirring for $1 \mathrm{~h} . \mathrm{NaBH}_{4}(23 \mathrm{mg}, 0.6 \mathrm{mmol})$ was added at room temperature, and the solution was stirred and monitored by TLC for 1 h . The solvent was evaporated and the residue was subjected to flash chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$-ethyl acetate 1:7) to obtain $13(0.12 \mathrm{~g}, 83 \%)$ as a yellow oil.

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