



Antiparasitic antioxidant phenylpropanoids and iridoid glycosides from *Tecoma mollis*

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ABSTRACT

A radical scavenging guided phytochemical study on the stem bark of *Tecoma mollis* afforded seven active phenylpropanoid glycosides (1–7), including a new one (4), and one iridoid (8). The structures of the isolated compounds were elucidated on the basis of spectroscopic evidences and correlated with known compounds. Compounds (1–7) displayed promising antioxidant activity (DPPH assay) in relation to ascorbic acid (positive control). The antimicrobial activity for compounds (1–8) was evaluated against five bacterial and five fungal strains. The isolated compounds exhibited nonselective weak to moderate antimicrobial activity. The highest antileishmanial activity against *Leishmania donovani* was observed for compound (7) with an IC₅₀ value of 6.71 µg/ml, using pentamidine and amphotericin B as drug controls. Compound (5) exhibited moderate antimalarial activity (45% inhibition) against chloroquine sensitive (D6) clones of *Plasmodium falciparum*.

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1. Introduction

Tecoma mollis Humb. and Bonpl., Family Bignoniaceae is an upright shrub to a large tree attaining 7 m in height. The plant is native to South United States, Mexico to Venezuela and Argentina and is known by the following synonyms: *Tecoma velutina*, Lindl.; and *Tecoma stans* var. *velutina*, Hort. [1]. The plant has been introduced to Egypt in 1982 as an ornamental plant for its timber and showy bell-shaped flowers and cultivated in the experimental station of Faculty of Agriculture, Assiut Univ., Egypt. *T. mollis* is a member of genus *Tecoma* which is of economic importance both for lumber and as

cultivated ornamentals [2,3]. The wood is widely used for timbering, construction and in building of ships and for export due to their hardness and resistance for environmental conditions [4–6]. To the best of our knowledge, little studies were focusing on the phytochemistry and biological activity of *T. mollis* [7], and this is the first study describing in details the chemistry of the polar constituents as well as the potential biological activities of *T. mollis*. The only reported phytochemical study of this plant exhibited the isolation of triterpenoid as well as flavonoid glycosides [7]. The reported biological studies of the leaves of *T. mollis* include analgesic, antipyretic and CNS depressant activities [7]. Also, the leaves of *T. mollis* have been used in traditional medicine as hypoglycaemic due to its alkaloidal contents [8,9]. In general, the genus *Tecoma* is rich in phytoconstituents and previous studies revealed the isolation and identification of many constituents of different chemical classes as phenylpropanoids [10,11], iridoids [11–16], flavonoids [7, 17–21], alkaloids [22–24], triterpenoids [7], phenolics [25] and indolic compounds [26]. Many biological activities were also reported for this genus such as anti-inflammatory,

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hypoglycaemic, spasmolytic, diuretic, antimicrobial and for protozoal diseases [27–35]. Other reported biological activities include antioxidant, anti-proliferative, insecticidal and as astringent for the treatment of diarrhea, dysentery and enteritis [36–40].

In the course of our ongoing research activities toward the isolation of biologically active compounds from plants growing in Egypt either wild or cultivated, in particular the species of diverse chemical constituents with various reported biological activities, and as a part of our collaboration with the National Center for Natural Products Research, School of Pharmacy, University of Mississippi, USA, for differential antimicrobial and antiparasitic screening, we had the opportunity to continue the work on the stem bark of the cultivated *T. mollis* to investigate its chemical constituents and potential biological activities.

In this study, we report the isolation of seven phenylpropanoid glycosides and one iridoid from the stem bark of *T. mollis*, one of which (**4**) is a new phenylpropanoid glycoside. All structures were established by spectroscopic analyses, as well as using accurate mass measurement.

