

Enantioselective Synthesis of $syn-\alpha$ -Aryl- β -hydroxy Weinreb Amides: Catalytic Asymmetric Roskamp Reaction of α -Aryl Diazo Weinreb **Amides**

Sung Ho Shin,[†] Eun Hee Baek,[†] Geum-Sook Hwang,*,[‡] and Do Hyun Ryu*,[†]

Supporting Information

ABSTRACT: A convenient one-pot procedure to synthesize a variety of highly optically active $syn-\alpha$ -aryl- β -hydroxy Weinreb amides using an asymmetric Roskamp/reduction strategy is described. An oxazaborolidinium ion catalyzed asymmetric Roskamp reaction of α -aryl diazo Weinreb amides with aldehydes produced α -phenyl- β -keto Weinreb amides, which were in situ reduced with zinc borohydride to give syn- α -aryl-

 β -hydroxy Weinreb amides in good yields (up to 87%) with high enantioselectivities (up to 99% ee) and syn stereoselectivities (>20:1).

nantioenriched α -alkyl- β -hydroxy carbonyl compounds are versatile synthetic intermediates for the synthesis of natural products and pharmaceutical agents. Because of their utility, the development of synthetic methods to prepare such compounds has been the subject of intense research. Among numerous methods, the asymmetric aldol reaction is generally regarded as one of the most efficient tools for their synthesis. In particular, since the mid-1990s, various methods for directed aldol reactions have been developed to produce chiral α -alkyl- β -hydroxy carbonyl compounds using various chiral auxiliaries and catalysts. 1,2 However, compared to α -alkyl carbonyl compounds, enantioselective syntheses of the corresponding α -aryl- β -hydroxy carbonyl compounds³ have rarely been reported, and to the best of our knowledge, only two catalytic enantioselective methods have been reported. The Benaglia group developed a chiral phosphine oxide catalyzed aldol reaction using thioesters and aldehydes (Scheme 1, eq 1).^{3g} In a complementary approach, the Davies group reported an asymmetric intermolecular C-H functionalization by a chiral dirhodium catalyst, which after TBS deprotection provided synaldol products with high enantioselectivities (Scheme 1, eq 2). 3h,i However, limited substrate scope and low yields have necessitated the development of new catalytic enantioselective methods.

Recently, Lewis acid catalyzed enantioselective Roskamp reactions using α -alkyl diazoesters were reported by both the Feng laboratory 4a and our group. 4b In this type of reaction, formal asymmetric C-H insertion into an aldehyde bond provided α -alkyl- β -ketoesters in high yields and with excellent enantioselectivities. Strategically, we envisioned that subsequent stereoselective reduction of the Roskamp product, α -aryl- β -keto carbonyl compounds 1, would generate chiral $syn-\alpha$ -aryl- β hydroxy carbonyl compounds 2 in a one-pot manner (Scheme 1, eq 3). We describe here a new method for the synthesis of

Scheme 1. Catalytic Asymmetric Methods To Prepare Chiral syn-α-Aryl-β-hydroxy Carbonyl Compounds

Organocatalytic direct aldol reaction of thioesters

C-H functionalization by rhodium catalyst

Roskamp reaction and reduction (this work)

highly optically active $syn-\alpha$ -aryl- β -hydroxy carbonyl compounds, which can be regarded as an alternative to the catalytic asymmetric aldol reaction.

Initially, asymmetric Roskamp reactions between ethyl or tert-butyl α-phenyl diazoesters and p-tolualdehyde were examined to find optimized conditions for the synthesis of α aryl- β -ketoesters 1a (Table 1, entries 1 and 2). However, when the reaction was carried out at -40 °C in the presence of 20 mol % oxazaborolidinium ion $4a^5$ activated by triflic imide, the desired α -phenyl- β -ketoester 1a was formed as a minor product in \sim 20% yield via 1,2-hydride shift (path a), and the major ptolyl migration product 3a was isolated in \sim 60% yield (Table 1, path b) 4c,d path b).4

