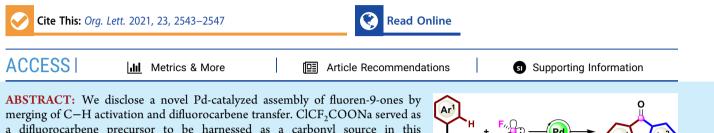


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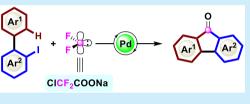


# Pd-Catalyzed Assembly of Fluoren-9-ones by Merging of C–H Activation and Difluorocarbene Transfer

Xiaobing Liu, Heyun Sheng, Yao Zhou, and Qiuling Song\*



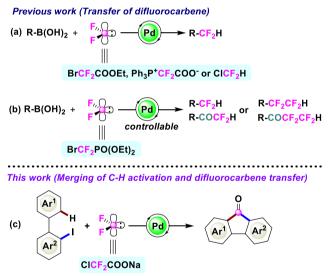
a difluorocarbene precursor to be harnessed as a carbonyl source in this transformation. The current protocol enables us to afford fluoren-9-ones in high yields with excellent functional group compatibility, which also represents the first example of using difluorocarbene as a coupling partner in transition-metal-catalyzed C-H activation.



**S** odium chlorodifluoroacetate (ClCF<sub>2</sub>COONa) has loomed as a promising reagent in organic synthesis, which can be incubated into difluorocarbene via C–Cl and C–C bond cleavage.<sup>1</sup> There are three cases for further utilization of the in situ generated difluorocarbene: (1) as the difluoromethyl reagent; (2) as the C1 synthon; and (3) as the carbonyl source. With respect to previous applications of sodium chlorodifluoroacetate, nucleophiles or amine derivatives are a prerequisite for the triumphant transformations of ClCF<sub>2</sub>COONa. To our knowledge, the utilization of electrophiles or nonamine derivatives to react with ClCF<sub>2</sub>COONa is still far from being supplemented. Especially, the treatment of ClCF<sub>2</sub>COONa as an efficient coupling partner in transitionmetal-catalyzed C–H activation has never been documented heretofore.

During the past years, a vast array of strategies have been developed for difluorocarbene transfer reactions. In sharp contrast to the nonfluorinated carbene complex, the catalytic formation of the metal=CF<sub>2</sub> complex and the application of carbon nucleophiles to transfer the difluorocarbene still remain as long-standing challenges in organic chemistry.<sup>2</sup> In this regard, Zhang's group disclosed an unprecedented Pdcatalyzed difluoromethylation of aryl boronic acids with bromodifluoroacetate (Scheme 1a), in which  $(L_n)Pd=CF_2$ was formed as the crucial intermediate.<sup>3</sup> Soon afterward, Xiao and co-workers also demonstrated an outstanding Pd-catalyzed transfer of difluorocarbene using Ph<sub>3</sub>P<sup>+</sup>CF<sub>2</sub>CO<sub>2</sub><sup>-</sup>(PDFA) as the difluorocarbene source (Scheme 1a).<sup>4</sup> In 2017, Zhang's group once again successfully developed a Pd-catalyzed ClCF<sub>2</sub>H-triggered formation of difluoromethylated arenes (Scheme 1a).<sup>5</sup> Recently, the same group described a preeminent difluorocarbene transfer reaction by using diethyl bromodifluoromethylphosphonate as a difluorocarbene precursor, in which four types of fluoroalkylated arenes were achieved in a controllable catalytic way (Scheme 1b).<sup>6</sup> Due to our persistent interest in the transformations of halogenated difluoro compounds,<sup>7</sup> herein we report an effective method-

# Scheme 1. Pd-Catalyzed Transfer of Difluorocarbene



ology by merging of C–H activation and difluorocarbene transfer to access fluoren-9-ones<sup>8</sup> (Scheme 1c). In this process, difluorocarbene acts as a carbonyl source and simultaneously forms two  $C(sp^2)-C(sp^2)$  bonds, which also represents the first example using difluorocarbene as a coupling partner in transition-metal-catalyzed C–H activation.

