

Palladium-Catalyzed *meta*-Selective C–H Functionalization by Noncovalent H-Bonding Interaction

Guoshuai Li, Yifei Yan, Pengfei Zhang, Xiaohua Xu,* and Zhong Jin*



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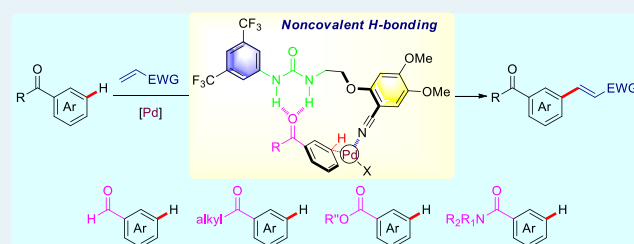
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ABSTRACT: Controlling positional selectivity represents one of the most important aspects in transition-metal-catalyzed C–H bond functionalization. However, the conventional directing template strategies via a covalent binding to the substrates are always hindered by prior stoichiometric installation and removal of the directing groups. Herein, we report a palladium-catalyzed *meta*-selective C–H olefination of aromatic carbonyl compounds by noncovalent hydrogen-bonding interaction. *N,N'*-Substituted ureas were engineered to serve as a H-bonding donor for binding to the substrates and, meanwhile, achieve site-selective control by the integrated directing group.

KEYWORDS: palladium, noncovalent H-bonding interaction, C–H activation, olefination, urea

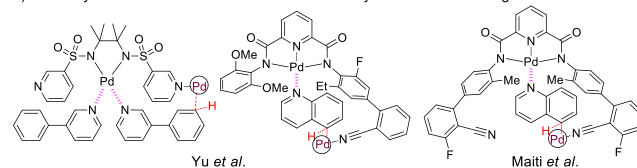


Transition-metal-catalyzed remote *meta*- or *para*-C–H bond functionalization of arenes^{1,2} has attracted the extensive interest from both chemical and medicinal communities in recent years owing to its vast potentials for step- and atom-economic synthesis of active pharmacophores, as well as late-stage diversification of pharmaceuticals. To render the catalyst proximal to the distal C–H bond efficiently, the C–H activation process frequently requires the participation of a deliberate σ -bonding directing group (DG) in the formation of a metal-embedded macrocyclic transition state. However, the stoichiometric installation and removal of a DG consistently influence the synthetic efficacy of the C–H functionalization strategy.^{3,4} Moreover, some kinds of substrates such as aromatic aldehydes/ketones and esters, among others, are incompatible with the DG's methodology because they lack a site for the installation of DG. In contrast, the employment of noncovalent bonding interactions to direct transition-metal-catalyzed C–H functionalization remarkably solves these irreconcilable conflicts. For instance, through various noncovalent bonding interactions,^{5–8} iridium-catalyzed *meta*-C–H borylation of arenes has been successfully achieved by several research groups to date.

Relative to the high reactivity of iridium catalysts, palladium-catalyzed remote C–H activation based on noncovalent bonding interactions still remains extremely challenging to date. In 2017, a class of bifunctional directing templates based on noncovalent Lewis acid/base coordination was elegantly designed by the Yu group for Pd-catalyzed distal C–H alkenylation of heteroarenes,⁹ for example, quinolines (Scheme 1a). In such a template, one palladium atom fixed in the template central scaffold is in charge of reversibly anchoring substrates by heteroatom coordination, and the other

Scheme 1. Noncovalent Bonding Strategies for Pd-Catalyzed Distal C–H Functionalization

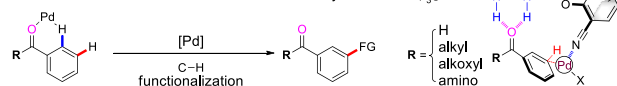
a) Pd-catalyzed remote C–H functionalization enabled by noncovalent bonding interaction



b) This work: Pd-catalyzed *meta*-C–H functionalization through hydrogen-bonding interactions

Challenges:

- ▽ competitive *ortho*-C–H activation by σ -chelation
- ▽ less reactivity of electron-poor phenyl ring
- ▽ weak coordination nitrile DG for Pd catalysts



regioselectively cleaves the distal C–H bonds by coordination to the DG. Subsequently, through utilizing symmetrically bidentate nitrile templates (Scheme 1a), the Maiti group also realized Pd-catalyzed distal C–H alkenylation and alkylation in the same heteroarenes based on the template designed by Yu's group.¹⁰ To date, the intermolecular hydrogen-bonding approach to achieve the palladium-catalyzed remote C–H

