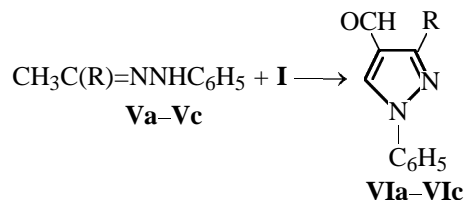




The product ratio **III** : **IV** varies from 50 : 1 for compounds **a** to 10 : 1 for compounds **b** and **c**. These data indicate that the electrophilic attack of **I** is mainly directed at the more electronegative carbon atom of the methylene group of hydrazones **IIa–IIc**. Formylation of **IIIa–IIIc** at the C<sup>5</sup> atom does not occur.

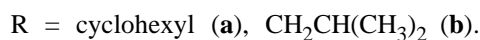
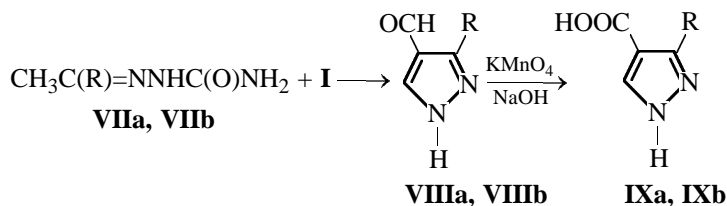
Phenylhydrazones **Va–Vc**, in which the methylene group at the C=N bond is either absent or shielded by a branched alkyl group, react differently. The major products are the corresponding 1-phenyl-3-alkylpyrazole-2-carbaldehydes **VIa–VIc**; thus, complex **I** mainly attacks the methyl group at the C=N bond.



The yield of pyrazolecarbaldehydes **VIa–VIc** was 72, 60, and 58%, respectively. Also detected were unchanged starting hydrazones and unidentified compounds with the structure differing from that of **III**.

Proceeding from the relationships revealed, we prepared pyrazolecarbaldehydes **VIIIa** and **VIIIb** con-

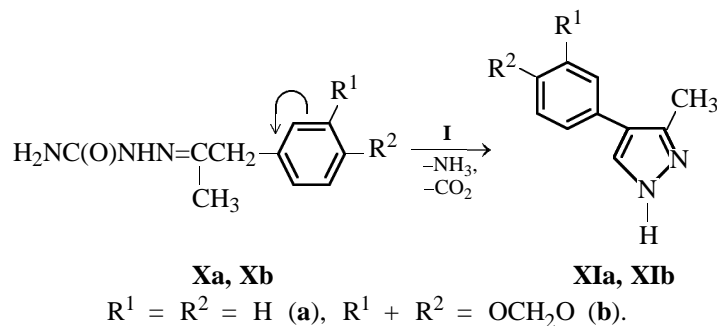
taining no substituent at the N atom by the reaction of complex **I** with semicarbazones derived from carbocyclic and aliphatic methyl ketones in which the methylene group at the C=N bond is either absent (**VIIa**) or sterically shielded (**VIIb**) compared to the methyl group.



Pyrazolecarbaldehydes **VIIIa** and **VIIIb**, which are readily soluble in alkalis, are readily oxidized into the corresponding pyrazole-4-carboxylic acids **IXa** and **IXb**.

As expected, under the conditions similar to those of preparation of **VIIIa** and **VIIIb**, semicarbazones **Xa**

and **Xb** derived from benzyl methyl ketones form the corresponding 3-methyl-4-arylpyrazoles **XIa** and **XIb**, which is due to the higher electron density on the methylene component in **Xa** and **Xb**, compared to the that on the methyl group, owing to the conjugation of the C=N bond with the benzene ring.



The structures of pyrazoles **IIIa–IIIc**, **IVa–IVc**, **VIa–VIc**, **VIIIa**, **VIIIb**, **IXa**, **IXb**, **XIa**, and **XIb** were confirmed by elemental analysis, mass spectrometry, and  $^1\text{H}$  NMR spectroscopy. It should be noted that the tautomerism of 3(5)-substituents in the diazole ring, typical of 1-unsubstituted pyrazoles and promoted by intermolecular N–H hydrogen bonding, is manifested only in **XIa** as characteristic splitting of the  $^1\text{H}$  NMR signals of the  $\text{CH}_3$ , CH, and NH groups of the diazole ring. The ratio of the 3- and 5- $\text{CH}_3$  isomers is 1.8 : 1.

