Regular Article

Novel 6-Phenylnicotinohydrazide Derivatives: Design, Synthesis and Biological Evaluation as a Novel Class of Antitubercular and Antimicrobial Agents

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In our ongoing efforts to develop potent antitubercular agents based on the 6-phenylnicotinohydrazide, herein we report the design, synthesis and biological evaluation of three sets of 6-phenylnicotinohydrazide derivatives 8a–g, 12 and 16a, b. The designed compounds were synthesized and *in vitro* evaluated for their antitubercular activity. In addition, their antifungal and antibacterial activities were evaluated as well. The nicotinohydrazide class displayed different levels of antimicrobial activity and possessed a distinctive pattern of selectivity against the tested microorganisms. However, the 2,6-dichlorobenzylidene counterpart 8b emerged as the most active one in this study, with superior antimycobacterial activities with MIC range of $0.24-1.95\mu$ g/mL. The structure–activity relationship for such nicotinohydrazides has been established. Further, the cytotoxicity of the most active antitubercular compounds 8b, d and g were tested against the normal breast cells WI-38; none of them displayed significant cytotoxic effect, thereby providing a good therapeutic index.

Key words 6-phenylnicotinohydrazide; antitubercular; antimicrobial; synthesis; cytotoxicity

Recently, the report of the WHO¹⁾ on universal surveillance of the antimicrobial resistance disclosed that antibiotic resistance is no longer a prediction for the future; it is occurring worldwide, and is putting at risk the ability to treat the major infections in the community and hospitals. Without an urgent and coordinated action, the world is moving towards a post-antibiotic era, in which common infections and minor injuries can once again kill.¹⁾ The increasingly rising prevalence of multi-drug resistant microbial infections in the past few decades has become a serious health care problem.¹⁾ The antibiotic resistance crisis could be substantially attributed to the overuse and misuse of these medications, as well as a lack of new drug development by the pharmaceutical industry because of the reduced economic momentum and exigent (demanding) regulatory requirements.²⁻⁷⁾ In this sense, treatment of infectious diseases still remains an important and challenging health problem. There is an urgent need for the discovery of new agents with antimicrobial activity that is distinct from the well-known classes of antimicrobial agents to which many clinically relevant pathogens have become resistant.^{8,9)}

Tuberculosis (TB) is a top infectious disease killer worldwide. In the twentieth global report on TB published by WHO in a series that started in 1997, provided evidence that in 2014, TB killed 1.5 million people (1.1 million human immunodeficiency virus (HIV)-negative and 0.4 million HIV-positive). Worldwide, 9.6 million people are estimated to have fallen ill with TB in 2014 and 12 million prevalent latent cases of TB reported. Globally, 12% of the 9.6 million new TB cases in 2014 were HIV-positive. Moreover, in 2014, an estimated 480000 people worldwide developed multidrug-resistant TB (MDR-TB).¹⁰⁾ The incidence of MDR-TB, is defined as resistance to the first-line drugs isoniazid and rifampicin, while extensively drug-resistant (XDR-TB), is defined as resistance to fluoroquinolones and to at least one of the injectable secondline drugs.^{11,12} The emergence of MDR and XDR strains of *Mycobacterium tuberculosis* renders disease control difficult.

From 2016, the goal is to end the global TB epidemic by implementing the End TB Strategy. The strategy serves as a blueprint for countries to reduce the number of TB deaths by 90% by 2030.13) Therefore, the resurgence of this disease spurred the coordinated urge to develop novel antimycobacterial and antimicrobial agents in order to combat such resistance. Isoniazid (INH) I (Fig. 1), is one of the first-line clinically approved drugs in TB treatment.¹⁴⁻¹⁸⁾ It was also considered as a starting point in the search for new active analogues such as hydrazones, which could be considered as an important pharmacophore for the antibacterial activities, such as Nifuroxazide II^{19-21} (Fig. 1), and also for antitubercular activities.^{22–25)} Additionally, some antitubercular drugs such as Phtivazid III, Verazide IV, Salinazide V and Opiniazide VI are isonicotinic acid hydrazones²⁴⁻²⁶⁾ (Fig. 1). Moreover, numerous studies pinpointed the importance of this scaffold for the development of unprecedented anti-TB agent.^{17,27-29)}

The present study is a part of our ongoing efforts towards



Fig. 1. Structures of Antitubercular Agents I-VI, and the Target Compounds (8a-g, 12 and 16a, b)

developing potent antimycobacterial agents,^{30–32)} by the utilization of the 6-phenylnicotinohydrazide scaffold. In a previous study³⁰⁾ we developed seven hydrazones based on the 6-phenylnicotinohydrazide scaffold as potential antitubercular agents. The structure–activity relationship (SAR) study showed that grafting large lipophilic groups on the pendant phenyl ring as 2,4-dichloro substitution, compound **8a**, could achieve an antimycobacterial activity. Motivated by this finding, we aimed to synthesis three different sets of 6-phenylnicotinohydrazide derivatives **8a–g**, **12** and **16a**, **b** (Fig. 1), with the prime goal of developing more potent and safe antimycobacterial agents and also screening their antifungal and antibacterial activities.

Four distinctive strategies were adopted to develop such derivatives. The first one focused on the usage of different substituents (disubstitution) on the pendant phenyl ring, compounds 8b-e, to ensure different lipophilic and electronic environments which may manipulate the activities of the target derivatives. In the second strategy, an extension approach was applied for compound 8e with the large lipophilic 4-chlorophenyl moiety through ketomethylene (-CH₂-CO-) linker, compound 12, to achieve possible additional binding interactions. The third strategy depends on the usage of the hydrophobic alicyclic secondary amines; morpholine or piperidine as in compounds 16a and b, respectively. Finally, a bioisosteric tactic was utilized to replace the pendant substituted phenyl ring in compounds 8a-e with 2-thiazolyl moiety or 3-pyridyl moiety, compounds 8f and g, respectively, to carry out further elaboration of the 6-phenylnicotinohydrazide scaffold and to investigate a valuable SAR.

