which reduces the effect of dose $2 x$ of the agonist to that of a dose $x$.

In the antagonistic experiments we used subthreshold doses for those analogues exhibiting agonism. Regarding analogues lacking agonism, maximal doses of $100 \mathrm{nmol} / \mathrm{kg}$ were tested. Arginine-vasopressin ( $408 \mathrm{IU} / \mathrm{mg}$ ) was used as the standard when agonism was estimated as well as the agonist in the antagonistic experiments.

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Registry No. 1, 112195-99-0; 2, 112196-00-6; 3, 112196-01-7; 4, 112196-02-8; 5, 112196-03-9; 6, 112196-04-0; 7, 112196-05-1; 8, 112196-06-2; I, 112196-07-3; II, 112196-08-4; III, 112196-09-5; IV, 112196-10-8; V, 112196-11-9; VI, 112196-12-0; VII, 112196-13-1; VIII, 112196-14-2; AVP, 113-79-1; BOC-Tyr(Bzl)-Phe-Gln-Asn-Cys(Bzl)-Pro-Arg(Tos)-Gly-NH2, 63491-78-1; BOC-Tyr(Me)-Phe-Gln-Asn-Cys(Bzl)-Pro-Arg(Tos)-Gly-NH 2 , 67230-57-3; 4-methyl-1-[(phenylmethyl)thio]cyclohexaneacetic acid, 112196-15-3; 4-tert-butyl-1-[(phenylmethyl)thio]cyclohexaneacetic acid, 112196-16-4; 4-phenyl-1-[(phenylmethyl)thio]cyclohexaneacetic acid, 112196-17-5; 1-methyl-4-[(phenylmethyl)thio]-4piperidineacetic acid hydrochloride, 112219-50-8.

# 6-(Alkylamino)-9-benzyl-9H-purines. A New Class of Anticonvulsant Agents 

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Several 9-alkyl-6-substituted-purines were synthesized and tested for anticonvulsant activity against maximal electroshock-induced seizures (MES) in rats. Most compounds were prepared in three steps from 5 -amino-4,6dichloropyrimidine or in two steps via alkylation of 6 -chloropurine. Potent anticonvulsant activity against MES resided in compounds that contain a benzyl substituent at the 9 -position of 6 -(methylamino)- or 6 -(dimethylamino)purine. Among commonly used agents for control of seizures, this type of structure represents a new class of potent anticonvulsant agents.

Several drugs are available for treatment of epilepsy, but many patients fail to experience satisfactory seizure control with them, or they do so at the expense of significant side effects. ${ }^{1}$ Despite the many side effects associated with phenytoin, it is still the drug of choice for the treatment of many epileptic seizures. ${ }^{2}$ Due to the need for new, improved antiepileptic drugs, a program was initiated to discover and develop candidate antiepileptic agents with improved properties. ${ }^{3,4}$ From this program emerged 9-(2-fluorobenzyl)-6-(methylamino)-9H-purine [BW A78U (1)], a novel, orally active anticonvulsant with potent ac-


BW A78U (1)
tivity against maximal electroshock-induced seizures (MES) in rats and mice. ${ }^{5,6}$ Compound 1 has activity against MES in the rat with an oral $E D_{50}=2.5 \mathrm{mg} / \mathrm{kg}$. The oral $\mathrm{LD}_{50}$ is greater than $1000 \mathrm{mg} / \mathrm{kg}$. Compound 1 did not produce tolerance to its anticonvulsant effect in mice under conditions where phenytoin was ineffective against MES. Compared to commonly used anticonvulsants, the structure of 1 is unique. ${ }^{7}$

A variety of biological activities have been reported for 9 -benzylpurines. Certain derivatives have antianginal, ${ }^{8}$ bronchodilating, ${ }^{8}$ or antiinflammatory ${ }^{9}$ activity, and 9-(2-chloro- 6 -fluorobenzyl)adenine is a potent anticoccidial agent. ${ }^{10}$ Other 9 -benzylpurines have activity as inhibitors of purine nucleoside phosphorylases ${ }^{11,12}$ or adenosine deaminase. ${ }^{13-16}$ Some of the earliest research on 9-benzylpurines was targeted to anticancer activity. ${ }^{17}$

[^0]Scheme I


We recently reported the weak antiviral activity of some $9-[(a m i n o a c y l a m i d o)$ benzyl $]$ purines. ${ }^{18}$ This class of com-
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Table I. Physical Properties of 4-Chloro-5,6-diaminopyrimidines


| no. | X | R | method | yield, \% | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | formula ${ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | $\mathrm{CH}\left(\mathrm{CH}_{3}\right)-R$ | H | $\mathrm{A}^{\text {b }}$ | 85 | 168-171 ${ }^{\text {c }}$ | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{ClN}_{4}$ |
| 3 | $\mathrm{CH}\left(\mathrm{CH}_{3}\right)-\mathrm{S}$ | H | $\mathrm{A}^{\text {d }}$ | 65 | 170-172 ${ }^{\text {e }}$ | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{ClN}_{4}$ |
| 4 | $\mathrm{CH}_{2}$ | 4-F | A | 84 | 240-242f | $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{ClFN}_{4}$ |
| 5 | $\mathrm{CH}_{2}$ | $3-\mathrm{Cl}$ | $\mathrm{A}^{\text {b }}$ | 85 | 197-200 ${ }^{\text {g }}$ | $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{Cl}_{2} \mathrm{~N}_{4}$ |
| 6 | $\mathrm{CH}_{2}$ | $3,4-\mathrm{Cl}_{2}$ | $\mathrm{A}^{\text {b }}$ | $23^{g}$ | 195-197 ${ }^{\text {h }}$ | $\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{Cl}_{3} \mathrm{~N}_{4}$ |
| 7 | $\mathrm{CH}_{2}$ | $3-\mathrm{CH}_{3}$ | $\mathrm{A}^{\text {b }}$ | 83 | 184-186 ${ }^{\text {i }}$ | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{ClN}_{4}$ |
| 8 | $\mathrm{CH}_{2}$ | 3-F | A | 75 | 197-200 ${ }^{\text {e }}$ | $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{ClFN}_{4}$ |
| 9 | $\mathrm{CH}_{2}$ | $3-\mathrm{OCH}_{3}$ | $\mathrm{A}^{\text {b }}$ | 84 | 131-133 ${ }^{\text {i }}$ | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{ClN}_{4} \mathrm{O}$ |
| 10 | $\mathrm{CH}_{2}$ | $2-\mathrm{Cl}$ | A | 79 | 197.5-199 ${ }^{\text {e }}$ | $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{Cl}_{2} \mathrm{~N}_{4}$ |
| 11 | $\mathrm{CH}_{2}$ | $2-\mathrm{CH}_{3}$ | A | 72 | 199-200.5 ${ }^{\text {e }}$ | $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{ClN}_{4}$ |
| 12 | $\mathrm{CH}_{2}$ | 2-F | A | 82 | 220-223f | $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{ClFN}_{4}$ |
| 13 | $\mathrm{CH}_{2}$ | $2-\mathrm{OCH}_{3}$ | A | 87 | 147-148 | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{ClN}_{4} \mathrm{O}$ |

${ }^{a}$ All compounds were analyzed for $\mathrm{C}, \mathrm{H}, \mathrm{N},{ }^{b}$ The cooled reaction solution was spin evaporated in vacuo to give the crude product. ${ }^{c}$ Recrystallized from EtOAc. ${ }^{d}$ The cooled reaction solution was spin evaporated in vacuo. The residual solid was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, washed with $\mathrm{H}_{2} \mathrm{O}$, and evaporated to give the crude product. ${ }^{e}$ Recrystallized from EtOAc-cyclohexane. ${ }^{f}$ Recrystallized from EtOH. ${ }^{8}$ Recrystallized from MeOH . ${ }^{h} \mathrm{Mp} 198-200^{\circ} \mathrm{C}$ reported for this compound in ref 22 . ${ }^{i}$ Recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$.