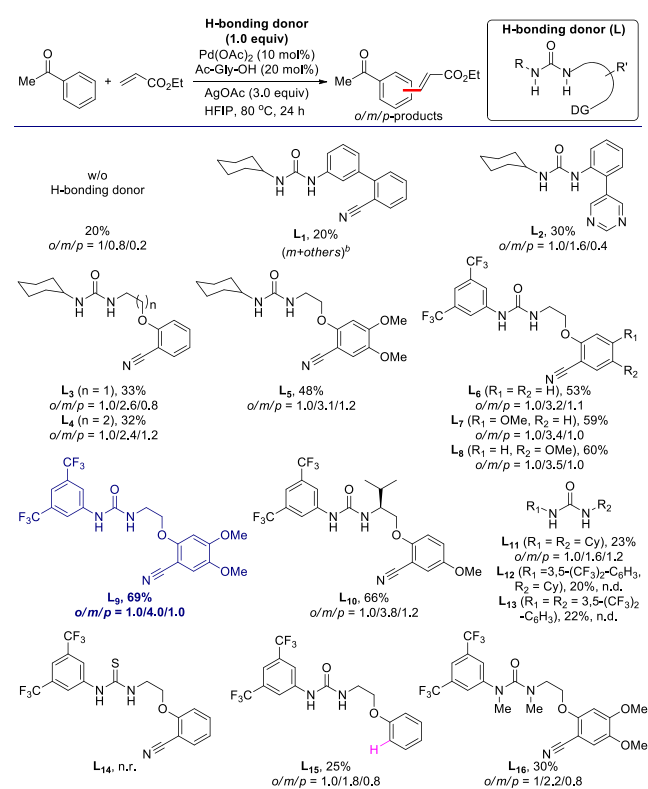
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activation of arenes is yet to be exploited probably because of the perceived lower reactivity. To address this, we sought to investigate aromatic carbonyl compounds in palladium-catalyzed remote *meta*-C–H activation, envisaging that the carbonyl group serving as a hydrogen-bond acceptor would be able to pair with a hydrogen-bond donor bearing a suitable directing template. *N,N'*-Substituted ureas behave as the strong double hydrogen-bond donors, shown thus far to be capable of assembling a lot of guest acceptor molecules including aldehydes, ketones, esters, imines, *N*-acyliminium ions, and nitro compounds.¹¹ Herein, we report that the *N,N'*-substituted urea scaffold is highly proficient as a hydrogen-bond donor to direct palladium-catalyzed C–H functionalization to the *meta* position in a series of arene substrates bearing the carbonyl group as hydrogen-bond acceptors.

Based on the Yu's pioneering work^{3a} on remote *meta*-selective C–H functionalization directed by a covalently U-shaped template, a range of DGs for transition-metal-catalyzed distal C–H functionalization have been developed since then. Over the past few years, We,¹² Yu,¹³ Maiti,¹⁴ and others¹⁵ have identified a substituted salicylonitrile moiety as the versatile template to direct palladium-catalyzed *meta*-C–H functionalization in various arene derivatives. In approaching the development of an efficient hydrogen-bonding donor template, we began by elaborating an appropriate *N,N'*-substituted urea framework for Pd-catalyzed remote C–H functionalization of arenes, according to the experimental and computational results established for the covalently binding template-s.^{12a,13b,16} By integrating a salicylonitrile-bearing tether into the urea backbone, a powerful double hydrogen-bonding interaction appears plausible with an aromatic carbonyl substrate promoting C–H activation at the *meta* position of the phenyl ring. Therefore, we first synthesized a series of *N*-cyclohexyl ureas, L₁–L₅, containing an additional directing functional group on the other nitrogen atom (Scheme 2). Using acetophenone as a model substrate, C–H olefination only gave the poor reactivity with a mixture of *o*-, *m*-, and *p*-products (*o/m/p* = 1/0.8/0.2) in the absence of a hydrogen-bonding donor. The major *o*-olefinated product obtained is probably attributed to the natively Pd-catalyzed carbonyl-directed *ortho*-C–H activation.¹⁷ While an equivalent H-bonding donor (L₂–L₄) bearing a *meta*-selective DG was added into the reaction, the desired products were afforded with significantly improved regioselectivity, albeit in just a slightly higher yield. By modifying the salicylonitrile moiety in hydrogen-bonding donors (L₅), C–H olefination achieved the *m*-product in up to 48% yield with a regioselectivity of *o/m/p* = 1.0/3.1/1.2. Pleasingly, replacing the cyclohexyl group in the urea with a 3,5-(bis)trifluoromethyl phenyl group led to a dramatic improvement in both reactivity and regioselectivity (L₆–L₁₀). Hydrogen-bonding donor L₉ was the most effective, affording the olefinated products at 69% yield with a regioselectivity of *o/m/p* = 1.0/4.0/1.0. A possible explanation for the enhanced reactivity of hydrogen-bonding donors (L₆–L₁₀) is that a very weak, not detectable by routine spectroscopic and crystallographic methods, intramolecular C–H...O interaction between the weakly acidic *ortho*-C–H proton and the urea carbonyl group takes place,^{11a–c} which drastically increases the binding potential of hydrogen-bonding donors to guest molecules. The dimethoxy substituent on the nitrile template (L₉) affords improved reactivity and regioselectivity, probably because of the enhanced electron density of the phenyl ring which prompts the metal binding of

