(2S,5R)-trans-2-Methyl-5-[(dimethylamino)methyl]-1,3oxathiolane Methiodide [(-)-10]. An excess of MeI (1 mL) was added to a solution of (-)-9 in anhydrous ether (0.5 g, 3 mmol, in 20 mL) and the mixture kept at room temperature overnight. The white solid obtained was crystallized from ethanol: yield 90%; mp 141-142 °C (lit.^{5,16} mp 152-153 °C for the racemate); $[\alpha]^{20}_{\rm D}$ -18.3° (EtOH); CD (EtOH) λ 245 nm, $\Delta\epsilon$ -0.506; NMR (Me₂SO-d₆) δ 1.53 (d, 3, 2-CH₃), 2.4-4.0 (m, 4, 4-H₂ and 5-CH₂), 3.25 (s, 9, N⁺(CH₃)₃), 4.95 (t, t, 1, 5 H), 5.48 (q, 1, 2 H).

In the same way, starting from (+)-9, (2*R*,5*S*)-trans-2methyl-5-[(dimethylamino)methyl]-1,3-oxathiolane methiodide [(+)-10] was obtained: mp 140–142 °C; $[\alpha]^{20}_{D}$ +17.7° (EtOH); CD (EtOH) λ 245 nm, $\Delta \epsilon$ +0.491. It shows the same IR and NMR spectra as the enantiomer.

2-Methyl-1,3-oxathiolane-5-carboxylic Acid (11). Racemic 5 (6 g, 40 mmol) was dissolved in 25 mL of 2 N NaOH and the resultant mixture refluxed for 2 h. Acidification of the cold solution with 2 N HCl and ether extraction gave 5 g of an oil that is suitable pure for the following reaction: IR (neat) ν 3500–2300 (OH), 1750 (CO) cm⁻¹; NMR (CDCl₃) δ 1.58 and 1.68 (d, 3, 2-CH₃, trans/cis isomers in 4:1 ratio), 3.36 (d, 2, 4-H₂), 4.56 and 4.94 (t, 1, 5 H, trans/cis isomers, $J_{\text{trans}} = 6$ and $J_{\text{cis}} = 7$ Hz), 5.46 (q, 1, 2 H). According to the NMR spectrum the pure *cis*-5 had isomerized to 4:1 mixture of *trans/cis*-11.

Pharmacology. Guinea Pig Ileum. Male guinea pigs (200–300 g) were killed by cervical dislocation. Segments of ileum 1.5–2.0 cm long were carefully removed and suspended in a 10 mL of organ bath containing a solution of the following composition (mM): NaCl, 137; NaHCO₃, 12; KCl, 2.7; MgSO₄, 1; NaHPO₄, 0.4; CaCl₂, 1.8; glucose, 5 (which was kept at 37 °C and aerated with O₂ containing 5% CO₂). Contractions were recorded isotonically at a loading tension of 1 g on an electromechanical transducer connected to a Gemini II poligraph. After equilibration for 30 min, cumulative concentration responses were obtained, taking care not to reach supramaximal concentration in order to

avoid desensitization. A full dose-response curve was then determined.

Frog Rectus Abdominis. The rectus abdominis muscle of frogs weighting 10–20 g was set up at room temperature in a 5-mL bath containing Clark frog Ringer solution of the following composition (mM): NaCl, 111; KCl, 1.88; CaCl₂, 1.08; NaHPO₄, 0.08; NaHCO₃, 2.38; glucose, 11.1 (which was aereated with oxygen as described by the Edimburgh Group).²² Contractions were recorded on a Gemini II poligraph. Cumulative dose-response curves were established after a 1-h period at which the tissue was allowed to stabilize.

Acknowledgment. We thank Cristina Bellucci and Maria Scarlattini, who is on leave from the CNR Fisiologia Clinica, Pisa, for their excellent technical assistence.

Registry No. (+)-1, 102046-04-8; (-)-1, 103066-66-6; (±)-2, 103004-06-4; (±)-2.oxalate, 103004-07-5; (+)-2.(+)-di-O,O'-p-toluyltartaric acid, 103066-63-3; (-)-2.(-)-di-O,O'-p-toluyltartaric acid, 103066-65-5; (+)-2, 103066-62-2; (-)-2, 103066-64-4; (S)-3 (S-benzyl).brucine, 103004-08-6; (R)-3 (S-benzyl).brucine, 30134-78-2; (S)-3 (S-benzyl), 30134-77-1; (R)-3 (S-benzyl), 30134-76-0; (+)-(S)-3, 30163-03-2; (-)-(R)-3, 30163-02-1; (+)-(S)-4, 103004-09-7; (-)-(R)-4, 103004-10-0; (±)-5, 103066-74-6; (+)-5, 103066-69-9; (-)-5, 103004-11-1; (+)-6, 103066-74-6; (+)-5, 103066-67-7; (+)-7, 103004-12-2; (-)-7, 103066-71-3; (+)-9, 103029-06-2; (-)-9, 103066-73-5; (+)-10, 103066-76-8; (-)-10, 103066-75-7; (±)-cis-11, 103004-14-4; (±)-trans-11, 103004-15-5; C₆H₃CH₂SCH₂CH(OH)CO₂H, 30134-75-9.

Synthesis and Dopaminergic Activity of Some Halogenated Mono- and Dihydroxylated 2-Aminotetralins

Joseph Weinstock,*[†] Dimitri E. Gaitanopoulos,[†] Hye-Ja Oh,[†] Francis R. Pfeiffer,[†] Carole B. Karash,[†] Joseph W. Venslavsky,[†] Henry M. Sarau,[‡] Kathryn E. Flaim,[‡] J. Paul Hieble,[§] and Carl Kaiser*[†]

Departments of Medicinal Chemistry, Molecular Pharmacology, and Pharmacology, Research and Development Division, Smith Kline & French Laboratories, Philadelphia, Pennsylvania 19101. Received December 27, 1985

In a series of 7,8-dihydroxy-1-phenyltetrahydro-3-benzazepine dopamine receptor agonists introduction of a chloro or fluoro substituent into the 6-position increases dopaminergic potency. Also, in this series replacement of the 7-hydroxyl group with a halogen results in inversion of activity from dopamine receptor agonist to antagonist. The present study was aimed at exploring the possibility that the structure-activity observations in the 3-benzazepine series of dopaminergic agents might be extrapolated to another class of dopamine receptor agonists, the 2-aminotetralins. Thus, a series of chloro- and fluoro-substituted mono- and dihydroxylated 2-aminotetralins was prepared and evaluated for dopaminergic properties in D-1 and D-2 receptor-related tests. Introduction of a chloro substituent into the 8-position of the prototype of this series, i.e. 2-amino-6,7-dihydroxytetralin (ADTN), resulted in a compound with a high degree of selectivity for the D-1 subpopulation of dopamine receptors; it was equally or more potent than ADTN in the D-1 receptor-related tests with greatly decreased effectiveness in the tests involving D-2 receptors. A similar effect was observed with 8-fluoro-ADTN; however, the N-(4-hydroxyphenethyl)-N-propyl derivative 4g of the 8-chloro-substituted ADTN showed marked D-2 binding affinity. Conversely, introduction of a chloro substituent into the 5-position of ADTN markedly decreased D-1 receptor affinity and efficacy. This effect was not seen with the related 5-fluoro derivative, suggesting D-1 receptors are more sensitive to bulk in the 5-position of ADTN than are the D-2 receptors. Replacement of either the 6- or 7-hydroxyl groups of ADTN with a chloro or fluoro substituent, in contrast, did not parallel the response seen in the benzazepine series (i.e., the compounds uniformly demonstrated less receptor affinity and did not have dopamine receptor antagonist activity); however, the decrease in agonist potency was less marked in the case of 2-amino-6-fluoro-7-hydroxytetralins than in the chlorinated monohydroxyaminotetralins. Thus, a parallelism in structure-activity relationships in the benzazepine and aminotetralin series of dopamine receptor agonists was not observed. The differences may reflect altered modes of receptor binding in the two series.

2-Amino-5,6-dihydroxytetralin (A-5,6-DTN, 1a)¹⁻⁵ and its 6,7-dihydroxy isomer (ADTN, 1b)²⁻¹⁰ have significant

dopamine-like activity in a number of pharmacological and biochemical test systems that measure responses resulting

⁽²²⁾ Staff of the Department of Pharmacology University of Edimburgh Pharmacological Experimental on Isolated Preparations; Churchill: Livingstone, Edinburg, London, New York, 1970.

[†]Department of Medicinal Chemistry.

[‡]Department of Molecular Pharmacology.

[§] Department of Pharmacology.

⁽¹⁾ Cannon, J. G.; Kim, J. C.; Aleem, M. A.; Long, J. P. J. Med. Chem. 1972, 15, 348.

from activation of D-1, D-2¹¹ or DA₁, DA₂¹² receptor subpopulations. The monohydroxylated 2-aminotetralins have also been the subject of extensive investigation. In various binding studies dopamine receptor affinity is in the order 7-OH (1c) > 5-OH (1d) > 6-OH (1e).¹³⁻¹⁶ The 8-monohydroxylated derivative 1f has little activity on dopamine receptors;¹³ however, its N,N-di-n-propyl derivative 1g is a potent agonist of central serotonin receptors.^{17,18}



In another series of dopamine receptor agonists, the tetrahydro-3-benzazepines, e.g. **2a**, ¹⁹⁻²⁵ introduction of a

- (2) McDermed, J. D.; McKenzie, G. M.; Phillips, A. P. J. Med. Chem. 1975, 18, 362.
- (3) Sheppard, H.; Burghardt, C. R.; Long, J. P. Res. Commun. Chem. Pathol. Pharmacol. 1978, 19, 213.
- (4) Freedman, S. B.; Templeton, W. M.; Poat, J. A.; Woodruff, G. N. Proc. Br. Pharmacol. Soc. 1981, 759P.
- (5) Horn, A. S.; Grol, C. J.; Dijkstra, D.; Mulder, A. H. J. Med. Chem. 1978, 21, 825.
- (6) Woodruff, G. N.; Elkhawad, A. O.; Pinder, R. M. Eur. J. Pharmacol. 1974, 25, 80.
- (7) Davis, A.; Roberts, P. J.; Woodruff, G. N. Br. J. Pharmacol. 1978, 63, 183.
- (8) Abel, M. S.; Clement-Cormier, Y. Pharmacologist 1978, 20, 241.
- (9) Woodruff, G. N. Trends Pharmacol. Sci. 1982, 3, 59.
- (10) Horn, A. S.; Rodgers, J. R. J. Pharm. Pharmacol. 1980, 32, 521.
- (11) Kebabian, J. W.; Calne, D. B. Nature (London) 1979, 277, 93.
- (12) Goldberg, L. I.; Kohli, J. D. Commun. Psychopharmacol. 1979, 3, 447.
- (13) Seiler, M. P.; Markstein, R. Mol. Pharmacol. 1982, 22, 281.
- (14) McDermed, J. D.; McKenzie, G. M.; Freeman, H. S. J. Med. Chem. 1976, 19, 547.
- (15) Tedesco, J. L.; Seeman, P.; McDermed, J. D. Mol. Pharmacol. 1979, 16, 369.
- (16) Feenstra, M. G. P.; Rollema, H.; Dijkstra, D.; Grol, C. J.; Horn, A. S.; Westerink, B. H. C. Naunyn-Schmiedeberg's Arch. Pharmacol. 1980, 313, 213.
- (17) Arvidsson, L.-E.; Hacksell, U.; Nilsson, J. L. G.; Hjorth, S.; Carlsson, A.; Lindberg, P.; Sanchez, D.; Wikström, H. J. Med. Chem. 1981, 24, 921.
- (18) Arvidsson, L.-E.; Hacksell, U.; Johansson, A. M.; Nilsson, J. L. G.; Lindberg, P.; Sanchez, D.; Wikström, H.; Svensson, K.; Hjorth, S.; Carlsson, A. J. Med. Chem. 1984, 27, 45.
- (19) Setler, P. E.; Sarau, H. M.; Zirkle, C. L.; Saunders, H. L. Eur. J. Pharmacol. 1978, 50, 419.
- (20) Pendleton, R. G.; Samler, L.; Kaiser, C.; Ridley, P. T. Eur. J. Pharmacol. 1978, 51, 19.
- (21) Freedman, S. B.; Wait, C. P.; Woodruff, G. N. Proc. Br. Pharmacol. Soc. 1979, 430P.



halogen, particularly a chloro, substituent into the catecholic system markedly affects dopaminergic potency. Thus, **2b** (fenoldopam, SK&F 82526)²⁶⁻²⁹ and **2c**^{24,30,31} are

1. C₆H₅CH₂NH₂

2.PtO₂, H₂ 3.Pd/C, H₂ CH₃Q

CH

BBra 4c

NHR

14a, R=CH₂C₆H₅ b, R=H

12

СН3О

сн₃о

13

- (22) Hahn, R. A.; Wardell, J. R., Jr. J. Cardiovasc. Pharmacol. 1980, 2, 583.
- (23) Goldberg, L. I.; Kohli, J. D. In *Dopamine Receptors*; Kaiser, C., Kebabian, J. W., Eds.; ACS Symposium Series 224; American Chemical Society: Washington, DC, 1983; pp 101-113.
- (24) (a) Kaiser, C.; Dandridge, P. A.; Weinstock, J.; Ackerman, D. M.; Sarau, H. M.; Setler, P. E.; Webb, R. L.; Foley, J. J.; Horodniak, J. W.; Matz, E. D. Acta Pharm. Seuc. 1983, Suppl. 2, 132. (b) Dandridge, P. A.; Kaiser, C.; Brenner, M.; Gaitanopoulos, D.; Davis, L. D.; Webb, R. L.; Foley, J. J.; Sarau, H. M. J. Med. Chem. 1984, 27, 28.
- (25) Hahn, R. A.; Wardell, J. R., Jr.; Sarau, H. M.; Ridley, P. T. J. Pharmacol. Exp. Ther. 1983, 223, 305.
- (26) Weinstock, J.; Wilson, J. W.; Ladd, D. L.; Brush, C. K.; Pfeiffer, F. R.; Kuo, G. Y.; Holden, K. G.; Yim, N. C. F.; Hahn, R. A.; Wardell, J. R., Jr.; Tobia, A. J.; Setler, P. E.; Sarau, H. M.; Ridley, P. T. J. Med. Chem. 1980, 23, 973.
- (27) Berkowitz, B. A.; Zabko-Potapovich, B.; Sherman, S.; Hieble, J. P.; Weinstock, J.; Ohlstein, E. H. Fed. Proc. 1984, 43, 743.
- (28) Ackerman, D. M.; Weinstock, J.; Wiebelhaus, V. D.; Berkowitz, B. Drug Dev. Res. 1982, 2, 283.
- (29) Stote, R. M.; Dubb, J. W.; Familiar, R. G.; Erb, B. B.; Alexander, F. Clin. Pharmacol. Ther. 1983, 34, 309.