It is noteworthy that no attention was paid to explore the antibacterial and antifungal activities for the 6-phenylnicotinohydrazide scaffold, so far. Therefore, the synthesized derivatives **8a–g**, **12** and **16a**, **b** not only were evaluated for their anti-tubercular activity, but also assessed for their antibacterial and antifungal activities. Also, the cytotoxicity of the most active molecules was tested against the normal breast cells WI-38.

RESULTS AND DISCUSSION

Discussion of Chemistry The synthetic strategy for preparing the target nicotinic acid hydrazides 8a-g, 12 and 16a, b was illustrated in Charts 1–3. In Chart 1, synthesis of target hydrazides required three successive reactions. The enaminone 3 reacted with ethyl acetoacetate 4 and ammonium acetate in hot glacial acetic acid, to furnish the ethyl 6-(4-bromophenyl)-2-methylnicotinate 5 in a three-component heterocyclocondensation process. This was followed by hydrazinolysis of the later ester 5 to yield the nicotinic acid hydrazide 6, in 90% yield. Condensation of the hydrazide 6 with different aldehydes 7a-g, in ethyl alcohol in the presence of a catalytic amount of acetic acid furnished the target compounds 8a-g (Chart 1).

IR spectra of compounds **8a–g** revealed the appearance of bands around 1660 and 3170 cm⁻¹ assigned to the carbonyl and NH groups, respectively. Also, ¹H-NMR spectra of **8a–g** displayed a singlet D₂O-exchangeable signal attributable to NH proton of hydrazine function (=N–NH–) around δ 12 *ppm*, while the methyl (–CH₃) and methine (–CH=N–) protons appeared around δ 2.60 and 8.40 *ppm*, respectively, as singlet signals. Also, additional singlet signals corresponding to methoxy protons resonating around δ 3.70 *ppm* in case of **8c** and **d** were observed. Furthermore, ¹³C-NMR showed the characteristic signals for the carbons of carbonyl groups around δ 170.30 *ppm*, whereas the carbons for the methyl groups ap-



Reagents and conditions: i, xylene, reflux 8h; ii, $NH_4OAc/AcOH/reflux 5h$ (86% yield); iii, $NH_2NH_2.H_2O/reflux 3h$ (90% yield); iv, EtOH/AcOH (catalytic)/reflux 4h (65–83% yield).

78%

80%

Chart 1



Reagents and conditions: i, DMF, K₂CO₃, r.t. 1 h (55% Yield); ii, EtOH/AcOH (catalytic)/reflux 4 h (64% yield). Chart 2

thiophen-2-yl

pyridin-3-yl

f

g



Reagents and conditions: i, DMSO, K₂CO₃, reflux 6h (65 and 70% yield, respectively); ii, EtOH/AcOH (catalytic)/reflux 4h (75 and 80% yield, respectively).

On the other hand, O-alkylation of 5-bromo-2-hydroxybenzaldehyde 9 with 2-bromo-1-(4-chlorophenyl)ethanone 10 in dry N,N-dimethylformamide (DMF) in presence of potassium carbonate yielded aldehyde derivative 11. The latter was heated under reflux with hydrazide 6 in absolute ethyl alcohol and catalytic amount of glacial acetic acid to afford the target derivative 12 (Chart 2).

IR spectrum of **12** confirmed the presence of two peaks at 1701 and 3155 cm^{-1} corresponding to the (C=O) and NH groups, respectively. The ¹H-NMR spectrum of **12** showed two singlet signals of the aliphatic (–OCH₂–) and methine (–CH=N–) protons at δ 5.72 and 8.00 *ppm*, respectively, also, ¹H-NMR spectrum revealed the presence of D₂O exchangable NH proton at a δ 12.14 *ppm*.

Finally, *p*-flourobenzaldehyde 13 was reacted with an appropriate alicyclic secondary amine 14a, b in dry dimethyl sulfoxide (DMSO) in presence of potassium carbonate to furnish aldehydes 15a, b (yield, 65 and 70%, respectively), which condensed with hydrazide 6 in absolute ethyl alcohol and catalytic amount of glacial acetic acid to afford the respective target derivative 16a, b in good yield (Chart 3).

¹H-NMR spectra for the latter products showed singlet D_2O -exchangeable signals attributable to NH proton of the hydrazine function (=N–NH–) around δ 11.70 *ppm*, while the methyl (–CH₃) protons appeared at δ 2.65 *ppm* as singlet signals. Also, their ¹H-NMR revealed the protons of morpholine and piperidine moiety in the aliphatic regions; (δ 1.53–1.59 and 3.19 *ppm* for piperidine moiety) and (δ 3.11 and 3.367 *ppm* for morpholine moiety).

The prepared hydrazones in this study appeared in the ¹H-NMR spectra (in the solution state, DMSO- d_6) as mixed *E* and *Z* geometrical isomers due to the imine C=N double bond. While, the X-ray single crystal analysis for compound **8a** illustrates the interesting property of such hydrazone in the crystalline state as an *E* geometrical isomer around (C=N) (Supplementary Materials).^{33–35})

Discussion of Biological Evaluation

Anti-microbial Activity

Antimycobacterial, antifungal and antibacterial activities for the target compounds **8a–g**, **12**, **16a**, **b** and reference drugs were *in vitro* evaluated at the Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University, Cairo, Egypt. The results are represented in Tables 1–3.

Antimycobacterial Activity

Target compounds **8a–g**, **12**, **16a**, **b** were *in vitro* evaluated for their antitubercular activity against *M. tuberculosis* (RCMB 010126) using the microplate Alamar blue assay (MABA).³⁶⁾ Isoniazide and pyrazinamide were used as reference drugs. The results of the *in vitro* antimycobacterial screening, as percent inhibition and minimum inhibitory concentration (MIC), are summarized in Table 1. Among the tested hydrazones, compound **8b** emerged as the most potent member with significant antimycobacterial activity with a MIC value of $3.9 \,\mu$ g/mL. Compound **8g** also exhibited moderate activity (MIC=15.63 μ g/mL). Otherwise, compounds **8c**, **8d** and **12** possessed modest anti-tubercular activity with MIC values range of: 62.5 to 125 μ g/mL.