We selected 2-iodobiphenyl (1a) and sodium chlorodifluoroacetate (2) as benchmark substrates to initiate our

Received: February 8, 2021 Published: March 22, 2021





research. Despondingly, when the reaction was conducted in the presence of  $Pd(OAc)_2$  by using  $Na_2CO_3$  as base, a stagnant reaction was observed (Table 1, entry 1). To our delight, 9*H*-

# Table 1. Optimization of the Reaction Conditions

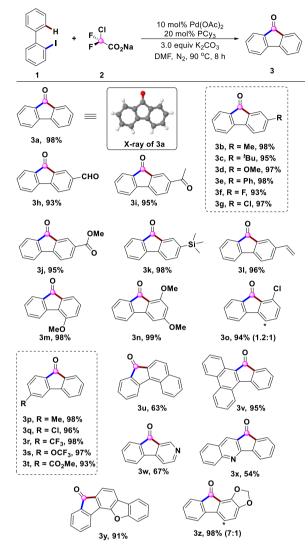
H Ia	+ F. CI F CO <sub>2</sub> Na 2	10 mol% Po 20 mol% I 2 equiv t solver	igand base	O J 3a
entry <sup>a</sup>	ligand	base	solvent	yield <sup>b</sup>
1		$Na_2CO_3$	DMF	trace
2	PPh <sub>3</sub>	Na <sub>2</sub> CO <sub>3</sub>	DMF	13%
3	p-OMe(C <sub>6</sub> H <sub>4</sub> ) <sub>3</sub> P	$Na_2CO_3$	DMF	34%
4	p-F(C <sub>6</sub> H <sub>4</sub> ) <sub>3</sub> P	$Na_2CO_3$	DMF	10%
5	dppe	$Na_2CO_3$	DMF	trace
6	PCy <sub>3</sub>	$Na_2CO_3$	DMF	73%
7	PCy <sub>3</sub>	$K_2CO_3$	DMF	87%
8	PCy <sub>3</sub>	K <sub>3</sub> PO <sub>4</sub>	DMF	69%
9	PCy <sub>3</sub>	NaOAc	DMF	trace
10	PCy <sub>3</sub>	Et <sub>3</sub> N	DMF	n.r.
11 <sup>c</sup>	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	DMF	98%
12	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	81%
13	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	1,4-dioxane	<10%
14	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	CHCl <sub>3</sub>	trace
15	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	toluene	n.r.
16 <sup>d</sup>	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	DMF	87%
17 <sup>e</sup>	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	DMF	38%
18 <sup>f</sup>	PCy <sub>3</sub>	K <sub>2</sub> CO <sub>3</sub>	DMF	n.r.

<sup>*a*</sup>Reaction conditions: 0.2 mmol of **1a**, 0.4 mmol of **2**, 10 mol % of Pd(OAc)<sub>2</sub>, 20 mol % of ligand, 2 equiv of base, 2 mL of solvent, at N<sub>2</sub> for 8 h under 90 °C. <sup>*b*</sup>Isolated yield. <sup>*c*</sup>3.0 equiv of K<sub>2</sub>CO<sub>3</sub>. <sup>*d*</sup>5 mol % of Pd(OAc)<sub>2</sub> + 10 mol % of PCy<sub>3</sub>. <sup>*e*</sup>Air atmosphere. <sup>*f*</sup>Without Pd(OAc)<sub>2</sub>.

fluoren-9-one 3a was achieved in 13% yield when triphenylphosphine served as the additive. Since the ligand could significantly enhance the reaction efficiency, we then investigated a suite of different phosphine ligands (Table 1, entries 2-6). Among the ligands inspected, tricyclohexylphosphine displayed the best catalytic partner, in which the desired 3a could be isolated in 73% yield (Table 1, entry 6). Subsequently, a battery of bases were examined (Table 1, entries 7-10). K<sub>3</sub>PO<sub>4</sub>, NaOAc, and Et<sub>3</sub>N were proved to be inefficient or even ineffective. On the contrary, K<sub>2</sub>CO<sub>3</sub> turned out to be the optimum accelerant to deliver the 9H-fluoren-9one 3a in 87% yield (Table 1, entry 7). The augment of the amount of K2CO3 was beneficial to this difluorocarbene transfer reaction, and the yield of 3a could be increased to 98% (Table 1, entry 11). As a follow-up optimization, we also screened a range of solvents, and the results showed that DMF was still the optimal reaction medium (Table 1, entries 12-15). When the loading of catalyst was reduced, an inferior yield of 3a was obtained (Table 1, entry 16). When the model reaction was conducted under air atmosphere, the yield of 9Hfluoren-9-one 3a was sharply decreased to 37% (Table 1, entry 17). The control experiment indicated that  $Pd(OAc)_2$  was indispensable for this transformation (Table 1, entry 18).