Table I. Halogenated Mono- and Dihydroxylated 2-Aminotetralins



				4				
				$starting^{a}$				
no.	Х	\mathbb{R}^1	\mathbb{R}^2	matl.	mp, °C	yield, %	recryst solvent	formula ^b
4a	5-F, 6,7-(OH) ₂	Н	Н	11a°	269-270 dec	94	MeOH-EtOAc	C ₁₀ H ₁₂ FNO ₂ ·HBr
4b	5-Cl, 6,7-(OH) ₂	Н	н	11 b °	250 dec	98	MeOH-EtOAc	C ₁₀ H ₁₂ ClNO ₂ ·HBr
4c	8-F, $6,7-(OH)_2$	Н	Н	14 b °	225 - 229	75	MeOH-EtOAc	C ₁₀ H ₁₂ FNO ₂ ·HBr
4d	8-Cl, $6,7-(OH)_2$	H	Н	18°	253-256 dec	90	MeOH-EtOAc	C ₁₀ H ₁₂ ClNO ₂ ·HBr
4e	8-Cl, $6,7-(OH)_2$	<i>n</i> -Pr	n-Pr	19°	190-191	96	MeOH-EtOAc	C ₁₆ H ₂₄ ClNO ₂ ·HBr ^d
4f	8-Cl, $6,7-(OH)_2$	$(CH_2)_2$ Ph-4-OH	Н	20b°	240-241 dec	100	MeOH	C ₁₈ H ₂₀ ClNO ₃ ·HBr
4g	8-Cl, $6,7-(OH)_2$	$(CH_2)_2$ Ph-4-OH	n-Pr	21°	195-200	50	EtOH	C ₂₁ H ₂₆ ClNO ₃ ·HBr ^e
4 h	6-F, 7-OH	H	н	26a	264-270 dec	63	MeOH-EtOAc	C ₁₀ H ₁₂ FNO·HBr
4i	6-F, 7-OH	CH_3	CH_3	27a	237 - 241	76	MeOH-EtOH	C ₁₂ H ₁₆ FNO·HBr ^f
4j	6-F, 7-OH	n-Pr	n-Pr	27b	221-223	11	$MeOH-Et_2O^g$	C ₁₆ H ₂₄ FNO·HCl ^h
4k	6-F, 7-OH	$NR^{1}R^{2} = CH_{2}N($	$(n-\Pr)_2$	29°	198-200	70	2-PrOH-	C ₁₇ H ₂₆ FNO·HBr
							MeOH-Et ₂ O	
41	6-Cl, 7-OH	H	Н	34c	>265	56	$MeOH-Et_2O$	C ₁₀ H ₁₂ CINO•HBr
4m	6-Cl, 7-OH	CH_3	CH_3	34d	246 - 247	73	$MeOH-Et_2O$	C ₁₂ H ₁₆ ClNO·HBr
4n	6-Cl, 7-OH	n-Pr	n-Pr	34e	227	83	$MeOH-Et_2O$	C ₁₆ H ₂₄ ClNO·HBr
40	6,8-Cl ₂ , 7-OH	Н	Н	34f	>250	85	MeOH-MeCN	C ₁₀ H ₁₁ Cl ₂ NO·HBr ^e
4p	6,8-Cl ₂ , 7-OH	CH_3	CH_3	34g	>265	91	$MeOH^i$	$C_{12}H_{15}Cl_2NO \cdot HBr^{j}$
4q	6,8-Cl ₂ , 7-OH	n-Pr	n-Pr	34h	246	77	$MeOH-Et_2O$	C ₁₆ H ₂₃ Cl ₂ NO·HBr
4r ^k	5-Cl, 6-OH	H	Н	34i	>275	34	MeOH-MeCN	C ₁₀ H ₁₂ ClNO·HBr
4s	5-Cl, 6-OH	CH_3	CH_3	34j	257 - 258	69	$MeOH-Et_2O$	$C_{12}H_{16}ClNO \cdot HBr^{h}$
4t ^k	5-Cl, 6-OH	n-Pr	n-Pr	34k	230-232	85	MeOH-Et ₂ O	C ₁₆ H ₂₄ ClNO·HBr ^j
4u	7-Cl, 6-OH	Н	н	341	>265	61	MeOH-Et ₂ O	C ₁₀ H ₁₂ ClNO•HBr
4 v	7-Cl, 6-OH	CH_3	CH_3	34m	231 - 232	87	$MeOH-Et_2O$	$C_{12}H_{16}CINO \cdot HBr^{h}$
4w	7-Cl, 6-OH	n-Pr	n-Pr	34n	190-192	81	MeOH-Et ₂ O	C ₁₆ H ₂₄ ClNO•HBr
dopamine	2-(3,4-dihydroxyphenyl)ethy	lamine						
ADTN	6,7-(OH) ₂	Н	н					

^aCompounds were prepared by HBr cleavage of the indicated corresponding methoxy derivative by general method B (Experimental Section), unless indicated otherwise. ^bAll compounds analyzed satisfactorily for C, H, N. ^cPrepared by general method 2, Experimental Section. ^dAnal. for 0.75 H₂O. ^eAnal. for 0.5 H₂O. ^fAnal. for 0.4 H₂O. ^fCrude HBr salt was purified by chromatography on SiO₂ with gradient of 0.5 to 4.5% MeOH in CH₂Cl₂ to give base that was converted to HCl salt. ^hAnal. for 0.25 H₂O. ⁱCompound purified by trituration with MeOH. ^jAnalyzed satisfactorily for Cl. ^kReferences 60–62.

more potent activators of dopamine receptors than is the prototype 2a. Substitution of position 6 of 2a with fluorine to give 2d likewise results in a potent dopaminomimetic with unique pharmacological actions.³¹ Equally of interest, replacement of the 7-hydroxyl group of 2a, or even more strikingly its N-methyl derivative, with either a chloro or bromo substituent, i.e. to afford 3a (SCH 23390),³²⁻³⁹ 3b (SK&F 83509),³⁴ or 3c (SK&F 83566),⁴⁰ results in com-

- (30) Blumberg, A. L.; Hieble, J. P.; McCafferty, J.; Hahn, R. A.; Smith, J., Jr. Fed. Proc. 1982, 41, 345.
- (31) Weinstock, J.; Wilson, J. W.; Ladd, D. L.; Brenner, M.; Ackerman, D. M.; Blumberg, A. L.; Hahn, R. A.; Hieble, J. P.; Sarau, H. M.; Wiebelhaus, V. D. In *Dopamine Receptors*; Kaiser, C., Kebabian, J. W., Eds.; ACS Symp. Ser. 224; American Chemical Society: Washington, DC, 1983; pp 157-173.
- (32) Iorio, L. C.; Houser, V.; Korduba, C. A.; Leitz, F.; Barnett, A. *Pharmacologist* 1981, 23, 136.
- (33) Iorio, L. C.; Barnett, A.; Leitz, F. H.; Houser, V. P.; Korduba, C. A. J. Pharmacol. Exp. Ther. 1983, 226, 462.
- (34) Itoh, Y.; Beaulieu, M.; Kebabian, J. W. Eur. J. Pharmacol. 1984, 100, 119.
- (35) Hyttel, J. Eur. J. Pharmacol. 1983, 91, 153.
- (36) Goldberg, L. I.; Glock, B. S.; Kohli, J. D.; Barnett, A. Hypertension 1984, 6, I-25.
- (37) Hilditch, A.; Drew, G. M.; Naylor, R. J. Eur. J. Pharmacol. 1984, 97, 333.
- (38) O'Boyle, K. M.; Waddington, J. L. Eur. J. Pharmacol. 1984, 98, 433.
- (39) Mailman, R. B.; Schulz, D. W.; Lewis, M. H.; Staples, L; Rollema, H.; Dehaven, D. L. Eur. J. Pharmacol. 1984, 101, 159.
- (40) Flaim, K. E.; Gessner, G. W.; Crooke, S. T.; Sarau, H. M.; Weinstock, J. Life Sci. 1985, 36, 1427.

pounds that are the most potent and selective antagonists of the D-1, or DA_1 , subpopulation of dopamine receptors presently known.

Consideration of the potent dopaminergic activity of the mono- and dihydroxylated 2-aminotetralins, coupled with the striking effect of halogen substitution in the benzazepine class of dopamine receptor agonists, led us to study the dopaminergic effects of chloro and fluoro substitution and hydroxyl replacement in the aminotetralins. The synthesis and results of pharmacological and biochemical studies of these halogenated mono- and dihydroxylated 2-aminotetralins (4, Table I) are the subject of this paper.

Chemistry. Synthesis of a series of halogenated monoand dihydroxylated 2-aminotetralins (4, Table I) was accomplished as outlined in Schemes I-VI.

The preparation of 5-fluoro (4a) and 5-chloro (4b) derivatives of ADTN is illustrated in Scheme I. Accordingly, (2-fluoro-3,4-dimethoxyphenyl)acetic acid (5a), prepared by hydrolysis of the corresponding nitrile,⁴¹ and (2chloro-3,4-dimethoxyphenyl)acetic acid (5b)⁴² were converted to the ethyl hydrogen malonates 9a and 9b by the general procedure of Eistetter and Wolf.⁴³ These malonates cyclized to 1-oxo-1,2,3,4-tetrahydronaphthoates 10a and 10b upon treatment with polyphosphate ester (PPE). Subsequent hydrogenolysis and hydrolysis gave tetrahydronaphthoic acids 11a and 11b that were converted to 4a and 4b via a modified Curtius reaction⁴⁴ and subsequent

- (41) Ladd. D. L.; Weinstock, J. J. Org. Chem. 1981, 46, 203.
- (42) Young, T. E.; Miziantz, M. E. J. Med. Chem. 1966, 9, 635.
- (43) Eistetter, K.; Wolf, H. P. O. J. Med. Chem. 1982, 25, 109.

1618 Journal of Medicinal Chemistry, 1986, Vol. 29, No. 9

Scheme III



Table II. Halogenated Mono- and Dimethoxylated 2-Aminotetralin Hydrochlorides

no.	X	R ¹	R ²	starting compd	mp, °C	yield, %	recryst solvent	formula ^a
34a	5-F, 6,7-(CH ₃ O) ₂	Н	Н	11 a	237-239	97	MeOH-Et ₂ O	C ₁₂ H ₁₆ FNO ₂ ·HCl
34b	5-Cl, 6,7-(CH ₃ O) ₂	н	н	11b	242-244	48	MeOH–EtÕAc	C ₁₂ H ₁₆ ClNO ₂ ·HCl
34c	6-Cl, 7-CH ₃ O	н	н	33	>255	47	MeOH-MeCN	C ₁₁ H ₁₄ ClNO·HCl ^b
34d	6-Cl, 7-CH ₃ O	CH_3	CH_3	33	224 - 225	44	$MeOH-Et_2O$	C ₁₃ H ₁₈ ClNO·HCl
34e	6-Cl, 7-CH ₃ O	<i>n</i> -Pr	n - Pr	34c	240	29	MeOH-Et ₂ O	C ₁₇ H ₂₆ ClNO·HCl
34f	6,8-Cl ₂ , 7-CH ₃ O	н	Н	35b	>250	53	EtOH	C ₁₁ H ₁₃ Cl ₂ NO·HCl
34g	6,8-Cl ₂ , 7-CH ₃ O	CH_3	CH_3	34f	197 - 199	65	2-PrOH	C ₁₃ H ₁₇ Cl ₂ NO·HCl
34h	6,8-Cl ₂ , 7-CH ₃ O	n-Pr	n-Pr	34f	188-189	60	2-PrOH	C ₁₇ H ₂₅ Cl ₂ NO·HCl ^b
34i	5-Cl, 6-CH ₃ O	н	н	38a	>270	28	MeOH	C ₁₁ H ₁₄ ClNO•HCl
34j	5-Cl, 6-CH ₃ O	CH_3	CH_3	34i	259 - 260	45	MeOH-Et ₂ O	C ₁₃ H ₁₈ ClNO•HCl
34k	5-Cl, 6-CH ₃ O	n-Pr	n-Pr	34i	194-197	53	2-PrOH	C ₁₇ H ₂₆ ClNO·HCl ^c
341	7-Cl, 6-CH ₃ O	н	Н	38b	>280	25	MeOH-MeCN	C ₁₁ H ₁₄ ClNO·HCl
34m	7-Cl, 6-CH ₃ O	CH_3	CH_3	38b	>270	60	MeOH-Et ₂ O	C ₁₃ H ₁₈ ClNO·HCl
34n	7-Cl, 6-CH ₃ O	n-Pr	n-Pr	341	216 - 218	58	MeOH-Et ₂ O	C ₁₇ H ₂₆ ClNO·HCl

^a All compounds analyzed satisfactorily for C, H, N. ^b Analyzed satisfactorily for Cl. ^c Anal. for 0.25 H₂O.

