

# The Convergent Synthesis of CI-981, an Optically Active, Highly Potent, Tissue Selective Inhibitor of HMG-CoA Reductase

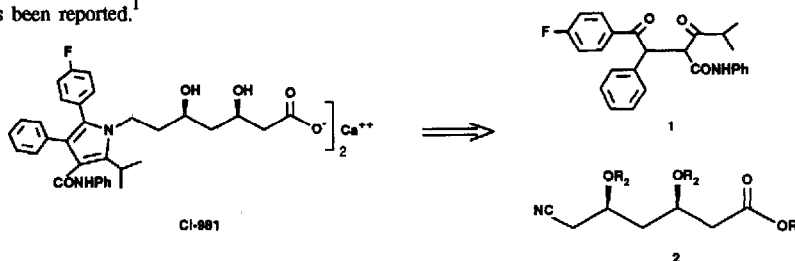
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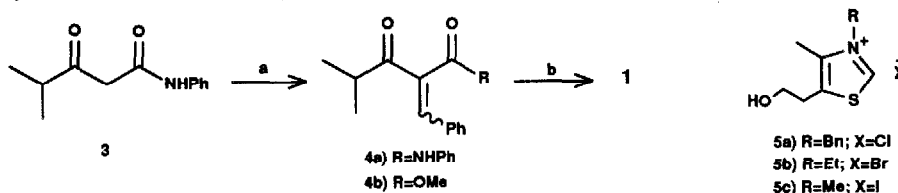
**Abstract:** The synthesis of CI-981 is described starting from isobutyrylacetanilide (3) and the key chiral intermediate 2.

CI-981, a potent and tissue selective inhibitor of HMG-CoA Reductase, is currently undergoing clinical trials and could prove to be an important addition to therapy for the treatment of hypocholesterolemia and the prevention of atherosclerosis.<sup>1,2</sup>

Economical, large scale syntheses of enantiomerically pure compounds as complex as CI-981 require, if possible, a convergent route.<sup>3</sup> A convergent synthesis of CI-981 required the preparation of two key intermediates; (1) and (2). However, a Paal-Knorr synthesis of such a complex pyrrole was by no means assured. The preparation of the 7-carbon side-chain acid derivative (2) has been reported.<sup>1</sup>



Diketone 1 was prepared in two steps from commercially available isobutyrylacetanilide (3). The addition of p-fluorobenzaldehyde to the Knoevenagel product 4a was carried out using the Stetter reaction.<sup>4</sup> The choice of catalyst (5) proved to be the key to the success of this reaction.

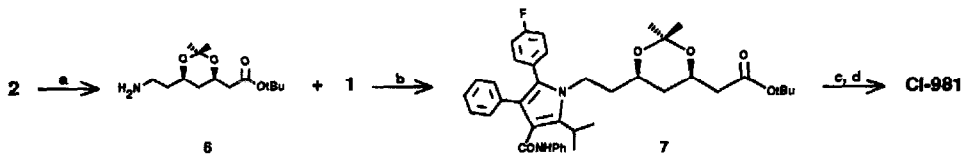


a.  $\beta$ -Alanine, AcOH, Hexane,  $\Delta$  (85%); b. 5b, EtOH,  $\Delta$ , then 4a, 4-F-C<sub>6</sub>H<sub>4</sub>CHO, NEt<sub>3</sub>,  $\Delta$  (80%)

Benzyl thiazolium catalyst (5a), which worked very well in the reaction of p-fluorobenzaldehyde with the methyl ester (4b)<sup>2a</sup>, gave the benzoin condensation product of p-fluorobenzaldehyde as the major product when used in the reaction with 4a.<sup>5</sup> However, use of either the ethyl (5b) or methyl thiazolium catalyst (5c) (20 mol. %) under anhydrous, concentrated conditions provided the diketone (1) in 80% yield.<sup>6</sup> To our knowledge this is the first example of the extension of the Stetter reaction to carboxamides.

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The preparation of the amine, (6)<sup>7</sup> from the intermediate nitrile (2) was accomplished in high yield using a Molybdenum doped Raney Nickel catalyst<sup>8</sup>, under 50 psig hydrogen pressure, in ammonia-methanol at room temperature.

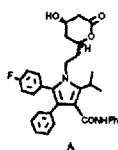


a.  $\text{Re-Ni}$ ,  $\text{MeOH}$ , 50 PSIG  $\text{H}_2$  (85%); b.  $(\text{CH}_3)_3\text{CO}_2\text{H}$ ,  $\text{C}_7\text{H}_8$ ,  $\text{C}_7\text{H}_{16}$ ,  $\text{C}_4\text{H}_8\text{O}$ ,  $\Delta$  (75%); c.  $\text{HCl}$ ,  $\text{MeOH}$  then  $\text{NaOH}$ ; d.  $\text{Ca}(\text{OAc})_2$

