## Synthesis of N-1-oxypyrimidine 1,3-dioxolane and 1,3-oxathiolane nucleosides

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## Two series of 1,3-dioxolanes and 1,3-oxathiolane nucleosides containing *N*-1-oxypyrimidine were synthesized as potential antiviral agents.

The potent activity displayed by 3'-azido-3'-deoxythymidine  $(AZT)^1$  against human immuno-deficiency virus (HIV) prompted further design and evaluation of nucleoside analogues.<sup>2</sup> However, the toxicities associated with these compounds as well as the development of resistant viral strains upon prolonged treatment indicates that there is still a need for novel therapeutic agents.<sup>3</sup> One approach is to replace the carbohydrate moiety of 2',3'-dideoxynucleoside analogues with other five membered rings.<sup>3b,4</sup> It has been demonstrated that hetero-substitution of these rings has a profound effect on the biological activity of the resulting nucleoside analogue<sup>5</sup> as displayed by (-)-2'-deoxy-3'-thiacytidine (3TC<sup>®</sup>, Epivir) 1<sup>5c,6</sup> and (+)-2'-deoxy-3'-oxacytidine (Troxacitabine) 2.<sup>7</sup>



As part of an ongoing search for new antiviral leads, we further explored this class of 3'-heterosubstituted nucleosides. We synthesized a novel class of these compounds where the 1,3-oxathiolane or 1,3-dioxolane ring is linked to the heterocyclic base through a nitrogen–oxygen bond. This is exemplified by the general structure **3**. The structure of these nucleosides are analogous to those of biologically active compounds<sup>8</sup> such as 1-( $\alpha$ -D-ribofuranosyloxy)uracil **4**,<sup>8b</sup> 9-(3-hydroxypropoxy)guanine **5**<sup>8c</sup> and 9-(2,3-dihydroxypropoxy)guanine **6**.<sup>8d</sup> As an example, **6** showed more potent and selective activity than acyclovir against HSV-1, HSV-2 but was less active than acyclovir against VZV.<sup>8c,d</sup>

The synthetic route to  $(\pm)$ -1,3-dioxolane and 1,3-oxathiolane nucleoside analogues **3** is based upon reaction of *N*-1-hydroxy-heterocycles **8** with a dioxolane or oxathiolane moiety **7** bearing a suitable leaving group Y (Scheme 1).



Our strategy was to build stepwise the N-1-hydroxypyrimidine base 8 since its direct synthesis by oxidation of the base has not yet been achieved. N-1-Hydroxyuracil 12 was selected as the key base in this series.9 The preparation of this compound is described in Scheme 2. Treatment of O-benzylhydroxylamine with aqueous acidic potassium cyanate afforded urea 9 in 80% yield. This was reacted with sodium hydride followed by ethyl 3,3-dimethoxypropionate to give the protected uracil 10. This compound was hydrogenated to give N-1-hydroxyuracil 12. Similarly, N-1-hydroxythymine 13 was prepared by reacting the urea 9 with ethyl 3,3-diethoxy-2-methylpropionate followed by deprotection of the benzyl group of compound 11. These pyrimidine bases 12 and 13 were also synthesized by Klötzer using a similar approach.<sup>10</sup> N-1-Hydroxycytosine was prepared from its precursor uracil. For example, reaction of N-1-benzyloxyuracil 10 with phosphorus oxychloride, triethylamine and triazole gave the triazolo derivative which was treated with ammonia then debenzylated to give N-1-hydroxycytosine 15 in 75% yield (Scheme 2).

Unlike 1-(benzyloxy)imidazole which is unstable under alkaline conditions,<sup>8a</sup> the *N*-1-benzyloxypyrimidines **10**, **11**, **14** are stable under a variety of conditions. These include acidic and basic conditions, as well as temperatures (<90 °C) and catalytic hydrogenation. In addition, the final free *N*-1-hydroxy compounds **12**, **13**, **15** can be stored for months at 0 °C without decomposition.

Two approaches were considered for the preparation of pyrimidine and purine nucleosides. The first route was based upon coupling of a suitably functionalised *N*-1-hydroxy base with 1,3-dioxolane or 1,3-oxathiolane sugars under Mitsunobu conditions. Unfortunately, this reaction appeared to be ineffective and resulted in low yield. In fact, the Mitsunobu reaction between acylated furanose and 1-hydroxybenzotriazole gave similar results.<sup>11</sup> The second approach offers a more general route for the synthesis of these nucleosides. Scheme 3 illustrates a representative example where the sugar moieties of 1,3-dioxolane or 1,3-oxathiolane were reacted with iodo- or bromotrimethylsilane then the solution was treated with a mixture of sodium hydride and *N*-1-hydroxyuracil **12** or



Scheme 2 Reagents and conditions: (a) 10% aq. acetic acid; (b) NaH, DMSO, 70 °C, 22 h; (c) Na, EtOH, (MeO)<sub>2</sub>CHCH<sub>2</sub>COOEt; (d) 10% Pd/C, cyclohexane, EtOH, 60 °C, 5 h; (e) POCl<sub>3</sub>, Et<sub>3</sub>N, triazole; (f) NH<sub>3</sub>; (g) H<sub>2</sub>/Pd-C.

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thymidine 13 in DMF. This gave the desired nucleoside 16 and 17 or 18 and 19, respectively, as a 1:1 mixture of *cis* and *trans* isomers in 72 and 66% yields, respectively.<sup>12</sup> Replacement of halotrimethylsilane with trimethylsilyl triflate or the base sodium hydride with triethylamine did not alter the ratio of isomers but reduced the yield. Separation of the isomers 16–19 by chromatography followed by deprotection with methanolic ammonia gave the expected nucleosides 20–27 in high yields.<sup>13</sup>

Similarly, cytosine derivatives 28–31 were produced in a 1:1



mixture of *cis* and *trans* isomers under the same conditions starting from N-1-hydroxycytosine **15**. Direct conversion of uracil nucleoside **16** or **17** to the corresponding cytosine using the triazolo-phosphorus oxychloride-ammonia procedure was not successful and gave low yield of the expected product **28–31**.

The anti HIV, HBV, HSV-1 and HSV-2 activities of  $(\pm)$ -1,3-dioxolane and 1,3-oxathiolane nucleoside analogues **20–27**, **28–31**, were evaluated and compared with 3TC<sup>®</sup> (Epivir) and AZT. All of them were found to be inactive and non-toxic, except the cytosine derivative **29** which displayed weak inhibition of extracellular HBV.

In summary, described herein is a novel class of  $(\pm)$ -1,3-dioxolane and 1,3-oxathiolane nucleoside analogues. The biological results demonstrate that linking the sugar to the heterocyclic base through an oxygen causes dramatic reduction in antiviral activity in this series of compounds.

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## Notes and references

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