## EXPERIMENTAL

The  $^1\text{H}$  NMR spectra were recorded on a Bruker AM-360 spectrometer (360.14 MHz, solvent  $\text{DMSO}-d_6$ ). The mass spectra were taken on a QP-5000 spectrometer (electron impact, 70 eV).

**Phenylhydrazones IIa–IIc and Va–Vc.** 2-Butanone, 2-pentanone, 2-hexanone, 4-methyl-2-pentanone, or 3-methyl-2-butanone was carefully mixed with an equimolar amount of freshly distilled phenylhydrazine. After the completion of the exothermic reaction, the mixture was cooled to  $45^\circ\text{C}$ , five drops of glacial acetic acid were added, and the mixture was stirred for 1 h on a boiling water bath. Then the mixture was cooled, a small amount of aqueous ethanol was added, the aqueous layer was separated, and the residue was distilled at a reduced pressure in a nitrogen flow to obtain compounds **IIa–IIc**, **Va**, and **Vb**. Below are the compound nos., bp [ $^\circ\text{C}$  (*p*, mm Hg)],  $n_D^{20}$ , and yield (%): **IIa**, 115–116 (3), 1.5742, 90; **IIb**, 130–133 (3), 1.5637, 80; **IIc**, 141–144 (3), 1.5530, 83; **Va**, 140–142 (3), 1.5494, 95; **Vb**, 102–104 (2), 1.5600, 93.

Compound **Vc** was obtained in 85% yield by dropwise addition of phenylhydrazine to an alcoholic solution of 1-adamantyl methyl ketone in the presence of several drops of glacial acetic acid, followed by refluxing for 2 h, cooling, filtration, washing with ethanol, and drying in a vacuum (decomposition point  $198\text{--}200^\circ\text{C}$ ).

**Semicarbazones VIIa, VIIb, Xa, and Xb.** A mixture of 100 g of semicarbazide hydrochloride and 100 g of anhydrous sodium acetate was thoroughly ground in a porcelain mortar, suspended in 1 l of isopropyl alcohol, refluxed for 30 min, and filtered while hot. A 0.8-mol portion of methyl cyclohexyl ketone, 4-methyl-2-pentanone, methyl benzyl ketone, or piperonyl methyl ketone was added to the refluxing mother liquor. The mixture was refluxed for 1 h, cooled to room temperature, allowed to stand for 10 h, and filtered. The precipitate on the filter was washed with cold isopropyl alcohol and dried at  $100^\circ\text{C}$ . Semicarbazones **VIIa**, **VIIb**, **Xa**, and **Xb** were obtained in 90% yield.

## Reactions of phenylhydrazones IIa–IIc with I.

A 153.5-g portion of freshly distilled  $\text{POCl}_3$  was added dropwise at  $0^\circ\text{C}$  to 220 g of anhydrous DMF; the mixture was left for 20 min. Under external cooling, 0.5 mol of freshly distilled phenylhydrazone **IIa–IIc** was slowly added dropwise, avoiding warming-up of the mixture above  $50^\circ\text{C}$ . After adding the whole amount of **IIa–IIc**, the mixture was stirred for 2 h at  $80^\circ\text{C}$  and poured onto 0.5 kg of ice. The resulting mixture was alkalinized with 30% NaOH to pH 8–9, left for 20 min, and neutralized with 20% HCl to pH 6–7. The mixture was extracted with chloroform ( $3 \times 180$  ml). After removing the solvent, the mixture was analyzed by  $^1\text{H}$  NMR spectroscopy and distilled in a vacuum with a long Vigreux column. Yields: **IIIa** 43 g (50%); **IIIb** 56 g (61%) + **IVb** 4.9 g (6%); **IIIc** 65 g (65%) + **IVc** 6.1 g (6.4%).

**1-Phenyl-3,4-dimethylpyrazole IIIa:** bp  $116\text{--}118^\circ\text{C}$  (3 mm Hg),  $n_D^{20}$  1.5860. Mass spectrum,  $m/z$ : 172 [ $M$ ] $^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 2.02 s (3H,  $\text{CH}_3$ ), 2.17 s (3H,  $\text{CH}_3$ ), 7.20 t (1H, CH), 7.43 t (2H, 2CH), 7.73 d (2H, 2CH), 8.13 s (1H, CHN). Found, %: C 76.59; H 7.04.  $\text{C}_{11}\text{H}_{12}\text{N}_2$ . Calculated, %: C 76.71; H 7.02.

