1358

Structure–Activity Relationships of Antibacterial 6,7- and 7,8-Disubstituted 1-Alkyl-1,4-dihydro-4-oxoquinoline-3-carboxylic Acids¹

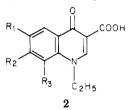
Hiroshi Koga,* Akira Itoh, Satoshi Murayama, Seigo Suzue, and Tsutomu Irikura

Central Laboratories, Kyorin Pharmaceutical Co., Ltd., Nogi-machi, Shimotsuga-gun, Tochigi-ken, 329-01, Japan. Received April 14, 1980

Previous quantitative and qualitative structure-activity studies in antibacterial monosubstituted 1-ethyl-1,4-dihydro-4-oxoquinoline-3-carboxylic acids prompted us to synthesize the 6,7,8-polysubstituted compounds. In this paper, the preparation and antibacterial activity of the 6,7- and 7,8-disubstituted compounds and their derivatives are described. Among these compounds, 1-ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)quinoline-3-carboxylic acid (34) possessed many significant activities and was more active than oxolinic acid (84) against Gram-positive and Gram-negative bacteria. Structure-activity relationships are discussed.

Since nalidixic acid (1, 1-ethyl-1,4-dihydro-7-methyl-4oxo-1,8-naphthyridine-3-carboxylic acid), which shows a good effect on Gram-negative bacteria, was introduced into therapy in 1963, a large number of its analogues have been synthesized and evaluated, some of which came into the market.²

We have been engaged for several years in the search for better drugs in this series and previously reported quantitative structure-activity relationships (QSAR) in 6-, 7-, or 8-monosubstituted 1-ethyl-1,4-dihydro-4-oxoquinoline-3-carboxylic acids (2) against *Escherichia coli*,



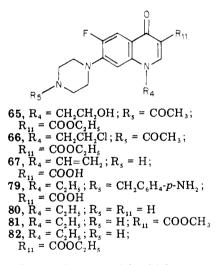
which is one of the representative species of Gram-negative bacteria.¹ The QSAR equation showed that the antibacterial activities of 2 were parabolically correlated with steric parameters for R_1 and R_3 (Es and B_4 , respectively). Although no relationship correlating physicochemical constants for R_2 with the activity of 2 was observed, it was found that among the substituents tested (nitro, acetyl, chloro, methyl, methoxy, dimethylamino, piperazinyl, and hydrogen groups), the piperazinyl group showed the most promise. It proved to have the most potent activity against Gram-negative bacteria, including Pseudomonas aeruginosa. The Hansch equation also revealed that the activities of some of the 6,7,8-polysubstituted derivatives of 2 might be more potent than those of the 6-, 7-, or 8-monosubstituted compounds. In particular, the activities of the 6-fluoro- and 6-chloro-7-(1-piperazinyl) derivatives (34 and 37) in the disubstituted analogues were expected to be very potent, namely, about 10 and 5 times, respectively, that of monosubstituted 3 (2, $R_1 = R_3 = H$ and $R_2 =$ piperazinyl),³ which had the most potent activity and the broadest spectrum of the compounds tested.

In this paper, structure-activity relationships (SAR) of 6,7- and 7,8-disubstituted 1-alkyl-1,4-dihydro-4-oxo-quinoline-3-carboxylic acid derivatives are reported.

Chemistry. The requisite 6,7- and 7,8-disubstituted 1-alkyl-1,4-dihydro-4-oxoquinoline-3-carboxylic acids and their derivatives were prepared by the usual method.²

Anilines (4) were heated with diethyl ethoxymethylenemalonate to give malonates (5) which, generally without purification, were cyclized to 4-hydroxyquinoline-3carboxylic acid ethyl esters (6–15) (Scheme I). Alkylation of the esters 6–16 by treatment with alkyl halides and anhydrous potassium carbonate gave 1-alkyl-1,4-dihydro-4-oxoquinoline esters (17). The 1-alkyl esters (17) were hydrolyzed with aqueous sodium hydroxide or hydrochloric acid to produce the carboxylic acids 18–34, whose N-alkyl-4-quinolinone structure was confirmed by spectral data. The 7-chloro-4-quinolinones 18–31 and 35 were allowed to react with amines in order to obtain the desired 7-amino derivatives 34, 36–40, and 42–63.

A 1-vinyl derivative (67) was readily prepared by using



15 or 77 as the starting material, which was converted to the 1-hydroxyethyl derivative (65) by hydroxyethylation or esterification. The 1-chloroethyl derivative (66) was obtained by treatment of 65 with thionyl chloride and treated with aqueous sodium hydroxide to give the desired 1-vinyl compound (67).

The 7-amino derivatives 34, 51, 63, and 67 were readily alkylated or acylated to afford 36 and 68–78. The N-pnitrobenzyl derivative (72) was reduced to the N-paminobenzyl derivative (79) by catalytic hydrogenation. Quinolinone 80 was prepared by acid-catalyzed decarboxylation of 34. The 4-oxoquinolinecarboxylic acid 34 was also esterified by adding thionyl chloride in the presence of appropriate alcohols to give 81 and 82.