Upon examining the MIC results in Table 1, worthy information about the structure activity relationships, of compounds 8a-g, 12, 16a, b with respect to their antitubercular activity, could be concluded. Firstly, we examined the effect of grafting different two substituents on the pendent phenyl ring on the prepared hydrazones activity. Incorporation of two lipophilic groups as chloro or methoxy groups led to compounds **8b–d** with potent to modest activity (MIC: $3.90-100 \mu$ g/mL). Conversely, introduction of hydrophilic group (–OH) and lipophilic one (–Br), as in compound **8e**, abolished the activity. Also, appending alicyclic secondary amines; morpholine or piperidine as in compounds **16a** and **b**, diminished the activity, suggesting that substitution of the pendant phenyl group with two lipophilic groups is crucial for the antimycobacterial activity.

On the other hand, the positional isomerism; moving the two chlorine atoms from 2,4 positions in compound **8a** to 2,6 positions in **8b**, resulted in about 25 fold increase in the antimycobacterial activity, hinting that the positions of such two lipophilic groups greatly affect the activity. The order of activity of the disubstituted hydrazones **8a–d** was 2,6-dichloro **8b**>3,4-dimethoxy **8c**>2,4-dichloro **8a** and 2,5-dimethoxy **8d** (MIC=3.90, 62.5, 100 and $100 \,\mu$ g/mL, respectively). Moreover, extending compound **8e** (MIC=>150 μ g/mL) with the large lipophilic 4-chlorophenyl moiety through ketomethylene (-CH₂-CO-) linker, led to a comparatively active counterpart **12** (MIC=125 μ g/mL), confirming that presence of hydrophilic group is not favorable for activity.

Finally, we examined the effect of bioisosteric replacement of the pendant substituted phenyl moiety in compounds **8a**-d. Interestingly, attaching 3-pyridyl group, compound **8g**, as bioisostere to the phenyl one maintained the activity (MIC= $62.5 \mu g/mL$), while 2-thienyl group, **8f**, abolished the activity.

Anti-fungal Activity

Target compounds **8a–g**, **12**, **16a**, **b** and amphotericin, a membrane-active polyene macrolide antibiotic and an antifungal reference drug, were *in vitro* evaluated for their antifungal activity, by inhibition zone (I.Z.) technique and MIC,³²⁾ towards *Aspergillus fumigatus* and *Candida albicans*, Table 2.

Aspergillus species, globally ubiquitous saprophytes, are found in a variety of ecological niches. More than 200 species of aspergilli have been identified, less than 20 of which are recognized to cause human disease. Among them, *A. fumigatus* was found to be the most prevalent and is predominantly responsible for the increased incidence of invasive aspergillosis in the immunocompromised patient.

As for the antifungal activity, data in Table 2 showed that the 2,5-dimethoxybenzylidene derivative **8d** demonstrated the best activity amongst the tested compounds against *A. fumigatus* (MIC=0.49 μ g/mL), which was 2-fold more potent than the reference dug, amphotericin (MIC=0.98 μ g/mL). While, compounds **8b** and **g** were equipotent to amphotericin against *A. fumigatus*. Furthermore, derivatives **8a**, **c**, **f**, **12** and **16b** showed moderate activity against *A. fumigatus* (MIC=7.81, 1.95, 7.81, 3.9 and 15.63 μ g/mL, respectively)

Regarding the activity against *Candida albicans*, the 2,6-dichlorobenzylidene counterpart **8b** was the most active one with MIC=0.98 μ g/mL. Also, compound **8d** displayed good activity and was equipotent to amphotericin against *Candida albicans*, (MIC=1.95 μ g/mL). Moreover, compounds **8a**, c, f, g, **12** and **16b** were moderately active, against *Candida albicans*, with MIC values ranging from 3.9 to 31.25 μ g/mL.

Table 1.	Structures,	Antitubercular	Activities,	$\log P$	Measurements	and 1	Drug-Likeness	Model	Score	of the	Synthesize	ed
Derivative	S											

Compound	Structure	Mean of inhibition %	MIC (μg/ml)	LogP ^a	Drug-likeness model score ^b
8a	Br Cl Cl	13.57±0.72	100	6.23	0.13
8b		79.21± 3.51	3.90	6.21	0.35
8c	Br N N N O	52.14± 2.28	62.5	4.60	0.66
8d	Br N N N P	27.33± 1.50	100	4.99	0.07
8e	Br N N OH	NA	NA	5.67	0.47
8f	Br N N S	NA	NA	4.85	0.20
8g	Br N N N N N N N N N N N N N N N N N N N	67.28± 0.58	15.63	3.71	0.78
12	Br Cr N Cr	22.43± 1.59	125	7.57	0.36
16a	Br N N N N N N N N N N N N N N N N N N N	NA	NA	4.90	0.20
16b	Br C N N N N N N N N N N N N N N N N N N	NA	NA	5.96	-0.09
Pyrazinamid		93.25±1.63	3.21		
INH		83.2 ± 2.16	3.90		

^a: Calculated by [www.molinspiration.com]; ^b: Calculated by ref. 48; NA: No activity (>150 µg/mL).

