The ratio of cholesterol 5,6-secosterols formed from ozone and singlet oxygen offers insight into the oxidation of cholesterol *in vivo*[†]

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Ongoing efforts to unravel the origins of the cholesterol 5,6-secosterols (1a and 1b) in biological systems have revealed that the two known chemical routes to these oxysterols, ozonolysis of cholesterol (3) and Hock-cleavage of $5-\alpha$ -hydroperoxycholesterol (4a), are distinguishable based upon the ratio of the hydrazone derivatives (2a and 2b) formed in each case and this ratio offers an insight into the chemical origin of the secosterols *in vivo*.

In a recent report, Pratt and co-workers¹ have shown that Hock-cleavage of 5α -hydroperoxycholesterol (**4a**), which can arise from the singlet oxygen ene reaction with cholesterol (**3**), occurs facilely under acidic conditions in organic solvents leading to the formation of primarily cholesterol 5,6-secosterol atheronal-B (**1b**). Atheronal-A (**1a**) is either not formed at all or is a minor component in participating solvents such as ethanol (Fig. 1).

We discovered the cholesterol 5,6-secosterols 1a and 1b within human atherosclerotic plaque material *in vivo*² and surmised that, based upon the wealth of literature in the field of chemical-, biological- and auto-oxidation of cholesterol at the time,^{3,4} that only ozone was capable of generating 1a from cholesterol (3).² In fact, we considered the presence of 1a [measured as 2a after extraction and derivatization with 2,4-dinitrophenyl (2,4-DNP) hydrazine] within inflammatory arteries as indirect evidence that an oxidant with the chemical signature of ozone may be generated during atherosclerosis progression. We also showed that 1a undergoes an almost instantaneous aldolization process to form 1b in whole blood and therefore the cholesterol 5,6-secosterols 1a and 1b were both potential signature molecules for cholesterol ozonolysis *in vivo.*²

The oxysterols 1a and 1b are proatherogenic⁵ and induce protein misfolding and amyloidogenesis⁶ in a number of biologically-relevant proteins.^{2,7} Given the pathological effects of 1a and 1b, we are striving to understand their chemical origin *in vivo*. Herein we have studied the formation of 1a and **1b**, analyzed as their 2,4-DNP hydrazones (**2a** and **2b**), during the oxidation of cholesterol (**3**) in biologically-relevant formats—human low-density lipoprotein (hLDL); human high-density lipoprotein (hHDL); and liposomes using cellular, biological and chemical oxidant systems in aqueous buffer (Table 1 and ESI[†]).

The 2,4-DNP hydrazone of atheronal-B (2b) is observed at levels above untreated hLDL ($\sim 0.01\%$ of 3), after oxidation by superoxide anion (Table 1, entries 1 and 2), by singlet oxygen generating systems (Table 1, entries 6 and 7), and by activated human polymorphonuclear leukocytes (hPMNs, Table 1, entries 8 and 9). In contrast, no elevation in 2b above untreated samples is observed after hLDL treatment with H₂O₂, HOCl or hydroxyl radical generating systems (Table 1, entries 3-5). The generation of 2b during hLDL oxidation by phorbol myristate acetate (PMA)-activated hPMNs, is proportional to cell density, is consistently higher in lipid extracts of hLDL rather than the PMN cell membranes and is elevated in D₂O relative to H₂O-containing buffer. These combined data suggest that the oxidant responsible for the generation of **2b** is both sufficiently long-lived to be released into the medium and its life-time is prolonged in deuterated buffers, hinting at roles for both ${}^{\bullet}O_2^{-}/HO_2^{\bullet}$ and ¹O₂ (Table 1, entries 8 and 9 and ESI,[†] Tables S1–S3).

The photosensitized and hPMN-mediated oxidation of cholesterol in hHDL in aqueous buffer (PBS, pH 7.4) parallels the observations with hLDL in that the 2,4-DNP hydrazone of cholesterol secosterol **2b** is elevated above untreated levels and



Fig. 1 Chemical routes from cholesterol 3 into the cholesterol secosterols 1a and 1b are being sought.