Table II. Physical Properties of 6-Chloropurines


| no. | R | isomer | method | yield, \% | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | formula ${ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | H | 9 | B | 54 | 84-86 ${ }^{\text {b,c }}$ | $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{ClN}_{4}$ |
| 15 | 4-Cl | 9 | B | 59 | 133-135 ${ }^{\text {b }}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{~N}_{4}$ |
| 16 | 4-Cl | 7 | C | 19 | 168-170 ${ }^{\text {b }}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{~N}_{4}$ |
| 17 | $4-\mathrm{CH}_{3}$ | 9 | B | 45 | 132-134 ${ }^{\text {d }}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4}$ |
| 18 | $4-\mathrm{CH}_{8}$ | 7 | C | 25 | 148-149 ${ }^{\text {e }}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4}$ |
| 19 | 4-F | 9 | D | 92 | 118-120 ${ }^{f}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{ClFN}_{4}$ |
| 20 | $4-\mathrm{OCH}_{3}$ | 9 | B8 | 28 | 122-124 ${ }^{\text {b }}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4} \mathrm{O}$ |
| 21 | $4-\mathrm{OCH}_{3}$ | 7 | $\mathrm{C}^{g}$ | 9 | 134-136 ${ }^{\text {h }}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4} \mathrm{O}$ |
| 22 | $3-\mathrm{CF}_{3}$ | 9 | B | $48^{\prime}$ | 116-117 | $\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{ClF}_{3} \mathrm{~N}_{4}$ |
| 23 | $3-\mathrm{CF}_{3}$ | 7 | C | $13^{i}$ | 185-186 | $\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{ClF}_{3} \mathrm{~N}_{4}$ |
| 24 | $3-\mathrm{I}$ | 9 | B | 26 | 156-156.5 ${ }^{\text {i }}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{ClIN}_{4}$ |
| 25 | 3-I | 7 | C | 10 | $170.5-171^{i}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{ClIN}_{4}$ |
| 26 | $3-\mathrm{Br}$ | 9 | B | 49 | $109-110^{i}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{BrClN}_{4}$ |
| 27 | $3-\mathrm{Br}$ | 7 | C | 13 | $151.5-153{ }^{\text {i }}$ | $\mathrm{C}_{22} \mathrm{H}_{8} \mathrm{BrClN}_{4}$ |
| 28 | $3-\mathrm{Cl}$ | 9 | D | $70^{j}$ | 90-92 | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{Cl}_{2} \mathrm{~N}_{4}$ |
| 29 | $3,5-\mathrm{Cl}_{2}$ | 9 | B | 27 | 164-165 ${ }^{\text {i }}$ | $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{Cl}_{3} \mathrm{~N}_{4}$ |
| 30 | $3,5-\mathrm{Cl}_{2}$ | 7 | C | 4 | $168-170^{i}$ | $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{Cl}_{3} \mathrm{~N}_{4}$ |
| 31 | $3,4-\mathrm{Cl}_{2}$ | 9 | D | 57 | 151-153 ${ }^{\text {d,k }}$ | $\mathrm{C}_{12} \mathrm{H}_{7} \mathrm{Cl}_{3} \mathrm{~N}_{4}$ |
| 32 | $3-\mathrm{CH}_{3}$ | 9 | D | 81 | 90-91 ${ }^{l}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4}$ |
| 33 | 3-F | 9 | D | 93 | 103-105 ${ }^{i}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{ClFN}_{4}$ |
| 34 | $3-\mathrm{OCH}_{3}$ | 9 | D | 92 | 103-104 ${ }^{\text {l }}$ | $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{ClN}_{4} \mathrm{O}$ |
| 35 | $3-\mathrm{OCH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | 9 | exp | 41 | 79-83 ${ }^{\text {m }}$ | $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{ClN}_{4} \mathrm{O}$ |
| 36 | $2-\mathrm{F}$ | 9 | D | 95 | 97-99 ${ }^{\prime}$ | $\mathrm{C}_{12} \mathrm{H}_{8} \mathrm{ClFN}_{4}$ |

${ }^{a}$ All compounds were analyzed for $\mathrm{C}, \mathrm{H}, \mathrm{N} .{ }^{b}$ Recrystallized from $\mathrm{C}_{6} \mathrm{H}_{6}$-petroleum ether ( $30-60{ }^{\circ} \mathrm{C}$ ). ${ }^{c} \mathrm{Mp} 86-87{ }^{\circ} \mathrm{C}$ reported for this compound in ref 17. ${ }^{d}$ Recrystallized from MeOH. ${ }^{e}$ Recrystallized from $2-\mathrm{PrOH}$. ${ }^{f}$ Recrystallized from cyclohexane. ${ }^{\boldsymbol{g}}$ For the starting 4 -methoxybenzyl bromide, see: Woodward, R. B. J. Am. Chem. Soc. 1940, 62, 1478. ${ }^{h}$ Recrystallized from EtOAc. ${ }^{i}$ Recrystallized from EtOAc-cyclohexane. ${ }^{j}$ Recrystallized from toluene-petroleum ether $\left(30-60^{\circ} \mathrm{C}\right) .{ }^{k} \mathrm{Mp} 149-151{ }^{\circ} \mathrm{C}$ reported for this compound in ref 22. ${ }^{l}$ Recrystallized from heptane. ${ }^{m}$ Recrystallized from pentane.
pound also binds to the benzodiazepine receptor. ${ }^{19} \mathrm{Be}$ cause some agents that bind to this receptor are used in the treatment of epilepsy, ${ }^{7,20}$ we tested a variety of purines for anticonvulsant activity. The chemistry and struc-
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ture-activity relationships associated with the development of 9 -benzylpurines with anticonvulsant activity are reported.

Chemistry. Most of the 9 -substituted purines were prepared by one of two methods (Scheme I). Amination of 5 -amino- 4,6 -dichloropyrimidine (I) with the appropriate amine by modification of a general literature method ${ }^{21,22}$ gave the 4 -substituted pyrimidines II (Table I). Condensation of II with triethyl orthoformate ${ }^{23}$ and ethane-
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Table III. Physical Properties of 6,9-Disubstituted Purines


| no. | $\mathrm{R}^{6}$ | $\mathrm{R}^{9}$ | method | yield, \% | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ | formula ${ }^{\text {a }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 37 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | E | $87^{\text {b }}$ | 126-128 ${ }^{\text {c }}$ | $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~N}_{5}$ |
| 38 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{3}$ | $\exp$ | $26^{\text {d }}$ | 113-114 ${ }^{\text {e }}$ | $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{~N}_{5}$ |
| 39 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{\dagger}$ | 89 | 161-163, ${ }^{\text {¢ }}$ | $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{5}$ |
| 40 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{i}$ | $82^{b}$ | 70-72 | $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{5}$ |
| 41 | $\mathrm{N}\left(\mathrm{CH}_{2}\right)_{2}$ | $\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{6} \mathrm{H}_{5} \cdot R^{j}$ | E | $48^{k}$ | 104-105 | $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{5}$ |
| 42 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{6} \mathrm{H}_{5} \cdot S^{l}$ | E | $59^{k}$ | 104-105.5 | $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{5}$ |
| 43 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{11}{ }^{m}$ | F, exp | $81^{n}$ | 78-80 | $\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{5}$ |
| 44 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{3} \mathrm{O}^{\circ}$ | $\mathrm{D}, \mathrm{E},{ }^{p} \exp$ | $57^{k}$ | 118-119 | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}$ |
| 45 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{3} \mathrm{~S}^{9}$ | B, E | $80^{k}$ | 128-129 | $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{~S}$ |
| 46 | $\mathrm{NHCH}_{3}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{r}$ | $81^{\text {b }}$ | 130-132 ${ }^{\text {s }}$ | $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{5}$ |
| 47 | $\mathrm{NH}_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $t$ | $34^{g}$ | 231-234 | $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{~N}_{5}$ |
| 48 | H | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\stackrel{u}{ }$ | $69^{*}$ | 96-98 | $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4}$ |
| 49 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{r}$ | $62^{w}$ | 55.5-57 | $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{5}$ |
| 50 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{x}$ | $43^{w}$ | 52-55.5 | $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{~N}_{5}$ |
| 51 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{5} \mathrm{H}_{9}{ }^{y}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{x}$ | $29^{\text {w }}$ | 58-60 | $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{~N}_{5}$ |
| 52 | $\mathrm{N}_{\left(\mathrm{CH}_{2}\right)_{3}}{ }^{\text {b }}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{\boldsymbol{z}}$ | $37^{a a}$ | 168-170 | $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{5}$ |
| 53 | $\mathrm{NHC}_{3} \mathrm{H}_{5}{ }^{\text {b }}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{E}^{x}$ | $61^{\text {ce }}$ | 158-159 | $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{5}$ |
| 54 | $\mathrm{NHC}_{6} \mathrm{H}_{5}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | F | $77^{5}$ | 187-189 | $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{~N}_{5}$ |
| 55 | $\mathrm{NHCH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $\mathrm{F}^{\text {dd }}$ | $89^{e e}$ | 118-120 | $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ |
| 56 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CHO}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | exp | $58 f$ | 162-164 | $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}$ |
| 57 | $\mathrm{OH}^{\text {ge }}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $h h$ | 54 | 295-298 ${ }^{8}$ | $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{O}$ |
| 58 59 | $\mathrm{OCH}_{3}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | ${ }_{\text {ii }}{ }^{\text {exp }}$ | $71^{d}$ | 114-116 | $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}$ |
| 59 | $\mathrm{SCH}_{3}$ | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | exp ${ }^{j j}$ | $64{ }^{\text {ff }}$ | $115-116^{k k}$ | $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{~S}$ |