Received: August 5, 2015

[†]Department of Chemistry, Sungkyunkwan University, Suwon 440-746, Korea

^{*}Western Seoul Center, Korea Basic Science Institute, Seoul 120-140, Korea

Organic Letters Letter

Table 1. Optimization of the Asymmetric Roskamp Reaction of α -Phenyl Diazo Weinreb Amides with p-Tolualdehyde^a

entry	cat.	X	Y	$1/3^{b}$	yield ^c (%)	ee ^d (%)
1	4a	OEt	Tf_2N	1:3	22	e
2	4a	O-t-Bu	Tf_2N	1:3	20	e
3	4a	NMe(OMe)	Tf_2N	>20:1	47	80
4	4a	NMe(OMe)	TfO	>20:1	51	82
5	4b	NMe(OMe)	TfO	>20:1	92	89
6	4c	NMe(OMe)	TfO	>20:1	87	91
7	4d	NMe(OMe)	TfO	>20:1	90	91

^aThe reaction of diazo compounds (0.18 mmol) with *p*-tolualdehyde (0.21 mmol) was performed in the presence of 4 (20 mol %) in 0.5 mL of toluene at −40 °C. ^bDetermined by ¹H NMR analysis of the crude product. ^cIsolated yield of 1. ^dThe ee of 1 was determined by chiral HPLC. ^eThe ee was not determined.

We next considered changing the ester group of the α -phenyl diazoester to alternative groups such as thioesters or amides. Since the N-methoxy-N-methylamide (Weinreb amide) group has many advantages in terms of facile transformation to ketones or aldehydes, 7a we applied the α -phenyl diazo Weinreb amide 7b to these catalytic conditions.

Gratifyingly, replacement of the ester group of the diazo compound with the Weinreb amide group afforded α -phenyl- β keto Weinreb amide 1b as the major product in 47% yield and 80% ee with an excellent 1/3 ratio (>20:1) (Table 1, entry 3). Due to the presence of an acidic proton in 1b, chromatography had to be carried out at -78 °C to avoid erosion of the ee value. 4b The enantioselectivity of the product increased slightly in the presence of the catalyst activated by triflic acid (Table 1. entry 4). We then investigated the effect of changing the boracycle catalyst substituents and found that the best boron aryl substituent was the 9-phenanthrenyl group (Table 1, entries 5-7). The reaction with catalyst 4d in toluene provided optically active α -phenyl- β -keto Weinreb amide **1b** in 90% yield and 91% ee (Table 1, entry 7). As far as we know, this is the first example of catalytic asymmetric Roskamp reaction with α aryl diazo carbonyl compound.

With optimized conditions for the synthesis of α -phenyl- β -keto Weinreb amide **1b** in hand, we investigated diaster-eoselective reduction conditions for the synthesis of syn- α -phenyl- β -hydroxy Weinreb amide **2ba**. Due to difficulty in the isolation of **1b** without loss of optical purity at rt, direct reduction of the ketone of **1b** in a one-pot manner was applied to the preparation of **2ba**. As reducing reagent, zinc borohydride appeared to be superior to lithium borohydride or sodium borohydride in terms of syn stereoselectivity. When zinc borohydride was added at -78 °C to **1b** prepared in situ by the optimized conditions in Table 1, syn- α -phenyl- β -hydroxyamide **2ba** was isolated in 85% yield and 91% ee with an excellent syn/anti ratio (>20:1) (Table 2, entry 1).

Table 2. Asymmetric Roskamp/Reduction Reaction of α -Phenyl Diazo Weinreb Amides with Aldehydes^a

			· -	-
entry	2	R	yield ^b (%)	ee ^c (%)
1	2ba	$4-MeC_6H_4$	85	91
2	2ca	C_6H_5	87	89
3^d	2da	$4-MeOC_6H_4$	75	99
4	2ea	4-BrC ₆ H ₄	80	93
5	2fa	4-ClC ₆ H ₄	82	94
6	2ga	$4-FC_6H_4$	84	92
7	2ha	$4-CF_3C_6H_4$	79	90
8	2ia	$4-NO_2C_6H_4$	74	91
9	2ja	$4-CNC_6H_4$	80	90
10	2ka	$2-MeC_6H_4$	72	94
11	2la	$2-FC_6H_4$	83	99
12	2ma	2-BrC ₆ H ₄	81	93
13	2na	1-Naph	70	98
14	2oa	2-Naph	72	89
15	2pa	Et	78	63
16	2qa	n-Pen	83	55

^aThe reaction of diazo Weinreb amides (0.18 mmol) with aryl aldehyde (0.21 mmol) was performed in the presence of **4d** (20 mol %) in toluene (0.5 mL) at -40 °C, and $Zn(BH_4)_2$ in ether was added at -78 °C. ^bIsolated yield of **2**. ^cThe ee of **2** was determined by chiral HPLC. ^dThe reaction was performed at -40 °C for 96 h.

