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Efficient Synthesis of (*S*)-(+)-Clopidogrel Bisulfate-Capped Silver Nanoparticles

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In this manuscript primarily one-pot synthetic development in the preparation of clopidogrel
bisulfate with a polymorphic crystalline form II in 90% yield was developed. This premade
antiplatelet drug has been used to protect starch-stabilized silver nanoparticles (AgNPs).

Keywords

(*S*)-(+)-clopidogrel bisulfate, silver nanoparticles, functionalized silver nanoparticles

Introduction

(S)-(+)-Clopidogrel bisulfate **1** (CLP) (Fig. 1), is a thienopyridine derivative as a potent oral antiplatelet agent often used in the treatment of coronary artery, peripheral vascular and cerebrovascular diseases.^[1] This drug is one of the biggest selling drugs in the world. It can crystallize in various polymorphic crystalline forms and amorphous forms, but only forms I and II are used in pharmaceutical products.^[2,3]

Nanoparticles (NPs) are promising as talented candidates for various biomedical applications such as enhanced resolution magnetic resonance imaging (MRI), drug delivery, tissue repair, cell and tissue targeting and transfection.^[4-20] Silver nanoparticles (AgNPs) are a significant class of nanomaterial for a wide range of industrial and biomedical applications. The unique chemical properties of AgNPs make it a promising targeted delivery approach for drugs or gene specific cells. AgNPs also effectively inhibited integrin-mediated platelet functional responses like aggregation, secretion, adhesion to immobilized fibrinogen or collagen and retraction of fibrin clot in a dose-dependent manner, irrespective of the nature of agonists used.^[21]

With increase use of AgNPs as drug carriers and their accepted antiplatelet property, it would be interesting to investigate its interaction with CLP. Here, synthetic improvement in the preparation of CLP also is described.

Experimental

Material and Measurements

AgNO₃ (99.9%), paraformaldehyde (PFA, 99%), *L*-camphorsulfonic acid (*L*-CSA), soluble starch and NaBH₄ (99%) were purchased from Merck. The FT-IR spectra for the samples were obtained using Shimadzu FT-IR-8900 spectrophotometer by using KBr pellets. All NMR data

were recorded in DMSO- d_6 using Bruker Avance 500 MHz spectrometer. Chemical shifts are reported in ppm (δ) using deuterated solvents as internal references. Elemental analyses were made by a Carlo-Erba EA1110 CNNO-S analyzer and agreed with the calculated values. UV-Vis absorption spectra in the range 200-500 nm in EtOH were measured with a Shimadzu UV-2100 spectrophotometer. Melting points are uncorrected and determined by electrothermal 9100 melting point apparatus. The X-ray diffraction (XRD) patterns were recorded in a wide angle range ($2\theta = 10-70$) by Phillips (pw-1840) X-ray diffractometer with Cu-K α radiation. The morphology and particle sizes of synthesized powder were characterized by transmission electron microscope (TEM) images on a Phillips CM-10 instrument with an accelerating voltage of 100 kV.

One-pot procedure for the synthesis of racemic clopidogrel bisulfate (2)

To a vessel equipped with Dean-Stark assembly, 2-(thiophen-2-yl)ethanamine **5** (0.48 g, 4 mmol), PFA (0.13 g, 4.3 mmol) and toluene (6 mL) were added and the reaction mixture was refluxed for 2 h to obtain a yellow solution contain imine. After cooling to 20 °C, a solution of 6N HCl (0.7 mL) in DMF was added into the reaction mixture and continued to heat at 50 °C for 1 h to produce **4** as a white precipitate. The reaction was cooled to 25 °C and aqueous solution of 10 % Na₂CO₃ was added and stirred for 30 min. Then a solution of methyl-2-bromo-2-(2-chlorophenyl)acetate **3** (1.04 g, 4 mmol) in toluene (7 mL) was added to the reaction mixture. The reaction mixture stirred at r.t for 12 h. The aqueous layer is discarded and the organic layer was washed with water. The organic phase was evaporated, to afford viscose oil. To this oil, conc. H₂SO₄ (0.42 mL) and EtOAc (8 mL) was added. The mixture was stirred for 1 h. The precipitated crystals filtered off and washed with cool acetone. The pure racemic clopidogrel

bisulfate **2** was dried in oven at 50 °C (1.35 g, 90% yield). White powder; mp 220-222 °C; IR: ν (KBr, cm^{-1}) 3434 (OH stretch), 3100, 2989, 2953, 1755 (C=O stretch), 1239, 1222 (C-O stretch), 1063.