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Entry	Oxidant system	Reactive species	Products ^{b} (% of 3)	
			2a	2b
1	Xan/XO ^c	•O ₂ ⁻ /HO ₂ •	n.d.	0.02
2	$Acet/XO^d$	$\bullet O_2^-/HO_2^{\bullet}$	n.d.	0.05
3	Glc/GlcOx ^e	H_2O_2	n.d.	> 0.01
4	Glc/GlcOx/MPO ^f	H ₂ O ₂ /HOCl	n.d.	> 0.01
5	Glc/GlcOx/FeCl2 ^g	H_2O_2/HO^{\bullet}	n.d.	> 0.01
6	$HPIX/h\nu^h$	${}^{1}O_{2}$	n.d.	2.30
7	Glc/GlcOx/HPIX/hv	$H_2O_2/^1O_2$	n.d.	2.61
8	hPMNs ⁱ	Multiple	n.d.	0.30
9	hPMNs/D ₂ O	Multiple	n.d.	0.35
10	O_3/O_2^{j}	0,	0.40	0.10

^{*a*} For full details see the ESI;[†] lipid extracts were derivatized with 2,4-DNP hydrazine (2 mM) in isopropanol–ethanol (9 : 1) containing HCl (0.08%) ^{*b*} Products were analyzed as their 2,4-DNP hydrazones **2a** or **2b** and are reported as the amount of secosterol formed as a percentage of the cholesterol in hLDL at t = 0. ^{*c*} Xanthine (Xan, 0.5 mM)/xanthine oxidase (XO, 0.4 mg ml⁻¹). ^{*d*} Acetaldehyde (Acet, 10 mM) incubated for 30 min. ^{*e*} Glucose (Glc, 2 mM)/glucose oxidase (GlcOx, 0.1 mg ml⁻¹) incubated for 14 h at 37 °C. ^{*f*} Same as e with myeloperoxidase (MPO, 0.2 μ M). ^{*g*} Same as e with FeCl₂ (20 μ M). ^{*h*} Hematoporphyrin IX (HPIX, 40 μ M), $h\nu$ (400–700 nm), 14 h incubation. ^{*i*} hPMNs, (5.4 × 10⁷ cells per ml) activated with PMA (0.1 mg ml⁻¹) in PB or PB in D₂O. ^{*j*} A stream of O₃/O₂ gas (5 ml min⁻¹) was passed over a solution of hLDL in PBS (pH 7.4) for 40 s.

no **2a** is observed (ESI,[†] Tables S4 and S5). Similarly, the photosensitized oxidation of **3** in liposomes (DOPC 80 mol%; **3** 5 mol%; PIP-2 15 mol%) in PBS (pH 7.4) implicates a ${}^{1}O_{2}$ pathway to **2b**, because the measured levels of **2b** are elevated in PBS containing D₂O, relative to H₂O (ESI,[†] Table S6).

A clear finding of the hLDL oxidation studies detailed in Table 1 is that the 2,4-DNP hydrazone of atheronal-A (2a) is only generated after ozonolysis (Table 1). Thus, passage of an ozone-oxygen mixture over the surface of an hLDL solution containing catalase in aqueous buffer (pH 7.4), followed by 2,4-DNP hydrazine derivatization of the lipid extract, yields 2a and **2b** in a ratio of ~ 4 : 1 (Table 1, entry 10, ESI,[†] Fig. S1). To validate this result, cholesterol 3, which had been exposed to ozone in aluminium oxide (Al₂O₃)-treated CDCl₃, was shaken in aqueous buffers at a range of physiological pHs (5.8, 6.5, 7.0 and 7.4) in the presence or absence of a reductant (Zn dust), and the resultant products were assessed by ¹H NMR (ESI,† Fig. S2). This analysis reveals that, in all cases tested, only cholesterol secosterol 1a is generated in this aqueous treatment of ozonolyzed 3. Secosterol 1b is not formed by this route as confirmed by 2,4-DNP hydrazine derivatization of the ¹H NMR samples.