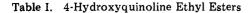
Results and Discussion

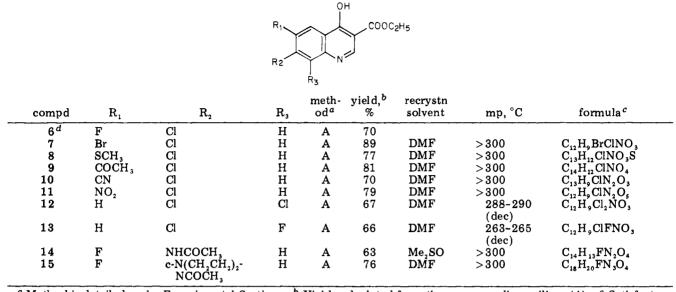
Table VI summarizes the in vitro antibacterial activity against Gram-positive (*Staphylococcus aureus* 209P) and Gram-negative bacteria (*Escherichia coli* NIHJ JC-2 and *Pseudomonas aeruginosa* V-1). The data for 1, 3, pipemidic acid [83, 8-ethyl-5,8-dihydro-5-oxo-2-(1-piperazinyl)-

⁽¹⁾ This work was presented in part at the 98th Annual Meeting of the Pharmaceutical Society of Japan, Okayama, Apr 1978, and at the 99th Annual Meeting of the Pharmaceutical Society of Japan, Sapporo, Aug 1979.

⁽²⁾ R. Albrecht, Prog. Drug Res., 21, 9 (1977).

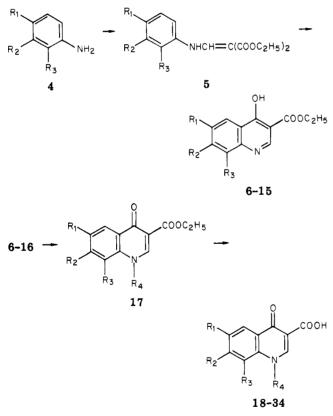
⁽³⁾ S. Minami, J. Matsumoto, M. Shimizu, and Y. Takase, German Offen. 2 362 553 (1974); Chem. Abstr., 81, 105562k (1974).





^a Method is detailed under Experimental Section. ^b Yields calculated from the corresponding anilines (4). ^c Satisfactory C, H, and N analyses (within ±0.4% of theoretical values) were obtained in each instance in which the formula is provided. ^d See the Experimental Section.

Scheme I



pyrido[2,3-*d*]pyrimidine-6-carboxylic acid],⁵ and oxolinic acid [84, 1-ethyl-1,4-dihydro-6,7-(methylenedioxy)-4-oxo-quinoline-3-carboxylic acid]⁶ are included for comparison.

The results for 6-substituted 7-piperazinyl derivatives (34, 37, 39, and 41-45) showed that fluorine was preferable for the 6-substituent of 3, and the activity against *Escherichia coli* NIHJ JC-2 of 34 was 16 times more potent

(5) J. Matsumoto and S. Minami, J. Med. Chem. 18, 74 (1975).

than that of 3, giving the reason for fixing fluorine for R_1 of 2. A series of 7-substituted 6-fluoroquinolinones (18, 32, 33, 36, 48-63, 67-76, and 78-82) was screened, and it was found that the SAR were comparable with those of piromidic acid (85, 8-ethyl-5,8-dihydro-5-oxo-2pyrrolidinopyrido [2,3-d] pyrimidine-6-carboxylic acid)⁷ and pipemidic acid $(83)^5$, although the activity was generally more potent. The replacement of the 7-chloro group of 18 by a methyl or amino group generally caused an increase in the activity. The substitution of the hydrogen of the piperazine NH group in 34 by an alkyl or acyl group reduced the activity against Gram-negative bacteria, particularly Pseudomonas aeruginosa V-1. The displacement of the 1-ethyl group in 34 by sterically comparable substituents, 2-fluoroethyl and vinyl groups (49 and 67), resulted in almost equal activity against Gram-negative bacteria, while the substitution by more or less sterically hindered groups (48 and 51-54) decreased the activity. Decarboxylated compound 80 and esters 81 and 82 did not show any significant activity.

Introduction of fluorine and chlorine (47 and 46) at the 8 position of 3 gave activity against *Escherichia coli* NIHJ JC-2 comparable to and twice that of 3, respectively.

Compound 34 was selected for clinical trial on the basis of the preclinical studies.⁸ SAR of 6,7,8-trisubstituted compounds and QSAR of 1,4-dihydro-4-oxoquinoline-3carboxylic acids will be reported in subsequent papers.

Experimental Section

Spectral data were obtained with the following instruments: IR, Hitachi 260-10 infrared spectrophotometer; NMR, JEOL JNM-4H-100 (using tetramethylsilane as internal standard); mass spectra, Hitachi RMU-6E. Melting points were determined on a Yanagimoto micro melting point apparatus and are uncorrected. Analyses are within $\pm 0.4\%$ of theoretical values when indicated by symbols of the elements. Solutions were dried over anhydrous Na₂SO₄.