With optimized one-pot reaction conditions in hand, we evaluated this methodology with a range of aldehydes. Regardless of the electronic properties of substituents on the aromatic aldehyde, the corresponding products 2 were obtained in good yields, high enantioselectivities, and excellent *synstereoselectivities* (Table 2, entries 1–14). While *orthosubstituted* benzaldehydes were not good substrates for the chiral oxazaborolidinium catalyst 4 in general, the corresponding in good yields with excellent enantioselectivities (Table 2, entries 10–12). By contrast, this protocol with aliphatic aldehydes provided the corresponding products 2 with moderate enantioselectivities (Table 2, entries 15 and 16).

Encouraged by the good results illustrated in Table 2, we applied this catalytic methodology to a variety of diazo-Weinreb amides. For *para*-substituted phenyl diazo Weinreb amides, substitution with electron-withdrawing groups, such as p-Br or p-CF₃, provided high yields and enantioselectivities regardless of the electronic properties of substituents on the aromatic aldehyde (Table 3, entries 1–7). However, substitution with electron-donating groups, such as a p-Me, gave syn- α -p-tolyl- β -hydroxy Weinreb amides 2 with lower enantioselectivities (Table 3, entries 8 and 9).

Further chemical transformations of the optically active α -aryl- β -hydroxy Weinreb amides are illustrated in Scheme 2. Grignard reaction of **2ca** with methylmagnesium chloride⁹ led to the α -aryl- β -hydroxy ketone **5** in 84% yield without the need to protect the chiral alcohol (Scheme 2). Spectral data for the purified **5** were in good accordance with literature data confirming the *syn*-stereochemistry of **2**. In addition, TBS protection of **2db** followed by DIBAL-H reduction provided

Organic Letters Letter

Table 3. Asymmetric Roskamp/Reduction Reaction of α -Aryl Diazo Weinreb Amides with Aryl Aldehydes^a

yield^b (%) R^1 R^2 2 entry ee^c (%) 2bb CF₃ 93 1 4-Me 80 2 CF₃ 95 2eb 4-Br 81 3 2ib CF₂ 4-CN 80 93 2bc Br 4-Me 87 88 2ec Br 4-Br 74 92 4-CN 94 6 2jc Br 82 79 4-OMe 99 2db Br 73 8^e 4-Me 74 82 2bd Me 9e 2ed 72

^aThe reaction of diazo Weinreb amide (0.18 mmol) with aldehyde (0.21 mmol) was performed in the presence of 4d (20 mol %) in toluene (1.0 mL) at -40 °C, and $Zn(BH_4)_2$ in ether was added at -78 °C. ^bIsolated yield of 2. ^cThe ee of 2 was determined by chiral HPLC. ^dThe Roskamp reaction was performed at -40 °C for 120 h. ^cThe Roskamp reaction was performed at -40 °C for 24 h.

Scheme 2. Transformations of Weinreb Amides 2 to Ketone, Aldehyde, and 1,3-Diol

the corresponding aldehyde 7,¹¹ which was reduced to alcohol 8 in 70% yield for the two steps. Finally, 1,3-diol 9 was prepared by deprotection of the TBS group with TBAF. Comparison of the optical rotation data of 9 confirmed the absolute (2*S*,3*S*) stereochemistry of 2.^{3h,i}

