

# Preparation of Thiazole-2-thiones through TBPB-Promoted Oxidative Cascade Cyclization of Enaminones with Elemental Sulfur

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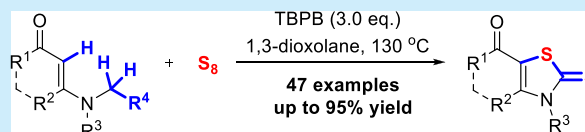


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**ABSTRACT:** An unprecedented method for the construction of thiazole-2-thiones via an oxidative cascade cyclization strategy is described. The novel protocol involves the simultaneous formation of two C—S bonds and a C=S bond on the structure of enaminones in a single operation through a cascade of C(sp<sup>2</sup>)—H/C(sp<sup>3</sup>)—H bond sulfurations and C(sp<sup>3</sup>)—H bond thiocarbonylation. This transformation allows for the efficient synthesis of thiazole-2-thiones with broad tolerance in moderate to excellent yields from simple enaminones with elemental sulfur.



Thiazoles belong to an important class of five-membered heterocyclic skeletons which ubiquitously occur in natural products,<sup>1</sup> bioactive molecules,<sup>2</sup> and functional materials,<sup>3</sup> as well as being widely used as synthetic intermediates in organic synthesis.<sup>1a,4</sup> Among them, thiazolone derivatives, especially thiazole-2-thiones, have widespread applications as valuable synthetic blocks that exhibit a variety of biological activities in drug discovery (Figure 1).<sup>2a,5</sup>

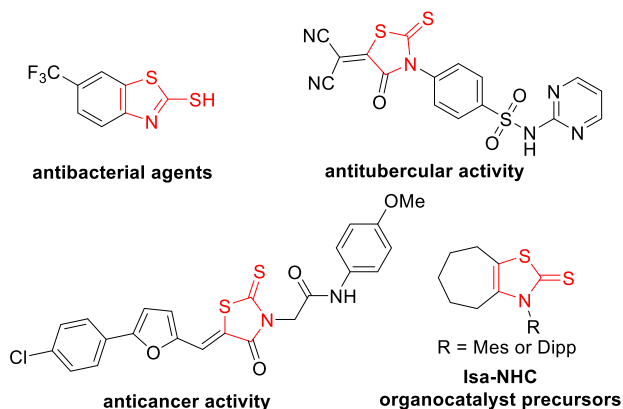


Figure 1. Selected examples of bioactive thiazole-2-thione molecules.

Furthermore, thiazole-2-thiones frequently serve as important synthetic precursors in organic synthesis (Figure 1).<sup>6</sup> Therefore, numerous synthetic methods have been developed toward the construction of structurally diverse thiazole-2-thiones.

Among all of the synthetic approaches for thiazole-2-thiones, the vast majority of applications are accomplished through three conventional synthetic strategies (Scheme 1a): (1) thiocarbonylation of thiazole-2-ones via Lawesson's reagent;<sup>5a,7</sup> (2) cyclocondensation of isothiocyanates with cyanide compounds;<sup>5b,8</sup> and (3) cyclization of *o*-haloanilines with

dithiocarbamate salts.<sup>5d,e,9</sup> Recently, Halimehiani and co-workers reported a facile and efficient method for the synthesis of thiazole-2(3*H*)-thiones from nitroepoxides and *in situ* generated dithiocarbamates in water (Scheme 1b).<sup>10</sup> Additionally, Liang and co-workers developed two elegant strategies for the synthesis of benzothiazolethiones through a three-component reaction of *o*-iodoanilines and K<sub>2</sub>S with two different carbon sources in the presence or absence of CuCl, respectively (Scheme 1c).<sup>11</sup> However, most of these approaches suffer from some limitations, such as restricted starting materials, severe preparative limitations, harsh reaction conditions, and poor efficiencies, thereby restricting their utilities. Therefore, more efficient and practical methods for the synthesis of thiazole-2-thiones from simple and readily available starting materials are of great value.