Preparation of (S)-(+)-clopidogrel camphorsulfonate salt

The racemic clopidogrel bisulfate **2** (1.5 g, 3.75 mmol) and *L*-CSA (0.88 g, 3.75 mmol) were dissolved in acetone and stirred at r.t for 12 h. The formed precipitate was filtered, washed and dried to give a white solid as a (*S*)-clopidogrel camphor sulfonate salt (0.96 g, 75% yield), $[\alpha]_D = +25$ ($c = 1.68$ % in CH_3OH); IR: ν (KBr, cm^{-1}) 3454 (N-H stretch), 2975, 1756 (C=O stretch), 1735, (C=O stretch), 1630, 1506, 1473, 1439, 1263, 1234, 1157, 1030, 753, 724, 701.

Preparation of (S)-(+)-clopidogrel bisulfate (1, CLP)

The (*S*)-(+)-clopidogrel camphorsulfonate salt obtained (0.96 g, 4.8 mmol) was suspended in EtOAc and stirred vigorously at 10 °C. Subsequently, H_2SO_4 (0.3 g, 3 mmol) was added dropwise into mixture and stirred for 4 h and (*S*)-(+)-clopidogrel bisulfate **1** was obtained by the usual filtering, washing and drying (0.78 g, 85% yield); mp 176-178 °C; $[\alpha]_D = +55.10$ ($c = 1.68$ % in CH_3OH); IR: ν (KBr, cm^{-1}) 3435 (O-H stretch), 3121, 2956, 2550, 2552, 1753, 1591, 1439, 1188, 1154, 1029; ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 3.10 (s, 2H), 3.49 (s, 2H), 3.76 (s, 3H), 4.24 (s, 2H), 5.67 (s, 1H), 6.89 (d, 1H, $J = 4.7$ Hz), 7.44 (d, 1H, $J = 5.1$ Hz), 7.52-7.58 (m, 2H), 7.66 (dd, 1H, $J = 1.5, 7.8$ Hz), 7.72 (dd, 1H, $J = 1.5, 7.4$ Hz); ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 23.3, 54.5, 66.4, 125.9, 126.3, 129.4, 131.1, 131.5, 132.5, 133.1, 135.1, 168.3; *Anal.* Calcd for $\text{C}_{16}\text{H}_{18}\text{ClNO}_6\text{S}_2$: C, 45.77; H, 4.32; N, 3.34. Found: C, 45.61; H, 4.39; N, 3.48.

Preparation of (S)-(+)-clopidogrel bisulfate-coated silver nanoparticles (Ag@CLP)

A solution of AgNO₃ (0.082 g, 0.48 mmol) in ultrapure H₂O (10 mL) was prepared. To this solution were added 0.03 g soluble starch in 5.0 mL ultrapure H₂O and then the mixture was stirred heavily for 30 min in an ice bath. A solution of NaBH₄ (0.018 g, 0.48 mmol) in ultrapure H₂O (6 mL) was added dropwise to the aqueous solution, the color of solution suddenly turned to bright yellow, attributed to the formation of AgNPs. Then, the color of solution turned to black. On complete addition of NaBH₄, the resulting mixture was further stirred for 30 min at r.t. Then, a solution of CLP (0.403 g, 0.96 mmol) in EtOH (10 mL) was added to the reaction vessel, and then the mixture was stirred for further 4 h at this stage the color of solution was turned to dark brown. The suspension obtained was then centrifuged at 10,000 rpm for 15 min and the precipitate washed three times with double distilled H₂O to remove any water soluble impurity. After that, the precipitate was washed 3 times by dispersion and centrifugation using EtOH to remove excess CLP and excess reducing agent. The precipitate was then dried in an oven at 60 °C for 10 h and the pale gray powder was obtained.

Results and Discussion

In continuation to our interest in the design and synthesis of chemically and biologically medicinal heterocycles,^[22-27] in this paper, we attempts to functionalized AgNPs covalently with the CLP (Ag@CLP). There are many routes to synthesis of CLP.^[28-32] Herein; first we report an efficient modified synthesis of CLP with excellent yield and in high optical purity resolution. The one-pot synthesis of CLP **1** is depicted in Scheme 1. In this effort 2-(thiophen-2-yl)ethanamine **5** upon reaction with paraformaldehyde (PFA) and HCl (6 N) in toluene followed by electrophilic cyclization afforded tetrahydrothieno pyridine **4**. Subsequently, this intermediate

undergoes reaction with methyl-2-bromo-2-(2-chlorophenyl)acetate **3** and acetone/conc. H₂SO₄ to obtain a pure racemic **2**, which was resolved using *L*-camphorsulfonic acid (*L*-CSA) in acetone to afford (*S*)-(+)-clopidogrel bisulfate **1**.

