# A Formal Synthesis of ( $\pm$ )-Cephalotaxine via Pauson-Khand Reaction 

Ping Xing, ${ }^{\text {a }}$ Zuo-gang Huang, ${ }^{\text {b }}$ Yun Jin, ${ }^{a}$ Biao Jiang*a,b

${ }^{\text {a }}$ Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, P. R. of China
${ }^{\text {b }}$ Shanghai Advanced Research Institute, Chinese Academy of Sciences, 99 Haike Road, Shanghai 201210, P. R. of China Fax $+86(21) 64166128$; E-mail: jiangb@mail.sioc.ac.cn
Received: 14.11.2012; Accepted after revision: 27.12.2012


#### Abstract

A concise route toward the formal synthesis of ( $\pm$ )-cephalotaxine has been developed. An intermolecular Pauson-Khand reaction was adopted to construct the cyclopentenone ring efficiently with high regioselectivity.


Key words: $( \pm)$-cephalotaxine, alkaloids, intermolecular PausonKhand reaction, cycloaddition, cyclopentenones, regioselectivity

Cephalotaxine is the parent structure of a class of complex natural products, Cephalotaxus alkaloids, which are isolated from the Asian plant Cephalotaxus drupacea or Cephalotaxus fortunei. ${ }^{1}$ The unique spiro-annulated polycyclic structure was elucidated by X-ray analysis (Figure 1). ${ }^{2}$ Several natural ester derivatives $\mathbf{2}-\mathbf{5}$ of cephalotaxine have demonstrated potent anti-leukemic activity, exhibiting acute toxicity against P388 and L1210 leukemia cells with $\mathrm{IC}_{50}$ values in the $\mathrm{ng} / \mathrm{mL}$ range. ${ }^{3}$ The outstanding bioactivity of these derivatives has attracted great interest from the chemical community in the synthesis of cephalotaxine. Since the first total synthesis of ( $\pm$ )-cephalotaxine by Weinreb in $1972,{ }^{4}$ a number of innovative synthetic strategies have been developed. ${ }^{5}$
Construction of the unique spirocyclic amine moiety within cephalotaxine is the most challenging part of the reported synthetic work. ${ }^{5 d, g, 6-8}$ The retrosynthetic analysis in Scheme 1 shows that the spirocyclic amine moiety might be constructed by an intramolecular Michael addition within 6 , and the cyclopentenone portion of 6 could be prepared by an intermolecular Pauson-Khand reaction from the 1,11-enyne 7. ${ }^{9}$ This appeared to be a straightforward and concise strategy, and employing the PausonKhand reaction for this purpose seemed to be worth attempting. ${ }^{10}$

Piperonal and chloromethyl methyl ether were used to synthesize 8 and 9 , respectively, and coupling of these gave enolic ether $\mathbf{1 0}$ via a Wittig reaction. After lithiumhalogen exchange and subsequent reaction with oxirane, two-carbon-atom-extended alcohol 11 was obtained. Reaction of $\mathbf{1 2}$ with pent- $4-\mathrm{yn}-1$-amine gave the enyne precursor 7 for the Pauson-Khand reaction. However, although the reaction was attempted under a variety of different conditions, neither $\mathbf{7}$ nor $\mathbf{1 3}$ underwent the intramolecular Pauson-Khand reaction to form cyclopentenone 6

## SYNTHESIS 2013, 45, 0596-0600

Advanced online publication: 31.01.2013
DOI: 10.1055/s-0032-1318116; Art ID: SS-2012-F0885-OP
© Georg Thieme Verlag Stuttgart • New York


Figure 1 The structure of cephalotaxine and its derivatives


Scheme 1 Retrosynthetic analysis of cephalotaxine
or 14. Constructing the cyclopentenone moiety bearing a ten-membered ring via a Pauson-Khand reaction from our substrates seemed to be difficult (Scheme 2).
We then decided to adopt an intermolecular PausonKhand reaction strategy. As also reported by Mariano, ${ }^{8}$ the spirocyclic amine moiety can be constructed by an intramolecular Michael addition reaction from 15 (Scheme


Scheme 2 Synthesis of 6 or 14. Reagents and conditions: (a) $\mathrm{Br}_{2}, \mathrm{Fe}$, AcOH , r.t., $68 \%$; (b) $\mathrm{Ph}_{3} \mathrm{P}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, quant.; (c) $t$-BuOK, THF, r.t., 1.5 $\mathrm{h}, 85 \%$; (d) $t$ - BuLi , oxirane, $-78^{\circ} \mathrm{C}$ to r.t., $70 \%$; (e) $\mathrm{TsCl}, \mathrm{Et}_{3} \mathrm{~N}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., $95 \%$; (f) pent-4-yn-1-amine, DIPEA, TBAI, THF, $74 \%$; (g) $\mathrm{Boc}_{2} \mathrm{O}, 5 \% \mathrm{NaOH}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, quant.

3 ). The cyclopentenone ring portion of $\mathbf{1 5}$ might be constructed via an intermolecular Pauson-Khand reaction with ethene.