The observed stereochemistry for the asymmetric Roskamp reaction of the α -aryl diazo Weinreb amide using oxazaborolidinium ion catalyst **4d** can be rationalized on the basis of the transition-state model shown in Scheme 3. The mode of coordination of aryl aldehyde to **4d** is the same as has been previously observed in enantioselective cyanosilylation, ¹² Roskamp reaction of diazoesters, ^{4b} and formal C–C insertion reactions. ^{4c} In the pre-transition-state assembly **10**, shown in Scheme 3, the aldehyde group is situated above the phenyl group, which effectively shields the re face (back) from attack by the α -aryl diazo Weinreb amide. Meanwhile, an apparent π - π interaction between the aryl ring of the aldehyde with the diazo Weinreb amide aryl group holds the two aryl rings together. ¹³ Thus, nucleophilic addition of the diazo Weinreb amide from the si face (front) of the aldehyde is facilitated,

Scheme 3. Transition-State Model for the Asymmetric Roskamp Reaction of α -Aryl Diazo Weinreb Amides with Ar-CHO Catalyzed by 4d and the Reduction Mechanism with Zinc Borohydride

leading to intermediate 11. Chemoselective hydride migration with loss of nitrogen provides the α -aryl- β -keto Weinreb amide S-1 as the major enantiomer, which is then reduced to form the syn (2S, 3S)- α -aryl- β -hydroxy Weinreb amide 2. The 3S stereochemistry of 2 can be explained by the chelation-controlled model 12 with $Zn(BH_4)_2$.

In summary, we have disclosed an oxazaborolidinium ion catalyzed asymmetric synthesis of $syn-\alpha$ -aryl- β -hydroxy Weinreb amides from various aldehydes and α -aryl diazo Weinreb amides that provides high enantioselectivities and excellent diastereoselectivities. The present method may be considered a new strategic reaction that is an alternative to the catalytic asymmetric aldol reaction. The absolute configuration of the reaction product was the same as that predicted by the transition state model in Scheme 3. Additional applications of this catalytic asymmetric transformation and extension of the substrate scope are in progress.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02268.

Experimental procedures and full analytical data (PDF)

AUTHOR INFORMATION

Corresponding Authors

*E-mail: gshwang@kbsi.re.kr. *E-mail: dhryu@skku.edu.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This research was supported by the Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education (NRF-2013R1A1A2059838) and the Ministry of Science, ICT & Future Planning (NRF-2013R1A1A2073207), the National Research Council of Science and Technology [the Creative

Organic Letters Letter

Allied Project (CAP)], and the Korea Basic Science Institute [C35705].