The mp, IR, ¹H NMR, ¹³C NMR, XRD, [α]_D and elemental analysis confirm that a polymorphic crystalline form II of **1** was produced.^[3]

The synthesis of tetrahydrothieno pyridine **4** was optimized and the results are shown in Table 1. When this reaction was carried out in the presence of DMF.HCl and toluene, the yield of **4** was increased up to 90% (Table 1, Entry 2).

In other efforts, we examined several other bases and solvents to optimize the reaction conditions for preparation of **1**. The results are listed in Table 2. It was found that using toluene along with Na₂CO₃ led to higher yields in shorter reaction times (Table 2, Entry 5).

The kinetic resolution of prepared racemic clopidogrel bisulfate **1** was completed by using *L*-CSA. An optimization of the reaction parameter such as utilized *L*-CSA (in mmol), solvent and yield with respect to the final product concentration were performed. It was found that using 0.012 mmol *L*-CSA in acetone led to higher yields (Table 3, Entry 3). Subsequently, the obtained (*S*)-clopidogrel camphor sulfonate salt, was converted to (*S*)-(+)-clopidogrel bisulfate **1** in H₂SO₄ / EtOAc condition (Table 4, Entry 5).

For preparation of Ag@CLP, solution of AgNO₃ in ultrapure H₂O and starch as a capping agent under vigorous stirring was added to the round bottom flask. Subsequently, a modified Brust reaction^[33] was carried out by addition of excess amount of NaBH₄ as reducing agent. In order to reduce all of the Ag⁺ ions to metallic silver, the molar ratio of metal to the reducing agent was selected as 1:2 respectively. Then, EtOH solution of CLP (molar ratio CLP/Ag = 2:1) was added

to the reaction vessel. The sulfur of CLP bind strongly to AgNPs surface by self-assembly.^{[25,34-}

^{35]} Formation of Ag@CLP was monitored by IR, X-ray diffraction, UV-Vis and TEM. To achieve a fine nanosized particle the amylose was used as a green capping agent.^[36]

The interaction between CLP and AgNPs was recognized by IR spectra of CLP and Ag@CLP (Figure S3 and S4). In both spectra the C=O, O-H of the HSO₄⁻, N⁺-H and aromatic C-H stretching vibration bands at 1753, 3435, 2552 and 3121 cm⁻¹ were observed respectively. The C=C_{sym} and C=C_{asym} bands of thiophene for CLP and Ag@CLP were observed at 1439, 1591 cm⁻¹ and 1444, 1590 cm⁻¹ respectively, indicating that sulfur is attached to the AgNPs surface.^[34]

The powder XRD patterns of the CLP and Ag@CLP are shown in Fig. 2. The peaks position of the CLP were observed at $2\theta = 8.87^\circ, 13.04^\circ, 17.76^\circ, 18.59^\circ, 20.24^\circ, 21.69^\circ, 23.04^\circ, 23.86^\circ, 24.79^\circ, 26.61^\circ, 27.72^\circ, 29.44^\circ, 34.30^\circ, 35.76^\circ, 41.15^\circ$ and 48.68° (Figure 2a). This pattern correlated well with the crystalline structure of form II.³ In Figure 2b, the characteristic peaks of AgNPs were at $2\theta = 38.24^\circ, 44.43^\circ$ and 64.56° respectively, assigned as (111), (200) and (220) reflection lines of the face centered cubic (fcc) structure of metallic silver. The average crystallite size, D , was calculated from Scherrer's equation to be 29 nm while $D = K\lambda/(\beta\cos\theta)$ and λ is the wavelength of Cu-K α radiation used ($\lambda = 1.54 \text{ \AA}$), β is the full width at half-maximum (FWHM) intensity (0.31) of the diffraction line, θ is the Bragg angle for the measured hkl peak, and K is a constant equal to 0.94. This value is in good agreement with the TEM image. The intensity of peaks reflected the high degree of crystallinity of the AgNPs.

UV-Vis Absorption Spectra

Fig. 3, illustrates the UV-Vis absorption spectra of CLP and Ag@CLP in EtOH solution. The pure CLP, shows a λ_{\max} at 243 nm corresponds to the π - π^* transition of the chlorobenzene and

thiophene moiety while, solution of Ag@CLP shows absorption peaks at 243 and 412 nm corresponds to CLP and surface plasmon resonance absorption band (SPRAB) of AgNPs, respectively. The weaker absorption band of Ag@CLP, compared to absorption band of free CLP at 243 nm, confirms a lower concentration of CLP on AgNPs. The appearance of two new peaks at 297 and 346 nm probably are related to the formation of Ag-CLP complex.