Scheme 3 Mariano's work on cephalotaxine ${ }^{8}$

We synthesized 15 from 16 (Scheme 4). Firstly, we protected the hydroxyl group of $\mathbf{1 6}$ by reaction with mesyl chloride, to form $\mathbf{1 7}$. We found that $\mathbf{1 7}$ tends to eliminate under traditional Sonogashira coupling reaction conditions; therefore, mild conditions at room temperature were applied, and the coupling product 18 could be obtained in $87 \%$ yield. Next, the intermolecular Pauson-Khand reaction of internal alkyne 18 was carried out. Although two isomeric products are possible, according to the direction of the carbonyl group, ${ }^{11}$ we were delighted to find that only the desired isomer 19 was obtained. Under optimized conditions, a satisfactory yield of $58 \%$ was obtained when $n$-butyl methyl sulfide was used as a promoter with dimethyl sulfoxide as the oxidant. Subsequently, the hydroxyl group in 19 was converted into the benzylamine group in 21 by two steps consisting of pyridinium chlorochromate oxidation and reductive amination. Then 21 was cyclized to the ten-membered ring compound 22, which
was further converted into the critical intermediate 15, which can be converted into cephalotaxine in five steps as described in the literature. ${ }^{5 \mathrm{~g}, 8}$ This straightforward synthesis of $\mathbf{1 5}$ from $\mathbf{1 6}$ was achieved in an overall yield of $20 \%$ over eight steps.


Scheme 4 Synthesis of key intermediate 15. Reagents and conditions: (a) $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}$, r.t., quant.; (b) $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$, CuI, TBAI, pent-4-yn-1-ol, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 87 \%$; (c) $\mathrm{Co}_{2}(\mathrm{CO})_{8}, n$-BuSMe, DMSO, ethene ( 50 atm ), toluene, $100^{\circ} \mathrm{C}, 58 \%$; (d) $\mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., $68 \%$; (e) $\mathrm{BnNH}_{2} \cdot \mathrm{HCl}, \mathrm{THF}-\mathrm{EtOH}$, then $\mathrm{NaOAc}, \mathrm{NaBH}_{3} \mathrm{CN}$, r.t.; (f) DIPEA, $\mathrm{MeCN}, 75^{\circ} \mathrm{C}, 72 \mathrm{~h}, 70 \%$ (2 steps); (g) 1. Pd/C, $\mathrm{H}_{2} ; 2 . \mathrm{Boc}_{2} \mathrm{O}, \mathrm{Et}_{3} \mathrm{~N}$, 83\% (2 steps).

In conclusion, we have developed a concise route for the formal synthesis of cephalotaxine via an intermolecular Pauson-Khand reaction to form the cyclopentenone core structure conveniently. This formal synthetic route to the key intermediate $\mathbf{1 5}$ is the shortest reported so far. It demonstrates an efficient way to construct a cyclopentenone group in some natural products from alkyne precursors.

Chemicals and solvents were either purchased from commercial suppliers or purified by standard techniques. Analytical TLC was performed on Silicycle silica gel plates with F-254 as an indicator and compounds were visualized by irradiation with UV light. Flash column chromatography was carried out on silica gel $\mathrm{H}(40 \mu \mathrm{~m}) .{ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Varian Mercury300 spectrometer ( $300 \mathrm{MHz},{ }^{1} \mathrm{H} ; 75 \mathrm{MHz},{ }^{13} \mathrm{C}$ ). The spectra were recorded of samples in $\mathrm{CDCl}_{3}$ or acetone- $\mathrm{d}_{6}$ as solvent at r.t., and ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR chemical shifts are reported in ppm relative to either the residual solvent peak $\left({ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right)$, or TMS $\left({ }^{1} \mathrm{H}\right)$ as an internal standard. IR spectra were recorded by using a Bio-Rad FTS-185 instrument. MS was performed on an Agilent 5973N or HP 5989A mass instrument (EI). HRMS was performed on a Waters Micromass GCT (EI).

## 6-Bromobenzo $[d][1,3]$ dioxole-5-carbaldehyde (8)

A suspension of $\mathrm{Fe}(11.72 \mathrm{~g}, 0.21 \mathrm{~mol})$ in $\mathrm{AcOH}(140 \mathrm{~mL})$ was treated dropwise with $\mathrm{Br}_{2}(17 \mathrm{~mL}, 0.33 \mathrm{~mol})$ and the mixture was then stirred for 20 min at $0^{\circ} \mathrm{C}$. Piperonal ( $30.00 \mathrm{~g}, 0.20 \mathrm{~mol}$ ) in $\mathrm{AcOH}(100 \mathrm{~mL})$ was added dropwise and the mixture was stirred for 5 min at $0{ }^{\circ} \mathrm{C}$; then $\mathrm{Br}_{2}(10 \mathrm{~mL}, 0.19 \mathrm{~mol})$ was added dropwise at $0^{\circ} \mathrm{C}$. The dark brown soln was stirred at r.t. until no piperonal was detected by TLC. The soln was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (to 1 L ) and
neutralized with sat. aq $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(500 \mathrm{~mL})$. The organic layer was neutralized with $10 \%$ aq $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and separated. The organic layer was washed with sat. aq $\mathrm{NaCl}(2 \times 300 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to dryness. The solid was recrystallized (EtOH); this gave 8 .