 BBr_3 cleavage⁴⁵ of the methyl ethers (see Table II). 2-Amino-8-fluoro-6,7-dihydroxytetralin (4c) was derived from 5a, which was transformed to 13 via the Burckhalter-Campbell reaction,⁴⁶ followed by reductive amination and methoxyl scission as shown in Scheme II.

- (44) Nichols, D. E.; Barfkneckt, C. F.; Long, J. P.; Standridge, R. T.; Howell, H. G.; Partyka, R. A.; Dyer, D. C. J. Med. Chem. 1974, 17, 161.
- (45) For example, see: Pfeiffer, F. R.; Wilson, J. W.; Weinstock, J.; Kuo, G. Y.; Chambers, P. A.; Holden, K. G.; Hahn, R. A.; Wardell, J. R., Jr.; Tobia, A. J.; Setler, P. E.; Sarau, H. M. J. Med. Chem. 1982, 25, 352.
- (46) Burckhalter, J. H.; Campbell, J. R. J. Org. Chem. 1961, 26, 4232.

As indicated in Scheme III, Stobbe condensation of 2-chloroveratraldehyde $(15)^{47}$ with dimethyl succinate, followed by reduction and methanesulfonic acid (MSA) cyclization, gave 16, which upon sequential reduction, ester hydrolysis, and Curtius⁴⁴ reaction gave 18. Cleavage of the 6,7-dimethoxyl substituents of 18 afforded 4d. Reductive dipropylation of 18 with excess propionaldehyde followed by ether cleavage produced the N,N-di-n-propyl derivative 4e. The other N-substituted derivatives of 4d, i.e. 4f and 4g, were obtained by conventional methods as outlined. Preparation of the 2-amino-6-fluoro-7-hydroxytetralins 4h-4j, as well as the (dipropylamino)methyl homologue

⁽⁴⁷⁾ Faulkner, J. K.; Woodstock, D. J. Chem. Soc. 1962, 4737.

Scheme IV



Scheme VI



4k, was achieved as shown in Scheme IV. Thus, the butyric acid 22, obtained by Friedel-Crafts acylation of 2fluoroanisole with succinic anhydride followed by reduction of the resulting succinic acid, was cyclized^{48,49} to a single product, the tetralone 23, whose structure was established by ¹H NMR (two doublets for ortho and meta H-F coupling with the expected coupling constants of 11.2 and 8.8 Hz, respectively). The tetralone 23 was converted to the aminotetralone 25 by Neber rearrangement^{50,51} of the intermediate oximino tosylate 24b. Hydrogenolysis of 25 to 26a was best accomplished with palladium-on-carbon catalyst in an acetic acid-aqueous hydrochloric acid solution. Purification of 26a was accomplished via the tertbutyl carbamate 26b, which was reconverted to 26a with concentrated hydrobromic acid. The N,N-dimethyl 27a and N.N-dipropyl 27b derivatives of 26a were obtained by reductive alkylation.^{48,52,53} Hydrobromic acid cleavage of 26a, 27a, and 27b produced 4h-4j, respectively. Carbethoxylation of 23 afforded 28a (85%) along with 15% of 28b that resulted from ethoxide displacement of the 6fluoro substituent. Catalytic hydrogenation of the ester 28a under conventional conditions was incomplete; however, the corresponding carboxylic acid was readily hydrogenolyzed to afford 28c. Conversion of 28c to its $N_{,-}$ N-di-n-propyl carboxamide, followed by borane reduction⁵⁴ and BBr₃ cleavage of the 7-methoxy group gave the (2aminomethyl)tetralin 4k.

The synthetic route employed for derivation of 41-4qis outlined in Scheme V; it is patterned after the sequence described by Cannon et al.⁴⁸ In this fashion 2-chloro-5methylphenol was converted via 30 and 31 to 32 and 33, which by various modifications of the Curtius reaction^{44,55-59} followed by several alkylation/reduction pro-

- (48) Cannon, J. G.; Brubaker, A. N.; Long, J. P.; Flynn, J. R.; Verimer, T.; Harnirattisai, P.; Costall, B.; Naylor, R. J.; Nohria, V. J. Med. Chem. 1981, 24, 149.
- (49) Davies, J. E.; King, F. E.; Roberts, J. C. J. Chem. Soc. 1955, 2782.
- (50) Nedelec, L.; Pierdet, A.; Fauveau, P.; Euvrard, C.; Proulx-Ferland, L.; Dumont, C.; Labrie, F.; Boissier, J. R. J. Med. Chem. 1983, 26, 522.
- (51) Sprenger, W. K.; Cannon, J. G.; Barman, B. K.; Burkman, A. M. J. Med. Chem. 1969, 12, 487.
- (52) Marchini, P.; Liso, G.; Reho, A.; Liberatore, F.; Moracci, F. M. J. Org. Chem. 1975, 40, 3453.
- (53) Gribble, G. W.; Lord, P. D.; Skotnicki, J.; Dietz, S. E.; Eaton, J. T.; Johnson, J. L. J. Am. Chem. Soc. 1974, 98, 7812.
- (54) Brown, H. C.; Heim, P. J. Am. Chem. Soc. 1964, 86, 3566.
- (55) Weinstock, J. J. Org. Chem. 1961, 26, 3511.
- (56) Kaiser, C.; Weinstock, J. Org. Synth. 1971, 51, 48.
- (57) Smith, P. A. S. Org. React. 1946, 3, 337.
- (58) Kaiser, C.; Lester, B. M.; Zirkle, C. L.; Burger, A.; Davis, C. S.; Delia, T. J.; Zirngibl, L. J. Med. Pharm. Chem. 1962, 5, 1243.

cesses⁵¹⁻⁵⁴ and HBr cleavage of the appropriate methoxylated derivatives 34 (see Table II) produced 41, 4m, and 4n. The intermediate oxonaphthoic acid 32 was chlorinated to afford an intermediate dichlorinated derivative 35a, whose structure was confirmed by ¹H NMR (singlet one proton signal at δ 7.95). Hydrogenolysis of 35a gave 35b, which was subjected to Curtius reaction, followed by appropriate alkylation/reduction and methoxy cleavage of the intermediate 34 to afford 40, 4p, and 4q. 2-Amino-5-chloro-6-hydroxytetralins 4r-4t⁶⁰⁻⁶² and the

isomeric 7-chloro-6-hydroxy derivatives 4u-4w were synthesized as outlined in Scheme VI. Accordingly, nitration of ethyl 1,2,3,4-tetrahydro-6-methoxynaphthoate (36)⁶³ afforded a mixture of 5- (37a) and 7-nitro (37b) derivatives, whose identity was established by ¹H NMR. The 5-nitro isomer had single proton doublets at δ 6.8 and 7.1, corresponding to the protons in positions 7 and 8, respectively. The 7-nitro isomer had one proton singlets at δ 6.8 and 7.7 attributable to the protons in positions 5 and 8, respectively. The individual isomers were catalytically hydrogenated, and the resulting amines were converted to the corresponding chloro derivatives 38a and 38b by diazotization followed by treatment with CuCl.⁶⁴ Curtius reaction and appropriate acylation/reduction gave methoxyl derivatives (Table II) that were cleaved with HBr to provide 4r-4w (see Table I).

Results and Discussion

The affinity of halogenated mono- and dihydroxy-2aminotetralins 4a-4w (Table III) for D-1 dopamine receptor binding sites was measured in a test involving displacement of [³H]fenoldopam from homogenized rat striatum.⁴⁰ Affinity for D-2 receptors was evaluated from the ability of the compounds to displace [³H]spiroperidol from homogenized bovine pituitary.40 The efficacies of some of the compounds having greatest affinity in the D-1 and D-2 receptor binding studies were examined. A test that measures the ability of a compound to stimulate dopamine-sensitive rat caudate adenylate cyclase or to antagonize the stimulant effect of dopamine in this preparation was used as a measure of D-1 efficacy.⁴⁰ Similarly, for selected compounds D-2 efficacy was evaluated in a previously described^{65,66} isolated, perfused rabbit ear artery test.

Results obtained with 5- and 8-fluoro and -chloro derivatives 4a-4g of ADTN are consistent with the conceptual model of the dopamine receptor proposed by McDermed et al.⁶⁷ Thus, as has been observed with the 5- and 8-*n*-propyl derivatives⁶⁸ of ADTN in a [³H]dopamine displacement test using rat striatal membrane, D-1

- (59) Pfister, J. R.; Wyman, W. E. Synthesis 1983, 38.
- (60) Pless, J.; Seiler, M. P. (Sandoz-Patent-G.m.b.H.) Ger. Offen. 2752659, June 9, 1978; Chem. Abstr. 1978, 89, 146658e.
- (61) Pless, J.; Seiler, M. P. (Sandoz-Patent-G.m.b.H.) Ger. Offen. 2803 582, Aug 2, 1979; Chem. Abstr. 1980, 92, 22303e.
- (62) Pless, J.; Seiler, M. P. (Sandoz A.-G.) Patentschrift (Switz.) CH 637 363, July 29, 1983; Chem. Abstr. 1983, 100, 51307w.
- (63) Itoh, K.; Miyake, A.; Tanabe, M.; Hirata, M.; Oka, Y. Chem. Pharm. Bull. (Jpn.) 1983, 31, 2006.
- (64) Fieser, L. F. Experiments in Organic Chemistry, 3rd ed.; Heath and Co.: Boston, 1957; p 196.
- (65) Hieble, J. P.; Pendleton, R. G. Naunyn-Schmiedeberg's Arch. Pharmacol. 1979, 309, 217.
- (66) Steinsland, O. S.; Hieble, J. P. Science (Washington, D.C.) 1979, 199, 443.
- (67) McDermed, J. D.; Freeman, H. S.; Ferris, R. M. In Catecholamines: Basic and Clinical Frontiers; Usdin, E., Kopin, I. J., Barchas, J. D., Eds.; Pergamon: New York, 1979; pp 568-570.
- (68) Freeman, H. S.; McDermed, J. D. Spec. Publ.—Chem. Soc. 1982, No. 42, 154-166.

dopamine receptor affinity and efficacy comparable to that of ADTN are retained by 2-amino-8-chloro-6,7-dihydroxytetralin (4d), whereas the 5-chloro-substituted isomer 4b had about 20-fold less affinity for the [3H]fenoldopam binding sites and did not stimulate dopaminesensitive adenvlate cyclase to a significant degree. In contrast, 4d was about one-tenth as effective as ADTN in displacing [³H]spiroperidol from its binding sites and it was only about $1/_{100}$ as effective in inhibiting the constrictor response of the isolated, perfused rabbit ear artery to nerve stimulation, whereas 4b was one-half and onetenth as potent as ADTN in these tests, respectively. These data suggest that 8-chloro, but not 5-chloro, substitution of ADTN results in a relatively selective agonist of the D-1 subpopulation of dopamine receptors. The relatively high degree of D-2 receptor potency produced by the 5-chloro derivative 4b may suggest bulk tolerance in this subtype of dopamine receptors that is not seen in the D-1 subpopulation. This has not been addressed in previous models of the dopamine receptor(s). Both the 5- and 8-fluoro derivatives, 4a and 4c, respectively, were approximately equipotent with ADTN in tests for D-1 receptor affinity and activity. This is in conformity with the hypothetical model⁶⁷ of the dopamine receptor if one considers that the bulk of the fluorine substituent is about the same as that of a hydrogen; i.e., the fluoro group does not abut the postulated site of bulk intolerance to prevent access of the molecule to the hypothesized primary binding sites on the receptor. In tests involving D-2 receptors, 4a and 4c were similar. Both were less potent than ADTN in displacing [³H]spiroperidol, but they were equally or slightly more potent than the parent in the rabbit ear artery paradigm.

The N-substituted derivatives 4e, 4f, and 4g of 2amino-8-chloro-6,7-dihydroxytetralin (4d), as noted for the N,N-dipropylated derivative of ADTN,¹³ had decreased effectiveness in D-1 receptor-related tests whereas they were more potent than 4d in the D-2 receptor-related tests. This was most strikingly observed with the N-propyl-N-(p-hydroxyphenethyl) derivative 4g, which was the most potent member of the series as a displacer of [³H]spiroperidol binding (7-fold more effective than ADTN) but was less than one-tenth as potent as ADTN in the rabbit ear artery test.⁶⁹

2-Amino-6-fluoro-7-hydroxytetralin (4h) was about one-fifth as potent as ADTN as a displacer of [³H]fenoldopam binding and about one-third as effective as the prototype in the [³H]spiroperidol displacement test. N,-N-Dimethylation of 4h, i.e. to give 4i, and N,N-dipropylation, i.e. to afford 4j, decreased affinity in the D-1 receptor binding test, but slightly increased it in the D-2 receptor binding procedure. The latter compound was about one-third as potent as ADTN in the rabbit ear artery test. The homologue of 4j, i.e., the aminomethylated tetralin 4k, which lacks a dopaminergic phenethylamine pharmacophore,⁷⁰ as anticipated, had little or no activity

⁽⁶⁹⁾ The difference in data obtained in the D-2 receptor-related tests, i.e. [³H]spiroperidol displacement and the rabbit ear artery test, may be reflective of a measure of overall (high and low) affinity binding in the former procedure and predominantly high-affinity D-2 receptor binding in the latter. In addition, binding reflects both agonist and antagonist potential, whereas the rabbit ear artery test indicates only agonist activity. Although the rabbit ear artery test also shows a positive response to α_2 agonists, aminotetralins described in this paper were demonstrated to have activity at D-2 receptors by their dissociation constants ($K_{\rm B}$) for (S)-sulpiride. These $K_{\rm B}$ values were consistently between 5 and 50 nM: Flaim, K. E.; Hieble, J. P., unpublished observations.

in either of the binding tests.