TEM Imaging

The TEM of Ag@CLP image reveals that particles are spherical shape with approximate size of < 40 nm; Scherrer's calc. = 29 nm (Fig. 4).

Conclusions

We have developed one-pot process for the synthesis of polymorphic crystalline form II CLP and utilizing a novel method for adsorption of premade CLP molecule on AgNPs surface. The sulfur of CLP bind strongly to AgNPs surface by self-assembly as confirmed by IR and UV-Vis. This Ag@CLP system may provide an advantage model system for the development of new effective antiplatelet drug.

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Table 1. Reaction conditions for the synthesis of tetrahydrothieno pyridine **4**.

Entry	Utilized Acid	Solvent	Yield (%)
1	DMF.HCl	C ₂ H ₄ Cl ₂	80
2	DMF.HCl	Toluene	90
3	<i>i</i> -Pr.HCl	C ₂ H ₄ Cl ₂	75
4	<i>i</i> -Pr.HCl	Toluene	80
5	<i>p</i> -TSA	Toluene	65

Table 2. Reaction conditions for the synthesis of racemic-CLP.

Entry	Solvent	Base	Time (h)	Yield (%)
1	C ₂ H ₄ Cl ₂	NEt ₃	6	80
2	C ₂ H ₄ Cl ₂	Na ₂ CO ₃	10	85
3	DMF	K ₂ CO ₃	5	80
4	Toluene	NEt ₃	5	85
5	Toluene	Na ₂ CO ₃	2	90
6	CH ₃ OH	Na ₂ CO ₃	4	88

Table 3. Reaction conditions for the kinetic resolution.

Entry	<i>L</i> -CSA (mmol)	Solvent	Time (h)	Optical rotation	Yield (%)
1	0.015	acetone	4	+23	72
2	0.012	acetone	4	+24	66
3	0.012	acetone	12	+25	75
4	0.015	Toluene	4	+22	70
5	0.012	Toluene	4	+23	75
6	0.009	Toluene	12	+24.5	74

Table 4. Variation of solvent for resolution of (*S*)-(+)-clopidogrel bisulfate.

Entry	Solvent	Time (h)	Yield (%)
1	MIPK*	8	80
2	THF	8	75
3	<i>i</i> -PrOH	12	80
4	acetone	5	80
5	EtOAc	4	85

*Methyl isopropyl ketone

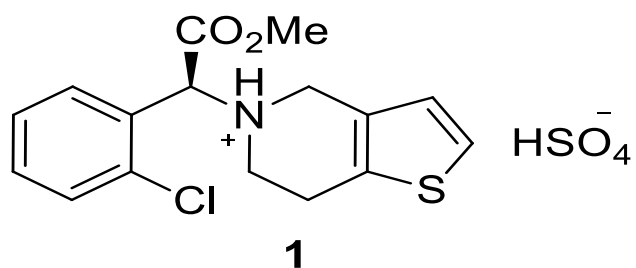


Fig.1. Chemical structure of (*S*)-(+)-clopidogrel bisulfate (CLP).

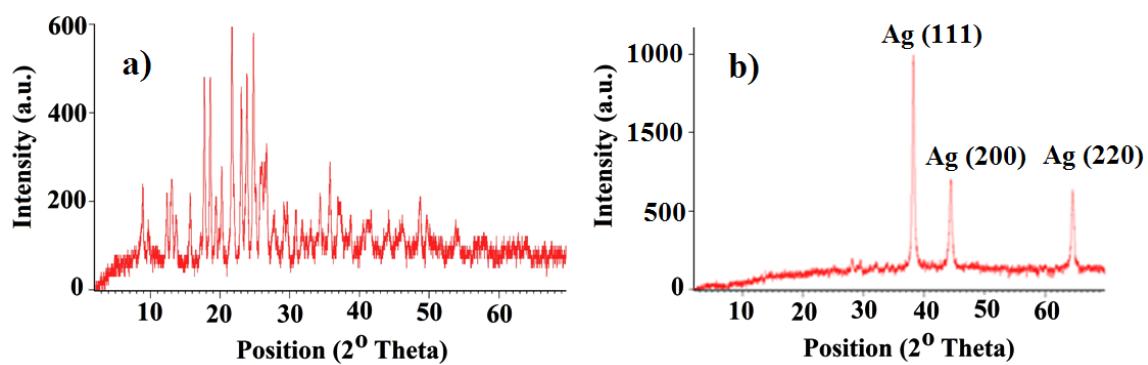


Fig. 2. X-ray diffraction patterns of the (a) CLP and (b) Ag@CLP.

