# STUDIES ON THE ASYMMETRIC TOTAL SYNTHESIS OF TRICHOTHECENES. STEREOSELECTIVE CONSTRUCTION OF THE C-RING FRAGMENT <br> Duy H. Hua* and S. Venkataraman <br> Department of Chemistry, Kansas State University <br> Manhattan, Kansas 66506 

Summary: A stereoselective construction of the C-ring fragment of trichothecenes from readily available 4 -cumyloxy-2-cyclopentenol in 12 steps ( $24 \%$ overall yield) is described.

We have recently described an asymmetric synthesis based on the reaction of enones with chiral sulfinylallyl anions. The utilization of this Michael-type addition reaction in the asymmetric synthesis of trichothecene mycotoxins ${ }^{2}$ is being developed. In this communication we describe a stereoselective construction of the $C$-ring fragment (1) ${ }^{3,4}$ of trichothecenes from the readily available 4-cumyloxy-2-cyclopenten-1-ol (2) . ${ }_{\sim}^{5}$

Oxidation of alcohol $\underset{\sim}{2}$ with 1.5 equiv. of pyridinium chlorochromate ${ }^{6}$ and 3 molecular sieves in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at r.t. for 1 h provided enone 3 in $95 \%$ yield. Treatment of enone 3 with 1.1 equiv. of $\mathrm{CH}_{3} \mathrm{Li}$ in THF ( $40 \mathrm{~mL} / \mathrm{g}$ of 3 ) at $-30^{\circ} \mathrm{C}$ for 1 h gave alcohol $\underline{4}^{7}$ in $86 \%$ yield and isomer $5^{8}$ in $6 \%$ yield. In this 1,2 addition reaction, methyllithium attacks the carbonyl group predominantly from the side trans to the cumyloxyl group. Epoxidation of 4 with 1.1 equiv. of m-chloroperbenzoic acid (MCPBA) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $r$.t. for 20 h produced epoxide $\underset{\sim}{6},{ }^{9}$ isolated in 85\% yield after column chromatographic separation. The peracid apparently approaches the double bond from the side trans to the cumyloxyl group despite the proximity of the allylic hydroxyl. The stereochemistry of $\underset{6}{6}$ is supported by the selective hydrolysis of the c-2 benzoate group of 11 to alcohol 12 and the formation of 16 from 12 (vide infra). Hydrolysis of the cumyloxyl group and regioselective epoxide opening with 1.5 equiv. of $\mathrm{Tl}\left(\mathrm{ONO}_{2}\right)_{3} \cdot 3 \mathrm{H}_{2} \mathrm{O}^{10}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL} / \mathrm{g})$ at r.t. for 2.5 h provided the nitrate triol (7). ${ }^{11}$ Benzoylation of ? with 2.5 equiv. of benzoyl cyanide and 5 equiv. of $\mathrm{Et}_{3} \mathrm{~N}$ in $\mathrm{CH}_{3} \mathrm{CN}$ at r.t. for 2 h furnished dibenzoate 8 in $60 \%$ overall yield from epoxide $\underset{\sim}{6}$. During the study of this oxirane-ring


Trichothecenes


1


2


3


4


5


6

$7: R^{1}=H, R^{2}=\mathrm{NO}_{2}$
$8: \mathbf{R}^{\mathbf{1}}=\mathrm{COPh}, \mathrm{R}^{\mathbf{2}}=\mathrm{NO}_{\mathbf{2}}$
$10: R^{1}=C O P h, R^{2}=H$
$11: R^{\prime}=\mathbf{C O P h}, R^{2}=t-\mathrm{BuMe}_{2} \mathrm{Si}$


9

$12: R^{1}=H, R^{3}=\mathrm{COPh}$
$13: R^{1}=t-\mathrm{BuMe}_{2} \mathrm{Si}, \mathrm{R}^{3}=\mathrm{COPh}$
$14: R^{1}=\mathrm{t}-\mathrm{BuMe}_{2} \mathrm{Si}, \mathrm{R}^{3}=\mathrm{H}$


