A New And Mild Procedure For The Preparation Of Vinyl Formamides From Thiooximes

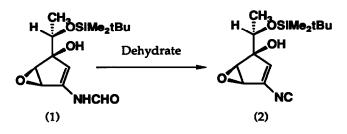
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<u>Summary</u>: Treatment of thiooximes with triphenylphosphine and acetic formic anhydride in dichloromethane at room temperature gives good yields of vinyl formamides under essentially neutral conditions.

Isonitrins A,B and C are isonitrile containing natural products isolated from fungi of the genus trichoderma¹. Their unusual and complex molecular structure coupled with their antimicrobial activity promted us to undertake their total synthesis².



In our approach to the total synthesis of deoxytrichoviridin we planned to prepare and dehydrate the vinyl formamide (1) as the key step. (scheme 1) We anticipated that the vinyl Scheme 1

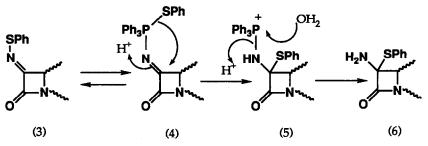


formamide (1) would be extremely sensitive to acid³ and as there were no methods for the

preparation of vinyl formamides under mild conditions at the outset of this work⁴ we decided to investigate alternative procedures for their synthesis. We noted that in a related synthetic problem,

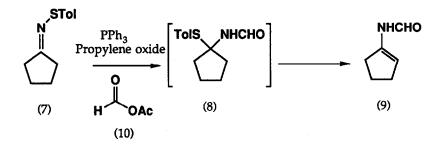
Gordon and co-workers⁵ had converted the cephalosporin thiooxime (3) to an α -thioamine (6) by treatment with PPh₃ and silica gel. He proposed the intermediacy of (4) which underwent migration of the thiophenyl to the imine carbon followed by protonation and loss of triphenylphosphine oxide to give the product (6).(scheme 2)

Scheme 2



We speculated that by replacing the silica gel with a formyl source that the product

would be an α -thioformamide, which would undergo loss of thiotolyl to give the vinyl formamide (scheme 3) Scheme 3



We tested our conjecture and found that vinyl formamides are isolated using this modified procedure. Experimentally the reaction consists of treating the cyclopentanone thiooxime (7) with PPh₃ (3eq.), acetic formic anhydride⁶ (3eq.) and propylene oxide (5eq.) (as an acid scavenger), and stirring at room temperature under an atmosphere of argon for 18 hours. The crude reaction mixture is washed with 5% aq. NaHCO₃, dried (Na₂SO₄) and then purified by chromatography on basic alumina to give the vinyl formamide (9) in 68%. We have assumed that

the α -thioformamide (8) is formed, although alternative intermediates are possible. This methodology has been applied to a variety of thiooximes and we have found it to a be very general procedure for the preparation of vinyl formamides under mild conditions. (see table). The precursor thiooximes are conveniently available from either the corresponding ketones⁷ or amines⁸ in one step transformations using readily available starting materials. The isolation of the less substituted

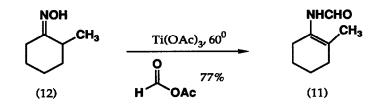
<u>Table</u>

a) Purified by recrystallization from pentane at -30°. b) Purified by chromatography on flash silica.

	<u> </u>	% YIELD		% YIELD
	THIOOXIME	THIOOXIME	VINYL FORMAMIDE	VINYL FORMAMIDE
1	NSTOI	89 ^a	МНСНО	68
2	NSTO	88 ^a	МНСНО	76
3		94 ^b		58
4	Ph	89 ª	Рһ	80
5	NSTol CH ₃	89 ^a	Вг Инсно	81
6		92 ^a	СІ	86
7	MeO NSTOI	96 ^a	мео	85
8	NSTOI CH3	66 ^b	ис инсно	68
9	NSTOI CH3	69ª	ЛНСНО	86

vinyl formamide in case 3 is noteworthy, giving a complimentary result to a recently published route to vinyl formamides⁴ in which only the more thermodynamically st²ble vinyl formamide (11) was isolated on treating oxime (12) with Ti(OAc)₃ and acetic formic anhydride. (scheme 4) We are currently investigating this unusual result further.

Scheme 4



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