

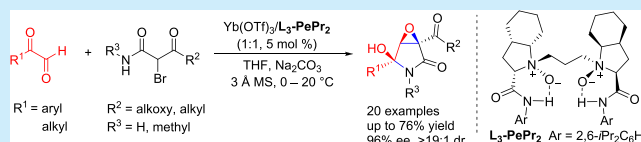
Asymmetric Synthesis of α,β -Epoxy- γ -lactams through Tandem Darzens/Hemiaminalization Reaction

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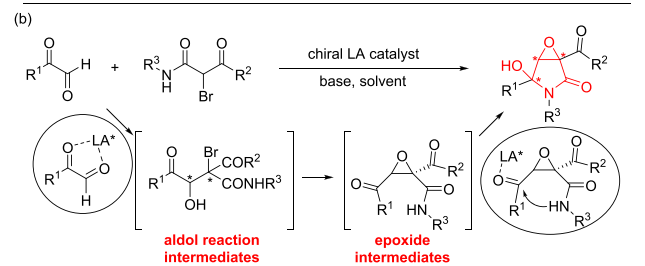
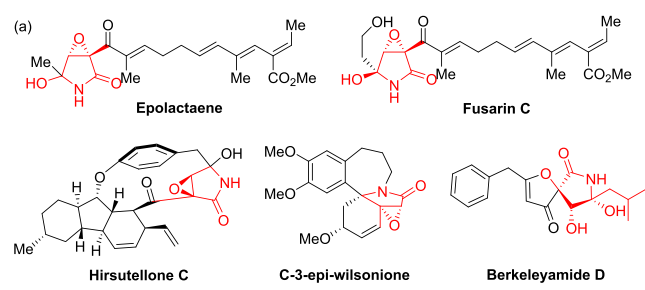
S Supporting Information

ABSTRACT: A catalytic asymmetric tandem Darzens/hemiaminalization reaction of glyoxals with α -bromo- β -esteramides or α -bromo- β -ketoamide was accomplished in the presence of a chiral N,N' -dioxide/Yb(III) complex. Various chiral α,β -epoxy- γ -lactams were obtained in moderate to good yields with excellent diastereo- and enantioselectivities. The versatility of the transformation is illustrated in the formal synthesis of berkeleyamide D.



α,β -Epoxy- γ -lactams are a highly valuable skeleton that is present in a plethora of biologically active molecules, such as epolactaene,¹ fusarin C,² hirsutellone C,³ and C-3-epi-wilsonione⁴ (Scheme 1a). They can also serve as attractive

Scheme 1. (a) Selected Natural Products Bearing α,β -Epoxy- γ -Lactam Structure; (b) Darzens/Hemiaminalization Reaction for the Construction of Epoxy pyrrolidinone Derivatives



building blocks to access more complex molecular architectures, including the synthesis of salinosporamide A,⁵ lucilactaene,⁶ azaspirene,⁷ and berkeleyamide D.⁸ The valuable biological activities of these related natural products, as well as a novel 3-acyl-5-hydroxyl-3,4-epoxy-2-pyrrolidinone structure characteristic, have attracted considerable interest from synthesis chemists. The Darzens/hemiaminalization reaction between glyoxals and α -bromo- β -ketoamide provides a

straightforward route for the construction of this key moiety,^{8,9} which contains three stereogenic centers (Scheme 1b). This method has been used for the short synthesis of (\pm)-berkeleyamide D; however, the optically enriched enantiomers were obtained by resolution with chiral HPLC.^{8a} As asymmetric catalysis is an extremely important method to generate chiral compounds, the development of an efficient asymmetric catalytic version of this Darzens/hemiaminalization reaction is useful but challenging.

Catalytic asymmetric Darzens reaction,¹⁰ which generally employs the base-induced sequence of aldol reaction/intramolecular cyclization, is demonstrated to be one of the powerful approaches for the preparation of chiral epoxy compounds.¹¹ Both chiral organocatalysis and Lewis acid catalysis have been successfully used in enantioselective Darzens reactions. However, the type of nucleophilic species is limited to α -halo carbonyl compounds, chloroacetonitrile, chloromethylsulfones, or 2-halo-1-indanone. The difficulties related to the target asymmetric tandem Darzens/hemiaminalization include the following: (1) α -bromo- β -ketoamides or α -bromo- β -esteramides are new nucleophiles in the Darzens reaction, and their steric hindrance might hamper the addition and S_N2 cyclization processes; (2) the diastereomeric selection in the aldol reaction is directly related to the following formation of epoxide and lactam due to stereochemistry requirements; (3) there are two ring structures and three stereogenic centers in the final product, including two four-substituted carbon centers; (4) both the base-involved background reaction and a retro-aldol process would erode the enantioselectivity of the reaction.¹²