15


16


17
cleavage reaction, we found that 6 could be treated with 2 equiv. of $\mathrm{TiCl}_{4}$ in $\mathrm{AcOH}-\mathrm{H}_{2} \mathrm{O}$ (50:1; bo mL/g) at r.t. for 20 min to give chloride $g^{12}$ in $85 \%$ yield. ${ }^{13}$ The regioselective attack at $C-3$ of $\underset{\sim}{5}$ by chloride ion was proven by using the ${ }^{1} H$ NMR decoupling experiments on 9 and its $\mathrm{C}-2, \mathrm{C}-4$ diacetate derivative. Similarly, the stereochemistry of $I$ was established by the decoupling experiment on it and its derivatives (i.e., 8, 10~14). It is assumed that hydrolysis of the cumyl group was followed by oxiranering cleavage. Silylether $1^{14}$ was obtained from $\underline{8}$ by the two-step sequence: (i) reduction with $\mathrm{Zn}-\mathrm{AcOH}$ at r.t. for 1 h ; $92 \%$ yield and (ii) silylation with 1.2 equiv. of $t-\mathrm{BuMe}_{2} \mathrm{SiCl}, 2$ equiv. of imidazole and 0.2 equiv. of p-dimethylaminopyridine (DMAP) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at r.t. for 6 h ; $98 \%$ yield. Selective debenzoylation of 11 (i.e., $\mathrm{C}-2$ benzoate) with 0.15 equiv. of $\mathrm{K}_{2} \mathrm{CO}_{3}$ in MeOH ( $10 \mathrm{~mL} / \mathrm{g}$ ) at $0^{\circ} \mathrm{C}$ for 12 h provided $85 \%$ yield of 12 . Transformation of diol 12 to enone $1^{15}$ was accomplished by the sequence: (i) silylation with 1.5 equiv. of $t-\mathrm{BuMe}_{2} \mathrm{SiCl}, 2.0$ equiv. of imidazole and 0.3 equiv. of DMAP in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $\mathrm{r} . \mathrm{t}$. for 15 n ; $90 \%$ yield, (ii) debenzoylation with 1 equiv. of $\mathrm{K}_{2} \mathrm{CO}_{3}$ in MeOH at r.t.; $98 \%$ yield, (iii) oxidation of $\mathrm{C}-4$ hydroxy group with pyridinium chlorochromate in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at r.t. for 4 n ; $90 \%$ yield and (iv) dehydration with 1.5 equiv. of methanesulfonyl chloride and 3.0 equiv. of $E t{ }_{3} N$ in ether at $0^{\circ} \mathrm{C}$; $95 \%$ yield.

The stereochemistry at $C-3$ and $C-4$ of nitrate 7 was proven by converting intermediate $\underset{\sim}{12}$ to cyclic carbonate $\underset{\sim}{6}$ by the sequence: (i) carbamoylation of $\underset{\sim}{2} \underset{\sim}{2}$ with $\mathrm{PhN}=\mathrm{C}=0$ and DMAP in
pyridine at $60^{\circ} \mathrm{C}$ for 10 h ; (ii) debenzoylation with $\mathrm{K}_{2} \mathrm{CO}_{3}$ in MeOH at r.t. and (iii) desilylation with HF in $\mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ at r.t. followed by methyl chloroformate $-\mathrm{Et}_{3} \mathrm{~N}$.

Finally disilylether 1 was deprotected to trans-4,5-dihydroxy-3-methyl-2-cyclopentenone (17), the acid hydrolyzed cleavage product $Z i$ of moenomycin, ${ }^{16}$ by treatment with $\underline{n}^{-B u_{4}} \mathrm{NF}$ in THF at r.t. in $90 \%$ yield. The above synthesis provides a general route for the stereoselective construction of highly oxygenated cyclopentanes.

