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## Graphical Abstract

Synthesis of the spiroacetal core of the
Leave this area blank for abstract info. cephalosporolide family of natural products
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penisporolide core

$\xrightarrow{\text { DHQD }}$

cephalosporolide core

# Synthesis of the spiroacetal core of the cephalosporolide family of natural products 

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ABSTRACT

The synthesis of four possible stereoisomers of the spiroacetal core of the natural products cephalosporolides H and I and penisporolides A and B is described. The key steps involve the use of Sharpless asymmetric dihydroxylation to install the desired stereochemistry of the $\gamma$-lactone ring and an oxidative radical cyclisation to form the spiroacetal ring system.

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## Keywords:

Cross metathesis;
Sharpless asymmetric dihydroxylation;
Oxidative radical cyclisation;
Spiroacetals;
Cephalosporolides.

[^0]
## 1. Introduction

The 5,5 -spiroacetal fused $\gamma$-lactone moiety is a structural feature of several biologically relevant natural products, including cephalosporolides $\mathrm{E} \quad$ (1) and $\mathrm{F} \quad$ (2), ${ }^{1-3}$ cephalosporolides H(3) and I (4), ${ }^{4}$ penisporolides A (5) and B $(6)^{5}$ and ascospiroketal $\mathrm{B}(7)^{6}$ (Figure 1). The simplest members of this family are cephalosporolides E (1) and F (2), which possess a methyl group at C-9 of the spiroacetal and differ only in the stereochemistry at the C-6 spirocentre. Cephalosporolides H (3) and I (4) were isolated in 2007 from
the marine-derived fungus, Penicillium sp. ${ }^{4}$ Both cephalosporolides H (3) and I (4) possess a gem-dimethyl group $\alpha$ to the $\gamma$-lactone and differ only in the substituent at C 9. Cephalosporolide $H(3)$ has a saturated heptyl side chain whereas cephalosporolide I (4) has a four carbon chain bearing a terminal carboxylic acid. Preliminary biological testing on these natural products demonstrated inhibition of the enzymes $3 \alpha$-hydroxysteroid dehydrogenase and xanthine oxidase, which implicates these natural products as potential antiinflammatory agents. ${ }^{7-9}$

cephalosporolide H (3)

cephalosporolide I (4)

cephalosporolide F (2)

ascospiroketal B (7)

Figure 1: Natural products possessing the 5,5 -spiroacetal fused $\gamma$-lactone structural moiety. ${ }^{1,4-6}$

Penisporolides A (5) and B (6) were isolated from the marine-derived Penicillium sp. ${ }^{5}$ and possess similar structural features to cephalosporolides $\mathrm{H}(\mathbf{3})$ and I (4). They differ in the C-3 and C-4 stereochemistry of the $\gamma$-lactone and the sidechain at C-9. Penisporolide A (5) possesses a 5-hydroxyheptyl side chain at C-9 whereas penisporolide B (6) bears a 3-oxohexyl substituent. The structures of these natural products as well as cephalosporolides $H(3)$ and I (4) were tentatively proposed based on comparison of their spectroscopic data and to the spectroscopic data reported for the confirmed structures of cephalosporolides E (1) and F (2). ${ }^{10}$ As a result of our interest in the total synthesis of spiroacetalcontaining natural products and our recent synthesis of cephalosporolides $\mathrm{E}(\mathbf{1})$ and $\mathrm{F}(\mathbf{2}){ }^{11}$ we extended our work to the synthesis of cephalosporolides $\mathrm{H}(\mathbf{5})$ and $\mathrm{I}(\mathbf{6})$.

Recent syntheses of cephalosporolide H (3) by Dudley et al. ${ }^{12-14}$ and Fernandes et al. ${ }^{15}$ have suggested that the proposed stereochemistry of the spiroacetal centre should be revised. With this question in mind, we devised a synthetic strategy to enable the synthesis of each of the four natural products: cephalosporolides $\mathrm{H}(\mathbf{3})$ and $\mathrm{I}(\mathbf{4})$ and penisporolides $\mathrm{A}(\mathbf{5})$ and B (6) from a common spiroacetal precursor.

## 2. Results and discussion

Our retrosynthesis (Scheme 1) allows for the preparation of two diastereomers of the common spiroacetal intermediate $\mathbf{8}$, that upon debenzylation and further side-chain extension should afford both natural products cephalosporolides H (5) and I (6). Oxidative radical cyclisation of hydroxyl-lactone 9 would provide spiroacetal 8. The lactone is accessible via a Sharpless asymmetric dihydroxylation of olefin $\mathbf{1 0}$ with concomitant lactonisation. The latter can be assembled via a cross metathesis reaction between olefins 11 and $\mathbf{1 2}$, which are
obtained from commercially available glycidol $\mathbf{1 3}$ and alcohol 14, respectively.


Scheme 1: Retrosynthesis of cephalosporolide H (3) and I (4)
A similar strategy for the synthesis of penisporolides A (5) and $B$ (6) from spiroacetal precursor 15, could also be employed simply by changing the Sharpless ligand used for the
asymmetric dihydroxylation of olefin $\mathbf{1 0}$ (Scheme 2). Oxidative radical cyclisation of lactone $\mathbf{1 6}$ then affords the two diastereomers of the spiroacetal precursor 15. Sharpless asymmetric dihydroxylation of olefin $\mathbf{1 0}$ enables installation of the desired stereochemistry at $\mathrm{C}-3$ and $\mathrm{C}-4$.


$(\mathrm{DHQ})_{2} \mathrm{PHAL}$
10

$\downarrow$


Scheme 2: Retrosynthesis of penisporolides A (5) and B (6)
Implementation of the synthetic strategy planned for cephalosporolides H (3) and I (4) summarized in Scheme 1 initially required preparation of olefin $\mathbf{1 2}$ in 5 steps from commercially available glycidol $\mathbf{1 3}$. Glycidol $\mathbf{1 3}$ was protected as a benzyl ether and treated with $(R, R)$-salen $\mathrm{Co}^{\mathrm{II}}$ complex $\mathbf{A}$ affording enantioenriched ( $R$ )-epoxide $\mathbf{1 7}$ which could easily be separated from the ( $S$ )-diol by column chromatography (Scheme 3). ${ }^{16}$ The newly formed epoxide 17 was opened at the terminal position by the addition of an allyl cuprate in THF to afford secondary alcohol 18. ${ }^{17}$ Ozonolysis of alkene 18 furnished lactol $\mathbf{1 9}$ as a 1:1 mixture of cis/trans isomers. This mixture was then treated with allyltrimethylsilane and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ to install the desired allyl side-chain, providing 12 as a $1: 1$ mixture of cis and trans isomers in $83 \%$ yield. ${ }^{18}$


Scheme 3: Reagents and conditions: a) $\mathrm{BnBr}, \mathrm{NaH}, \mathrm{DMF}, 0^{\circ} \mathrm{C}$ to rt, $18 \mathrm{~h}, 84 \%$; b) AcOH, THF, $\mathrm{H}_{2} \mathrm{O},(R, R)$-salen $\mathrm{Co}^{\text {II }}$ complex $\mathbf{A}, 0$ ${ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 16 \mathrm{~h}, 48 \%$; c) CuI, allyl $\mathrm{MgBr}, \mathrm{THF},-40^{\circ} \mathrm{C}$ to rt, 3 h , $70 \%$; d) $\mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{MeOH}, \mathrm{PPh}_{3},-78{ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 15 \mathrm{~h}, 71 \%$, 1:1 cis:trans mixture; e) allyltrimethylsilane, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}, \mathrm{rt}$, $3 \mathrm{~h}, 83 \%, 1: 1$ cis:trans mixture.

With olefin 12 in hand, attention turned to the synthesis of olefin 11, the cross metathesis coupling partner (Scheme 4). Commercially available ester $\mathbf{1 4}$ was deprotected to give the carboxylic acid which was converted to benzyl ester
20._ENREF_15 ${ }^{19}$ Swern oxidation of alcohol 20 to the corresponding aldehyde ${ }^{20}$ followed by Wittig olefination yielded olefin 11.


Scheme 4: Reagents and conditions: a) $\mathrm{LiOH}, \mathrm{THF}, \mathrm{MeOH}, \mathrm{H}_{2} \mathrm{O}$, $0{ }^{\circ} \mathrm{C}$ to rt, $1.5 \mathrm{~h}, 73 \%$; b) BnBr, DMF, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{rt}, 5 \mathrm{~h}, 86 \%$; c) DMSO, $\mathrm{CH}_{2} \mathrm{Cl}_{2},\left(\mathrm{COCl}_{2}, \mathrm{NEt}_{3},-7{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}, 82 \%\right.$; d) LiHMDS, $\mathrm{MePh}_{3} \mathrm{P}^{+} \mathrm{Br}^{-}, \mathrm{THF}, 0^{\circ} \mathrm{C}$ to rt, $16 \mathrm{~h}, 46 \%$.