Antibacterial Activity

The results in Table 3 showed that the compound **8b** demonstrated broad and excellent activity among all the tested compounds, it was twice the activity of ampicillin against *Staphylococcus aureus* and *Bacillus subtilis* (0.49, $0.24 \,\mu$ g/mL, ampicillin, 0.98, $0.49 \,\mu$ g/mL), it was also equipotent to Gentamicin against *Salmonella typhimurium*, *Klebsi*-

ella penumoniae and Escherichia coli (0.98, 0.49, 0.49 μ g/mL). The (2,4-dichlorobenzylidene)-2-methylnicotinohydrazide **8a** was equipotent to ampicillin against *S. aureus*, (0.98 μ g/mL), it also possessed the same activity against *Streptococus pneumoniae* and *B. subtilis* as ciprofloxacin (1.95 μ g/mL). As for the Gram-negative (g-ve) activity, it exhibited 4-foldsthe activity of ciprofloxacin against *K. penumoniae* and possessed

Table 2. Antifungal Activity of the Tested Standards and Synthesized Compounds; Expressed as Inhibition Diameter Zones (I.Z.) in mm, and Minimum Inhibitory Concentrations (MIC) in μ g/mL



Compound	D/A r/V	I.Z.	(mm)	MIC (µg/mL)			
Compound	K/AI/A —	Af	Са	Af	Ca		
8a	2,4-(Cl ₂)-C ₆ H ₃ -	18.2±0.38	17.3±0.58	7.81	15.63		
8b	2,6-(Cl ₂)-C ₆ H ₃ -	23.4 ± 0.25	24.1±1.5	0.98	0.98		
8c	3,4-(OCH ₃) ₂ -C ₆ H ₃ -	21.2 ± 0.58	18.3 ± 0.58	1.95	7.81		
8d	2,5-(OCH ₃) ₂ -C ₆ H ₃ -	24.7 ± 0.38	20.8±1.2	0.49	1.95		
8e	5-Br-2-OH-C ₆ H ₃ -	20.1 ± 0.63	18.3 ± 0.58	62.5	125		
8f	Thiophen-2-yl	18.2 ± 0.58	15.4±0.34	7.81	31.25		
8g	Pyridin-3-yl	22.9±0.25	19.3±1.53	0.98	3.9		
12	_	13.6±0.37	11.4 ± 1.20	3.9	7.81		
16a	0	14.2 ± 0.25	13.6±1.25	62.5	62.5		
16b	CH ₂	16.9±0.58	16.4±1.33	15.63	31.25		
AB	2	23.1 ± 0.90	21.3±0.72	0.98	1.95		

The screening organisms, Aspergillus fumigatus (RCMB 02568, Af), Candida albicans (RCMB 05036, Ca), AB: Amphotericin B.

Table 3.	Antibacterial Activity	of the Test	ed Standards	and Synthesized	Compounds;	Expressed a	as Inhibition	Diameter	Zones ((.Z.) in	mm, a	and Mini-
mum Inhib	itory Concentrations	(MIC) in $\mu g/$	mL									

	Gram-positive bacteria						Gram-negative bacteria								
Comp.	I.Z. (mm)			MIC (µg/mL)			I.Z. (mm)				MIC (µg/mL)				
	Sa	Sp	Bs	Sa	Sp	Bs	Ра	St	Kp	Ec	Ра	St	Кр	Ec	
8a	22.4	21.3	20.6	0.98	1.95	1.95	18.3	21.3	22.4	22.6	7.81	1.95	0.98	0.98	
8b	25.2	24.1	26.4	0.49	0.49	0.24	21.3	22.7	25.4	26.3	1.95	0.98	0.49	0.49	
8c	21.3	22.1	20.3	1.95	0.98	3.9	NA	18.4	20.3	21.2	NA	7.81	3.9	1.95	
8d	20.4	22.6	19.8	3.9	0.98	3.9	21.3	23.4	24.9	26.4	1.95	0.98	0.49	0.49	
8e	18.7	21.3	20.2	7.81	7.81	31.25	NA	18.3	21.2	21.9	NA	31.25	15.63	7.81	
8f	13.6	16.4	12.4	62.5	31.25	125	NA	13.2	15.7	17.3	NA	62.5	31.25	15.63	
8g	24.2	23.6	21.3	0.49	0.49	1.95	19.6	22.4	24.6	25.1	3.9	0.98	0.49	0.49	
12	18.2	17.6	16.1	3.9	1.95	3.9	NA	15.6	17.1	17.9	NA	7.81	1.95	62.5	
16a	17.3	17.9	13.6	15.63	7.81	62.5	NA	11.2	13.4	16.2	NA	125	62.5	31.25	
16b	21.9	21.2	20.1	0.98	1.95	3.9	16.3	20.4	21.6	22.2	31.25	3.9	0.98	0.98	
Ар	26.4	25.3	28.9	0.98	0.49	0.49									
GM							26.3	23.2	27.5	27.3	0.98	0.98	0.49	0.49	
CF	22.2	23.3	25.4	1.95	1.95	1.95	23.1	21.8	24.7	24.3	3.90	3.90	3.90	1.95	

NA: No Activity. The screening organisms, Gram-positive bacteria: *Staphylococcus aureus* (RCMB 010028, Sa), *Streptococus pneumoniae* (RCMB 010010, Sp), and *Bacillus subtilis* (RCMB 010069, Bs). Gram-negative bacteria: *Pseudomonas aeruginosa* (RCMB 010043, Pa), *Salmonella typhimurium* (RCMB 010315, St), *Klebsiella penumoniae* (RCMB 0010093, Kp) and *Escherichia coli* (RCMB 010052, Ec), AP: Ampicillin, GM: Gentamicin, CF: Ciprofloxacin.

twice its activity against *S. typhimurium* and *E. coli*. On the other hand, compound **8g** demonstrated twice the activity of ampicillin against *S. aureus*, the same potency as ampicillin against (*S. pneumoniae*) as well as (the effect of gentamicin on) the g-ve organisms, *S. typhimurium, K. penumoniae* and *E. coli*. Furthermore, the benzylidene derivative **8c** showed equipotent activity to ciprofloxacin against *S. aureus, K. penu*-

moniae, E. coli and was almost 2-folds its activity against S. pneumoniae. Regarding the activity of piperidinylbenzylidene counterpart **16b**, it exhibited g-ve and Gram-positive (g+ve) activity, it was found to demonstrate the same activity as ampicillin against S. aureus (0.98 μ g/mL) and same potency as ciprofloxacin against S. pneumoniae (1.95 μ g/mL). The g-ve activity of this compound **16b**, was also demonstrated by its



Fig. 2. Antitubercular Activity (MIC) and Cytotoxicity (IC₅₀) of Compounds 8b, c and g against Normal Breast Cells WI-38

activity against *S. typhimurium* with the same potency as ciprofloxacin ($3.9\,\mu$ g/mL), 4-folds its activity against *K. penumoniae* and 2-folds the activity against *E. coli*. In particular, the arylidene-hydrazide derivatives **8b**, **d** and **g** were the only derivatives in this series that showed activity against the g-ve resistant, *Pseudomonas aeruginosa*, where **8b** and **d** were twice as potent as ciprofloxacin (1.95 and $3.9\,\mu$ g/mL) and **8g** was equipotent.