2. Materials and methods

2.1. General experimental procedures

IR spectra were measured on an Ati Mattson Genesis Series FTIR machine. UV absorption was done on a UV-vis spectrometer (Perkin-Elmer Instruments, Lambda 25). The 1D and 2D-NMR experiments were recorded on Varian Unity INOVA 400 MHz NMR systems. Low resolution ESI-MS data were obtained using a Perseptive Biosystems Mariner LC-MS, and high-resolution ESI-MS data were obtained on a Finnigan MAT 900 XLT system. HPLC separations were carried out using a Phenomenex RP column (C18, 250 × 10 mm, 5 μm) and an Agilent 1200 series gradient pump monitored using a DAD G1315B variable-wavelength UV detector. Column chromatography (CC) was performed using a silica gel (Kieselgel 60 Å, 40–63 μm mesh size, Fluorochem, UK), size exclusion chromatography (Sephadex LH-20, 25–100 mm mesh size, SIGMA, Germany) and using TLC pre-coated silica-gel 60 F254 (0.25 mm, ALUGRAM® SIL G/UV₂₅₄, Macherey-Nagel, Germany) and RP-18 F254S plates (0.25 mm, Merck, Germany). All flash chromatography was performed on Biotage cartridge flash columns (Si 40 M and C18 HS 40 M) using a Biotage Flash system, Charlottesville, USA.

2.2. Plant materials

The stem barks were collected in June 2005 from the trees cultivated in the experimental station of Faculty of Agriculture, Assiut Univ., Egypt. The plant was authenticated by Dr. Gamal Taha, Dept. of Horticulture, Faculty of Agriculture, Assiut Univ., Egypt. A voucher specimen (no. 2011TM) was deposited at the Pharmacognosy Dept., Faculty of Pharmacy, Assiut Univ., Egypt.

2.3. Extraction and isolation

One kg of the air-dried powdered stem barks of *T. mollis* were extracted by maceration in 70% EtOH (2 L × 3). The alcoholic extract was concentrated and the solvent free

residue (185 g, 18.5%) was mixed with 500 mL of distilled H₂O, and defatted with CHCl₃. The aqueous layer was subjected to DIAION HP20 column (25 × 1.5 cm) and eluted with H₂O, MeOH and Me₂CO, successively. The MeOH eluant (31 g) was subjected to medium pressure liquid chromatography (MPLC) on silica gel column using CHCl₃–MeOH mixtures in a manner of increasing polarities. Hundred and five fractions (100 mL each) were collected and monitored on TLC (silica gel) using CHCl₃–MeOH–H₂O (80:20:0.5), (70:30:3) and (55:40:5), as well as on RP-C18 using H₂O–MeOH (70:30), (50:50) and (30:70) as solvent systems and Anisaldehyde/H₂SO₄ as a spraying reagent. Similar fractions were combined to yield five groups.

Group 1 (fractions 1–22, eluent: CHCl₃–MeOH, 8:2) (3.6 g) was re-chromatographed on RP-C18 column using eluent MeOH–H₂O (70:30) to provide seven sub-fractions. Part (40 mg) of sub-fraction 1–3 (900 mg) was purified by HPLC (Luna 5 μm C18 100 Å, 10 μm × 10 mm × 250 mm) using a gradient of 0–100% CH₃CN–H₂O over 40 min to give compound (**2**) (9.7 mg, yellow powder).

Group 2 (fractions 23–60, eluent: CHCl₃–MeOH, 7:3) (8.3 g) was re-chromatographed on silica gel column using EtOAc–MeOH in a manner of increasing polarities to provide five sub-fractions. Part (65 mg) of sub-fraction 2–4 (3.9 g) was purified by RP HPLC using a gradient of 0–100% CH₃CN–H₂O over 40 min to afford compound (**1**) (13.6 mg, yellow powder) and compound (**6**) (19.2 mg, yellow powder).

Group 3 (fractions 61–78, eluent: CHCl₃–MeOH, 7:3) (6.9 g) was subjected to gel filtration chromatography using a Sephadex LH-20 column with CHCl₃–MeOH (3:2) to provide nine fractions. Part (80 mg) of sub-fraction 3–6 (3.4 g) was purified by RP HPLC using a gradient of 0–100% CH₃CN–H₂O over 40 min to give compound (**5**) (14.1 mg, yellow powder) and (**4**) (16.4 mg, yellow powder).

Group 4 (fractions 79–96, eluent: CHCl₃–MeOH, 6:4) (5.1 g) was re-chromatographed on RP-C18 column using 5–70% aq. MeOH as eluent to provide six sub-fractions. Part (60 mg) of sub-fractions 2–3 (1.8 g) were purified by RP HPLC using a gradient of 0–100% CH₃CN–H₂O over 40 min to give compound (**3**) (7.1 mg, yellow powder) and (**7**) (13.1 mg, yellow powder).