New C—C/C—X bond formation via direct transformations of inert C—H bonds, especially C(sp<sup>3</sup>)—H bonds with poorer acidity and higher bond energy, has emerged as a powerful strategy for the construction of valuable molecules.<sup>12</sup> On the other hand, enaminone chemistry has increasingly attracted the interest of synthetic chemists because of versatile reactive sites in their structures as building blocks for organic synthesis, which allows the construction of valuable heterocycles with structural diversity.<sup>13</sup> Direct transformations of C(sp<sup>3</sup>)—H bonds in the enaminone structure have greatly enriched the diversity of enaminone-involved transformations. However, reports to describe the use of this strategy are scarce,<sup>14</sup> and this transformation involving the formation of

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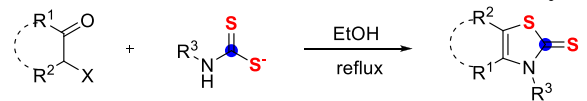
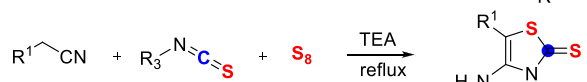
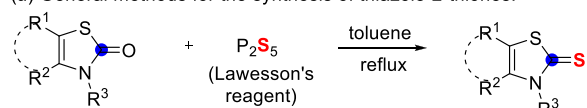
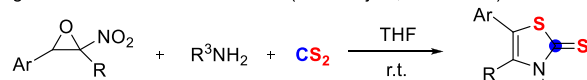
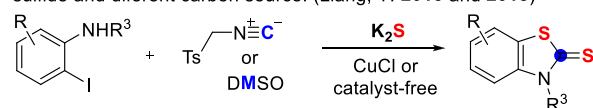
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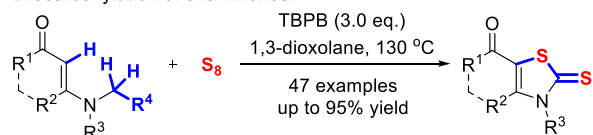
## Scheme 1. Synthetic Methods of Thiazole-2-thiones

## Previous work:

(a) General methods for the synthesis of thiazole-2-thiones.

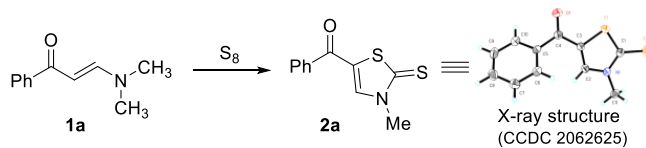
(b) Thiazole-2-thiones synthesis from nitroepoxides and *in situ* generated dithiocarbamic acids. (Halimehiani, A. Z. 2017)(c) Benzothiazolethiones synthesis from *o*-iodoanilines, potassium sulfide and different carbon source. (Liang, Y. 2015 and 2018)

## This work:

(d) Thiazole-2-thiones synthesis by *in situ* sulfur cyclization and thiocarbonylation of enaminones.

multiple new bonds in enaminones still remains with considerable challenges that lie ahead. For example, C(sp<sup>3</sup>)-H bonds on the aliphatic amine group in the enaminone structure could be achieved by this transformation to construct heterocycles, through inserting a heteroatom into the higher-electron-density double bonds of the  $\alpha$ -carbon.<sup>14a-d</sup> With our continued interest in enaminone chemistry and direct transformations of C(sp<sup>3</sup>)-H bonds,<sup>15</sup> herein, we first report a new approach for the construction of thiazole-2-thiones through *tert*-butyl peroxybenzoate (TBPB)-promoted oxidative cascade cyclization of enaminones with elemental sulfur. In this transformation, two C-S bonds and a C=S bond are efficiently formed via a sequenced C(sp<sup>2</sup>)-H/C(sp<sup>3</sup>)-H bond sulfuration, cyclization, and C(sp<sup>3</sup>)-H thiocarbonylation process under metal-free conditions.