With the optimized reaction conditions in hand, the substrate scope of this novel Pd-catalyzed assembly of fluoren-9-ones was then investigated (Scheme 2). The electronic effect of the substituents installed on 2-iodinebi-

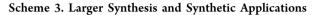
# Scheme 2. Substrate Scope with Respect to 2-Iodobiphenyls<sup>a</sup>

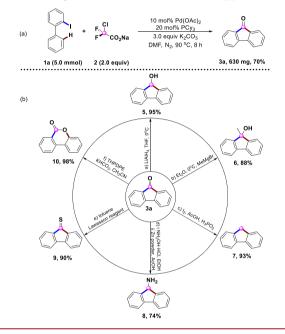


"All reactions unless otherwise stated were carried out with 1 (0.2 mmol), 2 (0.4 mmol), Pd(OAc)<sub>2</sub> (10 mol %), PCy<sub>3</sub> (20 mol %), and  $K_2CO_3$  (3.0 equiv) in DMF (2 mL) under N<sub>2</sub> at 90 °C for 8 h.

phenyls had no evident distinctions. Regardless of the 2iodinebiphenyls bearing the electron-withdrawing or electrondonating groups on the aromatic ring, the 2-iodinebiphenyls can almost be converted equivalently, enabling the formation of the corresponding substituted fluorenones in excellent yields (3a-3t). The structure of 3a was definitely confirmed by X-ray single-crystal diffraction. It should be noted that a succession of sensitive functional groups, such as chlorine, formyl, ketone, ester, TMS, and vinyl, were well amenable to this Pd-catalyzed C-H activation, which makes the further structural elaboration of the fluoren-9-ones readily available. In addition to parasubstituted 2-iodinebiphenyls, meta- and ortho-substituted 2iodinebiphenyls were also compatible in this Pd-catalyzed difluorocarbene transfer reaction. The targeted products (3m-30) were obtained in 93%-99% yields. A range of 2iodinebiphenyls with different substituents installed on the ring of iodobenzene were also certified to be good candidates to assemble the fluorenones (3p-3t) in 93%-98% yields. Gladly, the subjection of fused 2-iodinebiaryls to this transformation was also successful, and the expected products 3u and 3v could be isolated in 63% and 95% yields, respectively. Aside from fused rings, heteroaromatic 2-iodinebiaryls (2u-2z) could be smoothly engaged in this Pd-catalyzed C-H activation as well, rendering the desired fluorenones 3u-3z in moderate to excellent yields.

In order to showcase the practicability of this Pd-catalyzed assembly of fluorenones, we executed the large-scale synthesis and further synthetic manipulations. When the reaction was scaled up to 5 mmol, the targeted **3a** was achieved in 70% yield without loss of the efficiency (Scheme 3a). As a versatile



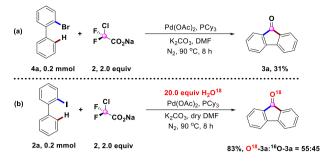


synthon, fluorenones can be transformed into a variety of functionalized organic molecules (Scheme 3b). Fluoren-9-one **3a** could be easily converted into 9*H*-fluoren-9-ol **5** and 9-methyl-9*H*-fluoren-9-ol **6** in 95% and 88% yields, respectively.<sup>9,10</sup> In addition, **3a** could also be reduced into 9*H*-fluorene 7 in high yield using the hypophosphorous acid–iodine reducing system.<sup>11</sup> According to the reported literature, 9*H*-fluoren-9-amine **8**, 9*H*-fluorene-9-thione **9**, and 6*H*-benzo-[c]chromen-6-one **10** could be readily acquired from **3a** in one or two steps as well.<sup>12–14</sup>