${ }^{a}$ All compounds were analyzed for $\mathrm{C}, \mathrm{H}, \mathrm{N} .{ }^{b}$ Recrystallized from toluene-petroleum ether $\left(30-60^{\circ} \mathrm{C}\right) .{ }^{c} \mathrm{Mp} 131-132{ }^{\circ} \mathrm{C}$ reported for this compound in ref $17 .{ }^{d}$ Recrystallized from heptane. ${ }^{e} \mathrm{Mp} 114-115{ }^{\circ} \mathrm{C}$ reported for this compound in ref 28 by a different method. ${ }^{\prime}$ For the starting purine see ref $22 .{ }^{s}$ Recrystallized from $\mathrm{EtOH} .{ }^{h} \mathrm{Mp} 168-169^{\circ} \mathrm{C}$ reported for this compound in ref 22 . ${ }^{i}$ For the starting purine, see: Schaeffer, H. J.; Johnson, R. N. J. Med. Chem. 1968, 11, 21. ${ }^{j}[\alpha]^{20}{ }_{D}-9.2^{\circ}(c 1.0, \mathrm{MeOH}) .{ }^{k}$ Recrystallized from cyclohexane. ${ }^{\ell}[\alpha]^{20}{ }^{2}+8.6^{\circ}$ (c 1.0, MeOH). ${ }^{m}$ Cyclohexylmethyl substituent. ${ }^{n}$ Recrystallized from pentane. ${ }^{\circ}$ Furfuryl substituent. ${ }^{p}$ The crude product was purified by flash chromatography as in method B. ${ }^{q} 2$-Thienyl substituent. ${ }^{r}$ The aqueous alkylamine was formed by dissolving a 3 -fold excess of alkylamine hydrochloride in 2 molar equiv of $1 \mathrm{~N} \mathrm{NaOH} .{ }^{s} \mathrm{Mp} 133-134^{\circ} \mathrm{C}$ reported for this compound by Fuji, T.; Saito, T. Chem. Pharm. Bull. 1985, 33, 3635. ${ }^{t}$ Prepared by the method of Carraway, K. L.; Huang, P. C.; Scott, T. G. In Synthetic Procedures in Nucleic Acid Chemistry; Zorbach, W. W., Tipson, R. S., Eds.; Interscience: New York, 1968; Vol. 1, p 3; mp $233.5-235.5^{\circ} \mathrm{C}$. "Prepared by the method of ref $17, \mathrm{mp} 100-101^{\circ} \mathrm{C} .{ }^{v}$ Recrystallized from $\mathrm{CCl}_{4}$-pentane. ${ }^{w}$ Recrystallized from petroleum ether ( $30-60^{\circ} \mathrm{C}$ ). ${ }^{x} \mathrm{~A} 3$-fold excess of the amine was used in the reaction. ${ }^{y}$ Cyclopentyl substituent. ${ }^{z}$ An equimolar amount of triethylamine and trimethyleneimine was used, and the product was purified by flash chromatography. ${ }^{a a}$ Recrystallized from cyclohexane-toluene. ${ }^{b b}$ Cyclopropyl substituent. ${ }^{c c}$ Recrystallized from EtOAc. ${ }^{d d}$ A 2 -fold excess of triethylamine was used in the reaction. ${ }^{e e}$ Recrystallized from toluene. "f Recrystallized from cyclo-hexane-EtOAc. ${ }^{\text {gg }}$ Present as the oxo tautomer. ${ }^{h h}$ Prepared by the method of ref $17, \mathrm{mp}>260^{\circ} \mathrm{C}$. ${ }^{\text {ii }}$ Prepared by the method of ref $17, \mathrm{mp}$ $128{ }^{\circ} \mathrm{C}$. ${ }^{j j}$ For the starting 6 -thionopurine see ref $29 .{ }^{k k} \mathrm{Mp} 117-118{ }^{\circ} \mathrm{C}$ reported for this compound in ref 30 by a different method.
sulfonic acid ${ }^{24}$ afforded the intermediate 6 -chloro- 9 -sub-stituted-purines III. Alternatively, 6 -chloropurine (IV) was alkylated with the appropriate benzyl halide to give a mixture of the 7 -benzyl-6-chloropurines ( V ) and 9-benzyl-6-chloropurines (III). ${ }^{17}$ Isomers were easily separated by flash chromatography ${ }^{25}$ to give the 9 -isomers III in $26-59 \%$ yields (Table II). The 6 -chloro- 9 -substitut-ed-purines III were reacted with the appropriate amine to give most of the 6-(alkylamino)-9-benzylpurines (VI) in Tables III and IV. The UV spectra of the 6-(dimethylamino) purines were compatible with 9 -substitution. ${ }^{26}$
The 6-( $N$-methylformamido)purine 56 (Table III) was prepared from 46 and acetic-formic anhydride. ${ }^{27}$ The 9 -(aminobenzyl)purines 68 and 87 (Table IV) were prepared from 64 and 86 , respectively, by catalytic hydrogenation. The 4 -hydroxy compound 66 was prepared from 67 by catalytic hydrogenolysis of the $O$-benzyl moiety. Compound 67 was prepared from 4 -(benzyloxy)benz-

[^1]aldehyde as described in the Experimental Section. The 3-benzyloxy compound 81, available in four steps from 3 -hydroxybenzyl alcohol, was cleaved with HBr in AcOH to give 80.

## Biological Results and Discussion

Compounds in Tables V-VII were evaluated for anticonvulsant activity in the maximal electroshock-induced seizure (MES) screen in Sprague-Dawley male rats. ${ }^{4}$ The compounds were tested initially at $25 \mathrm{mg} / \mathrm{kg} \mathrm{ip}$, and if activity was high, an $\mathrm{ED}_{50}$ was determined. The most active compounds were also tested orally.

The parent 9-benzylpurine 37 (Table V) protected rats against MES with an oral $E D_{50}=12 \mathrm{mg} / \mathrm{kg}$ and an ip $E D_{50}$ $=6 \mathrm{mg} / \mathrm{kg}$. Variation of the purine 9 -substituent had a substantial effect on anticonvulsant activity. The 9 -methyl substituent (see 38) gave a compound with about one-fifth the activity of 37 . The 9 -phenyl compound 39 was not active as it gave no protection at $12.5 \mathrm{mg} / \mathrm{kg}$. Substitution of the one-carbon bridge with a two-carbon bridge between the phenyl and purine rings (see 40) resulted in about a 5 -fold loss in activity. Substitution of a methyl group on the methylene bridge (see 41 and 42) gave compounds that were less active than 37 . The 9 -cyclohexylmethyl compound 43 was completely inactive at $25 \mathrm{mg} / \mathrm{kg}$ ip. How ever, the furfuryl and thenyl analogues 44 and 45 showed one-third and about one-half the activity, respectively, of

Table IV. Physical Properties of 9-Benzyl-6-(dimethylamino)purines

${ }^{a}$ All compounds were analyzed for $\mathrm{C}, \mathrm{H}, \mathrm{N},{ }^{b} \mathrm{~A}$ solution of 2.2 M dimethylamine and ethanol was used instead of $40 \%$ aqueous dimethylamine. ${ }^{c}$ Recrystallized from EtOH. ${ }^{d}$ Recrystallized from toluene-petroleum ether ( $30-60{ }^{\circ} \mathrm{C}$ ). ${ }^{e}$ Recrystallized from cyclohexane. $f$ For the starting purine see ref $13 .{ }^{8} \mathrm{Mp} 211{ }^{\circ} \mathrm{C}$ reported for this compound in ref $13 .{ }^{h}$ For the starting purine see ref $16 .{ }^{i}$ Recrystallized from $\mathrm{CHCl}_{3}-\mathrm{EtOH} .{ }^{j}$ Recrystallized from EtOAc. ${ }^{k}$ Recrystallized from cyclohexane-EtOAc. ${ }^{l}$ Melts with resolidification and remelts $93-95$ ${ }^{\circ} \mathrm{C}$. ${ }^{m}$ Recrystallized from heptane. ${ }^{n}$ Recrystallized from MeOH . ${ }^{\circ}$ For the starting purine see ref $14 .{ }^{p} \mathrm{Mp} 156-157{ }^{\circ} \mathrm{C}$ reported for this compound in ref $14 .{ }^{q}$ Recrystallized from toluene. ${ }^{r} \mathrm{Mp} 169-171^{\circ} \mathrm{C}$ reported in ref 15.
the parent 9 -benzylpurine 37 . Compounds that contain a flat, aromatic ring bonded to the 9 -substituted purine by a methylene group possessed good anticonvulsant activity.