Yield: 31.40 g ( $68 \%$ ); colorless crystals.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=10.17(\mathrm{~s}, 1 \mathrm{H}), 7.36(\mathrm{~s}, 1 \mathrm{H}), 7.26$ (s, 1 H ), 6.08 ( $\mathrm{s}, 2 \mathrm{H}$ ).

## (Methoxymethyl)triphenylphosphonium Chloride (9)

A soln of $\mathrm{Ph}_{3} \mathrm{P}(10.00 \mathrm{~g}, 38.10 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was treated with $\mathrm{MOMCl}(3.20 \mathrm{~mL}, 42.60 \mathrm{mmol})$ and the mixture was stirred overnight under reflux until no $\mathrm{Ph}_{3} \mathrm{P}$ was detected by TLC. The soln changed to a suspension after cooling to r.t., and benzene ( 10 mL ) was added. $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and MOMCl were removed by vacuum distillation. Product 9 was collected by filtration and washed with petrol ether.

Yield: 14.50 g (100\%); white solid.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.87-7.61(\mathrm{~m}, 15 \mathrm{H}), 5.89(\mathrm{~d}$, $J=4.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.69(\mathrm{~s}, 3 \mathrm{H})$.

## 5-Bromo-6-(2-methoxyvinyl)benzo $[d][1,3]$ dioxole (10)

A suspension of $9(1.84 \mathrm{~g}, 5.38 \mathrm{mmol})$ in THF $(20 \mathrm{~mL})$ was treated with $t$ - $\mathrm{BuOK}(0.57 \mathrm{~g}, 5.38 \mathrm{mmol})$ and the mixture was stirred for 20 min at r.t. under an argon atmosphere. The color of the suspension changed from white to brown. Then $\mathbf{8}(0.69 \mathrm{~g}, 3.00 \mathrm{mmol})$ was added and the mixture was stirred for 1.5 h at r.t. until no $\mathbf{8}$ was detected. The soln was quenched with sat. aq $\mathrm{NH}_{4} \mathrm{Cl}$ soln $(40 \mathrm{~mL})$, the organic soln was separated, and the aqueous layer was extracted with EtOAc $(3 \times 30 \mathrm{~mL})$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to dryness; the residue was purified by flash chromatography (silica gel, PE-EtOAc, 100:1); this gave the products $(Z) \mathbf{- 1 0}$ and $(E) \mathbf{- 1 0}$.
(Z)-10: yield: 330 mg (43\%); light yellow liquid.
( $E$ )-10: yield: 330 mg (42\%); light gray solid.
IR (KBr): (Z)-10: 2937, 1648, 1477, 1090, $1040 \mathrm{~cm}^{-1}$; (E)-10: 3001, $1639,1481,1135,1038 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(Z)=7.63(\mathrm{~s}, 1 \mathrm{H}), 7.00(\mathrm{~s}, 1 \mathrm{H}), 6.13$ (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.92(\mathrm{~s}, 2 \mathrm{H}), 5.52(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.73(\mathrm{~s}$, $3 \mathrm{H}) ; \delta(E)=6.98(\mathrm{~s}, 1 \mathrm{H}), 6.85(\mathrm{~d}, J=12.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{~s}, 1 \mathrm{H})$, $6.01(\mathrm{~d}, J=12.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.93(\mathrm{~s}, 2 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H})$
${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(Z)=148.0,146.9,146.1,128.5$, $113.3,112.3,109.6,103.8,101.5,60.7 ; \delta(E)=149.4,147.4,146.4$, $129.4,113.1,112.5,105.1,104.4,101.5,56.4$.

MS (EI): $m / z(\%)=256(100)\left[\mathrm{M}^{+}\right]$.
HRMS (EI): $m / z$ calcd for $\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{O}_{3} \mathrm{Br}: 255.9730$; found: 255.9747 .
2-\{6-(2-Methoxyvinyl)benzo[d][1,3]dioxol-5-yl\}ethanol (11)
A soln of $\mathbf{1 0}(514 \mathrm{mg}, 2.00 \mathrm{mmol})$ in THF $(10 \mathrm{~mL})$ was treated with $1.5 \mathrm{M} t-\mathrm{BuLi}$ in pentane $(3.6 \mathrm{~mL}, 4.00 \mathrm{mmol})$ and the mixture was stirred for 1 h at $-78^{\circ} \mathrm{C}$ under argon atmosphere. The color of the soln changed from colorless to orange. Oxirane ( $0.35 \mathrm{ml}, 6.00$ mmol ) was added and the mixture was stirred for 2 h at $-78^{\circ} \mathrm{C}$ and 16 h at r.t. The soln was quenched with sat. aq $\mathrm{NH}_{4} \mathrm{Cl}$ soln ( 1 mL ) and filtered. The organic soln was concentrated to dryness, and the residue was purified by flash chromatography (silica gel, PEEtOAc, $4: 1$ ); this gave the products $(Z) \mathbf{- 1 1}$ and $(E) \mathbf{- 1 1}$.
( $Z$ )-11: yield: 156 mg (35\%); light yellow liquid.
(E)-11: yield: $156 \mathrm{mg}(35 \%)$; light yellow liquid.