Armed with coupling partners $\mathbf{1 1}$ and 12, attention turned to the cross metathesis reaction (Scheme 5). Grubbs' II catalyst was employed due to its known propensity for trans selectivity in cross metathesis reactions involving alkenes with a fully substituted $\alpha$-quaternary centre. ${ }^{21}$ Using a variety of conditions only moderate yields of cross-coupled olefin $\mathbf{1 0}$ were obtained. The reaction was carried out in the microwave in the presence of chlorodicyclohexyl borane (catalytic quantity) and $10 \mathrm{~mol} \%$ Grubbs' II in toluene at $90^{\circ} \mathrm{C}$ for 9 h to give the desired $(E)$ olefin 10 in $22 \%$ yield. ${ }^{*}$


Scheme 5: Reagents and conditions: a) Grubbs' II, toluene, $\mathrm{cy}_{2} \mathrm{BCl}, 90^{\circ} \mathrm{C}$, microwave, $9 \mathrm{~h}, 22 \%$.

Olefin 10 was next subjected to Sharpless asymmetric dihydroxylation using AD-mix $\beta$ to afford lactones 9 a ( $\alpha_{\mathrm{D}}$ $\left.+12.1, c 1.0 \mathrm{CHCl}_{3}\right)$ and $9 \mathrm{~b}\left(\alpha_{\mathrm{D}}+7.0, c 0.3 \mathrm{CHCl}_{3}\right)$ as a separable 1:1 mixture of diastereomers (Scheme 6). The newly formed lactones exhibit the cis stereochemistry between C-3 and C-4 as desired for cephalosporolides H 5 and I 6 and differ only in the stereochemistry at C-6.


Scheme 6: Reagents and conditions: a) ( DHQD$)_{2} \mathrm{PHAL}$, methanesulfonamide, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{~K}_{3} \mathrm{Fe}(\mathrm{CN})_{6}, \mathrm{OsO}_{4}, t$ - $\mathrm{BuOH}, \mathrm{H}_{2} \mathrm{O}$, rt, $15 \mathrm{~h}, 73 \%, 9 \mathrm{a}: 9 \mathbf{b}$ (1:1).

Sharpless asymmetric dihydroxylation of olefin $\mathbf{1 0}$ was also carried out using AD-mix $\alpha$ furnishing a 1:1 separable mixture of lactones 16a $\left(\alpha_{D}-35.7, c 1.0 \mathrm{CHCl}_{3}\right)$ and 16b $\left(\alpha_{D}-18.0, c 0.3\right.$

[^1]$\mathrm{CHCl}_{3}$ ) (Scheme 7). The lactones formed in this reaction possess the desired stereochemistry at C-3 and C-4 for penisporolides $A(3)$ and $B(4)$.


Scheme 7: Reagents and conditions: a) (DHQ) ${ }_{2} \mathrm{PHAL}$, methanesulfonamide, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{~K} 3 \mathrm{Fe}(\mathrm{CN})_{6}, \mathrm{OsO}_{4}, t-\mathrm{BuOH}, \mathrm{H}_{2} \mathrm{O}$, rt, $15 \mathrm{~h}, 76 \%$, 16a:16b (1:1).

The key oxidative radical cyclisation reaction was separately carried out on lactones 9a and 9b (Scheme 8). Treatment of lactone $9 \mathbf{a}$ with iodosobenzene diacetate and iodine afforded a separable $1: 1$ mixture of spiroacetals $\mathbf{8 a}\left(\alpha_{D}\right.$ $\left.+42.0, c 0.26 \mathrm{CHCl}_{3}\right)$ and $\mathbf{8 b}\left(\alpha_{\mathrm{D}}-2.9, c 1.0 \mathrm{CHCl}_{3}\right)$. The same 1:1 ratio of spiroacetals $\mathbf{8 a}$ and $\mathbf{8 b}$ was obtained when lactone 9 b was employed in the oxidative cyclisation reaction.




Scheme 8: Reagents and conditions: a) $\mathrm{PhI}(\mathrm{OAc})_{2}, \mathrm{I}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, hexane, $75 \mathrm{~W}, 1 \mathrm{~h}, 51 \%$, 8a:8b (1:1); b) $\mathrm{PhI}(\mathrm{OAc})_{2}, \mathrm{I}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, hexane, $75 \mathrm{~W}, 1 \mathrm{~h}, 52 \%, 8 \mathrm{a}: 8 \mathrm{~b}$ (1:1).

The oxidative cyclisation reaction was also carried out on lactone 16a affording a $1: 1$ mixture of spiroacetals $\mathbf{1 5 a}\left(\alpha_{D}\right.$ $\left.-41.2, c \quad 0.4 \mathrm{CHCl}_{3}\right)$ and 15b $\left(\alpha_{\mathrm{D}}-20.4, c \quad 0.25 \mathrm{CHCl}_{3}\right)$. Similarly, a $1: 1$ mixture of spiroacetals $\mathbf{1 5 a}$ and $\mathbf{1 5 b}$ was obtained when the same procedure was carried out on lactone 16b.


Figure 2: Comparison of H-4 NMR of the known natural products with the four stereoisomers synthesised. ${ }^{10-12}$



Scheme 9: Reagents and conditions: a) $\mathrm{PhI}(\mathrm{OAc})_{2}, \mathrm{I}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, hexane, $75 \mathrm{~W}, 1 \mathrm{~h}, 51 \%, 15 \mathrm{a}: 15 \mathrm{~b}(1: 1)$; b) $\mathrm{PhI}(\mathrm{OAc})_{2}, \mathrm{I}_{2}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, hexane, $75 \mathrm{~W}, 1 \mathrm{~h}, 51 \%$, 15a:15b (1:1).

Recently, the stereochemistry of cephalosporolide H 3a and its C-6 epimer 3 (Figure 2), synthesised by Dudley et al. ${ }^{12-14}$ and Fernandes et al. ${ }^{15}$ was determined by comparison of the ${ }^{1} \mathrm{H}$ NMR data reported for cephalosporolides E (1) and F (2) and their enantiomers 21 and 22 for which X-ray crystallographic data was available (Figure 2). ${ }^{10-11}$ The reassignment of the stereochemistry of the isolated cephalosporolide $\mathrm{H}(\mathbf{3})$ to that of the synthetic 3a was based primarily on $\alpha_{D}$ values and comparison of the characteristic coupling pattern for $\mathrm{H}-4$ observed in ent-cephalosporolides E (21) and F (22).

The same approach was used to determine the stereochemistry of the spirocentre in spiroacetals $\mathbf{8 a}, \mathbf{8 b}, \mathbf{1 5 a}$ and 15b. Given that the C-4 stereochemistry in these four diastereomeric spiroacetals was installed unequivocally via Sharpless asymmetric dihydroxylation, the characteristic H-4 coupling pattern could then be used to determine the stereochemistry of the spirocentre (Figure 2 and 3). In spiroacetal 8a, H-4 resonated as a triplet at $\delta 5.06 \mathrm{ppm}$ with a coupling constant, $J 4.8 \mathrm{~Hz}$, establishing ( $R$ )-stereochemistry at the spirocentre. Conversely, H-4 in spiroacetal $\mathbf{8 b}$ resonates as a doublet of doublet of doublets at $\delta 5.02 \mathrm{ppm}$ with coupling constant, $J 6.6,3.7,1.6 \mathrm{~Hz}$ establishing ( $S$ )-stereochemistry at the spirocentre.

Similarly, spiroacetals $\mathbf{1 5 a}$ and $\mathbf{1 5 b}$ were assigned as $6 S$ and $6 R$ based on comparison of the ${ }^{1} \mathrm{H}$ NMR coupling patterns observed for $\mathrm{H}-4$ to the same resonance in cephalosporolides E (1) and F (2).


Figure 3: Characteristic H-4 coupling pattern for spiroacetals 8a, $\mathbf{8 b}, \mathbf{1 5 a}$ and $\mathbf{1 5 b}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right)$

The cephalosporolide family poses some interesting questions concerning the stereochemistry of the natural products. Dudley et al. ${ }^{14}$ has synthesised four possible diastereomers of cephalosporolide H (3-3c). Unfortunately without access to an authentic sample of the natural product definitive confirmation of the stereochemistry cannot be substantiated. Therefore uncertainty still exists regarding the
configuration of cephalosporolide H and the other members of the cephalosporolide family.

Deprotection of spiroacetals $\mathbf{1 5 a}$ and $\mathbf{1 5 b}$ was carried out using $10 \% \mathrm{Pd} / \mathrm{C}$ in EtOAc to give the corresponding primary alcohols $23\left(\alpha_{\mathrm{D}}+0.91, c 0.22 \mathrm{CHCl}_{3}\right)$ and $24\left(\alpha_{\mathrm{D}}-1.4, c 0.13\right.$ $\mathrm{CHCl}_{3}$ ). Further elaboration of 23 and 24 should afford the natural products penisporolides $\mathrm{A}(\mathbf{3})$ and B (4).


Scheme 10: Reagents and conditions: a) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOAc}, \mathrm{H}_{2}$, $18 \mathrm{~h}, 87 \%$; b) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOAc}, \mathrm{H}_{2}, 16 \mathrm{~h}, 68 \%$.

## 3. Conclusion

The synthetic work described herein provides access to the four stereoisomeric spiroacetals 8a, 8b, 15a, 15b which comprise the spiroacetal core structures of the natural products cephalosporolides $\mathrm{H}(\mathbf{3})$ and $\mathrm{I}(\mathbf{4})$ as well as penisporolides A (5) and B (6). Use of Sharpless asymmetric dihydroxylation with the appropriate chiral ligand enabled establishment of the desired absolute stereochemistry in the $\gamma$-lactone ring. Oxidative radical cyclisation was then used to form the two possible configurations at the spirocentre. The synthetic work undertaken also facilitates biological evaluation of the four individual spiroacetal core structures present in this important family of natural products.