We started our investigation by choosing enaminones **1a** and elemental sulfur as model substrates. To our delight, the desired thiazole-2-thione **2a** was obtained in 83% yield in the presence of dicumyl peroxide (DCP) with 4.0 equiv in 1,3-dioxolane at 130 °C for 12 h (Table 1, entry 1), and its structure was unambiguously confirmed by X-ray crystallographic analysis (CCDC 2062625). Other solvents were screened, and it was revealed that 1,4-dioxane, acetonitrile, toluene, dimethyl sulfoxide (DMSO), and *N,N*-dimethylformamide (DMF) resulted in a lower yield than in 1,3-dioxolane (entries 2–6), while dichloromethane (DCM) and tetrahydrofuran (THF) gave unsuccessful results (entries 7 and 8). Next, a handful of oxidants were examined. *tert*-Butyl hydroperoxide (TBHP), di-*tert*-butyl peroxide (DTBP), and cumyl hydroperoxide (CHP) proved to be ineffective for this cyclization reaction (entries 9–11). Dibenzoyl peroxide

Table 1. Optimization of Reaction Conditions<sup>a,b</sup>

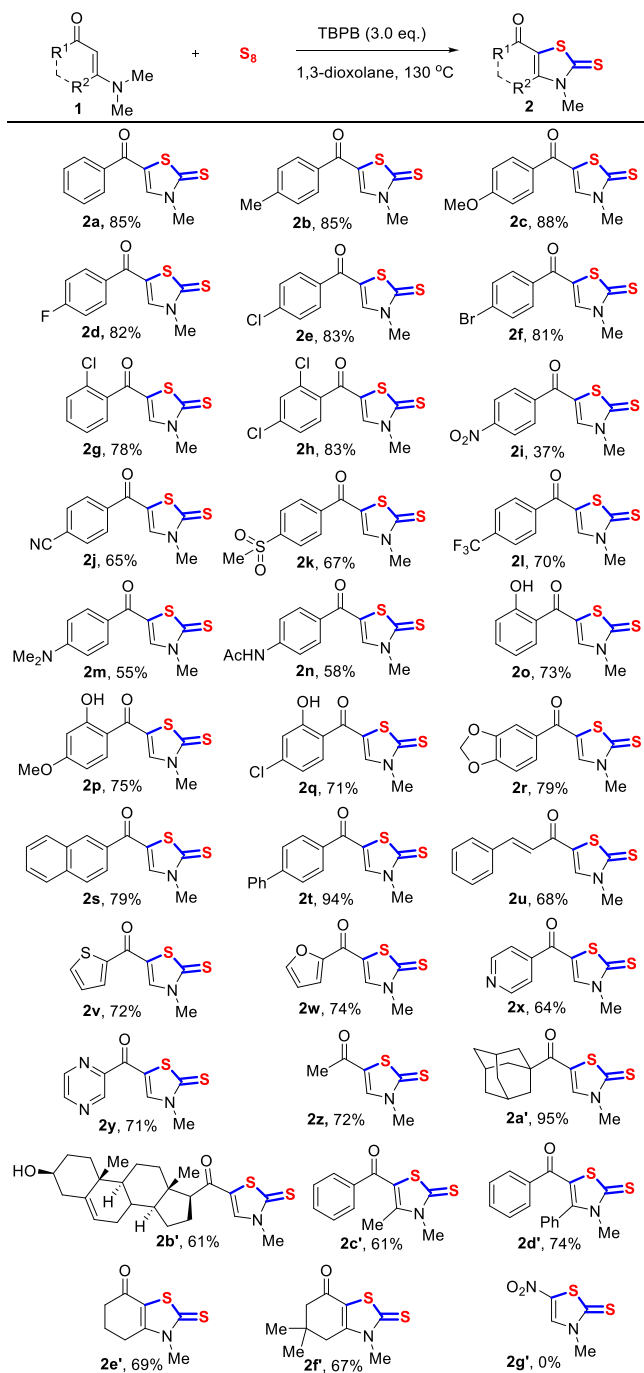
entry	oxidant	solvent	T (°C)	time (h)	yield (%)
1	DCP (4 equiv)	1,3-dioxolane	130	12	83
2	DCP (4 equiv)	1,4-dioxane	130	12	55
3	DCP (4 equiv)	acetonitrile	130	12	52
4	DCP (4 equiv)	toluene	130	12	23
5	DCP (4 equiv)	DMSO	130	12	27
6	DCP (4 equiv)	DMF	130	12	36
7	DCP (4 equiv)	DCM	130	12	n.r. <sup>c</sup>
8	DCP (4 equiv)	THF	130	12	n.r.
9	TBHP (4 equiv)	1,3-dioxolane	130	12	n.r.
10	DTBP (4 equiv)	1,3-dioxolane	130	12	n.r.
11	CHP (4 equiv)	1,3-dioxolane	130	12	n.r.
12	BPO (4 equiv)	1,3-dioxolane	130	12	65
13	TBPB (4 equiv)	1,3-dioxolane	130	3	84
14	TBPB (4 equiv)	1,3-dioxolane	120	12	47
15	TBPB (4 equiv)	1,3-dioxolane	140	3	83
16	TBPB (2 equiv)	1,3-dioxolane	130	12	65
17	TBPB (3 equiv)	1,3-dioxolane	130	3	85
18	TBPB (5 equiv)	1,3-dioxolane	130	3	83