IR (KBr): (Z)-11: 3369, 2939, 2884, 1648, 1484, 1099, $1041 \mathrm{~cm}^{-1}$; (E)-11: 3387, 2939, 2886, 1639, 1504, $1485 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(Z)=7.48(\mathrm{~s}, 1 \mathrm{H}), 6.67(\mathrm{~s}, 1 \mathrm{H}), 6.11$ (d, $J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.92(\mathrm{~s}, 2 \mathrm{H}), 5.32(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.84$ $3.66(\mathrm{~m}, 5 \mathrm{H}), 2.85(\mathrm{t}, J=6.7 \mathrm{~Hz}, 2 \mathrm{H}) ; \delta(E)=6.75(\mathrm{~s}, 1 \mathrm{H}), 6.73$
(d, $J=13.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~s}, 1 \mathrm{H}), 5.93(\mathrm{~d}, J=12.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.88$ ( $\mathrm{s}, 2 \mathrm{H}$ ), $3.75(\mathrm{t}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H}), 2.80(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $2 \mathrm{H}), 1.93$ (br, 1 H ).
${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(Z)=146.9,145.7,145.4,128.7$, $127.5,109.8,109.3,101.9,100.6,62.8,60.3,36.5 ; \delta(E)=148.9$, $146.3,145.9,128.4,128.3,110.1,105.6,102.7,100.7,62.8,56.6$, 36.3.
$\operatorname{MS}(\mathrm{EI}):(Z)-11: m / z(\%)=222(88.13)\left[\mathrm{M}^{+}\right], 161(100) ;(E)-11: m / z$ $(\%)=222(100)\left[\mathrm{M}^{+}\right]$.
HRMS (EI) (Z)-11: calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{Na}$ : 245.0784; found: 245.0788; ( $E$-11: calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{4} \mathrm{Na}$ : 245.0784; found: 245.0782 .

## 2-\{6-(2-Methoxyvinyl)benzo[d][1,3]dioxol-5-yl\}ethyl 4-Methylbenzenesulfonate (12)

A soln of $\mathbf{1 1}(27 \mathrm{mg}, 0.12 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was treated with $\mathrm{Et}_{3} \mathrm{~N}(0.1 \mathrm{~mL})$ and $\mathrm{TsCl}(30 \mathrm{mg}, 0.16 \mathrm{mmol})$ and the mixture was stirred overnight at $\mathrm{r} . \mathrm{t}$. The soln was quenched with $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})$, the organic soln was separated, and the aqueous layer was extracted by $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 3 \mathrm{~mL})$. The combined organic layer was concentrated to dryness and the residue was purified by flash chromatography (silica gel, PE-EtOAc, 6:1); this gave products $(Z)$-12 and $(E)$-12.
(Z)-12: yield: 22 mg (47\%); light yellow liquid.
(E)-12: yield: 23 mg (48\%); light yellow liquid.

IR (KBr): (Z)-12: 2900, 1486, 1358, 1176, $1098 \mathrm{~cm}^{-1} ;(E) \mathbf{- 1 2}: 2900$, 1639, 1487, 1357, $1176 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(Z)=7.71(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.37$ (s, 1 H), $7.29(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.51(\mathrm{~s}, 1 \mathrm{H}), 6.07(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 5.90(\mathrm{~s}, 2 \mathrm{H}), 5.12(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.10(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2$ H), $3.73(\mathrm{~s}, 3 \mathrm{H}), 2.91(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}) ; \delta(E)=7.66$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.26(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.65(\mathrm{~s}, 1 \mathrm{H}), 6.64(\mathrm{~d}$, $J=12.6 \mathrm{~Hz}, 1 \mathrm{H}) 6.49(\mathrm{~s}, 1 \mathrm{H}), 5.85(\mathrm{~s}, 1 \mathrm{H}), 5.75(\mathrm{~d}, J=12.7 \mathrm{~Hz}$, $1 \mathrm{H}), 4.07(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(Z)=147.3,146.1,145.4,144.4$, $132.6,129.5,127.6,127.5,125.9,109.7,109.3,101.2,100.6,69.8$, $60.3,32.9,21.4 ; \delta(E)=149.3,146.7,145.8,144.5,132.7,129.5$, $128.5,127.5,125.6,109.9,105.6,101.7,100.7,69.5,56.4,32.8$, 21.4.
$\operatorname{MS}(\mathrm{EI}):(Z) \mathbf{- 1 2}: m / z(\%)=376(51.34)\left[\mathrm{M}^{+}\right], 161(100) ;(E) \mathbf{- 1 2 :} \mathrm{m} / \mathrm{z}$ $(\%)=376(4.29)\left[\mathrm{M}^{+}\right], 161$ (100).
HRMS (MALDI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{O}_{6} \mathrm{~S}: 377.1069$; found: 377.1053.