6-Fluoro-1,4-dihydro-7-methyl-4-oxoquinoline-3-carboxylic acid ethyl ester (16)⁹ and 6,7-dichloro-1-ethyl-1,4-dihydro-4-oxo-

⁽⁴⁾ R. Kobayashi, K. Isagai, Y. Naka, and M. Hosoya, Japan Kokai 88973 (1976); Chem. Abstr., 86, 121368k (1977).

⁽⁶⁾ D. Kaminsky and R. I. Meltzer, J. Med. Chem., 11, 160 (1968).

 ⁽⁷⁾ S. Minami, T. Shono, and J. Matsumoto, Chem. Pharm. Bull., 19, 1482 (1971); ibid., 19, 1426 (1971).

⁽⁸⁾ A. Ito, K. Hirai, M. Inoue, H. Koga, S. Suzue, T. Irikura, and S. Mitsuhashi, Antimicrob. Agents Chemother., 17, 103 (1980).

quinoline-3-carboxylic acid $(35)^{10}$ were synthesized by the methods in the literature.

5-Amino-2-fluoroacetanilide (4, $R_1 = F$, $R_2 = NHCOCH_3$, and $R_3 = H$) was obtained from 2-fluoro-5-nitroacetanilide¹¹ by reduction (Fe-HCl in aqueous EtOH) in the usual manner in 74% yield. The acetanilide was recrystallized from *i*-PrOH and gave mp 138–140 °C. Anal. (C₈H₉FN₂O) C, H, N.

4-Fluoro-3-(4-acetyl-1-piperazinyl)aniline (4, $\mathbf{R}_1 = \mathbf{F}$, $\mathbf{R}_2 = 4$ -Acetyl-1-piperazinyl, and $\mathbf{R}_3 = \mathbf{H}$). A mixture of o-fluorophenylpiperazine¹² (11.4 g, 0.063 mol), acetic anhydride (12.9 g, 0.126 mol), and DMF (10 mL) was heated at 80–90 °C with stirring. After 1 h, the mixture was evaporated to dryness and made basic with aqueous K_2CO_3 . The solution was extracted with CH_2Cl_2 . The CH_2Cl_2 layer was washed with H_2O , dried, and evaporated. Distillation of the residual oil gave 1-acetyl-4-(o-fluorophenyl)piperazine (12.3 g, 87%), bp 185 °C (6 mm). Anal. $(C_{12}H_{15}FN_2O)$ C, H, N.

To a stirred solution of 1-acetyl-4-(o-fluorophenyl)piperazine (12.0 g, 0.054 mol) in concentrated H_2SO_4 (40 mL) was added dropwise a solution of 60% HNO₃ (5.8 g, 0.055 mol) and concentrated H_2SO_4 (5 mL) at 5–10 °C, and the mixture was stirred at 10 °C for 2 h. The acidic solution was poured onto ice, neutralized with concentrated NH₄OH, and extracted with benzene. After working up, the residue was recrystallized from EtOH to give 1-acetyl-4-(2-fluoro-5-nitrophenyl)piperazine (6.3 g, 44%): mp 132–134 °C. Anal. (C₁₂H₁₄FN₃O₃) C, H, N.

A mixture of 1-acetyl-4-(2-fluoro-5-nitrophenyl)piperazine (6.0 g, 0.0225 mol), EtOH (100 mL), and 10% palladium on charcoal (2.0 g) was hydrogenated at room temperature until hydrogen uptake ceased. The mixture was filtered and the filtrate evaporated to dryness. The residue was recrystallized from benzene to give 4 ($R_1 = F, R_2 = 4$ -acetyl-1-piperazinyl, and $R_3 = H$; 5.3 g, quant), mp 132 °C. Anal. ($C_{12}H_{16}FN_3O$) C, H, N.

7-Chloro-6-fluoro-4-hydroxyquinoline-3-carboxylic Acid Ethyl Ester (6). Method A. A mixture of 3-chloro-4-fluoroaniline (4, $R_1 = F$, $R_2 = Cl$, and $R_3 = H$; 1.46 g, 0.01 mol) and diethyl ethoxymethylenemalonate (2.16 g, 0.01 mol) was heated at 120-130 °C. After 2 h, the resulting EtOH was evaporated off. The crude malonate was used in the successive reaction without further purification. The residue was recrystallized from *n*-hexane to give 5 ($R_1 = F$, $R_2 = Cl$, and $R_3 = H$; 3.16 g, quant): mp 55-57 °C; ¹H NMR (CDCl₃) δ 1.2-1.45 (6 H, m, 2 CH₃), 4.1-4.4 (4 H, m, 2 CH₂), 6.85-7.25 (3 H, m, aromatic H), 8.33 (1 H, d, $J_{H-H} =$ 13 Hz, NCH), 10.99 (1 H, d, $J_{H-H} = 13$ Hz, NH); IR (KBr) 1685 cm⁻¹ (ester). Anal. ($C_{1-H_{15}}CFNO_4$) C, H, N.