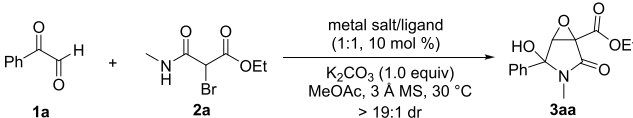
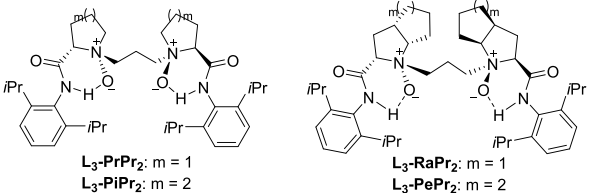
Our research group has achieved good results in a number of asymmetric catalytic transformations, including the Darzens reaction between phenacyl bromides and N -protected isatins,^{10h} as well as a diastereodivergent formal cyclo-

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propanation reaction between 3-Cl oxindoles and β,γ -unsaturated- α -ketoesters.¹³ These results indicate that chiral Lewis acid catalysts derived from N,N' -dioxide and metal salts would enable the titled reaction effectively. The Lewis acid catalyst could activate the carbonyl group of glyoxals¹⁴ and the epoxide intermediate (Scheme 1b), accelerating the initial aldol reaction and the final nucleophilic ring closure, subsequently. The diastereo- and enantioselectivity of the whole process could be controlled by meticulous modification of the chiral N,N' -dioxide subunits. Herein, we report a new chiral N,N' -dioxide-ytterbium(III) complex¹⁵ catalyzed diastereo- and enantioselective Darzens/hemiaminalization reaction, allowing efficient and direct construction of a number of 3-acyl-5-hydroxyl-3,4-epoxy-2-pyrrolidinone derivatives, including the optically enriched intermediate for the synthesis of berkeleyamide D.

Initially, phenyl glyoxal **1a** and ethyl 2-bromo-3-(methylamino)-3-oxopropanoate **2a** were selected as the model substrates to optimize the reaction conditions (Table 1).

Table 1. Optimization of Reaction Conditions^a

entry	metal salt	ligand	yield (%) ^b	ee (%) ^c
1	Mg(OTf) ₂	L ₃ -RaPr ₂	21	12
2	Ni(OTf) ₂	L ₃ -RaPr ₂	29	10
3	Sc(OTf) ₃	L ₃ -RaPr ₂	30	2
4	Y(OTf) ₃	L ₃ -RaPr ₂	27	68
5	Yb(OTf) ₃	L ₃ -RaPr ₂	57	81
6	Yb(OTf) ₃	L ₃ -PiPr ₂	50	12
7	Yb(OTf) ₃	L ₃ -PrPr ₂	56	26
8	Yb(OTf) ₃	L ₃ -PePr ₂	33	84
9 ^d	Yb(OTf) ₃	L ₃ -PePr ₂	65	88
10 ^{d,e}	Yb(OTf) ₃	L ₃ -PePr ₂	64	90
11 ^{d-f}	Yb(OTf) ₃	L ₃ -PePr ₂	71	90
12 ^{d-g}	Yb(OTf) ₃	L ₃ -PePr ₂	71	90

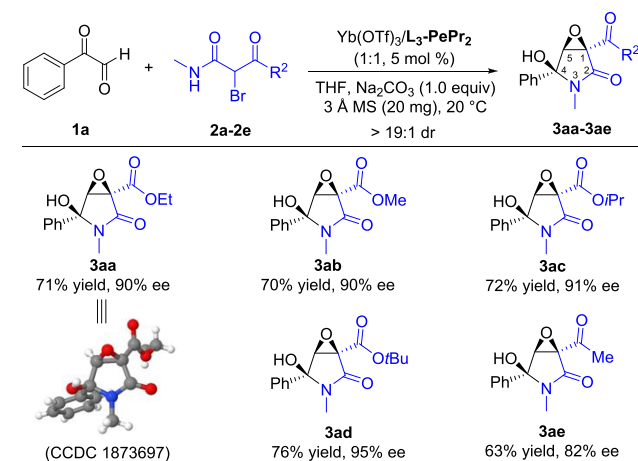
^aUnless otherwise noted, the reactions were performed with metal salt/ligand (1:1, 10 mol %), **1a** (0.10 mmol), **2a** (0.10 mmol), K₂CO₃ (1.0 equiv), and 3 Å MS (20 mg) in MeOAc (1.0 mL) at 30 °C for 16 h. ^bIsolated yield; >19/1 dr was obtained. ^cDetermined by HPLC analysis on a chiral stationary phase. ^dNa₂CO₃ was used instead of K₂CO₃. ^eTHF (1.2 mL) was used instead of MeOAc. ^f20 °C for 24 h. ^g5 mol % catalyst loading for 48 h.