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## References and Notes

1. Hua, D. H.; Sinai-Zingde, G.; Venkataraman, S. J. Am. Chem. Soc. 1985, 107, 0000. The conjugate addition reaction of sulfinylallyl anions has also been reported recently: a) Binns, M. R.; Haynes, R. K.; Katsifis, A. A.; Schober, P. A.; Vonwiller, S. C. Tetrahedron Lett. 1985, 1565. b) Binns, M. R.; Chai, O. L.; Haynes, R. K.; Katsifis, A. A.; Schober, P. A.; Vonwiller, S. C. Tetrahedron Lett. 1985, 1569.
2. The history, structure, biological significance and anticancer activity of naturally occurring trichothecenes have been reviewed: a) Doyle, T. W.; Bradner, W. T. "Anticancer Agents based on Natural Product Models," a series of monographs of medicinal chemistry, vol. 16, edited by Cassady, J. M.; Douros, J. D.; Academic Press: New York, 1980, p. 4372. b) Jarvis, B. B.; Mazzola, E. P. Acc. Chem. Res. 1982, 15, 388-395. c) Tamm, Ch. "Chemistry and Biotechnology of Biologically Active Natural Products"; Szantay, Cs. Ed.: Elsevier Science Pub., New York, 1984; pp. 59-77 and references therein.
3. A general scheme for assembling the trichothecene skeleton involving the addition of an A-ring unit to a c-ring unit followed by an intramolecular cyclization providing the Bring has been described: a) Brooks, D. W.; Grothaus, P. G.; Palmer, J. T. Tetrahedron Lett. 1982, 4187-4190. b) Brooks, D. W.; Grothaus, P. G.; Mazdiyasni, H. j. Am. Chem. Soc. 1983, 105, 4473-4474 and references therein.
4. The asymmetric total synthesis of trichothecenes will be discussed at a later date.
5. Alcohol 2 was prepared from cyclopentadiene and cumyl hydroperoxide: Stork, G.; Isobe, M. J. Am. Chem. Soc. 1975, 97, 6260-6261.
6. Corey, E. J.; Suggs, J. W. Tetrahedron Lett. 1975, 2647-2650.
7. All new compounds displayed satisfactory ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ), ${ }^{13} \mathrm{C}$ NMR ( 100 MHz ), UV, IR and low-resolution mass spectra and satisfactory elemental analysis or chemical ionization MS. 4: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 7.5 \sim 7.2(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ph}), 5.81[\mathrm{dd}, \mathrm{J}=5.5, \mathrm{~J}=1.2,1 \mathrm{H}$, $=\mathrm{CHC}(\mathrm{OH}) \mathrm{CH}_{3} \mathrm{~J}, 5.71(\mathrm{dd}, \mathrm{J}=6, \mathrm{~J}=2,1 \mathrm{H},=\mathrm{CH}), 4.15(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHO}), 2.25(\mathrm{dd}, \mathrm{J}=14, \mathrm{~J}=$ $\left.7,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.92(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 1.88\left(\mathrm{dd}, \mathrm{J}=14, \mathrm{~J}=4,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.57(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CMe} \mathrm{Ph})$, $1.55(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CMePh}), 1.24[\mathrm{~s}, 3 \mathrm{H}, \mathrm{C}(\mathrm{OH}) \mathrm{Me}] .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 146.8(\mathrm{~s}, \mathrm{Ph}), 140.5(\mathrm{~d}$, $=\mathrm{CHCOH}), 134(\mathrm{~d},=\mathrm{CH}), 128.2(\mathrm{~d}, \mathrm{o}-\mathrm{Ph}), 127(\mathrm{~d}, \mathrm{p}-\mathrm{Ph}), 126(\mathrm{~d}, \mathrm{~m}-\mathrm{Ph}), 80.9(\mathrm{~s}, \mathrm{C}-\mathrm{Ph})$, $7 \overline{7} .7(\mathrm{~s}, \mathrm{CMeOH}), 76.5(\mathrm{~d}, \mathrm{C}-0), 50.4\left(\mathrm{t}, \mathrm{CH}_{2}\right), 29.8(\mathrm{q}, \mathrm{CMePh}), 28.6(\mathrm{q}, \mathrm{CMePh}), 27(\mathrm{q}$, Me).
