

Cite this: *Org. Biomol. Chem.*, 2011, **9**, 5104

www.rsc.org/obc

PAPER

Exploiting Sm(II) and Sm(III) in SmI₂-initiated reaction cascades: application in a tag removal–cyclisation approach to spirooxindole scaffolds†

Susannah C. Coote, Seidjolo Quenum and David J. Procter*

Received 14th February 2011, Accepted 4th May 2011

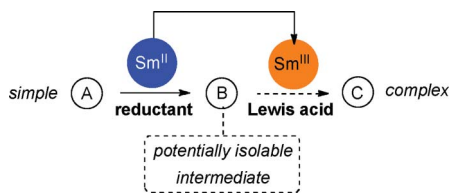
DOI: 10.1039/c1ob05710c

A tag removal–cyclisation sequence is described that is initiated by reduction using a Sm(II) species and completed by a Sm(III) Lewis acid that is formed in an earlier stage. Therefore, the reaction cascade utilises both oxidation states of a samarium reagent in discrete steps and allows access to privileged, pyrrolidinyl-spirooxindole scaffolds and analogues inspired by the anti-cancer natural product spirotryprostatin A.

Introduction

Cascade reactions in which multiple synthetic tasks are carried out selectively in a single reaction flask represent the holy grail for the synthetic chemist. The lanthanide reagent, samarium diiodide (SmI₂),¹ is one of the most important chemical reductants and is often the reagent of choice for the orchestration of reductive reaction cascades due to the high chemo-, regio- and stereoselectivity it exhibits.²

Here we describe an example of a sequence that utilises both oxidation states of a samarium reagent in discrete steps. In contrast to reduction-only cascades, the exploitation of additional Lewis acid-mediated processes to terminate reaction sequences, mediated by Sm(III) species formed in earlier reductive steps, greatly expands the chemical space that can be accessed. In the new reaction cascades, Sm(II)-mediated reduction of A generates an otherwise stable product B that is then transformed by the Lewis acidic Sm(III) by-product to give C (Scheme 1).

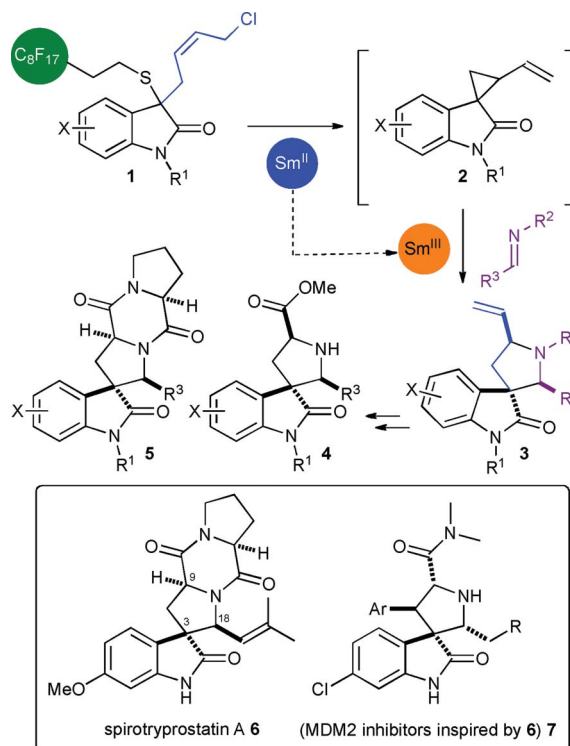
Scheme 1 Exploiting Sm(II) and Sm(III) in SmI₂-initiated cascades.