## $N$-(2-\{6-(2-Methoxyvinyl)benzo[ $d][1,3]$ dioxol-5-yl\}ethyl)pent-

 4-yn-1-amine (7)A soln of $12(142 \mathrm{mg}, 0.32 \mathrm{mmol})$, TBAI ( $117 \mathrm{mg}, 0.32 \mathrm{mmol}$ ), and DIPEA $(0.13 \mathrm{~mL}, 0.74 \mathrm{mmol})$ in THF $(1.5 \mathrm{~mL})$ was treated dropwise with pent-4-yn-1-amine ( $80 \mathrm{mg}, 0.21 \mathrm{mmol}$ ) in THF ( 0.5 mL ) and the mixture was stirred for 3 h under reflux. The soln was concentrated to dryness and the residue was purified by flash chromatography (silica gel, EtOAc-MeOH, 50:1); this gave 7.
Yield: 45 mg (74\%); red solid.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=6.75(\mathrm{~s}, 1 \mathrm{H}), 6.74(\mathrm{~d}, J=12.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.64(\mathrm{~s}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=12.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.88(\mathrm{~s}, 2 \mathrm{H}), 3.66(\mathrm{~s}$, $3 \mathrm{H}), 2.83-2.67(\mathrm{~m}, 6 \mathrm{H}), 2.23(\mathrm{dt}, J=2.7,7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.94$ (t, $J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.68(5, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.43(\mathrm{br}, 1 \mathrm{H})$.
ESI-MS: $m / z=288.1[\mathrm{M}+\mathrm{H}]^{+}$.
HRMS (EI): $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{O}_{3} \mathrm{~N}$ : 288.1585; found: 288.1594 .

## tert-Butyl (2-\{6-(2-Methoxyvinyl)benzo[d] [1,3]dioxol-5-yl\}eth-

 yl)(pent-4-yn-1-yl)carbamate (13)An emulsion of $7(29 \mathrm{mg}, 0.10 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}$ $(1 \mathrm{~mL})$ was treated with $\mathrm{NaOH}(4.40 \mathrm{mg}, 0.11 \mathrm{mmol})$ and $\mathrm{Boc}_{2} \mathrm{O}$ $(24 \mathrm{mg}, 0.11 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.00 \mathrm{~mL})$ and the mixture was stirred for 2 h at r.t. The emulsion was separated and the aqueous
layer was extracted by $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 3 \mathrm{~mL})$. The combined organic layer was concentrated to dryness and the residue was purified by flash chromatography (silica gel, PE-EtOAc, 10:1); this gave 13.
Yield: 39 mg (100\%); brown solid.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=6.76(\mathrm{~s}, 2 \mathrm{H}), 6.60(\mathrm{~d}, J=14.4 \mathrm{~Hz}$, $1 \mathrm{H}), 6.08(\mathrm{~d}, J=12.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.89(\mathrm{~s}, 2 \mathrm{H}), 3.70(\mathrm{br}, 3 \mathrm{H}), 3.29$ (dd, $J=8.5,6.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 3.21 (br, 2 H ), 2.78 (br, 2 H ), 2.17 (br, 2 H), $1.96(\mathrm{t}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.71(\mathrm{br}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H})$.

MS (EI): $m / z(\%)=387$ (57.23) [M $\left.{ }^{+}\right], 57$ (100).
HRMS (MALDI): $m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{NO}_{5} \mathrm{Na}$ : 410.1938; found: 410.1937.

2-\{6-Iodobenzo $[d][1,3]$ dioxol-5-yl\}ethyl Methanesulfonate (17) A soln of $16(190 \mathrm{mg}, 0.65 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ was treated with $\mathrm{Et}_{3} \mathrm{~N}(0.20 \mathrm{~mL}, 1.60 \mathrm{mmol})$ and $\mathrm{MsCl}(141 \mathrm{mg}, 1.23 \mathrm{mmol})$ and the mixture was stirred for 1 h at r.t. until no 16 was detected. The soln was quenched with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to dryness; then the residue was purified by flash chromatography (silica gel, PEEtOAc, 5:1); this gave 17.
Yield: 241 mg (100\%); light brown solid.
IR (KBr): 2903, 1478, 1354, $1174 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=7.43(\mathrm{~s}, 1 \mathrm{H}), 7.00(\mathrm{~s}, 1 \mathrm{H}), 6.16(\mathrm{~s}$, $2 \mathrm{H}), 4.55(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.30(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.14(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=148.5,147.6,131.8,118.6,110.2$, 101.7, 87.8, 68.6, 39.9, 37.2.

MS (EI): $m / z(\%)=370(27.68)\left[\mathrm{M}^{+}\right], 274$ (100).
HRMS (MALDI): $m / z$ calcd for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{NO}_{5} \mathrm{IS}$ : 369.9372 ; found: 369.9379.