After extensive optimization of the Paal-Knorr pyrrole formation,<sup>9</sup> conditions using a ternary solvent mixture of toluene-heptane-tetrahydrofuran (1:4:1) and pivalic acid catalysis, provided a 75% yield of 7.<sup>10</sup> Conversion of 7 to CI-981 was carried out without isolation, by deprotection of the acetal using aqueous  $\text{HCl}$ /methanol, dilute base hydrolysis of the tert-butyl ester (anchimeric assistance) and treatment of the derived sodium salt with  $\text{Ca}(\text{OAc})_2$ . The hemi-calcium salt, CI-981, was isolated as an amorphous solid in an overall yield of 60% from 2, with an enantiomeric purity of  $\geq 99.5\%$ .<sup>11,12</sup>

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- For alternative approaches to the synthesis of CI-981 see reference 2a and U.S. Patent 5,003,080.
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- It is reported that benzoin formation is a fast reversible reaction that precedes addition and that use of the appropriate benzoin in place of the aldehyde provides the same product. In our experience, when using p-fluorobenzaldehyde, the use of (5a) as the catalyst provided excellent conversion to the benzoin which, even under forcing conditions, could not be converted, in acceptable yield, to the required diketone (1). However use of the N-alkyl thiazolium catalysts (5b or c) provided excellent conversion to 1, with very little benzoin formation observed (HPLC). Noteably, when using 5b or c, benzoin formation is dramatically increased by dilution of the reaction mixture.
- <sup>1</sup>H NMR spectrum of 1 (200 MHz, in  $\text{CDCl}_3$ )  $\delta$  1.03 (d, 3H), 1.22 (d, 3H), 2.98 (quin., 1H), 4.91 (d,  $J=11$  Hz, 1H), 5.51 (d,  $J=11$  Hz, 1H), 6.98 - 7.43 (m, 12H), 8.17 (dd, 2H), 9.41 (br s, 1H); m.p. 208-209.5 C. Elemental analysis and FTIR were in accord with structure.
- <sup>1</sup>H NMR spectrum of 6 (200 MHz, in  $\text{CDCl}_3$ )  $\delta$  1.0-1.2 (m, 1H), 1.22 (s, 3H), 1.31 (s, 12H), 1.35-1.45 (m, 3H), 2.15 (dd,  $J=6.2$  and 15.1 Hz, 1H), 2.29 (dd,  $J=7.0$  and 15.1 Hz, 1H), 2.66 (t,  $J=6.6$  Hz, 2H), 3.82 (m, 1H), 4.12 (m, 1H);  $[\alpha]_D^{25} +14.3^\circ$  ( $c=1$ ,  $\text{CHCl}_3$ ). <sup>13</sup>C NMR, FTIR and Mass Spectral data were in accord with structure.
- Available as A7000<sup>®</sup> from Activated Metals or 3100<sup>®</sup> from Grace Co.
- Paal, C., *Ber.*, 1885, 18, 367; Knorr, L., *Ber.*, 1885, 18, 299.
- <sup>1</sup>H NMR spectrum of 7 (200 MHz, in  $\text{CDCl}_3$ )  $\delta$  1.0-1.7 (m, 5H), 1.30 (s, 3H), 1.36 (s, 3H), 1.43 (s, 9H), 1.53 (d,  $J=7.1$  Hz, 6H), 2.23 (dd,  $J=6.3$  and 15.3 Hz, 1H), 2.39 (dd,  $J=6.3$  and 15.3 Hz, 1H), 3.5-3.9 (m, 3H), 4.0-4.2 (m, 2H), 6.8-7.3 (m, 14H);  $[\alpha] +4.71^\circ$  ( $c=1$ ,  $\text{CHCl}_3$ ). <sup>13</sup>C NMR data was in accord with structure.
- <sup>1</sup>H NMR spectrum of CI-981 (200 MHz, in  $\text{CDCl}_3$ )  $\delta$  1.26 (m, 2H), 1.37 (m, 6H), 1.59 (m, 2H), 2.04 (m, 2H), 3.24-3.96 (m, 5H), 4.80 (brs, 1H), 5.75 (brs, 1H), 7.00-7.22 (m, 12H), 7.52 (d, 2H), 9.82 (s, 1H);  $[\alpha]_D -7.4^\circ$  ( $c=1$ , DMSO). Elemental analysis, <sup>13</sup>C NMR, FTIR and Mass Spectral data were in accord with structure.
- The enantiomers of CI-981 have not been resolved by chiral HPLC. Conversion to lactone A allows determination of enantiomeric purity, by HPLC at 254 nm using a 25 cm Chiralcel OF column, with a mobile phase of hexane-IPA (82:18) at a flow rate of 1 ml/min.



<sup>1</sup>H NMR spectrum of A (200 MHz, in  $\text{CDCl}_3$ )  $\delta$  1.37 (s, 3H), 1.40 (s, 3H), 1.5-1.8 (m, 4H), 2.25-2.66 (m, 2H), 3.24 (m, 1H), 3.9-4.1 (m, 3H), 4.49 (m, 1H), 5.18 (d,  $J=3.1$  Hz, 1H), 6.9-7.4 (m, 12H), 7.52 (d,  $J=7.8$  Hz, 2H), 9.82 (s, 1H); m.p. 159.2-160.7°C;  $[\alpha]_D^{25} +26.05^\circ$  ( $c=1$ ,  $\text{CHCl}_3$ ). Elemental analysis, <sup>13</sup>C NMR, and FTIR data were in accord with structure.