REFERENCES

- (1) Modern Aldol Reactions; Mahrwald, R., Ed.; Wiley-VCH: Weinheim, 2004.
- (2) For reviews, see: (a) Arya, P.; Qin, H. Tetrahedron 2000, 56, 917.
 (b) Trost, B. M.; Brindle, C. S. Chem. Soc. Rev. 2010, 39, 1600.
 (c) Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. Chem. Rev. 2007, 107, 5471. (d) Mlynarski, I.; Baś, S. Chem. Soc. Rev. 2014, 43, 577.
- (3) For stoichiometric asymmetric methods, see: (a) Russell, M. G. N.; Baker, R.; Castro, J. L. Tetrahedron Lett. 1999, 40, 8667. (b) Baker, R.; Cooke, N. G.; Humphrey, G. R.; Wright, S. H. B.; Hirshfield, J. J. Chem. Soc., Chem. Commun. 1987, 1102. (c) Okazaki, M.; Shuto, Y. Biosci., Biotechnol., Biochem. 2001, 65, 1134. (d) Churcher, I.; Williams, S.; Kerrad, S.; Harrison, T.; Castro, J. L.; Shearman, M. S.; Lewis, H. D.; Clarke, E. E.; Wrigley, J. D. J.; Beher, D.; Tang, Y. S.; Liu, W. J. Med. Chem. 2003, 46, 2275. (e) Ahn, M.; Tanaka, K.; Fuji, K. J. Chem. Soc., Perkin Trans. 1 1998, 185. (f) Fringuelli, F.; Piermatti, O.; Pizzo, F. J. Org. Chem. 1995, 60, 7006. For catalytic asymmetric methods, see: (g) Rossi, S.; Benaglia, M.; Cozzi, F.; Genoni, A.; Benincori, T. Adv. Synth. Catal. 2011, 353, 848. (h) Wang, H.; Li, G.; Engle, K. M.; Yu, J.-Q.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 6774. (i) Davies, H. M. L.; Hedley, S. J.; Bohall, B. R. J. Org. Chem. 2005, 70, 10737.
- (4) (a) Li, W.; Wang, J.; Hu, X.; Shen, K.; Wang, W.; Chu, Y.; Lin, L.; Liu, X.; Feng, X. J. Am. Chem. Soc. 2010, 132, 8532. (b) Gao, L.; Kang, B. C.; Hwang, G.-S.; Ryu, D. H. Angew. Chem., Int. Ed. 2012, 51, 8322. For related regioselective examples, see: (c) Gao, L.; Kang, B. C.; Ryu, D. H. J. Am. Chem. Soc. 2013, 135, 14556. (d) Hashimoto, T.; Naganawa, Y.; Maruoka, K. J. Am. Chem. Soc. 2008, 130, 2434.
- (5) (a) Corey, E. J. Angew. Chem., Int. Ed. 2009, 48, 2100. (b) Senapati, B. K.; Hwang, G.-S.; Lee, S.; Ryu, D. H. Angew. Chem., Int. Ed. 2009, 48, 4398. (c) Senapati, B. K.; Gao, L.; Lee, S. I.; Hwang, G.-S.; Ryu, D. H. Org. Lett. 2010, 12, 5088. (d) Lee, S. I.; Kang, B. C.; Hwang, G.-S.; Ryu, D. H. Org. Lett. 2013, 15, 1428. (e) Lee, S. I.; Hwang, G.-S.; Ryu, D. H. J. Am. Chem. Soc. 2013, 135, 7126.
- (6) Asymmetric Roskamp reaction between piperidine α -phenyl diazoamide and p-tolualdehyde provided 1,2-hydride shift product 1 as a major product in \sim 50% yield with a 1/3 ratio of 6/1. For the diastereocontrol effect of diazoamide, see: Marcoux, D.; Goudreau, S. R.; Charette, A. B. *J. Org. Chem.* **2009**, 74, 8939.
- (7) (a) Mentzel, M.; Hoffmann, H. M. R. *J. Prakt. Chem./Chem.-Ztg.* **1997**, 339, 517. (b) α -Aryl diazo-Weinreb amides were prepared from arylacetyl chloride in two steps. (See the Supporting Information.)
- (8) Gensler, W. J.; Johnson, F.; Sloan, D. B. J. Am. Chem. Soc. 1960, 82, 6074.
- (9) Jiao, X.; Wang, L.; Xiao, Q.; Xie, P.; Liang, X. J. Asian Nat. Prod. Res. 2009, 11, 274.
- (10) (a) Kang, B. C.; Shim, S. Y.; Ryu, D. H. Org. Lett. 2014, 16, 2077. (b) Mahrwald, R.; Schetter, B. Org. Lett. 2006, 8, 281. (c) Schetter, B.; Ziemer, B.; Schnakenburg, G.; Mahrwald, R. J. Org. Chem. 2008, 73, 813.
- (11) Maggiotti, V.; Wong, J.-B.; Razet, R.; Cowley, A. R.; Gouverneur, V. Tetrahedron: Asymmetry 2002, 13, 1789.
- (12) Ryu, D. H.; Corey, E. J. J. Am. Chem. Soc. 2005, 127, 5384.
- (13) For selected reviews on $\pi-\pi$ interaction, see: (a) Waters, M. L. Curr. Opin. Chem. Biol. **2002**, 6, 736. (b) Hunter, C. A.; Lawson, K. R.; Perkins, J.; Urch, C. J. J. Chem. Soc., Perkin Trans. 2 **2001**, 651. (c) Salonen, L. M.; Ellermann, M.; Diederich, F. Angew. Chem., Int. Ed. **2011**, 50, 4808. For selected paper on $\pi-\pi$ interaction, see: (d) Hunter, C. A.; Sanders, J. K. M. J. Am. Chem. Soc. **1990**, 112, 5525.
- (14) Oishi, T.; Nakata, T. Acc. Chem. Res. 1984, 17, 338.