Variation of the 6 -substituent of $\mathbf{3 7}$ had a substantial effect on the relative potency of 9-benzylpurines in protecting rats against MES (Table VI). Removal of a single methyl group (see 46) had no effect on activity; however, the 6 -aminopurine 47 and 6 -hydrogen purine 48 were about 5 -fold less active. Of eight other 6 -substituted aminopurines, only $49\left(\mathrm{~N}^{\left.\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 53\left(\mathrm{NHC}_{3} \mathrm{H}_{5}\right) \text {, and } 55}\right.$ ( $\mathrm{NHCH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ ) retained significant anticonvulsant activity. Although the hypoxanthine derivative 57 was essentially inactive, the 6 -methoxypurine 58 had activity comparable to that of phenytoin.
Analogues of 37 that contain substitutions in the phenyl moiety are listed in Table VII. Analogues that contain a substituent in the para position were almost uniformly inactive at $25 \mathrm{mg} / \mathrm{kg} \mathrm{ip}$. Only the small fluoro (62) and hydrophilic amino (68) substituents gave compounds with detectable anticonvulsant activity at $25 \mathrm{mg} / \mathrm{kg}$. Thus, with the exception of the amino and fluoro, substituents in the para position of 37 did not impart anticonvulsant activity to the compounds.

Several meta-substituted analogues of 37 had good anticonvulsant activity. The 3 -chloro analogue 72 was slightly more active than 37 when administered ip. Addition of a second chlorine in the para position (see 74) resulted in complete loss of activity. This result further substantiated
the lack of tolerance for substituents in the para position. Substitution of trifluoromethyl (69), iodo (70), bromo (71), fluoro (76), or methoxy (77) for the chloro in 72 resulted in only a 2 - to 3 -fold loss in anticonvulsant activity. The less lipophilic nitro (78), cyano (79), and hydroxy (80) meta-substituted analogues of 37 had little or no anticonvulsant activity at $25 \mathrm{mg} / \mathrm{kg}$. In general, meta-substituted analogues of 37 retained good anticonvulsant activity when the substituent was electron withdrawing and lipophilic.

Introduction of a second meta chloro group gave 73, which had good anticonvulsant activity with an ip $\mathrm{ED}_{50}$ $=7 \mathrm{mg} / \mathrm{kg}$. However, 73 was extremely toxic to mice. Its ip $\mathrm{LD}_{50}$ was less than $0.5 \mathrm{mg} / \mathrm{kg}$. The monochloro analogue 72 was 100 -fold less toxic with an ip $\mathrm{LD}_{50}=54$ $\mathrm{mg} / \mathrm{kg}$, whereas the parent 9 -benzylpurine 37 was relatively innocuous with an ip $\mathrm{LD}_{50}>500 \mathrm{mg} / \mathrm{kg}$.

Several ortho-substituted analogues of 37 retained good anticonvulsant activity. There was no apparent correlation of activity with electronic, lipophilic, or size parameters of the ortho substituents. The fluoro- and methyl-substituted analogues ( 84 and 83 , respectively) were only about $50 \%$ less active than the parent 37.

Potent anticonvulsant activity resided in compounds that contained a benzyl group at the 9 -position of 6 -(me-thylamino)- or 6-(dimethylamino)purine. There was little or no tolerance for para substitution, whereas analogues with electron-withdrawing, lipophilic, meta substituents retained substantial anticonvulsant activity.

Table V. Anticonvulsant Activity of 9-Substituted 6-(Dimethylamino)purines ${ }^{a}$


| no. | R | $\begin{gathered} \mathrm{MES} \mathrm{ED} \mathrm{ED}_{50}, \\ \mathrm{mg} / \mathrm{kg}^{b-d} \end{gathered}$ |  |
| :---: | :---: | :---: | :---: |
|  |  | ip | po |
| 37 | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | $6 \pm 1$ | $12 \pm 3$ |
| 38 | $\mathrm{CH}_{3}$ | (33) |  |
| 39 | $\mathrm{C}_{6} \mathrm{H}_{5}$ | ( $0^{\text {e }}$ ) |  |
| 40 | $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | (33) |  |
| 41 | $\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{6} \mathrm{H}_{5} \cdot R$ | (33) |  |
| 42 | $\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{6} \mathrm{H}_{5} \cdot \mathrm{~S}$ | (33) |  |
| 43 | $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{11}{ }^{\text {I }}$ | (0) |  |
| 44 | $\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{3} \mathrm{O}^{8}$ | 18 |  |
| 45 | $\mathrm{CH}_{2} \mathrm{C}_{4} \mathrm{H}_{3} \mathrm{~S}^{h}$ | $10 \pm 2.5$ |  |
| phenytoin |  | $9.6 \pm 2.0$ | $20 \pm 3$ |

${ }^{a}$ The compounds were tested for their ability to protect Sprague-Dawley male rats against maximal electroshock-induced seizures (MES) as described in ref 4. The $\mathrm{ED}_{50}$ was the dose needed to protect $50 \%$ of the animals against the hind-limb extensor component and were calculated by the method of Miller, L. C.; Tainter, M. L. Proc. Soc. Exp. Biol. Med. 1944, 57, 261. ${ }^{6}$ The compounds were administered as solutions or fine dispersions in water or $5 \%$ methylcellulose. ${ }^{c}$ Values in parentheses are percent inhibition at $25 \mathrm{mg} / \mathrm{kg}$. ${ }^{d}$ Where $\mathrm{ED}_{50}$ values are presented with a standard error a minimum of 12 animals were used per dose level with four doses per compound. $\mathrm{ED}_{50}$ values without standard error were determined by using three doses of compound with six animals per point. ${ }^{e}$ Inhibition at $12.5 \mathrm{mg} / \mathrm{kg}$. ${ }^{f}$ Cyclohexyl substituent. ${ }^{8}$ Furfuryl substituent. ${ }^{h} 2$-Thenyl substituent.

Table VI. Anticonvulsant Activity of 6-Substituted 9 -Benzylpurines ${ }^{a}$


| no. | R | MES ED ${ }_{50}, \mathrm{mg} / \mathrm{kg}^{b-d}$ |  |
| :---: | :---: | :---: | :---: |
|  |  | ip | po |
| 37 | $\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$ | $6 \pm 1$ | $12 \pm 3$ |
| 46 | $\mathrm{NHCH}_{3}$ | $5 \pm 1$ | $12 \pm 2$ |
| 47 | $\mathrm{NH}_{2}$ | (33) |  |
| 48 | H | (33) |  |
| 49 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{2} \mathrm{CH}_{3}$ | 8 | 30 |
| 50 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ | (17) |  |
| 51 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{C}_{6} \mathrm{H}_{9}{ }^{e}$ | (17) |  |
| 52 | $\mathrm{N}\left(\mathrm{CH}_{2}\right)_{3}$ | (17) |  |
| 53 | $\mathrm{NHC}_{3} \mathrm{H}_{5}{ }^{\text {f }}$ | $8.6 \pm 0.7$ |  |
| 54 | $\mathrm{NHC}_{6} \mathrm{H}_{5}$ | (0) |  |
| 55 | $\mathrm{NHCH}_{2} \mathrm{CH}_{2} \mathrm{OH}$ | (84) | (17) |
| 56 | $\mathrm{N}\left(\mathrm{CH}_{3}\right) \mathrm{CHO}$ | (0) |  |
| 57 | OH (oxo) | (17) |  |
| 58 | $\mathrm{OCH}_{3}$ | $10 \pm 1.6$ | $21 \pm 3.6$ |
| 59 | $\mathrm{SCH}_{3}$ | (30) |  |

${ }^{a-d}$ See footnotes $a-d$ in Table V. ${ }^{e}$ Cyclopentyl substituent. ${ }^{f}$ Cyclopropyl substituent.