Compared to the 2-amino-6-fluoro-7-hydroxytetralins, 4h, 4i, and 4j, the related 2-amino-6-chloro-7-hydroxytetralin (41) was relatively weakly effective in both binding studies, and affinity was decreased further by N.N-dimethylation (4m) or N,N-dipropylation (4n), although the latter compound had weak activity in the isolated perfused rabbit ear artery preparation. None of the 2-amino-6chloro-7-hydroxytetralins, i.e. 4l, 4m, or 4n, were significantly effective in stimulating or antagonizing dop-amine-sensitive adenylate cyclase. The reason for decreased dopamine receptor affinity of the chlorinated derivatives, 41, 4m, and 4n, as compared to their fluoro counterparts, 4h, 4i, and 4j, is not certain. For example, it may be the consequence of unfavorable steric effects of the larger chloro substituent, the effects of the halogen substitution on the acidity of the o-hyroxyl group, or the result of the fluoro substituent's ability to form hydrogen bonds with appropriate sites on the receptor. The 2amino-6,8-dichloro-7-hydroxytetralins, 40, 4p, and 4q, were even less effective than their monochlorinated counterparts as displacers of [³H]fenoldopam and [³H]spiroperidol binding. In addition, the N,N-dipropyl derivative 4q neither stimulated adenylate cyclase nor affected the activation of the enzyme by dopamine.

The 5- (4r, 4s, 4t) and 7- (4u, 4v, 4w) chlorinated 6hydroxy-2-aminotetralins, which lack the dopaminergic *m*-hydroxyphenethylamine pharmacophore⁷⁰ (the hydroxyl group is para to the embedded ethylamine chain) were generally weakly effective in the test for displacement of [³H]fenoldopam. Further, the 7-chloro derivatives, 4u, 4v, and 4w, did not stimulate or inhibit dopamine-sensitive adenylate cyclase. The N,N-dipropylated compounds 4t and 4w did show significant D-2 receptor related activity. They were about one-fourth as effective as ADTN in displacing [³H]spiroperidol from bovine pituitary homogenate binding sites, and they were 10- to 15-fold less potent than the prototype in the rabbit ear artery test.

In summary, the results of our study of halogenated mono- and dihydroxylated 2-aminotetralins indicate that in contrast to the 3-benzazepine series of dopamine receptor agonists²⁶⁻³¹ introduction of a properly located halogen, i.e. in the 8-position of ADTN, results in a slight increase in affinity for dopaminergic receptors of the D-1 subtype. In addition, 8-chloro substitution of the parent, perhaps as a consequence of increasing the acidity of the o-hydroxyl group or by binding to a complementary lipophilic site on the receptor, greatly enhances selectivity for the D-1 vs. D-2 dopamine receptor subpopulation. These observations are in general agreement with hypothetical models suggested for the dopamine receptor(s). $^{13,67-70}$ In contrast to the 3-benzazepine series of dopaminergic agents,³²⁻⁴⁰ however, replacement of either the 6- or 7hydroxy group of ADTN or various N-substituted derivatives with a halogen does not change the activity of the compound from dopamine receptor agonist to antagonist. The reason for this difference in structure-activity properties in the two series of dopamine receptor modulators is not readily apparent. Possibly, this difference may result from the tightness with which the two classes bind to D-1 receptors. 1-Phenylbenzazepines bind more tightly to D-1 receptors than do the 2-aminotetralins perhaps as a consequence of the attraction of the properly oriented 1phenyl group to an accessory site on this receptor sub-type.²⁴ Thus, the K_{Bind} in a test for displacement of [³H]fenoldopam binding to rat caudate homogenate is 5.1 nM for **2a**, the prototype of benzazepine D-1 receptor agonists, whereas for ADTN it is 70 nM. Consequently, the halogenated benzazepine D-1 antagonists, which lack the catecholic system needed for efficacy, may bind tightly to the receptor to prevent the effector event from occurring. In contrast, binding of the halogenated monohydroxylated 2-aminotetralins to the receptors may be insufficiently tight to enable blockade of these receptors.

Experimental Section

Melting points were determined in open capillary tubes on a Thomas-Hoover Uni-Melt apparatus; they are uncorrected. Elemental analyses were carried out by the Analytical, Physical and Structural Chemistry Department of Smith Kline & French Laboratories. Where analyses are reported by symbols of the elements, results were within 0.4% of the calculated value unless indicated otherwise. IR, NMR, and MS data were determined for all numbered compounds in this section and were evaluated as consistent with the indicated structures. These data are presented only where required for structural assignment. IR spectra were obtained using a Perkin-Elmer 727 IR spectrophotometer. NMR spectra were determined using an Hitachi Perkin-Elmer R-24 spectrometer with Me₄Si as an internal standard. Mass spectra were obtained on an Hitachi Perkin-Elmer 6E spectrometer. TLCs were carried out on Analtech Uniplates, silica gel GF, 250 µm. The acronyms TFA and MSA are used for trifluoroacetic and methanesulfonic acids, respectively.

Chemistry. General Methods. Methoxyl Cleavage. Method A (Boron Tribromide).⁴⁵ A 0.2 M solution of BBr₃ in CH_2Cl_2 (100 mL) was added dropwise to a stirred mixture of 20 mmol of the methoxylated or dimethoxylated 2-aminotetralin in 100 mL of CH_2Cl_2 at -78 °C. The cooling bath was removed, and the reaction was stirred at ambient temperature for 24 h. After the mixture was cooled in an ice-H₂O bath, 100 mL of MeOH was added dropwise. The resulting solution was again concentrated. The residue was dissolved in 100 mL of MeOH, and the solution was agan concentrated. This procedure was repeated three times, and then the solid residue was recrystallized to give the amine hydrobromide (see Table I).

Method B (Hydrobromic Acid). A stirred solution of 10 mmol of the appropriate methoxylated 2-aminotetralin derivative and 30 mL of freshly distilled concentrated HBr was heated at 110 °C for 2.5 h. After the solution was concentrated in vacuo to about 10–20 mL, it was cooled to 0–10 °C and the crystalline hydrobromide was filtered and recrystallized from the appropriate solvent (Table I). In some cases the hydrobromide was converted to the free base, chromatographed, and then treated with HCl to give a hydrochloride.

(2-Fluoro-3,4-dimethoxyphenyl)acetic Acid (5a). A mixture of 50.4 g (0.26 mol) of (2-fluoro-3,4-dimethoxyphenyl)acetonitrile,⁴¹ KOH (254.7 g, 4.5 mol), 500 mL of EtOH, and 500 mL of H₂O was stirred and refluxed for 20 h. The resulting solution was concentrated to a volume of about 700 mL. After being cooled at 25 °C, the solution was extracted twice with Et₂O. The aqueous phase was filtered, and the filtrate was poured slowly into a stirred solution of 450 mL of 12 N HCl in 600 mL of H₂O. The resulting suspension was stirred for 1 h at -10 to 0 °C, and then it was filtered. After the solid was washed with H₂O, it was dried to give 44.0 g (80%) of colorless crystals, mp 100-101 °C. Anal. (C₁₀H₁₁FO₄) C, H.

(2-Fluoro-3,4-dimethoxyphenyl)ethanol (6a). To a stirred solution of 52.4 g (0.25 mol) of (2-fluoro-3,4-dimethoxyphenyl)-acetic acid (5a) in 500 mL of THF, under Ar, was added dropwise during 1 h 300 mL (0.3 mol) of a 1 M solution of BH₃ in THF. After being stirred at 25 °C for 24 h, the reaction mixture was cooled to 0-5 °C and 160 mL of MeOH was added dropwise. The resulting solution was concentrated in vacuo to leave a liquid residue. A solution of this residue in 300 mL of Et₂O was washed twice with a saturated solution of NaHCO₃. The ethereal solution was dried (Na₂SO₄) and concentrated, and the residue was distilled to give 46.6 g (98%) of a colorless liquid, bp 128–130 °C (0.15 torr). Anal. (C₁₀H₁₃FO₃) C, H.

2-(2-Chloro-3,4-dimethoxyphenyl)ethanol (6b) was prepared by BH₃ reduction of (2-chloro-3,4-dimethoxyphenyl)acetic acid

⁽⁷⁰⁾ Kaiser, C.; Jain, T. Med. Res. Rev. 1985, 5, 145.

⁽⁷¹⁾ Gunter, M. J.; Mander, L. N. Austr. J. Chem. 1981, 34, 695.

 $(\mathbf{5b})^{42}$ in the same manner as described for the fluoro analogue 5a. The yield of colorless liquid, bp 128–132 °C (0.2 torr), was 94%. Anal. (C₁₀H₁₃ClO₃) C, H.

2-(2-Fluoro-3,4-dimethoxyphenyl)ethyl Tosylate (7a).⁷² Tosyl chloride (8.1 g, 0.04 mol) was added in small portions to a stirred solution of **6a** (4.2 g, 0.02 mol) in 15 mL of dry pyridine. The resulting suspension was stirred at 5 °C for 16 h, and then 5 mL of H₂O was added slowly. After the mixture was stirred at 5 °C for 1 h, it was diluted with 50 mL of cold H₂O. The mixture was extracted with Et₂O, and the ethereal extracts were washed with 0.1 N HCl, H₂O, and finally a saturated aqueous solution of NaHCO₃. The Et₂O solution was dried (Na₂SO₄) and concentrated to leave a 6.6 g (89%) of a solid residue, mp 72–73 °C, after trituration with petroleum ether. Anal. (C₁₇H₁₉FO₅S) C, H.

2-(2-Chloro-3,4-dimethoxyphenyl)ethyl Bromide (7b). PBr_3 (3.8 g, 14 mmol) in 30 mL of Et₂O was added to a stirred solution of 6.03 g (28 mmol) of **6b** in 30 mL of Et₂O. The solution was refluxed for 1 h, cooled to 5 °C, and *cautiously* poured onto 60 g of crushed ice. After the Et₂O phase was separated and washed successively with a saturated aqueous solution of NaHCO₃, H₂O, and brine, it was dried (Na₂SO₄) and concentrated. The residual liquid was distilled, and the fraction (5.3 g, 68%), bp 112–115 °C (0.15 torr), was collected. Anal. (C₁₀H₁₂BrClO₂) C, H.

Diethyl α -[2-(2-Fluoro-3,4-dimethoxyphenyl)ethyl]malonate (8a). Diethyl malonate (13.7 g, 86 mmol) was added dropwise at 20 °C to a stirred suspension of 1.4 g (56 mmol) of NaH in 55 mL of THF under an atmosphere of Ar. To the resulting reaction mixture was added dropwise a solution of 15.2 g (43 mmol) of 7a in 50 mL of THF. The mixture, under Ar, was stirred and refluxed for 40 h, and then it was concentrated in vacuo. The residue was partitioned between 50 mL of Et₂O and 50 mL of H₂O. After the combined Et₂O extracts were washed with H₂O and brine, they were dried (Na₂SO₄) and concentrated. The residue was fractionally distilled to give 9.8 g (67%) of a colorless liquid, bp 175 °C (0.13 torr). Anal. (C₁₇-H₂₃FO₆) C, H.

Diethyl α -[2-(2-chloro-3,4-dimethoxyphenyl)ethyl]malonate (8b) was prepared from 7b and diethyl malonate by essentially the same procedure described for preparation 32a (48 h reflux period). The product (83%) was obtained as a colorless liquid, bp 172 °C (0.15 torr). Anal. (C₁₇H₂₃ClO₆) C, H.

Ethyl Hydrogen α -[2-(2-Fluoro-3,4-dimethoxyphenyl)ethyl]malonate (9a). A solution of 1.6 g (28 mmol) of KOH in EtOH was added dropwise, over a period of 1 h, to a stirred solution of 9.6 g (28 mmol) of 8a in 25 mL of EtOH. After being stirred at ambient temperature for 60 h, the mixture was concentrated in vacuo. The residue was diluted with H₂O, and the resulting mixture was extracted with Et₂O. The aqueous phase was acidified with 2.5 N HCl, and the mixture was extracted with Et₂O. After being washed with H₂O and brine, the Et₂O extract was dried and concentrated in vacuo to give 7.3 g (83%) of a colorless syrup. Anal. (C₁₅H₁₉FO₆) C, H.

Ethyl hydrogen α -[2-(2-chloro-3,4-dimethoxyphenyl)ethyl]malonate (9b) was prepared from 8b by the same procedure described for synthesis of 9a from 8a. The product (81%) was obtained as a viscous, colorless liquid; TLC (silica gel GF, 95:5:0.2 CHCl₃-CH₃OH-HCOOH) gave a single spot, R_f 0.9. It was employed for further reaction without additional purification.

Ethyl 5-Fluoro-6,7-dimethoxy-1-oxo-1,2,3,4-tetrahydro-2naphthoate (10a). A solution of 6.7 g (21 mmol) of 9a in 35 mL of approximately 3 M PPE in CHCl₃ was stirred under Ar at ambient temperature for 24 h. The reaction mixture was poured onto 50 g of crushed ice with vigorous stirring. After the layers were separated, the aqueous phase was extracted with Et₂O. The combined organic extracts were washed successively with H₂O, 5% aqueous NaHCO₃, H₂O, and brine. After being dried (Na₂SO₄), the organic extract was concentrated to give 3.1 g of a viscous liquid that solidified on standing at 25 °C. Unreacted 9a (2.6 g) was recovered from the NaHCO₃ extracts. This was treated with PPE as described initially to give an additional 1.9 g of crude product. The combined crude product (4.9 g) was chromatographed by MPLC using a 2.5 × 50 cm column packed with 124 g of 230-400-mesh silica gel and a mobile phase of 3.3 mL of 97% formic acid/L of CHCl₃. A total of 54 fractions (15 mL) were collected at a flow rate of 5 mL/min. Fractions 6-34 were combined and concentrated to give 1.8 g of a viscous liquid that crystallized on standing at 25 °C; TLC (silica gel GF, 100:0.2 CHCl₃-HCOOH) showed one spot at R_f 0.35. Recrystallization from EtOH gave 1.2 g (20%) of colorless needles, mp 77-79 °C. Anal. (C₁₅H₁₇FO₅) C, H.