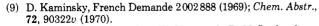
cm⁻¹ (ester). Anal. (C₁₄H₁₅ClFNO₄) C, H, N. The crude 5 (R₁ = F, R₂ = Cl, and R₃ = H; 5.4 g, 0.017 mol) was added to diphenyl ether (50 mL) and refluxed for 1 h. After the solution cooled, the resulting precipitate was filtered off, washed with benzene, and dried. The solid was recrystallized from DMF to give 6 (3.2 g, 70%): mp >300 °C; ¹H NMR (CF₃COOD) δ 1.56 (3 H, t, J_{H-H} = 7 Hz, CH₃), 4.73 (2 H, q, J_{H-H} = 7 Hz, CH₂), 8.35 (1 H, d, J_{H-F} = 8 Hz, aromatic H), 8.37 (1 H, d, J_{H-F} = 6 Hz, aromatic H), 9.35 (1 H, s, 2-H); IR (KBr) 1690 cm⁻¹ (ester). Anal. (C₁₂H₉ClFNO₃) C, H, N.

The 4-hydroxyquinolines 7–15, found in Table I, were prepared by this method from the corresponding anilines (4).

7-Chloro-1-ethyl-6-fluoro-1,4-dihydro-4-oxoquinoline-3carboxylic Acid (18). Method B. A mixture of 6 (2.7 g, 0.01 mol), K_2CO_3 (3.45 g, 0.025 mol), EtI (4 mL, 0.05 mol), and DMF (20 mL) was heated at 80-90 °C with stirring. After 10 h, the mixture was evaporated to dryness and extracted with CH_2Cl_2 . The CH_2Cl_2 layer was washed with H_2O , dried, and evaporated to dryness. The crude ester (3.0 g, quant) was used in the successive reaction without further purification. The residue was recrystallized from EtOH to yield 17 ($R_1 = F, R_2 = Cl, R_3 = H$, and $R_4 = C_2H_5$; 2.7 g, 90%): mp 142-143 °C; ¹H NMR (CDCl₃)

4-Oxoquinoline-3-carboxylic Acids

Table II.



- (10) N. Barton, A. F. Crowther, W. Hepworth, D. N. Rechardson, and G. W. Driver, British Patent 830 832 (1960); Chem. Abstr., 55, 7442e (1961).
- (11) J. J. Blanksma, W. J. van den Broek, and D. Hoegen, Rec. Trav. Chim. Pays-Bas, 65, 329 (1946).
- (12) R. Ratouis, J. R. Boissier, and C. Dumont, J. Med. Chem., 8, 104 (1965).

compd	$\mathbf{R_{i}}$	${ m R_2}$	${ m R_{3}}$	${ m R_4}$	alkylating agent	method ^a	yield, ^b %	recrystn solvent	mp, °C	formula ^c
1 8 d	L L	G	H	C.H.	EtI	B, C	90, ^e 88 ^f			
10	Br	50	H	C.H.	EtI	B.	92	DMF	>300	C ₁₂ H, BrCINO3
00	SCH	50	Η	C.H.	EtI	B	47	DMF	270–273 dec	C ₁₃ H ₁₂ CINO ₃ S
916	COCH	50	Η	C.H.	EtI	C	63	DMF	272-275 dec	$C_{14}H_{12}CINO_4$
25	CIN NO	50	Η	C,H,	EtI	C	60	DMF	>300	C ₁₃ H ₆ CIN ₂ O ₃
23	NON	50	Η	C.H.	EtI	C	60	DMF	291-294 dec	C ₁₂ H ₆ CIN ₂ O ₅
24	H 12		5	C,H.	EtI	В	26	DMF	205 - 208	$C_{12}H_{*}Cl_{2}NO_{3}$
25	H	50	ſz.	C,H,	EtI	ы	62	DMF	236-238	C ₁₂ H, CIFNO3
26	F	10	Η	CH,	MeI	12	06	DMF	>300	C.,H,CIFNO3
22	, F±	5	H	CHLCHLF	FCH, CH, I	C	64	DMF	262 - 264	C ₁₂ H ₈ CIF ₂ NO ₃
28	, F .	50	Η	CH, CH, OH	HOCH,CH,Br	B	91	DMF	256-259	C ₁₂ H, CIFNO4
20	, <u>F</u>	50	Η	4	n-C,H,Br	B	92	DMF	254 - 255	C ₁₃ H ₁₁ CIFNO ₃
30	, fr	50	Η	CH, CH=CH,	CH,=CHCH,Br	в	89	DMF	234 - 237	C13H,CIFNO3
31	, ſz		Η	CH,C,H,	C, H, CH, CI	в	96	DMF	252-253	C,H,,CIFNO,
50	, fx	HU	Ξ	C.H.	Eŭ '	c	68	DMF	287 - 290	C, H, FNO
1 6	4 [¥	NH	H	C.H.	EtI	В	34	DMF	>300	C,,H,,FN,O,
34	ч Ч	c-N(CH ₂ CH ₂) ₂ NH	Η	C_2H_3	EtI	B	74	CH ₂ Cl ₂ -EtOH	221-223	C ₁₆ H ₁₈ FN ₃ O ₃
^a Methods i perimental Se	are detailed u	^a Methods are detailed under Experimental Section. ^b Yields calculated from the corresponding 4-hydroxyquinoline esters (6-16). ^c See Table I, footnote c . perimental Section. ^e Yield in method B. ^f Yield in method C.	ction. eld in m	^b Yields calculated ethod C.	from the correspond	ing 4-hy dro	ky quinoline	sters (6-16). ^c See	Table I, footnote c.	d See the Ex-