## 4. Experimental

### 4.1 General methods

All reactions were carried out in oven-dried glassware which was further dried under high vacuum whilst heating with a heat gun. Reactions were carried out under an atmosphere of argon or nitrogen dried by passing through a cylinder of calcium chloride. Solvents were dried under standard conditions. Analytical thin layer chromatography (TLC) was performed on 0.2 mm aluminium plates of silica gel $60 \mathrm{~F}_{254}$ (Merck) and compounds were visualised by ultra-violet fluorescence or by staining with potassium permanganate or vanillin solutions, followed by heating the plate as appropriate. Flash column chromatography was carried out using 40-63 $\mu \mathrm{m}$, 230-430 mesh silica gel with the solvent indicated. Infrared spectra were obtained using a Perkin Elmer Spectrum One Fourier Transform infrared spectrometer with a universal ATR sampling accessory. Optical rotations were measured at $20^{\circ} \mathrm{C}$ using either a Perkin Elmer 341 polarimeter or a Rudolph Research Analytical Autopol IV at $\lambda=598 \mathrm{~nm}$ and are given in units of $10^{-1}$ deg cm $\mathrm{g}^{-1}$. Nuclear magnetic resonance (NMR) spectra were recorded as indicated on a Bruker AVANCE DRX400 spectrometer operating at 400 MHz for ${ }^{1} \mathrm{H}$ nuclei and 100 MHz for ${ }^{13} \mathrm{C}$ nuclei. All chemical shifts are recorded in parts per million ( ppm ) relative to tetramethylsilane $(\delta 0.00$ ppm) or deuterated chloroform ( $\delta 7.26 \mathrm{ppm}$ or $\delta 77.0 \mathrm{ppm}$ ). Coupling constant $J$ values are given in Hertz (Hz). Melting
points were determined on an Electrothermal ${ }^{\circledR}$ melting point apparatus and are uncorrected. Infrared spectra were obtained on a Perkin Elmer Spectrum One Universal ATR Sampling Accessory Fourier Transform IR spectrometer, neat, over a crystal plate or using a Perkin Elmer Spectrum 1000 series Fourier Transform Optical rotations of chiral compounds were obtained using a Perkin Elmer 341 polarimeter in the solvent indicated. High-resolution mass spectra were obtained by electron spray ionisation (ESI) using a microTOF-Q mass spectrometer. Microwave reactions were performed on the Biotage Initiator.

### 4.1.8 Benzyl 2,2-dimethylbut-3-enoate 11

To a solution of methyl triphenylphosphonium bromide ( 5.2 $\mathrm{g}, 14.5 \mathrm{mmol})$ in THF ( 40 mL ) was added LiHMDS ( 17.1 mL , 0.85 M in THF, 14.5 mmol ) at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 30 min and a solution of benzyl 2,2-dimethyl-3oxopropanoate ( $2 \mathrm{~g}, 9.71 \mathrm{mmol}$ ) in THF ( 10 ml ) was added. The mixture was allowed to warm to rt overnight and quenched by the addition of sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(15 \mathrm{~mL})$. The reaction mixture was extracted with $\operatorname{EtOAc}(3 \times 20 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 20 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using 20:1 hexanes/EtOAc as eluent gave the title compound $11(0.75 \mathrm{~g}$, $38 \%$ ) as a yellow oil. $\mathrm{R}_{\mathrm{f}}$ : 0.77 ( $90 \%$ hexanes/EtOAc); $v_{\text {max }}$ (film) $/ \mathrm{cm}^{-1}: 2977,1728,1259,1132,695 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right)$ : 7.37-7.29 (5H, m, Ph), $6.05(1 \mathrm{H}, \mathrm{dd}, J=17.4,10.5$ $\mathrm{Hz}, \mathrm{H}-3), 5.11\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}, \mathrm{H}-4\right), 5.06(1 \mathrm{H}, \mathrm{d}, J=10.5 \mathrm{~Hz}$, $\mathrm{H}-4), 1.33\left(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 176.1$ (C=O, C-1), $142.4(\mathrm{CH}, \mathrm{C}-3), 136.3(\mathrm{C}, \mathrm{Ph}), 128.5(2 \times \mathrm{CH}$, $\mathrm{Ph}), 127.9(\mathrm{CH}, \mathrm{Ph}), 127.7(2 \times \mathrm{CH}, \mathrm{Ph}), 113.0\left(\mathrm{CH}_{2}, \mathrm{C}-4\right)$, $66.3\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 44.9(\mathrm{C}, \mathrm{C}-2), 24.6\left(2 \times \mathrm{CH}_{3}\right)$. HRMS (ESI): calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{NaO}_{2}$ : 227.1043, found: 227.1040.
4.1.9 Benzyl 5-((5S)-5-((benzyloxy)methyl)tetrahydrofuran-2-yl)-2,2-dimethylpent-3-enoate 10

A solution of alkene $\mathbf{1 2}(0.1 \mathrm{~g}, 0.43 \mathrm{mmol})$ and alkene $\mathbf{1 1}$ $(0.35 \mathrm{~g}, 1.7 \mathrm{mmol})$ in toluene ( 3 mL ) in a microwave vessel was degassed using the freeze-pump-thaw method. A solution of Grubbs' II catalyst ( $0.04 \mathrm{~g}, 0.04 \mathrm{mmol}$ ) in toluene $(1 \mathrm{~mL})$ under argon was added. Chlorodicyclohexyl borane ( 0.04 mL , 1 M in hexane, 0.04 mmol ) was added and the solution heated in the microwave at $90^{\circ} \mathrm{C}$ for 9 h . Purification by column chromatography afforded the title compound $\mathbf{1 0}(0.04 \mathrm{~g}, 22 \%)$ as a $1: 1$ mixture of diastereomers as a yellow oil. $\mathrm{R}_{\mathrm{f}}: 0.51(60 \%$ hexanes $/ \mathrm{EtOAc}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1}: 2930,1726,1455,1130$, 697; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-7.25(10 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 5.69$ ( $1 \mathrm{H}, \mathrm{dt}, J=15.6,1.4 \mathrm{~Hz}, \mathrm{H}-3$ ), $5.53-5.45$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4$ ), 5.09 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.61-4.52 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.20-4.14 ( 0.4 H , $\mathrm{m}, \mathrm{H}-9)$, 4.10-4.03 ( $0.6 \mathrm{H}, \mathrm{m}, \mathrm{H}-9$ ), 4.01-3.94 ( $0.4 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), 3.91-3.84 ( $0.6 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), 3.48-3.40 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-10$ ), 2.40-2.32 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5$ ), 2.24-2.13 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5$ ), 1.98-1.78 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), 1.67-1.58 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-7$ ), $1.30\left(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 176.3(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.4(\mathrm{C}, \mathrm{Ph}), 136.7(2 \times \mathrm{CH}, 2 \times$ $\mathrm{C}-3), 136.3(\mathrm{C}, \mathrm{Ph}), 128.4(2 \times \mathrm{CH}, \mathrm{Ph}) 128.3(2 \times \mathrm{CH}, \mathrm{Ph})$, $127.9(\mathrm{CH}, \mathrm{Ph}), 127.7(2 \times \mathrm{CH}, \mathrm{Ph}), 127.6(2 \times \mathrm{CH}, \mathrm{Ph}), 127.5$ $(\mathrm{CH}, \mathrm{Ph}), 124.9(2 \times \mathrm{CH}, 2 \times \mathrm{C}-4), 79.3(\mathrm{CH}, \mathrm{C}-9), 78.8(\mathrm{CH}$, C-9), $78.1(\mathrm{CH}, \mathrm{C}-6), 77.8(\mathrm{CH}, \mathrm{C}-6), 73.3\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 73.0$ $\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 72.9\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 66.2\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 44.2(\mathrm{C}, \mathrm{C}-$ 2), $38.8\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 38.7\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 30.8\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 29.9$ $\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 28.5\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 28.1\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 25.1\left(2 \times \mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{26} \mathrm{H}_{32} \mathrm{NaO}_{4}$ : 431.2193, found: 431.2194.
4.1.10 (4R,5R)-5-(((2R,5S)-5-
((benzyloxy)methyl)tetrahydrofuran-2-yl)methyl)-4-hydroxy-3,3-dimethyldihydrofuran-2(3H)-one 9a and (4R,5R)-5-
(((2S,5S)-5-((benzyloxy)methyl)tetrahydrofuran-2-yl)methyl)-4-hydroxy-3,3-dimethyldihydrofuran-2(3H)-one $9 \boldsymbol{b}$

A solution of (DHQD) ${ }_{2}$ PHAL $(0.009 \mathrm{~g}, 0.01 \mathrm{mmol})$, methanesulfonamide $(0.023 \mathrm{~g}, 0.02 \mathrm{mmol})$, $\mathrm{OsO}_{4}(0.02 \mathrm{~mL}$, $2.5 \%$ solution in $t$-BuOH, $2.4 \mu \mathrm{~mol}), \mathrm{K}_{3} \mathrm{Fe}(\mathrm{CN})_{6}(0.23 \mathrm{~g}, 0.73$ $\mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.09 \mathrm{~g}, 0.73 \mathrm{mmol})$ in $t-\mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(1: 1,1$ $\mathrm{mL})$ was stirred for 30 min . A solution of alkene $10(0.10 \mathrm{~g}$, $0.24 \mathrm{mmol})$ in $t-\mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(1: 1,0.3 \mathrm{~mL})$ was added. The reaction mixture was stirred at rt overnight and $\mathrm{Na}_{2} \mathrm{SO}_{3}(0.02$ g) was added. The suspension was stirred for 30 min . The mixture was extracted with EtOAc ( $3 \times 1 \mathrm{~mL}$ ). The combined organic extracts were washed with $2 \mathrm{~N} \mathrm{KOH}(0.01 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using $9: 1$ hexanes/EtOAc as eluent gave a separable 1:1 mixture of 9a:9b as yellow oils.