In order to gain further insight into the mechanism for this Pd-catalyzed C–H activation by merging of difluorocarbene transfer, we conducted the control experiments and labeling experiments. When 2-iodo-1,1'-biphenyl **2a** was replaced by 2-bromo-1,1'-biphenyl **4a**, the yield of **3a** was sharply decreased to 31%, indicating that the reactivity of iodobenzenes is much higher than that of bromobenzenes in this transformation (Scheme 4a). To notarize the origination of the carbonyl oxygen atom of the targeted 9*H*-fluoren-9-one, an isotope labeling experiment was executed. When 20 equiv of  $H_2^{18}O$  was added to the reaction, 55% of <sup>18</sup>O-labeled product **3a** was detected by GC-MS (Scheme 4b). This result showed that the oxygen atom in the desired product should originate from the water in the reaction system, which facilitated the hydrolysis of difluorocarbene.

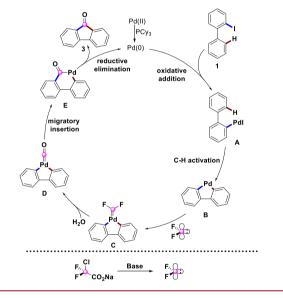
Based on the existing literature and our previous work,<sup>1,7,8,15</sup> we proposed a possible mechanism for this Pd-catalyzed

Scheme 4. Control Experiment and Labeling Experiment



construction of fluorenones. As showcased in Scheme 5, first, oxidative addition of 1 gives rise to the formation of aryl

## Scheme 5. Primary Mechanism Studies



palladium species **A**, which subsequently undergoes C-H bond activation to render the intermediate **B**. With the assistance of base, difluorocarbene is generated from the sodium difluorochloroacetate in situ, which is then captured by metal palladium to deliver the intermediate **C**. The hydrolysis of intermediate **C** produces the palladium complex **D**. The active species **D** goes through the migration insertion to provide the six-membered palladacycle **E**. Finally, intermediate **E** suffers from reductive elimination to furnish the desired product **3**.

In summary, we have successfully established an efficacious Pd-catalyzed difluorocarbene transfer for the construction of fluoren-9-ones. In the current process, difluorocarbene acts as a carbonyl source and simultaneously forms two  $C(sp^2)-C(sp^2)$  bonds, which also represents the first example using difluorocarbene as a coupling reagent in C–H activation. A vast array of functionalized fluoren-9-ones were achieved in high yields via C–H activation with wide substrate scope. This methodology is readily scaled up for the construction of fluoren-9-ones without loss of efficiency. Further applications of this synthetic strategy by merging of C–H activation and difluorocarbene transfer are underway in our group.

## ASSOCIATED CONTENT

#### **Supporting Information**

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.1c00467.

General experimental procedures, crystallographic data, and spectroscopic data for the corresponding products (PDF)

# **Accession Codes**

CCDC 2049630 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

#### AUTHOR INFORMATION

### **Corresponding Author**

Qiuling Song – Institute of Next Generation Matter Transformation, College of Materials Science & Engineering and College of Chemical Engineering at Huaqiao University, Xiamen 361021, Fujian, P. R. China; Fujian University Key Laboratory of Molecule Synthesis and Function Discovery, College of Chemistry, Fuzhou University, Fuzhou 350108, Fujian, P. R. China; State Key Laboratory of Organometallic Chemistry and Key Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, P. R. China; orcid.org/0000-0002-9836-8860; Email: qsong@hqu.edu.cn

## Authors

- Xiaobing Liu Institute of Next Generation Matter Transformation, College of Materials Science & Engineering and College of Chemical Engineering at Huaqiao University, Xiamen 361021, Fujian, P. R. China
- **Heyun Sheng** Institute of Next Generation Matter Transformation, College of Materials Science & Engineering and College of Chemical Engineering at Huaqiao University, Xiamen 361021, Fujian, P. R. China
- Yao Zhou College of Chemistry and Chemical Engineering, Hubei Normal University, Huangshi 435002, Hubei, P. R. China

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.orglett.1c00467

# Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

Financial support from the National Natural Science Foundation (21772046) and the Natural Science Foundation of Fujian Province (2016J01064) is gratefully acknowledged. We also thank the Instrumental Analysis Center of Huaqiao University for analysis support.

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