)quinolines	
7-(1-Piperazinyl	
Table III.	

^c A, reprecipitation from aqueous $K_2 CO_{3^{-}}$

 b See Table I, footnote c.

^{*a*} Method is detailed under Experimental Section. AcOH.

 δ 1.3–1.65 (6 H, m, 2 CH₃), 4.15-4.5 (4 H, m, 2 CH₂), 7.52 (1 H, d, $J_{\rm H-F}$ = 6 Hz, 8-H), 8.14 (1 H, d, $J_{\rm H-F}$ = 8 Hz, 5-H), 8.42 (1 H, s, 2-H); IR (KBr) 1720 (ester), 1615 cm⁻¹ (C=O). Anal. (C₁₄-H₁₃ClFNO₃) C, H, N.

A mixture of crude 17 (R₁ = F, R₂ = Cl, R₃ = H, and R₄ = C₂H₅; 2.7 g, 0.0091 mol) and 2 N NaOH (25 mL, 0.05 mol) was refluxed with stirring. After 2 h, the mixture was acidified with AcOH, and the resulting precipitate was filtered off, washed with H₂O, and dried. The solid was recrystallized from DMF to yield 18 (2.2 g, 90%): mp 284–285 °C; ¹H NMR (CF₃COOD) δ 1.81 (3 H, t, J_{H-H} = 7 Hz, CH₃), 4.92 (2 H, q, J_{H-H} = 7 Hz, CH₂), 8.39 (1 H, d, J_{H-F} = 5 Hz, aromatic H), 8.41 (1 H, d, J_{H-F} = 8 Hz, aromatic H), 9.39 (1 H, s, 2-H); IR (KBr) 1715 (COOH), 1610 cm⁻¹ (C=O); MS, m/e 269 (M⁺). Anal. (C₁₂H₉ClFNO₃) C, H, N.

Method C. Compound 18 was also obtained in comparable yield (88%) when 2 N HCl was used as hydrolyzing agent.

The 4-oxoquinolines **19–34**, shown in Table II, were prepared by these methods from the corresponding 4-hydroxyquinolines 7–16.

1-Ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)quinoline-3-carboxylic Acid (34). Method D. A mixture of 18 (2.7 g, 0.01 mol) and piperazine (4.3 g, 0.05 mol) was heated at 130-140 °C with stirring. After 5 h, the mixture was evaporated to dryness and H₂O was added to the residue. The resulting solid was filtered off, washed with H₂O, dried, and recrystallized from CH₂Cl₂-MeOH or purified by reprecipitation from aqueous AcOH-aqueous NaOH to give 34 (2.1 g, 66%): mp 227-228 °C; ¹H NMR (CF₃COOD) δ 1.78 (3 H, t, J_{H-H} = 7 Hz, CH₃), 3.7-4.1 (8 H, m, piperazine CH₂), 4.88 (2 H, q, J_{H-H} = 7 Hz, CH₃), 3.7-4.1 (8 H, m, piperazine CH₂), 4.88 (1 H, d, J_{H-F} = 12.5 Hz, 5-H), 9.32 (1 H, s, 2-H); IR (KBr) 1730 (COOH), 1620 cm⁻¹ (C=O); MS, m/e 319 (M⁺), 277, 275, 233. Compound 34 was readily converted to the hydrochloride in the usual way and recrystallized from H₂O-EtOH. The hydrochloride of 34 had mp >300 °C. Anal. (C₁₆H₁₈FN₃O₃·HCl) C, H, N.

The above aqueous filtrate was acidified with concentrated HCl, and the resulting crystals were filtered off. The solid was dissolved in aqueous NaOH. The solution was neutralized by adding aqueous AcOH, and the precipitate was filtered off, washed with H₂O, and dried. The solid was recrystallized from DMF to give 7-chloro-1-ethyl-1,4-dihydro-4-0x0-6-(1-piperazinyl)quinoline-3-carboxylic acid [64; 0.84 g, 25%; mp 272-273 °C. Anal. (C₁₆-H₁₈ClN₃O₃) C, H, N], which was converted to the hydrochloride in the usual way and recrystallized from H₂O. The hydrochloride of **64** had mp >300 °C; ¹H NMR (CF₃COOD) δ 1.80 (3 H, t, J_{H-H} = 7 Hz, CH₃), 3.6–3.9 (8 H, m, piperazine CH₂), 4.87 (2 H, q, J_{H-H} = 7 Hz, CH₂), 8.29 (1 H, s, aromatic H), 8.40 (1 H, s, aromatic H), 9.40 (1 H, s, 2-H); IR (KBr) 1720 (COOH), 1608 cm⁻¹ (C=O); MS, *m/e* 335 (M⁺-HCl). Anal. (C₁₆H₁₈ClN₃O₃·HCl-0.25H₂O) C, H, N.