## 2-\{6-(5-Hydroxypent-1-yn-1-yl)benzo [d] [1,3]dioxol-5-yl\}ethyl Methanesulfonate (18)

A soln of 17 ( $629 \mathrm{mg}, 1.70 \mathrm{mmol})$, TBAI ( $940 \mathrm{mg}, 2.54 \mathrm{mmol}$ ), $\operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{4}(196 \mathrm{mg}, 0.17 \mathrm{mmol}), \mathrm{CuI}(34 \mathrm{mg}, 0.18 \mathrm{mmol})$ and pent-4-yl-1-ol (275 g, 3.27 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{~mL})$ and $\mathrm{Et}_{3} \mathrm{~N}(1 \mathrm{~mL})$ was stirred overnight at r.t. under an argon atmosphere. The soln was filtered and concentrated to dryness and the residue was purified by flash chromatography (silica gel, $\mathrm{PE}-\mathrm{EtOAc}, 1: 1$ ); this gave 18.
Yield: 483 mg ( $87 \%$ ); light yellow liquid.
IR (KBr): 3380, 2940, 1486, 1352, $1174 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=6.81(\mathrm{~s}, 1 \mathrm{H}), 6.66(\mathrm{~s}, 1 \mathrm{H}), 5.91$ ( $\mathrm{s}, 2 \mathrm{H}$ ), $4.37(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.76(\mathrm{t}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.10(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.90(\mathrm{~s}, 1 \mathrm{H}), 2.52(\mathrm{t}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.21(\mathrm{br}, 1$ H), $1.82(\mathrm{p}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=147.4,146.4,132.1,116.5,111.8$, $109.8,101.3,92.4,78.6,69.3,61.2,37.2,34.4,31.2,15.8$.
ESI-MS: $m / z=349.1[\mathrm{M}+\mathrm{Na}]^{+}$.
HRMS (MALDI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{6} \mathrm{SNa}$ : 349.0716; found: 349.0711.

2-\{6-[2-(3-Hydroxypropyl)-5-oxocyclopent-1-en-1-yl]ben-zo[d][1,3]dioxol-5-yl\}ethyl Methanesulfonate (19)
A soln of $18(258 \mathrm{mg}, 0.79 \mathrm{mmol})$ and $\mathrm{Co}_{2}(\mathrm{CO})_{8}(297 \mathrm{mg}, 0.87$ $\mathrm{mmol})$ in toluene ( 10 mL ) was stirred at r.t. under an argon atmosphere until no 18 was detected. $n-B u S M e(82 \mathrm{mg}, 0.79 \mathrm{mmol})$ and DMSO ( $450 \mathrm{mg}, 5.76 \mathrm{mmol}$ ) were added and the soln was transferred to a high-pressure reactor. The soln was stirred overnight at $100^{\circ} \mathrm{C}$ under an ethene atmosphere ( 50 atm ). The soln was concentrated to dryness and the residue was purified by flash chromatography (silica gel, PE-EtOAc, 1:4); this gave 19.
Yield: 175 mg (58\%); colorless liquid.
IR (KBr): 3384, 2938, 1486, 1353, $1174 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=6.76$ (s, 1 H ), 6.44 (s, 1 H ), 5.96 $(\mathrm{d}, J=12.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.20(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.61(\mathrm{br}, 1 \mathrm{H}), 2.91$ (s, 3 H ), 2.80-2.69 (m, 4 H ), 2.60-2.53 (m, 2 H ), 2.53-2.32 (m, 2 $\mathrm{H}), 1.83-1.70(\mathrm{~m}, 2 \mathrm{H}), 1.55(\mathrm{br}, 1 \mathrm{H})$.
ESI-MS: $m / z=383.1[\mathrm{M}+\mathrm{Na}]^{+}$.
HRMS (MALDI): $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{O}_{7} \mathrm{SNa}$ : 405.0978; found: 405.0999.

2-\{6-[5-Oxo-2-(3-oxopropyl)cyclopent-1-en-1-yl]ben-zo[d][1,3]dioxol-5-yl\}ethyl Methanesulfonate (20)
A suspension of $19(46 \mathrm{mg}, 0.12 \mathrm{mmol})$ and silica gel $(50 \mathrm{mg}, 200-$ $300)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.0 \mathrm{~mL})$ was treated with PCC ( $\left.52 \mathrm{mg}, 0.24 \mathrm{mmol}\right)$ at $0^{\circ} \mathrm{C}$ and the mixture was stirred for 2 h at r.t. until no 19 was detected. The black suspension was filtered though Celite and the filtrate was concentrated to dryness; the residue was purified by flash chromatography (silica gel, PE-EtOAc, 1:2); this gave 20.
Yield: 31 mg (68\%); colorless liquid.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=9.75$ (s, 1 H ), 6.78 ( $\mathrm{s}, 1 \mathrm{H}$ ), 6.46 (s, 1 H), 5.97 (dd, $J=10.4,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.20(\mathrm{br}, 2 \mathrm{H}), 2.91$ (s, 3 H), 2.82-2.52 (m, 10 H$)$.
${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=208.1,199.9,175.1,147.9,141.8$, $128.5,125.0,110.0,101.3,70.0,40.6,37.1,34.5,33.3,29.6,29.3$, 24.0.