**1-Phenyl-3-methyl-4-ethylpyrazole IIIb:** bp  $140\text{--}142^\circ\text{C}$  (4 mm Hg),  $n_D^{20}$  1.5750. Mass spectrum,  $m/z$ : 186 [ $M$ ] $^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.17 t (3H,  $\text{CH}_3$ ), 2.18 s (3H,  $\text{CH}_3$ ), 2.42 q (2H,  $\text{CH}_2$ ), 7.21 t (1H, CH), 7.43 t (2H, 2CH), 7.75 d (2H, 2CH), 8.16 s (1H, CHN). Found, %: C 77.30; H 7.60.  $\text{C}_{12}\text{H}_{14}\text{N}_2$ . Calculated, %: C 77.38; H 7.58.

**1-Phenyl-3-methyl-4-propylpyrazole IIIc:** bp  $148\text{--}150^\circ\text{C}$  (3 mm Hg),  $n_D^{20}$  1.5664. Mass spectrum,  $m/z$ : 200 [ $M$ ] $^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.08 t (3H,  $\text{CH}_3$ ), 1.30–1.58 m (2H,  $\text{CH}_2$ ), 2.19 s (3H,  $\text{CH}_3$ ), 2.38 q (2H,  $\text{CH}_2$ ), 7.20 t (1H, CH), 7.44 t (2H, 2CH), 7.78 d (2H, 2CH), 8.15 s (1H, CHN). Found, %: C 78.02; H 8.02.  $\text{C}_{13}\text{H}_{16}\text{N}_2$ . Calculated, %: C 77.96; H 8.05.

**1-Phenyl-3-ethylpyrazole-4-carbaldehyde IVa.** Yield according to  $^1\text{H}$  NMR spectrum 1%.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.28 t (3H,  $\text{CH}_3$ ), 2.90 q (2H,  $\text{CH}_2$ ), 7.06 t (1H, CH), 7.52 t (2H, 2CH), 7.87 d (2H, 2CH), 9.14 s (1H, CHN), 9.95 s (1H, CHO).

**1-Phenyl-3-propylpyrazole-4-carbaldehyde IVb:** bp  $151\text{--}153^\circ\text{C}$  (4 mm Hg),  $n_D^{20}$  1.5840. Mass spectrum,  $m/z$ : 214 [ $M$ ] $^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.96 t (3H,  $\text{CH}_3$ ), 1.69 m (2H,  $\text{CH}_2$ ), 2.85 t (2H,  $\text{CH}_2$ ), 7.30 t (1H, CH), 7.52 t (2H, 2CH), 7.86 d (2H, 2CH), 9.14 s (1H, CHN), 9.93 s (1H, CHO). Found, %: C 72.82; H 6.60.  $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}$ . Calculated, %: C 72.87; H 6.59.

**1-Phenyl-3-butylpyrazole-4-carbaldehyde IVc:** bp 160–162°C (3 mm Hg),  $n_D^{20}$  1.5790. Mass spectrum,  $m/z$ : 228  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.94 t (3H,  $\text{CH}_3$ ), 1.25–1.40 m (2H,  $\text{CH}_2$ ), 1.60–1.80 m (2H,  $\text{CH}_2$ ), 2.83 t (2H,  $\text{CH}_2$ ), 7.36 t (1H, CH), 7.49 t (2H, 2CH), 7.87 d (2H, 2CH), 9.15 s (1H, CHN), 9.94 s (1H, CHO). Found, %: C 73.51; H 7.14.  $\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}$ . Calculated, %: C 73.66; H 7.06.

**Reactions of phenylhydrazones Va–Vc with I.** A 153.5-g portion of freshly distilled  $\text{POCl}_3$  was added dropwise at 0°C to 220 g of anhydrous DMF; the mixture was left for 20 min. Under external cooling, 0.5 mol of freshly distilled phenylhydrazone was slowly added dropwise (for **Va**, **Vb**) or in portions (for **Vc**), avoiding warming-up of the mixture above 55°C. After adding the whole amount of **Va–Vc**, the mixture was stirred for 2 h at 80°C and poured while hot onto 0.5 kg of ice. The resulting mixture was alkalinized with 30% NaOH to pH 8–9, left for 20 min, and neutralized with 20% HCl to pH 7. The mixture was extracted with chloroform (3 × 180 ml). After removing the solvent, the mixture was analyzed by  $^1\text{H}$  NMR spectroscopy and distilled in a vacuum with a Vigreux column (for **VIa**, **VIb**). Compound **VIb** slowly crystallized, after which it was recrystallized from hexane. Pyrazole **VIa** was purified by additional vacuum distillation. Crude pyrazole **VIc** was dissolved in pure chloroform and treated with a solution of sodium bisulfite at 50°C. The resulting precipitate of the bisulfite derivative was filtered off, washed with hot chloroform, dried, treated with 20%  $\text{H}_2\text{SO}_4$ , and extracted with chloroform (2 × 150 ml). After removing the solvent, pyrazole **VIc** was recrystallized from isopropyl alcohol. Yields: **VIa** 82.1 g (72%), **VIb** 64.2 g (60%), and **VIc** 88.7 g (58%).