Data for 9a; ( $0.03 \mathrm{~g}, 36 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.36$ ( $60 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}+12.1\left(c 1.0, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1}: 3438,2926,1765$, 1456, 1095; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.37-7.27(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH})$, $4.55\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.54-4.50(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4), 4.27-4.21(1 \mathrm{H}$, $\mathrm{m}, \mathrm{H}-9), 4.07-4.00(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6, \mathrm{H}-3), 3.73(1 \mathrm{H}, \mathrm{d}, J=2.6 \mathrm{~Hz}$, $\mathrm{OH}), 3.48(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-10), 2.21-2.13(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 2.08-1.97$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), 1.18-1.59 (3H, m, H-8, H-7), $1.26(6 \mathrm{H}, \mathrm{s}, 2 \times$ $\left.\mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $181.0(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.1(\mathrm{C}, \mathrm{Ph})$, $128.4(2 \times \mathrm{CH}, \mathrm{Ph}), 127.6(\mathrm{CH}, \mathrm{Ph}), 127.5(2 \times \mathrm{CH}, \mathrm{Ph}), 80.1$ ( $\mathrm{CH}, \mathrm{C}-4$ ), 78.6 (CH, C-9), 76.4 (CH, C-6), 75.7 (CH, C-3), $73.2\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 72.4(\mathrm{CH}, \mathrm{C}-10), 45.0(\mathrm{C}, \mathrm{C}-2), 34.5\left(\mathrm{CH}_{2}\right.$, $\mathrm{C}-5), 32.9\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 28.2\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.2\left(\mathrm{CH}_{3}\right), 17.8$ $\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{NaO}_{5}: 357.1672$, found: 357.1677.

Data for 9b; ( $0.03 \mathrm{~g}, 36 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.25$ ( $60 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}+7.0\left(\right.$ c $\left.0.3, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1}: 3439,2920,1759$, 1458, 1088; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.37-7.27(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH})$, 4.67-4.62 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4$ ), $4.54\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), ~ 4.14-4.09(2 \mathrm{H}$, $\mathrm{m}, \mathrm{H}-9, \mathrm{H}-6), 3.95(1 \mathrm{H}, \mathrm{d}, J=3.4 \mathrm{~Hz}, \mathrm{H}-3), 3.58(1 \mathrm{H}, \mathrm{dd}, J=$ $10.1,3.4 \mathrm{~Hz}, \mathrm{H}-10 \mathrm{a}), 3.51(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 3.46(1 \mathrm{H}, \mathrm{dd}, J=10.2$, $4.7 \mathrm{~Hz}, \mathrm{H}-10 \mathrm{~b}), 2.37-2.29(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 2.15-2.09(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ 5), 2.03-1.92 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), 1.90-1.83 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7$ ), 1.76-1.65 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7$ ), $1.21\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.09\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $181.0(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.1(\mathrm{C}, \mathrm{Ph}), 128.4(2 \times$ $\mathrm{CH}, \mathrm{Ph}), 127.7(\mathrm{CH}, \mathrm{Ph}), 127.6(2 \times \mathrm{CH}, \mathrm{Ph}), 78.1(\mathrm{CH}, \mathrm{C}-9)$, $77.8(\mathrm{CH}, \mathrm{C}-4), 76.3(\mathrm{CH}, \mathrm{C}-3), 76.1(\mathrm{CH}, \mathrm{C}-6), 73.4\left(\mathrm{CH}_{2}\right.$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), $72.2(\mathrm{CH}, \mathrm{C}-10), 45.0(\mathrm{C}, \mathrm{C}-2), 31.3\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 29.1$ $\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 27.7\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.2\left(\mathrm{CH}_{3}\right), 17.9\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{NaO}_{5}$ : 357.1672 , found: 357.1682 .
4.1.11 (4S,5S)-5-(((2R,5R)-5-
((benzyloxy)methyl)tetrahydrofuran-2-yl)methyl)-4-hydroxy-3,3-dimethyldihydrofuran-2(3H)-one 16a and (4S,5S)-5-(((2S,5S)-5-((benzyloxy)methyl)tetrahydrofuran-2-yl)methyl)-4-hydroxy-3,3-dimethyldihydrofuran-2(3H)-one 16b

A solution of ( DHQ$)_{2}$ PHAL $(0.016 \mathrm{~g}, 0.02 \mathrm{mmol})$, methanesulfonamide ( $0.04 \mathrm{~g}, 0.04 \mathrm{mmol})$, $\mathrm{OsO}_{4}(0.4 \mathrm{~mL}, 2.5 \%$ solution in $t$ - $\mathrm{BuOH}, 0.04 \mathrm{mmol}$ ), $\mathrm{K}_{3} \mathrm{Fe}(\mathrm{CN})_{6}(0.4 \mathrm{~g}, 1.3 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.18 \mathrm{~g}, 1.3 \mathrm{mmol})$ in $t-\mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(1: 1,1 \mathrm{~mL})$ was stirred for 30 min . A solution of alkene $10(0.18 \mathrm{~g}, 0.43 \mathrm{mmol})$ in $t-\mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(1: 1,0.3 \mathrm{~mL})$ was added and the reaction mixture was stirred at rt overnight before $\mathrm{Na}_{2} \mathrm{SO}_{3}(0.02 \mathrm{~g})$ was added. The suspension was stirred for 30 min and the mixture was extracted with EtOAc $(3 \times 3 \mathrm{~mL})$, the combined organic extracts were washed with $2 \mathrm{~N} \mathrm{KOH}(0.01 \mathrm{~mL})$, dried over
$\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using $9: 1$ hexanes/EtOAc as eluent gave a separable $1: 1$ mixture of 16a:16b as colourless solids.
Data for 16a; ( $0.05 \mathrm{~g}, 38 \%), \mathrm{R}_{\mathrm{f}}$ : $0.36(60 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}-37.5\left(c \quad 1.0, \mathrm{CHCl}_{3}\right) ; \mathrm{mp}: 60-64{ }^{\circ} \mathrm{C}$; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1}$ : $3440,2930,1762,1457,1086 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-$ $7.26(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 4.54\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.54-4.50(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ 4), 4.18-4.12 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9$ ), 4.04-4.02 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6, \mathrm{H}-3$ ), 3.99 $(1 \mathrm{H}, \mathrm{d}, J=2.9 \mathrm{~Hz}, \mathrm{OH}), 3.52(1 \mathrm{H}, \mathrm{dd}, J=10.0,3.6 \mathrm{~Hz}, \mathrm{H}-10)$, $3.38(1 \mathrm{H}, \mathrm{dd}, J=10.1,5.7 \mathrm{~Hz}, \mathrm{H}-10), 2.17-2.04(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-5$, H-7), 1.98-1.89 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), 1.87-1.78 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), $1.74-$ $1.64(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7), 1.28\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 181.1 (C=O, C-1), 137.5 (C, Ph), 128.4 ( 2 $\times \mathrm{CH}, \mathrm{Ph}), 128.1(2 \times \mathrm{CH}, \mathrm{Ph}), 127.8(\mathrm{CH}, \mathrm{Ph}), 80.4(\mathrm{CH}, \mathrm{C}-$ 4), $78.8(\mathrm{CH}, \mathrm{C}-9), 76.5(\mathrm{CH}, \mathrm{C}-6), 76.5(\mathrm{CH}, \mathrm{C}-3), 73.5\left(\mathrm{CH}_{2}\right.$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 71.8 (CH, C-10), 45.0 (C, C-2), $34.9\left(\mathrm{CH}_{2}, \mathrm{C}-5\right.$ ), 32.1 $\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 27.4\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.2\left(\mathrm{CH}_{3}\right), 17.9\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{KO}_{5}: 373.1412$, found: 373.1407.

Data for 16b; ( $0.05 \mathrm{~g}, 38 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.25$ ( $60 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}-18.0\left(\mathrm{c} 0.3, \mathrm{CHCl}_{3}\right) ; \mathrm{mp}: 59-65{ }^{\circ} \mathrm{C}$; $v_{\text {max }}(\mathrm{film}) / \mathrm{cm}^{-1}$ : 3441, 2929, 1763, 1466, 1095; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-$ $7.26(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), ~ 4.71-4.66(1 \mathrm{H}, \mathrm{H}-4), 4.55\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right)$, 4.26-4.19 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6, \mathrm{H}-9$ ), 3.92 ( $1 \mathrm{H}, \mathrm{t}, J=3.6 \mathrm{~Hz}, \mathrm{H}-3$ ), $3.61(1 \mathrm{H}, \mathrm{d}, J=3.6 \mathrm{~Hz}, \mathrm{OH}), 3.46(2 \mathrm{H}, \mathrm{d}, J=4.9, \mathrm{H}-10), 2.31-$ $2.24(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 2.14-2.02$ ( $3 \mathrm{H}, \mathrm{m}, \mathrm{H}-5, \mathrm{H}-8$ ), 1.79-1.63 $(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-7), 1.26\left(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : 180.8 (C=O, C-1), 138.1(C, Ph), 128.4 ( $2 \times \mathrm{CH}, \mathrm{Ph}$ ), 127.6 (2 $\times \mathrm{CH}, \mathrm{Ph}), 127.6(\mathrm{CH}, \mathrm{Ph}), 78.3(\mathrm{CH}, \mathrm{C}-9), 77.8(\mathrm{CH}, \mathrm{C}-4)$, $76.5(\mathrm{CH}, \mathrm{C}-3), 75.4(\mathrm{CH}, \mathrm{C}-6), 73.4\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 72.6$ $\left(\mathrm{CH}_{2}, \mathrm{C}-10\right)$, $45.2(\mathrm{C}, \mathrm{C}-2), 31.8\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 30.4\left(\mathrm{CH}_{2}, \mathrm{C}-7\right)$, $28.5\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.3\left(\mathrm{CH}_{3}\right), 17.9\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{NaO}_{5}: 357.1672$, found: 357.1675.