To evaluate the feasibility of exploiting both oxidation states of a samarium reagent in cascade reactions we designed a phase

School of Chemistry, University of Manchester, Oxford Road, Manchester, UK M13 9PL. E-mail: david.j.procter@manchester.ac.uk; Fax: +44 (0)161 275 4939; Tel: +44 (0)161 2751425

† Electronic supplementary information (ESI) available: Experimental procedures, characterisation data and ¹H and ¹³C spectra, X-ray crystallographic data. CCDC reference numbers 780379–780382 and 791298. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c1ob05710c

tag removal–cyclisation sequence initiated by reduction of tagged oxindoles **1** using a Sm(II) species and completed by a Sm(III) Lewis acid that is formed in an earlier step. The fluororous³ approach would allow access to the biologically significant pyrrolidinyl-spirooxindoles **3**, with purification being conveniently carried out using fluororous solid-phase extraction (FSPE)⁴ (Scheme 2). The pyrrolidinyl-spirooxindole unit forms the core of an extensive family of alkaloid natural products, many of whose members possess bioactivity.⁵ For example, spirotryprostatins A and B⁶



Scheme 2 A Sm(II)/Sm(III)-mediated approach to the pyrrolidinyl-spirooxindole scaffold.

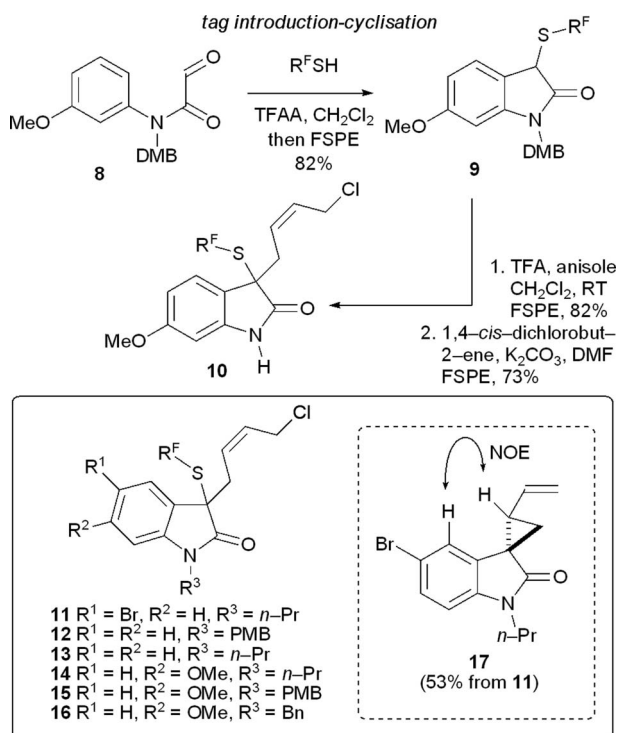
exhibit anti-cancer activity and analogues of spirotryprostatin **6** have improved activity against breast cancer cell lines.^{6b} Furthermore, simplified analogues such as **7** are promising MDM2 inhibitors.⁷ We proposed that manipulation of **3** could give natural product analogues **5** via intermediates **4**, which are related to MDM2 inhibitors **7** (Scheme 2).

New tagging strategies⁸ continue to play a key role in the quest for more efficient synthesis. In particular, the use of fluororous tags and associated purification technologies³ now enjoys widespread use in chemistry and chemical biology. In addition to the development of new Sm-mediated cascade processes, our studies will show how the unavoidable tag introduction and tag removal steps can be used to trigger important synthetic events in a route and thus the use of a phase tag to assist purification need not lengthen a synthesis.

Results and discussion

Development and scope of Sm(II)/Sm(III)-mediated reaction cascades

To investigate our proposal to exploit a Sm(II)/Sm(III)-mediated reaction cascade, tagged oxindoles **10–16** were prepared using a modification of our tag introduction–cyclisation process based on a Pummerer-type⁹ reaction, and were purified conveniently using fluororous solid-phase extraction (FSPE).⁴ For example, tag introduction–cyclisation^{10,11} involving glyoxamide **8** followed by removal of the DMB protection and alkylation with *cis*-1,4-dichlorobut-2-ene gave **10**. *N*-Alkyl substrates **11–16** were prepared in two steps from substrates analogous to **8**. FSPE was used to purify products in all steps *en route* to **10–16** (Scheme 3). Pleasingly, treatment of tagged substrate **11** with SmI₂¹² triggered