**1-Phenyl-3-isobutylpyrazole-4-carbaldehyde VIa:** bp 173–174°C (4 mm Hg),  $n_D^{20}$  1.5815. Mass spectrum,  $m/z$ : 228  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.92 d (6H, 2 $\text{CH}_3$ ), 1.97–2.09 m (1H, CH), 2.77 d (2H,  $\text{CH}_2$ ), 7.39 t (1H, CH), 7.53 t (2H, 2CH), 7.90 d (2H, 2CH), 9.16 s (1H, CHN), 9.95 s (1H, CHO). Found, %: C 73.63; H 7.05.  $\text{C}_{14}\text{H}_{16}\text{N}_2\text{O}$ . Calculated, %: C 73.66; H 7.06.

**1-Phenyl-3-isopropylpyrazole-4-carbaldehyde VIb:** bp 140–150°C (2 mm Hg), mp 58–59°C. Mass spectrum,  $m/z$ : 214  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.28 d (6H, 2 $\text{CH}_3$ ), 3.37–3.49 m (1H, CH), 7.37 t (1H, CH), 7.52 t (2H, 2CH), 7.84 d (2H, 2CH), 9.14 s (1H, CHN), 9.95 s (1H, CHO). Found, %: C 72.83; H 6.57.  $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}$ . Calculated, %: C 72.87; H 6.59.

**1-Phenyl-3-(1'-adamantyl)pyrazole-4-carbaldehyde VIc:** mp 116–118°C. Mass spectrum,  $m/z$ : 306  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.75–2.20 m (15H,

$\text{C}_{10}\text{H}_{15}$ ), 7.30 t (1H, CH), 7.43 t (2H, 2CH), 7.68 d (2H, 2CH), 8.40 s (1H, CHN), 10.18 s (1H, CHO). Found, %: C 78.35; H 7.28.  $\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}$ . Calculated, %: C 78.40; H 7.24.

**Reactions of semicarbazones VIIa, VIIb, Xa, and Xb with I.** A 153.5-g portion of freshly distilled  $\text{POCl}_3$  was added dropwise at 0°C to 220 g of anhydrous DMF; the mixture was left for 20 min. Under external cooling, 0.5 mol of freshly distilled **VIIa**, **VIIb**, **Xa**, or **Xb** was slowly added dropwise in portions, avoiding warming-up of the mixture above 50°C. After adding the whole amount of semicarbazones, the mixture was stirred for 1 h at 80°C and poured while hot onto 0.5 kg of ice. The resulting mixture was alkalinized with 30% NaOH to pH 8–9, left for 20 min, and neutralized with 20% HCl to pH 7. In the case of **VIIa** and **VIIb**, the mixture was extracted with chloroform (3 × 250 ml). After removing the solvent, the residue was analyzed by  $^1\text{H}$  NMR spectroscopy and recrystallized from hexane (for **VIIa**) or immediately subjected to oxidation (for **VIIb**). In formylation of **Xa** and **Xb**, the precipitates formed upon neutralization were filtered off and recrystallized from water. Yields: **VIIIa** 59.7 g (70%), **VIIIb** ~45 g (62%), **XIa** 26.55 g (35%), and **XIb** 38.8 g (40%).

**3-Cyclohexylpyrazole-4-carbaldehyde VIIIa:** mp 111–112°C. Mass spectrum,  $m/z$ : 178  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.15–1.40 m (3H,  $\text{C}_6\text{H}_{11}$ ), 1.45–1.58 q (2H,  $\text{C}_6\text{H}_{11}$ ), 1.66–1.74 d (1H,  $\text{C}_6\text{H}_{11}$ ), 1.74–1.86 t (4H,  $\text{C}_6\text{H}_{11}$ ), 3.12 t (1H,  $\text{C}_6\text{H}_{11}$ ), 8.05 br (1H, CHN), 9.86 s (1H, CHO), 13.15 br (1H, NH). Found, %: C 67.30; H 7.96.  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}$ . Calculated, %: C 67.39; H 7.92.