In vitro Cytotoxicity

The cytotoxic activity of the most active antitubercular derivatives **8b**, **c** and **g** were tested against normal breast cells WI-38 using 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay.³⁷⁾ The results are expressed as growth inhibitory concentration values (IC₅₀) which represent the concentration of the compound needed to make a 50% inhibition of cell growth after 48h of incubation compared to untreated control. None of the three compounds displayed significant cytotoxic effect (IC₅₀=127.7, 98.2 and 146.04 μ M, respectively), thereby providing a good therapeutic index (Fig. 2).

CONCLUSION

In conclusion, three sets of 6-phenylnicotinohydrazide derivatives 8a-g, 12 and 16a, b were designed, synthesized and tested in vitro for their activity against M. tuberculosis. Also, the antifungal and antibacterial activities for such compounds were evaluated. Compound 8b was found to be the most promising member in this study, with superior antimycobacterial activity (MIC= $3.90 \,\mu \text{g/mL}$) and potent broad-spectrum antimicrobial activities with MIC range of 0.24-1.95 µg/mL. The impact of incorporation of two large lipophilic groups, extension approach and bioisosteric replacement of the pendant phenyl moiety, was explored by the SAR study. The cytotoxicity of the most active antitubercular compounds 8b, d and g were tested against the normal breast cells WI-38; none of them displayed significant cytotoxic effect, thereby providing a good therapeutic index. Structures of compounds 8a and 11 were further substantiated via X-ray single crystal analysis.

MATERIALS AND METHODS

General Chemistry Melting points were measured with a Stuart melting point apparatus and were uncorrected. IR spectra were recorded as potassium bromide discs on Shimadzu FT-IR 8400S spectrophotometer and expressed in wave number (cm⁻¹). The NMR spectra were recorded by Bruker spectrophotometer at 400 MHz. ¹H-NMR spectra were run at 400 MHz, while ¹³C-NMR spectra were run at 100 MHz in deuterated dimethylsulphoxide (DMSO-d₆). Chemical shifts ($\delta_{\rm H}$) are reported relative to TMS as internal standard. All coupling constant (J) values are given in hertz. Chemical shifts (δ_c) are reported relative to DMSO- d_6 as internal standards. The abbreviations used are as follows: s, singlet; d, doublet; m, multiplet. Mass spectra were measured on a GCMS-QP1000 EX and Helwett Packard 5988 spectrometers at 70 e.V. Elemental analyses was carried out at the Regional Center for Microbiology and Biotechnology, Al-Azhar University, Cairo, Egypt. Reaction courses and product mixtures were routinely monitored by TLC on silica gel precoated F₂₅₄ Merck plates. Unless otherwise noted, all solvents and reagents were commercially available and used without further purification.

1-(4-Bromophenyl)-3-(dimethylamino)prop-2-en-1-one 3

Compound **3** was prepared according to reported procedure³⁸⁾ (mp 79–80°C, reported 75–77°C³⁸⁾).

Ethyl 6-(4-Bromophenyl)-2-methylnicotinate 5

Compound **5** was prepared according to reported procedure³⁹ (mp 74–75°C, reported $72-74^{40}$).

6-(4-Bromophenyl)-2-methylnicotinohydrazide 6

Compound **6** was prepared according to reported procedure³⁰ (mp 239–241°C, as reported³⁰).

General Procedure for Preparation of *N'*-Arylidene-6-(4-bromophenyl)-2-methylnicotinohydrazides 8a-g The appropriate aldehyde derivative 7a-g (1 mmol) was added to a solution of the 6-(4-bromophenyl)-2-methylnicotinohydrazide 6 (0.31 g, 1 mmol) in absolute ethyl alcohol (10 mL) and a catalytic amount of glacial acetic acid. The reaction mixture was heated under reflux for 3 h. The formed precipitate was filtered off while hot, washed with methanol, and finally crystallized from ethanol/dioxane to furnish the target compounds 8a-g with 65-83% yield.

6-(4-Bromophenyl)-*N*'-(2,4-dichlorobenzylidene)-2methylnicotinohydrazide (**8a**)

White crystals (yield 70%), mp 269-271°C.³⁰⁾

6-(4-Bromophenyl)-*N*'-(2,6-dichlorobenzylidene)-2methylnicotinohydrazide (**8b**)

White crystals (yield 73%), mp 261–263°C; IR (KBr, $v \text{ cm}^{-1}$): 3167 (NH) and 1651 (C=O); ¹H-NMR (DMSO- d_6) δ ppm: 2.53, 2.65 (2 s, 3H, CH₃), 7.35–7.42 (m, 3H, Ar-H), 7.71 (d, 2H, Ar-H, J=8.40 Hz), 7.85–8.00 (m, 2H, Ar-H), 8.09 (d, 2H, Ar-H, J=8.48 Hz), 8.34, 8.62 (2 s, 1H, Ar-H), 11.96, 12.03 (2 s, 1H, NH, D₂O exchangeable); *Anal.* Calcd for C₂₀H₁₄BrCl₂N₃O: C, 51.87; H, 3.05; N, 9.07. Found: C, 51.54; H, 3.09; N, 9.15.