Scheme 2. Evaluation of Hydrogen-Bonding Donors for Pd-Catalyzed *meta*-C–H Olefination^a



^aNMR yield, and regioselectivity was determined by ¹H NMR analysis with reference to an internal standard. ^bC–H olefination occurring in the template was also observed. n.d.: regioselectivity not detected, n.r.: no reaction.

the cyano group.¹⁸ The use of symmetrical *N,N'*-dicyclohexyl, *N,N'*-diaryl ureas (L₁₁ and L₁₃), or unsymmetrical urea (L₁₂) only provided comparable yields to the blank experiment but with slightly higher *meta* regioselectivities. C–H olefination in the presence of thiourea (L₁₄) failed to give the desired products, which is certainly due to the poisoning deactivation of Pd catalysts by the coordination of sulfur atoms in the thiourea molecules. As the control experiments, removal of the nitrile group in the hydrogen-bonding donor (L₁₅ vs L₆) led to a significant decline in both reactivity and regioselectivity, indicating the essential directing effect of the nitrile group. Replacement of two hydrogen atoms on the urea N–H bonds by two methyl groups (L₁₆) also afforded a diminished yield and *meta* selectivity. As observed in the previous reports,¹⁹ as a solvent hexafluoroisopropanol (HFIP) shows a significant influence on the yield and regioselectivity of C–H olefination (for details, see Supporting Information).

It is well-established that *N,N'*-disubstituted ureas as hydrogen-bond donors may associate with guest molecules, such as triphenylphosphine oxide (TPPO), dimethyl sulfoxide (DMSO), and benzophenone,^{11b} to form 1:1 complexes through double hydrogen bonds. The solution NMR and IR analyses proved to be particularly useful for hydrogen-bonding interaction. As compared with L₉ (5.81 ppm, 7.62 ppm) in *d*-chloroform solution, the chemical shifts of two urea N–H protons move downfield to 6.62 ppm, 9.18 ppm in L₉/TPPO 1:1 complex and 6.70 ppm, 9.45 ppm in *d*₆-DMSO solution, respectively (Figure 1), consistent with the previous

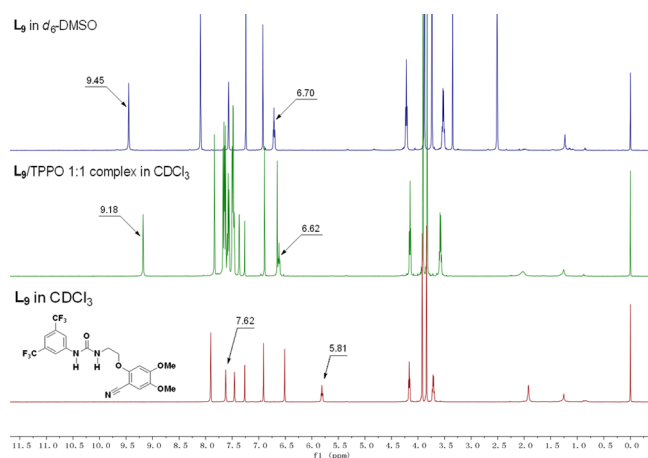


Figure 1. Chemical shift variations of urea L_9 N–H protons resulted from intermolecular H-bonding interaction.