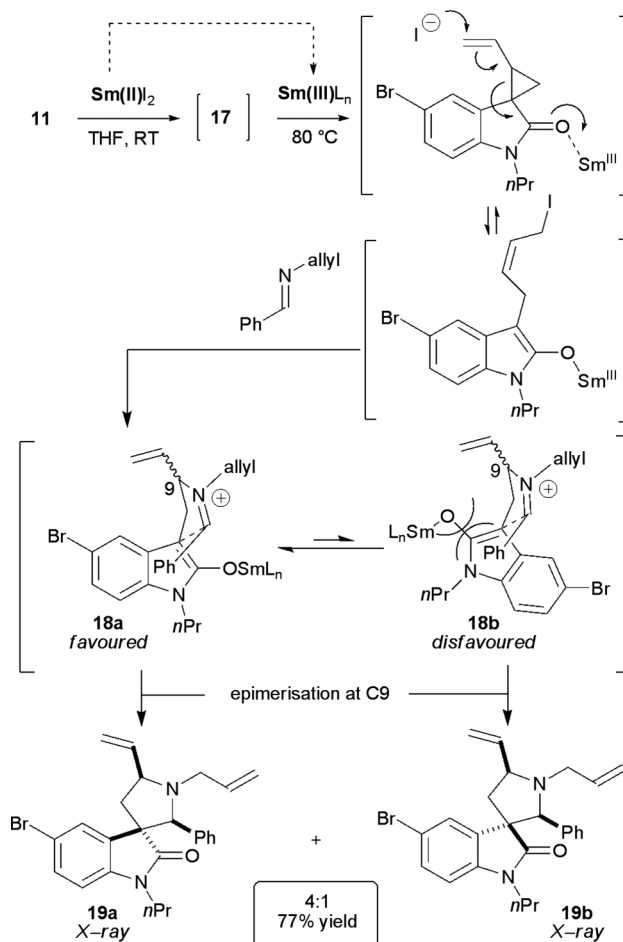


Scheme 3 Substrates for Sm(II)/Sm(III)-mediated cascades ($R^F = -CH_2CH_2C_8F_{17}$).

expulsion of the fluororous tag and intramolecular Sm(III)-enolate¹³ alkylation to give vinylcyclopropane **17** (*cf.* **2**) in 53% isolated yield as a single diastereoisomer (Scheme 3).

We postulated that **17** would react with an imine in the presence of a Sm(III) Lewis acid, formed during tag removal, to give an iminium ion intermediate (*cf.* **18**) that would undergo subsequent Mannich-type cyclisation. Carreira has previously employed oxindole cyclopropanes in analogous cycloadditions mediated by MgX_2 .^{6h,6i,14}

We were pleased to find that treatment of **11** with SmI₂ in THF (2 equiv) at room temperature followed by addition of *N*-allylphenylimine and heating to 80 °C resulted in removal of the fluororous tag and cyclisation to give pyrrolidinyl-spirooxindole **19** as a separable 4 : 1 mixture of diastereoisomers in 77% yield after purification by FSPE. The relative stereochemistry of diastereoisomers **19a** and **19b** was confirmed by X-ray crystallographic analysis¹⁵ (Scheme 4).



Scheme 4 Mechanism and proposed origin of diastereoselectivity in the Sm(II)/Sm(III)-mediated reaction cascade.

The intermediacy of spirocyclopropyloxindole **17** was confirmed by monitoring the reaction by ¹H NMR spectroscopy and by the observation that heating **17** at 80 °C with *N*-allylphenylimine in the presence of SmI₃ gave **19** in 68% yield and a similar diastereoisomeric ratio to that obtained from the sequential reaction. The evolution of the reagent was crucial to

the success of the cascade, as heating **17** with the imine in the absence of Sm(III) led to the recovery of starting materials.