The oxidation of **3** by ${}^{1}O_{2}$ is a well-understood process that was first reported by Schenck.⁸ In brief, the 5 α -hydroperoxide **4a** is formed as the major product, which can rearrange to the 7 α -hydroperoxide **4b** under certain conditions.⁹ Therefore, the observation that ${}^{1}O_{2}$ and superoxide anion-mediated biological- and chemical-oxidations of hLDL, hHDL and liposome-derived **3** in aqueous buffers result in elevated levels of **2b** could be a result of the following possibilities: (1) conversion of **4a** or **4b** into **1a** and subsequent aldolization of **1a** into **1b** *in aqua* (followed by 2,4-DNP hydrazine derivatization after extraction); (2) conversion of **4a** or **4b** directly into **1b** *in aqua* (followed by 2,4-DNP hydrazine derivatization after extraction); or (3) conversion of **4a** or **4b** into **2b** during the 2,4-DNP hydrazine derivatization process.

A number of experiments have been performed that help to deconvolute these alternatives. When authentic 4a is incubated in aqueous buffers for 20 h (pH 5.8-7.5), the allylic hydroperoxide undergoes a low percentage isomerization into 4b, but there is no formation of 1a or 1b (ESI,† Table S7 and Fig. S3). In addition, we have shown previously that authentic 1a does not undergo measurable aldolization to 1b under the conditions of these studies.⁷ Finally, Pratt and co-workers¹ have recently observed that Hock-cleavage of 4a occurs in the presence of 2,4-DNP hydrazine to give 2b, with a trace of 2a being formed. Using authentic 4a, we find that this hydroperoxide undergoes Hock-cleavage during the hydrazine derivatization process utilized in this study (2 mM 2,4-DNP hydrazine, 0.08% HCl in isopropanol-ethanol (9:1)) to yield quantitative amounts of 2b (no 2a is detected, ESI, † Fig. S4). In addition, 7α -hydroperoxide **4b** does not undergo conversion into either 2a or 2b during the 2,4-DNP hydrazine derivatization process vide supra. Therefore, the **2b** that arises after ${}^{\bullet}O_2^{-}$, ${}^{1}O_2$ and PMA-activated hPMN-mediated oxidation of hLDL, hHDL and liposomes (Table 1) is most likely a result of cholesterol peroxidation to generate 4a, followed by Hockcleavage of 4a during the 2,4-DNP hydrazine derivatization process.

We have studied the singlet oxygen route to the cholesterol secosterols further: photochemical oxidation of 3 in Al₂O₃treated CDCl₃ with in situ acid (TFA, 1% v/v) leads to a good yield of **1b** (>69%), and surprisingly, albeit in low yield, of 1a ($\sim 3.5\%$). Inclusion of participating solvents, such as methanol, or increasing acid concentration does not significantly increase the yield of isolated secosterol 1a (ESI,† Table S8). When the concentration of TFA is lowered sufficiently (0.02% v/v), the progress of Hock-cleavage of 4a to form 1b can be followed by ¹H NMR, reinforcing that 1a is not formed even as a discrete intermediate, during this process on a time-scale where allylic isomerization of 4a to 4b competes measurably (ESI,† Table S9 and Fig. S5). Furthermore, when authentic 4a is incubated in isopropanol-ethanol (9:1) with 0.08% HCl for 3 h, ¹H NMR reveals a complex mixture of 1b and 1b-derived dehydration and acetal products but no 1a or 1a-derived products. Derivatization of this 4a-derived mixture with 2,4-DNP hydrazine then yields 2b and hydrazones of 1b-derived dehydration products with no measurable 2a or 2a-derived dehydration or acetal products in either case (ESI,† Fig. S6). Thus, the summary of this singlet oxygen/Hock-cleavage route to the cholesterol secosterols in organic solvents is that in all the systems we have studied, 1b is formed in vast excess over 1a.