### 4.1.12 2R,3a'R,5S,6a'R)-5-((benzyloxy)methyl)-6', $6^{\prime}-$ dimethyltetrahydro-3H,3'H-spiro[furan-2,2'-furo[3,2-b]furan]$5^{\prime}\left(3 a^{\prime} H\right)$-one $8 a$ and ( $\left.2 S, 3 a^{\prime} R, 5 S, 6 a^{\prime} R\right)-5-(($ benzyloxy)methyl)-6',6'-dimethyltetrahydro-3H,3'H-spirolfuran-2,2'-furo[3,2-b]furan]-5'(3a'H)-one $\mathbf{8 b}$

A solution of lactone $9 \mathrm{a}(0.02 \mathrm{~g}, 0.06 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.1$ $\mathrm{mL})$ was added to a stirring solution of $\mathrm{PhI}(\mathrm{OAc})_{2}(0.04 \mathrm{~g}, 0.12$ $\mathrm{mmol})$ and $\mathrm{I}_{2}(0.04 \mathrm{~g}, 0.14 \mathrm{mmol})$ in hexane $(0.2 \mathrm{~mL}) . \mathrm{N}_{2}$ gas was bubbled through the solution for 5 min and the reaction was then irradiated with a 75 W desk lamp for 1 h . The solution was diluted with $\mathrm{Et}_{2} \mathrm{O}(0.2 \mathrm{~mL})$ and saturated aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(0.1 \mathrm{~mL})$ was added, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 1 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification on deactivated silica using 9:1 to 3:2 hexanes/EtOAc gave colourless oils.

Data for 8b; ( $0.005 \mathrm{~g}, 25 \%$ ), R $\mathrm{R}_{\mathrm{f}}: 0.69$ ( $50 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}+42.0\left(\mathrm{c} 0.26, \mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1}: 2930,1776$, 1050,$754 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-7.7$ ( $5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}$ ), 5.02 ( 1 H , ddd, $J=6.4,3.7,1.6 \mathrm{~Hz}, \mathrm{H}-4$ ), 4.56 ( 2 H , dd, $J=12.2,9.0$ $\left.\mathrm{Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.32(1 \mathrm{H}, \mathrm{d}, J=3.8 \mathrm{~Hz}, \mathrm{H}-3)$, $4.30-4.26(1 \mathrm{H}, \mathrm{m}$, H-9), 3.48-3.41 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-10$ ), $2.58(1 \mathrm{H}, \mathrm{dd}, J=15.0,6.5 \mathrm{~Hz}$, H-5), $2.36(1 \mathrm{H}, \mathrm{dd}, J=15.1,1.5 \mathrm{~Hz}, \mathrm{H}-5), 2.17-1.96(4 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-7, \mathrm{H}-8), 1.74-1.66(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8), 1.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.22$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 175.0(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.2$ $(\mathrm{C}, \mathrm{Ph}), 128.4(3 \times \mathrm{CH}, \mathrm{Ph}), 127.6(2 \times \mathrm{CH}, \mathrm{Ph}), 116.0(\mathrm{C}, \mathrm{C}-$ 6), $85.3(\mathrm{CH}, \mathrm{C}-9), 80.5(\mathrm{CH}, \mathrm{C}-4), 77.7(\mathrm{CH}, \mathrm{C}-3), 73.4\left(\mathrm{CH}_{2}\right.$, $\mathrm{C}-10), 72.4\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 44.4(\mathrm{C}, \mathrm{C}-2), 41.9\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 35.2$ $\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 26.8\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.0\left(\mathrm{CH}_{3}\right), 18.0\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NaO}_{5}: 355.1516$, found: 355.1528.

Data for 8a; ( $0.005 \mathrm{~g}, 25 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.56$ ( $50 \%$ hexanes: EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}-2.9$ (c 0.17 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}$ (film) $/ \mathrm{cm}^{-1}: 2926,1773$,

1214, 746; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ): 7.35-7.27 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}$ ), $5.06(1 \mathrm{H}, \mathrm{t}, J=4.8 \mathrm{~Hz}, \mathrm{H}-4), 4.53(2 \mathrm{H}, \mathrm{dd}, J=29.3,12.1 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 4.28(1 \mathrm{H}, \mathrm{d}, J=4.5 \mathrm{~Hz}, \mathrm{H}-3), 4.25-4.19(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9)$, $3.56(1 \mathrm{H}, \mathrm{dd}, J=9.9,5.4 \mathrm{~Hz}, \mathrm{H}-10), 3.48(1 \mathrm{H}, \mathrm{dd}, J=9.8,5.8$ $\mathrm{Hz}, \mathrm{H}-10), 2.51$ ( $1 \mathrm{H}, \mathrm{d}, J=14.2 \mathrm{~Hz}, \mathrm{H}-5$ ), 2.18-1.99 ( $5 \mathrm{H}, \mathrm{m}$, H-5, H-7, H-8), 1.92-1.84 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7$ ), 1.27 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}$ ), 1.21 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 178.8(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.6$ $(\mathrm{C}, \mathrm{Ph}), 128.3(2 \times \mathrm{CH}, \mathrm{Ph}), 127.6(2 \times \mathrm{CH}, \mathrm{Ph}), 127.4(\mathrm{CH}$, $\mathrm{Ph}), 115.7$ (C, C-6), 87.3 (CH, C-9), 85.3 (CH, C-4), 79.7 (CH, $\mathrm{C}-3), 73.5\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 73.3\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 44.6(\mathrm{C}, \mathrm{C}-2), 41.7$ $\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 36.0\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 27.9\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 24.6\left(\mathrm{CH}_{3}\right), 18.3$ $\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{O}_{5}: 333.1697$, found: 333.1694.
4.1.13 (2S,3a'S,5S,6a'S)-5-((benzyloxy)methyl)-6',6'-dimethyltetrahydro-3H, $3^{\prime}$ 'H-spiro[furan-2,2'-furo[3,2-b]furan]$5^{\prime}\left(3 a^{\prime} H\right)$-one 15a and ( $\left.2 R, 3 a^{\prime} \mathrm{S}, 5 S, 6 a^{\prime} S\right)$-5-((benzyloxy)methyl)-6',6'-dimethyltetrahydro-3H,3'H-spirolfuran-2,2'-furo[3,2-
blfuran]-5' $\left(3 a^{\prime} H\right)$-one $\mathbf{1 5 b}$
A solution of lactone $\mathbf{1 6 a}(0.03 \mathrm{~g}, 0.09 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.1$ mL ) was added to a stirring solution of $\mathrm{PhI}(\mathrm{OAc})_{2}(0.06 \mathrm{~g}, 0.18$ $\mathrm{mmol})$ and $\mathrm{I}_{2}(0.05 \mathrm{~g}, 0.21 \mathrm{mmol})$ in hexane $(0.2 \mathrm{~mL}) . \mathrm{N}_{2}$ gas was bubbled through the solution for 5 min and the reaction was then irradiated with a 75 W desk lamp for 1 h . The solution was diluted with $\mathrm{Et}_{2} \mathrm{O}(0.2 \mathrm{~mL})$ and saturated aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(0.1 \mathrm{~mL})$ was added, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 1 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification on deactivated silica using 9:1 to 3:2 hexanes/EtOAc gave a colourless oil.
Data for 15b; ( $0.007 \mathrm{~g}, 25 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.69$ ( $50 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}-41.5\left(\mathrm{c} 0.4, \mathrm{CHCl}_{3}\right) ; v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1}: 2933,1776,1138$, 1089,$1052 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-7.26(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH})$, $5.00(1 \mathrm{H}$, ddd, $J=6.3,3.9,1.5 \mathrm{~Hz}, \mathrm{H}-4), 4.58\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right)$, 4.34-4.27 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9$ ), $4.24(1 \mathrm{H}, \mathrm{d}, J=3.7 \mathrm{~Hz}, \mathrm{H}-3), 3.49$ $(1 \mathrm{H}, \mathrm{dd}, J=9.9,7.7 \mathrm{~Hz}, \mathrm{H}-10), 3.42(1 \mathrm{H}, \mathrm{dd}, J=9.9,4.0 \mathrm{~Hz}$, $\mathrm{H}-10), 2.57(1 \mathrm{H}, \mathrm{dd}, J=15.0,6.4 \mathrm{~Hz}, \mathrm{H}-5), 2.34(1 \mathrm{H}, \mathrm{dd}, J=$ $15.0,1.4 \mathrm{~Hz}, \mathrm{H}-5), 2.10-1.90$ ( $3 \mathrm{H}, \mathrm{m}, \mathrm{H}-7, \mathrm{H}-8$ ), 1.81-1.70 $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8), 1.19\left(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : 180.8 (C=O, C-1), 138.2 (C, Ph), 128.4 ( $2 \times \mathrm{CH}, \mathrm{Ph}$ ), 127.8 ( 2 $\times \mathrm{CH}, \mathrm{Ph}), 127.6(\mathrm{CH}, \mathrm{Ph}), 115.7(\mathrm{C}, \mathrm{C}-6), 85.1(\mathrm{CH}, \mathrm{C}-9)$, $80.3(\mathrm{CH}, \mathrm{C}-4), 79.0(\mathrm{CH}, \mathrm{C}-3), 74.2\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 73.2\left(\mathrm{CH}_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 44.3(\mathrm{C}, \mathrm{C}-2), 41.5\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 36.1\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 27.3$ $\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 23.0\left(\mathrm{CH}_{3}\right), 17.9\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NaO}_{5}: 355.1516$, found: 355.1516 .
Data for 15a; ( $0.007 \mathrm{~g}, 25 \%$ ), $\mathrm{R}_{\mathrm{f}}: 0.57$ ( $50 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}-20.4$ (c $0.25, \mathrm{CHCl}_{3}$ ); $v_{\text {max }}$ (film) $/ \mathrm{cm}^{-1}: 2936,1774$, 1120,$1105 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-7.26(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH})$, $5.10(1 \mathrm{H}, \mathrm{t}, J=5.2 \mathrm{~Hz}, \mathrm{H}-4), 4.53\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.35(1 \mathrm{H}, \mathrm{d}$, $J=5.0 \mathrm{~Hz}, \mathrm{H}-3), 4.29-4.25(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9), 3.52-3.42(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ 10), $2.47(1 \mathrm{H}, \mathrm{d}, J=14.1 \mathrm{~Hz}, \mathrm{H}-5), 2.15-2.01(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-5, \mathrm{H}-$ 7, H-8), 1.84-1.80 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ), $1.26\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.22(3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{3}$ ); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 180.8(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 138.5(\mathrm{C}, \mathrm{Ph})$, $128.3(2 \times \mathrm{CH}, \mathrm{Ph}), 127.6(2 \times \mathrm{CH}, \mathrm{Ph}), 127.5(\mathrm{CH}, \mathrm{Ph}), 115.7$ (C, C-6), 86.9 (CH, C-3), 79.9 (CH, C-4), 78.6 (CH, C-9), 73.3 $\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 72.1\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 44.3(\mathrm{C}, \mathrm{C}-2), 41.4\left(\mathrm{CH}_{2}, \mathrm{C}-\right.$ 5), $34.2\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 26.7\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 25.2\left(\mathrm{CH}_{3}\right), 18.4\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{KO}_{5}: 371.1255$, found: 371.1268.
4.1.14 (2R,3a'S,5S,6a'S)-5-(hydroxymethyl)-6',6'-
dimethyltetrahydro-3H,3'H-spiro[furan-2,2'-furo[3,2-b]furan]$5^{\prime}\left(3 a^{\prime} H\right)$-one 23