Group 5 (fractions 97–105, eluent: CHCl₃–MeOH, 6:4) (3.4 g) was re-chromatographed on Sephadex LH-20 column with CHCl₃–MeOH (1:1) to provide seven sub-fractions. Part (50 mg) of sub-fractions 4–6 (826 mg) was purified by RP HPLC using a gradient of 0–80% CH₃CN–H₂O over 40 min to give compound (**8**) (16.8 mg, yellow powder).

2.4. Acid hydrolysis of isolated glycosides [41]

Isolated glycosides (5 mg, each) were dissolved in 0.2 N H₂SO₄ (5 mL) and heated at 95 °C for 1 h. After cooling, the reaction mixture was extracted with Et₂O (2 mL × 3). The aqueous layer was neutralized with NaHCO₃, evaporated to dryness, and extracted with pyridine. The pyridine extract

was then analyzed on silica gel TLC (EtOAc–MeOH–H₂O–AcOH 13:3:3:4). Rhamnose (R_f 0.52), Apiose (R_f 0.48) and glucose (R_f 0.32) were detected by comparison with authentic samples.

2.5. DPPH radical-scavenging assay

Firstly, radical scavenging activity of the isolated glycosides against stable DPPH[•] was performed with a rapid TLC screening method using 0.2% DPPH in MeOH. 30 min after spraying, the active compounds appear as yellow spots against purple background [42,43]. In a second experiment, spectrophotometric assay was carried out according to the method of Yen and Chen [44]. Briefly, 2.0 mL of a wide range of concentrations (2.5–120 μ M) of test sample (in MeOH) was added to 2.0 mL of 100 μ M of methanolic solution of DPPH. The mixture was vortexed for 1 min and then left to stand at room temperature for 30 min in dark. When DPPH[•] reacts with an antioxidant compound, it changes its color from deep-violet to light-yellow which is measured at 517 nm on a UV/visible light spectrophotometer. Absorption of blank sample containing the same amount of MeOH and DPPH solution was prepared and measured daily. The experiment was carried out in triplicate, using ascorbic acid as a positive control. The percentage reduction of the DPPH, Q , referred to “inhibition or quenching” was calculated by the following formula [45]:

$$Q(\% \text{ inhibition}) = [(A_B - A_A) / A_B] \times 100$$

where: A_B —absorption of blank sample ($t = 0$ min);
 A_A —absorption of tested extract solution ($t = 30$ min).

2.6. Biological assays

2.6.1. Antimicrobial assay

All organisms are obtained from the American Type Culture Collection (Manassas, VA). They include the fungi *Candida albicans* ATCC 90028, *Candida glabrata* ATCC 90030, *Candida krusei* ATCC 6258, *Cryptococcus neoformans* ATCC 90113, and *Aspergillus fumigatus* ATCC 90906 and the bacteria *Staphylococcus aureus* ATCC 29213, methicillin-resistant *S. aureus* ATCC 43300 (MRSA), *Pseudomonas aeruginosa* ATCC 27853, *Mycobacterium intracellulare* ATCC 23068 and *Escherichia coli* (ATCC 35218). Susceptibility testing was performed for all organisms, except *M. intracellulare*, using modified versions of the CLSI/NCCLS methods [46]. *M. intracellulare* susceptibility was tested using a modified Franzblau-method [47]. Samples dissolved in DMSO were serially diluted in DMSO/saline (20%/0.9%) and transferred in duplicate to 96-well flat bottom microplates. Microbial inocula were prepared by correcting the OD₆₃₀ of microbe suspensions in incubation broth to afford final target inocula. Controls [fungi: amphotericin B; bacteria: ciprofloxacin (ICN Biomedicals, OH)] were included in each assay. All organisms were read at 630 or 544(ex)/590(em) nm (*M. intracellulare* and *A. fumigatus*) prior to and after incubation. Percent growth was plotted versus test concentration to afford the IC₅₀. Minimum fungicidal or bactericidal concentrations were determined by removing 5 μ L from each clear well, transferring to agar, and incubating until growth

was seen. The MFC/MBC is defined as the lowest test concentration that kills the organism (allows no growth on agar).