6-(4-Bromophenyl)-N'-(3,4-dimethoxybenzylidene)-2methylnicotinohydrazide (8c)

White crystals (yield 83%), mp 233–235°C; IR (KBr, $v \text{cm}^{-1}$): 3150 (NH) and 1651 (C=O); ¹H-NMR (DMSO- d_6) δ ppm: 2.53, 2.65 (2 s, 3H, CH₃), 3.58, 3.72, 3.81 (3 s, 6H, 2-OCH₃), 6.92–7.20 (m, 2H, Ar-H), 7.37 (s, 1H, Ar-H), 7.70 (d, 2H, Ar-H, *J*=8.44Hz), 7.85–8.00 (m, 2H, Ar-H), 8.09 (d, 2H, Ar-H, *J*=8.52Hz), 8.24 (s, 1H, Ar-H), 11.80, 11.91 (2 s, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6) δ ppm: 23.43 (CH₃), 55.52, 55.92 (OCH₃), 55.98, 56.04 (OCH₃), 108.73, 109.28, 111.93, 112.10, 117.06, 117.58, 120.89, 122.49, 123.41, 123.63, 127.25, 129.11, 129.24, 129.85, 132.24, 137.31, 137.51, 137.56, 137.79, 144.24, 148.58, 149.26, 149.55, 150.88, 151.35, 154.72, 155.16, 155.41, 156.40, 164.21, 170.21; *Anal.* Calcd for C₂₂H₂₀BrN₃O₃: C, 58.16; H, 4.44; N, 9.25. Found: C, 58.34; H, 4.48; N, 9.16.

6-(4-Bromophenyl)-*N*'-(2,5-dimethoxybenzylidene)-2methylnicotinohydrazide (**8d**)

White crystals (yield 80%), mp 235–237°C; IR (KBr, $v \text{ cm}^{-1}$): 3170 (NH) and 1662 (C=O); ¹H-NMR (DMSO-*d*₆) δ *ppm*: 2.52, 2.66 (2 s, 3 H, CH₃), 3.51, 3.81 (2 s, 3 H, OCH₃), 3.77 (s, 3 H, OCH₃), 6.79–7.39 (m, 3 H, Ar-H), 7.70 (d, 2 H, Ar-H, *J*=8.52 Hz), 7.83–8.00 (m, 2H, Ar-H), 8.10 (d, 2H, Ar-H, *J*=8.40 Hz), 8.37, 8.65 (2 s, 1H, Ar-H), 11.90, 11.96 (2 s, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d*₆) δ *ppm*: 23.43, 23.50 (CH₃), 55.46, 55.94 (OCH₃), 56.64, 56.73 (OCH₃), 109.55, 109.94, 113.73, 113.95, 117.01, 117.10, 117.54, 118.42, 123.11, 123.43, 123.65, 129.13, 129.27, 129.65, 130.60, 132.24, 137.39, 137.54, 139.83, 143.77, 152.51, 152.84, 153.42, 153.74, 154.72, 155.07, 155.46, 156.56, 164.23, 170.35; *Anal.* Calcd for C₂₂H₂₀BrN₃O₃: C, 58.16; H, 4.44; N, 9.25. Found: C, 57.82; H, 4.49; N, 9.18.

N'-(4-Bromo-2-hydroxybenzylidene)-6-(4-bromophenyl)-2methylnicotinohydrazide (**8e**)

White crystals (yield 70%), mp 241–242°C; IR (KBr, $v \text{ cm}^{-1}$): 3182 (NH) and 1651 (C=O); ¹H-NMR (DMSO- d_6) δ ppm: 2.52, 2.67 (2 s, 3H, CH₃), 6.78, 6.90 (2 d, 1H, Ar-H, J=8.68 & 8.76 Hz), 7.31, 7.42 (2 dd, 1H, Ar-H, J=8.68, 2.36 Hz), 7.70 (d, 2H, Ar-H, J=8.04 Hz), 7.82–7.96 (m, 2H, Ar-H), 8.01 (d, 1H, Ar-H, J=8.04 Hz), 8.09 (d, 2H, Ar-H, J=8.44 Hz), 8.32, 8.51 (2 s, 1H, Ar-H), 10.26, 11.14 (2 s, 1H, OH, D₂O exchangeable), 12.09, 12.20 (2 s, 1H, NH, D₂O exchangeable); Anal. Calcd for C₂₀H₁₅Br₂N₃O₂: C, 49.11; H, 3.09; N, 8.59. Found: C, 48.83; H, 3.13; N, 8.68.

6-(4-Bromophenyl)-2-methyl-*N*'-(thiophen-2-ylmethylene)nicotinohydrazide (**8f**)

White crystals (yield 78%), mp 245-246°C; IR (KBr,

νcm⁻¹): 3170 (NH) and 1658 (C=O); ¹H-NMR (DMSO-*d*₆) *δ ppm*: 2.52, 2.65 (2s, 3H, CH₃), 7.04, 7.15 (2t, 1H, Ar-H, *J*=4.40 & 4.76 Hz), 7.37, 7.48 (2 d, 1H, Ar-H, *J*=3.00 & 2.64 Hz), 7.69–7.72 (m, 3H, Ar-H), 7.81–7.98 (m, 2H, Ar-H), 8.09 (d, 2H, Ar-H, *J*=8.52 Hz), 8.28, 8.53 (2s, 1H, Ar-H), 11.87, 11.95 (2s, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d*₆) *δ ppm*: 23.45, 23.55 (CH₃), 117.12, 117.58, 123.46, 123.67, 128.31, 128.39, 129.00, 129.18, 129.26, 129.60, 129.70, 130.25, 130.93, 131.73, 132.25, 137.35, 137.53, 137.60, 137.69, 139.15, 139.33, 139.82, 143.48, 154.77, 155.15, 155.49, 156.47, 164.21, 170.08; *Anal.* Calcd for C₁₈H₁₄BrN₃OS: C, 54.01; H, 3.53; N, 10.50. Found: C, C, 53.82; H, 3.56; N, 10.33.