A solution of spiroacetal $\mathbf{1 5 a}(6.0 \mathrm{mg}, 0.018 \mathrm{mmol})$ was stirred in EtOAc ( 0.5 mL ) over $10 \% \mathrm{Pd} / \mathrm{C}(0.01 \mathrm{~g})$ under $\mathrm{H}_{2}$ overnight. The mixture was filtered through a plug of Celite ${ }^{\text {® }}$
and concentrated under reduced pressure to give the title compound $\mathbf{2 3}$ as a colourless oil ( $3 \mathrm{mg}, 0.012 \mathrm{mmol}, 68 \%$ ). $\mathrm{R}_{\mathrm{f}}$ : 0.13 ( $50 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}-1.4$ (c $0.13, \mathrm{CHCl}_{3}$ ); $v_{\text {max }}$ (film) $/ \mathrm{cm}^{-1}: 3432,2923,1772,1460,1125,828 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 5.14(1 \mathrm{H}, \mathrm{t}, J=5.1 \mathrm{~Hz}, \mathrm{H}-4), 4.39(1 \mathrm{H}, \mathrm{d}, J=4.9 \mathrm{~Hz}$, $\mathrm{H}-3), 4.24-4.18(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-9), 3.68(1 \mathrm{H}, \mathrm{dd}, J=11.8,2.9 \mathrm{~Hz}$, $\mathrm{H}-10), 3.47(1 \mathrm{H}, \mathrm{q}, J=5.8 \mathrm{~Hz}, \mathrm{H}-10), 2.48(1 \mathrm{H}, \mathrm{d}, J=14.1$ $\mathrm{Hz}, \mathrm{H}-5), 2.17(1 \mathrm{H}, \mathrm{dd}, J=14.2,5.5 \mathrm{~Hz}, \mathrm{H}-5), 2.14-2.02(4 \mathrm{H}$, $\mathrm{m}, \mathrm{H}-5, \mathrm{H}-7, \mathrm{H}-8), 1.27\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.21\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 180.9 ( $\mathrm{C}=\mathrm{O}, \mathrm{C}-1$ ), 115.4 (C, C-6), 86.8 ( $\mathrm{CH}, \mathrm{C}-3$ ), $79.8(\mathrm{CH}, \mathrm{C}-3), 79.8(\mathrm{CH}, \mathrm{C}-4), 64.7\left(\mathrm{CH}_{2}, \mathrm{C}-10\right)$, $44.4(\mathrm{C}, \mathrm{C}-2), 41.5\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 34.5\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 25.5\left(\mathrm{CH}_{2}, \mathrm{C}-\right.$ 8), $25.5\left(\mathrm{CH}_{3}\right), 18.2\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{KO}_{5}: 281.0786$, found: 281.0790.
4.1.14 (2S,3a'S,5S,6a'S)-5-(hydroxymethyl)-6',6'-
dimethyltetrahydro-3H,3'H-spiro[furan-2,2'-furo[3,2-b]furan]-5'(3a'H)-one 24

A solution of spiroacetal $\mathbf{1 5 b}(0.011 \mathrm{~g}, 0.03 \mathrm{mmol})$ was stirred in EtOAc ( 1 mL ) over $10 \% \mathrm{Pd} / \mathrm{C}(0.02 \mathrm{~g})$ under $\mathrm{H}_{2}$ overnight. The mixture was filtered through a plug of Celite ${ }^{\circledR}$ and concentrated under reduced pressure to give the title compound $\mathbf{2 4}$ as a colourless oil ( $7.0 \mathrm{mg}, 87 \%$ ). $\mathrm{R}_{\mathrm{f}}: 0.13(50 \%$ hexanes $/ \mathrm{EtOAc}$ ); $[\alpha]_{\mathrm{D}}+0.91\left(\mathrm{c} 0.22, \mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1}$ : 3432, 2923, 1772, 1460, 1125, 828; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $5.05(1 \mathrm{H}$, ddd, $J=6.0,3.7,1.4 \mathrm{~Hz}, \mathrm{H}-4), 4.39-4.31(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ $3, \mathrm{H}-9), 3.75(1 \mathrm{H}, \mathrm{dd}, J=9.9 \mathrm{~Hz}, \mathrm{H}-12), 3.51-3.48(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ $10), 2.58(1 \mathrm{H}, \mathrm{dd}, J=15.2,6.1 \mathrm{~Hz}, \mathrm{H}-5), 2.50(1 \mathrm{H}, \mathrm{dd}, J=$ $15.2,0.9 \mathrm{~Hz}, \mathrm{H}-5), 2.20-1.96(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-7, \mathrm{H}-8), 1.26(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{3}\right), 1.23\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 180.4(\mathrm{C}=\mathrm{O}$, C-1), 115.7 (C, C-6), 85.4 (CH, C-9), 80.9 (CH, C-3), 80.1 ( $\mathrm{CH}, \mathrm{C}-4), 64.7\left(\mathrm{CH}_{2}, \mathrm{C}-10\right), 44.5(\mathrm{C}, \mathrm{C}-2), 41.1\left(\mathrm{CH}_{2}, \mathrm{C}-5\right)$, $37.2\left(\mathrm{CH}_{2}, \mathrm{C}-7\right), 24.8\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 22.8\left(\mathrm{CH}_{3}\right), 17.9\left(\mathrm{CH}_{3}\right)$. HRMS (ESI): calcd. for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{KO}_{5}$ : 281.0786, found: 281.0790.