Ethyl 5-chloro-6,7-dimethoxy-1-oxo-1,2,3,4-tetrahydro-2naphthoate (10b) was prepared from 9b in the same manner as described for the preparation of 10a from 9a. In this case, however, the crude product was purified by silica gel column chromatography using CHCl₃ for elution. All fractions that gave a single spot at R_f 0.3 (silica gel GF, CHCl₃) were combined and concentrated. The residue was crystallized from aqueous EtOH to give colorless crystals (41% yield), mp 79-83 °C. Anal. (C₁₅H₁₇ClO₅) C, H.

5-Fluoro-6,7-dimethoxy-1,2,3,4-tetrahydro-2-naphthoic Acid (11a). A mixture of 1.9 g (6.3 mmol) of 10a, 0.4 g of 10% Pd/C catalyst, 0.4 mL of 70% HClO₄, and 25 mL of HOAc was hydrogenated for 4 h on a Parr apparatus at 25 °C and an initial H_2 pressure of 45 psi. NaOAc (0.4 g) was added, and the mixture was stirred for 5 min. The mixture was filtered, and the filtrate was concentrated in vacuo. The residue was taken into Et₂O. After the ethereal solution was washed with a saturated aqueous solution of NaHCO₃, it was dried and concentrated to give 1.7 g (96%) of crude ethyl 5-fluoro-6,7-dimethoxy-1,2,3,4-tetrahydro-2-naphthoate. To this ester (1.5 g, 5.4 mmol) in 10 mL of MeOH was added a solution of 0.6 g (11 mmol) of KOH in 10 mL of H_2O . The resulting mixture was stirred and refluxed for 2 h. After the MeOH was distilled from the reaction mixture, the remaining solution was diluted with 50 mL of H_2O . The resulting solution was washed with Et₂O, and then it was acidified with 2.5 N HCl to give 1.3 g (93%) a crystalline product, mp 178–181 °C. Anal. (C₁₃H₁₅FO₄) C, H.

5-Chloro-6,7-dimethoxy-1,2,3,4-tetrahydro-2-naphthoic acid (11b) was prepared from 10b in the same fashion as that described for conversion of 10a to 11a. Crude intermediate *ethyl* 5*chloro-6*,7-*dimethoxy-1,2,3,4-tetrahydro-2-naphthoate* (96% yield) was obtained as colorless crystals, mp 56-57 °C. The acid 35b (97% yield) was a colorless solid, mp 172-174 °C. Anal. $C_{13}H_{15}ClO_4$) C, H.

2-Amino-5-fluoro-6,7-dimethoxytetralin Hydrochloride (34a).⁴² A suspension of 1.2 g (4.7 mmol) of 11a in 2 mL of SOCl₂ was heated at 60 °C for 4 h. The resulting solution was concentrated and stripped twice with toluene. The resulting acid chloride was dissolved in 2 mL of acetone, and the solution was added dropwise to a stirred solution of 0.5 g (7.2 mmol) of NaN₃ in 2 mL of H₂O at 0 °C. After the resulting suspension was stirred for 30 min at 0–20 °C, it was diluted with 50 mL of H_2O and cooled to 0-5 °C. The solid acid azide was filtered, washed with H_2O_1 and dried in the air at 25 °C. A solution of the azide in 3 mL of toluene was heated at 100 °C for 1 h, and then it was concentrated. The resulting isocyanate was dissolved in 2.3 mL of benzyl alcohol, and the resulting solution was heated at 100 $^{\circ}\mathrm{C}$ for 6 h. After standing at 25 °C, the mixture solidified. It was triturated with n-C₆H₁₄ to give 1.53 g (90%) of tan crystalline benzyl carbamate, mp 128-130 °C. Hydrogenolysis of a mixture of the carbamate, 0.7 mL of concentrated HCl, 300 mg of Pd/C and 100 mL of EtOH at an initial pressure of 50 psi of H_2 at 25 °C, following filtration, concentration and trituration of the residue with Et₂O gave 1.06 g (97%) of colorless crystals, mp 237-239 °C dec (Table II).

2-Amino-5-chloro-6,7-dimethoxytetralin Hydrochloride (34b).⁵⁸ A solution of ethyl chloroformate (1.1 g, 10.1 mmol) in 2 mL of acetone was added slowly to a stirred solution of 11b (2.5 g, 9.2 mmol) and Et₃N (10.2 mmol) in 17 mL of acetone at -5 °C. The mixture was stirred at 0 °C, and a solution of NaN₃ (1.2 g, 18.3 mmol) in 34 mL of H₂O was added *dropwise* during 30 min. After the mixture was stirred at 0 °C for 1 h, it was poured into 34 mL of ice-H₂O. The mixture was extracted with toluene. After the toluene extracts were dried (MgSO₄), they were added dropwise to 100 mL of stirred toluene at 100 °C. Heating at 100 °C was continued for 1 h, and then the solution was concentrated. To the residual liquid was added 18.6 mL of 8 N HCl, and the

mixture was stirred at 100 °C for 1 h. The mixture was concentrated, and the residue was dissolved in H₂O. The aqueous solution was made alkaline with 2 N NaOH. After the mixture was extracted with Et₂O, the extracts were dried (MgSO₄) and concentrated. A solution of the residual liquid in Et₂O was acidified with HCl. The crystalline solid was washed with Et₂O to give 0.73 g (48%) of colorless crystals after recrystallization from MeOH–EtOAc (Table II).

(2-Fluoro-3,4-dimethoxyphenyl)acetyl Chloride (12). A stirred suspension of 43 g (0.2 mol) of 5 and 72 g (0.6 mol) of SOCl₂ in 460 mL of toluene was heated at 100 °C for 2.5 h. Stirring was continued at 25 °C for 16 h, and then the solution was concentrated under reduced pressure. Distillation of the residue afforded 45.7 (97%) of a pale pink liquid, bp 128–132 °C (0.26 torr). Anal. ($C_{10}H_{10}$ ClFO₃) H; C: calcd, 51.63; found, 51.03.

8-Fluoro-6,7-dimethoxy-2-tetralone (13) was prepared according to the general method of Burckhalter and Campbell.⁴⁶ A solution of 45.5 g (0.2 mol) of 12 in 200 mL of CH_2Cl_2 was added dropwise during 1.5 h to a stirred suspension of $AlCl_3$ (52.2 g, 0.39 mol) in 800 mL of CH₂Cl₂ at -78 °C. Ethylene was bubbled rapidly through the mixture for 25 min. After removal of the cooling bath, stirring was continued for 3.5 h and 250 mL of H₂O was added dropwise with caution. The organic layer was separated and washed with 2 N HCl, a saturated aqueous solution of NaHCO₃, and brine. After being dried (Na₂SO₄), the solution was concentrated in vacuo to give 44.1 g of a viscous liquid that was triturated with a small volume of cold Et_2O to give 11.3 g of a crude monohydroxy monomethyl derivative. A mixture of 10.3 g (49 mmol) of this crude material, 7.4 g (59 mmol) of dimethyl sulfate and 31.3 g (0.23 mol) of K₂CO₃ in 80 mL of DMF was stirred at 100 °C for 1.5 h. The cooled mixture was diluted with H_2O and extracted with $Et_2O.$ The Et_2O extracts were washed with $H_2O,$ dried, and concentrated. The residue was chromatographed by MPLC (silica gel 60, 230-400 mesh; mobile phase 98:2 CH₂Cl₂-MeOH). Fractions giving a single spot at R_f 0.3 (70:30 c-C₆ H_{12} -Et₂O) were combined and concentrated, and the residual solid was recrystallized from $c-C_6H_{12}-n-C_6H_{14}$ to give 2.1 g (19%) of colorless crystals, mp 94–95 °C. Anal. (C₁₂H₁₃FO₃) C, H.

2-(Benzylamino)-8-fluoro-6,7-dimethoxytetralin Hydrochloride (14a). A solution of 2.0 g (9 mmol) of 13b, p-toluenesulfonic acid (0.06 g), and benzylamine (1.2 g, 11 mmol) in 100 mL of toluene was refluxed azeotropically under Ar for 2 h. After CO_2 was bubbled through the solution, it was washed successively with H₂O, a saturated aqueous NaHCO₃ solution, and brine, dried, and concentrated. A mixture of the residual imine, 280 mg of PtO₂, and 100 mL of EtOH was hydrogenated at 25 °C and an initial pressure of 50 psi of H₂ for 1.5 h. After the mixture was filtered, the filtrate was made acidic with HCl and concentrated in vacuo. The residual solid was recrystallized from MeOH–EtOH to give 2.1 g (70%) of a white solid, mp 247–250 °C. Anal. ($C_{19}H_{22}FNO_2$ ·HCl) C, H, N.

2-Amino-8-fluoro-6,7-dimethoxytetralin Hydrochloride (14b). A suspension of 2.1 g (5.9 mmol) of 14a, 0.5 mL of 12 N HCl, and 100 mL of EtOH was hydrogenated at 60 °C and an initial H₂ pressure of 45 psi for 5 h. The mixture was filtered and the filtrate concentrated. Recrystallization of the solid residue from MeOH-EtOAc gave 1.5 g (94%) of colorless crystals, mp 225-228 °C dec. Anal. ($C_{12}H_{16}FNO_2$ ·HCl) C, H, N.

Methyl 8-Chloro-6,7-dimethoxy-4-oxo-1,2,3,4-tetrahydro-2-naphthoate (16). A warm solution of 2-chloroveratraldehyde⁴⁴ (100.3 g, 0.5 mol) and dimethyl succinate (83 g, 0.57 mol) in 170 mL of MeOH was added to a refluxing solution of NaOMe in MeOH, prepared by the cautious portionwise addition of 13.8 g (0.6 mol) of Na to 500 mL of MeOH under N₂. After the mixture was refluxed for 2 h, 350 mL of solvent was distilled off. The remaining solution was cooled to 10 °C, made acidic with 2 N HCl, and diluted with 1.5 L of H_2O . Precipitated solid was filtered and partitioned between equal volumes of Et₂O and a saturated aqueous solution of NaHCO₃. The aqueous NaHCO₃ solution was acidified with 12 N HCl. After the resulting mixture was extracted with EtOAc, the organic extract was washed (H_2O) , dried $(MgSO_4)$, and concentrated to give 52.8 g of a solid benzylidene derivative, mp 140–143 °C, after recrystallization from EtOAc–n-C₆H₁₄. This was dissolved in EtOAc and hydrogenated in the presence of 10% Pd/C catalyst at 25 °C and 65 psi of H_2 to give 52.1 g of a *benzyl*

derivative, mp 118–121 °C. A mixture of this material and 375 mL of MSA was heated at 100 °C for 15 min. The resulting solution was poured onto 2 L of ice–H₂O. Precipitated solid was filtered, washed with H₂O, dried in vacuo at 80 °C, and recrystallized from aqueous MeOH to give 43.1 g (92%) of crystalline solid, mp 146–147 °C. Anal. ($C_{14}H_{15}ClO_5$) C, H.

Methyl 8-Chloro-6,7-dimethoxy-1,2,3,4-tetrahydro-2naphthoate (17a) was prepared by reductive hydrogenolysis of 16 in the same manner as described for reduction of 34a to 35a. The crystals, mp 92–93 °C (99% yield), were recrystallized from MeOH; TLC (silica gel GF, CHCl₃) showed a single spot at R_f 0.4. Anal. (C₁₄H₁₇ClO₄) C, H.

8-Chloro-6,7-dimethoxy-1,2,3,4-tetrahydro-2-naphthoic Acid (17b). A suspension of 29.6 g (97 mmol) of 17a in a solution of 10.9 g (0.184 mol) of KOH in 200 mL of 50% aqueous MeOH was stirred and refluxed for 24 h. The solution was concentrated to 100 mL in vacuo, cooled, washed with Et₂O, and made acidic with 12 N HCl to give 25.26 g of colorless crystals, mp 212–214 °C, after recrystallization from EtOAc. Anal. ($C_{13}H_{15}ClO_4$) C, H.

2-Amino-8-chloro-6,7-dimethoxytetralin hydrochloride (18) was prepared from 17b by the same procedure as that described for conversion of 11a to 34a. The intermediate benzyl carbamate (83%) melted at 119–121 °C after recrystallization from 2-PrOH. The yield of colorless crystals, mp 269–271 °C dec, after trituration with Et_2O was 96%. Anal. ($C_{12}H_{16}CINO_2$ ·HCl) C, H, N.

8-Chloro-6,7-dimethoxy-2-(di-n-propylamino)tetralin Hydrochloride (19). A mixture of 18 (5.6 g, 20 mmol), 5.81 g (0.1 mol) of propionaldehyde, 1.6 g (20 mmol) of NaOAc, and 2 g of 10% Pd/C catalyst was hydrogenated for 10 h at 25 °C and an initial H₂ pressure of 60 psi. The mixture was filtered, and the filtrate was concentrated in vacuo. The residue was suspended in H₂O, and the mixture was made alkaline with 10 N NaOH and extracted into Et₂O. After the ethereal solution was washed (H₂O) and dried (MgSO₄), it was concentrated. The residue was taken into Et₂O and acidified with gaseous HCl. The resulting hydrochloride was recrystallized from MeCN-Et₂O to give 3.8 g (52%) of crystals, mp 158-160 °C. Anal. (C₁₈H₂₈ClNO₂·HCl) C, H, N.