## Experimental Section

Melting points were taken in capillary tubes on a Mel-Temp block or a Thomas-Hoover Unimelt and were uncorrected. UV spectra were measured on a Unicam SP 800 or a Cary 118 UV-vis spectrophotometer. NMR data were recorded on a Varian XL-100-15-FT, a Varian FT-80A, a Varian T-60, or a Hitachi Per-kin-Elmer R-24 spectrometer with $\mathrm{Me}_{4} \mathrm{Si}$ as an internal standard. Each analytical sample had spectral data compatible with its assigned structure and moved as a single spot on TLC. TLC's

Table VII. Anticonvulsant Activity of 9-Benzyl-6-(dimethylamino)purines ${ }^{a}$

|  |  |  |  |
| :---: | :---: | :---: | :---: |
| no. | R | MES ED $50, \mathrm{mg} / \mathrm{kg}^{\text {b-d }}$ |  |
|  |  | ip | po |
| 37 | H | $6 \pm 1$ | $12 \pm 3$ |
| 60 | 4-Cl | (0) |  |
| 61 | $4-\mathrm{CH}_{3}$ | (0) |  |
| 62 | 4-F | (33) |  |
| 63 | $4-\mathrm{OCH}_{3}$ | (0) |  |
| 64 | $4-\mathrm{NO}_{2}$ | (0) |  |
| 65 | $4-\mathrm{CN}$ | (0) |  |
| 66 | 4-OH | (0) |  |
| 68 | $4-\mathrm{NH}_{2}$ | 25 |  |
| 69 | $3-\mathrm{CF}_{3}$ | 12 |  |
| 70 | 3-I | 11 |  |
| 71 | $3-\mathrm{Br}$ | 13 |  |
| 72 | $3-\mathrm{Cl}$ | $4.5 \pm 0.6$ | $18 \pm 2$ |
| 73 | $3,5-\mathrm{Cl}_{2}$ | 7 |  |
| 74 | $3,4-\mathrm{Cl}_{2}$ | (0) |  |
| 75 | $3-\mathrm{CH}_{3}$ | 25 |  |
| 76 | 3-F | 8.4 |  |
| 77 | $3-\mathrm{OCH}_{3}$ | 12.5 |  |
| 78 | $3-\mathrm{NO}_{2}$ | (17) |  |
| 79 | $3-\mathrm{CN}$ | (17) |  |
| 80 | $3-\mathrm{OH}$ | (0) |  |
| 81 | $3-\mathrm{OCH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}$ | (33) |  |
| 82 | $2-\mathrm{Cl}$ | (84) |  |
| 83 | $2-\mathrm{CH}_{3}$ | 9 |  |
| 84 | 2-F | $8.8 \pm 1.7$ |  |
| 85 | $2-\mathrm{OCH}_{3}$ | (0) |  |
| 86 | $2-\mathrm{NO}_{2}$ | 25 |  |
| $87$ | $2-\mathrm{NH}_{2}$ | $20 \pm 2$ |  |
| phenytoin |  | $9.6 \pm 2.0$ | $20 \pm 3$ |

${ }^{a-d}$ See footnotes $a-d$ in Table V.
were developed on Whatman $200 \mu$ MK6F plates of silica gel with fluorescent indicator. Preparative flash chromatography ${ }^{25}$ was performed on silica gel $60(40-63 \mu \mathrm{M}$, E.Merck, no. 9385). All compounds were analyzed for $\mathrm{C}, \mathrm{H}$, and N and gave values within $0.4 \%$ of theoretical values. Elemental analyses were performed by Atlantic Microlab, Inc.

Method A. 5-Amino-4-chloro-6-[(3-fluorobenzyl)amino]pyrimidine (8). A mixture of 5 -amino-4,6-dichloropyrimidine ( $10.0 \mathrm{~g}, 61.0 \mathrm{mmol}$ ), ( 3 -fluorobenzyl)amine ( $97 \%$ ) ( $7.9 \mathrm{~g}, 61.2$ mmol ), 1-BuOH ( 150 mL ), and $\mathrm{Et}_{3} \mathrm{~N}(6.5 \mathrm{~g}, 64.2 \mathrm{mmol}$ ) was refluxed with stirring for 21 h . The dark solution was cooled to give a solid that was collected on a Büchner funnel and washed with cyclohexane. The white solid was dispersed in $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$, collected, and dried to give $11.5 \mathrm{~g}(75 \%)$ of $8, \mathrm{mp} 196-200^{\circ} \mathrm{C}$. Recrystallization of a portion from EtOAc-cyclohexane gave the analytical sample: $\operatorname{mp} 197-200^{\circ} \mathrm{C}$; UV $(0.1 \mathrm{~N} \mathrm{HCl}+10 \%$ EtOH $)$ $\lambda_{\text {max }} 304 \mathrm{~nm}(\epsilon 12500)$; UV ( $0.1 \mathrm{~N} \mathrm{NaOH}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 263$ $\mathrm{nm}(\epsilon 9200), 269$ (8900), 291 (8900); NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 7.73$ (s, 1 H , pyrimidine H ), 7.5-7.0 (complex m, 5 H , aromatic $\mathrm{H}+\mathrm{NH}$ ), 5.05 (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 4.65 (d, collapsed to s with $\mathrm{D}_{2} \mathrm{O}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ).

Method B. 9-(3-Bromobenzyl)-6-chloro-9H-purine (26). A mixture of 6 -chloropurine ( $3.09 \mathrm{~g}, 20 \mathrm{mmol}$ ), DMSO ( 50 mL ), anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(3.45 \mathrm{~g}, 25 \mathrm{mmol})$, and 3-bromobenzyl bromide ( $5.0 \mathrm{~g}, 20 \mathrm{mmol}$ ) was stirred at ambient temperature for 39 h . The reaction solution was decanted from the solids, poured into ice water, and acidified to pH 5 with HOAc ( 0.5 mL ). The mixture was extracted with EtOAc $(4 \times 200 \mathrm{~mL})$. The combined extracts were washed with an equal volume of $\mathrm{H}_{2} \mathrm{O}$, dried $\left(\mathrm{MgSO}_{4}\right)$, and spin evaporated in vacuo. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 50 mL ) and added to silica gel $60(5 \mathrm{~g})$. This mixture was spin evaporated in vacuo, and the residual solid was introduced on a column of silica gel 60 wetted with EtOAc-cyclohexane, 1:2. The column was eluted with EtOAc-cyclohexane, 1:2, by using the flash chromatography technique. The fractions containing the
higher $R_{f}$, major component were combined and spin evaporated in vacuo to give $3.20 \mathrm{~g}(49 \%)$ of 26 , which was a single spot on TLC. Recrystallization from EtOAc-cyclohexane gave the analytical sample: $\mathrm{mp} 109-110^{\circ} \mathrm{C}$; UV ( $\mathrm{pH} 7+10 \% \mathrm{EtOH}$ ) $\lambda_{\text {max }}$ $265.5 \mathrm{~nm}(\epsilon 10200)$; NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 8.86$ ( $\mathrm{s}, 1 \mathrm{H}$, purine H ), 8.80 (s, 1 H , purine H ), 7.7-7.2 (m, $4 \mathrm{H}, \mathrm{Ar} \mathrm{H}$ ), $5.55\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$ ).

Method C. 7-(3-Bromobenzyl)-6-chloro-7H-purine (27). After elution of the 9 -isomer 26 , the 7 -isomer was isolated by changing the elution solvent to EtOAc-cyclohexane, 1:1. Fractions containing the lower $R_{f}, 7$-isomer were collected, combined, and spin evaporated in vacuo to give $1.24 \mathrm{~g}(19 \%)$ of 27 . Recrystallization from EtOAc-cyclohexane gave the analytical sample: $\mathrm{mp} 151.5-153^{\circ} \mathrm{C}$; UV (pH $7+10 \% \mathrm{EtOH}$ ) $\lambda_{\text {max }} 268.5 \mathrm{~nm}(\epsilon 8500)$; NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 9.02(\mathrm{~s}, 1 \mathrm{H}$, purine H ), $8.84(\mathrm{~s}, 1 \mathrm{H}$, purine $\mathrm{H}), 7.7-7.0(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar} \mathrm{H}), 5.79\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$.