observation by Kuninobu and Kanai.^{5a} The ^{31}P chemical shift of TPPO also moves downfield by 2.32 ppm in the complex, which is obviously resulted from deshielding by hydrogen-bonding interactions. Carbonyl stretching frequencies for L_9 are highly sensitive to hydrogen bonding, varying from 1658 cm^{-1} in L_9 to 1712 cm^{-1} in both 1:1 complexes of L_9 with TPPO and benzophenone (Figure 2). The high-frequency

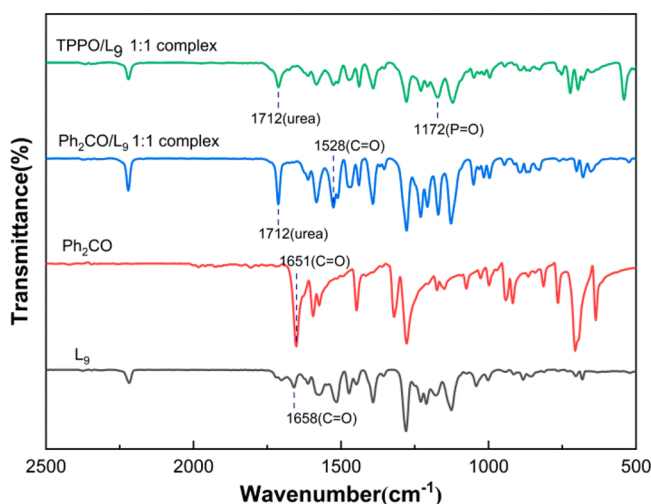


Figure 2. IR analyses of intermolecular H-bonding interaction.

bands occur in both complexes where the P=O or carbonyl oxygen forms two strong hydrogen bonds to the N–H protons of urea L_9 . Meanwhile, the carbonyl stretching frequency of benzophenone varies from 1651 to 1528 cm^{-1} in L_9 /benzophenone 1:1 complex. Similarly, urea carbonyl stretching frequencies in 1:1 cocrystals of L_9 and aromatic aldehyde, ester and amide also vary from 1658 cm^{-1} in L_9 to 1712 cm^{-1} (Figure 3). All IR and NMR data unambiguously supports the occurrence of hydrogen-bonding interaction between L_9 and hydrogen-bond acceptors.

Next, we investigate the application of hydrogen-bond donor L_9 in Pd-catalyzed *meta*-C–H olefination of aromatic carbonyl derivatives. For acetophenones and benzaldehydes (Scheme 3), the methoxyl substituents on the *ortho*, *para* positions of the phenyl ring were well tolerated, giving similar yields and excellent *meta* selectivities (3b, 3c, 3m, 4b, and 4c). Substrates

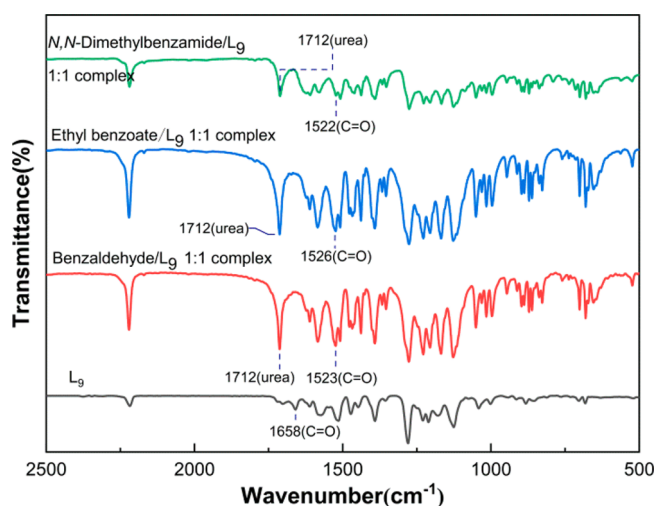
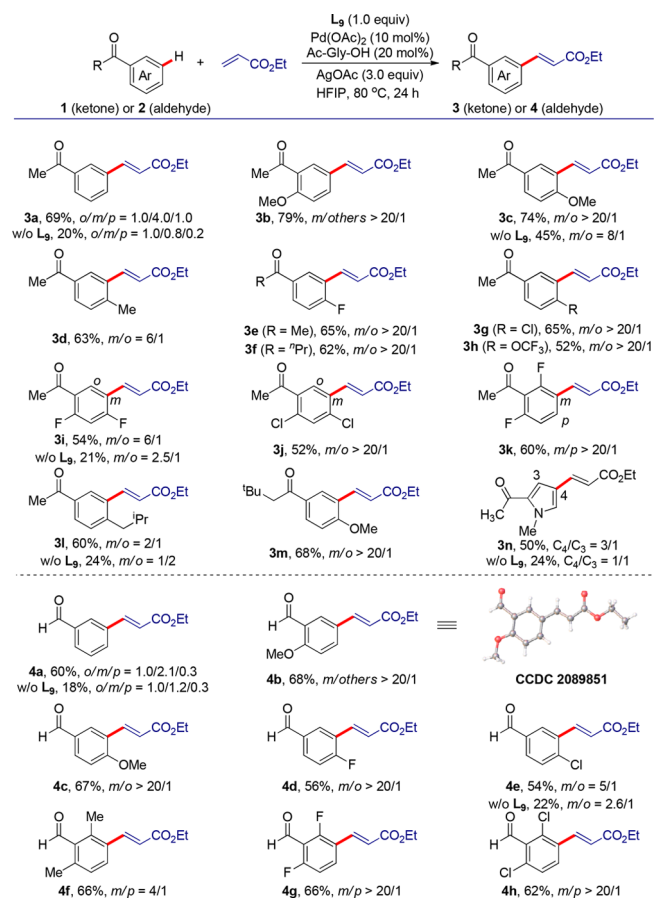