8-Chloro-6,7-dimethoxy-2-[[(4-methoxyphenyl)acetyl]amino]tetralin (20a). To a suspension of 18 (2.5 g, 9 mmol) in 50 mL of THF was added 2.17 g (21.4 mmol) of Et_3N . The mixture was stirred for 10 min at ambient temperature, and then it was cooled to 0 °C and 2.2 g (12 mmol) of 4-methoxyphenylacetyl chloride in 10 mL of THF was added dropwise. The mixture was stirred at 25 °C for 16 h, and then it was filtered. The filtrate was concentrated in vacuo. A solution of the residual solid in CHCl₃ was washed (H₂O, saturated aqueous NaHCO₃ solution, H₂O), dried (MgSO₄), and concentrated. The residue was triturated with a small volume of cold Et_2O to give 2.4 g (67%) of colorless crystals, mp 144–147 °C. Anal. (C₂₁H₂₄ClNO₄) C, H, N.

8-Chloro-6,7-dimethoxy-2-[[2-(4-methoxyphenyl)ethyl]amino]tetralin Hydrochloride (20b).⁵⁴ A solution of 2.3 g (5.9 mmol) of 20a in 20 mL of THF was added dropwise at 0 °C to a stirred solution of 14 mL of a 1 M solution of BH₃ in THF under Ar. After the reaction mixture was refluxed for 2 h, it was cooled, 4 mL of 3 N HCl was added, and the THF was removed by distillation. The resulting suspension was cooled to 10 °C, 30 mL of THF was added, and the crystals (2.1 g, 87%), mp 268-270 °C, were filtered. Anal. ($C_{12}H_{26}CINO_3\cdotHCl)$ C, H, N.

8-Chloro-6,7-dimethoxy-2-[N-[2-(4-methoxyphenyl)ethyl]-N-n-propylamino]tetralin (21). To a stirred suspension of 20b (1.7 g, 4.1 mmol) and K_2CO_3 (3 g, 22 mmol) in 10 mL of H_2O and 10 mL of CHCl₃ was added 0.9 g (10 mmol) of propionyl chloride. After being stirred at 25 °C for 1 h, the mixture was filtered and the organic phase was separated. The CHCl₃ solution was washed (2 N HCl, H_2O), dried (MgSO₄), and concentrated to give 1.5 g of a viscous amide. This amide was reduced with BH₃ as described in the preceding example. An aqueous solution of the resulting crude hydrochloride was made alkaline with 2.5 N NaOH. After the mixture was extracted with Et₂O, the extracts were washed (H₂O), dried (MgSO₄), and concentrated to give 1.4 g (96%) of a crude product; TLC (silica gel GF, 98:2 CHCl₃-MeOH) R_f 0.45. This material was used for conversion to 4g without further purification.

4-(3-Fluoro-4-methoxyphenyl)butyric Acid (22). To a stirred mixture of 113 g (0.9 mol) of 2-fluoroanisole, 1 L of 1,2dichloroethane, and 99 g (0.1 mol) of succinic anhydride at 0-10 °C was added, in portions, during 30 min, 293 g (2.2 mol) of AlCl₃. The mixture was gradually heated to 65 °C during 40 min. After 20 min at 65 °C, the mixture was allowed to return to ambient temperature during 3 h, and then it was poured into a mixture of 1 kg of ice and 500 mL of 12 N HCl. The solid was filtered and dried. Recrystallization from EtOH afforded 153 g (76%) of 3-(3-fluoro-4-methoxybenzoyl)propionic acid, mp 168-170 °C. A mixture of 27 g (0.12 mol) of this acid, 200 mL of EtOH, 60 mL of HOAc, and 3 mL of 10% HClO₄ was warmed to effect solution. After being cooled slightly, the solution was added to 3 g of 10% Pd/C catalyst (wetted with a small volume of EtOAc), and the mixture was hydrogenated for 3.5 h at ambient temperature at an initial pressure of 60 psi of H_2 . The mixture was filtered, and the filtrate was adjusted to pH 5.5 with NaOAc. After concentration of the solution in vacuo, the residue was dissolved in a mixture of equal volumes of EtOAc and H_2O . The organic phase was washed with brine, dried, and concentrated. The residual syrup was recrystallized from $Et_2O-n-C_6H_{14}$ to give 15.7 g (62%) of colorless crystals, mp 61-62 °C. Anal. $(C_{11}H_{13}FO_3)$ C, H.

6-Fluoro-7-methoxy-1-tetralone (23). A solution of 20 g (0.094 mol) of **22**, 30 mL of MSA, and 100 mL of TFA was stirred at 100 °C for 45 min. It was then poured onto 500 mL of ice-H₂O, and the mixture was extracted with EtOAc. The extracts were washed (H₂O, saturated aqueous NaHCO₃, brine), dried (MgSO₄), and concentrated. The residue was crystallized from 50 mL of EtOH to afford 11.85 g (65%) of colorless crystals: mp 93.5-95 °C; TLC (silica gel, 50:50 EtOAc-c-C₆H₁₂) gave a single spot having R_f 0.67; NMR (270 MHz) (CDCl₃) δ 6.91 [d, 1 H, J = 11.2 Hz, ArH (5)], 7.62 [d, 1 H, J = 8.8 Hz, ArH (8)]. Anal. (C₁₁H₁₁FO₂) C, H.

6-Fluoro-7-methoxy-1-oximinotetralin (24a). To a stirred solution of 1.2 g (6.2 mmol) of **23** in 15 mL of warm EtOH was added a solution of 0.64 g (9.3 mmol) of hydroxylamine hydrochloride and 0.43 g (3.1 mmol) of K_2CO_3 in 6 mL of H_2O . The mixture was stirred at reflux temperature for 2 h, and then it was concentrated. The residual solid was triturated with H_2O and recrystallized from EtOH to give 1.2 g (94%) of colorless crystals, mp 166–168 °C. Anal. ($C_{11}H_{12}FNO_2$) C, H, N.

6-Fluoro-7-methoxy-1-oximinotetralin Tosylate (24b). Freshly crystallized tosyl chloride (1.7 g, 8.7 mmol) was added to a solution of 1.2 g (5.8 mmol) of **24a** in 10 mL of pyridine. The resulting suspension was stirred at 25 °C for 3 h and poured into ice–H₂O, and the white precipitate was filtered. Rapid recrystallization (the product decomposes rapidly) from EtOH gave 2 g (97%) of crystals, mp 147–149 °C. Anal. (C₁₈H₁₈FNO₄S) C, H, N.

2-Amino-6-fluoro-7-methoxy-1-tetralone Hydrochloride (25). At 0 °C, under N₂, a solution of 2 g (5.6 mmol) of 24b in 40 mL of CH₂Cl₂ and 5 mL of EtOH was added to a stirred solution of 0.78 g (6.4 mmol) of t-BuOK in 15 mL of EtOH. After the mixture was stirred at 0 °C for 24 h, it was filtered. The filtrate was diluted with 50 mL of 6 N HCl. The acid layer was separated, washed with CH₂Cl₂, and then concentrated in vacuo at 40 °C. Azeotroping several times with EtOH gave 0.75 g of off-white crystals: mp 224-226 °C dec; MS (CH₄, CI) m/e 210 (supports M + 1 structure of the base); TLC (silica gel, 5:95:1 CH₃OH-CH₂Cl₂-concentrated aqueous NH₃) gave a single spot having R_f 0.38. Anal. (C₁₁H₁₂FNO₂·HCl·0.75H₂O) C, H; N: calcd, 5.40; found, 6.01.

2-Amino-6-fluoro-7-methoxytetralin Hydrochloride (26a). A mixture of 8.4 g (34 mmol) of 25, 6.7 g of 10% Pd/C catalyst, 800 mL of 3 N HCl, and 150 mL of HOAc was hydrogenated at 60 °C and an initial H₂ pressure of 60 psi for 30 h. The mixture was filtered and filtrate concentrated. The residue was azeotroped several times with EtOH-toluene to give 7.5 g (94%) crude 26a as an amorphous solid that was used for further reaction without additional purification.

tert-Butyl 6-fluoro-7-methoxytetralin-2-carbamate (26b) was prepared to separate 26a from minor impurities. Crude 26a (2.5 g, 10.8 mmol) was dissolved in 50 mL of DMF and 3 mL of Et₃N, 4.7 g (21 mmol) of di-tert-butyl dicarbonate was added, and the mixture was stirred at 25 °C for 17 h. The mixture was

concentrated in vacuo, and the residue was partitioned between EtOAc and H₂O. The organic layer was separated, washed with cold 1.5 N HCl, H₂O, 5% aqueous NaHCO₃, H₂O, and brine, dried, and concentrated. The residue was triturated with n-C₆H₁₄ to give 0.9 g (29%) of crystals: mp 116–118.5 °C; TLC (silica gel 66:33 c-C₆H₁₂-EtOAc) showed a single spot at R_f 0.81; MS, m/e 266 (M + 1). The material was used for conversion to 4**h**.

6-Fluoro-7-methoxy-2-(dimethylamino)tetralin (27a). Crude **26a** (3.5 g, 15 mmol) was added (exotherm) to a mixture of 4.5 g (72 mmol) of NaBH₃CN, 5.1 mL of a 37% solution of formaldehyde, and 35 mL of MeOH. After the stirred mixture was maintained at 25 °C and pH 6.2–6.5 (by addition of HOAc) for 20 h, it was gradually acidified to pH 1 with 3 N HCl and volatiles were removed at 30 °C in vacuo. The residue was partitioned between H₂O and EtOAc, and the aqueous phase was made basic with 2.5 N NaOH. The mixture was extracted with EtOAc. The extracts were washed (H₂O), dried (MgSO₄), and concentrated to give 1.2 g (35%) of an oily product that showed a single spot at R_f 0.48 on TLC (silica gel, 15:85:1 CH₃OH-CH₂Cl₂-concentrated aqueous NH₃). This material was used for conversion to **4i**.

6-Fluoro-7-methoxy-2-(di-n-propylamino)tetralin Hydrochloride (27b). To a stirred solution of 43 mL (0.57 mol) of propionic acid in 300 mL of toluene under N₂ was added, in portions, 7.2 g (0.19 mol) of NaBH₄ with intermittent cooling to maintain the temperature at 15–20 °C. The mixture was stirred at 15–20 °C until H₂ evolution had ceased (about 1.5 h). Then, 4 g (17 mmol) of crude 26a was added. After the stirred mixture, under N₂, was heated at 80–85 °C for 22 h, it was cooled, excess 2.5 N NaOH was added, and it was extracted with EtOAc. The extracts were washed (brine), dried (MgSO₄), and concentrated. The residual liquid in EtOAc was treated with HCl to give crude 27b that was used for preparation of 4j.

2-Carboethoxy-6-fluoro-7-methoxy-1-tetralone (28a). To a stirred suspension of 1 g (21 mmol) of NaH (50% dispersion in mineral oil) in 40 mL of THF under N2 was added dropwise 2.36 g (20 mmol) of diethyl carbonate, followed by 1.9 g (10 mmol) of 23 in 10 mL of THF. After being refluxed for 17 h, the mixture was cooled and 5 mL of HOAc was added dropwise with caution. The mixture was extracted with Et_2O . The Et_2O extracts were washed (saturated aqueous NaHCO₃ solution, brine), dried (MgSO₄), and concentrated. The residual liquid was chromatographed over silica gel, eluting with a gradient of CH_2Cl_2 to 0.5% MeOH in CH_2Cl_2 . The first fraction gave 2.2 g (85%) of crystals, mp 70-72 °C, after crystallization from EtOH: NMR (CDCl₃) δ 1.29 (t, 3 H, CH₂CH₃), 2.41 (m, 2 H), 2.96 (t, 2 H), 3.54 (q, 1 H), 3.91 (s, 3 H, OCH₃), 4.24 (q, 2, H, OCH₂CH₃), 6.91 (d, 1 H, J = 12 Hz, ArH), 7.61 (d, 1 H, J = 8 Hz, ArH). Anal. (C₁₄H₁₅FO₄) C, H.

Further elution gave 0.44 g of a second solid, mp 92–94 °C, after recrystallization from EtOH. This was identified as 2-carboethoxy-6-ethoxy-7-methoxy-1-tetralone (**28b**): MS (CH₄, CI) m/e 293 (M + 1); NMR (CDCl₃) δ 1.29 (t, 3 H, CO₂CH₂CH₃), 1.48 (t, 3 H, ArOCH₂CH₃), 6.75 [s, 1 H, ArH (5)], 7.54 [s, 1 H, ArH (8)]. Anal. (C₁₆H₂₀O₅) C, H.

6-Fluoro-7-methoxy-1,2,3,4-tetrahydronaphthoic Acid (28c). A mixture of 1.6 g (6 mmol) of 28a, 35 mL of EtOH, 15 mL of HOAc, 16 drops of 70% HClO₄, and 2 g of 10% Pd/C catalyst at 25 °C was hydrogenated for 6 h at an initial H₂ pressure of 60 psi. The mixture was filtered, and the filtrate was brought to pH 4.5 with NaOAc. The solution was concentrated in vacuo. A solution of the residue in 10 mL of 6 N HCl and 10 mL of HOAc was heated at 100 °C for 4 h, and 5 mL of H₂O was added. Upon being cooled to -5 °C, 0.66 g (49%) of crystals, mp 165-169 °C, after recrystallization from EtOH, deposited from the solution. Anal. (C₁₂H₁₃FO₃) C, H.