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## SUPPORTING INFORMATION

## for

Synthesis of the spiroacetal core of the cephalosporolide family of natural products
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## 2-((Benzyloxy)methyl)oxirane

$\mathrm{NaH}(60 \%$ dispersion in oil, $5.40 \mathrm{~g}, 135 \mathrm{mmol})$ was added to a stirred solution of glycidol $\mathbf{1 3}(10 \mathrm{~g}, 135 \mathrm{mmol})$ and $\mathrm{BnBr}(10.5$ $\mathrm{mL}, 176 \mathrm{mmol})$ in THF $(300 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was allowed to reach rt and stirred overnight. $\mathrm{H}_{2} \mathrm{O}(200 \mathrm{~mL})$ was added and the mixture extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 100 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 100 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure to give a crude yellow oil. Purification by flash chromatography using 9:1 hexanes/EtOAc as eluent afforded the title compound ( $18.8 \mathrm{~g}, 84 \%$ ) as a colourless oil. $\mathrm{R}_{\mathrm{f}}: 0.5\left(80 \%\right.$ hexanes/EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.35-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H), 4.61\left(1 \mathrm{H}, \mathrm{d}, J=11.9 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.55(1 \mathrm{H}, \mathrm{d}, J=$ $\left.11.9 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 3.76(1 \mathrm{H}, \mathrm{dd}, J=3.1,11.4 \mathrm{~Hz}, \mathrm{H}-3), 3.44(1 \mathrm{H}, \mathrm{dd}, J=5.8,11.4 \mathrm{~Hz}, \mathrm{H}-3), 3.21-3.16(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 2.81-2.78$ ( 1 H , $\left.\mathrm{m}, \mathrm{H}-1^{\prime}\right), 2.62-2.60\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1^{\prime}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 137.8(\mathrm{C}, \mathrm{Ph}), 128.4(2 \times \mathrm{CH}, \mathrm{Ph}), 127.7(3 \times \mathrm{CH}, \mathrm{Ph}), 73.3\left(\mathrm{CH}_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 70.8\left(\mathrm{CH}_{2}, \mathrm{C}-1\right), 50.8(\mathrm{CH}, \mathrm{C}-2), 44.3\left(\mathrm{CH}_{2}, \mathrm{C}-3\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{22}$

## (R)-2-((benzyloxy)methyl)oxirane 17

( $R, R$ )-N,N-Bis-(3,5-di-tert-butylsalicyclidene)-1,2-cyclohexanesanediaminocobalt (II) ( $0.029 \mathrm{~g}, 0.05 \mathrm{mmol}$ ) was added to a mixture of 2-((benzyloxy)methyl)oxirane ( $1.63 \mathrm{~g}, 9.91 \mathrm{mmol}$ ), AcOH ( $0.011 \mathrm{~mL}, 0.198 \mathrm{mmol}$ ) and THF ( 0.15 mL ). The mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and treated with $\mathrm{H}_{2} \mathrm{O}(0.098 \mathrm{~mL}, 5.45 \mathrm{mmol})$. The reaction was allowed to reach rt overnight. Purification by flash chromatography using 9:1 hexanes/EtOAc as eluent gave the title compound $\mathbf{1 7}(0.795 \mathrm{~g}, 4.84 \mathrm{mmol}, 48 \%)$ as a colourless oil. $\mathrm{R}_{\mathrm{f}}: 0.5\left(80 \%\right.$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}+6.03\left(c\right.$ 1.76, $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ lit. $[\alpha]_{\mathrm{D}}+5.4\left(c 1.76\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{16} \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.35-7.25$ $(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph} H), 4.61\left(1 \mathrm{H}, \mathrm{d}, J=11.9 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.55\left(1 \mathrm{H}, \mathrm{d}, J=11.9 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 3.76(1 \mathrm{H}, \mathrm{dd}, J=3.1,11.4 \mathrm{~Hz}, \mathrm{H}-3), 3.44(1 \mathrm{H}$, dd, $J=5.8,11.4 \mathrm{~Hz}, \mathrm{H}-3), 3.21-3.16(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 2.81-2.78\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1^{\prime}\right), 2.62-2.60\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1^{\prime}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 137.8$ $(\mathrm{C}, \mathrm{Ph}), 128.4(2 \times \mathrm{CH}, \mathrm{Ph}), 127.7(3 \times \mathrm{CH}, \mathrm{Ph}), 73.3\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 70.8\left(\mathrm{CH}_{2}, \mathrm{C}-1\right), 50.8(\mathrm{CH}, \mathrm{C}-2), 44.3\left(\mathrm{CH}_{2}, \mathrm{C}-3\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{16}$

## (2S)-1-(benzyloxy)hex-5-en-2-ol 18

Allylmagnesium bromide ( $4.57 \mathrm{~mL}, 1 \mathrm{M}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) was added to a stirring solution of $(R)$-benzyl glycidyl ether $\mathbf{1 7}(0.5 \mathrm{~g}, 3.05$ $\mathrm{mmol})$ and $\mathrm{CuI}(0.058 \mathrm{~g}, 0.305 \mathrm{mmol})$ in THF $(20 \mathrm{~mL})$ at $-40^{\circ} \mathrm{C}$. The black solution was allowed to warm to rt over 3 h and quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$. The reaction mixture was extracted with $\mathrm{EtOAc}(3 \times 20 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 40 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give a yellow oil. Purification by flash chromatography using $4: 1$ hexanes/EtOAc as eluent gave the title compound $\mathbf{1 8}(0.44 \mathrm{~g}, 2.13 \mathrm{mmol}, 70 \%)$ as a yellow oil. $\mathrm{R}_{\mathrm{f}}: 0.38$ ( $80 \%$ hexanes/EtOAc); $[\alpha]_{\mathrm{D}}{ }^{20}+7.36$ (c $1.00, \mathrm{CHCl}_{3}$ ); lit. $[\alpha]_{\mathrm{D}}{ }^{20}-7.28$ (c $1.00, \mathrm{CHCl}_{3}$ ) (for enantiomer); ${ }^{23} \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : 7.38-7.29 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}$ ), 5.87-5.77 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5$ ), 5.06-4.95 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), $4.56\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 3.87-3.80(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 3.50(1 \mathrm{H}$, dd, $J=9.4,3.3 \mathrm{~Hz}, \mathrm{H}-1), 3.34(1 \mathrm{H}, \mathrm{dd}, J=7.9,9.4 \mathrm{~Hz}, \mathrm{H}-1), 2.31(1 \mathrm{H}, \mathrm{d}, J=3.5 \mathrm{~Hz}, \mathrm{OH}), 2.27-2.08(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4), 1.62-1.47$ ( 2 H , $\mathrm{m}, \mathrm{H}-3) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $138.4(\mathrm{CH}, \mathrm{C}-5), 138.0(\mathrm{C}, \mathrm{Ph}), 128.5(2 \times \mathrm{CH}, \mathrm{Ph}), 127.8(\mathrm{CH}, \mathrm{Ph}), 127.7(2 \times \mathrm{CH}, \mathrm{Ph}), 115.0$ $\left(\mathrm{CH}_{2}, \mathrm{C}-6\right), 74.5\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 73.4\left(\mathrm{CH}_{2}, \mathrm{C}-1\right), 70.0(\mathrm{CH}, \mathrm{C}-2), 32.3\left(\mathrm{CH}_{2}, \mathrm{C}-3\right), 29.8\left(\mathrm{CH}_{2}, \mathrm{C}-4\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{17}$

## (2S)-Tetrahydro-5-hydroxy-2-(benzyloxymethyl)furan 19

To a solution of alkene $\mathbf{1 8}(0.2 \mathrm{~g}, 0.97 \mathrm{mmol})$ in dioxane $/ \mathrm{H}_{2} \mathrm{O}(3: 1,4 \mathrm{~mL})$ was added 2,6-lutidine ( $\left.0.11 \mathrm{~mL}, 0.97 \mathrm{mmol}\right)$, OsO ${ }_{4}$ $(0.25 \mathrm{~mL}, 2.5 \%$ solution in $t-\mathrm{BuOH}, 2 \mu \mathrm{~mol})$ and $\mathrm{NaIO}_{4}(0.42 \mathrm{~g}, 1.94 \mathrm{mmol})$ respectively. The suspension was stirred for 2 h and sat. aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(2 \mathrm{~mL})$ was added. The reaction mixture was extracted with EtOAc $(3 \times 5 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 10 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by flash chromatography using 9:1 hexanes/EtOAc as eluent gave the title compound $\mathbf{1 9}(0.14 \mathrm{~g}, 0.694 \mathrm{mmol}, 71 \%)$ as a colourless oil ( $1: 1$ mixture of diastereomers). $\mathrm{R}_{\mathrm{f}}: 0.23\left(90 \%\right.$ hexanes $/ \mathrm{EtOAc}$ ); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.37-7.27(10 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 5.60-5.58(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ 4), 5.45-5.42 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4^{\prime}$ ), 4.60-4.56 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.47-4.40 (1H, m, H-1), 4.33-4.27 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1^{\prime}$ ), 3.70-3.56 (2H, m, H-5), 3.51-3.42 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-5^{\prime}$ ), 2.66-1.66 ( $8 \mathrm{H}, \mathrm{m}, \mathrm{H}-2, \mathrm{H}-2^{\prime}, \mathrm{H}-3, \mathrm{H}-3^{\prime}$ ), $1.95(2 \mathrm{H}, \mathrm{br} \mathrm{s}, 2 \times \mathrm{OH}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 138.2(\mathrm{C}, \mathrm{Ph})$, $137.6\left(\mathrm{C}^{\prime}, \mathrm{Ph}\right), 128.5(2 \times \mathrm{CH}, \mathrm{Ph}), 128.3(\mathrm{CH}, \mathrm{Ph}), 127.9(2 \times \mathrm{CH}, \mathrm{Ph}), 127.8(2 \times \mathrm{CH}, \mathrm{Ph}), 127.7(\mathrm{CH}, \mathrm{Ph}), 127.6(\mathrm{CH}, \mathrm{Ph}), 127.6$ (CH, Ph), $\left.98.9(\mathrm{CH}, \mathrm{C}-4), 98.8\left(\mathrm{CH}, \mathrm{C}-4^{\prime}\right), 78.9(\mathrm{CH}, \mathrm{C}-1), 78.7(\mathrm{CH}, \mathrm{C}-1)^{\prime}\right), 73.5\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 72.5\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 34.6\left(\mathrm{CH}_{2}, \mathrm{C}-\right.$ 3), $32.7\left(\mathrm{CH}_{2}, \mathrm{C}-3\right), 25.8\left(\mathrm{CH}_{2}, \mathrm{C}-2\right), 24.5\left(\mathrm{CH}_{2}, \mathrm{C}-2\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{24}$