Method D. 6-Chloro-9-(3-fluorobenzyl)-9H-purine (33). A mixture of $8(10.0 \mathrm{~g}, 39.6 \mathrm{mmol}), \mathrm{HC}(\mathrm{OEt})_{3}(100 \mathrm{~mL})$, and $\mathrm{EtSO}_{3} \mathrm{H}(100 \mathrm{mg})$ was stirred at ambient temperature for 41 h . The solution was spin evaporated in vacuo. The residual solid was dissolved in EtOAc ( 200 mL ), washed with $5 \%$ aqueous $\mathrm{NaHCO}_{3}(50 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}(3 \times 50 \mathrm{~mL})$, and brine ( 50 mL ), and then dried $\left(\mathrm{MgSO}_{4}\right)$. The solution was spin evaporated in vacuo to a residue that was collected and washed with cyclohexane to give $9.77 \mathrm{~g}(94 \%)$ of $33, \mathrm{mp} 100-103^{\circ} \mathrm{C}$. Recrystallization from cy-clohexane-ethyl acetate gave the analytical sample: mp 103-105 ${ }^{\circ} \mathrm{C}$; UV (pH $\left.7+10 \% \mathrm{EtOH}\right) \lambda_{\max } 265 \mathrm{~nm}(\epsilon 11500), 269$ (sh) (11300); NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 8.87$ ( $\mathrm{s}, 1 \mathrm{H}$, purine H ), 8.82 ( $\mathrm{s}, 1$ H , purine H), 7.6-7.0 (complex m, $4 \mathrm{H}, \mathrm{Ar})$, $5.59\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$.

Method E. 9-(4-Fluorobenzyl)-6-(dimethylamino)-9Hpurine (62). A solution of $19(4.01 \mathrm{~g}, 15.2 \mathrm{mmol}), \mathrm{EtOH}(20 \mathrm{~mL})$, and $40 \%$ aqueous $\mathrm{Me}_{2} \mathrm{NH}(10 \mathrm{~mL})$ was stirred at ambient temperature for 15 h . The reaction mixture was spin evaporated in vacuo to remove the volatiles. The residue was dispersed in $\mathrm{H}_{2} \mathrm{O}$ $(50 \mathrm{~mL})$ and stirred for several minutes. The solids were collected, washed with $\mathrm{H}_{2} \mathrm{O}$, and sucked dry to give $3.90 \mathrm{~g}(94 \%)$ of $62, \mathrm{mp}$ $130-132{ }^{\circ} \mathrm{C}$. Recrystallization from cyclohexane gave $3.25 \mathrm{~g}(78 \%)$ of 62: $\mathrm{mp} 131-133^{\circ} \mathrm{C}$; UV ( $0.1 \mathrm{~N} \mathrm{HCl}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 270$ $\mathrm{nm}(\epsilon 19200)$; UV ( $0.1 \mathrm{~N} \mathrm{NaOH}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 277 \mathrm{~nm}(\epsilon$ 19100); NMR ( $\mathrm{Me}_{2} \mathrm{SO}-\mathrm{d}_{6}$ ) $\delta 8.28$ (s, 2 H , purine Hs ), $7.55-7.0$ ( m , $4 \mathrm{H}, \mathrm{Ar}), 5.40\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.47\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$.
Method F. 9-(Cyclohexylmethyl)-6-(dimethylamino)-9Hpurine (43). A mixture of $89(4.96 \mathrm{~g}, 19.8 \mathrm{mmol})$ and 125 mL of a solution of $\mathrm{Me}_{2} \mathrm{NH}$ in $\mathrm{EtOH}(100 \mathrm{~g} / \mathrm{L})$ was heated on a steam bath to effect dissolution. The reaction was left at ambient temperature for 17 h and spin evaporated in vacuo to remove the volatiles. The residue was dissolved in EtOAc and filtered to remove the insolubles. The EtOAc was evaporated to an oil that was dissolved in petroleum ether ( $30-60^{\circ} \mathrm{C}$ ), filtered through Celite (Preiser Scientific, Inc.) and spin evaporated in vacuo. The residual solid was recrystallized from pentane to give $4.19 \mathrm{~g}(81 \%)$ of 43: $\mathrm{mp} 78-80^{\circ} \mathrm{C}$; UV ( $0.1 \mathrm{~N} \mathrm{HCl}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 270 \mathrm{~nm}$ ( $\epsilon 17900$ ); $\mathrm{UV}(0.1 \mathrm{~N} \mathrm{NaOH}+10 \% \mathrm{EtOH}) \lambda_{\max } 278 \mathrm{~nm}(\epsilon 18500)$.
Method G. 9-(4-Aminobenzyl)-6-(dimethylamino)-9Hpurine (68). A mixture of $64(5.00 \mathrm{~g}, 16.7 \mathrm{mmol})$, HOAc ( 200 $\mathrm{mL})$, and $10 \% \mathrm{Pd} / \mathrm{C}(0.20 \mathrm{~g})$ was shaken in the presence of hydrogen at 2-3 atm for 1 h . The reaction mixture was filtered through Celite and concentrated under reduced pressure. The residue was dispersed in a minimum of EtOH and collected to give $4.14 \mathrm{~g}(92 \%)$ of $68, \mathrm{mp} 190-192{ }^{\circ} \mathrm{C}$. Recrystallization of a portion from EtOH gave the analytical sample: mp 193-196 ${ }^{\circ} \mathrm{C}$; UV ( $0.1 \mathrm{~N} \mathrm{HCl}+10 \%$ EtOH) $\lambda_{\text {max }} 268 \mathrm{~nm}(\epsilon 19300)$ UV ( 0.1 $\mathrm{N} \mathrm{NaOH}+10 \% \mathrm{EtOH}) \lambda_{\text {max }} 277 \mathrm{~nm}(\epsilon 21000) ; \mathrm{NMR}\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right)$ $\delta 8.25(\mathrm{~s}, 1 \mathrm{H}$, purine H), $8.15(\mathrm{~s}, 1 \mathrm{H}$, purine H), $7.07(\mathrm{~d}, 2 \mathrm{H}$, $J=8 \mathrm{~Hz}, \mathrm{Ar}), 6.52(\mathrm{~d}, 2 \mathrm{H}, J=8 \mathrm{~Hz}, \mathrm{Ar}), 5.18\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 5.07$ (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), $3.45\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$.
9-[3-(Benzyloxy)benzyl]-6-chloro-9H-purine (35). A mixture of 3-hydroxybenzyl alcohol ( $9.30 \mathrm{~g}, 75 \mathrm{mmol}$ ), powdered anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(10.5 \mathrm{~g}, 76 \mathrm{mmol})$, benzyltrimethylammonium hydroxide ( $1.5 \mathrm{~mL}, 40 \%$ solution in MeOH), and DMF ( 90 mL ) was stirred under nitrogen at ambient temperature. After 30 min , benzyl chloride ( $97 \%$ ) ( $8.3 \mathrm{~mL}, 70 \mathrm{mmol}$ ) was added, and the reaction was heated to $70^{\circ} \mathrm{C}$ for 18 h . The volatiles were removed by spin evaporation in vacuo, and the residual oil was dissolved in $\mathrm{Et}_{2} \mathrm{O}$ and washed with $\mathrm{H}_{2} \mathrm{O}, 1 \mathrm{~N} \mathrm{NaOH}$, and again with $\mathrm{H}_{2} \mathrm{O}$. The solution was dried $\left(\mathrm{MgSO}_{4}\right)$ and spin evaporated in vacuo to give $13.7 \mathrm{~g}(85 \%)$ of 3 -(benzyloxy)benzyl alcohol as an oil, which
was used without further purification. The benzyl alcohol (combined product from two reactions) was converted to the benzyl chloride and reacted with 6 -chloropurine as described for the para isomer 67 to give $14.5 \mathrm{~g}(30 \%)$ of $35, \mathrm{mp} 70-80^{\circ} \mathrm{C}$. Recrystallization of a portion from petroleum ether ( $60-68$ ${ }^{\circ} \mathrm{C}$ )-ethyl acetate gave the analytical sample: $\mathrm{mp} 79-83^{\circ} \mathrm{C}$; UV ( $\mathrm{pH} 7+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 268 \mathrm{~nm}(\epsilon 9800)$; NMR ( $\left.\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta$ 8.77 ( $\mathrm{s}, 1 \mathrm{H}$, purine H ), $8.74(\mathrm{~s}, 1 \mathrm{H}$, purine H ), $7.34(\mathrm{~s}, 5 \mathrm{H}, \mathrm{Ar})$, 7.02 (br m, $4 \mathrm{H}, \mathrm{Ar}$ ), 5.48 (s, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), 5.05 ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ).