2.6.2. In vitro antileishmanial assay

The *in vitro* antileishmanial activity was evaluated against a culture of *L. donovani* promastigotes grown in RPMI 1640 medium supplemented with 10% GIBCO fetal calf serum at 26 °C [48,49]. A three-day-old culture was diluted to 5×10^5 promastigotes/ml. Drug dilutions (50–3.1 μ g/ml) were prepared directly in cell suspension in a 96-well plate, followed by incubation (26 °C, 48 h). Growth of leishmanial promastigotes was determined by the Alamar Blue assay (BioSource International, Camarillo, CA) [50–52].

Standard fluorescence was measured by a Fluostar Galaxy plate reader (excitation wavelength, 544 nm; emission wavelength, 590 nm). Pentamidine and amphotericin B were used as the drug controls. Percent growth was calculated and plotted against the tested concentrations in order to determine the IC₅₀ and IC₉₀ values.

2.6.3. In vitro antimalarial activity

The assay for antimalarial activity was based on the determination of plasmodial LDH activity. For the assay, a suspension of red blood cells infected with D6 strain of *Plasmodium falciparum* (200 μ L, with 2% parasitemia and 2% hematocrit in RPMI 1640 medium supplemented with 10% human serum and 60 μ g/mL amikacin) was added to the wells of a 96-well plate containing 10 μ L of serially diluted test samples. The plate was flushed with a gas mixture of 90% N₂, 5% O₂, and 5% CO₂ and incubated at 37 °C, for 72 h in a modular incubation chamber (Billups-Rothenberg, CA). Parasitic LDH activity was determined according to the procedure of Makler and Hinrichs [53]. Briefly, 20 μ L of the incubation mixture was mixed with 100 μ L of the Malstat™ reagent (Flow Inc., Portland, OR) and incubated at room temperature for 30 min. Twenty microliters of a 1:1 mixture of NBT/PES (Sigma, St. Louis, MO) was then added and the plate is further incubated in the dark for 1 h. The reaction was then stopped by the addition of 100 μ L of a 5% acetic acid solution. The plate was read at 650 nm. Artemisinin and chloroquine were included in each assay as antimalarial drug controls. IC₅₀ values were computed from the dose response curves. To determine the selectivity index of antimalarial activity of compounds their *in vitro* cytotoxicity to mammalian cells was also determined. The assay was performed in 96-well tissue culture-treated plates. Vero cells (monkey kidney fibroblasts) were seeded to the wells at a density of 25,000 cells/well and incubated for 24 h. Diluted samples were added and plates were again incubated for 48 h. The number of viable cells was determined by Neutral Red assay as described earlier [54]. IC₅₀ values were obtained from dose response curves. Doxorubicin was used as a positive control for cytotoxicity.

Verpectoside B (Crassoside) (**3**): yellow amorphous powder (7.1 mg). $[\alpha]_D^{23} +133$ (c 0.08, MeOH); UV: λ_{max} (MeOH): 218, 301, 348 nm; IR (KBr) ν_{max} 3450, 1693, 1617, 1513 cm^{-1} ; HRESIMS m/z 809.24707 ($[M+Na]^+$, calcd for C₃₅H₄₆NaO₂₀, 809.24801, $\Delta = -1.0$ ppm); ¹H (400 MHz, DMSO-*d*₆) and ¹³C (100 MHz, DMSO-*d*₆) NMR data: see Table 1.

Table 1¹H (400 MHz) and ¹³C-NMR (100 MHz) data of **3** and **4** in DMSO-*d*₆.

