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Nucleophilic difluoromethylation and difluoromethylenation of aldehydes and ketones using diethyl difluoromethylphosphonate

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ABSTRACT

New methodology for difluoromethylation and difluoromethylenation of aldehydes and ketones based on nucleophilic fluorination using diethyl difluoromethylphosphonate (1) was developed. © 2008 Elsevier Ltd. All rights reserved.

1. Introduction

There is a growing interest in the development of new methodologies for the introduction of fluorine containing groups such as CF₃, CF₂H, and CF₂ into organic compounds due to their potential use in material science, pharmaceutical, and agrochemical industries.¹ Trifluoromethylations have been extensively studied and many efficient and selective methods are now available² (e.g., nucleophilic trifluoromethylation with TMSCF₃).³

In contrast to trifluoromethylation, methods for the analogous direct difluoromethylation and difluoromethylenation are just emerging. A number of geminal difluoromethylated compounds display unique biological activities such as enzyme inhibition⁴ and therefore they are valued targets in rational drug design.⁵ New synthetic methods for the convenient and efficient preparation of geminal difluoromethylated molecules will expedite their development.

Radical⁶ or nucleophilic⁷ difluoromethylation has been achieved using organoselenium precursors. Sulfur containing reagents⁸ were developed for electrophilic,⁹ radical,¹⁰ or nucleophilic difluoromethylation.¹¹ Difluoromethyl phenyl sulfone (PhSO₂CF₂H) can act as not only [CF₂H]⁻ but also a [CF₂]²⁻ equivalent.¹² PhSCF₂TMS served as a difluoromethylene radical anion synthon [CF₂]^{-.13} Fluorinated phosphonates have been recognized as non-hydrolyzable mimics of phosphates in numerous applications.¹⁴ Diethyl difluoromethylphosphonate (**1**) has been used for nucleophilic introduction of difluoromethylphosphonate moiety by a base deprotonation and reaction with various electrophiles including carbonyl compounds,^{15,16} imines,¹⁷ carboxylic acid esters,^{18,19} and chlorides,²⁰ unsaturated compounds²¹ and triflates.²²

Recently, in a preliminary study, we have identified the phosphonate **1** as a source of $[CF_2H]^-$ and $[CF_2]^{2-}$ moieties.²³ In this report, we have utilized phosphonate **1** as a versatile synthon for nucleophilic difluoromethylation and difluoromethylenation of aldehydes and ketones and disclose optimized reaction conditions for these transformations. The products of these reactions are difluoromethyl containing alcohols and 2,2-difluoro-1,3-diols. The later compounds have been previously prepared by nucleophilic difluoromethylation using difluoromethyl phenyl sulfone¹² or sulfoxide²⁴ giving mostly *anti* diols and electrophilic fluorination of 1,3-dicarbonyl compounds followed by reduction.²⁵ Asymmetric reduction with moderate enantioselectivity was also achieved.²⁶

2. Results and discussion

Phosphonates **3** were prepared as a versatile intermediate using a modified procedure starting from diethyl difluoromethylphosphonate (1)²⁷ and carbonyl compounds **2** in the presence of LDA as a base (Table 1). The reaction is efficient and gives good yields for a range of aldehydes and ketones. With





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Table 1

Preparation of phosphonates ${\bf 3}$ from diethyl difluoromethylphosphonate $({\bf 1})$ with carbonyl compounds ${\bf 2}^a$



Entry		R ¹	R ²	<i>t</i> (h)	Product	Yield (%)
1	2a	Ph	Н	0.7	3a	93
2	2b	4-ClC ₆ H ₄	Н	0.7	3b	84
3	2c	4-Me ₂ NC ₆ H ₄	Н	4	3c	70
4	2d	(E)-PhCH=CH	Н	0.7	3d	78
5	2e	$Ph(CH_2)_2$	Н	2	3e	81
6	2f	1-Naphthyl	Н	0.7	3f	93
7	2g	Me	Me	2	3g	80
8	2h	Ph	Me	2	3h	76
9	2i	Ph	Ph	6	3i	60
10	2j	-(CH ₂) ₅ -		4	3j	89
11	2k	-(CH ₂) ₃ CH=CH-		4	3k	91

^a Reaction conditions: **1** (1 equiv), LDA (1.05 equiv), THF, -74 °C, 30 min; **2** (1.3 equiv), 0.7-6 h; 0 °C, 10 min.

benzophenone (**2i**) the product **3i** was isolated in only 60% yield (entry 9); under the basic reaction conditions some product was transformed to the corresponding phosphate **4i** (Scheme 1).



Phosphonates **3** rearrange to phosphates **4** under various basic conditions (Scheme 1).¹⁵ Thus, **3a** reacts with potassium carbonate at room temperature, sodium dimsyl or *t*-BuOK at $-50 \,^{\circ}$ C (using DMF as a solvent in all cases) to give phosphate **4a**.²³ Hydrolysis of **4a** to difluoromethyl alcohol **5a** could be performed using LiOH·H₂O in refluxing ethanol, but the conversion to **5a** reaches only 20–30%. A more efficient procedure is the methanolysis with MeONa in methanol,²⁹ which furnishes the product **5a** in quantitative yield. Pleasingly, the two-step reaction can be carried out in one pot using simple reaction conditions: an excess of MeONa in methanol at room temperature to slightly elevated temperature in THF as a solvent. Our results are highlighted in Table 2. Isolated yields of difluoromethylated alcohols **5** were excellent except for **5k** (entry 7). In this case mostly formation of phosphate **4k** was observed (increasing the amount of base was not helpful).

Table 2

Preparation of difluoromethyl alcohols 5 from phosphonates 3^a

HO_CF ₂ PO(OEt) ₂	MeONa, MeOH	HO_CHF2
$R^{1} R^{2}$	THF	$R^1 R^2$
3		5

Entry		R^1 R^2		MeONa (equiv)	$T(^{\circ}C)$	<i>t</i> (h)	Product	Yield (%)
1	3a	Ph	Н	1.5	rt	1	5a	83
2	3b	4-ClC ₆ H ₄	Н	1.5	rt	1	5b	96
3	3e	$Ph(CH_2)_2$	Н	2	30	2	5e	90
4	3f	1-Naphthyl	Н	2	30	1	5f	95
5	3h	Ph	Me	4	40	1	5h	82
6	3i	Ph	Ph	2	rt	1	5i	86
7	3k	-(CH ₂) ₃ CH=CH-		2	rt	3	5k	35

 $^{\rm a}\,$ Reaction conditions: **3** (1 equiv), MeONa (1.5–4 equiv) in MeOH (4 M solution), THF, rt–40 $^\circ$ C, 1–3 h.

Table 3

Optimization of the reaction conditions for the preparation of difluoromethyl phosphate 6a from phosphonate $3a^a$



Entry	Base (equiv)	Solvent	T (°C)	t (min)	Product conversion (%)		
					6a	4a	
1	TMS ₂ NLi (2)	DMF	-60	30	0	0	
2	MeONa (4)	DMF	-50	30	0	75	
3	NaH (4)	DMF	-50	30	12	77	
4	t-BuOK (4)	THF	-60	60	20	0	
5	t-BuOK (4)	DMF	-10	10	36	0	
6	t-BuOK (4)	DMF	-50	10	95	0	
7	t-BuOK (4)	DMF	-65	10	92	0	
8	t-BuOK (2)	DMF	-65	10	86	0	
9	t-BuOK (2)	DMF	-65	40	100	0	
10	t-BuOK (1)	DMF	-65	40	52	0	

^a Reaction conditions: **3a** (1 equiv), PhCHO (2 equiv), base.

The presented methodology provides a simple and practical route toward nucleophilic introduction of difluoromethyl group into aldehydes and ketones. In the process the $P-CF_2$ bond of phosphonate **3** is cleaved and the generated difluoromethyl anion is quenched with the protic solvent to give the difluoromethyl functionality.

Next, we examined the reaction in a non-protic solvent with the aim of synthetically using the generated difluoromethyl anion for reaction with a second electrophile. Therefore, we turned our attention to a reaction of phosphonate **3a** with benzaldehyde in the presence of a base. We found that while in the presence of sodium methoxide or sodium hydride the main product is phosphate **4a**, the presence of *t*-BuOK in DMF led to the formation of the difluoromethylene containing phosphate **6a**. The results of the optimization of the type and amount of base, reaction time, and temperature are presented in Table 3.

Quantitative formation of the desired product **6a** was achieved using 2 equiv of *t*-BuOK in DMF at $-65 \,^{\circ}$ C (entry 9). Under these conditions, the product **6a** was formed as a 1:1 mixture of isomers *syn* and *anti*. *syn* and *anti* isomers of **6a** were identified by NMR (e.g., in ¹⁹F NMR of *syn* two signals can be observed: ddd and dd, while *anti* shows one distorted triplet). Additionally methanolysis of *anti*-**6a** gives known *anti*-**7a** (vide infra). The observed low diastereoselectivity in the formation of **6a** is in sharp contrast to the observation made by one of us in a difluoromethylenation using difluoromethyl phenyl sulfone, where the product 2,2-



Scheme 2.

Ĩ		HO $CF_2PO(OEt)_2$ $R^1 R^2$ + $R^3CHO \xrightarrow{t-BuOK, DMF}$					$ \begin{array}{c} POO(OEt)_2 \ (EtO)_2 OPO \ OH \\ R^2 & R^3 & R^3 & R^2 \\ F & F & F \end{array} $	$\xrightarrow{\text{MeOH}}_{i, 1 \text{ h}} R^3 \xrightarrow{\text{OH}}_{F \text{ F}} R^2$		
		3		2		6	6'		7	
Entry		R ¹	\mathbb{R}^2		R ³	<i>t</i> (h)	Phosphate yield (%) ^a 6 + 6 '	6:6 ′	Diol yield (%) ^b 7	Diol yield (%) ^c 7
1	3a	Ph	Н	2a	Ph	0.7	46 (anti)+46 (syn)	_	80 (anti); 80 (syn)	45 (anti)+44 (syn)
2	3b	4-ClC ₆ H ₄	Н	2b	4-ClC ₆ H ₄	0.7	43 (anti)+42 (syn)	_	79 (anti); 95 (syn)	45 (anti)+46 (syn)
3	3b	4-ClC ₆ H ₄	Н	21	4-MeOC ₆ H ₄	2	83 (anti+syn)	8:1	81 (anti+syn)	81 (anti+syn)
4	3g	Me	Me	2b	4-ClC ₆ H ₄	2	42	10:1	80	_
5	3j	-(CH ₂) ₅ -		2b	4-ClC ₆ H ₄	2	58	2.45:1	83	_

^a Reaction conditions: **3** (1 equiv), *t*-BuOK (2 equiv), **2** (2 equiv), DMF, -55 °C to -65 °C, 0.7-2 h.

^b Yield of diols 7 prepared from phosphates 6. Reaction conditions: 6 (1 equiv), MeONa (1 equiv) in MeOH (4 M solution), THF, rt, 1 h.

^c Yield of diols **7** prepared from phosphonates **3** using one pot procedure.

Preparation of phosphates 6 and diols 7 from phosphonates 3

Table 4

difluoro-1,3-diphenylpropane-1,3-diol (**7a**) was obtained as mostly *anti* isomer with a d.r. of 97:3.¹² In this special case the high d.r. was interpreted by a charge–charge repulsion effect between the reactive dianion species **8** and developing charge during the product (diolate) formation. In our case, the reaction probably proceeds through the anion **10** formed from the intermediate **9** (Scheme 2).