6-(4-Bromophenyl)-2-methyl-*N*'-(pyridin-3-ylmethylene)nicotinohydrazide (**8**g)

White crystals (yield 80%), mp 247–249°C; IR (KBr, $v \, {\rm cm}^{-1}$): 3190 (NH) and 1662 (C=O); ¹H-NMR (DMSO- d_6) δ ppm: 2.53, 2.66 (2 s, 3H, CH₃), 7.34, 7.48 (2 dd, 1H, Ar-H, J=7.76, 4.84 Hz), 7.70 (d, 2H, Ar-H, J=8.44 Hz), 7.85–8.01 (m, 2H, Ar-H), 8.10 (d, 2H, Ar-H, J=8.44 Hz), 8.15 (d, 1H, Ar-H, J=8.56 Hz), 8.38 (s, 1H, Ar-H), 8.62 (d, 1H, Ar-H, J=4.80 Hz), 8.88 (s, 1H, Ar-H), 12.09, 12.16 (2 s, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6) δ ppm: 23.42, 23.49 (CH₃), 117.22, 117.58, 123.49, 123.70, 124.40, 124.49, 129.22, 129.27, 129.45, 130.23, 130.29, 130.52, 132.20, 132.25, 133.54, 134.03, 137.42, 137.51, 137.66, 142.09, 145.72, 148.78, 149.30, 150.97, 151.33, 154.86, 155.08, 155.58, 156.54, 164.50, 170.52; Anal. Calcd for C₁₉H₁₅BrN₄O: C, 57.74; H, 3.83; N, 14.17. Found: C, 58.08; H, 3.87; N, 14.08.

5-Bromo-2-(2-(4-chlorophenyl)-2-oxoethoxy)benzaldehyde 11 To a stirred mixture of 5-bromosalicylaldehyde **9** (0.2 g, 1 mmol) and anhydrous potassium carbonate (0.27 g, 2 mmol) in absolute ethyl alcohol, a solution of 2-bromo-1-(4-chlorophenyl)ethan-1-one **10** (0.23 g, 1 mmol) in absolute ethyl alcohol was added. The reaction mixture was stirred for 2 h at room temperature. The solid product was collected by filtration, washed with water and ether, and crystallized from isopropanol to afford white crystals of compound **11** (yield 55%), mp 165–166°C; IR (KBr, *v* cm⁻¹): 1670, 1715 (2C=O); ¹H-NMR (DMSO-*d*₆) *δ ppm*: 5.83 (2 s, 2H, $-OCH_2-$), 7.65 (d, 2H, Ar-H, *J*=8.48 Hz), 7.76–7.79 (m, 2H, Ar-H), 8.02 (d, 2H, Ar-H, *J*=8.72 Hz), 8.07 (s, 1H, Ar-H), 10.41 (s, 1H, CHO); *Anal.* Calcd for C₁₅H₁₀BrClO₃: C, 50.95; H, 2.85. Found: C, 51.27; H, 2.79.

N'-(5-Bromo-2-(2-(4-chlorophenvl)-2-oxoethoxy)benzylidene)-6-(4-bromophenyl)-2-methylnicotinohydrazide 12 Compound 11 (0.35 g, 1 mmol) was added to a suspension of 6-(4-bromophenyl)-2-methylnicotinohydrazide 6 (0.31 g, 1 mmol) in absolute ethyl alcohol (10 mL) and a catalytic amount of glacial acetic acid. The reaction mixture was heated under reflux for 2h. The solid formed was filtered, washed with hot ethanol then crystallized from ethanol/DMF to furnish the target compound 12 with (yield 64%), mp 236–238°C; IR (KBr, $v \text{ cm}^{-1}$): 3155 (NH) and 1701 (C=O); ¹H-NMR (DMSO-d₆) δ ppm: 2.54, 2.67 (2s, 3H, CH₃), 5.72, 5.74 (2s, 2H, -OCH₂-), 7.02, 7.12 (2d, 1H, Ar-H, J=8.92 & 8.96 Hz), 7.41, 7.53 (2 dd, 1H, Ar-H, J=8.92, 2.48 Hz), 7.53 (d, 2H, Ar-H, J=8.44 Hz), 7.70 (d, 2H, Ar-H, J=8.48 Hz), 7.88-8.04 (m, 5H, Ar-H), 8.10 (d, 2H, Ar-H, J=8.48 Hz), 8.48, 8.73 (2s, 1H, Ar-H), 12.14 (s, 1H, NH, D₂O exchangeable); Anal. Calcd for C₂₈H₂₀Br₂ClN₃O₃: C, 52.41; H, 3.14; N, 6.55. Found: C, 52.18; H, 3.17; N, 6.69.

4-(Morpholino/1-piperidino)benzaldehyde 15a, b

Compound **15a**, **b** was prepared according to reported procedure⁴¹ (mp 66–69°C, 61–64°C, respectively).

General Procedure for Preparation of the Target Compounds 16a, b The appropriate aldehyde derivative 15a, b (1 mmol) was added to a solution of the 6-(4-bromophenyl)-2-methylnicotinohydrazide 6 (0.31 g, 1 mmol) in absolute ethyl alcohol (10 mL) and a catalytic amount of glacial acetic acid. The reaction mixture was heated under reflux for 4h. The formed precipitate was filtered off while hot, washed with methanol, and finally crystallized from ethanol/dioxane to give the target compounds 16a, b.

6-(4-Bromophenyl)-2-methyl-*N*'-(4-morpholinobenzylidene)nicotinohydrazide (**16a**)

White crystals (yield 75%), mp>300°C; IR (KBr, $v \text{ cm}^{-1}$): 3185 (NH) and 1658 (C=O); ¹H-NMR (DMSO-*d*₆) δ *ppm*: 2.52, 2.65 (2 s, 3H, CH₃), 3.11, 3.21 (2 t, 4H, morpholino-H, *J*=4.52 Hz), 3.67, 3.73 (2 t, 4H, morpholino-H, *J*=4.68 Hz), 6.87 (d, 1H, Ar-H, *J*=8.72 Hz), 7.00 (d, 1H, Ar-H, *J*=8.80 Hz), 7.26 (d, 0.5H, Ar-H, *J*=8.68 Hz), 7.58 (d, 1.5H, Ar-H, *J*=8.72 Hz), 7.70 (d, 2H, Ar-H, *J*=8.52 Hz), 7.81 (d, 0.5H, Ar-H, *J*=8.04 Hz), 7.90–7.99 (m, 2H, Ar-H), 8.09–8.13 (m, 2H, Ar-H), 8.19 (s, 0.5H, Ar-H), 11.69, 11.77 (2 s, 1H, NH, D₂O exchangeable); *Anal.* Calcd for C₂₄H₂₃BrN₄O₂: C, 60.13; H, 4.84; N, 11.69. Found: C, 59.88; H, 4.89; N, 11.74.