<sup>a</sup>Reaction conditions: **1a** (0.5 mmol), S<sub>8</sub> (1.5 mmol), oxidant in 2 mL of solvent. <sup>b</sup>Isolated yields. <sup>c</sup>n.r.: no reaction.

(BPO) afforded a low yield (entry 12). Gratifyingly, *tert*-butyl peroxybenzoate (TBPB) resulted in a sharp decrease of the reaction time, but the yield of **2a** was not affected (entry 13). When the reaction temperature was decreased to 120 °C, a significant reduction in the yield of product **2a** was observed (entry 14). Increasing the reaction temperature did not lead to any improvement (entry 15). Additionally, the effect of different oxidant loading was investigated, with 3.0 equiv found to give the best result in 85% yield (entries 16–18).

Under the optimized reaction conditions (Table 1, entry 17), various enaminone derivatives were investigated to expand the substrate scope of the TBPB-mediated cascade cyclization, and all results are summarized in Schemes 2–4.

Generally, when enaminones bearing *N,N*-dimethyl were examined in this transformation, the desired thiazole-2-thione products were obtained in satisfying yields (Scheme 2). A series of *N,N*-dimethyl enaminones containing either electron-donating (EDG) or electron-withdrawing (EWG) groups or halogen on the phenyl rings were well-tolerated, (**2a–2q**), though the nitro group afforded the desired product **2i** in slightly lower yield. Notably, the phenyl ring bearing free hydrogen groups, such as OH and AcNH, also successfully resulted in good reactivity and gave the desired products in 58–75% yields (**2n–2q**). Furthermore, enaminones with sterically hindered polycyclic and conjugated moieties, such as 2-naphthyl, 4-biphenyl, and styryl, were also compatible with this transformation, providing the corresponding products **2r–2u** in good to excellent yields. Satisfactorily, enaminones with heteroarene moieties, such as 2-thienyl, 2-furyl, 4-pyridyl, and 2-pyrazinyl, did not have an adverse effect on the reaction (**2v–2y**). To our delight, aliphatic enaminones, including methyl and 1-adamantyl, were successfully converted to the desired products **2z–2a'** in excellent yield. Importantly, to further explore the potential applicability of this method to the