Using the optimized conditions, we have synthesized phosphates 6 on preparative scale starting from different phosphonates **3** and carbonyl compounds **2**. The results are presented in Table 4. Phosphonate **3a** reacted with benzaldehvde (**2a**) in the presence of *t*-BuOK in DMF to give a mixture of phosphates **6a** in total yield of 92%. The syn and anti isomers of 6a were easily separated using flash chromatography and each isomer was converted to diol 7a in very good yields. The removal of the phosphate moiety from 6 deserves some comment. Aqueous alkaline conditions (e.g., 5 equiv LiOH \cdot H₂O, in ethanol) gave a maximum conversion of 30% (60 °C, 2 h) and various modifications of the conditions (amount of base, temperature, time) did not improve the result. The use of a diluted methanol solution of sodium methoxide (0.1-10 equiv) afforded maximum of 50% conversion as determined using GC-MS. Finally, we found that the amount of the protic solvent should be kept to minimum. The quantitative formation of diol 7a was achieved employing 4 M methanolic solution of MeONa (1 equiv) in THF.

This methodology allowed the preparation of other phosphates 6 and diols 7 in very good yields by the reaction of phosphonates derived from aldehydes with aromatic aldehydes (Table 4, entries 1–3). However, the presence of strong base precluded reaction with enolizable aldehydes such as 2e or *n*-heptanal. The reaction of phosphonate derived from ketones (such as **3g** or **3i**) with aldehydes was possible, but gave only moderate yields of phosphates 6 (entries 4 and 5) probably due to steric bulk of the nucleophilic difluoromethyl anion. In all cases, the crude reaction mixture contained phosphates 6 as a mixture of diastereomers in equal amounts. In the case of non-symmetrical products (entries 3-5) some rearrangement of the phosphate group was observed giving rise to isomeric mixture of phosphates 6 and 6' (on methanolysis both isomers give the same diol compound). Employing ketones as electrophiles did not lead to any products 6 but only to phosphates 4 (e.g., 3a with acetone, acetophenone, or benzophenone, 3b with trifluoromethyl phenyl ketone, 3c with cyclohexanone, 3g with acetophenone).

Results presented in Table 4 show that the isolation of intermediate phosphates **6** is not necessary. Diols **7** were obtained from phosphonates **3** with better yields using one pot procedure. This modification represents real improvement because it eliminates one isolation step.²⁸

3. Conclusions

In summary, we have utilized diethyl difluoromethylphosphonate (1) for $[CF_2H]^-$ and $[CF_2]^{2-}$ transfer in nucleophilic reactions with carbonyl compounds. A simple one step procedure was developed for the preparation of difluoromethyl alcohols **5** from phosphonates **3**. The same starting compounds can be also used for the preparation of gem-difluoro-1,3-diols **7** either with the isolation of intermediate phosphates **6** or in one pot reaction.

4. Experimental section

4.1. General

NMR spectra $({}^{1}H, {}^{13}C, {}^{19}F, {}^{31}P{}^{1}H)$ were measured on a 400 MHz or 500 MHz instrument using CDCl₃ or CD₃OD as solvents. The chemical shifts (δ) are reported in parts per million (ppm) relative to Me₄Si (0 ppm, for ¹H NMR), residual CHCl₃ peak (7.26 ppm for ¹H NMR and 77.0 ppm for ¹³C NMR), residual CD₂HOD peak (3.31 ppm for ¹H NMR and 49.05 ppm for ¹³C NMR), internal CFCl₃ (0 ppm for ¹⁹F NMR), and external H₃PO₄ in water (0 ppm for ³¹P NMR). GC-MS spectra were recorded using a gas chromatograph coupled with a quadrupole mass selective electron impact (EI) detector. High resolution mass spectra (HRMS) were recorded using electrospray (ESI) or electron impact (EI) ionization. Infrared spectra were measured on a FTIR instrument as a KBr discs, chloroform solutions or films. Solvents (DMF, THF, and methanol) were dried using activated molecular sieves (3 Å). Reactions were done under atmosphere of dry argon. All chemicals were used as received. The product purifications were performed by column chromatography using silica gel 60. Yields refer to isolated material judged to be >95% pure by ¹H NMR spectroscopy and GC-MS analysis.

4.2. General procedure for the preparation of phosphonates 3

A solution of *n*-BuLi (8.6 mL, 1.5 M, 12.84 mmol) in cyclohexane was added dropwise to a stirred solution of *i*-Pr₂NH (1.80 mL, 12.84 mmol) in THF (30 mL) cooled to 0 °C. The mixture was stirred at 0 °C for 20 min and then cooled to -74 °C. A solution of diethyl difluoromethylphosphonate (1) (2.3 g, 12.23 mmol) in THF (8 mL) was added dropwise, followed by stirring at this temperature for 30 min. A solution of carbonyl compound **2** (15.9 mmol) in THF (5 mL) was added, the mixture was stirred for appropriate time at -74 °C before being warmed up to 0 °C over 10 min. Saturated aqueous solution of NH₄Cl (20 mL) was added, the product was extracted into ethyl acetate or diethyl ether (4×40 mL). The combined organic extract was washed with saturated NaCl solution

(10 mL), dried (MgSO₄), filtered, concentrated under reduced pressure, and purified on silica gel to yield phosphonate 3.

4.2.1. Diethyl 1,1-difluoro-2-hydroxy-2-phenylethylphosphonate $(3a)^{15a}$

White solid (3.35 g, 93%). Chromatography 35-50% EtOAc/hexanes; R_f (40% EtOAc/hexanes) 0.25; mp 74–75 °C. ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 1.30 (\text{dt}, 3\text{H}, {}^3J_{\text{HH}} = 7.1 \text{ Hz}, {}^4J_{\text{HP}} = 0.5 \text{ Hz}, CH_3), 1.32$ (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =0.4 Hz, CH₃), 4.15–4.28 (m, 5H, 2×CH₂, OH), 5.08–5.16 (m, 1H, CH), 7.33–7.40 (m, 3H, Ph), 7.47–7.51 (m, 2H, Ph); 13 C NMR (100.6 MHz, CDCl₃) δ 16.1–16.2 (m), 64.8–65.1 (m), 73.0-73.6 (m), 114.3-121.7 (m), 128.0, 128.1, 128.7, 134.9-135.0 (m); ¹⁹F NMR (470 MHz, CDCl₃) δ –125.5 (ddd, 1F, ²J_{FF}=304.9 Hz, $^{2}J_{FP}$ =104.9 Hz, $^{3}J_{FH}$ =20.3 Hz), -114.9 (ddd, $^{2}J_{FF}$ =304.9 Hz, $^{2}J_{FP}$ = 100.2 Hz, ${}^{3}J_{FH}$ =6.4 Hz); ${}^{31}P$ NMR (202 MHz, CDCl₃) δ 6.95 (dd, $^{2}J_{\text{PF}}$ =104.9, 100.2 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3336 (s), 1493 (w), 1254 (s), 1172 (m), 1103 (m), 1068 (s), 1015 (s), 730 (m), 701 (m); GC-MS (EI) *m*/*z* 77 (20%), 107 (20), 109 (20), 132 (100), 133 (20), 140 (20), 160 (30), 161 (70), 188 (80), 221 (10), 294 (M⁺, 15); HRMS (ESI⁺) calculated for C₁₂H₁₈F₂O₄P: 295.0911, found: 295.0918.

4.2.2. Diethyl 2-(4-chlorophenyl)-1,1-difluoro-2hydroxyethylphosphonate (**3b**)

White solid (1.29 g, 84%). Chromatography 35-50% EtOAc/hexanes; *R*_f (40% EtOAc/hexanes) 0.40; mp 98–100 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.30-1.38 (m, 6H, 2×CH₃), 4.17-4.31 (m, 4H, 2×CH₂), 5.05-5.12 (m, 1H, CH), 7.34-7.37 (m, 2H, C_{Ar}H), 7.41-7.43 (m, 2H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.2–16.3 (m), 65.0– 65.3 (m), 72.5-73.1 (m), 114.0-121.4 (m), 128.3, 129.5, 133.3-133.4 (m), 134.7; ¹⁹F NMR (470 MHz, CDCl₃) δ –126.0 (ddd, 1F, ²J_{FF}=305.2 Hz, ²J_{FF}=104.8 Hz, ³J_{FH}=20.2 Hz), –115.0 (ddd, ${}^{2}J_{FF}$ =305.2 Hz, ${}^{2}J_{FP}$ =104.8 Hz, ${}^{3}J_{FH}$ =20.2 Hz), -115.0 (ddd, ${}^{2}J_{FF}$ =305.2 Hz, ${}^{2}J_{FP}$ =99.0 Hz, ${}^{3}J_{FH}$ =5.6 Hz); ${}^{31}P$ NMR (202 MHz, CDCl₃) δ 7.23 (dd, ²*J*_{PF}=104.8, 99.0 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3315 (s), 1492 (m), 1250 (m), 1174 (m), 1048 (s), 1020 (s), 783 (m); GC-MS (EI) *m*/*z* 77 (20%), 132 (100), 160 (30), 161 (70), 188 (85), 255 (10), 328 (M⁺, 10); HRMS (ESI⁺) calculated for C₁₂H₁₇F₂O₄PCl: 329.0521, found: 329.0530.

4.2.3. Diethyl 2-(4-(dimethylamino)phenyl)-1,1-difluoro-2hydroxyethylphosphonate $(3c)^{29}$

Yellow solid (470 mg, 70%). Chromatography 30-60% EtOAc/ hexanes; R_f (40% EtOAc/hexanes) 0.12; mp 92–94 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.31–1.35 (m, 6H, 2×CH₃), 2.95 (s, 6H, NMe₂), 3.61 (br s, 1H, OH), 4.18-4.29 (m, 4H, 2×CH₂), 4.98-5.04 (m, 1H, CH), 6.72 (d, 2H, ${}^{3}J_{HH}$ =8.7 Hz, C_{Ar}H), 7.33 (d, 2H, ${}^{3}J_{HH}$ =8.7 Hz, $C_{Ar}H$); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.2–16.3 (m), 40.5, 64.6– 64.9 (m), 73.0-73.6 (m), 112.0, 114.4-122.3 (m), 129.0, 150.9; ¹⁹F NMR (470 MHz, CDCl₃) δ –125.6 (ddd, 1F, ²*J*_{FF}=303.4 Hz, ²*J*_{FF}=106.2 Hz, ³*J*_{FH}=20.2 Hz), –115.2 (ddd, ²*J*_{FF}=303.4 Hz, ²*J*_{FF}=100.9 Hz, ³*J*_{FH}=6.2 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 7.90 (dd, ²*J*_{FF}=106.2, 100.9 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3336 (s), 2808 (w), 1613 (s), 1523 (s), 1257 (s), 1178 (m), 1165 (m), 1084 (m), 1057 (s), 1025 (s), 786 (m); GC-MS (EI) m/z 150 (100%), 182 (15), 183 (15), 337 (M⁺, 10); HRMS (ESI⁺) calculated for C₁₄H₂₃F₂NO₄P: 338.1333, found: 338.1342.

4.2.4. (E)-Diethyl 1,1-difluoro-2-hydroxy-4-phenylbut-3enylphosphonate (3d)^{16b}

Colorless oil (475 mg, 78%). Chromatography 30–50% EtOAc/ hexanes; R_f (30% EtOAc/hexanes) 0.30. ¹H NMR (400 MHz, CDCl₃) δ 1.32–1.40 (m, 6H, 2×CH₃), 3.47 (br s, 1H, OH), 4.22–4.35 (m, 4H, 2×CH₂), 4.64–4.74 (m, 1H, CHOH), 6.29 (dd, 1H, ³J_{HH}=16.0, 6.2 Hz, =CHCH), 6.83 (dd, 1H, ³*J*_{HH}=16.0 Hz, ⁴*J*_{HH}=1.0 Hz, PhCH=), 7.25-7.29 (m, 1H, Ph), 7.31–7.35 (m, 2H, Ph), 7.41–7.44 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.3–16.4 (m), 65.0 (d, ²J_{CP}=6.8 Hz), 72.6-73.2 (m), 114.6-119.9 (m), 122.1 (m), 126.8, 128.2, 128.6, 135.0,

136.0; ¹⁹F NMR (470 MHz, CDCl₃) δ –123.6 (ddd, 1F, ²J_{FF}=304.6 Hz, ${}^{2}J_{FP}$ =102.0 Hz, ${}^{3}J_{FH}$ =16.3 Hz), -116.5 (ddd, 1F, ${}^{2}J_{FF}$ =304.6 Hz, ${}^{2}J_{FP}$ =102.0 Hz, ${}^{3}J_{FH}$ =8.2 Hz); 31 P NMR (202 MHz, CDCl₃) δ 7.25 (t, ${}^{2}J_{PF}$ =102.0 Hz); GC-MS (EI) m/z 55 (18%), 77 (20), 103 (16), 115 (42), 133 (100), 146 (15), 161 (83), 188 (83), 320 (M⁺, 15); HRMS (ESI⁺) calculated for C14H20F2O4P: 321.1067, found: 321.1065.