6-(Dimethylamino)-9-methyl-9H-purine (38). A solution of 6 -(dimethylamino) purine ( $6.00 \mathrm{~g}, 36.8 \mathrm{mmol}$ ), $\mathrm{NaOMe}(2.08$ $\mathrm{g}, 38.5 \mathrm{mmol}$ ), and DMSO ( 13 mL ) was stirred at ambient temperature under a blanket of nitrogen for 1 h . Methyl iodide ( 2.30 $\mathrm{mL}, 3.0 \mathrm{mmol}$ ) was added, and the resultant mixture was stirred for 1 h . The reaction was diluted with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 700 \mathrm{~mL})$. The combined extracts were dried $\left(\mathrm{CaCl}_{2}\right)$ and spin evaporated in vacuo to give $5.06 \mathrm{~g}(78 \%)$ of crude 38. This material was introduced on a column of silica gel ( 55 $\mathrm{g}, 70-325 \mathrm{mesh}$ ) and eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}, 1: 1$. The appropriate fractions were combined and spin evaporated in vacuo to give $2.40 \mathrm{~g}(36 \%)$ of 38 , which was a single spot on TLC. Recrystallization from heptane gave 1.75 g ( $26 \%$ ) of analytically pure material: mp $113-114^{\circ} \mathrm{C}$ (lit. $.^{28} \mathrm{mp} 114-115{ }^{\circ} \mathrm{C}$ ); UV ( 0.1 $\mathrm{N} \mathrm{HCl}+10 \% \mathrm{EtOH}) \lambda_{\text {max }} 269.5 \mathrm{~nm}(\epsilon 17600)$; UV ( 0.1 N NaOH $+10 \% \mathrm{EtOH}) \lambda_{\max } 276 \mathrm{~nm}(\epsilon 18500)$.

9-Benzyl-6-( $\boldsymbol{N}$-methylformamido)-9H-purine (56). To a stirred, ice-bath cooled solution of 46 ( $3.50 \mathrm{~g}, 14.6 \mathrm{mmol}$ ), 4(dimethylamino) pyridine ( $1.80 \mathrm{~g}, 14.7 \mathrm{mmol}$ ), and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL}$ ) was added the acetic-formic anhydride ${ }^{27}$ prepared from 10 mL of $\mathrm{Ac}_{2} \mathrm{O}$ and 5 mL of $100 \% \mathrm{HCO}_{2} \mathrm{H}$. The reaction was stirred at ambient temperature for 16 h and then spin evaporated in vacuo. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(200 \mathrm{~mL})$, washed with water ( $3 \times 50 \mathrm{~mL}$ ), and filtered through glass wool. The solution was spin evaporated in vacuo, and the residue was recrystallized from cyclohexane-EtOAc to give $2.30 \mathrm{~g}(58 \%)$ of 56 : $\mathrm{mp} 162-164{ }^{\circ} \mathrm{C}$; NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 10.32$ (s, $1 \mathrm{H}, \mathrm{CHO}$ ), 8.72 (s, 1 H , purine H ), $8.70(\mathrm{~s}, 1 \mathrm{H}$, purine H ), $7.34(\mathrm{~s}, 5 \mathrm{H}, \mathrm{Ar}), 5.53$ (s, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), 3.44 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).

9-Benzyl-6-(methylthio)-9H-purine (59). To a vigorously stirred solution of 9-benzyl-6-mercaptopurine ${ }^{29}(9.70 \mathrm{~g}, 40.0 \mathrm{mmol})$ in 1 N sodium hydroxide ( 42 mL ) was added methyl iodide ( 10 $\mathrm{mL})$ dropwise. A thick precipitate formed, and $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and $\mathrm{EtOH}(20 \mathrm{~mL})$ were added to facilitate stirring. After 1 h , the solids were collected, washed with $\mathrm{H}_{2} \mathrm{O}$, and recrystallized from cyclohexane-EtOAc to give $6.58 \mathrm{~g}(64 \%)$ of $59: \mathrm{mp} 115-116^{\circ} \mathrm{C}$ (lit. ${ }^{30} \mathrm{mp} 117-118{ }^{\circ} \mathrm{C}$ ); UV ( $0.1 \mathrm{~N} \mathrm{HCl}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 294.5$ $\mathrm{nm}(\epsilon 18800), 286$ (sh) ( 16200 ), 305 (sh) ( 11000 ); UV ( $\mathrm{pH} 7+$ $10 \% \mathrm{EtOH}) \lambda_{\max } 287 \mathrm{~nm}(\epsilon 20300)$, 293 (20200); NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 8.68(\mathrm{~s}, 1 \mathrm{H}$, purine H$), 8.55(\mathrm{~s}, 1 \mathrm{H}$, purine H$), 7.32(\mathrm{~s}, 5 \mathrm{H}, \mathrm{Ar})$, 5.48 (s, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), 2.68 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).

6-(Dimethylamino)-9-(4-hydroxybenzyl)-9H-purine (66). A mixture of $67(3.15 \mathrm{~g}, 8.77 \mathrm{mmol})$, HOAc ( 200 mL ), and $5 \%$ $\mathrm{Pd} / \mathrm{C}(0.6 \mathrm{~g})$ was shaken in the presence of hydrogen at $2-3 \mathrm{~atm}$ for 20 h . An additional 200 mg of catalyst was added, and the reaction was shaken for an additional 60 h . The reaction mixture was filtered through Celite, and the filtrate was spin evaporated in vacuo. The residue was slurried with EtOH and reevaporated. The crystalline solid was collected and washed with EtOH to give $1.83 \mathrm{~g}(77 \%)$ of $66, \mathrm{mp} 215-218{ }^{\circ} \mathrm{C}$. Recrystallization from EtOAc gave $1.44 \mathrm{~g}(61 \%)$ of analytically pure $66: \mathrm{mp} 214-217^{\circ} \mathrm{C}$; NMR ( $\mathrm{Me}_{2} \mathrm{SO}-\mathrm{d}_{6}$ ) $\delta 9.38$ (br s, $1 \mathrm{H}, \mathrm{OH}$ ), 8.22 (s, 1 H , purine H ), 8.17 (s, 1 H , purine H), 7.17 (d, $2 \mathrm{H}, J=8 \mathrm{~Hz}, \mathrm{Ar}$ ), $6.70(\mathrm{~d}, 2 \mathrm{H}, J=$ $8 \mathrm{~Hz}, \mathrm{Ar}), 5.23\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.43\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$.