8. Isomer 5: ${ }^{1}{ }_{\mathrm{H}}$ NMR $7.5 \sim 7.2(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ph}), 5.78(\mathrm{~s}, 2 \mathrm{H}, \mathrm{HC}=\mathrm{CH}), 4.50(\mathrm{dd}, \mathrm{J}=6.5, \mathrm{~J}=5.2$, $1 \mathrm{H}, \mathrm{CH}-\mathrm{O}), 2.20\left(\mathrm{dd}, \mathrm{J}=14, \mathrm{~J}=7,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.87\left(\mathrm{dd}, \mathrm{J}=14, \mathrm{~J}=5,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.60(\mathrm{~s}$,
$1 \mathrm{H}, \mathrm{OH}), 1.55\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CMe}_{2} \mathrm{Ph}\right), 1.46\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}_{\mathrm{C}} \mathrm{NMR} 146.8(\mathrm{~s}, \mathrm{Ph}), 139.3(\mathrm{~d}$, $=\mathrm{CHCOH}), 136.2(\mathrm{~d},=\mathrm{CH}), 128.2(\mathrm{~d}, \mathrm{o}-\mathrm{Ph}), 127.0(\mathrm{~d}, \mathrm{p}-\mathrm{Ph}), 126.0(\mathrm{~d}, \mathrm{~m}-\mathrm{Ph}), 81.8(\mathrm{~s}$, $\mathrm{C}=\mathrm{Ph}), 77.8\left(\mathrm{~d}, \mathrm{C}-0-\mathrm{CMe}_{2} \mathrm{Ph}\right), 77.6(\mathrm{~s}, \mathrm{COHMe}), 49.5\left(\mathrm{t}, \mathrm{CH}_{2}\right), 29.6,28.7(\mathrm{q}, \mathrm{CMe} 2), 28.5$ ( $\mathrm{q}, \mathrm{Me}$ ).
9. Epoxide 6: ${ }^{1}{ }_{\mathrm{H}}$ NMR $7.5-7.2(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ph}), 3.55\left(\mathrm{t}, \mathrm{d}, \mathrm{J}=8, \mathrm{~J}=1.3,1 \mathrm{H}, \mathrm{CHOCMe}{ }_{2} \mathrm{Ph}\right)$, $3.32(\mathrm{dd}, \mathrm{J}=2.2, \mathrm{~J}=1.3,1 \mathrm{H}, \mathrm{C}-3 \mathrm{H}), 3.16(\mathrm{~d}, \mathrm{~J}=2.2,1 \mathrm{H}, \mathrm{C}-2 \mathrm{H}), 2.5(\mathrm{broad} \mathrm{s}, 1 \mathrm{H}$, $\mathrm{OH}), 1.80\left(\mathrm{dd}, \mathrm{J}=13, \mathrm{~J}=7.5,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.55\left(1 \mathrm{H}, \mathrm{CH}_{2}\right.$, overlap with $\left.\mathrm{CH}_{3}\right), 1.61$ ( $\mathrm{s}, 3 \mathrm{H}$, CMePh), $1.55(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CMePh}), 1.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{CNMR} 146.3$ ( $\left.\mathrm{s}, \mathrm{Ph}\right), 128.2$ (d, o-Ph), 127.2 (d, $\left.\underline{p}^{-} \mathrm{Ph}\right), 125.9\left(\mathrm{~d}, \mathrm{~m}^{-} \mathrm{Ph}\right), 77.6\left(\mathrm{~s}, \mathrm{CMe}_{2} \mathrm{Ph}\right), 75.0(\mathrm{~s}, \underline{\mathrm{CMeOH}}), 71.7$ (d, $\underline{C}^{-}$ $\mathrm{OCMe}_{2} \mathrm{Ph}$ ) , $60.8(\mathrm{~d}, \mathrm{C}-2), 58.3(\mathrm{~d}, \mathrm{c}-3), 41.6\left(\mathrm{t}, \mathrm{CH}_{2}\right), 29.5$ ( $\mathrm{q}, \mathrm{CMePh}$ ), 28.6 ( $\mathrm{q}, \mathrm{CMe} \mathrm{Ph}$ ), $23.7\left(\mathrm{q}, \mathrm{CH}_{3}\right)$.
10. Mincione, E.; Lanciano, F. Tetrahedron Lett. 1980, 21, 1149-1150.
 $[\mathrm{d}, \mathrm{J}=5.4,1 \mathrm{H}, \mathrm{CH}(\mathrm{OH})], 2.70(\operatorname{broad} \mathrm{~s}, 3 \mathrm{H}, \mathrm{OH}), 2.08\left(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=1.5,1 \mathrm{H}, \mathrm{CH}_{2}\right)$, $2.01\left(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=6,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.35\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C} \operatorname{NMR} 96.7\left(\mathrm{~d}, \mathrm{C}-\mathrm{ONO}_{2}\right), 80.3[\mathrm{~d}$, $\mathrm{CH}(\mathrm{OH}) \mathrm{CMeOH}, 78.8(\mathrm{~s}, \underline{\mathrm{CMeOH}}), 73.6(\mathrm{~d}, \mathrm{C}-\mathrm{OH}), 43.1\left(\mathrm{t}, \mathrm{CH}_{2}\right), 24.2\left(\mathrm{q}, \mathrm{CH}_{3}\right)$.