4.2.5. Diethyl 1,1-difluoro-2-hydroxy-4-phenylbutylphosphonate (**3e**)³⁰

Colorless oil (561 mg, 81%). Chromatography 20-40% EtOAc/ hexanes; R_f (40% EtOAc/hexanes) 0.16. ¹H NMR (400 MHz, CDCl₃) δ 1.37 (t, 6H, ³*I*_{HH}=7.1 Hz, 2×CH₃), 1.94–2.04 (m, 3H, PhCH₂, OH), 2.68–2.76 (m, 1H, CH_aH_bCH), 2.91–2.98 (m, 1H, CH_aH_bCH), 3.90–4.01 (m, 1H, CH), 4.23–4.33 (m, 4H, 2×CH₂), 7.18–7.23 (m, 3H, Ph), 7.27– 7.31 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.3 (d, ³ J_{CP} =5.6 Hz), 30.3, 31.2, 64.8-64.9 (m), 70.4-71.0 (m), 115.4-122.7 (m), 126.0, 128.4, 128.5, 141.2; ¹⁹F NMR (470 MHz, CDCl₃) δ –125.9 (ddd, 1F, ${}^{2}J_{FF}$ =304.1 Hz, ${}^{2}J_{FP}$ =105.2 Hz, ${}^{3}J_{FH}$ =18.3 Hz), -117.7 (ddd, ${}^{2}J_{FF}$ =304.1 Hz, ${}^{2}J_{FP}$ =101.4 Hz, ${}^{3}J_{FH}$ =7.2 Hz); ${}^{31}P$ NMR (202 MHz, CDCl₃) δ 7.59 (dd, ²*J*_{PF}=105.2, 101.4 Hz); FTIR (film, ν_{max} cm⁻¹) 3368 (s), 1604 (m), 1497 (m), 1254 (s), 1165 (s), 1057 (s), 1020 (s), 752 (m), 701 (m); GC-MS (EI) m/z 65 (30%), 91 (100), 111 (40), 132 (50), 138 (90), 161 (50), 166 (60), 188 (30), 198 (30), 284 (20), 322 (M⁺, 50); HRMS (ESI⁺) calculated for C₁₄H₂₂F₂O₄P: 323.1224, found: 323.1234.

4.2.6. Diethyl 1,1-difluoro-2-hydroxy-2-(naphthalen-1yl)ethylphosphonate (**3f**)

White solid (604 mg, 93%). Chromatography 20-40% EtOAc/ hexanes; *R_f* (30% EtOAc/hexanes) 0.20; mp 110–112 °C. ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 1.31 (\text{dt}, 3\text{H}, {}^{3}\text{J}_{\text{HH}} = 7.1 \text{ Hz}, {}^{4}\text{J}_{\text{HP}} = 0.6 \text{ Hz}, \text{CH}_3), 1.34$ (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =0.6 Hz, CH₃), 4.20–4.32 (m, 4H, 2×CH₂), 6.04-6.11 (m, 1H, CH), 7.47-7.56 (m, 3H, C_{Ar}H), 7.86-7.89 (m, 3H, C_{Ar}H), 8.07–8.09 (m, 1H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.2– 16.3 (m), 64.8-65.2 (m), 68.7-69.4 (m), 114.8-122.2 (m), 123.5, 125.1, 125.4, 126.2, 126.9, 128.7, 129.3, 131.0-131.1 (m), 131.7, 133.5; ¹⁹F NMR (470 MHz, CDCl₃) δ –126.1 (ddd, 1F, ²*J*_{FF}=305.6 Hz, ${}^{2}J_{\text{FP}}=106.3 \text{ Hz}, {}^{3}J_{\text{FH}}=21.4 \text{ Hz}), -113.5 \text{ (dd, } {}^{2}J_{\text{FF}}=305.6 \text{ Hz},$ $^{2}J_{FP}$ =99.8 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 7.69 (dd, $^{2}J_{PF}$ =106.3, 99.8 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3318 (s), 1598 (w), 1513 (w), 1252 (m), 1177 (m), 1063 (s), 1029 (s), 799 (m), 789 (m); GC–MS (EI) *m/z* 127 (35%), 126 (40), 129 (50), 131 (75), 157 (40), 160 (30), 161 (70), 170 (30), 188 (100), 189 (20), 190 (30), 344 (M⁺, 40); HRMS (ESI⁺) calculated for C₁₆H₂₀F₂O₄P: 345.1067, found: 345.1074.

4.2.7. Diethyl 1,1-difluoro-2-hydroxy-2-methylpropylphosphonate (3g)

Colorless oil (374 mg, 80%). Chromatography 25-50% EtOAc/ hexanes; R_f (30% EtOAc/hexanes) 0.35. ¹H NMR (400 MHz, CDCl₃) δ 1.39 (m, 12H, 4×CH₃), 4.26–4.34 (m, 4H, 2×CH₂); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.3 (d, ³*J*_{CP}=5.7 Hz, CH₂CH₃), 23.2–23.2 (m, Me_2C), 64.9 (d, ² J_{CP} =7.1 Hz, CH₂), 72.9 (dt, ² J_{CF} =22.3 Hz, ${}^{2}J_{CP}$ =14.4 Hz, COH), 120.0 (dt, ${}^{1}J_{CF}$ =270.8 Hz, ${}^{1}J_{CP}$ =201.0 Hz, CF₂); ${}^{19}F$ NMR (470 MHz, CDCl₃) δ –119.7 (d, ²J_{FP}=105.2 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 7.84 (t, ²J_{PF}=105.2 Hz); FTIR (film, ν_{max} cm⁻¹) 3392 (s), 1261 (s), 1165 (m), 1059 (s), 1024 (s); GC-MS (EI) m/z 43 (12%), 65 (15), 81 (15), 132 (100), 161 (65), 175 (28), 188 (45); HRMS (ESI⁺) calculated for C₈H₁₈F₂O₄P: 247.0911, found: 247.0918.

4.2.8. Diethyl 1,1-difluoro-2-hydroxy-2-phenylpropylphosphonate (**3h**)

Colorless oil (443 mg, 76%). Chromatography 20-40% EtOAc/ hexanes; R_f (30% EtOAc/hexanes) 0.33; ¹H NMR (400 MHz, CDCl₃) δ 1.04 (dt, 3H, ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=0.7 Hz, *CH*₃CH₂), 1.37 (dt, 3H, ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=0.7 Hz, CH₃CH₂), 1.71–1.72 (m, 3H, CH₃C), 3.53– 3.63 (m, 1H, CH_aH_b), 3.80–3.89 (m, 1H, CH_aH_b), 4.23–4.32 (m, 2H, 2×CH_aH_b), 4.50 (br s, 1H, OH), 7.29–7.39 (m, 3H, Ph), 7.57–7.59 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.9 (d, ³*J*_{CP}=5.9 Hz, CH₃CH₂), 16.3 (d, ³*J*_{CP}=5.6 Hz, CH₃CH₂), 23.1 (CH₃C), 64.4 (d, ²*J*_{CP}=7.1 Hz, CH₂), 65.1 (d, ²*J*_{CP}=7.1 Hz, CH₂), 75.9 (dt, ²*J*_{CF}=21.6 Hz, ²*J*_{CP}=14.3 Hz, COH), 115.3-122.8 (m), 126.6, 127.9, 140.1; ¹⁹F NMR (470 MHz, CDCl₃) δ -118.0 (dd, 1F, ²*J*_{FF}=301.9 Hz, ²*J*_{FP}=106.7 Hz), -117.1 (dd, 1F, ²*J*_{FF}=301.9 Hz, ²*J*_{FF}=100.4 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 7.80 (dd, ²*J*_{PF}=106.7, 100.4 Hz); FTIR (CHCl₃, ν_{max} cm⁻¹) 3598 (m), 3436 (s), 1603 (w), 1496 (m), 1449 (m), 1395 (m), 1249 (s), 1164 (m), 1132 (s), 1076 (s), 1061 (s), 1031 (s); GC-MS (EI) *m*/*z* 43 (35%), 77 (20), 105 (15), 121 (45), 132 (100), 161 (90), 188 (95), 308 (M⁺, 10); HRMS (ESI⁺) calculated for C₁₃H₂₀F₂O₄P: 309.1067, found: 309.1073.

4.2.9. Diethyl 1,1-difluoro-2-hydroxy-2,2-diphenylethyl-phosphonate (**3i**)

Colorless oil (719 mg, 60%). Chromatography 17–20% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.50; ¹H NMR (400 MHz, CDCl₃) δ 1.30–1.39 (m, 6H, 2×CH₃), 3.91–4.11 (m, 4H, 2×CH₂), 5.33 (br s, 1H, OH), 7.46–7.54 (m, 6H, Ph), 7.81–7.83 (m, 4H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.1 (d, ${}^{3}J_{CP}$ =5.8 Hz, CH₃), 64.8 (d, ${}^{2}J_{CP}$ =7.0 Hz, CH₂), 79.0 (dt, ${}^{2}J_{CF}$ =20.6 Hz, ${}^{2}J_{CP}$ =15.1 Hz, COH), 119.2 (dt, ${}^{1}J_{CF}$ =276.2 Hz, ${}^{1}J_{CP}$ =200.7 Hz, CF₂), 127.5, 127.8, 127.9, 139.5; ¹⁹F NMR (470 MHz, CDCl₃) δ –111.7 (d, ${}^{2}J_{FP}$ =102.8 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 8.24 (t, ${}^{2}J_{PF}$ =102.8 Hz); FTIR (film, ν_{max} cm⁻¹) 3380 (s), 1495 (m), 1450 (m), 1245 (s), 1163 (s), 1152 (s), 1072 (s), 1056 (s), 1031 (s), 725 (m), 699 (m); GC–MS (EI) *m*/*z* 77 (20%), 105 (40), 132 (30), 165 (30), 183 (100), 216 (30); HRMS (ESI⁺) calculated for C₁₈H₂₁F₂O₄PNa: 393.1043, found: 393.1040.

4.2.10. Diethyl difluoro(1-hydroxycyclohexyl)-methylphosphonate $(\mathbf{3j})^{29}$

White solid (1.032 g, 89%). Chromatography 15–40% EtOAc/hexanes; R_f (40% EtOAc/hexanes) 0.31; mp 50–52 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.37 (t, 6H, ${}^3J_{HH}$ =7.1 Hz, 2×CH₂CH₃), 1.49–1.67 (m, 8H, 4×CH₂), 1.84–1.87 (m, 2H, CH₂), 2.82 (br s, 1H, OH), 4.24–4.32 (m, 4H, 2×CH₂CH₃); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.3, 16.3, 20.4, 25.3, 29.6–29.7 (m), 64.8, 64.8, 73.9 (dt, ${}^2J_{CF}$ =21.0 Hz, ${}^2J_{CP}$ =13.5 Hz, COH), 120.3 (dt, ${}^1J_{CF}$ =270.9 Hz, ${}^1J_{CP}$ =200.4 Hz, CF₂); ¹⁹F NMR (470 MHz, CDCl₃) δ –121.8 (d, ${}^2J_{FP}$ =106.2 Hz); ³¹P NMR (202 MHz, CDCl₃) δ 8.03 (t, ${}^2J_{FF}$ =106.2 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3356 (s), 1251 (s), 1159 (m), 1056 (s), 1019 (s); GC–MS (EI) *m*/*z* 81 (30%), 109 (20), 132 (100), 133 (25), 160 (30), 161 (80), 188 (100), 215 (10), 243 (15), 266 (M⁺–F, 15); HRMS (ESI⁺) calculated for C₁₁H₂₂F₂O₄P: 287.1224, found: 287.1230.