The diastereoselectivity observed in the reaction cascade to give **19** can be explained by considering transition structures **18** (Scheme 4).¹⁶ Transition structure **18b** is disfavoured due to steric interactions arising from the *syn* relationship between the Sm(III)-enolate and its associated ligands and the substituent on the imine. The *syn* relationship between the substituents α to nitrogen in the pyrrolidine ring in both products appears to result from rapid epimerisation of the stereocentre α to nitrogen bearing the vinyl group (C9 in Scheme 4), post cyclisation.

The approach to oxindole cyclopropanes embedded in our new cascade reaction has several advantages over literature routes: mild, reductive conditions are used, thereby avoiding the use of strong bases and low temperatures or diazocompounds.^{6h,6j,14} In addition, the *in situ* generation of these relatively unstable species renders their isolation and purification unnecessary.

The scope of the Sm(II)/Sm(III)-mediated reaction cascade was explored using oxindole substrates **10–16** and a range of imines. Pyrrolidinyl-spirooxindoles **19–30** were obtained in good overall yield (58–77%) (Fig. 1). The moderate to good diastereoselectivity (3 : 1 dr to 11 : 1 dr) observed is comparable to that reported for the related stepwise process mediated by Mg(II).¹⁷ The process is compatible with some imines bearing α -hydrogens (e.g. preparation of **27** and **28**) and oxindoles bearing a free N–H (e.g. preparation of **29** and **30**). In all cases, FSPE can be used to purify products by removing any unreacted starting material and the fluorosulfide by-product resulting from removal of the fluorosulfide tag. Pleasingly, the fluorosulfide can be recovered from these cascades in ~70% yield using FSPE and can be reduced to the fluorosulfide thiol (*n*Bu₃P, H₂O, THF, 70%) and reused.¹⁸

Manipulation of products from the Sm(II)/Sm(III)-mediated reaction cascades

In our route to pyrrolidinyl-spirooxindoles, the introduction (see **8** to **9** in Scheme 3) and removal of the tag are used to trigger key cyclisation events in the fluorosulfide synthesis and no additional steps result from the use of a tag as a purification handle.

The products of the Sm(II)/Sm(III)-mediated reaction cascade can be readily converted to compounds related to MDM2 inhibitors **7** and spirotryprostatin A **6**. In the latter case, Danishefsky has shown that analogues of spirotryprostatin A bearing different sidechains at C18 can have activity greater than that exhibited by the natural products.^{6b} Attractively, our approach allows the synthesis of a wide range of analogues by introducing diversity at C18 and on the oxindole ring, and by manipulation of the terminal alkene in the pyrrolidinyl-spirooxindoles.

Oxidative cleavage of the alkenes in pyrrolidinyl-spirooxindoles **25/29** gave aldehydes **31/32** in good yield (Scheme 5). Subsequent oxidation, esterification and debenzoylation gave methyl esters **35/36** (cf. MDM2 inhibitors **7**). The relative stereochemistry of **35** was confirmed by X-ray crystallographic analysis.¹⁵ Coupling with Troc-(*S*)-proline chloride allowed a late-stage resolution to be exploited for the introduction of stereochemical diversity in our library approach. Removal of the Troc protecting group from **37/38** triggered cyclisation to give **39/40a,b** (Scheme 5). The stereochemistry of the four analogues was confirmed by X-ray crystallographic analysis of **40b**¹⁵ and NOE studies on **39a,b**.