Compound 64 did not possess any significant activities (Table VI).

The 7-amino-4-oxoquinolines 36-40 and 42-63, exhibited in Tables III and IV, were prepared from the corresponding 4-oxoquinolines 18-31 and 35 by this method using appropriate amines.

6-Fluoro-1,4-dihydro-1-(2-hydroxyethyl)-4-oxo-7-(4acetyl-1-piperazinyl)quinoline-3-carboxylic Acid Ethyl Ester (65). A mixture of 15 (3.0 g, 0.0083 mol), K_2CO_3 (5.7 g, 0.0413 mol), 2-bromoethanol (10.4 g, 0.083 mol), and DMF (40 mL) was heated at 90–100 °C with stirring. After 21 h, the mixture was evaporated to dryness. The residue was extracted with CH₂Cl₂, washed with H₂O, dried, and evaporated. The residue was recrystallized from CH₂Cl₂-AcOEt to give 65 (2.6 g, 77%): mp 221–223 °C (dec); ¹H NMR (CDCl₂) δ 1.33 (3 H, t, J_{H-H} = 7 Hz, CH₃), 2.15 (3 H, s, CH₃), 3.15–3.35, 3.6–3.9 (8 H, m, piperazine CH₂), 4.05–4.4 (6 H, m, 3 CH₂), 6.78 (1 H, d, J_{H-F} = 7 Hz, 8-H), 7.12 (1 H, d, J_{H-F} = 13 Hz, 5-H), 8.30 (1 H, s, 2-H); IR (KBr) 1705 (ester), 1650, 1620 cm⁻¹ (C=O). Anal. (C₂₀H₂₄FN₃O₅) C, H, N.

To an ice-cooled mixture of 77 (0.38 g, 0.001 mol) and absolute EtOH (20 mL) was added dropwise thionyl chloride (2.4 g, 0.02 mol). After the addition was completed, the mixture was refluxed with stirring. After 5.5 h, the mixture was evaporated to dryness. The residue was neutralized with aqueous K_2CO_3 and extracted with CHCl₃. After working up, the residue was recrystallized from CH₂Cl₂-AcOEt to yield **65** (0.36 g, 89%).

	formula ^b	C ₁₈ H ₂ FN ₃ O ₃ ·0.25H ₂ O C ₁₈ H ₂ FN ₃ O ₄ C ₁₈ H ₂ FN ₃ O ₄ C ₁₉ H ₂ FN ₃ O ₄ C ₂₃ H ₂₃ FN ₃ O ₃ ·0.25H ₂ O C ₂₃ H ₂₃ FN ₃ O ₄ C ₁₇ H ₁₈ FN ₃ O ₄ C ₁₈ H ₂₀ FN ₃ O ₄ C ₁₆ H ₁₈ FN ₃ O ₅ C ₁₆ H ₁₈ FN ₃ O ₅	n method F.
	mp, $^{\circ}C$	251-253 229-230 232-233 214-215 230-231 230-231 230-231 230-231 230-231 230-231 230-231 230-231 242-243 244-245 dec	E. ^e Yield i
	recrystn solvent	CHCI,-C ₆ H ₆ DMF DMF DMF DMF DMF DMF DMF DMF DMF DMF	^a Yield in method
СООН	yield, %	$\begin{array}{c} 30,^{d} 85^{e} \\ 72 \\ 60 \\ 60 \\ 83 \\ 73 \\ 83 \\ 83 \\ 83 \\ 83 \\ 83 \\ 83$	ntal Section.
	method ^a	យ្យកាយកាលល្ណាល ល ក	e Experime
	reagent	Mel, HCHO-HCOOH EtI HOCH,CH,Br CH,=CHCH,Br C,H,CH,Cl p-N0,C,H,CH,Br HCOOH Ac,0 C,H,COCI HCHO-HCOOH Ac,0 C,H,COCI HCHO-HCOOH Ac,0	le I, footnote $c.^{-c}$ See the Experimental Section. ^{a} Yield in method E. ^{e} Yield in method F.
	${ m R}_{10}$	CH, C,H, C,H, CH, CH, OH CH, CH, OH CH, CH, CH, CH, C, H, $P-NO_2$ CHO CHO COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, COCH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CH, CC, CH, CH, CH, CH, CH, CH, CH, CH,	^a Methods are detailed under Experimental Section. ^b See Table I,
	\mathbf{R}_{s} \mathbf{R}_{s}	$\begin{array}{c} -(CH_2)_2 -\\ -(CH_2)_2 -$	er Experiment:
	${ m R}_4$	С, H, С, C, H, С, C, H, C, H, C, H, C, H, C, C, H, C, C, H, C, C, H, C, C, H,	are detailed und
	compd	36 ^c 68 71 72 73 73 73 73 73 73 73 73 73 73 73 73 73	^a Methods :

0

7-Aminoquinoline

⊳.