No.	3		4	
	^δ ¹ H/ppm, mult, J/Hz	^δ ¹³ C/ppm, mult.	^δ ¹ H/ppm, mult, J/Hz	^δ ¹³ C/ppm, mult.
<i>Aglycone</i>				
1	–	129.1, s	–	129.5, s
2	6.62 (1H, brs)	116.3, d	6.61 (1H, d, 1.8)	116.3, d
3	–	145.0, s	–	145.0, s
4	–	143.5, s	–	143.5, s
5	6.62 (1H, d, 8.0)	115.5, d	6.62 (1H, d, 8.0)	115.5, d
6	6.49 (1H, d, 8.0)	119.6, d	6.49 (1H, dd, 1.8, 8.0)	119.5, d
7	2.68 (2H, t, 7.8)	35.0, t	2.69 (2H, m)	35.0, t
8	a. 3.88 (1H, m) b.3.62 (1H, m)	70.3, t	a. 3.82 (1H, m) b.3.63 (1H, m)	70.3, t
<i>Sugar 1</i>				
	Glucose	Glucose	Glucose	Glucose
1'	4.57 (1H, d, 7.8)	101.3, d	4.65 (1H, d, 7.2)	101.2, d
2'	3.56 (1H, m)	81.1, d	3.62 (1H, m)	80.6, d
3'	3.86 (1H, m)	77.7, d	3.91 (1H, m)	77.2, d
4'	4.77 (1H, m)	69.2, d	4.90 (1H, m)	68.8, d
5'	3.50 (1H, m)	74.3, d	3.81 (1H, m)	70.7, d
6'	a. 3.38 (1H, m) b. 3.34 (1H, m)	60.8, t	a. 4.03 (1H, dd, 5.0, 12.0) b. 3.95 (1H, m)	62.5, t
C=O	–	–	–	170.1, s
CH ₃	–	–	1.96 (3H, s)	20.6, q
<i>Caffeoyl moiety</i>				
1''	–	125.5, s	–	125.9, s
2''	7.00 (1H, brs)	114.7, d	7.02 (1H, brs)	114.8, d
3''	–	145.5, s	–	145.6, s
4''	–	148.5, s	–	148.5, s
5''	6.75 (1H, d, 8.0)	115.8, d	6.76 (1H, d, 8.1)	115.8, d
6''	6.98 (1H, d, 8.0)	121.4, d	6.98 (1H, d, 8.1)	121.5, d
7''	7.46 (1H, d, 16.0)	145.6, d	7.46 (1H, d, 15.9)	145.8, d
8''	6.20 (1H, d, 16.0)	113.6, d	6.19 (1H, d, 15.9)	113.2, d
C=O	–	165.7, s	–	165.6, s
<i>Sugar 2</i>				
	Rhamnose	Rhamnose	Rhamnose	Rhamnose
1'''	5.03 (1H, brs)	101.3, d	5.02 (1H, brs)	101.1, d
2'''	3.81 (1H, m)	70.4, d	3.81 (1H, m)	70.4, d
3'''	3.27 (1H, m)	70.4, d	3.27 (1H, m)	70.2, d
4'''	3.11 (1H, m)	71.8, d	3.11 (1H, m)	71.7, d
5'''	3.34 (1H, m)	68.9, d	3.34 (1H, m)	69.0, d
6'''	0.96 (3H, d, 6.2)	18.2, q	0.98 (3H, d, 6.2)	18.2, q
<i>Sugar 3</i>				
	Glucose	Glucose	Glucose	Glucose
1''''	4.54 (1H, d, 7.5)	101.8, d	4.53 (1H, d, 7.5)	101.8, d
2''''	2.99 (1H, t, 7.6)	74.3, d	2.99 (1H, t, 7.6)	74.3, d
3''''	3.12 (1H, m)	77.1, d	3.12 (1H, m)	77.1, d
4''''	3.11 (1H, m)	69.7, d	3.12 (1H, m)	69.7, d
5''''	3.16 (1H, m)	77.1, d	3.16 (1H, m)	77.1, d
6''''	a. 3.68 (1H, m) b. 3.52 (1H, m)	61.0, t	a. 3.68 (1H, m) b. 3.51 (1H, m)	60.9, t

Tocomoloside (6'-*O*-acetyl verpectoside B) (**4**): yellow amorphous powder (16.4 mg). $[\alpha]_D^{23} +116.3$ (*c* 0.2, MeOH); UV: $\lambda_{\text{max}}^{\text{MeOH}}$: 220, 291, 345 nm; IR (KBr) ν_{max} 3433, 1730, 1688, 1624, 1523 cm^{-1} ; HRESIMS *m/z* 851.25623 ($[\text{M} + \text{Na}]^+$, calcd for $\text{C}_{37}\text{H}_{48}\text{NaO}_{21}$, 851.25858, $\Delta = -3.0$ ppm); ¹H (400 MHz, DMSO-*d*₆) ¹³C (100 MHz, DMSO-*d*₆) NMR data: see Table 1.

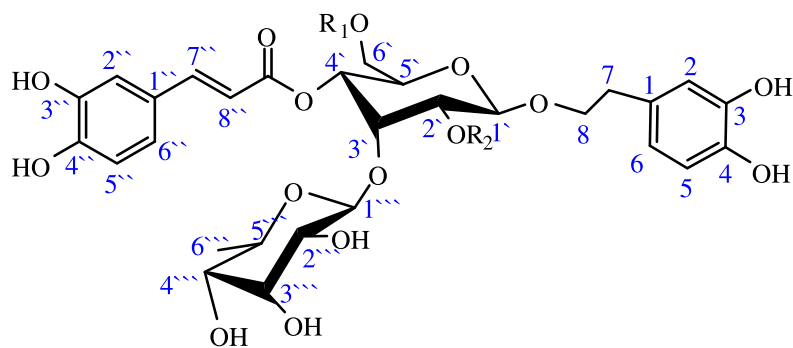
3. Results and discussion

A combination of flash or medium-pressure liquid chromatography (MPLC), gel filtration and HPLC of the MeOH fraction obtained from the stem bark of *T. mollis* afforded seven phenylpropanoids (**1–7**) and one iridoid (**8**) (Fig. 1).

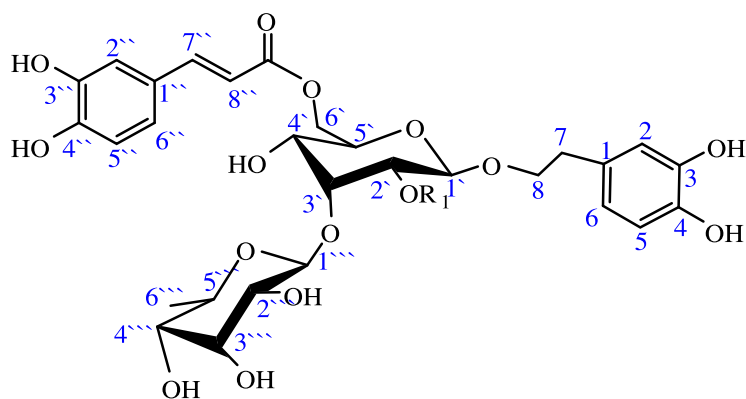
Their structures were elucidated by extensive 1D, 2D NMR analysis, accurate mass measurements and by comparing with the reported data.

Six known phenylpropanoids (**1–3**, **5–7**) were identified as acteoside (Verbascoside) (**1**) [55,56], 6'-*O*-acetylverbascoside (**2**) [57], verpectoside B (Crassoside) (**3**) [58], luteoside A (**5**) [41,59], isoacteoside (isoverbascoside) (**6**) [57,60], luteoside B (**7**) [41,61] and one iridoid, ixoside (**8**) [61–63]. All physical and spectral data of these compounds were in agreement with the respective published data.

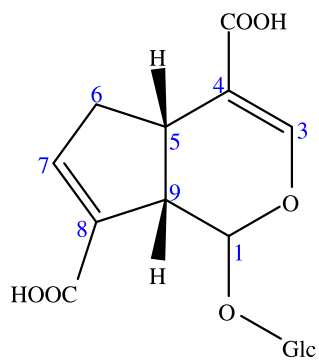
Compound **4** was obtained as a yellow amorphous powder ($[\alpha]_D^{23} +116.3$ (*c* 0.2, MeOH)). HRESIMS showed pseudomolecular ion peaks at *m/z* 829.27435 $[\text{M} + \text{H}]^+$ and 851.25623 $[\text{M} + \text{Na}]^+$ consistent with a molecular weight of 828 amu.



- | | | |
|----------|----------------|-------------|
| 1 | $R_1 = H$ | $R_2 = H$ |
| 2 | $R_1 = COCH_3$ | $R_2 = H$ |
| 3 | $R_1 = H$ | $R_2 = Glc$ |
| 4 | $R_1 = COCH_3$ | $R_2 = Glc$ |
| 5 | $R_1 = COCH_3$ | $R_2 = Api$ |



- | | |
|----------|-------------|
| 6 | $R_1 = H$ |
| 7 | $R_1 = Api$ |



8

Fig. 1. Chemical structures of isolated compounds **1–8** (Glc = Glucose, Api= Apiose).

Table 2
DPPH radical scavenging activity of fractions and isolated compounds.

Sample	IC ₅₀ in µg/ml (µM)
Total ext.	99.3
Chloroform fraction	>200
Hydro methanolic fraction	52.8
Group 1	39.6
Group 2	18.7
Group 3	28.8
Group 4	14.1
Group 5	128.6
Verbascoside (1)	6.1 (9.8)
6'-O-acetylverbascoside (2)	7.5 (11.2)
Verpectoside B (3)	8.6 (10.9)
Tocomoloside (4)	12.8 (15.5)
Luteoside A (5)	11.7 (14.6)
Isoverbascoside (6)	8.5 (13.7)
Luteoside B (7)	12.7 (16.8)
Ixoside (8)	42.4 (109.2)
Ascorbic acid	2.1 (11.9)

The molecular formula was established as C₃₇H₄₈O₂₁, thus implying 14 degrees of unsaturation. The UV:λ_{max}^{MeOH}: 220, 291, 345 nm, while IR spectrum showed strong absorption bands attributed to the hydroxyl and ester groups at 3433 and 1730, 1688 cm⁻¹, respectively. The 1D and 2D NMR spectra in DMSO-*d*₆ (Table 1) of **4** revealed the phenylpropanoid glycosides pattern and were very similar to those of **3** [58]. The ¹H NMR spectral data showed the presence of two sets of ABX system for β-(3,4-dihydroxyphenyl) ethoxy moiety and caffeoyl moiety, as well as two trans-olefinic protons, three anomeric protons for two β-glucopyranosyl and one α-rhamnopyranosyl units, and an acetyl signal. The ¹H and ¹³C NMR spectra were very similar to those of **3** except for the additional signal of acetyl group (δ_H 1.96 for COCH₃), (δ_C 20.6 for CH₃) and (δ_C 170.1 for CO). Comparison of the ¹³C NMR spectral data of **4** with those of **3** (Table 1) revealed downfield shift in the position of C-6' (+1.7 ppm) and upfield shift of C-5' (3.6 ppm) of the glucosyl moiety indicating that the acetyl group is at C-6' of the glucosyl moiety. Furthermore, the HMBC spectrum showed the three bond correlation between H-6' (δ_H 3.95 and 4.03) of the glucosyl moiety and carbonyl carbon (δ_C 170.1) of the acetyl group. Acid hydrolysis of **4** yielded sugars rhamnose and glucose. Therefore, the structure of compound **4** was elucidated as 6'-O-acetyl-verpectoside B. Compound **4** was identified as a new natural product for which we propose the name tocomoloside.

Phenylpropanoid glycosides are water soluble natural products widely distributed in the plant kingdom. Their significant biological activities such as enzyme inhibition, immunomodulatory, antibacterial and cytotoxic activities have been reported [64]. Recently, phenylpropanoid glycosides were demonstrated *in vitro* assays to have antioxidative and free radical scavenging activities [65–70]. The antioxidant activity of the isolated phenylpropanoids (**1–7**) was determined by using a 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging system. The IC₅₀s ranged from 9.8 to 16.8 µM (Table 2). The antioxidative effect of these compounds was related to the number of free phenolic hydroxyl groups in the form of 3,4-dihydroxy (catechol) moiety in their structures which explain the close similarity of their antioxidant activity. Absence or blocking of the hydroxyl groups by methyl group leads to dramatic decrease of the antioxidant activity [71]. The sequence of the strength of the antioxidant activity was shown to be **1**>**3**>**2**>**6**>**5**>**4**>**7**. The iridoid glycoside (ixoside) (**8**) exhibited very weak antioxidant activity with IC₅₀ = 109.2 µM. The free radical scavenging effects of the phenylpropanoids proven here may play an important role in the actions of the *Tecoma* species, and partly explain the mechanisms of the activities of phenylpropanoids against neoplasma [66,72] and inflammation [72,73] in which free radical are seriously involved.

The antifungal and antibacterial activity for the isolated compounds was evaluated against Gram positive, Gram negative bacteria and fungi. The percentage of inhibition (% inh.) of each strain was determined and compared with those for commercially available antibiotics (Table 3). In general, weak activity was observed for isolated compounds against fungal strains particularly *A. fumigatus* ATCC 90906. The highest activity was observed for compounds (**6**) and (**7**) with inhibition of 30% and 40% respectively. Similar was observed against Gram +ve and -ve bacteria with highest inhibition occurred for *Mycobacterium intracellulare* by compound (**7**).

Compounds (**1–8**) were tested against the protozoan parasite *L. donovani*, using pentamidine and amphotericin B as drug controls (Table 4). The highest *in vitro* antileishmanial activity was observed for compounds **1**, **5**, and **7** with percentage of inhibition [83% (IC₅₀ 30.08), 92% (IC₅₀ 15.07) and 92% (IC₅₀ 6.71), respectively] (Table 4). Compounds **2** and **6** showed relatively moderate activity with percentage of inhibition [62% (IC₅₀ >40), and 78% (IC₅₀ >40), respectively], while the least active compounds are **3**, **4** and **8** with percentage of inhibition (19%, 21% and 13%, respectively). It is obvious from Table 4

Table 3
Antimicrobial activity of isolated compounds (**1–8**) against different bacteria and fungi.

Compound number	Compound conc. (µg/ml)	<i>S. aureus</i> % inh.	MRS % inh.	<i>E. coli</i> % inh.	<i>P. aeruginosa</i> % inh.	<i>M. intracellulare</i> % inh.	<i>C. Albicans</i> % inh.	<i>C. glabrata</i> % inh.	<i>C. krusei</i> % inh.	<i>A. fumigatus</i> % inh.	<i>C. neoformans</i> % inh.
1	13	7	5	6	2	8	16	11	12	1	14
2	13	11	0	14	6	6	9	2	5	0	3
3	13	9	5	10	0	0	21	14	8	0	6
4	13	8	8	2	6	4	7	6	5	1	12
5	13	11	0	5	3	16	8	9	1	4	4
6	13	10	8	3	4	6	11	30	9	0	11
7	13	9	10	7	0	33	14	41	0	8	6
8	13	8	1	1	4	6	17	0	11	0	10

Table 4
In vitro antileishmanial activity of isolated compounds (1–8).

Compound Number	Compound conc. (µg/ml)	<i>L. donovani</i> % inh.	IC ₅₀ (µg/ml)	IC ₉₀ (µg/ml)
1	80	83	30.08	>40
2	80	62	>40	>40
3	80	19	ND	ND
4	80	21	ND	ND
5	80	92	15.07	>40
6	80	78	>40	>40
7	80	92	6.71	22.28
8	80	13	ND	ND
Amphotericin B ^a	80	100	0.34	0.43
Pentamidine ^a	80	100	1.37	1.62

IC₅₀: concentration causing 50% growth inhibition. IC₉₀: concentration causing 90% growth inhibition.

ND: not determined. a: control.

that introduction of five membered ring sugar moiety, such as apiose, in the position 2' lead to improve the antileishmanial activity, while introduction of six membered sugar moiety such as glucose lead to dramatic decrease in the activity. Acetylation at position 6' has neglectable effect. Not many phenylpropanoids have been reported for antileishmanial activity and little records considered the antileishmanial activity of phenylpropanoids [74]. The present study tries to highlight on the potential antileishmanial activity for this class of compounds and further investigation is required to establish and optimize their activity.

The *in vitro* antimalarial activity of (1–8) against chloroquine sensitive (D6) clones of *P. falciparum* was evaluated based on the determination of plasmodial LDH activity (Table 5). Compound (5) showed moderate antimalarial activity with (45%) inhibition against chloroquine sensitive (D6) clones of *P. falciparum*. The rest of the compounds exhibited weak antimalarial activity.

In conclusion, we report here the isolation of 7 phenylpropanoid glycosides and 1 iridoid from the stem bark of *T. mollis* cultivated in Egypt, one of which (4) is a new phenylpropanoid glycoside.

The phenylpropanoids with ortho phenolic dihydroxy group showed significant antioxidant activity comparable with ascorbic acid as natural antioxidant while the iridoid exhibited very weak antioxidant activity. None of the isolated compounds exhibited significant antimicrobial and antimalarial activity. Compounds 5 and 7 showed the highest antileishmanial activity, and further investigation is required for this class

Table 5
Antimalarial activity of isolated compounds (1–8).

Compound number	<i>P. falciparum</i> (D6 clone) % inh.
1	22
2	0
3	2
4	5
5	45
6	10
7	0
8	9

of compounds to use as antileishmanial drugs. Moderate antimalarial activity was observed with compound 5 against chloroquine sensitive (D6) clones of *P. falciparum*.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at doi:10.1016/j.fitote.2011.12.025.

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