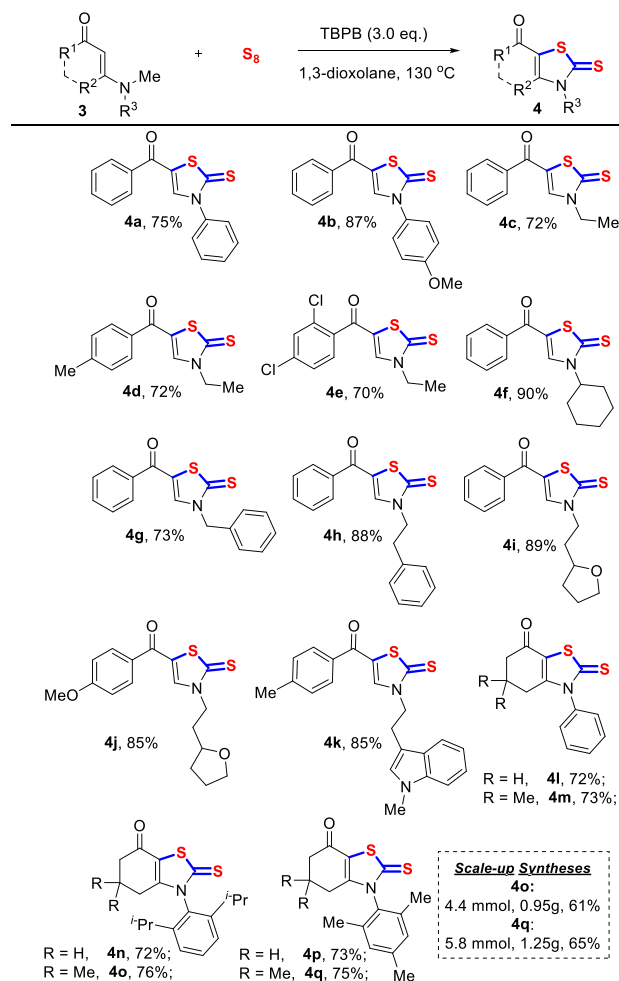
Scheme 2. Substrate Scope of *N,N*-Dimethyl Enaminones<sup>a,b</sup>

<sup>a</sup>Reaction conditions: **1** (0.5 mmol),  $S_8$  (1.5 mmol), and TBPB (3.0 equiv) in 2 mL of 1,3-dioxolane, stirred at 130 °C for 3.0 h. <sup>b</sup>Isolated yields.

late-stage modification of complex molecules, we prepared a natural product-derived pregnenolone **1b'**, which reacted with elemental sulfur under the standard conditions to obtain the pregnenolone derivative **2b'** in moderate yield. Next, we examined the effect at the  $\beta$ -position substituents of enaminones in this transformation, affording the desired thiazole-2-thione products **2c'** and **2d'** in good yields, separately. Finally, the cascade cyclization of cyclic enaminones were carried out identically well to give the corresponding cyclic thiazole-2-thiones products **2e'** and **2f'** in moderate

yields. However, with the nitro instead of the ketone of enaminones, the corresponding target product **2g'** was not observed.

After successful investigation of *N,N*-dimethyl enaminones, we wish to examine the utility of enaminones containing a wide variety of amine groups in our protocol (Schemes 3 and 4).

Scheme 3. Substrate Scope of *N*-Methyl-*N*-substituted Enaminones<sup>a,b</sup>

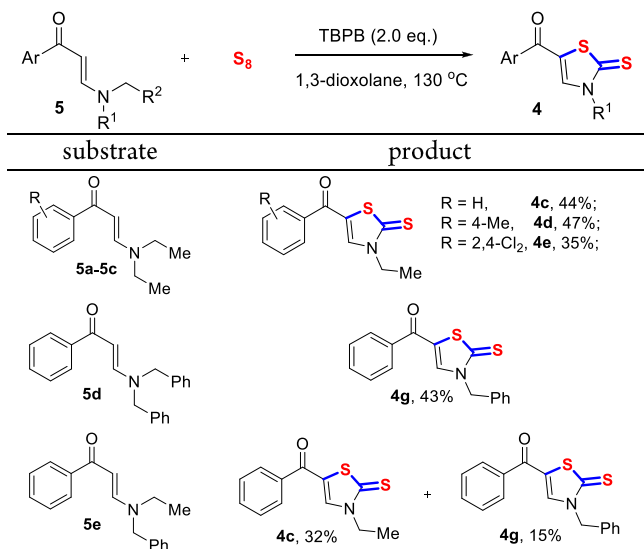
<sup>a</sup>Reaction conditions: **3** (0.5 mmol),  $S_8$  (1.5 mmol), and TBPB (3.0 equiv) in 2 mL of 1,3-dioxolane, stirred at 130 °C for 3.0 h. <sup>b</sup>Isolated yields.