4.2.11. Diethyl difluoro(1-hydroxycyclohex-2-enyl)methyl-phosphonate (**3k**)

Colorless oil (1.09 g, 91%). Chromatography 40–60% Et₂O/hexanes; R_f (60% Et₂O/hexanes) 0.33. ¹H NMR (400 MHz, CDCl₃) δ 1.36–1.40 (m, 6H, 2×CH₃), 1.73–2.13 (m, 6H, 3×CH₂), 3.17 (br s, 1H, OH), 4.25–4.35 (m, 4H, 2×CH₂CH₃), 5.88–5.91 (m, 1H, CH=CH), 6.09–6.13 (m, 1H, CH=CH); ¹³C NMR (100.6 MHz, CDCl₃) δ 16.3, 16.3, 17.4, 24.9, 29.4, 64.7–65.0 (m), 72.0 (dt, ²J_{CF}=21.9 Hz, ²J_{CP}=14.4 Hz, COH), 120.1 (dt, ¹J_{CF}=271.0 Hz, ¹J_{CP}=201.7 Hz, CF₂), 123.9–124.0 (m), 135.1; ¹⁹F NMR (470 MHz, CDCl₃) δ –119.8 (dd, 1F, ²J_{FF}=302.8 Hz, ²J_{FP}=105.3 Hz), –119.1 (dd, 1F, ²J_{FF}=302.8 Hz, ²J_{FP}=105.3 Hz), –119.1 (dd, 1F, ²J_{FF}=302.8 Hz, ²J_{FP}=105.3, 103.9 Hz); FTIR (film, ν_{max} cm⁻¹) 3364 (s), 1652 (w), 1254 (s), 1162 (m), 1048 (s), 1017 (s), 736 (m); GC–MS (EI) *m*/*z* 79 (20%), 97 (80), 109 (20), 132 (100), 133 (25), 160 (35), 161 (80), 188 (90); HRMS (ESI⁺) calculated for C₁₁H₂₀F₂O₄P: 285.1067, found: 285.1062.

4.3. General procedure for the preparation of difluoromethyl alcohols 5

A solution of sodium methoxide (0.19 mL, 4 M, 0.75 mmol) in methanol was added to a solution of phosphonate **3** (0.5 mmol) in

THF (8 mL). The mixture was stirred at rt to 40 °C for 1–3 h followed by the addition of saturated solution of NH₄Cl (10 mL). The product was extracted into ethyl acetate or dichloromethane (4×15 mL). The combined organic extract was washed with saturated NaCl solution (6 mL), dried (MgSO₄), filtered, concentrated under reduced pressure, and purified on silica gel to yield difluoromethyl alcohol **5**.

4.3.1. 2,2-Difluoro-1-phenylethanol (5a)^{11c}

Colorless oil (66 mg, 83%). Chromatography 15% Et₂O/hexanes; $R_f(20\% \text{ EtOAc/hexanes}) 0.31$; ¹H NMR (400 MHz, CDCl₃) $\delta 2.52$ (br s, 1H, OH), 4.81 (ddd, 1H, ³ J_{HF} =10.8, 9.5 Hz, ³ J_{HH} =4.7 Hz, CH), 5.77 (dt, 1H, ² J_{HF} =55.8 Hz, ³ J_{HH} =4.7 Hz, CF₂H), 7.38–7.43 (m, 5H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 73.6 (t, ² J_{CF} =24.5 Hz, CH), 115.8 (t, ¹ J_{CF} =245.4 Hz, CF₂), 127.1, 128.6, 129.0, 135.8 (t, ³ J_{CF} =3.2 Hz, C_{Ar}CH); ¹⁹F NMR (470 MHz, CDCl₃) δ –128.4 (ddd, 1F, ² J_{FF} =284.1 Hz, ² J_{FH} =55.8 Hz, ³ J_{FH} =10.8 Hz), –127.7 (ddd, 1F, ² J_{FF} =284.1 Hz, ² J_{FH} =55.8 Hz, ³ J_{FH} =9.5 Hz); FTIR (film, ν_{max} cm⁻¹) 3401 (s), 1607 (w), 1497 (m), 1456 (m), 1139 (m), 1121 (m), 1071 (s), 1028 (m), 701 (s); GC–MS (EI) m/z 51 (20%), 77 (50), 79 (70), 107 (100), 158 (M⁺, 20); HRMS (EI) calculated for C₈H₈F₂O: 158.0543, found: 158.0548.

4.3.2. 2,2-Difluoro-1-(4-chlorophenyl)ethanol (5b)

Colorless oil (123 mg, 96%). Chromatography 10–30% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.76. ¹H NMR (400 MHz, CDCl₃) δ 2.50 (br s, 1H, OH), 4.82 (ddd, 1H, ${}^3J_{\rm HF}$ =10.5, 9.5 Hz, ${}^3J_{\rm HH}$ =4.7 Hz, CH), 5.73 (dt, 1H, ${}^2J_{\rm HF}$ =56.0 Hz, ${}^3J_{\rm HH}$ =4.7 Hz, CF₂H), 7.33–7.43 (m, 4H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 73.0 (t, ${}^2J_{\rm CF}$ =24.7 Hz, CH), 115.5 (t, ${}^1J_{\rm CF}$ =245.7 Hz, CF₂), 128.4, 128.9, 134.1 (t, ${}^3J_{\rm CF}$ =3.2 Hz, C_{Ar}CH), 134.9. ¹⁹F NMR (470 MHz, CDCl₃) δ –128.5 (ddd, 1F, ${}^2J_{\rm FF}$ =285.2 Hz, ${}^2J_{\rm FH}$ =56.0 Hz, ${}^3J_{\rm FH}$ =9.5 Hz), –127.8 (ddd, 1F, ${}^2J_{\rm FF}$ =285.2, ${}^2J_{\rm FH}$ =56.0 Hz, ${}^3J_{\rm FH}$ =10.5 Hz); FTIR (CHCl₃, $\nu_{\rm max}$ cm⁻¹) 3609 (s), 1602 (m), 1496 (s), 1090 (s), 1079 (s), 1054 (s); GC–MS (EI) m/z 51 (15%), 77 (85), 113 (25), 141 (100), 192 (M⁺, 15); HRMS (EI) calculated for C₈H₆ClF₂O: 191.0070, found: 191.0076.

4.3.3. 1,1-Difluoro-4-phenylbutan-2-ol (5e)^{11f}

Colorless oil (62 mg, 90%). Chromatography 10–15% EtOAc/ hexanes; R_f (40% EtOAc/hexanes) 0.79. ¹H NMR (400 MHz, CDCl₃) δ 1.81–2.00 (m, 2H, PhCH₂), 2.10 (d, 1H, ³J_{HH}=5.1 Hz, OH), 2.72–2.79 (m, 1H, CH_aH_bCH), 2.88–2.95 (m, 1H, CH_aH_bCH), 3.69–3.80 (m, 1H, CHOH), 5.63 (dt, 1H, ²J_{HF}=56.1 Hz, ³J_{HH}=4.2 Hz, CF₂H), 7.21–7.24 (m, 3H, Ph), 7.30–7.34 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 30.9, 31.4 (t, ³J_{CF}=3.2 Hz, CH₂CH), 70.2 (t, ²J_{CF}=23.3 Hz, CH), 116.3 (t, ¹J_{CF}=243.7 Hz, CF₂), 126.2, 128.4, 128.5, 140.9; ¹⁹F NMR (470 MHz, CDCl₃) δ –130.1 (dd, ²J_{FH}=56.1 Hz, ³J_{FH}=10.7 Hz); FTIR (film, ν_{max} cm⁻¹) 3401 (s), 1604 (m), 1497 (m), 1455 (m), 1139 (s), 1063 (s), 751 (s), 700 (s); GC–MS (EI) *m*/*z* 51 (10%), 65 (15), 77 (15), 91 (100), 92 (40), 105 (25), 117 (60), 168 (10), 186 (M⁺, 40); HRMS (EI) calculated for C₁₀H₁₂F₂O: 186.0856, found: 186.0852.

4.3.4. 2,2-Difluoro-1-(naphthalen-1-yl)ethanol (**5f**)

Colorless oil (89 mg, 95%). Chromatography 15% EtOAc/hexanes; R_f (20% EtOAc/hexanes) 0.47. ¹H NMR (400 MHz, CDCl₃) δ 2.87 (d, 1H, ³ $J_{\rm HH}$ =4.0 Hz, OH), 5.57–5.64 (m, 1H, CH), 6.02 (ddd, 1H, ² $J_{\rm HF}$ =56.4, 55.1 Hz, ³ $J_{\rm HH}$ =4.6 Hz, CF₂H), 7.49–7.59 (m, 3H, C_{Ar}H), 7.70–7.71 (m, 1H, C_{Ar}H), 7.86–7.93 (m, 2H, C_{Ar}H), 8.05–8.07 (m, 2H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 70.4 (dd, ² $J_{\rm CF}$ =26.1, 23.2 Hz, CH), 115.8 (dd, ¹ $J_{\rm CF}$ =246.4, 245.3 Hz, CF₂), 122.8, 125.1, 125.2, 125.8, 126.6, 128.9, 129.5, 130.9, 131.6–131.7 (m), 133.7; ¹⁹F NMR (470 MHz, CDCl₃) δ –127.5 (ddd, 1F, ² $J_{\rm FF}$ =282.7 Hz, ² $J_{\rm FH}$ =56.4 Hz, ³ $J_{\rm FH}$ =12.5 Hz), –125.8 (ddd, 1F, ² $J_{\rm FF}$ =282.7 Hz, ² $J_{\rm FH}$ =55.1 Hz, ³ $J_{\rm FH}$ =7.2 Hz); FTIR (film, $\nu_{\rm max}$ cm⁻¹) 3401 (s), 1599 (m), 1513 (m), 1172 (m), 1139 (m), 1118 (m), 1068 (s), 779 (s); GC–MS (EI) m/z 127 (40%), 128 (55), 129 (100), 157 (90), 208 (M⁺, 50); HRMS (EI) calculated for C₁₂ H_{10} F₂O: 208.0699, found: 208.0703.

4.3.5. 1,1-Difluoro-2-phenylpropan-2-ol (5h)

Colorless oil (120 mg, 82%). Chromatography 20–40% EtOAc/ hexanes; R_f (30% EtOAc/hexanes) 0.40. ¹H NMR (400 MHz, CDCl₃) δ 1.67 (t, 3H, ³*J*=1.6 Hz, CH₃), 2.27 (br s, 1H, OH), 5.73 (dd, 1H, $^2J_{HF}$ =56.9, 56.0 Hz, CF₂H), 7.32–7.42 (m, 3H, Ph), 7.51–7.53 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 22.3 (t, ³*J*_{CF}=2.5 Hz, CH₃), 74.2 (t, $^2J_{CF}$ =22.0 Hz, COH), 116.9 (t, ¹*J*_{CF}=249.2 Hz, CF₂), 125.7, 128.1, 128.4, 140.2–140.3 (m); ¹⁹F NMR (470 MHz, CDCl₃) δ –131.1 (dd, $^2J_{FF}$ =276.3 Hz, ²*J*_{FH}=56.9 Hz), –130.1 (dd, ²*J*_{FF}=276.3 Hz, $^2J_{FH}$ =56.0 Hz); FTIR (CHCl₃, ν_{max} cm⁻¹) 3598 (s), 3471 (m), 1604 (w), 1497 (m), 1449 (s), 1387 (m), 1098 (s), 1074 (s), 1052 (s), 1029 (s); GC–MS (EI) *m/z* 43 (80%), 51 (20), 77 (22), 109 (15), 121 (100); HRMS (EI) calculated for C₉H₁₀F₂O: 172.0700, found: 172.0697.