**3-Isobutylpyrazole-4-carbaldehyde VIIIb.**  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.85 d (6H, 2 $\text{CH}_3$ ), 1.95 m (1H, CH), 2.73 d (2H,  $\text{CH}_2$ ), 7.95 s (1H, CHN), 9.83 s (1H, CHO), 13.25 br (1H, NH).

**3-Methyl-4-phenylpyrazole XIa:** mp 144–145°C. Mass spectrum,  $m/z$ : 158  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 2.32 and 2.38 d (3H,  $\text{CH}_3$ ), 7.22 t (1H, CH), 7.38 t (2H, 2CH), 7.43 d (2H, 2CH), 7.67 and 7.91 d (1H, CHN), 12.65 d (1H, NH). Found, %: C 75.90; H 6.36.  $\text{C}_{10}\text{H}_{10}\text{N}_2$ . Calculated, %: C 75.92; H 6.37.

**3-Methyl-4-(3',4'-methylenedioxyphenyl)pyrazole XIb:** mp 97–99°C. Mass spectrum,  $m/z$ : 202  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 2.31 s (3H,  $\text{CH}_3$ ), 6.00 s (2H,  $\text{OCH}_2\text{O}$ ), 6.90 q (2H, 2CH), 7.00 s (1H, CH), 7.67 s (1H, CHN). Found, %: C 65.24; H 4.92.  $\text{C}_{11}\text{H}_{10}\text{N}_2\text{O}_2$ . Calculated, %: C 65.34; H 4.98.

**Oxidation of aldehydes VIIIa and VIIIb.** Aldehyde **VIIIa** (23.32 g) or a reaction mixture from synthesis of **VIIIb** (70% main substance, 26.55 g,

0.131 mol of **VIIIb**) was dissolved in 500 ml of water containing 25 g of NaOH. The mixture was cooled to 15°C, and a solution of 17 g of  $\text{KMnO}_4$  in 500 ml of water was quickly added. The mixture was stirred for 30 min and then heated to 96–98°C to complete decolorization of the solution. The solution after cooling was filtered to remove  $\text{MnO}_2$  and acidified with HCl to pH 3 (for **IXa**) or 6 (for **IXb**). The precipitate that formed was filtered off, washed with water (as far as possible), and dried at 110°C, after which it was washed and dried again. Yields: **IXa** 23.3 g (92%) and **IXb** 22.1 g (90%); colorless crystalline substances.

**3-Cyclohexylpyrazole-4-carboxylic acid IXa:** mp > 270°C. Mass spectrum,  $m/z$ : 194  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.15–1.33 m (3H,  $\text{C}_6\text{H}_{11}$ ), 1.35–1.55 q (2H,  $\text{C}_6\text{H}_{11}$ ), 1.65–1.75 d (1H,  $\text{C}_6\text{H}_{11}$ ), 1.75–1.85 t (4H,  $\text{C}_6\text{H}_{11}$ ), 3.25 t (1H,  $\text{C}_6\text{H}_{11}$ ), 7.82 s (1H, CHN), 12.5 br (2H, NH and COOH). Found, %: C

61.80; H 7.24.  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}_2$ . Calculated, %: C 61.84; H 7.26.

**3-Isobutylpyrazole-4-carboxylic acid IXb:** mp > 270°C. Mass spectrum,  $m/z$ : 168  $[M]^+$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.85 d (6H,  $2\text{CH}_3$ ), 1.90–2.04 m (1H, CH), 2.72 d (2H,  $\text{CH}_2$ ), 7.80 s (1H, CHN), 12.50 br (2H, NH and COOH). Found, %: C 57.10; H 7.15.  $\text{C}_8\text{H}_{12}\text{N}_2\text{O}_2$ . Calculated, %: C 57.13; H 7.19.

## REFERENCES

1. Kira, M.A. and Bruckner-Wilhelm, A., *Acta Chim. Acad. Sci. Hung.*, 1968, vol. 56, p. 47.
2. Kira, M.A., Bruckner-Wilhelm, A., and Ruff, F., *Acta Chim. Acad. Sci. Hung.*, 1968, vol. 56, p. 189.
3. Kira, M.A., Abdel-Raeman, M.O., and Gadalla, K.Z., *Tetrahedron Lett.*, 1969, no. 2, p. 109.
4. Kira, M.A., Aboul-Enein, M.N., and Korkor, M.I., *J. Heterocyclic Chem.*, 1970, vol. 7, no. 1, p. 25.