6-Fluoro-7-methoxy-2-(di-n-propylamino)tetralin Hydrochloride (29). A solution of 0.6 g (2.7 mmol) of 28c, 0.74 g (5.4 mmol) of 1-hydroxybenzotriazole, 0.28 g (2.7 mmol) of din-propylamine, and 0.56 g (2.7 mmol) of dicyclohexylcarbodiimide in 15 mL of THF was stirred at ambient temperature for 17 h. The resulting mixture was filtered, the filtrate was concentrated, and the residue was partitioned between EtOAc and H₂O. The organic layer was washed (1 N HCl, H₂O, 5% aqueous NaHCO₃ solution, brine), dried (MgSO₄), and concentrated to give 0.9 g of a semisolid. This material was chromatographed on a silica

gel column with a gradient of CH₂Cl₂ to 0.5% MeOH in CH₂Cl₂ to provide 0.8 g of the tertiary carboxamide as a colorless liquid; MS (CH₄, CI) gave m/e 308 (M + 1). To a stirred solution of this material (0.8 g, 2.6 mmol) in 20 mL of THF under N₂ at 0–5 °C was added dropwise 20 mL of 1 M BH₃ in THF. After being stirred at 25 °C for 5 h, the mixture was again cooled to 0–5 °C, 20 mL of MeOH was added dropwise, and it was concentrated in vacuo. The residue was azeotroped several times with MeOH. Next, it was dissolved in MeOH, made acidic with HCl, and concentrated to give an amorphous solid: TLC (silica gel, 3:97 MeOH–CH₂Cl₂) gave a single spot having R_f 0.11; MS, m/e 294 (M + 1). This was employed for conversion to **4k** without additional purification.

4-Chloro-3-methoxybenzyl Bromide (30). A stirred mixture of 2-chloro-5-methylphenol (43 g, 0.3 mol), dimethyl sulfate (41.6 g, 0.33 mol), and powdered K₂CO₃ (103.5 g, 0.75 mol) in 500 mL of Me₂CO was heated at reflux for 16 h. After the mixture was concentrated, 500 mL of H₂O was added to the residue. The resulting mixture was extracted with Et_2O . The Et_2O extracts were washed (H₂O, brine), dried, and concentrated. Distillation of the residue afforded 45 g (95%) of 2-chloro-5-methylanisole, bp 75 °C (0.5 torr). A solution of 30 g (0.19 mol) of this anisole derivative and 37.3 g (0.21 mol) of NBS in 250 mL of CCl₄ was stirred and refluxed, while being irradiated with a 350-W IR flood lamp, for 3.5 h. After the mixture was cooled to 0 °C, it was filtered. The filtrate was concentrated and the residue distilled to give a colorless liquid: bp 115 °C (0.4 torr); NMR (CDCl₃) δ 3.73 (s, 3 H, CH₃O), 4.40 (s, 2 H, CH₂), 6.73-7.38 (m, 3 H, ArH). This material was used for conversion to 31 without additional characterization.

3-(4-Chloro-3-methoxyphenyl)-1,2,2-propanetricarboxylic Acid (31).⁴⁸ A dispersion of 4.5 g (0.19 mol) of NaH in mineral oil was washed several times with $n-C_6H_{14}$ and dried in vacuo. Triethyl 1,1,2-ethanetricarboxylate (45 g, 0.183 mol) in 70 mL of Me₂SO was added dropwise to a stirred suspension of the NaH in 20 mL of Me₂SO. After addition was completed, the mixture was stirred and heated at 60 °C for 1 h. After the stirred solution was cooled to 20 °C, a solution of 43 g (0.183 mol) of 4-chloro-3-methoxybenzyl bromide in 70 mL of Me₂SO was added slowly. The mixture was stirred at 60 °C for 1.5 h, and then the resulting solution was poured into 2 L of ice-H₂O. The mixture was extracted with Et_2O . The combined extracts were washed (H_2O). dried (MgSO₄), and concentrated. Distillation of the residual liquid afforded 64.7 g of triester, bp 200 °C (0.3 torr). The triester (10.5 g, 26.2 mmol) and a solution of 3.5 g (87 mmol) of NaOH in 40 mL of H_2O were stirred and refluxed for 4 days. The solution was diluted with 100 mL of ice- H_2O and extracted with Et_2O . The aqueous phase was made acidic with 12 N HCl, and the resulting mixture was extracted with EtOAc. After being washed $(H_2O, brine)$ and dried $(MgSO_4)$, the extracts were concentrated. Trituration of the residue with $n-C_6H_{14}$ gave 5.0 g (60%) of colorless crystals, mp 161-163 °C, after recrystallization from MeCN. Anal. (C₁₃H₁₃ClO₇) C, H.

6-Chloro-7-methoxy-4-oxo-1,2,3,4-tetrahydro-2-naphthoic Acid (32). Decarboxylation of 31 (4.8 g, 15 mmol) to (4chloro-3-methoxybenzyl)succinic acid, mp 157–160 °C (81% yield), was carried out at 180–185 °C by the general procedure of Cannon et al.⁴⁸ This succinic acid (20 g, 73 mmol) was added slowly to 140 mL of stirred MSA. After being stirred at 100 °C for 30 min, the solution was poured into 1 L of ice-H₂O. The solid was filtered and recrystallized from Me₂CO to afford 16.7 g (90%) of colorless crystals: mp 233–235 °C; ¹H NMR (Me₂SO-d₆) δ 6.97 (s, 1 H, ArH (8)), 7.88 (s, 1 H, ArH (5)). Anal. (C₁₂H₁₁ClO₄) C, H.

6-Chloro-7-methoxy-1,2,3,4-tetrahydronaphthoic Acid (33). A mixture of 2 g (7.9 mmol) of 32, 40 mL of HOAc, 0.5 mL of 70% HClO₄, and 0.5 g of 10% Pd/C catalyst was hydrogenated at 25 °C and an initial H₂ pressure of 60 psi for 2.5 h. The mixture was filtered, and 1 g of NaOAc was added to the filtrate. Most of the solvent was evaporated in vacuo. The residue was diluted with 100 mL of H₂O. The solid was filtered and recrystallized from EtOH to give 1.5 g (79%) of colorless crystals: mp 169–171 °C; ¹H NMR (Me₂SO-d₆) δ 6.65 (s, 1 H, ArH (8)), 6.97 (s, 1 H, ArH (5)). Anal. (C₁₂H₁₃ClO₃) C, H, Cl.

2-Amino-6-chloro-7-methoxytetralin hydrochloride (34c) was prepared from 33 by the same procedure described for conversion of 11b to $34b^{53,56,57}$ (see Table II).

6-Chloro-7-methoxy-2-(dimethylamino)tetralin Hydrochloride (34d). A solution of 2.37 g, 10 mmol of the isocyanate derived from 33 (by the procedure described for conversion of 11b to its isocvanate) in 30 mL of Et₂O was added dropwise to a stirred suspension of $LiAlH_4$ (0.76 g, 20 mmol) in 50 mL of Et_2O . After the mixture was stirred and refluxed for 2 h, excess LiAlH₄ was decomposed by *cautious* addition of 1 mL of H₂O, followed by 1 mL of 2.5 N NaOH and 3 mL of H₂O. The mixture was filtered and the filtrate concentrated to give the N-monomethyl derivative. To a stirred solution of this secondary amine (2.25)g, 10 mmol) in 20 mL of $CHCl_3$ was added a solution of K_2CO_3 (1.4 g, 10 mmol) in 20 mL of H₂O. To the stirred mixture was added methyl chloroformate (1.56 g, 12 mmol) in 5 mL of CHCl₃. After being stirred for 15 min, the CHCl₃ layer was separated, washed with H_2O , dried (MgSO₄), and concentrated to give the corresponding N-methyl methyl carbamate, which was reduced with $LiAlH_4$ as described previously to afford the N,N-dimethylamine which was converted to the hydrochloride in the solvent indicated in Table II.

6-Chloro-7-methoxy-2-(di-n-propylamino)tetralin Hydrochloride (34e). The base derived from 34c (4.96 g, 20 mmol) by treatment with 2 N NaOH followed by Et₂O extraction, drying $(MgSO_4)$, and concentration was heated with 20 mL of propionic anhydride at 100 °C for 2 h. The solution was poured into H₂O, and the mixture was extracted with CH_2Cl_2 . The organic extract was washed (5% aqueous NaHCO3, H2O, brine), dried (MgSO4), and concentrated. To a stirred solution of the residual Npropionamide in 2 mL of THF, under N₂, was added 30 mL of 1 M BH₃ in THF. The stirred mixture was refluxed for 4 h and cooled, and 2.5 N HCl was added until the evaporation of gas stopped. The mixture was heated at 100 °C, allowing the THF to evaporate. The resulting solid was collected. This hydrochloride was converted to its base, which was again propionylated and reduced in the same manner to give the N.N-dipropyl derivative. The resulting aqueous solution of hydrochloride was converted to the base by addition of 2 N NaOH. The base was extracted with EtOAc. The extracts were washed with H₂O, dried $(MgSO_4)$, and concentrated to afford the tertiary amine, which was converted into a hydrochloride in the solvent indicated in Table II.

6,8-Dichloro-7-methoxy-4-oxo-1,2,3,4-tetrahydro-2naphthoic Acid (35a). tert-Butyl hypochlorite (5.4 g, 50 mmol) was added to a suspension of 11.5 g (45 mmol) of **32** in 110 mL of HOAc. After being stirred at 60 °C for 20 h, the mixture was poured into 250 mL of ice-H₂O. The resulting solid was filtered and recrystallized from EtOH to give 10 g (77%) of colorless crystals: mp 184-185 °C; ¹H NMR (Me₂SO-d₆ + CDCl₃) δ 7.95 (s, 1 H, ArH (5)). Anal. (C₁₂H₁₀Cl₂O₄) C, H.

6,8-Dichloro-7-methoxy-1,2,3,4-tetrahydronaphthoic acid (**35b**) was prepared by hydrogenolysis of **35a** in the same manner as described for the preparation of **28c**. In this experiment it was necessary to extract the filtered catalyst with warm HOAc. The combined filtrates and HOAc extracts were diluted with ice-H₂O. The precipitated solid was filtered and recrystallized from MeOH to give 9.5 g (90%) of colorless crystals, mp 175 °C. Anal. ($C_{12}H_{12}Cl_2O_3$) C, H; Cl: calcd, 25.77; found, 25.15.

2-Amino-6,8-dichloro-7-methoxytetralin Hydrochloride $(34f).^{59}\,$ To a suspension of 2.2 g (8 mmol) of 35b and 0.9 mL of SOCl₂ in 60 mL of CH₂Cl₂ was added 2 drops of DMF. After being stirred at 25 °C for 4 h, the resulting solution was concentrated and the residue was stripped twice with toluene. The resulting acid chloride was dissolved in 12 mL of CH_2Cl_2 containing 8 mg of tetrabutylammonium bromide (TBAB). The solution was stirred at 0 °C as a solution of NaN_3 (0.65 g, 10 mmol) in H_2O (2 mL) was added dropwise. After the mixture was stirred at 0 °C for 1 h, additional CH_2Cl_2 was added. The CH_2Cl_2 solution was washed with H_2O and brine, and then it was concentrated in vacuo at 20 °C. A solution of the resulting carboxylic acid azide in 50 mL of toluene was refluxed for 1 h, and then it was concentrated in vacuo. The resulting isocyanate was dissolved in CH₂Cl₂ (30 mL) and TFA (0.84 mL) was added. After the solution was refluxed for 16 h, it was cooled to 25 °C and washed successively with a saturated solution of NaHCO₃, H₂O, and brine. The solution was dried (MgSO₄) and concentrated. The residual trifluoroacetamide was dissolved in a solution of 1.5 g of K₂CO₃ in 20 mL of MeOH. After the solution was refluxed for 1 h, it

no.	[³ H]fenoldopam rat striatum: K _{Bind} ± SE,ª µM (N)	dopamine-sens adenylate cyclase stim: ED ₅₀ , ^a µM	[3 H]spiroperidol, bovine pituitary: K_{Bind} , μ M (N) a	rabbit ear artery test: ED ₅₀ ± SE, ^a nM (K_B)
4a	0.074 (2)	1.6	0.52 ± 0.23 (3)	$2.6 \pm 0.8 (13)$
4b	0.84 ± 0.15 (3)	b	0.47 ± 0.19 (3)	$76.5 \pm 32 (10)$
4c	0.062 (2)	2.2	1.27 ± 0.30 (3)	4.5 ± 1.4 (6)
4d	0.043 ± 0.003 (3)	1.4	3.16 ± 0.82 (3)	$536 \pm 131 (43)$
4e	0.62 ± 0.17 (3)		0.24 ± 0.06 (3)	17 ± 2.3
4 f	0.145(2)		0.41(2)	
4g	0.27 ± 0.06 (3)		0.033 (2)	62 ± 44
4 h	0.37 (2)		0.66 ± 0.13 (3)	
4i	0.76 ± 0.12 (4)		0.47 (2)	
4j	1.74 ± 0.50 (3)		0.43 (2)	13 ± 1.1
4k	>50 (2)		14.11 ± 1.96 (3)	
41	1.44 (2)	Ь	1.05 (2)	С
4m	2.63 (2)	Ь	1.55(2)	
4n	4.71 (2)	Ь	1.63 ± 0.40 (3)	126 ± 48
40	2.97 (2)		$30 \pm 5\% @ 10 \ \mu M (4)$	
4p	>20 (2)		$29 \pm 3\% @ 10 \ \mu M \ (4)$	
4 q	>10 (2)	<i>b</i> .	$10.4 \pm 1.5 (3)$	
4 r	>10 (2)		$11.61 \pm 4.12 (3)$	
4s	5.95 (2)		3.5(2)	
4t	2.25 (2)		0.85 ± 0.06 (3)	53 ± 11
4u	1.97 ± 0.59 (3)	b	4.47 ± 0.91 (4)	
4v	2.30 (2)	b	0.09 ± 0.45 (3)	
4w	8.23 (2)	Ь	0.90 (2)	58 ± 17 (5)
dopamine	0.15	3.5	2.35	$37 \pm 6 (35)$
ADTN	0.07	2.7	0.21	$4.6 \pm 0.8 (10.5)$

^a See the Experimental Section for description of test and calculation of K_{Bind} , K_i , ED₅₀, and K_B values \pm standard error (SE). Although K_B values are presented for only selected compounds, all those for which an ED₅₀ is presented had values consistent with a D-2 action, i.e. between 5 and 50 nM vs. (S)-sulpiride. ^bStimulation and inhibition of adenylate cyclase was determined at two to six compound concentrations each done in triplicate in a single experiment. This compound did not stimulate dopamine-sensitive adenylate cyclase or inhibit stimulation in the presence of 50 μ M dopamine by >20%; therefore, it was considered inactive. ^cNo response at 1 μ M.

was filtered and the filtrate was concentrated. The residue was suspended in 20 mL of H_2O , and the mixture was extracted with Et_2O . After the Et_2O extracts were washed with H_2O and brine, they were concentrated. The residue was dissolved in MeOH- Et_2O and treated with HCl to give **34f** (Table II).