9-[4-(Benzyloxy)benzyl]-6-(dimethylamino)-9H-purine (67). A solution of 4 -(benzyloxy)benzaldehyde ( $10.61 \mathrm{~g}, 50.0$ $\mathrm{mmol})$, $\mathrm{EtOH}(100 \mathrm{~mL})$, and $\mathrm{NaBH}_{4}(1.00 \mathrm{~g})$ was stirred at ambient temperature for 16 h . The dark solution was spin evaporated in vacuo to give a solid residue that was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 50 mL ) and spin evaporated to dryness. This procedure was repeated
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twice to give crude 4-(benzyloxy) benzyl alcohol. To an ice-bath-cooled solution of this benzyl alcohol in $\mathrm{CH}_{2} \mathrm{CL}_{2}(200 \mathrm{~mL})$ was added $\mathrm{SOCl}_{2}(5 \mathrm{~mL})$ dropwise. After 0.5 h , the reaction solution was spin evaporated in vacuo. The residue was covered twice with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and evaporated to give crude 4 -(benzyloxy)benzyl chloride. A mixture of this benzyl chloride, 6 -chloropurine ( $7.72 \mathrm{~g}, 50.0 \mathrm{mmol}$ ), anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(8.20 \mathrm{~g}, 59.3 \mathrm{mmol})$, and DMF ( 50 mL ) was stirred at ambient temperature for 16 h . The reaction mixture was poured over 200 g of crushed ice and extracted twice with dichloromethane $(200,100 \mathrm{~mL})$. The combined extracts were washed with $\mathrm{H}_{2} \mathrm{O}$, filtered through glass wool, and spin evaporated in vacuo. Ethanol was added to the residue, and the mixture was spin evaporated at $90^{\circ} \mathrm{C}$. This procedure was repeated several times to remove residual DMF. The mixture was purified by the flash chromatography technique as described in method B to give 4.00 g ( $22 \%$ ) of 9-[4-(benzyloxy)benzyl]-6-chloro- $9 H$-purine, $\mathrm{mp} 121-125^{\circ} \mathrm{C}$, which was a single spot on TLC. This 6 -chloropurine was reacted with $40 \%$ aqueous $\mathrm{Me}_{2} \mathrm{NH}$ as in method E to give $3.70 \mathrm{~g}(99 \%)$ of $67, \mathrm{mp} 87-89^{\circ} \mathrm{C}$. Recrystallization of a sample from cyclohexane-EtOAc gave analytically pure material in $63 \%$ yield and an unchanged melting point: UV $(0.1 \mathrm{~N} \mathrm{HCl}+10 \% \mathrm{EtOH}) \lambda_{\text {max }} 270 \mathrm{~nm}(\epsilon 21000)$; UV ( $\mathrm{pH} 7+$ $10 \% \mathrm{EtOH}) \lambda_{\max } 277 \mathrm{~nm}(\epsilon 21000)$; NMR ( $\left.\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 8.18$ (s, 2 H , purine H 's), $7.33\left(\mathrm{~s}, 5 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.27(\mathrm{~d}, 2 \mathrm{H}, J=8.5 \mathrm{~Hz}$, Ar), 6.92 (d, $2 \mathrm{H}, J=8.5 \mathrm{~Hz}, \mathrm{Ar}$ ), $5.28\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{~N}\right), 5.03$ (s, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{O}\right), 3.40\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$.

6-(Dimethylamino)-9-(3-hydroxybenzyl)-9H-purine Hydrobromide (80). A solution of $81(5.50 \mathrm{~g}, 15.3 \mathrm{mmol})$ in $30 \%$ HBr in $\mathrm{HOAc}(50 \mathrm{~mL}$ ) was stirred at ambient temperature for 30 min . The volatiles were removed by spin evaporation in vacuo. The residue was recrystallized from EtOH to give $4.23 \mathrm{~g}(79 \%)$ of $80, \mathrm{mp} 260-262^{\circ} \mathrm{C}$. This impure solid was dissolved in 1 N NaOH and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The aqueous layer was adjusted to pH 7 with HOAc, and the white precipitate was collected. Recrystallization from a dilute solution of $30 \% \mathrm{HBr}$ in HOAc EtOH gave $2.75 \mathrm{~g}(51 \%)$ of analytically pure $80: \mathrm{mp} 260-262^{\circ} \mathrm{C}$; UV ( $\mathrm{pH} 7+10 \% \mathrm{MeOH}$ ) $\lambda_{\text {max }} 276.5 \mathrm{~nm}(\epsilon 21900)$; NMR $\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 10.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}, \mathrm{ArOH}+\mathrm{HBr}), 8.64(\mathrm{~s}, 1 \mathrm{H}$, purine H ), 8.45 ( $\mathrm{s}, 1 \mathrm{H}$, purine H), 7.2-6.6 ( $\mathrm{br} \mathrm{m}, 4 \mathrm{H}, \mathrm{Ar}$ ), 5.42 ( $\mathrm{s}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ) $3.58\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right)$.
5-Amino-4-chloro-6-[(cyclohexylmethyl)amino]pyrimidine (88). A mixture of 5 -amino- 4,6 -dichloropyrimidine $(7.00 \mathrm{~g}, 42.7$ mmol ), $\mathrm{Et}_{3} \mathrm{~N}$ ( $4.34 \mathrm{~g}, 42.9 \mathrm{mmol}$ ), (cyclohexylmethyl)amine ( 4.84 $\mathrm{g}, 42.8 \mathrm{mmol}$ ), and 1-BuOH ( 70 mL ) was heated at reflux with stirring for 18 h . The cooled reaction was spin evaporated in vacuo to a dark residue. The residue was dissolved in EtOAc and filtered to remove the insoluble hydrochloride. The filtrate was evaporated to give a solid that was dispersed in $\mathrm{CCl}_{4}$ and collected. The filtrate was diluted with hexane to give a second crop of crystals for a total yield of $9.0 \mathrm{~g}(88 \%)$ of $88, \mathrm{mp} 136-142^{\circ} \mathrm{C}$. Recrystallization from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ gave the analytical sample: mp $143-145{ }^{\circ} \mathrm{C}$; UV ( $0.1 \mathrm{~N} \mathrm{HCl}+10 \% \mathrm{EtOH}$ ) $\lambda_{\max } 306 \mathrm{~nm}(\epsilon 12000)$; UV ( $0.1 \mathrm{~N} \mathrm{NaOH}+10 \% \mathrm{EtOH}) \lambda_{\max } 266 \mathrm{~nm}(\epsilon 9300), 292$ ( 9100 ); NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 7.70(\mathrm{~s}, 1 \mathrm{H}$, pyrimidine H ), 6.70 (br t, 1 H , NH ), 4.99 (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 3.25 (t, $2 \mathrm{H}, \mathrm{NCH}_{2}$ ), 1.9-0.9 (br, 11 $\mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{11}$ ). Anal. $\left(\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{ClN}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

6-Chloro-9-(cyclohexylmethyl)-9H-purine (89). This compound was prepared from 88 and $\mathrm{HC}(\mathrm{OEt})_{3}$ by method D to give $5.09 \mathrm{~g}(77 \%)$ of $89: \mathrm{mp} 75-76^{\circ} \mathrm{C}$; UV $(\mathrm{pH} 7+10 \% \mathrm{EtOH}) \lambda_{\max }$ $266.5 \mathrm{~nm}(\epsilon 8600)$; NMR ( $\mathrm{Me}_{2} \mathrm{SO}-d_{6}$ ) $\delta 8.75$ ( $\mathrm{s}, 1 \mathrm{H}$, purine H), $8.68(\mathrm{~s}, 1 \mathrm{H}$, purine H$), 4.18\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.9-0.9(\mathrm{br}, 11 \mathrm{H}$, $\mathrm{C}_{6} \mathrm{H}_{11}$ ). Anal. $\left(\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{ClN}_{4}\right) \mathrm{C}, \mathrm{H}, \mathrm{N}$.

5-Amino-4-chloro-6-(furfurylamino) pyrimidine (90). A mixture of 5-amino-4,6-dichloropyrimidine ( $10.0 \mathrm{~g}, 61.0 \mathrm{mmol}$ ), furfurylamine ( $8.00 \mathrm{~g}, 82.4 \mathrm{mmol}$ ), 1-BuOH ( 100 mL ), and $\mathrm{Et}_{3} \mathrm{~N}$ $(10 \mathrm{~mL}, 71.4 \mathrm{mmol})$ was refluxed with stirring for 21 h . The dark solution was cooled and spin evaporated in vacuo. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(250 \mathrm{~mL})$ and washed with water ( $3 \times$ 50 mL ). The solution was filtered through glass wool and spin evaporated to dryness. The dark residue was recrystallized from cyclohexane-EtOAc to give $7.50 \mathrm{~g}(54 \%)$ of crude $\mathbf{9 0}, \mathrm{mp} 116-121$
C. The analytical sample was prepared by purification of 1.88 g of crude 90 by flash chromatography as described in method B. Evaporation of the appropriate fractions and recrystallization from EtOH-cyclohexane gave $1.48 \mathrm{~g}(78 \%)$ of $90: \mathrm{mp}$ 136-138 ${ }^{\circ} \mathrm{C}$; NMR $\left(\mathrm{Me}_{2} \mathrm{SO}-d_{6}\right) \delta 7.78(\mathrm{~s}, 1 \mathrm{H}$, pyrimidine H$), 7.57(\mathrm{~d}, 1 \mathrm{H}$, $J=2 \mathrm{~Hz}, \mathrm{OCH}), 7.23(\mathrm{t}, 1 \mathrm{H}, \mathrm{NH}), 6.35(\mathrm{~m}, 2 \mathrm{H}$, furan Hs$), 5.07$ (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 4.63 (d, $2 \mathrm{H}, \mathrm{CH}_{2}$ ). Anal. $\left(\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{ClN}_{4} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$, N.

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