4.3.6. 2,2-Difluoro-1,1-diphenylethanol (5i)³¹

Colorless oil (131 mg, 86%). Chromatography 10% EtOAc/hexanes; $R_f(20\%$ EtOAc/hexanes) 0.68. ¹H NMR (400 MHz, CDCl₃) δ 2.88 (br s, 1H, OH), 6.26 (t, 1H, ² J_{HF} =55.1 Hz, CF₂H), 7.34–7.42 (m, 6H, Ph), 7.49–7.52 (m, 4H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 78.0 (t, ² J_{CF} =21.1 Hz, COH), 116.8 (t, ¹ J_{CF} =250.3 Hz, CF₂), 127.0, 128.2, 128.3, 140.4; ¹⁹F NMR (470 MHz, CDCl₃) δ –128.1 (d, ² J_{FH} =55.1 Hz); FTIR (film, ν_{max} cm⁻¹) 3551 (s), 3468 (s), 1601 (m), 1495 (s), 1450 (s), 1073 (s), 1049 (s), 760 (s), 721 (s); GC–MS (EI) *m*/*z* 51 (10%), 77 (40), 105 (80), 183 (100), 184 (14); HRMS (ESI⁺) calculated for C₁₄H₁₂F₂ONa: 257.0754, found: 257.0745.

4.3.7. 1-(Difluoromethyl)cyclohex-2-enol (5k)

Colorless liquid (89 mg, 35%). Chromatography 20% Et₂O/hexanes; R_f (20% Et₂O/hexanes) 0.25. ¹H NMR (400 MHz, CDCl₃) δ 1.68–1.79 (m, 4H, 2×CH₂), 1.96–2.14 (m, 3H, CH₂, OH), 5.56 (t, 1H, ²J_{HF}=56.3 Hz, CF₂H), 5.64–5.67 (m, 1H, CH=CH), 6.09–6.13 (m, 1H, CH=CH); ¹³C NMR (100.6 MHz, CDCl₃) δ 17.4, 25.0, 29.2 (t, ⁵J_{CF}=2.3 Hz, CH₂CH=), 69.8 (t, ²J_{CF}=21.2 Hz, COH), 117.0 (t, ¹J_{CF}=247.5 Hz, CF₂), 124.0 (t, ³J_{CF}=3.5 Hz), 135.3; ¹⁹F NMR (470 MHz, CDCl₃) δ –133.3 (dd, ²J_{FF}=278.3 Hz, ²J_{FH}=56.3 Hz), –132.6 (dd, ²J_{FF}=278.3 Hz, ²J_{FH}=56.3 Hz); FTIR (film, ν_{max} cm⁻¹) 3592 (m), 3401 (s), 3033 (m), 1652 (m), 1070 (s), 989 (m), 734 (m); GC-MS (EI) *m*/*z* 41 (10%), 51 (15), 79 (20), 97 (100); HRMS (EI) calculated for C₇H₁₀F₂O: 148.0700, found: 148.0707.

4.4. General procedure for the preparation of phosphates 6

A solution of *t*-BuOK (224.4 mg, 2 mmol) in DMF (3 mL) was added dropwise to a stirred solution of phosphonate **3** (1 mmol) and carbonyl compound **2** (2 mmol) in DMF (3 mL) cooled to -65 °C. The mixture was stirred at -55 °C to -65 °C for appropriate time and then aqueous solution of NH₄Cl (8 mL) was added, the product was extracted into ethyl acetate or dichloromethane (3×15 mL). The combined organic extract was washed with saturated NaCl solution (7 mL), dried (MgSO₄), filtered, concentrated under reduced pressure, and purified on silica gel to yield phosphate **6**.

4.4.1. anti-2,2-Difluoro-3-hydroxy-1,3-diphenylpropyl diethyl phosphate (anti-**6a**)

White gummy solid (184 mg, 46%). Chromatography 20–70% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.20. ¹H NMR (400 MHz, CDCl₃) δ 1.12 (dt, 3H, ³ J_{HH} =7.1 Hz, ⁴ J_{HP} =1.1 Hz, CH₃), 1.40 (dt, 3H, ³ J_{HH} =7.1 Hz, ⁴ J_{HP} =1.1 Hz, CH₃), 3.82–4.02 (m, 2H, CH₂), 4.12–4.30 (m, 2H, CH₂), 5.16–5.23 (m, 1H, CH), 5.30 (d, 1H, ³ J_{HH} =5.9 Hz, OH), 5.76–5.85 (m, 1H, CH), 7.30–7.40 (m, 6H, Ph), 7.46–7.51 (m, 4H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.7 (d, ³ J_{CP} =6.7 Hz, CH₃), 16.0 (d, ³ J_{CP} =6.7 Hz, CH₃), 65.0 (d, ² J_{CP} =13.1 Hz, CH₂), 65.1 (d, ² J_{CP} =13.1 Hz, CH₂), 70.8 (dd, ² J_{CF} =28.7, 26.6 Hz, CHOH), 75.0–75.7 (m, CH), 119.1 (dt, ¹ J_{CF} =254.0 Hz, ³ J_{CP} =7.4 Hz, CF₂), 127.9, 128.0, 128.1, 128.2, 128.3, 129.3, 132.5 (d, ³ J_{CF} =3.2 Hz, CCH), 135.5; ¹⁹F NMR (470 MHz, CDCl₃)

 δ – 122.4 to – 122.3 (m); ³¹P NMR (162 MHz, CDCl₃) δ 0.24 (s); FTIR (film, ν_{max} cm⁻¹) 3339 (s), 1496 (m), 1456 (m), 1254 (s), 1166 (s), 1038 (s), 1025 (s), 745 (m), 718 (m), 699 (s); GC–MS (EI) *m*/*z* 77 (10%), 105 (10), 140 (100), 141 (10), 274 (15); HRMS (ESI⁺) calculated for C₁₉H₂₄F₂O₅P: 401.1329, found: 401.1320.

4.4.2. syn-2,2-Difluoro-3-hydroxy-1,3-diphenylpropyl diethyl phosphate (syn-**6a**)

White solid (185 mg, 46%). Chromatography 20-70% EtOAc/ hexanes; R_f (30% EtOAc/hexanes) 0.10; mp 74–76 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.03 (dt, 3H, ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=1.1 Hz, CH₃), 1.23 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.1 Hz, *CH*₃), 3.66–3.81 (m, 2H, *CH*₂), 3.95–4.12 (m, 2H, *CH*₂), 4.35 (d, 1H, ${}^{3}J_{HH}$ =5.2 Hz, *OH*), 4.47–4.54 (m, 1H, CHOH), 5.96-6.04 (m, 1H, CHOP), 7.29-7.31 (m, 3H, Ph), 7.34-7.38 (m, 2H, Ph), 7.41–7.44 (m, 3H, Ph), 7.64–7.66 (m, 2H, Ph); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.6 (d, ³J_{CP}=7.2 Hz, CH₃), 15.8 (d, ${}^{3}J_{CP}=7.2$ Hz, CH₃), 63.8 (d, ${}^{2}J_{CP}=5.8$ Hz, CH₂), 64.1 (d, ${}^{2}J_{CP}=5.8$ Hz, CH₂), 72.3 (dd, ²J_{CF}=30.3, 23.7 Hz, CHOH), 77.7–78.2 (m, CH), 119.7 (ddd, ¹*J*_{CF}=257.0, 250.9 Hz, ³*J*_{CP}=7.0 Hz, CF₂), 127.9, 128.0, 128.4, 128.5, 128.6, 129.4, 133.8–133.9 (m), 136.3; ¹⁹F NMR (470 MHz, CDCl₃) δ –123.9 (ddd, 1F, ²*J*_{FF}=250.7 Hz, ³*J*_{FH}=19.0, 6.3 Hz), –122.9 $(dd, {}^{2}J_{FF}=250.7 \text{ Hz}, {}^{3}J_{FH}=16.9 \text{ Hz}); {}^{31}P \text{ NMR} (162 \text{ MHz}, \text{ CDCl}_{3})$ $\delta - 1.56$ (s); FTIR (KBr, ν_{max} cm⁻¹) 3314 (s), 1495 (w), 1246 (m), 1163 (m), 1045 (s), 1021 (m), 721 (m), 701 (m); GC-MS (EI) m/z 77 (10%), 79 (10), 105 (10), 108 (10), 140 (100), 141 (10), 155 (15), 226 (10), 274 (15); HRMS (ESI⁺) calculated for C₁₉H₂₄F₂O₅P: 401.1329, found: 401.1313.

4.4.3. 1,3-Bis(4-chlorophenyl)-2,2-difluoro-3-hydroxypropyl diethyl phosphate (anti-**6b**)

Colorless oil (123 mg, 43%). Chromatography CH₂Cl₂/hexanes/ EtOAc 20:20:1; *R*_f (CH₂Cl₂/hexanes/EtOAc) 0.33. ¹H NMR (400 MHz, CDCl₃) δ 1.16 (dt, 3H, ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=1.0 Hz, CH₃), 1.41 (dt, 3H, ${}^{3}J_{\text{HH}}$ =7.1 Hz, ${}^{4}J_{\text{HP}}$ =1.1 Hz, CH₃), 3.85–4.05 (m, 2H, CH₂), 4.16–4.31 (m, 2H, CH_2), 5.11–5.19 (ddd, 1H, ${}^{3}J_{HF}$ =17.9, 12.1 Hz, ${}^{3}J_{HH}$ =5.9 Hz, CHOH), 5.46 (d, 1H, ³J_{HH}=5.9 Hz, OH), 5.71–5.80 (m, 1H, CHOP), 7.31–7.43 (m, 8H, $C_{AT}H$); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.8 (d, ${}^{3}J_{CP}$ =6.6 Hz, CH₃), 16.0 (d, ${}^{3}J_{CP}$ =6.7 Hz, CH₃), 65.1 (d, ${}^{2}J_{CP}$ =6.4 Hz, CH₂), 65.3 (d, ²*J*_{CP}=6.2 Hz, *C*H₂), 70.1 (dd, ²*J*_{CF}=27.8, 27.4 Hz, *C*HOH), 74.2–74.9 (m, CH), 118.9 (dt, ¹*J*_{CF}=254.1 Hz, ³*J*_{CP}=7.2 Hz, CF₂), 128.1, 128.5, 129.3, 129.6, 131.0 (d, ³*J*_{CF}=3.3 Hz, CCH), 134.0, 134.2, 135.5; 19 F NMR (470 MHz, CDCl₃) δ –122.5 to –122.4 (m); 31 P NMR (162 MHz, CDCl₃) δ 0.43 (s); FTIR (film, ν_{max} cm⁻¹) 3328 (m), 1599 (w), 1581 (w), 1493 (m), 1256 (m), 1164 (m), 1033 (s), 1014 (s), 804 (m), 766 (m); GC-MS (EI) m/z 77 (10%), 139 (12), 141 (10), 155 (10), 174 (100), 176 (40); HRMS (ESI⁺) calculated for C₁₉H₂₁Cl₂F₂NaO₅P: 491.0364, found: 491.0363.