Various metal salts coordinated with chiral N,N' -dioxide ligand L₃-RaPr₂ were evaluated in this reaction (Table 1, entries 1–5). The corresponding α,β -epoxy- γ -lactam **3aa** could be formed smoothly in the presence of K₂CO₃ as the base. It seemed that lanthanide metal salts as Y(OTf)₃ and Yb(OTf)₃ showed higher enantioselectivity than other commonly used metals salts, including Mg(OTf)₂, Ni(OTf)₂, and Sc(OTf)₃ in terms of enantioselectivity. Especially, the metal salt of Yb(OTf)₃ could promote the reaction to give the desired

product **3aa** in 57% yield with excellent diastereoselectivity (>19/1) and good enantioselectivity (81% ee) (Table 1, entry 5). Next, the structure of the N,N' -dioxide ligands was investigated. It was found that the chiral pool of the ligands greatly affected the enantioselectivity of the reaction. The screening of the chiral backbone of the ligands revealed L-perindopril derived L₃-PePr₂ provided higher enantioselectivity (84% ee) than L₃-PiPr₂ (12% ee) derived from S-pipecolic acid, L₃-PrPr₂ (26% ee) derived from L-proline, and L₃-RaPr₂ (81% ee) derived from L-ramipril (Table 1, entry 8 vs entries 5–7). The steric hindrance of the aniline subunits of the ligand had a small amount of influence on the enantioselectivity (see Supporting Information for details). When a weaker base as Na₂CO₃ was used instead of K₂CO₃, both the yield and the ee value had an obvious increase (Table 1, entry 8 vs entry 9). In addition, the enantioselectivity was further improved to 90% ee with THF as the solvent (Table 1, entry 10). Lowering the temperature to 20 °C resulted in a slightly higher yield to 71% (Table 1, entry 11). The yield and ee value could be maintained with a lower catalyst loading (5 mol %) albeit with a prolonged time (Table 1, entry 12). Therefore, the optimal reaction conditions were established as **1a** and **2a** (1.0 equiv), Yb(OTf)₃/L₃-PePr₂ (1:1, 5 mol %), 3 Å MS, and Na₂CO₃ (1.0 equiv) in THF at 20 °C for 48 h.

With the optimized conditions in hand, the substrate generality of the reaction was surveyed. As shown in Scheme 2, α -bromo- β -esteramides **2a–2d** bearing different ester

Scheme 2. Substrate Scope of the α -Bromoamides^a



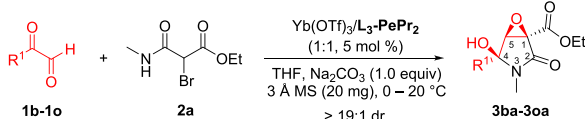
^aUnless otherwise noted, the reactions were performed with Yb(OTf)₃/L₃-PePr₂ (1:1, 5 mol %), **1a** (0.10 mmol), **2** (1.0 equiv), Na₂CO₃ (1.0 equiv), and 3 Å MS (20 mg) in THF (1.2 mL) at 20 °C for 48 h. >19/1 dr.

groups, including methyl, ethyl, isopropyl, and tertbutyl substituents, could react with **1a** smoothly to produce the corresponding products **3aa–3ad** in good yields (70–76% yield) with excellent ee values (90–95% ee). Moreover, the α -bromo- β -ketoamide **2e** was also tolerated and transformed into **3ae** in 63% yield with 82% ee. The absolute configuration of the product **3aa** was determined to be (1*R*,4*S*,5*S*) by X-ray crystallography analysis (CCDC 1873697).

Subsequently, we turned our attention to the scope of glyoxals. The electronic properties or steric hindrance of the substituents on the aromatic ring had a small effect on the stereoselectivity of the tandem reaction, and the desired α,β -

epoxy- γ -lactams **3ba**–**3ma** were obtained with 89–96% ee (Table 2, entries 1–12). The substrate containing electron-

Table 2. Substrate Scope of Glyoxals^a



entry	R ¹	T (°C)	yield (%) ^b	ee (%) ^c
1	3-MeC ₆ H ₄	20	53 (3ba)	91
2	3-MeOC ₆ H ₄	20	71 (3ca)	92
3	3-ClC ₆ H ₄	10	70 (3da)	96
4	3,4-Cl ₂ C ₆ H ₃	20	52 (3ea)	93
5	4-MeC ₆ H ₄	20	71 (3fa)	92
6	4-FC ₆ H ₄	10	65 (3ga)	90
7 ^d	4-ClC ₆ H ₄	0	74 (3ha)	89
8	4-BrC ₆ H ₄	10	62 (3ia)	91
9	4-CF ₃ C ₆ H ₄	10	62 (3ja)	94
10	4-PhC ₆ H ₄	20	75 (3ka)	93
11	4-tBuC ₆ H ₄	20	74 (3la)	90
12 ^d	4-NO ₂ C ₆ H ₄	0	61 (3ma)	91
13	2-Naphthyl	20	62 (3na)	93
14	Cyclohexyl	20	53 (3oa)	70

^aUnless otherwise noted, the reactions were performed with Yb(OTf)₃/L₃-PePr₂ (1:1, 5 mol %), **1** (1.0 equiv), **2a** (0.10 mmol), Na₂CO₃ (1.0 equiv) and 3 Å MS (20 mg) in THF (1.2 mL) at 20 °C for 48 h. All the products were obtained with >19/1 dr. ^bIsolated yield. ^cDetermined by HPLC analysis on a chiral stationary phase. ^d0 °C for 72 h.