Figure 3. IR spectra of 1:1 cocrystals of L_9 and aromatic aldehyde, ester, and amide.

Scheme 3. Pd-Catalyzed *meta*-C–H Olefination of Aromatic Ketones and Aldehydes^a

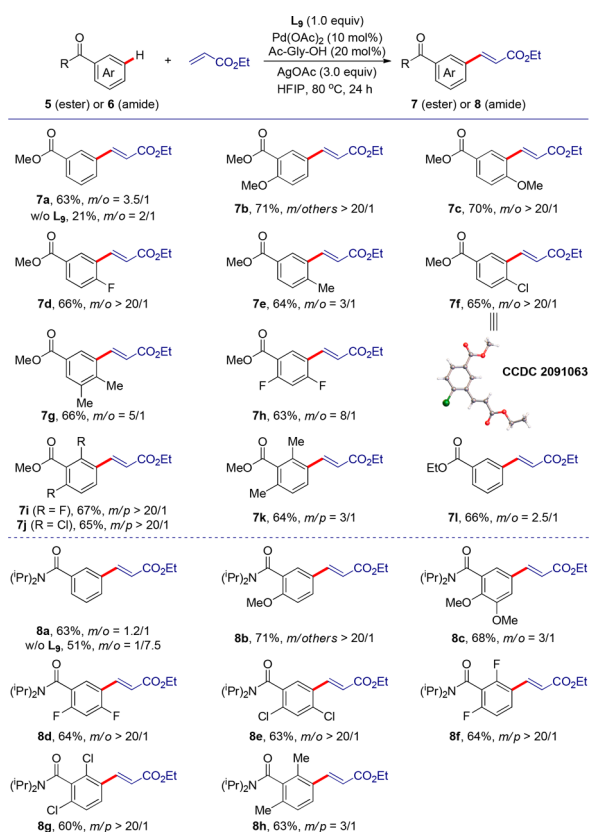


bearing electron-withdrawing fluorine, chlorine, and trifluoromethoxyl groups on the *para* position of the phenyl ring are compatible with the protocol, affording the *meta* products in yields of 52%–65% with regioselectivities of up to >20/1 (3e–3h, 4d, and 4e). C–H olefination proceeds smoothly with the substrates owning difluoro or dichloro groups in different substitution patterns to provide the desired *meta* products in

52%–66% yields (**3i**–**3k**, **4g**, and **4h**). Compared with 2,4-dichloro benzophenone (**3j**), 2,4-difluoro benzophenone (**3i**) gave rise to a decreased *meta* selectivity. Notably, *N*-methyl-2-acetyl pyrrole (**3n**) is also compatible with the protocol, providing the olefinated product in 50% yield with a regioselectivity of $C_4/C_3 = 3/1$. In general, a *meta* group on the phenyl ring led to the drastic decrease of *meta* selectivity in the C–H olefination process. This phenomenon was previously observed in template-enabled *meta*-C–H olefination of 3-arylpyridines by Yu,⁵ probably attributed to the *meta* substitution in substrates interfering with the optimal conformational orientation because of steric repulsion. Besides acetophenones, other aromatic ketones, such as **3f** and **3m**, are also suitable substrates for the present protocol. The presence of a bulky substituent like the methyl (**3d**) and *iso*-butyl (**3l**) group led to the obvious decline of product regioselectivity due to the steric effect. Owing to the enhanced electron-deficient nature, the substrates bearing an additionally electron-withdrawing group such as NO_2 , CF_3 , and ester group only provided a trace amount of olefinated products.