To a solution of alcohol $\mathbf{1 9}(5.34 \mathrm{~g}, 25.9 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(125 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was added allyltrimethylsilane ( 12.4 mL , $77.7 \mathrm{mmol})$ followed by dropwise addition of $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(3.49 \mathrm{~mL}, 28.5 \mathrm{mmol})$. The reaction mixture was stirred for 3 h , sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$ added and the mixture extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The combined organic extracts were washed with brine $(20 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using 9:1 hexanes/EtOAc as eluent gave a $1: 1$ mixture of cis/trans isomers of the title compound $\mathbf{1 2}(5.0 \mathrm{~g}, 21.5 \mathrm{mmol}, 83 \%)$ as a yellow oil. $\mathrm{R}_{\mathrm{f}}: 0.38\left(90 \%\right.$ hexanes/EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 7.36-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{PhH}), 5.81(1 \mathrm{H}, \mathrm{ddt}, J=17.2,10.2,7.0 \mathrm{~Hz}, \mathrm{H}-8), 5.10-$ $5.02(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9), 4.57\left(2 \mathrm{H}, \mathrm{dd}, J=5.6,12.3 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.23-4.17$ ( $0.3 \mathrm{H}, \mathrm{m}, \mathrm{H}-2$ ), 4.11-4.00 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2^{\prime}, \mathrm{H}-5^{\prime}$ ), 3.97-3.90 $(0.7 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 3.51-3.42(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 2.42-2.35(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7), 2.27-2.19\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-7{ }^{\prime}\right), 2.03-1.87$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4, \mathrm{H}-3$ ), 1.71-1.64 $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3^{\prime}\right), 1.59-1.51\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4^{\prime}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 138.4(\mathrm{C}, \mathrm{Ph}), 135.0\left(2 \times \mathrm{CH}, \mathrm{C}-7, \mathrm{C}^{\prime} 7^{\prime}\right), 128.3(2 \times \mathrm{CH}, \mathrm{Ph}), 127.6$ $(2 \times \mathrm{CH}, \mathrm{Ph}), 127.5(\mathrm{CH}, \mathrm{Ph}), 116.8\left(\mathrm{CH}_{2}, \mathrm{C}-8\right), 116.8\left(\mathrm{CH}_{2}, \mathrm{C}-8^{\prime}\right), 79.2(\mathrm{CH}, \mathrm{C}-4), 78.7\left(\mathrm{CH}, \mathrm{C}-4^{\prime}\right), 78.1(\mathrm{CH}, \mathrm{C}-1), 77.7(\mathrm{CH}, \mathrm{C}-$ $\left.1^{\prime}\right), 73.3\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 73.0\left(\mathrm{CH}_{2}, \mathrm{C}-5\right), 72.9\left(\mathrm{CH}_{2}, \mathrm{C}-5^{\prime}\right), 40.2\left(\mathrm{CH}_{2}, \mathrm{C}-6\right), 40.1\left(\mathrm{CH}_{2}, \mathrm{C}-6^{\prime}\right), 31.1\left(\mathrm{CH}_{2}, \mathrm{C}-2\right), 30.2\left(\mathrm{CH}_{2}, \mathrm{C}-2^{\prime}\right)$, $28.5\left(\mathrm{CH}_{2}, \mathrm{C}-3\right), 28.1\left(\mathrm{CH}_{2}, \mathrm{C}-3^{\prime}\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{18}$

## Benzyl 3-hydroxy-2,2-dimethylpropanoate 20

$\mathrm{LiOH}(15.9 \mathrm{~g}, 37.9 \mathrm{mmol})$ in $\mathrm{THF} / \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(1: 1: 0.5,62.5 \mathrm{~mL})$ was added to a solution of 3-hydroxy-2,2-dimethylpropionate $14(5 \mathrm{~g}, 37.9 \mathrm{mmol})$ in $\mathrm{THF} / \mathrm{MeOH}(1: 1,25 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The solution was warmed to rt and stirred for 1.5 h , the reaction mixture was adjusted to pH 2 by addition of $\mathrm{H}_{2} \mathrm{SO}_{4}$. The solution was concentrated under reduced pressure to remove THF and the residue washed with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$, extracted with $\mathrm{EtOAc}(3 \times 12 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure to give the acid ( $3.3 \mathrm{~g}, 27.9 \mathrm{mmol}, 73 \%$ ) as a white solid. Benzyl bromide ( $1.9 \mathrm{~mL}, 16.1 \mathrm{mmol}$ ) was added to a solution of crude acid ( 2.0 $\mathrm{g}, 16.9 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(2.57 \mathrm{~g}, 18.6 \mathrm{mmol})$ in DMF $(20 \mathrm{~mL})$. The reaction mixture was stirred for 5 h then quenched by the addition of $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{~mL})$. The reaction mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 10 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 10 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using 9:1 to $3: 2$ hexanes/EtOAc as eluent gave the title compound $20(3.04 \mathrm{~g}, 86 \%)$ as a yellow oil. $\mathrm{R}_{\mathrm{f}}: 0.48$ ( $60 \%$ hexanes/EtOAc); $\delta_{\mathrm{H}}(400$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 7.39-7.29 (5H, m, Ph), $5.14\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 3.57(2 \mathrm{H}, \mathrm{d}, J=6.4 \mathrm{~Hz}, \mathrm{H}-3), 2.54(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.21(6 \mathrm{H}, \mathrm{s}, 2 \times$ $\left.\mathrm{CH}_{3}\right)$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 176.9(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 135.8(\mathrm{C}, \mathrm{Ph}), 128.3(2 \times \mathrm{CH}, \mathrm{Ph}), 127.9(\mathrm{CH}, \mathrm{Ph}), 127.5(2 \times \mathrm{CH}, \mathrm{Ph}), 69.3\left(\mathrm{CH}_{2}\right.$, $\mathrm{C}-3)$, $66.1\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 44.2(\mathrm{C}, \mathrm{C}-2), 21.8\left(2 \times \mathrm{CH}_{3}\right)$. The spectroscopic data were in agreement with those reported in the literature. ${ }^{19}$

## Benzyl 2,2-dimethyl-3-oxopropanoate

Oxalyl chloride ( $2.5 \mathrm{~mL}, 29.2 \mathrm{mmol}$ ) was added dropwise to a solution of DMSO ( $4.14 \mathrm{~mL}, 58.4 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 100 mL ) at - 78 ${ }^{\circ} \mathrm{C}$. The reaction mixture was stirred for 10 min and a solution of alcohol $20(5.0 \mathrm{~g}, 24.3 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added. The solution was stirred for $1 \mathrm{~h}, \mathrm{NEt}_{3}(13.5 \mathrm{~mL}, 97.3 \mathrm{mmol})$ added and the solution allowed to warm to $\mathrm{rt} . \mathrm{H}_{2} \mathrm{O}(25 \mathrm{~mL})$ was added and the mixture extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50 \mathrm{~mL})$. The combined organic extracts were washed with brine ( 30 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure. Purification by column chromatography using 9:1 hexanes/EtOAc as eluent gave the title compound $(4.13 \mathrm{~g}, 82 \%)$ as a yellow oil. $\mathrm{R}_{\mathrm{f}}: 0.58$ ( $80 \%$ hexanes/EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 9.68(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-3), 7.39-$ $7.31(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.19\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 1.37\left(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 198.9(\mathrm{C}=\mathrm{O}, \mathrm{C}-3), 172.5(\mathrm{C}=\mathrm{O}, \mathrm{C}-1), 135.3$ $(\mathrm{C}, \mathrm{Ph}), 128.6(2 \times \mathrm{CH}, \mathrm{Ph}) 128.3(\mathrm{CH}, \mathrm{Ph}), 127.9(2 \times \mathrm{CH}, \mathrm{Ph}), 67.1\left(\mathrm{CH}_{2}, \mathrm{CH}_{2} \mathrm{Ph}\right), 45.9(\mathrm{C}, \mathrm{C}-2), 19.6\left(2 \times \mathrm{CH}_{3}\right)$. The spectroscopic data were in agreement with those reported in the literature._ENREF_20 ${ }^{20}$
$11{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):




${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):



9a ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):


9b ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):


16a ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):



16b ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):


16b

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):


8a ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):


${ }^{13}$ C NMR (100 MHz, $\left.\mathrm{CHCl}_{3}\right)$ :



15a ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):




15b ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):



${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

$24{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):

${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CHCl}_{3}$ ):




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[^1]:    * Recently Fernandes et al. ${ }^{15}$ reported that a very similar cross metathesis reaction did not proceed, due to steric crowding near the olefin bond caused by geminal methyl substituents. In the absence of these geminal dimethyl groups, the reaction proceeded with high yield and $E$ selectivity.