4.4.4. 1,3-Bis(4-chlorophenyl)-2,2-difluoro-3-hydroxypropyl diethyl phosphate (syn-**6b**)

White solid (120 mg, 42%). Chromatography 30–50% EtOAc/ hexanes; *R*_f (30% EtOAc/hexanes) 0.25; mp 109–111 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.07 (dt, 3H, ³J_{HH}=7.1 Hz, ⁴J_{HP}=1.1 Hz, CH₃), 1.27 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.1 Hz, CH₃), 3.70–3.84 (m, 2H, CH₂), 3.99-4.13 (m, 2H, CH₂), 4.44-4.49 (m, 1H, CHOH), 4.61 (br s, 1H, OH), 5.99 (ddd, 1H, ³*J*_{HF}=16.8, 6.2 Hz, ³*J*_{HP}=9.8 Hz, CHOP), 7.27–7.32 (m, 4H, C_{Ar}H), 7.40–7.43 (m, 2H, C_{Ar}H), 7.59 (d, 2H, *J*=8.4 Hz, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.7 (d, ³J_{CP}=7.0 Hz, CH₃), 15.9 (d, ${}^{3}J_{CP}$ =7.1 Hz, CH₃), 64.1 (d, ${}^{2}J_{CP}$ =6.0 Hz, CH₂), 64.5 (d, ${}^{2}J_{CP}$ =5.9 Hz, CH₂), 71.9 (dd, ²*J*_{CF}=30.8, 23.8 Hz, CH), 116.8–121.9 (m, CF₂), 128.3, 128.8, 129.4, 130.0, 132.2 (dd, ³*J*_{CF}=5.4, 2.2 Hz, CCH), 134.5, 134.7, 135.6; ¹⁹F NMR (470 MHz, CDCl₃) δ –124.1 (ddd, 1F, ²*J*_{FF}=252.0 Hz, ${}^{3}J_{FH}$ =19.7, 6.2 Hz), -122.5 (dd, 1F, ${}^{2}J_{FF}$ =252.0 Hz, ${}^{3}J_{FH}$ =16.8 Hz); ${}^{31}P$ NMR (162 MHz, CDCl₃) δ –1.64 (s); FTIR (KBr, ν_{max} cm⁻¹) 3290 (s), 1601 (w), 1581 (w), 1493 (m), 1260 (s), 1166 (m), 1049 (s), 1033 (s), 1015 (s), 802 (w), 764 (m); GC-MS (EI) m/z 77 (12%), 139 (15), 141

(13), 155 (25), 174 (100), 176 (30); HRMS (ESI⁺) calculated for $C_{19}H_{21}Cl_2F_2NaO_5P$: 491.0364, found: 491.0363.

4.4.5. 1-(4-Chlorophenyl)-2,2-difluoro-3-hydroxy-3-(4methoxyphenyl)propyl diethyl phosphate (**6bl**); 3-(4-chlorophenyl)-2,2-difluoro-3-hydroxy-1-(4-methoxyphenyl)propyl diethyl phosphate (**6'bl**)

Ratio (8:1): colorless liquid (225 mg, 83%). Chromatography 20-70% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.10. ¹H NMR (400 MHz, CDCl₃) δ 1.05 (dt, 3H, ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=1.0 Hz, *CH*₃), 1.12 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.0 Hz, CH₃), 1.25 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.0 Hz, CH₃), 1.38 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.0 Hz, CH₃), 3.71– 4.25 (m, 14H, 4×CH₂, 2×OMe), 4.44-4.51 (m, 1H), 4.56-4.59 (m, 1H), 5.10–5.17 (m, 1H, CH), 5.49 (d, 1H, ³J_{HH}=5.8 Hz, OH), 5.67–5.76 (m, 1H, CH), 5.87-5.96 (m, 1H, CH), 6.87-6.94 (m, 4H, C_{Ar}H), 7.28-7.32 (m, 6H, C_{Ar}H), 7.36–7.42 (m, 4H, C_{Ar}H), 7.53–7.56 (m, 2H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.7 (d, ³J_{CP}=7.0 Hz, CH₃), 15.7 (d, ³ J_{CP} =6.7 Hz, CH₃), 15.8 (d, ³ J_{CP} =7.0 Hz, CH₃), 16.0 (d, ³ J_{CP} =6.7 Hz, CH₃), 15.8 (d, ³ J_{CP} =7.0 Hz, CH₃), 16.0 (d, ³ J_{CP} =6.7 Hz, CH₃), 55.2, 55.2, 63.8 (d, ² J_{CP} =5.7 Hz, CH₂), 64.2, (d, ² J_{CP} =6.0 Hz, CH₂), 64.9 (d, ² J_{CP} =6.4 Hz, CH₂), 65.0 (d, ² J_{CP} =6.2 Hz, CH₂), 70.3 (dd, ²J_{CF}=28.1, 27.6 Hz, CH), 71.8 (dd, ²J_{CF}=23.7, 30.3 Hz, CH), 74.7–75.4 (m), 77.1-77.8 (m), 113.7, 113.9, 116.6-122.2 (m), 128.0, 128.1, 129.3, 129.4, 129.7, 130.0, 134.0, 134.3, 134.9, 160.3, 160.4; ¹⁹F NMR (470 MHz, CDCl₃) δ -124.0 (ddd, 1F, ²J_{FF}=251.0 Hz, ³J_{FH}=19.5, 6.0 Hz, syn), -122.8 (dd, 1F, ${}^{2}J_{FF}=251.0$ Hz, ${}^{3}J_{FH}=19.5$ Hz, syn), -122.3 (dd, 2F, ³J_{FH}=12.4, 11.5 Hz, anti); ³¹P NMR (162 MHz, CDCl₃) $\delta - 1.49$ (s, P', syn), -1.41 (s, P, syn), -0.02 (s, P', anti), 0.20 (s, P, anti); FTIR (film, *v*_{max} cm⁻¹) 3320 (s), 2840 (m), 1614 (s), 1587 (m), 1516 (s), 1492 (m), 1253 (s), 1178 (m), 1164 (m), 1031 (s), 808 (m); GC-MS (EI) m/z 77 (10%), 111 (10), 119 (10), 127 (20), 135 (40), 136 (40), 139 (20), 155 (25), 170 (25), 174 (100), 175 (10), 176 (40); HRMS (ESI⁺) calculated for C₂₀H₂₄ClF₂O₆PNa: 487.0865, found: 487.0873.

4.4.6. 4-(4-Chlorophenyl)-3,3-difluoro-4-hydroxy-2-methylbutan-2-yl diethyl phosphate (**6gb**); 1-(4-chlorophenyl)-2,2-difluoro-3hydroxy-3-methylbutyl diethyl phosphate (**6'gb**)

Ratio (10:1); colorless oil (96 mg, 42%). Chromatography 25-60% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.40. ¹H NMR (400 MHz, CDCl₃) δ 1.08 (dt, 3H', ³J_{HH}=7.1 Hz, ⁴J_{HP}=1.1 Hz, CH₃CH₂), 1.30 (dt, 3H', ³*J*_{HH}=7.1 Hz, ⁴*J*_{HP}=1.2 Hz, CH₃CH₂), 1.35 (dt, 3H, ³*J*_{HH}=7.1 Hz, ${}^{4}J_{HP}$ =1.1 Hz, CH₃CH₂), 1.36 (dt, 3H, ${}^{3}J_{HH}$ =7.1 Hz, ${}^{4}J_{HP}$ =1.1 Hz, CH₃CH₂), 1.42 (s, 3H', CH₃), 1.60 (d, 3H, J=1.2 Hz, CH₃), 1.77 (s, 3H, CH₃), 1.87 (s, 3H', CH₃), 3.19 (br s, 1H', OH), 3.76–3.86 (m, 2H', CH₂), 4.04–4.19 (m, 4H, 2H', CH₂), 4.40 (d, 1H, ³J_{HH}=5.1 Hz, OH), 5.11–5.18 (m, 1H, CH), 5.63–5.72 (m, 1H', CH), 7.31–7.44 (m, 4H, 4H', C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.7 (d, ³*J*_{CP}=6.9 Hz, CH₃CH₂), 16.0 (d, ${}^{3}J_{CP}$ =7.1 Hz, CH₃CH₂), 22.7, 23.4, 24.4, 64.1 (d, ${}^{2}J_{CP}$ =6.3 Hz, CH₂), 64.2 (d, ${}^{2}J_{CP}$ =5.9 Hz, CH₂), 64.3 (d, ${}^{2}J_{CP}$ =5.9 Hz, CH₂), 64.6 (d, ${}^{2}J_{CP}$ =5.9 Hz, CH₂), 71.8 (dd, ${}^{2}J_{CP}$ =33.5, 23.8 Hz, CH), 84.0-84.7 (m), 119.3 (ddd, ${}^{1}J_{CF}$ =260.3, 269.9 Hz, ${}^{3}J_{CP}$ =9.1 Hz, CF₂), 128.1, 128.5, 129.5 (dd, ${}^{3}J_{CP}$ =6.1 Hz, CF₂), 129.5 (dd, ${}^{1}J_{CF}$ =260.3, 269.9 Hz, ${}^{3}J_{CP}$ =9.1 Hz, CF₂), 128.1, 128.5, 129.5 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =2.5 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =2.5 (dd, ${}^{3}J_{CP}$ =1.2 (dd, ${}^{3}J_{CP}$ =2.5 (dd, ${}^{$ 129.5, 130,0, 132.7, 134.2, 135.3, 135.5 (d, $^3\!J_{CF}\!\!=\!\!1$ Hz, CCH); $^{19}\!F$ NMR (470 MHz, CDCl₃) δ –126.8 (dd, 1F, ²J_{FF}=263.9 Hz, ³J_{FH}=21.1 Hz), -123.8 (dd, 1F', ${}^{2}J_{FF}=263.9$ Hz, ${}^{3}J_{FH}=16.9$ Hz), -114.5 (d, 1F', $^{2}J_{FF}$ =263.5 Hz), -113.0 (d, 1F, $^{2}J_{FF}$ =263.9 Hz); ³¹P NMR (162 MHz, CDCl₃) δ –5.17 (s, P), –1.20 (s, P'); FTIR (CHCl₃, ν_{max} cm⁻¹) 3595 (m), 3337 (m), 1600 (w), 1494 (m), 1152 (m), 1031 (s); GC-MS (EI) m/z 77 (25%), 99 (50), 126 (33), 141 (25), 155 (100), 174 (30) (major); 59 (25%), 77 (20), 92 (20), 109 (22), 141 (30), 155 (28), 174 (100), 176 (35), 277 (20) (minor); HRMS (ESI⁺) calculated for C₁₅H₂₃F₂O₅PCl: 387.0940, found: 387.0937.

4.4.7. 1-(2-(4-Chlorophenyl)-1,1-difluoro-2-hydroxyethyl)cyclohexyl diethyl phosphate (**6jb**); 1-(4-chlorophenyl)2,2-difluoro-2-(1-hydroxycyclohexyl)ethyl diethyl phosphate (**6'jb**)

Ratio (2.45:1); colorless oil (103 mg, 58%). Chromatography 20– 40% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.25. ¹H NMR

(400 MHz, CDCl₃) δ 1.09 (dt, 3H', ³J_{HH}=6.0 Hz, ⁴J_{HP}=1.1 Hz, CH₃), 1.30 (dt, 3H', ${}^{3}J_{HH}=7.1$ Hz, ${}^{4}J_{HP}=1.1$ Hz, CH_{3}), 1.35–1.41 (m, 6H, 2×CH₃), 1.53-2.55 (m, 10H, 10H', 10×CH₂), 2.81 (s, 1H', OH), 3.76-3.86 (m, 2H', CH₂CH₃), 4.05-4.22 (m, 4H, 2H', CH₂CH₃), 5.12-5.21 (m, 2H, CH, OH), 5.64-5.72 (m, 1H', CH), 7.31-7.42 (m, 4H, 4H', C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.7–16.0 (m), 16.0–16.1 (m), 20.6-20.7 (m), 20.8-20.9 (m), 24.6, 25.2, 29.6, 30.4-30.7 (m), 64.2-64.6 (m), 71.7-72.3 (m), 74.2-74.7 (m), 75.3-75.9 (m), 87.6 (dt, ${}^{2}I_{CF}=27.3$, ${}^{2}I_{CP}=9.1$ Hz, COP), 117.0–122.1 (m), 128.0, 128.4, 129.6, 130.0, 133.0, 134.0, 135.2, 135.9; ¹⁹F NMR (470 MHz, CDCl₃) δ –125.6 (dd, 1F', ${}^{2}J_{FF}$ =264.5 Hz, ${}^{3}J_{FH}$ =16.9 Hz), -124.0 (dd, 1F, ${}^{2}J_{FF}$ =268.4 Hz, ${}^{3}J_{FH}$ =22.2 Hz), -116.0. (d, 1F', ${}^{2}J_{FF}$ =264.5 Hz), -109.5 (d, 1F, ${}^{2}J_{FF}$ =268.4 Hz); ${}^{31}P$ NMR (162 MHz, CDCl₃) δ -5.58 (s, P), -1.29 (s, P'); FTIR (film, ν_{max} cm⁻¹) 3339 (s), 2984 (m), 2939 (s), 2867 (m), 1599 (w), 1581 (w), 1493 (m), 1452 (m), 1265 (s), 1250 (s), 1090 (s), 1028 (s), 807 (m); GC-MS (EI) m/z 81 (15%), 99 (15), 141 (25), 155 (20), 174 (100), 176 (30), 149 (40), 151 (12), 277 (10), 308 (7); HRMS (ESI⁺) calculated for C₁₈H₂₆ClF₂O₅PNa: 449.1067, found: 449.1068.