11. Chloride 9: $1_{\mathrm{H}}$ NMR $4.17\left[\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}(\mathrm{OH})-\mathrm{CH}_{2}\right], 4.04(\mathrm{dd}, \mathrm{J}=7, \mathrm{~J}=4,1 \mathrm{H}, \mathrm{CHCl}), 3.76[\mathrm{~d}$, $\mathrm{J}=7,1 \mathrm{H}, \mathrm{CH}(\mathrm{OH}) \mathrm{CMe}], 3.03(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OH}), 2.88(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 2.21(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=7,1 \mathrm{H}$, $\mathrm{CH}_{2}$ ), $1.98\left(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=3,1 \mathrm{H}, \mathrm{CH}_{2}\right), 1.34(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH} 3) .{ }^{13} \mathrm{C} \mathrm{NMR} 84.5[\mathrm{~d}$, $\underline{\mathrm{CH}}(\mathrm{OH}) \mathrm{CMeOH}], 77.50(\mathrm{~s}, \underline{\mathrm{CMeOH}}), 77.13(\mathrm{~d}, \mathrm{C}-\mathrm{OH}), 70.61(\mathrm{~d}, \mathrm{C}-\mathrm{Cl}), 43.9\left(\mathrm{t}, \mathrm{CH}_{2}\right), 25.6(\mathrm{q}$, $\mathrm{CH}_{3}$ ).
12. As far as we are aware, this is the first example of an oxirane ring-opening reaction with $\mathrm{TiCl}_{4}$. A full study with various epoxides will be reported in due course.
 $J=7.4,1 H, C-2 H), 5.17(d d d, J=8, J=5.3, J=4,1 H, C-4 H), 4.79(d d, J=7.4, \mathrm{~J}=$ 5.3, $1 \mathrm{H}, \mathrm{CHOSi}), 2.56\left(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=8,1 \mathrm{H}, \mathrm{CH}_{2}\right), 2.02\left(\mathrm{dd}, \mathrm{J}=15, \mathrm{~J}=4,1 \mathrm{H}, \mathrm{CH}_{2}\right)$, $1.61(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 1.38\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.78(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{Bu}), 0.03(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SiCH}), 0.00(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{SiCH} 3$ ). ${ }^{13} \mathrm{C}_{\mathrm{C}}$ NMR $166.1(\mathrm{~s}, \mathrm{C}=0$ ), $165.8(\mathrm{~s}, \mathrm{C}=0$ ), $133.4,133.1$ (ss, quarternary C of $\mathrm{Ph}), 130.0,129.9(\mathrm{~d}, \mathrm{p}-\mathrm{Ph}), 129.8,129.7(\mathrm{~d}, \mathrm{o}-\mathrm{Ph}), 128.5,128.4(\mathrm{~d}, \mathrm{~m}-\mathrm{Ph}), 82.7(\mathrm{~d}, \mathrm{c}-$ 4), $79.4(\mathrm{~d}, \mathrm{C}-2), 77.8(\mathrm{~s}, \mathrm{C}-\mathrm{OH}), 75.9(\mathrm{~d}, \mathrm{COOSi}), 42.5\left(\mathrm{t}, \mathrm{CH}_{2}\right), 26.8(\mathrm{q}, \mathrm{CMeOH}), 2 b .5$ ( $\mathrm{q}, 3 \mathrm{C}, \mathrm{CMe}_{3}$ ), $18.5\left(\mathrm{~s}, \mathrm{SiCMe}_{3}\right),-4.7,-4.8\left(\mathrm{qq}, \mathrm{SiMe}_{2}\right)$.
13. Enone 1: ${ }^{1} \mathrm{H}$ NMR $5.9(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH}), 4.52(\mathrm{~m}, 1 \mathrm{H}, \mathrm{C}=\mathrm{C}-\mathrm{CHOSi}), 4.15(\mathrm{~d}, \mathrm{~J}=2.7,1 \mathrm{H}, 0=\mathrm{C}-$ CHOSi), 2.06 ( $\mathrm{dd}, \mathrm{J}=1.1, \mathrm{C}=\mathrm{C}-\mathrm{Me}$ ), $0.94\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe}_{3}\right.$ ), 0.93 ( $\mathrm{s}, 9 \mathrm{H}, \mathrm{CMe}_{3}$ ), 0.19, 0.18, $0.17,0.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SiMe}_{2}\right) .{ }^{13} \mathrm{C}$ NMR $201.7(\mathrm{~s}, \mathrm{C}=0), 172.8(\mathrm{~s},=\mathrm{C}-\mathrm{Me}), 128.4(\mathrm{~d},=\mathrm{CH})$, $82.9(\mathrm{~d}, \mathrm{C}=\mathrm{C}-\mathrm{COSi}), 80.3(\mathrm{~d}, \mathrm{C}-\mathrm{OSi}), 26.0,25.8\left(\mathrm{q}, \mathrm{CMe}_{3}\right), 18.4,18.0\left(\mathrm{~s}, \mathrm{CMe}_{3}\right), 16.5(\mathrm{q}$, $\mathrm{CH}_{3}$ ), $-3.7,-4.0,-4.5,-4.6$ ( $\mathrm{q}, \mathrm{SiMe}_{2}$ ).
14. Langenfeld, N.; Welzel, P. Tetrahedron Lett. 1978, 1833-1836, and references cited therein.
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