## 7-Benzyl-2,3,4,5,6,7,8,9-octahydro-1H-[1,3]dioxo-

lo $\left[4^{\prime}, 5^{\prime}: 4,5\right]$ benzo $[1,2-d]$ cyclopenta [ $\left.f\right]$ azecin-1-one (22)
A soln of $20(76 \mathrm{mg}, 0.20 \mathrm{mmol})$ in anhyd THF $(4.5 \mathrm{~mL})$ and anhyd EtOH $(2 \mathrm{~mL})$ was treated with $\mathrm{BnNH}_{2} \cdot \mathrm{HCl}(117 \mathrm{mg}, 0.81 \mathrm{mmol})$ and the mixture was stirred for 1 h at r.t. The soln was sequentially treated with $\mathrm{NaOAc}(67 \mathrm{mg}, 0.40 \mathrm{mmol})$ and $\mathrm{NaBH}_{3} \mathrm{CN}(26 \mathrm{mg}$, 0.40 mmol ) and then stirred overnight at r.t. until no imine was detected. The soln was quenched with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and NaCl (to saturate the soln), and then separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to dryness to give 21. A soln of 21 and DIPEA ( 0.15 mL ) in MeCN ( 50 mL ) was stirred for 72 h at $75^{\circ} \mathrm{C}$ under an argon atmosphere in a sealed tube. The soln was concentrated to dryness and the residue was purified by flash chromatography (silica gel, PE-EtOAc, 8:1); this gave 22.
Yield: 52 mg ( $70 \%$ ); white solid.
IR (KBr): 2953, 2789, 1697, 1500, 1484, $1030 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.20-7.05(\mathrm{~m}, 3 \mathrm{H}), 6.79-6.72(\mathrm{~m}$, $2 \mathrm{H}), 6.42(\mathrm{~s}, 1 \mathrm{H}), 6.30(\mathrm{~s}, 1 \mathrm{H}), 6.02(\mathrm{~d}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.92(\mathrm{~d}$, $J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.70(\mathrm{~d}, J=14.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.13(\mathrm{dt}, J=12.8,4.2$ $\mathrm{Hz}, 1 \mathrm{H}), 3.04(\mathrm{~d}, J=14.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.86-2.73(\mathrm{~m}, 1 \mathrm{H}), 2.62-2.55$ (m, 3 H), 2.55-2.06 (m, 7 H), 1.85 (dt, $J=13.9,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.69$ (br, 1 H ).
${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=208.7,177.1,147.2,145.4,144.3$, $139.4,134.0,128.2,127.7,126.4,125.8,109.4,108.5,100.7,58.4$, 54.6, 48.8, 34.6, 30.8, 28.3, 27.9, 23.4.

ESI-MS: $m / z=376.2[\mathrm{M}+\mathrm{H}]^{+}$.
HRMS (MALDI): m/z calcd for $\mathrm{C}_{24} \mathrm{H}_{26} \mathrm{NO}_{3}: 376.1907$; found: 376.1922.

## tert-Butyl 1-Oxo-2,3,5,6,8,9-hexahydro-1H-[1,3]dioxo-

lo $\left.4^{\prime}, 5^{\prime}: 4,5\right]$ benzo $[1,2-d]$ cyclopenta $[f]$ azecine- $7(4 H)$-carboxylate (15)
A suspension of $22(36 \mathrm{mg}, 0.10 \mathrm{mmol})$ and $\mathrm{Pd} / \mathrm{C}(4 \mathrm{mg})$ in $i-\mathrm{PrOH}$ $(5 \mathrm{~mL})$ was stirred overnight at $50^{\circ} \mathrm{C}$ under a $\mathrm{H}_{2}$ atmosphere (1 $\mathrm{atm})$ until no 22 was detected. The suspension was filtered and concentrated to give 23. A soln of $23(0.05 \mathrm{mmol}), \mathrm{Et}_{3} \mathrm{~N}(0.1 \mathrm{~mL})$, and $\mathrm{Boc}_{2} \mathrm{O}(0.1 \mathrm{~mL}, 0.40 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was stirred at r.t. for 2 h until no 23 was detected; then the soln was concentrated to dryness and the residue was purified by flash chromatography (silica gel, $\mathrm{PE}-\mathrm{EtOAc}, 2: 1$ ); this gave 15.
Yield: 16 mg (83\%); white solid.

IR (KBr): 2921, 1695, 1484, $1160 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=6.76$ (s, 1 H ), 6.37 (s, 1 H ), 5.92 $(\mathrm{d}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{br}, 2 \mathrm{H}), 3.02(\mathrm{br}, 2 \mathrm{H}), 2.93-2.76(\mathrm{~m}, 1$ H), 2.64-2.37 (m, 7 H), 2.14 (br, 2 H), 1.72 (br, 2 H), 1.34 (s, 9 H).
${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=208.0,178.1,156.0,147.6,146.1$, $142.4,132.7,125.9,109.8,108.6,100.9,79.4,51.4,47.6,34.7$, 32.1, 29.6, 29.0, 28.2, 23.9.

MS (EI): $m / z(\%)=385(6.04)\left[\mathrm{M}^{+}\right], 57(100)$.
HRMS (EI): $m / z$ calcd for $\mathrm{C}_{24} \mathrm{H}_{27} \mathrm{NO}_{5}: 385.1889$; found: 385.1894

## Acknowledgment

We are grateful to the National Basic Research Program of China (2010CB833200 and 2010CB833300), the National Natural Science Foundation of China (20832007 and 21102167), the Science and Technology Commission of Shanghai Municipality (12DZ1930902) and the Knowledge Innovation Program of the Chinese Academy of Sciences.

Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synthesis.