4.5. General procedure for the preparation of difluoromethyl diols 7 from phosphates 6

A solution sodium methoxide (0.125 mL, 4 M, 0.5 mmol) in methanol was added to a solution of phosphate **6** (0.5 mmol) in THF (4 mL). The mixture was stirred at room temperature for 1 h and then aqueous solution of NH₄Cl (4 mL) was added, the product was extracted into ethyl acetate or dichloromethane (4×15 mL). The combined organic extract was washed with saturated NaCl solution (6 mL), dried (MgSO₄), filtered, concentrated under reduced pressure, and purified on silica gel to yield difluoromethyl diols **7**.

4.5.1. anti-2,2-Difluoro-1,3-diphenylpropane-1,3-diol (anti-7a)¹²

White solid (102 mg, 80%). Chromatography 20–60% EtOAc/ hexanes; R_f (40% EtOAc/hexanes) 0.57; mp 129–131 °C. ¹H NMR (400 MHz, CD₃OD) δ 4.88 (br s, 2H, 2×OH), 5.16 (t, 2H, ³*J*_{HF}=12.5 Hz, 2×CH), 7.27–7.35 (m, 6H, Ph), 7.43–7.46 (m, 4H, Ph); ¹³C NMR (100.6 MHz, CD₃OD) δ 72.9 (t, ²*J*_{CF}=28.1 Hz, CH), 121.9 (t, ¹*J*_{CF}=250.7 Hz, CF₂), 128.9, 129.1, 129.4, 139.2; ¹⁹F NMR (470 MHz, CD₃OD) δ –121.5 (t, ³*J*_{FH}=12.5 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3460 (s), 3318 (s), 1495 (m), 1456 (m), 1160 (m), 1153 (m), 1042 (s), 1025 (m), 745 (m), 728 (m), 695 (m); GC–MS (EI) *m*/*z* 51 (10%), 77 (30), 79 (30), 105 (15), 107 (30), 140 (100), 141 (15); HRMS (ESI⁻) calculated for C₁₅H₁₃F₂O₂: 263.0884, found: 263.0872.

4.5.2. syn-2,2-Difluoro-1,3-diphenylpropane-1,3-diol (syn-7a)

White solid (105 mg, 80%). Chromatography 20–60% EtOAc/ hexanes; R_f (40% EtOAc/hexanes) 0.41; mp 143–145 °C. ¹H NMR (400 MHz, CD₃OD) δ 4.82 (dd, 2H, ³ J_{HF} =14.1, 11.4 Hz, 2×CH), 4.87 (br s, 2H, 2×OH), 7.30–7.37 (m, 6H, Ph), 7.44–7.46 (m, 4H, Ph); ¹³C NMR (100.6 MHz, CD₃OD) δ 74.1 (dd, ² J_{CF} =26.6, 24.2 Hz, CH), 122.2 (dd, ¹ J_{CF} =255.2, 247.9 Hz, CF₂), 129.0, 129.3, 129.3, 139.2; ¹⁹F NMR (470 MHz, CD₃OD) δ –124.8 (dt, 1F, ² J_{FF} =250.9 Hz, ³ J_{FH} =14.1 Hz), –122.8 (dt, 1F, ² J_{FF} =250.9 Hz, ³ J_{FH} =11.4 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3397 (s), 1496 (m), 1457 (m), 1199 (m), 1144 (m), 1060 (s), 1034 (m), 1023 (m), 759 (m), 730 (m), 704 (m); GC–MS (EI) *m/z* 77 (25%), 79 (30), 107 (30), 140 (100), 141 (15); HRMS (ESI⁻) calculated for C₁₅H₁₃F₂O₂: 263.0884, found: 263.0896.

4.5.3. anti-1,3-Bis(4-chlorophenyl)-2,2-difluoropropane-1,3-diol (anti-**7b**)³²

White solid (69 mg, 79%). Chromatography 20–40% EtOAc/hexanes; R_f (33% EtOAc/hexanes) 0.26; mp 149–150 °C. ¹H NMR (400 MHz, CD₃OD) δ 4.88 (br s, 2H, 2×OH), 5.19 (t, 2H, ³*J*_{HF}=12.4 Hz, 2×CH), 7.32–7.34 (m, 4H, C_{Ar}H), 7.41–7.43 (m, 4H, C_{Ar}H); ¹³C NMR (100.6 MHz, CD₃OD) δ 72.0 (t, ²*J*_{CF}=28.4 Hz, CH), 121.8 (dd, ¹*J*_{CF}=250.8, 250.5 Hz, CF₂), 129.0, 130.9, 135.0, 138.1; ¹⁹F NMR

 $(470 \text{ MHz, CD}_3\text{OD}) \delta - 121.5 \text{ (t, }{}^{3}\!J_{\text{FH}} = 12.4 \text{ Hz}); \text{FTIR (KBr, }\nu_{\text{max}} \text{ cm}^{-1}) \\ 3593 \text{ (s), } 1596 \text{ (w), } 1578 \text{ (w), } 1493 \text{ (m), } 1090 \text{ (s), } 1074 \text{ (s), } 1035 \text{ (m), } \\ 1016 \text{ (m), } 795 \text{ (s), } 756 \text{ (s); } \text{GC}-\text{MS (EI) }m/z \text{ 77 } (23\%), \text{ 78 } (20), \text{ 113 } \\ (10), 141 \text{ (40), } 142 \text{ (42), } 143 \text{ (20), } 144 \text{ (18), } 174 \text{ (100), } 176 \text{ (30). } \text{HRMS } \\ \text{(ESI^-) calculated for } C_{15}\text{H}_{11}\text{Cl}_2\text{F}_2\text{O}_2\text{: } 331.0099, \text{ found } 331.0101. \\ \end{cases}$

4.5.4. syn-1,3-Bis(4-chlorophenyl)-2,2-difluoropropane-1,3-diol (syn-**7b**)

Viscous oil (81 mg, 95%). Chromatography 20–40% EtOAc/hexanes; R_f (33% EtOAc/hexanes) 0.38. ¹H NMR (400 MHz, CD₃OD) δ 4.81 (dd, 2H, ³ J_{HF} =13.8, 11.2 Hz, 2×CH), 4.88 (br s, 2H, 2×OH), 7.34–7.37 (m, 4H, C_{Ar}H), 7.43–7.45 (m, 4H, C_{Ar}H); ¹³C NMR (100.6 MHz, CDCl₃) δ 73.5 (dd, ² J_{CF} =28.8, 24.8 Hz, CH), 119.2 (dd, ¹ J_{CF} =256.6, 247.9 Hz, CF₂), 128.6, 129.3, 134.3 (d, ³ J_{CF} =2.7 Hz, CCH), 135.0; ¹⁹F NMR (470 MHz, CD₃OD) δ –124.5 (dt, 1F, ² J_{FF} =251.9 Hz, ³ J_{FH} =13.8 Hz), –122.2 (dt, 1F, ² J_{FF} =251.9 Hz, ³ J_{FH} =11.2 Hz); FTIR (film, ν_{max} cm⁻¹) 3573 (w), 3387 (m), 1598 (m), 1580 (w), 1493 (s), 1193 (m), 1092 (s), 1060 (s), 1031 (m), 1015 (s), 844 (m), 802 (m), 763 (s); GC–MS (EI) m/z 77 (50%), 113 (15), 141 (85), 143 (37), 174 (100), 176 (38). HRMS (ESI⁻) calculated for C₁₅H₁₁Cl₂F₂O₂: 331.0099, found 331.0103.

4.5.5. 1-(4-Chlorophenyl)-2,2-difluoro-3-(4-methoxyphenyl)-propane-1,3-diol (7bl)

Colorless thick oil (85 mg, 81%). Chromatography 25–55% EtOAc/hexanes; R_f (40% EtOAc/hexanes) 0.31. ¹H NMR (400 MHz, CD₃OD) δ 3.77 (s, 3H, OMe), 3.78 (s, 3H, OMe), 4.73–4.86 (m, 2H, 2×CH), 4.88 (br s, 4H, 4×OH), 5.08–5.19 (m, 2H, 2×CH), 6.86–6.92 (m, 4H, C_{Ar}H), 7.31–7.43 (m, 12H, C_{Ar}H); ¹³C NMR (100.6 MHz, CD₃OD) δ 55.7, 55.7, 71.9–72.7 (m), 73.2–74.0 (m), 114.3, 114.5, 119.4–124.7 (m), 128.9, 129.1, 130.5, 130.5, 130.9–131.0 (m), 134.9, 135.1, 138.1–138.2 (m), 161.1, 161.3; ¹⁹F NMR (470 MHz, CD₃OD) δ –124.8 (dt, 1F, ²*J*_{FF}=250.8 Hz, ³*J*_{FH}=14.0 Hz, *syn*), –122.8 (dt, 1F, ²*J*_{FF}=250.8 Hz, ³*J*_{FH}=14.0 Hz, *syn*), –122.8 (dt, 1F, ²*J*_{FF}=250.8 Hz, ³*J*_{FH}=14.0 Hz, *syn*), 1515 (s), 1493 (m), 1252 (s), 1179 (s), 1067 (s), 1034 (s), 840 (m), 802 (m); GC–MS (EI) *m/z* 77 (15%), 94 (10), 109 (10), 137 (100), 170 (10), 174 (10), 328 (5); HRMS (ESI⁻) calculated for C₁₆H₁₄F₂O₃Cl: 327.0600, found: 327.0602.

4.5.6. 1-(4-Chlorophenyl)-2,2-difluoro-3-methylbutane-1,3-diol (**7gb**)

White solid (50 mg, 80%). Chromatography 25–40% EtOAc/hexanes; R_f (30% EtOAc/hexanes) 0.45; mp 97–99 °C. ¹H NMR (400 MHz, CD₃OD) δ 1.36 (s, 3H, CH₃), 1.37 (s, 3H, CH₃), 4.88 (br s, 2H, 2×OH), 5.13 (dd, 1H, ³ J_{HF} =21.4 Hz, ³ J_{HH} =3.9 Hz, CH), 7.33–7.36 (m, 2H, C_{Ar}H), 7.43 (d, 2H, *J*=8.4 Hz, C_{Ar}H); ¹³C NMR (100.6 MHz, CD₃OD) δ 24.3, 25.1, 73.3 (dd, ² J_{CF} =33.7, 24.0 Hz, CH), 74.6 (t, ² J_{CF} =26.3 Hz, COH), 122.0 (dd, ¹ J_{CF} =259.1, 247.7 Hz, CF₂), 128.9, 131.1, 135.0, 138.6; ¹⁹F NMR (470 MHz, CD₃OD) δ –127.8 (dd, 1F, ² J_{FF} =260.6 Hz, ³ J_{FH} =21.4 Hz), –113.2 (d, 1F, ² J_{FF} =260.6 Hz); FTIR (KBr, ν_{max} cm⁻¹) 3369 (s), 3247 (s), 1598 (w), 1493 (s), 1409 (m), 1150 (m), 1088 (s), 1072 (s), 780 (s); GC–MS (EI) *m/z* 43 (20%), 60 (25), 78 (55), 92 (80), 142 (100), 174 (40); HRMS (ESI⁻) calculated for C₁₁H₁₂ClF₂O₂: 249.0494, found: 249.0497.