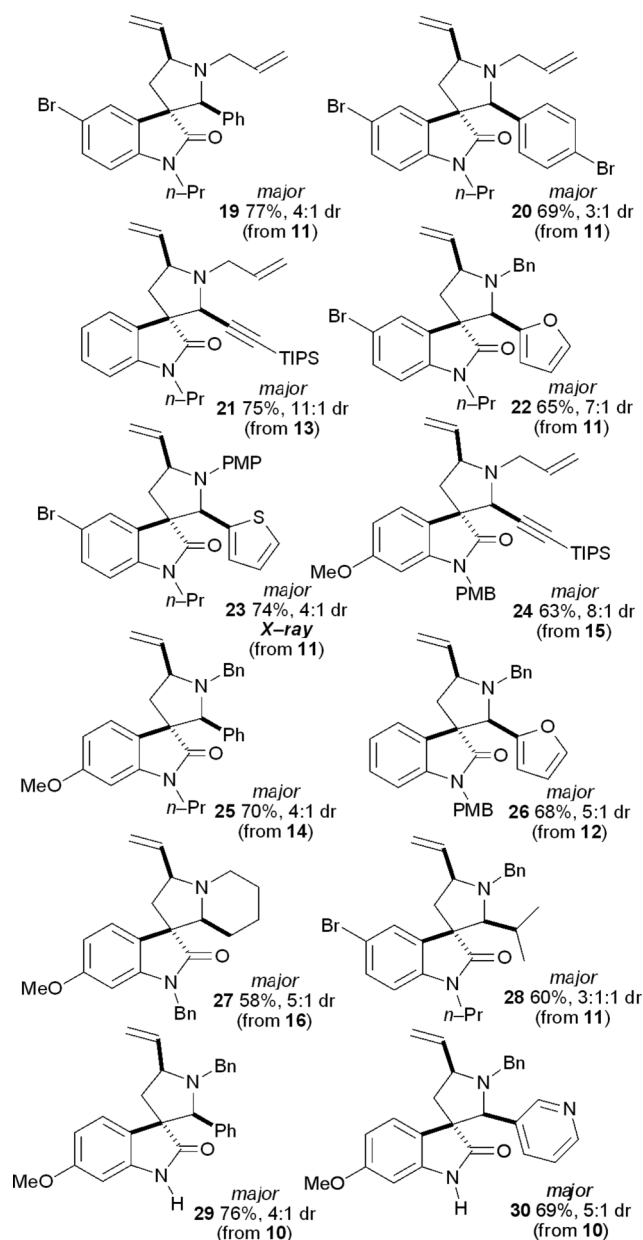
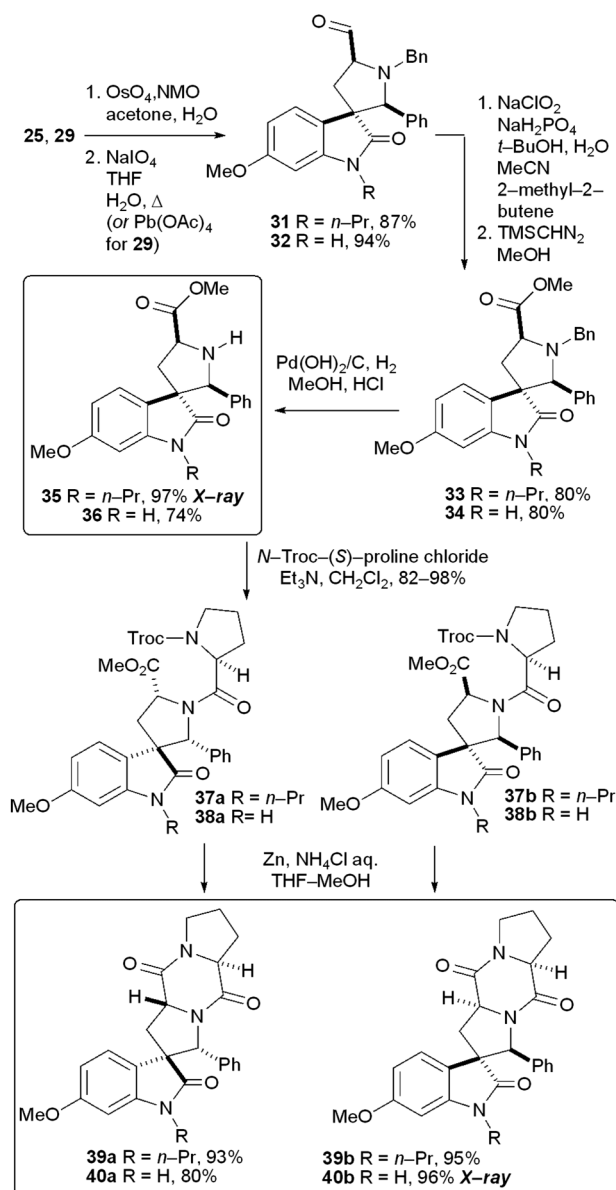


Fig. 1 Scope of the Sm(II)/Sm(III)-mediated reaction cascade.

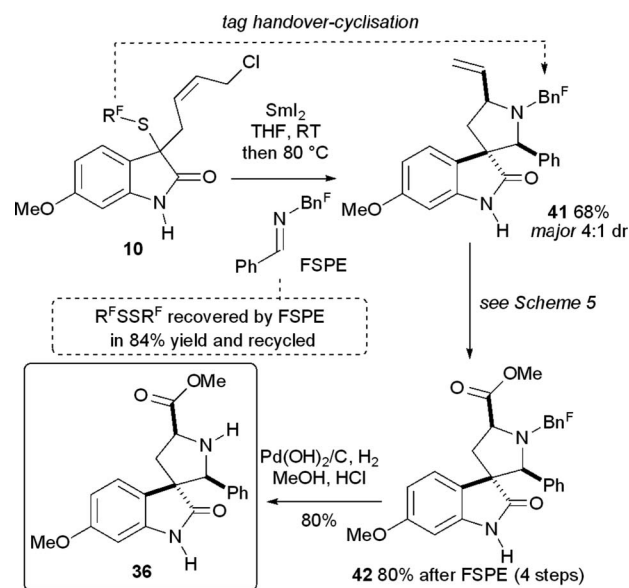
The Sm(II)/Sm(III)-mediated reaction cascade has also been exploited in a tag handover–cyclisation strategy for the preparation of compounds related to MDM2 inhibitors **7** and spirotryprostatin A **6**. A reaction cascade employing tagged oxindole **10** and an imine bearing a fluorosulfide protecting group¹⁹ allowed tags to be exchanged during the cascade to generate **41** thus avoiding the need to remove an old tag and introduce a new tag in a stepwise fashion. The fluorosulfide tag that is removed can be recovered and recycled as before (*vide supra*) as the R^FSSR^F elutes more slowly than **41** on fluorosulfide silica gel due to its higher fluorine content.¹⁸ Straightforward manipulation of **41** gave **42**, after purification by FSPE. Finally, removal of the fluorosulfide benzyl group under acidic hydrogenolysis conditions gave **36** (Scheme 6). The conversion of **36** to **40a,b** has been illustrated in Scheme 5.



Scheme 5 Manipulation of pyrrolidinyl-spirooxindoles.

Conclusions

SmI_2 -initiated reaction cascades have been developed in which the samarium reagent's change from reductant to Lewis acid is exploited. The concept has been exemplified in a phase tag removal–cyclisation process which is initiated by reduction using a $\text{Sm}(\text{II})$ species and completed by a $\text{Sm}(\text{III})$ Lewis acid formed during the reductive step. Thus, the sequence utilises both oxidation states of a samarium reagent in discrete steps. The sequence has been exploited in a fluororous synthesis of pyrrolidinyl-spirooxindoles in which the introduction and removal of the fluororous tag are used to trigger key cyclisation events and thus no additional steps result from the use of a tag as a purification handle. The route has been extended to prepare analogues inspired by the anti-cancer natural product spirotryprostatin A. The biological evaluation of these compounds is underway and preliminary studies have shown that

Scheme 6 A tag handover–cyclisation strategy ($\text{Bn}^{\text{F}} = -\text{CH}_2(4\text{-C}_6\text{H}_4\text{CH}_2-\text{CH}_2\text{C}_8\text{F}_{17})$).

analogues such as **36** display levels of activity similar to that of spirotryprostatin A.²⁰

Acknowledgements

We thank the EPSRC (S. C. C and project studentship to S. Q) and the University of Manchester for financial support. We also thank Sylvia Gang for preliminary studies.

Notes and references

- For a review of metal-mediated radical reactions: (a) A. Gansäuer and H. Bluhm, *Chem. Rev.*, 2000, **100**, 2771; For recent reviews on the use of samarium(II) iodide: (b) D. J. Procter, R. A. Flowers II and T. Skrydstrup, *Organic Synthesis using Samarium Diodide: A Practical Guide*, RSC Publishing, Cambridge, 2010; (c) K. C. Nicolaou, S. P. Ellery and J. S. Chen, *Angew. Chem., Int. Ed.*, 2009, **48**, 7140; (d) R. A. Flowers II, *Synlett*, 2008, 1427; (e) K. Gopalaiiah and H. B. Kagan, *New J. Chem.*, 2008, **32**, 607; (f) D. J. Edmonds, D. Johnston and D. J. Procter, *Chem. Rev.*, 2004, **104**, 3371; (g) A. Dahlén and G. Hilmersson, *Eur. J. Inorg. Chem.*, 2004, 3393; (h) H. B. Kagan, *Tetrahedron*, 2003, **59**, 10351; (i) P. G. Steel, *J. Chem. Soc., Perkin Trans. 1*, 2001, 2727; (j) G. A. Molander and C. R. Harris, *Tetrahedron*, 1998, **54**, 3321; (k) T. Skrydstrup, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 345.
- See ref. 1b,j,k for general discussions. For selected, recent examples of SmI_2 -mediated reaction cascades, see (a) M. D. Helm, D. Sucunza, M. Da Silva, M. Helliwell and D. J. Procter, *Tetrahedron Lett.*, 2009, **50**, 3224; (b) M. D. Helm, M. Da Silva, D. Sucunza, M. Helliwell and D. J. Procter, *Tetrahedron*, 2009, **65**, 10816; (c) M. D. Helm, M. Da Silva, D. Sucunza, T. J. K. Findley and D. J. Procter, *Angew. Chem., Int. Ed.*, 2009, **48**, 9315; (d) T. M. Baker, L. A. Sloan, L. H. Choudhury, M. Murai and D. J. Procter, *Tetrahedron: Asymmetry*, 2010, **21**, 1246; (e) C. Beemelmanns and H-U. Reissig, *Angew. Chem., Int. Ed.*, 2010, **49**, 8021; (f) D. Parmar, K. Price, M. Spain, H. Matsubara, P. A. Bradley and D. J. Procter, *J. Am. Chem. Soc.*, 2011, **133**, 2418.
- For a discussion of fluororous tagging, see (a) A. Studer, S. Hadida, R. Ferritto, S.-Y. Kim, P. Jeger, P. Wipf and D. P. Curran, *Science*, 1997, **275**, 823; for recent reviews see: (b) D. P. Curran, in *The Handbook of Fluorous Chemistry*, ed. J. A. Gladysz, D. P. Curran and I. T. Horváth, Wiley-VCH, Weinheim, 2004; (c) W. Zhang, *Tetrahedron*, 2003, **59**, 4475; (d) W. Zhang, *Chem. Rev.*, 2004, **104**, 2531; for a recent example of tag removal–functionalisation, see: (e) R. Bejot, T. Fowler, L. Carroll, S. Boldon, J. E. Moore, J. Declerck and V. Gouverneur, *Angew. Chem., Int. Ed.*, 2009, **48**, 586.

- 4 D. P. Curran and Z. Luo, *J. Am. Chem. Soc.*, 1999, **121**, 9069.
- 5 For excellent reviews, see: (a) C. V. Galliford and K. A. Scheidt, *Angew. Chem., Int. Ed.*, 2007, **46**, 8748; (b) C. Marti and E. M. Carreira, *Eur. J. Org. Chem.*, 2003, 2209; (c) B. M. Trost and M. K. Brennan, *Synthesis*, 2009, 3003.
- 6 For syntheses of spirotryprostatin A and B, see: (a) S. D. Edmondson and S. J. Danishefsky, *Angew. Chem. Int. Ed.*, 1998, **37**, 1138; (b) S. D. Edmondson, S. J. Danishefsky, L. Sepp-Lorenzino and N. Rosen, *J. Am. Chem. Soc.*, 1999, **121**, 2147; (c) F. von Nussbaum and S. J. Danishefsky, *Angew. Chem. Int. Ed.*, 2000, **39**, 2175; (d) P. R. Sebahar and R. M. Williams, *J. Am. Chem. Soc.*, 2000, **122**, 5666; (e) L. E. Overman and M. D. Rosen, *Angew. Chem. Int. Ed.*, 2000, **39**, 4596; (f) H. Wang and A. Ganesan, *J. Org. Chem.*, 2000, **65**, 4685; (g) P. R. Sebahar, H. Osada, T. Usui and R. M. Williams, *Tetrahedron*, 2002, **58**, 6311; (h) C. Meyers and E. M. Carreira, *Angew. Chem. Int. Ed.*, 2003, **42**, 694; (i) F. Y. Miyake, K. Yakushijin and D. A. Horne, *Angew. Chem. Int. Ed.*, 2004, **43**, 5357; (j) C. Marti and E. M. Carreira, *J. Am. Chem. Soc.*, 2005, **127**, 11505; (k) B. M. Trost and D. T. Stiles, *Org. Lett.*, 2007, **9**, 2763.
- 7 For selected examples, see: (a) M. M.-C. Lo, C. S. Neumann, S. Nagayama, E. O. Perlstein and S. L. Schreiber, *J. Am. Chem. Soc.*, 2004, **126**, 16077; (b) K. Ding, Y. Lu, Z. Nikolovska-Coleska, S. Qiu, Y. Ding, W. Gao, J. Stuckey, K. Krajewski, P. P. Roller, Y. Tomita, D. A. Parrish, J. R. Deschamps and S. Wang, *J. Am. Chem. Soc.*, 2005, **127**, 10130; (c) S. Shangary, D. Qin, D. McEachern, M. Liu, R. S. Miller, S. Qiu, Z. Nikolovska-Coleska, K. Ding, G. Wang, J. Chen, D. Bernard, J. Zhang, Y. Lu, Q. Gu, R. B. Shah, K. J. Pienta, X. Ling, S. Kang, M. Guo, Y. Sun, D. Yang and S. Wang, *Proc. Natl. Acad. Sci. USA*, 2008, **105**, 3933; (d) S. Yu, D. Qin, S. Shangary, J. Chen, G. Wang, K. Ding, D. McEachern, S. Qiu, Z. Nikolovska-Coleska, R. Miller, S. Kang, D. Yang and S. Wang, *J. Med. Chem.*, 2009, **52**, 7970.
- 8 J. Yoshida and K. Itami, *Chem. Rev.*, 2002, **102**, 3693.
- 9 For reviews on the Pummerer reaction, see: (a) S. K. Bur and A. Padwa, *Chem. Rev.*, 2004, **104**, 2401; (b) K. S. Feldman, *Tetrahedron*, 2006, **62**, 5003; (c) S. Akai and Y. Kita, *Top. Curr. Chem.*, 2007, **274**, 35; (d) L. H. S. Smith, S. C. Coote, H. F. Sneddon and D. J. Procter, *Angew. Chem. Int. Ed.*, 2010, **49**, 5832.
- 10 (a) M. Miller, W. Tsang, A. Merritt and D. J. Procter, *Chem. Commun.*, 2007, 498; (b) M. Miller, J. C. Vogel, W. Tsang, A. Merritt and D. J. Procter, *Org. Biomol. Chem.*, 2009, **7**, 589; (c) C. Ovens, N. G. Martin and D. J. Procter, *Org. Lett.*, 2008, **10**, 1441.
- 11 (a) L. A. McAllister, R. A. McCormick, S. Brand and D. J. