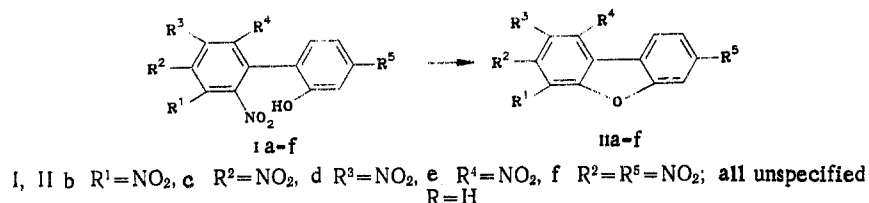


## SYNTHESIS OF DIBENZOFURAN AND ITS NITRO-SUBSTITUTED DERIVATIVES

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The intramolecular nucleophilic substitution of the ortho-nitro group in 2'-nitro-2-hydroxybiphenyl in the presence of alkali agents (KOH, NaH, t-BuOK) in polar aprotic solvents (DMF, hexametapol, DMSO) leads to dibenzofuran. This reaction is a further example of the intramolecular nucleophilic substitution of the ortho-nitro group in the biphenyl nucleus [1, 2].



The solution of 0.01 mole of 2'-nitro-2-hydroxybiphenyl (Ia-f) in 30-45 ml of the aprotic solvent is stirred for 2-6 h in the presence of 0.02 mole of the alkali agent at a raised temperature; the mixture is cooled and poured into dilute hydrochloric acid. The precipitate crystals are filtered off and purified by chromatography on a column with alumina prior to the isolation of the dibenzofurans (IIa-f).

The compounds are presented with the characteristics of reaction temperature (°C), yield (%), utilizing hexametapol or DMSO as the solvent and NaH as the alkali agent), and mp (°C): (IIa), 125-130, 70, and 82-82.5 (from acetone); (IIb), 110-120, 72, and 138-138.5 (from acetone); (IIc), 110-120, 75, and 182-183 (from ethanol); (IId), 110-120, 70, and 151.5-152 (from benzene); (IIe), 100, 89, and 121-122 (from acetic acid); (II f), 40-60, 83, and 327-328 (from acetic acid).

The data of the elemental analysis of the compounds (IIa-f) correspond to the calculated data. The individuality of the compounds was evaluated by the method of thin layer chromatography on Silufol.

## LITERATURE CITED

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