Besides ketones and aldehydes, we also examined the generality of hydrogen-bond donor L_9 in Pd-catalyzed *meta*-C–H olefination of benzoate esters and benzamides (Scheme 4). Similar to ketones and aldehydes, the methoxyl group at either *ortho* or *para* position of the phenyl ring was reactive to provide the olefination products in more than 70% isolated yields and excellent regioselectivities (**7b**, **7c**, and **8b**). Substrates incorporating difluoro or dichloro groups in different substitution patterns are compatible with the

Scheme 4. Pd-Catalyzed *meta*-C–H Olefination of Benzoate Esters and Benzamides^a

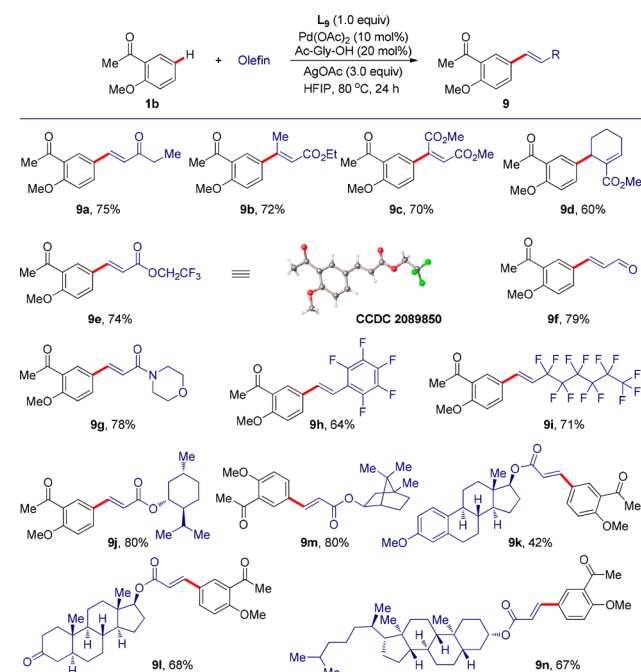


^aIsolated yield.

protocol, giving rise to the desired olefinated products in 60%–67% yields with regioselectivities of 8/1 to $>20/1$ (**7h**–**7j** and **8d**–**8g**). In addition to methyl ester, other alkyl esters are also suitable substrates in C–H olefination, albeit in decreased *meta* selectivity (**7l**). For benzamide substrates (**8a**), in the absence of L_9 , *ortho*-olefinated product was obtained as the major product ($o/m = 7.5/1$) due to the proximal σ -chelation of the amide group. In contrast, regioselectivity of C–H olefination was improved to $o/m = 1/1.2$ with L_9 as a hydrogen-bond donor.

The scope of C–H olefination was further studied using different olefin partners with acetophenone **1b** (Scheme 5). To

Scheme 5. Scope of the Olefins^a



^aIsolated yield.

our delight, in the presence of hydrogen-bond donor L_9 , *meta*-selective C–H olefination was shown to be compatible with a diverse set of olefins containing ketones (**9a** and **9l**), esters (**9c**), aldehydes (**9f**), amides (**9g**), multiple-fluoro substituted aryl and alkyl (**9e**, **9h**, and **9i**) functional groups. The diverse olefins derived from naturally occurring terpenes were also well tolerated in the protocol (**9j**–**9n**).