Accordingly, the scope and limitations of enaminones bearing an *N*-methyl group were evaluated (Scheme 3). Similar to the above-mentioned cascade cyclization reaction, both the *N*-aryl and *N*-alkyl group on the amine moiety were well-tolerated under optimal reaction conditions, affording the desired products (**4a**–**4k**) in good to excellent yields. Notably, when the amine moieties were all *N*-alkyl groups, including two  $C(sp^3)$ –H bond sulfuration reaction sites, this transformation was regioselectively carried out on the *N*-methyl group, due to the lower  $C(sp^3)$ –H bond strength of *N*-methyl than the  $C(sp^3)$ – $C(sp^3)$  bond of another *N*-alkyl group in the final thiocarbonylation process. Moreover, the *N*-aryl groups on cyclic enaminones were well-tolerated, giving the corresponding cyclic thiazole-2-thiones products (**4l**–**4q**) in moderate yields. To establish the feasibility of this approach for practical applications, scale-up reactions were then carried out,

delivering the expected products **4o** (0.95 g) and **4q** (1.25 g) in 61% yield and 65% yield, respectively.

Considering the potential for the selective cleavage of two C(sp<sup>3</sup>)-C(sp<sup>3</sup>) bonds in the final thiocarbonylation process of enamines bearing two *N*-alkyl groups, we next turned our attention to the compatibility and competitiveness of this transformation (Scheme 4). To our delight, enamines

#### Scheme 4. Substrate Scope of *N,N*-Disubstituted Enaminones<sup>a,b</sup>

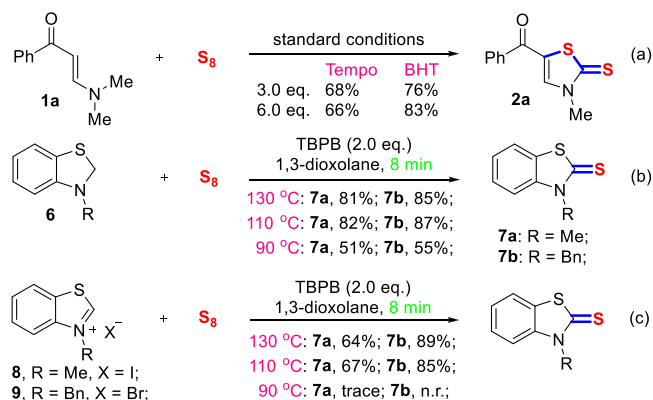


<sup>a</sup>Reaction conditions: **5** (0.5 mmol), S<sub>8</sub> (1.5 mmol), and TBPB (3.0 equiv) in 2 mL of 1,3-dioxolane, stirred at 130 °C for 3.0 h. <sup>b</sup>Isolated yields.

bearing two *N*-ethyl or *N*-benzyl groups participated in this cascade cyclization smoothly to provide the desired products (**4c–4e** and **4g**). Interestingly, this cascade cyclization reaction achieving on *N*-ethyl was superior to that with the *N*-benzyl group, when the amine moiety of enamines **5e** were *N*-ethyl and *N*-benzyl, providing **4c** and **4g** in 32% and 15% yields, respectively. It should be stated that the relatively low yields of these desired products were mainly due to the need for the cleavage of the C(sp<sup>3</sup>)-C(sp<sup>3</sup>) bond on the *N*-alkyl group during the transformation. In addition, the reaction of elemental sulfur with in situ prepared enamines or enaminones was also investigated, but the corresponding target products were not observed (see the Supporting Information).

To elucidate the reaction mechanism of this cascade cyclization reaction, a series of control experiments were performed, as shown in Scheme 5. When 3.0 or 6.0 equiv of the radical inhibitors 2,2,6,6-tetramethylpiperidin-1-oxyl (TEMPO) or 2,6-di-*tert*-butyl-4-methylphenol (BHT) was added under the standard conditions, the reaction was almost unaffected (Scheme 5a). These results ruled out the possibility that this transformation was unlikely to involve a radical pathway. Subsequently, the thiocarbonylation of *N*-substituted 2,3-dihydrobenzo[*d*]thiazoles **6** with sulfur element under the optimal conditions in only 8 min afforded *N*-substituted benzo[*d*]thiazole-2(3*H*)-thione **7a** and **7b** in 81% yield and 85% yield, respectively. They could still proceed smoothly in 8 min when these reactions were carried out at 110 °C, but their yield decreased significantly at 90 °C (Scheme 5b). Unfortunately, this analogue could not be found in our