Table '

1-(2-Chloroethyl)-6-fluoro-1,4-dihydro-4-oxo-7-(4-acetyl-1-piperazinyl)quinoline-3-carboxylic Acid Ethyl Ester (66). To an ice-cooled mixture of 65 (0.405 g, 0.001 mol), pyridine (0.095 g, 0.0012 mol), and CHCl₃ (10 mL) was added dropwise a solution of SOCl₂ (1.19 g, 0.01 mol) in 5 mL of CHCl₃. The mixture was left overnight at room temperature. The solution was evaporated and H₂O added. The aqueous mixture was neutralized with aqueous K₂CO₃ and extracted with CHCl₃. After working up, the solid was recrystallized from EtOH to give 66 (0.364 g, 86%): mp 218–219 °C; ¹H NMR (CDCl₃) δ 1.37 (3 H, t, J_{H-H} = 7 Hz, CH₃), 2.13 (3 H, s, CH₃), 3.1–3.35, 3.55–4.05, 4.45–4.65 (12 H, m, 6 CH₂), 4.33 (2 H, q, J_{H-H} = 7 Hz, OCH₂), 6.71 (1 H, d, J_{H-F} = 7 Hz, 8-H), 7.91 (1 H, d, J_{H-F} = 13 Hz, 5-H), 8.34 (1 H, s, 2-H); IR (KBr) 1735 (ester), 1622 cm⁻¹ (C==O). Anal. (C₂₀H₂₃ClFN₃O₄) C, H, N. 6-Fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)-1-vinyl-

6-Fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)-1-vinylquinoline-3-carboxylic Acid (67). A mixture of 66 (0.266 g, 0.00063 mol), NaOH (0.252 g, 0.0063 mol), H₂O (5 mL), and EtOH (5 mL) was heated at 95-100 °C with stirring. After 3 h, the solution was concentrated and neutralized with aqueous AcOH. The precipitate was filtered off, washed with H₂O, and dried. The solid was recrystallized from DMF to give 67 (0.173 g, 87%): mp 256-257 °C (dec); ¹H NMR (CF₃COOD) δ 3.5-4.1 (8 H, m, piperazine CH₂), 6.0-6.25 (2 H, m, vinyl H), 7.3-7.7 (2 H, m, vinyl H and 8-H), 8.30 (1 H, d, J_{H-F} = 13 Hz, 5-H), 9.22 (1 H, s, 2-H); IR (KBr) 1618 cm⁻¹ (C=O). Anal. (C₁₆H₁₆FN₃O₃·0.25H₂O) C, H, N.

Alkylation of 7-Amino-4-oxoquinolines (34 and 67; Table V). Method E. A mixture of 34 (3.2 g, 0.01 mol), Et₃N (1.5 g, 0.015 mol), alkyl halide (0.012-0.02 mol) shown in Table V, and DMF (40 mL) was heated at 80–90 °C with stirring. After 2 h, the mixture was concentrated to dryness. The residue was recrystallized from an appropriate solvent to give the corresponding alkylpiperazine derivative (36, 68–71, or 72).

Method F. To a solution of 87% HCOOH (10 mL) and 37% HCHO (10 mL) was added 0.01 mol of 34 or 67. The mixture was refluxed with stirring. After 4–7 h, the mixture was evaporated to dryness and the residue was dissolved in H_2O , neutralized with aqueous NaOH, and extracted with CH_2Cl_2 . After working up, the solid was recrystallized from DMF to yield 36 or 76.

Acylation of 7-Amino-4-oxoquinolines (34, 51, and 63; Table V). Method G. A mixture of 0.01 mol of 34, 51, or 63, Et_3N (1.0–1.5 g, 0.01–0.015 mol), and acylating agent (0.01–0.5 mol), shown in Table V, was heated at 90–100 °C with stirring. After 2–5 h, the mixture was evaporated to dryness and the residue was treated with H_2O and filtered off. The solid was washed with H_2O , dried, and recrystallized from an appropriate solvent to give the corresponding acyl derivative (73–75, 77, or 78).

1-Ethyl-6-fluoro-1,4-dihydro-4-oxo-7-[4-(p-aminobenzyl)-1-piperazinyl]quinoline-3-carboxylic Acid (79). A mixture of 72 (2.0 g, 0.0044 mol), AcOH (50 mL), and 5% palladium on charcoal (0.4 g) was hydrogenated at room temperature until about 300 mL of hydrogen was taken up. The slurry was filtered and the filtrate concentrated to dryness. To the residue were added concentrated HCl and EtOH. The resulting solid was filtered off and recrystallized from H₂O-EtOH to yield 79 (1.4 g, 64%): mp 220-223 °C (dec). Anal. ($C_{22}H_{25}FN_4O_3$ ·2HCl·0.5H₂O) C, H, N.