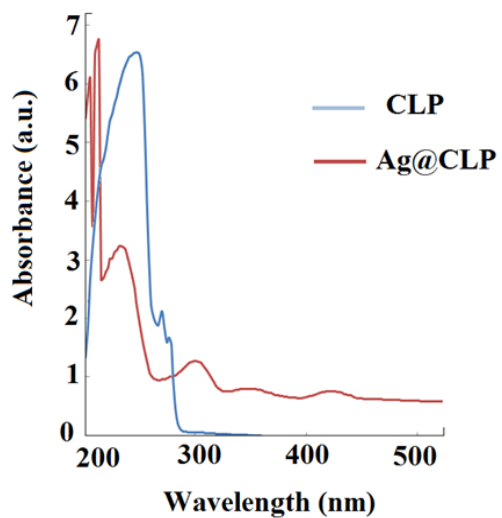


Fig. 3 UV-Vis absorption spectra of CLP and Ag@CLP (EtOH, $c = 1.0 \times 10^{-4}$ M, 293 K).

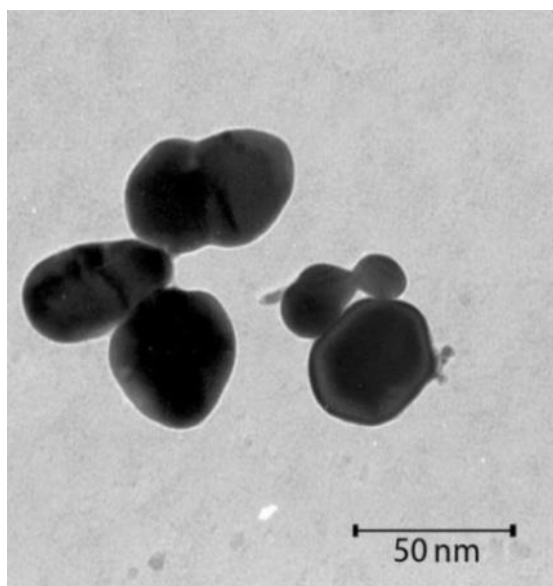
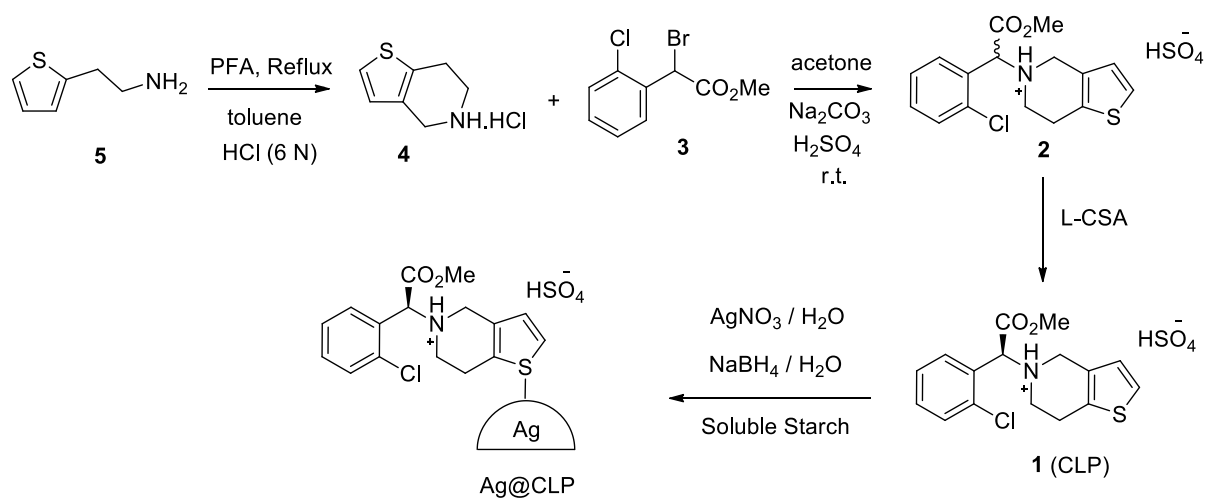


Fig. 4. TEM image of the Ag@CLP.



Scheme 1. Preparation of Ag@CLP.