6,8-Dichloro-7-methoxy-2-(dimethylamino)tetralin Hydrochloride (34g). 48,52,53 The hydrochloride **34f** (1.2 g, 4.2 mmol) was converted to base in the usual fashion (NH₄OH, CH₂Cl₂). A solution of the base in 5 mL of MeOH was added slowly to a solution of 0.8 g (12.3 mmol) of NaCNBH₃ and 2.2 mL (0.021 mol) of 37% aqueous formaldehyde in 10 mL of MeOH. After the mixture was stirred at 25 °C for 2 h (following an initial slight exotherm), it was adjusted to pH 6 with AcOH and stirred for an additional 16 h. The solution was concentrated, and the residue was made alkaline with 2.5 N NaOH. The aqueous mixture was extracted with Et₂O. After being washed with H₂O and brine, the Et₂O extracts were dried (MgSO₄) and concentrated. The residual liquid in MeOH–Et₂O was treated with HCl to afford **34g** (Table II).

6,8-Dichloro-7-methoxy-2-(di-*n***-propylamino)tetralin hydrochloride (34h)** was prepared from the base derived from 34f in the same manner as described for conversion of 34c to 34e. The hydrochloride was prepared in the solvent indicated in Table II.

Ethyl 6-Methoxy-5-nitro-1,2,3,4-tetrahydronaphthoate (37a) and Ethyl 6-Methoxy-7-nitro-1,2,3,4-tetrahydronaphthoate (37b). To a stirred solution of 100 g (0.427 mol) of ethyl 1,2,3,4-tetrahydro-6-methoxynaphthoate (36)⁶³ in 220 mL of Ac₂O at -5 °C was added dropwise 27 mL of 70% HNO₃. After being stirred at -5 to 0 °C for 1 h, the mixture was held at -10 °C for 20 h. The crystalline 7-nitro isomer 37b was filtered and washed with EtOH to give 32.0 g (27%) of pale yellow crystals: mp 119-121 °C; ¹H NMR δ 6.8 (s, 1 H, ArH (5)), 7.7 (s, 1 H, ArH (8)). Anal. $(C_{14}H_{17}NO_5)$ C, H, N. The Ac₂O filtrate was poured into 500 mL of ice- H_2O . The liquid phase was decanted from the resulting viscous residue that was triturated with EtOH to give 17.5 g (15%) of yellow crystals of the 5-nitro isomer 37a, mp 92–95 °C, after recrystallization from EtOH: ¹H NMR δ 6.8 (d, 1 H, ArH (7)), 7.1 (d, 1 H, ArH (8)). Anal. (C₁₄H₁₇NO₅) C, H, N.

5-Chloro-6-methoxy-1,2,3,4-tetrahydronaphthoic Acid (38a). A mixture of 24.4 g (87.5 mmol) of 37a in 200 mL of HOAc and 0.9 g of 10% Pd/C catalyst was hydrogenated for 1 h at 25 °C and an initial pressure of 60 psi of H₂. After the mixture was filtered, the filtrate was concentrated to afford crude 5-amino derivative (16.9 g, 78%), mp 75-76 °C, after recrystallization from EtOH. This crude amino ester (10 g, 40 mmol) was suspended in 15 mL of H₂O and 60 mL of 12 N HCl. The mixture was heated until solution occurred. After the resulting solution was cooled to 0 °C, 4.1 g (60 mmol) of NaNO₂ in 30 mL of H₂O was added dropwise into the bottom of the reaction vessel while the temperature was maintained at 0 °C. After the reaction was stirred at 30 °C for 30 min, it was added slowly to a solution of 7.92 g (80 mmol) of freshly prepared CuCl⁶⁴ in 30 mL of 12 N HCl. The reaction mixture was stirred at 25 °C for 10 min and then at 100 °C for 1 h. The resulting mixture was diluted with 200 mL of H_2O and extracted with EtOAc. The extracts were washed (H_2O , brine), dried (MgSO₄), and concentrated to give 4.6 g (48%) of crystals, mp 195-197 °C. Anal. (C₁₂H₁₃ClO₃) C, H.

2-Amino-5-chloro-6-methoxytetralin hydrochloride (34i) was prepared from 38a in the same manner as that described for conversion of 35b to 34f (see Table II).

5-Chloro-6-methoxy-2-(dimethylamino)tetralin hydrochloride (34j) was obtained from 34i in the same fashion as that detailed for conversion of 34f to 34g (see Table II).

5-Chloro-6-methoxy-2-(di-*n*-propylamino)tetralin hydrochloride (34k) was derived from 34i by the same procedure as that described for conversion of 34c to 34e (see Table II).

7-Chloro-6-methoxy-1,2,3,4-tetrahydronaphthoic acid (38b) was prepared from 37b in the same manner as described for conversion of 37a to 38a. The intermediate 7-amino derivative (77% crude yield), mp 78–81 °C, gave 73% of crystals, mp 163–165 °C. Anal. ($C_{12}H_{13}ClO_3$) C, H, Cl.

2-Amino-7-chloro-6-methoxytetralin hydrochloride (341) was prepared from 38b by a procedure identical with that described for conversion of 35b to 34f (see Table II).

7-Chloro-6-methoxy-2-(dimethylamino)tetralin hydrochloride (34m) was obtained from 34l by a procedure identical with that detailed for conversion of the isocyanate derived from 33 to 34d (see Table II).

7-Chloro-6-methoxy-2-(di-*n*-propylamino)tetralin hydrochloride (34n) was prepared from 34l by a procedure identical with that employed for the conversion of 34c to 34e (see Table II).

Pharmacology. Competition for [³H]Fenoldopam in Rat Striatum.40 This assay performed with homogenized and washed membrane preparations from rat caudate nuclei was carried out as described previously.⁴⁰ In each experiment the amount of [³H]fenoldopam bound was determined in the absence (total) and presence (nonspecific) of 10^{-6} M (+)-butaclamol, the difference yielding specific [³H]fenoldopam binding. The ability of each compound to compete with [3H]fenoldopam (approximately 2.0 nM) was tested at concentrations of 10^{-7} and 10^{-6} M. If a compound displaced fenoldopam by 50% at a concentration of 10^{-6} M, it was considered to have significant activity and was further tested to obtain an IC_{50} for competition against fenoldopam. The K_{Bind} of a compound was calculated from the equation $K_{\text{Bind}} =$ $IC_{50}/(1 + L/K_D)$ where L is the concentration of [³H]fenoldopam and $K_{\rm D}$ is the equilibrium dissociation constant for fenoldopam $(2.3 \pm 0.1 \text{ nM}).$

Stimulation or inhibition of dopamine-sensitive adenylate cyclase was carried out as described previously.⁴⁰ EC_{50} values refer to the concentration of compound required to produce 50% of the maximum stimulation attainable with the test compound.

Competition for [³H]spiroperidol in bovine anterior pituitary was performed as described previously.⁴⁰ The ability of each compound to compete with [³H]spiroperidol (approximately 0.25 nM) was tested at concentrations of 10⁻⁵ and 10⁻⁷ M. Compounds displacing [³H]spiroperidol by 50% or more were tested further to obtain an IC₅₀. The K_{Bind} of a compound equals IC₅₀/(1 + L/K_{D}) where L is the concentration of [³H]spiroperidol and K_{D} is the dissociation constant for spiroperidol (0.3 nM).

Isolated Perfused Rabbit Ear Artery Test.^{65,66} This test was conducted as described perviously.^{65,66} The concentrationdependent inhibition of the constrictor response of the rabbit ear artery to brief intermittent periods of nerve stimulation was measured. The EC₅₀ is the concentration of drug required to produce 50% inhibition of the neuroeffector response (constriction) to nerve stimulation. Dissociation constants (K_B) for (S)-sulpiride, a selective DA₂ receptor antagonist, are expressed in nM concentrations and were determined according to the method of Furchgott.⁷²

Registry No. 4a, 103347-58-6; 4a·HBr, 103347-79-1; 4b, 103347-59-7; 4b·HBr, 103347-38-2; 4c, 103347-60-0; 4c·HBr, 103347-39-3; 4d, 103347-61-1; 4d·HBr, 103347-40-6; 4e, 103347-

Journal of Medicinal Chemistry, 1986, Vol. 29, No. 9 1627

62-2; 4e·HBr, 103347-41-7; 4f, 103347-63-3; 4f·HBr, 103347-42-8; 4g, 103366-72-9; 4g·HBr, 103347-43-9; 4h, 103347-64-4; 4h·HBr, 103347-44-0; 4i, 103347-65-5; 4i·HBr, 103347-45-1; 4j, 103366-73-0; 4j·HBr, 103347-46-2; 4k, 103347-66-6; 4k·HBr, 103347-47-3; 4l, 103347-67-7; 4l·HBr, 103347-48-4; 4m, 103347-68-8; 4m·HBr, 103366-71-8; 4n, 103347-69-9; 4n·HBr, 103347-49-5; 4o, 103347-70-2; 40·HBr, 103347-50-8; 4p, 103347-71-3; 4p·HBr, 103347-51-9; 4q, 103347-72-4; 4q·HBr, 103347-52-0; 4r, 103347-73-5; 4r·HBr, 67544-45-0; 4s, 103347-74-6; 4s-HBr, 103347-53-1; 4t, 103347-75-7; 4t·HBr, 103347-54-2; 4u, 103347-76-8; 4u·HBr, 103347-55-3; 4v, 103347-77-9; 4v·HBr, 103347-56-4; 4w, 103347-78-0; 4w·HBr, 103347-57-5; 5a, 78495-65-5; 5b, 6834-51-1; 6a, 103346-70-9; 6b, 103346-71-0; 7a (Y = OTs), 103346-72-1; 7b (Y = Br), 103346-73-2; 8a, 103346-74-3; 8b, 103346-75-4; 9a, 103346-76-5; 9b, 103346-77-6; 10a, 103346-78-7; 10b, 103346-79-8; 11a, 103346-80-1; 11a ethyl ester, 103346-81-2; 11a acid chloride, 103346-85-6; 11a acid azide, 103346-86-7; 11b, 103346-82-3; 11b ethyl ester, 103346-83-4; 12, 103346-89-0; 13, 103346-90-3; 14a·HCl, 103346-91-4; 14b·HCl, 103346-92-5; 15, 5417-17-4; 16, 103346-95-8; 17a, 103346-96-9; 17b, 103346-97-0; 18·HCl, 103346-98-1; 19·HCl, 103346-99-2; 20a, 103347-00-8; 20b-HCl, 103347-01-9; 21, 103347-02-0; 22, 331-42-0; 23, 345-08-4; 24a, 103347-03-1; 24b, 103347-04-2; 25-HCl, 103347-05-3; 26a·HCl, 103347-06-4; 26b, 103347-07-5; 27a, 103347-08-6; 27b, 103347-09-7; 28a, 103347-10-0; 28b, 103347-11-1; 28c, 103347-12-2; 29·HCl, 103347-13-3; 30, 103347-14-4; 31, 103347-15-5; 31 (triethyl ester), 103347-16-6; 32, 103347-17-7; 33, 103347-18-8; 33 (isocyanate), 103347-21-3; 34 (X = 5-F; $6,7-(MeO)_2$; $R^1 = H; R^2 = Cbz$, 103346-87-8; 34 (X = 6,7-(MeO)₂; 8-Cl; R¹ = H; R² = Cbz), 103347-80-4; 34a·HCl, 103346-84-5; 34b·HCl. 103346-88-9; 34c·HCl, 103347-19-9; 34d·HCl, 103347-20-2; 34e·HCl, 103347-22-4; 34f·HCl, 103347-25-7; 34g·HCl, 103347-26-8; 34h·HCl, 103347-27-9; 34i·HCl, 103347-31-5; 34i·HCl, 103347-32-6; 34k·HCl, 103347-33-7; 341·HCl, 103347-34-8; 34m·HCl, 103347-35-9; 34n·HCl, 103347-36-0; 35a, 103347-23-5; 35b, 103347-24-6; 36, 88285-43-2; 37a, 103347-28-0; 37a (5-amino deriv.), 103347-81-5; 37b, 103347-29-1; 38a, 103347-30-4; 38b, 103347-37-1; CH (CO₂Et)₂, 105-53-3; CO2MeCH2CH2CO2Me, 106-65-0; CH3CH2CHO, 123-38-6; 4-MeOC₆H₄CH₂COCl, 4693-91-8; 2-MeOC₆H₄F, 321-28-8; 4-MeO-3-F-C₆H₃-CO(CH₂)₂CO₂H, 347-63-7; NH(Pr)₂, 142-84-7; 3,4-dimethoxy-2-fluorobenzeneacetonitrile, 7537-08-8; 2-(2chloro-3.4-dimethoxybenzylidene)succinic acid dimethyl ester. 103346-93-6; 2-(2-chloro-3,4-dimethoxybenzyl)succinic acid dimethyl ester, 103346-94-7; succinic anhydride, 108-30-5; 2chloro-5-methylanisole, 73909-16-7; 2-chloro-5-methylphenol, 615-74-7; triethyl 1,1,2-ethanetricarboxylate, 7459-46-3; 6(or 7)-methoxy-8-fluoro-2-tetralone, 103347-83-7.