To elucidate the roles of hydrogen-bond donors in the reaction, we further carried out three sets of parallel experiments (Figure 4). In the absence of hydrogen-bond donor L_9 , the olefination reaction gave no more than 20% of the olefinated products **3a** after 20 h, while under the same conditions, close to 65% of the products were obtained with 1.0 equiv of L_9 . Both the reactivity and *meta* selectivity (53% yield, $o/m/p = 1/3.3/1$) declined obviously when reducing the loadings of hydrogen-bond donor L_9 to 0.5 equiv. It was found that with a loading of 0.5 equiv L_9 , C–H olefination products were provided in slightly higher yield (58% yield) when further extending the reaction time to 48 h. In regard to *meta*-C–H olefination products, kinetic isotope effect (KIE) was also measured, and the value of $k_{\text{H}}/k_{\text{D}}$ was 3.93 (Scheme 6), indicating that C–H-bond cleavage is probably the rate-determining step as observed in the previous reports.^{12,13b}

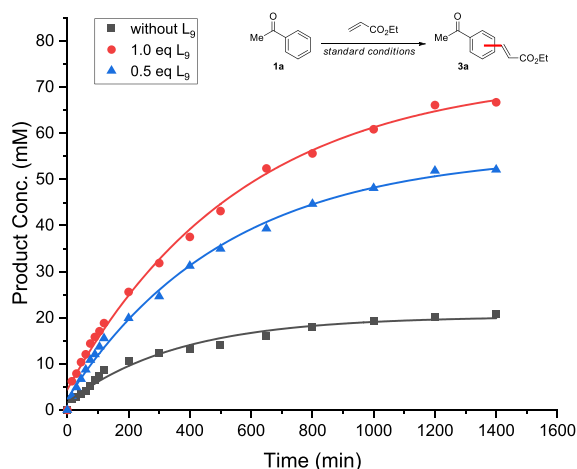
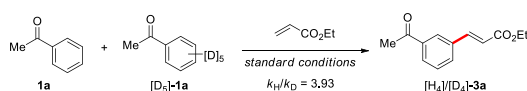


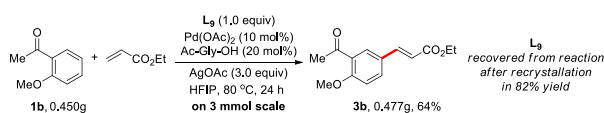
Figure 4. Effect of H-bonding interaction on C–H olefination.

Scheme 6. KIE Experiment



Finally, hydrogen-bonding donor enabled C–H olefination was carried out on a 3 mmol scale to provide the *meta*-olefinated product **3b** in 64% isolated yield, accompanied by 82% recovery of hydrogen-bond donor **L₉** from the reaction (Scheme 7).

Scheme 7. Up-Scaled Reaction and Recovery of H-Bonding Donor



In conclusion, we have described the first example of Pd-catalyzed *meta*-selective C–H functionalization of aromatic carbonyl compounds through noncovalent hydrogen-bonding interaction. The *N,N'*-substituted ureas as hydrogen-bond donors were demonstrated to be capable of forming the double hydrogen bonds with aromatic ketones, aldehydes, benzoate esters, and benzamides, thereby achieving C–H functionalization on the *meta* position of the phenyl ring by an embedded nitrile directing group. We believe that the present noncovalent hydrogen-bonding strategy for Pd-catalyzed distal C–H bond functionalization will prove to be applicable in other types of substrates and/or reactions.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acscatal.1c02974>.

General information, experimental procedures, analytical data, and NMR spectra for products (PDF)

AUTHOR INFORMATION

Corresponding Authors

Zhong Jin – State Key Laboratory of Elementoorganic Chemistry, College of Chemistry, Nankai University, Tianjin 300071, China; Laboratory of Xinjiang Native Medicinal and Edible Plant Resource Chemistry, College of Chemistry and Environmental Science, Kashi University, Kashgar 844007, China; orcid.org/0000-0001-5722-7175; Email: zjin@nankai.edu.cn

Xiaohua Xu – State Key Laboratory of Elementoorganic Chemistry, College of Chemistry, Nankai University, Tianjin 300071, China; Email: xiaohuaxu@nankai.edu.cn

Authors

Guoshuai Li – State Key Laboratory of Elementoorganic Chemistry, College of Chemistry, Nankai University, Tianjin 300071, China

Yifei Yan – State Key Laboratory of Elementoorganic Chemistry, College of Chemistry, Nankai University, Tianjin 300071, China

Pengfei Zhang – State Key Laboratory of Elementoorganic Chemistry, College of Chemistry, Nankai University, Tianjin 300071, China

Complete contact information is available at:

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Notes

The authors declare no competing financial interest.

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DEDICATION

Dedicated to the 100th anniversary of chemistry at Nankai University and 60th anniversary of Kashi University.

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