6-(4-Bromophenyl)-2-methyl-*N*'-(4-(piperidin-1-yl)benzylidene) Nicotinohydrazide (16b)

White crystals (yield 80%), mp 283–285°C; IR (KBr, $v \text{ cm}^{-1}$): 3170 (NH) and 1661 (C=O); ¹H-NMR (DMSO- d_6) δ ppm: 1.53–1.59 (m, 6H, piperidino-H), 2.64 (s, 3H, CH₃), 3.19 (t, 4H, piperidino-H), 6.84 (d, 1H, Ar-H, J=8.00Hz), 6.97 (d, 1H, Ar-H, J=8.40Hz), 7.21 (d, 0.5H, Ar-H, J=8.28Hz), 7.54 (d, 1.5H, Ar-H, J=8.52Hz), 7.71 (d, 2H, Ar-H, J=8.16Hz), 7.81 (d, 0.5H, Ar-H, J=8.04Hz), 7.90–7.97 (m, 2H, Ar-H), 8.09–8.16 (m, 2.5H, Ar-H), 11.64, 11.73 (2 s, 1H, NH, D₂O exchangeable); Anal. Calcd for C₂₅H₂₅BrN₄O: C, 62.90; H, 5.28; N, 11.74. Found: C, 62.69; H, 5.33; N, 11.81.

Single Crystal X-Ray Crystallography The characterization of compounds **8a** and **11** were also conducted by a single crystal X-ray structural analysis. The structures were solved with direct method and refined by SHELXTL.^{42,43} The measurements were carried out on a Bruker SMART APEX IID8 Venture diffractometer. Crystallographic data of compounds **8a** and **11** is deposited with the Cambridge Crystallographic Data Center with deposition number CCDC 1430062 and 1431357, respectively. The crystallographic structures, crystal packing, crystallographic data, refinement, selected geometric parameters and hydrogen-bond geometry of both compounds are illustrated in the Supplementary Materials.

Antimycobacterial Activity *M. tuberculosis* (RCMB 010126) strain was provided from culture collection of the RCMB, Al-Azhar University, Cairo, Egypt. Isoniazide and pyrazinamide were used as reference drugs. Antimycobacterial activity of the synthesized compounds was evaluated using the microplate Alamar blue assay (MABA)³⁶⁾ which was performed in black, clear-bottomed, 96-well microplates (in order to minimize background effect). Outer perimeter wells were filled with sterile water to prevent dehydration in experimental wells. Initial compounds dilutions were prepared in dimethyl sulfoxide and subsequent twofold dilutions were performed in the microplates. 0.1 mL of $10^5 \text{ CFU/mL} M.$ tuberculosis in-

oculum was added to wells, additional control wells consisted of bacteria only (B). Plates were incubated at 37°C. Starting at day 4 of incubation, 20μ L of alamarBlue solution (Alamar Biosciences/Accumed, Westlake, OH, U.S.A.) and 12.5μ L l of 20% Tween 80 were added added to the entire plate. Plates were then incubated at 37°C, and results were recorded at 24h post-reagent addition at 590nm. Percent inhibition was defined as: 1–(mean of test well/mean of B wells)×100. Visual MICs were defined as the lowest concentration of drug that prevented a color change.

Antimicrobial Activity All strains were provided from culture collection of the RCMB, Al-Azhar University, Cairo, Egypt. Antibacterial and antifungal activities were expressed as the diameter of inhibition zones; Agar well diffusion method was used as described earlier.^{44,45} Holes (1 cm diameter) were digger in the agar using sterile cork borer in sterile malt agar plates for fungi and sterile nutrient agar plates for bacteria, which had previously been uniformly seeded with tested microorganisms. The holes were filled by fungal filtrates (100 μ L). Plates were left in a cooled incubator at 4°C for one hour for diffusion and then incubated at 37°C for tested bacteria and 28°C for tested fungi. Inhibition zones developed due to active antimicrobial metabolites were measured after 24h of incubation for bacteria and 48h of incubation for fungi. Amphotericin B and ciprofloxacin were used as antifungal and antibacterial positive control, respectively. The experiment was performed in triplicate and the average zone of inhibition was calculated.

MIC

MIC was performed by a serial dilution technique described by Irobi *et al.*,⁴⁶⁾ starting with 100-mmol concentration of all compounds dissolved in 1 mL DMSO and then reduced by successive twofold dilutions of stock solution using a calibrated micropipette. Amphotericin B and ciprofloxacin were used as the reference compounds for fungi and bacteria; respectively. The final solutions concentrations were 125, 62.50, 31.25, 15.63, 7.81, 3.90, 1.95, 0.98, 0.49, 0.24 and 0.12μ mol/mL. The microtiter plates were incubated at 37°C for tested bacteria and 28°C for tested fungi and were readied using microplate reader after 24h for bacteria and after 48h for fungi. In each case, triplicate tests were performed and the average was taken as final reading. MIC was expressed as the lowest concentration inhibiting test organism's growth.

Cytotoxicity against WI-38 WI-38 cells (normal breast cells), were obtained from American Type Culture Collection. The cells were propagated in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% heat-inactivated fetal bovine serum (FBS) (Hyclone), 10 ug/mL of insulin (Sigma), and 1% penicillin–streptomycin. All of the other chemicals and reagents were from Sigma, or Invitrogen. Cytotoxicity was determined using MTT assay following a reported procedure.⁴⁷⁾ The IC₅₀ was estimated from graphic plots of the dose response curve for each conc. using Graphpad Prism software (San Diego, CA, U.S.A.). The data presented are the mean of at least three separate experiments.

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Conflict of Interest The authors declare no conflict of interest.

Supplementary Materials The online version of this article contains supplementary materials.

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