1-Ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)quinoline (80). A solution of 34 (10 g, 0.031 mol) in 600 mL of 2 N HCl was refluxed. After 50 h, the aqueous solution was concentrated and made strongly basic with aqueous 10% NaOH. The precipitate was extracted with CH₂Cl₂. After working up, the solid was recrystallized from H₂O to give 80 (5.94 g, 69%), mp 209-211 °C. Anal. ($C_{15}H_{18}FN_{3}O$) C, H, N.

1-Ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)quinoline-3-carboxylic Acid Methyl Ester (81) and Ethyl Ester (82). To an ice-cooled mixture of the hydrochloride of 34 (3.56 g, 0.01 mol) and absolute MeOH (100 mL) was added dropwise SOCl₂ (24 g, 0.2 mol). The mixture was refluxed for 12.5 h and evaporated to dryness. The residue was made basic with aqueous K₂CO₃ and extracted with CH₂Cl₂. After working up, the solid was recrystallized from CH₃CN to give 81 (0.80 g, 47%): mp 189–190 °C; ¹H NMR (CDCl₃) δ 1.50 (3 H, t, J_{H-H} = 7 Hz, CH₃), 2.08 (1 H, s, NH), 3.0–3.3 (8 H, m, piperazine CH₂), 3.89 (3 H, s, OCH₃), 4.16 (2 H, q, J_{H-H} = 7 Hz, CH₂), 6.67 (1 H, d, J_{H-F} = 7 Hz, 8-H), 7.94 (1 H, d, J_{H-F} = 13 Hz, 5-H), 8.33 (1

Table VI. In Vitro Antibacterial Activity

·	min inhibitory concn, µg/mL		
compd	S. aureus 209 P	E. coli NIHJ JC-2	P. aeruginosa V-1
18	12.5	1.56	100
32	6.25	0.39	50
33	>100	3.13	>100
34	0.39	0.05	0.39
36	0.39	0.10	1.56
37	1.56	0.20	3.13
38	1.56	0.78	25
39	3.13	0.39	12.5
40	1.56	0.39	100
41	3.13	0.39	6.25
42	25	0.78	12.5
43	100	100	>100
44	12.5	0.39	6.25
45	25	0.78	12.5
46 47	3.13	0.39	6.25
47	$\begin{array}{r}12.5\\6.25\end{array}$	0.78	1.56
48 49	6.25 1.56	0.39 0.10	1.56
49 50	0.39	0.10	$\begin{array}{c} 0.78\\ 3.13\end{array}$
50 51	1.56	0.10	3.13
52	1.56	0.20	3.13
53	3.13	0.20	1.56
54	1.56	0.78	1.56
55	0.78	0.39	50
56	0.20	0.39	12.5
57	0.78	1.56	50
58	0.78	0.20	12.5
59	0.39	0.20	12.5
60	1.56	1.56	100
61	0.39	0.10	3.13
62	3.13	0.39	12.5
63	>100	6.25	50
64	>100	>100	>100
67	3.13	0.10	0.39
68	0.39	0.10	3.13
6 9	0.78	0.10	6.25
70	0.39	0.39	6.25
71	0.39	0.78	50
72	1.56	6.25	>100
73	1.56	0.39	6.25
74	0.78	1.56	25
75 76	$1.56 \\ 1.56$	3.13	25
78 78	>100	$\begin{array}{c} 0.10\\ 25\end{array}$	3.13
79			>100
80	0.39 >100	0.39 >100	12.5 >100
81	100	12.5	>100 50
82	50	12.5 12.5	50
1	>100	3.13	100
3	12.5	0.78	3.13
83	25	1.56	12.5
84	3.13	0.10	25

H, s, 2-H); IR (KBr) 3330 (NH), 1725 (ester), 1628 cm⁻¹ (C=O); MS, m/e 333 (M⁺), 291. Anal. (C₁₇H₂₀FN₃O₃) C, H, N.

When the hydrochloride of 34 (3.56 g, 0.01 mol), absolute EtOH (100 mL), and SOCl₂ (24 g, 0.2 mol) were treated under the above conditions, 82 (3.20 g, 92%) was obtained after recrystallization from CH₃CN. 82: mp 179–180 °C. Anal. ($C_{18}H_{22}FN_3O_3$) C, H, N.

In Vitro Antibacterial Activity. The MIC $(\mu g/mL)$ of compounds was determined by means of a standard twofold serial dilution method using agar media.¹³

Acknowledgment. We are indebted to Dr. S. Sato, Y. Abe, and their members for their helpful discussion and valuable technical assistance. We thank the staff of the analytical center for spectral measurement and elemental analysis.

⁽¹³⁾ M. Ohgoshi, Chemotherapy, 22, 1126 (1974).