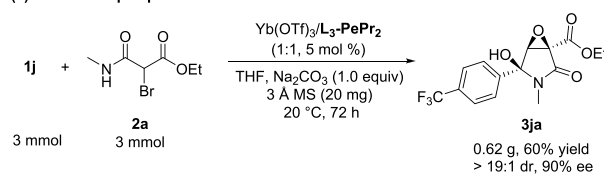
withdrawing substituents at the 4-position of the aromatic groups exhibited higher reactivity than that with electron-donating groups (Table 2, entries 6–9, 12 vs entries 5 and 11). In addition, 2-naphthyl- and cyclohexyl-containing glyoxals were also suitable substrates, giving the products **3na** and **3oa** in 62% yield with 93% ee and 53% yield with 70% ee, respectively (Table 2, entries 13 and 14). It is noteworthy that only one diastereoisomer of the epoxy-pyrrolidinone products was isolated in all cases. The absolute configuration of the products **3ab**–**3ae**, as well as **3ba**–**3na**, were determined to be uniform by comparing the circular-dichroism spectra with that of **3aa**.

To show the practicability of this methodology, a scale-up synthesis of the product **3ja** was carried out. In the presence of 5 mol % of Yb(OTf)₃/L₃-PePr₂, glyoxal **1j** (3 mmol) reacted with α -bromo- β -esteramide **2a** (3 mmol) smoothly and delivered **3ja** in 60% yield, >19:1 dr with 90% ee (Scheme 3a). Berkeleyamide D, originally isolated from the extracts of a fungus, shows inhibition of matrix metalloproteinase-3 and caspase-1.¹⁶ Kuramochi and co-workers realized the synthesis of (\pm)-berkeleyamide D involving the formation of α,β -epoxy- γ -lactam through the Darzens/ring-closure reaction of isobutyglyoxal and α -bromo- β -ketoamide.^{8a} After the screening of the reaction conditions in our catalytic system, the reaction between isobutyglyoxal **1p** and α -bromo- β -ketoamide **2f** proceeded smoothly, generating the desired α,β -epoxy- γ -lactam **3pf** in 55% yield with 82% ee. Accordingly, the access to chiral berkeleyamide D is feasible from the chiral **3pf** (Scheme 3b).^{8a,17}

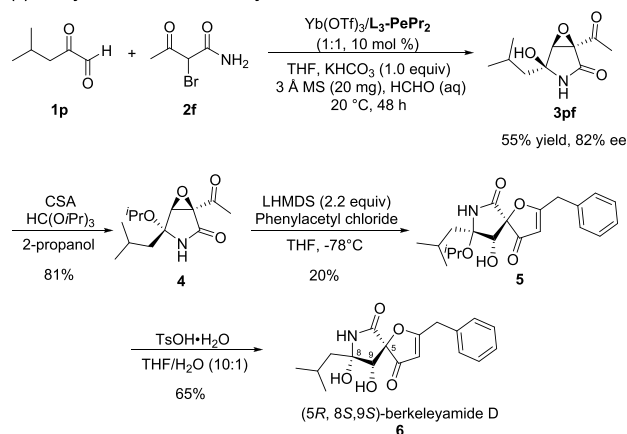
In summary, we have presented a highly enantioselective tandem Darzens/ring-closure reaction of glyoxals with α -

Scheme 3. Scale-up Experiment and Synthesis of Berkeleyamide D

(a) The scale-up experiment



(b) The synthesis of chiral berkeleyamide D



bromo- β -esteramides or α -bromo- β -ketoamide catalyzed by a chiral N,N' -dioxide–Yb(OTf)₃ complex. A series of desired α,β -epoxy- γ -lactams were afforded in up to 76% yield with 96% ee and >19:1 dr. In addition, this protocol has been successfully applied to the formal enantioselective synthesis of the biologically active molecule berkeleyamide D.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.9b01589.

Experiment procedures, full spectroscopic data for all new compounds, and copies of ¹H, ¹³C{¹H}, ¹⁹F{¹H} NMR, and HPLC spectra (PDF)

Accession Codes

CCDC 1873697 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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