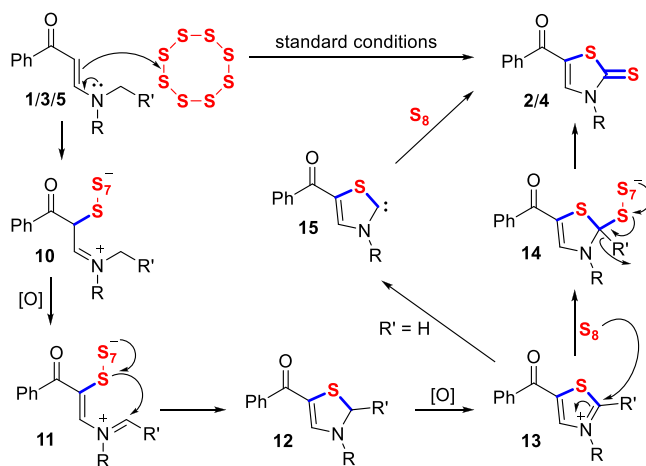
#### Scheme 5. Control Experiments



reaction system, which may not be stable because of the high reactivity of such intermediates. Furthermore, similar to the above-mentioned thiocarbonylation of *N*-substituted 2,3-dihydrobenzo[*d*]thiazoles, *N*-substituted benzo[*d*]thiazol-3-iums **8** and **9** worked well in 8 min at 130 and 110 °C, giving the expected products **7a** and **7b** in moderate to excellent yields (Scheme 5c). These results suggested that the thiocarbonylation process might not be a speed-determining step in the transformation, and the carbene intermediate might be formed in the thiocarbonylation process.

On the basis of the above-mentioned control experiments and previous studies of C(sp<sup>3</sup>)-H bond transformation<sup>14</sup> and C-S bond formation<sup>16</sup> in enamine chemistry, we proposed a plausible mechanism for this TBPB-promoted oxidative cascade cyclization (Scheme 6). Initially, sulfurated imine ion

#### Scheme 6. Possible Mechanism



species **10** was generated via the tautomerization of enamines **1/3/5** and the nucleophilic attack of elemental sulfur (S<sub>8</sub>), which undergo subsequent oxidation with oxidant to afford another imine ion species **11**.<sup>17a,b</sup> An intramolecular-cyclization on the *N*-alkyl group took place by the formation of a new C(sp<sup>3</sup>)-S bond and delivered 4-thiazoline intermediate **12** with the release of sulfur (S<sub>7</sub>). Thiazolinium intermediate **13** was produced after intermediate **12** proceeded in the second oxidation. Intermediate **13** proceeded in the second nucleophilic attack of S<sub>8</sub> and formed intermediate **14**. Finally, the desired products **2/4** would be obtained by the elimination of intermediate **14**. Furthermore, a simultaneous process

promoted by thiazole carbenes **15** to generate the desired products **2/4** was also possible.<sup>17c–e</sup>

In summary, we have developed a novel TBPB-promoted oxidative cascade cyclization strategy for the efficient construction of thiazole-2-thione skeletons from enaminones and elemental sulfur in moderate to excellent yields. The protocol includes C(sp<sup>2</sup>)-H/C(sp<sup>3</sup>)-H bond sulfuration of enaminones between the alkenyl and *N*-alkyl moieties, and the subsequent C(sp<sup>3</sup>)-H thiocarbonylation of the *N*-alkyl moieties, allowing for ready access to thiazole-2-thiones with a broad scope. Further efforts on the synthetic application of this strategy are ongoing in our lab.

## ■ ASSOCIATED CONTENT

### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c00751>.

Experimental procedures, characterization data, and copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra (PDF)

### Accession Codes

CCDC 2062625 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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### Notes

The authors declare no competing financial interest.

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