The ${}^{1}O_{2}$ route to secosterol hydrazone **2b** has a number of implications for our quantification method of atheronals in biological systems. Analytical methods to quantify cholesterol secosterols from biological extracts must be able to deconvolute the 5 α -hydroperoxycholesterol **4a** from the cholesterol secosterol **1b** (the 2,4-DNP hydrazine method does not). To this end, we have developed an isotope dilution (ID) GC-MS assay, which measures the levels of **1b** in biological extracts after treatment with bis(trimethylsilyl) trifluoroacetamide (BSTFA)

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in pyridine, conditions where **4a** is completely stable (ESI,[†] Table S10, Fig. S7-S8)). A comparison of this IDGCMS assay with the 2,4-DNP hydrazine method in a plasma analysis of **1b**-derivatives in the apo $E^{-/-}$ mouse model of atherosclerosis reveals that the measured levels are not significantly different [GCMS method = 7.5 ± 1.3 nM (n = 4); 2,4-DNP hydrazine method = 9.3 ± 2.4 nM (n = 6); p > 0.05], suggesting that **4a** is not contributing significantly to the measured amounts of **2b** in this system (ESI,[†] Fig. S9).

Analysis of human arterial plaque extracts by IDGCMS reveals a mean [1a(b)] of 32 ± 15 pmol mg⁻¹ of tissue which is similar to levels we previously reported for 2b in plaque extracts² (ESI,†Fig. S10–S12). Interestingly, treatment of these arterial plaque extracts with Zn/acetic acid prior to lipid extraction results in an increase in the measured 1a(1b) levels, suggesting that either 4a or cholesterol ozonides are present in these atheroma samples and are being converted into 1a and 1b during tissue processing. We are currently investigating this finding (ESI,† Fig. S11 and S12).

We are also interested in understanding to what extent the acid-catalyzed rearrangement of **4a** or a chemical oxidant with the chemical signature of ozone may be contributing to cholesterol secosterol biogenesis *in vivo*. As described *vide supra* authentic **4a** is stable at physiological pH for >20 h and 5 α -hydroperoxide **4a** has been shown to be isolable from biological samples by a number of groups.^{4,10} However, one cannot discount that the formation of **1b**, which is ubiquitous in all biological samples we have tested thus far, does not, at least in part, arise from this route.

Pratt and co-workers¹ have shown that Hock-cleavage of authentic 4a in organic solvents leads to levels of $1b \gg 1a$ and we herein have shown that 4a, generated by chemical- and biological-oxidations of cholesterol in aqueous buffers, generates 2b (with no measurable 2a) upon 2,4-DNP hydrazine derivatization. Ozonolysis of 3 in aqueous buffers generates $1a \gg 1b$ (at all physiologically-relevant pHs either in the presence or absence of added chemical reductant). We have also studied the ozone-mediated oxidation of hLDL-bound cholesterol in aqueous buffer over a range of pH values, temperatures and durations of reaction and found that the ratio 1a: 1b varies from $\sim 15: 1$ to $\sim 4: 1$, but in all cases, the level of atheronal-A is greater than atheronal-B (ESI,† Table S12). An analysis of our original report of the levels of cholesterol secosterols (measured as their hydrazones 2a and **2b**) in human arterial plaque samples reveals that the ratio of 2a : 2b varies greatly (61.3 : 1 to 0.06 : 1, n = 28) (ESI, \dagger Table S11). However, the ratio of **2a** : **2b** is >1 in 20 of the 28 samples and > 10 in 10 of the 28 samples. Therefore, the dominant form of cholesterol 5,6-secosterols in arterial plaque extracts, where aldolization is minimal, is **1a**.² Based upon our

new understanding of the ozone and singlet oxygen pathways from **3** into **2a** and **2b** (after 2,4-DNP hydrazine derivatization), this clinical data offers support for an oxidant with the chemical signature of ozone playing a composite role in arterial plaque cholesterol secosterol **1a** biogenesis.

While the debate as to the relevance of ozone generation in biological systems continues,¹¹ we continue to search for more direct markers of this oxidant in biological extracts. The terminal products from cholesterol ozonolysis, **1a** and **1b**, may not discriminate the pathways to their formation under all conditions, however, a ratio of **1a** to **1b** of >1 is strong support for oxidation of **3** by an oxidant with the chemical signature of ozone. Our current search is aimed at quantifying more direct species that originate as a result of the primary chemical reaction of ozone with biological molecules, such as 1,2,4-trioxolanes.

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