4.5.7. 1-(2-(4-Chlorophenyl)-1,1-difluoro-2-hydroxyethyl)cyclohexanol (**7jb**)

White solid (48 mg, 83%). Chromatography 20–40% EtOAc/hexanes; R_f (40% EtOAc/hexanes) 0.63; mp 140–142 °C. ¹H NMR (400 MHz, CDCl₃) δ 1.55–1.96 (m, 10H, 5×CH₂), 2.39 (br s, 1H, OH), 4.27 (br s, 1H, OH), 5.07–5.13 (m, 1H, CH), 7.33–7.36 (m, 2H, C_{Ar}H), 7.39–7.42 (m, 2H, C_{Ar}H); ¹³C NMR (125.7 MHz, CDCl₃) δ 20.3, 20.7, 25.1, 29.7, 30.7, 72.8 (dd, ²J_{CF}=32.9, 24.7 Hz, CH), 76.1 (dd, ²J_{CF}=26.9, 24.8 Hz, COH), 119.0 (dd, ¹J_{CF}=258.9, 250.7 Hz, CF₂), 128.2, 129.6, 134.4, 134.9; ¹⁹F NMR (470 MHz, CDCl₃) δ –135.2 to –135.1 (m, 1F),

-134.6 to -134.5 (m, 1F), -114.8 (s, 1F), -114.2 (s, 1F); FTIR (KBr, $\nu_{\rm max}$ cm $^{-1}$) 3404 (s), 3253 (s), 1598 (w), 1491 (m), 1449 (m), 1414 (m), 1079 (s), 1071 (s), 985 (m), 812 (m); GC–MS (EI) m/z 77 (40%), 81 (35), 132 (80), 141 (100), 143 (30), 172 (20), 176 (5); HRMS (ESI $^-$) calculated for C14H16F2O2CI: 289.0801, found: 289.0813.

4.6. General procedure for the preparation of difluoromethyl diols 7 from phosphonates 3 (one pot procedure)

A solution of *t*-BuOK (224.4 mg, 2 mmol) in DMF (3 mL) was added dropwise to a stirred solution of phosphonate **3** (1 mmol) and carbonyl compound **2** (2 mmol) in DMF (3 mL) cooled to -65 °C. The mixture was stirred at -55 °C to -65 °C for appropriate time and then at this temperature a solution sodium methoxide (0.25 mL, 4 M, 1 mmol) in methanol was added. The mixture was warmed up to rt and stirred for 1 h. Aqueous solution of NH₄Cl (4 mL) was added, the product was extracted into dichloromethane (4×15 mL). The combined organic extract was washed with saturated NaCl solution (6 mL), dried (MgSO₄), filtered, concentrated under reduced pressure, and purified on silica gel to yield difluoromethyl diols **7**.

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References and notes

- (a) Uneyama, K. Organofluorine Chemistry; Blackwell: Oxford, 2006; (b) Kirsch, P. Modern Fluoroorganic Chemistry; Wiley-VCH: Weinheim, 2004; (c) Chambers, R. D. Fluorine in Organic Chemistry; Blackwell: Oxford, 2004.
- (a) Ma, J.-A.; Cahard, D. J. Fluorine Chem. 2007, 128, 975; (b) Prakash, G. K. S.; Hu, J. In Science of Synthesis (Trihalomethyl Compounds); Charette, A. B., Ed.; Thieme: New York, NY, 2005; Vol. 22; (c) Ma, J.-A.; Cahard, D. Chem. Rev. 2004, 104, 6119.
- (a) Prakash, G. K. S.; Mandal, M. J. Fluorine Chem. 2001, 112, 123; (b) Singh, R. P.; Shreeve, J. M. Tetrahedron 2000, 56, 7613; (c) Prakash, G. K. S.; Yudin, A. K. Chem. Rev. 1997, 97, 757.
- (a) Müller, K.; Faeh, C.; Diederich, F. Science 2007, 317, 1881; (b) Silva, A. M.; Cachau, R. E.; Sham, H. L.; Erickson, J. W. J. Mol. Biol. 1996, 255, 321.
- 5. Tozer, M. J.; Herpin, T. F. *Tetrahedron* **1996**, *52*, 8619.
- 6. Murakami, S.; Ishii, H.; Tajima, T.; Fuchigami, T. Tetrahedron 2006, 62, 3761.
- 7. Qin, Y.-Y.; Qiu, X.-L.; Yang, Y.-Y.; Meng, W.-D.; Qing, F.-L. J. Org. Chem. 2005, 70, 9040.
- 8. Prakash, G. K. S.; Hu, J. Acc. Chem. Res. 2007, 40, 921.
- (a) Prakash, G. K. S.; Weber, C.; Chacko, S.; Olah, G. A. Org. Lett. 2007, 9, 1863; (b) Prakash, G. K. S.; Weber, C.; Chacko, S.; Olah, G. A. J. Comb. Chem. 2007, 9, 920.
 Li, Y.; Liu, J.; Zhang, L.; Zhu, L.; Hu, J. J. Org. Chem. 2007, 72, 5824.
- (a) Ni, C.; Liu, J.; Zhang, L.; Hu, J. Angew. Chem., Int. Ed. 2007, 46, 786; (b) Li, Y.;
 (a) Ni, C.; Liu, J.; Zhang, L.; Hu, J. Angew. Chem., Int. Ed. 2007, 46, 786; (b) Li, Y.;
 Hu, J. Angew. Chem., Int. Ed. 2005, 44, 5882; (c) Prakash, G. K. S.; Hu, J.; Wang,
 Y.; Olah, G. A. Eur. J. Org. Chem. 2005, 2218; (d) Ni, C.; Hu, J. Tetrahedron Lett.
 2005, 46, 8273; (e) Prakash, G. K. S.; Hu, J.; Wang, Y.; Olah, G. A. J. Fluorine
 Chem. 2005, 126, 529; (f) Prakash, G. K. S.; Wang, Y.; Hu, J.; Olah, G. A.
 J. Fluorine Chem. 2005, 126, 1361; (g) Prakash, G. K. S.; Hu, J.; Wang, Y.; Olah,
 G. A. Org. Lett. 2004, 6, 4315; (h) Prakash, G. K. S.; Hu, J.; Olah, G. A. J. Org.
 Chem. 2003, 68, 4457.
- 12. Prakash, G. K. S.; Hu, J.; Mathew, T.; Olah, G. A. Angew. Chem., Int. Ed. 2003, 42, 5216.
- 13. Li, Y.; Hu, J. Angew. Chem., Int. Ed. 2007, 46, 2489.
- (a) Romanenko, V. D.; Kukhar, V. P. Chem. Rev. 2006, 106, 3868 and references cited therein; (b) Blackburn, G. M. Chem. Ind. (London) 1981, 134; (c) McKenna, C. E.; Shen, P. D. J. Org. Chem. 1981, 46, 4573.
- (a) Obayashi, M.; Ito, E.; Matsui, K.; Kondo, K. *Tetrahedron Lett.* **1982**, *23*, 2323;
 (b) Piettre, S. R.; Cabanas, L. *Tetrahedron Lett.* **1996**, *37*, 5881; (c) Piettre, S. R.; Girol, C.; Schelcher, C. G. *Tetrahedron Lett.* **1996**, *37*, 4711.
- (a) Martin, S. F.; Dean, D. W.; Wagman, A. S. *Tetrahedron Lett.* **1992**, *33*, 1839; (b) Pajkert, R.; Koroniak, H. J. *Fluorine Chem.* **2007**, *128*, 1260.
- Röschenthaler, G.-V.; Kukhar, V.; Barten, J.; Gvozdovska, N.; Belik, M.; Sorochinsky, A. Tetrahedron Lett. 2004, 45, 6665.
- (a) Phillion, D. P.; Cleary, D. G. J. Org. Chem. **1992**, *57*, 2763; (b) Lequeux, T. P.; Percy, J. M. J. Chem. Soc., Chem. Commun. **1995**, 2111; (c) Blades, K.; Lequeux, T. P.; Percy, J. M. Tetrahedron **1997**, *53*, 10623; (d) Li, X.; Bhandari, A.; Holmes, C. P.; Szardenings, A. K. Bioorg. Med. Chem. Lett. **2004**, *14*, 4301.

- 19. (a) Bouvet, D.; O'Hagan, D. Tetrahedron 1999, 55, 10481; (b) Fox, D. T.; Poulter, C. D. J. Org. Chem. 2005, 70, 1978.
- 20. Blackburn, G. M.; Brown, D.; Martin, S. J.; Parratt, M. J. J. Chem. Soc., Perkin Trans. 1 **1987**, 181.
- 21. (a) Howson, W.; Hills, J. M. Bioorg. Med. Chem. Lett. 1991, 1, 501; (b) Lequeux, T. P.; Percy, J. M. Synlett **1995**, 361; (c) Blades, K.; Lapôtre, D.; Percy, J. M. Tetrahedron Lett. **1997**, 38, 5895; (d) Blades, K.; Percy, J. M. *Tetrahedron Lett*. **1998**, 39, 9085; (e) Murano, T.; Muroyama, S.; Yokomatsu, T.; Shibuya, S. *Synlett* **2002**, 1657.
- Berkowitz, D. B.; Eggen, M. J.; Shen, Q.; Sloss, D. G. J. Org. Chem. 1993, 58, 6174.
 Zibinsky, M.; Beier, P.; Prakash, G. K. S. Khimiya Interesakh Ustoichivogo Razvitiya
- 2008. 16. 71.
- 24. Zhu, L.; Li, Y.; Ni, C.; Hu, J.; Beier, P.; Wang, Y.; Prakash, G. K. S.; Olah, G. A. J. Fluorine Chem. **2007**, 128, 1241.
- 25. Kuroboshi, M.; Ishihara, T. Bull. Chem. Soc. Jpn. 1990, 63, 1185.
- 26. Clarke, M. L.; France, M. B.; Knight, F. R.; Frew, J. J. R.; Roff, G. J. Synlett **2007**, 1739.
- 27. Diethyl difluoromethyl phosphonate (1) was prepared in one step on ca 50 g scale from diethyl phosphite and chlorodifluoromethane in 84% isolated yield following published procedure: Bergstrom, D. E.; Shum, P. W. J. Org. Chem. 1988, 53, 3953.
- 28. We failed to perform the one pot reaction of phosphonate 1 to diols 7 under various conditions: *t*-BuOK or *i*-Pr₂NK+*t*-BuOLi in DMF (-60 °C).
- 29. Waschbuesch, R.; Samadi, M.; Savignac, P. J. Organomet. Chem. **1997**, 529, 267.
- Waschbuesch, K., Sahladi, W., Savigna, F.J. Organomez, Chem. 2007,
 Cox, R. J.; Gibson, J. S.; Martín, M. B. M. ChemBioChem 2002, 3, 874.
 Stavber, G.; Zupan, M.; Jereb, M.; Stavber, S. Org. Lett. 2004, 6, 4973.
- 32. Li, J.-L.; Liu, J.-T. Tetrahedron **2007**, 63, 898.