## References

(1) For reviews on the Cephalotaxus alkaloids, see: (a) Miah, M. A. J.; Hudlicky, T.; Reed, J. W. Cephalotaxus alkaloids, In The Alkaloids; Vol. 51; Academic Press: New York, 1998, 199-269. (b) Huang, L.; Xue, Z. Cephalotaxus alkaloids, In The Alkaloids; Vol. 23; Academic Press: New York, 1984, 157-226.
(2) Abraham, D. J.; Rosenstein, R. D.; McGandy, E. L. Tetrahedron Lett. 1969, 10, 4085
(3) Morita, H.; Arisaka, M.; Yoshida, N.; Kobayashi, J. Tetrahedron 2000, 56, 2929.
(4) Auerbach, J.; Weinreb, S. M. J. Am. Chem. Soc. 1972, 94, 7172.
(5) (a) Yasuda, S.; Yamada, T.; Hanaoka, M. Tetrahedron Lett. 1986, 27, 2023. (b) Kuehne, M. E.; Bornmann, W. G.; Parsons, W. H.; Spitzer, T. D.; Blount, J. F.; Zubieta, J. J. Org. Chem. 1988, 53, 3439. (c) Burkholder, T. P.; Fuchs, P. L. J. Am. Chem. Soc. 1988, 110, 2341. (d) Burkholder, T. P.; Fuchs, P. L. J. Am. Chem. Soc. 1990, 112, 9601.
(e) Ikeda, M.; Okano, M.; Kosaka, K.; Kido, M.; Ishibashi, H. Chem. Pharm. Bull. 1993, 41, 276. (f) Isono, N.; Mori, M. J. Org. Chem. 1995, 60, 115. (g) Lin, X.; Kavash, R. W.; Mariano, P. S. J. Org. Chem. 1996, 61, 7335. (h) Tietze, L. F.; Schirok, H. Angew. Chem. Int. Ed. 1997, 36, 1124. (i) Nagasaka, T.; Sato, H.; Saeki, S. I. Tetrahedron: Asymmetry 1997, 8, 191. (j) Tietze, L. F.; Schirok, H. J. Am. Chem. Soc. 1999, 121, 10264. (k) Suga, S.; Watanabe, M.; Yoshida, J. J. Am. Chem. Soc. 2002, 124, 14824. (1) Koseki, Y.; Sato, H.; Watanabe, Y.; Nagasaka, T. Org. Lett. 2002, 4, 885. (m) Li, W.-D. Z.; Wang, Y.-Q. Org. Lett. 2003, 5, 293. (n) Eckelbarger, J. D.; Wilmot, J. T.; Gin, D. Y. J. Am. Chem. Soc. 2006, 128, 10370. (o) Li, W.-D. Z.; Wang, X.W. Org. Lett. 2007, 9, 1211. (p) Liu, Q.; Ferreira, E. M.; Stoltz, B. M. J. Org. Chem. 2007, 72, 7352. (q) Hameed, A.; Blake, A. J.; Hayes, C. J. J. Org. Chem. 2008, 73, 8045.
(r) Esmieu, W. R.; Worden, S. M.; Catterick, D.; Wilson, C.; Hayes, C. J. Org. Lett. 2008, 10, 3045. (s) Taniguchi, T.; Ishibashi, H. Org. Lett. 2008, 10, 4129. (t) Sun, M.-R.; Lu, Y.-Z.; Yang, H.; Liu, H.-M. J. Org. Chem. 2009, 74, 2213. (u) Li, W.-D. Z.; Duo, W.-G.; Zhuang, C.-H. Org. Lett. 2011, 13, 3538.
(6) Parry, R. J.; Schwab, J. M. J. Am. Chem. Soc. 1975, 97, 2555.
(7) Schwab, J. M.; Parry, R. J.; Foxman, B. M. J. Chem. Soc., Chem. Comтип. 1975, 906.
(8) Lin, X.; Kavash, R. W.; Mariano, P. S. J. Am. Chem. Soc. 1994, 116, 9791.
(9) Shore, N. E. In Comprehensive Organic Synthesis; Vol. 5; Trost, B. M.; Fleming, I., Eds.; Pergamon: Oxford, 1991, 1037-1064.
(10) For some examples, see: (a) Ahmar, M.; Locatelli, C.; Colombier, D.; Cazes, B. Tetrahedron Lett. 1997, 38, 5281. (b) Perez-Serrano, L.; Casarrubios, L.; Dominguez, G.; Perez-Castells, J. Chem. Commun. 2001, 2602. (c) PerezSerrano, L.; Dominguez, G.; Perez-Castells, J. J. Org. Chem. 2004, 69, 5413. (d) Mukai, C.; Nomura, I.; Yamanishi, K.; Hanaoka, M. Org. Lett. 2002, 4, 1755. (e) Mukai, C.; Nomura, I.; Kitagaki, S. J. Org. Chem. 2003, 68, 1376. (f) Mukai, C.; Inagaki, F.; Yoshida, T.; Kitagaki, S. Tetrahedron Lett. 2004, 45, 4117. (g) Krafft, M. E.; Fu, Z.; Bonaga, L. V. R. Tetrahedron Lett. 2001, 42, 1427.
(11) For a review, see: Laschat, S.; Becheanu, A.; Bell, T.; Baro, A. Synlett 2005, 2547.

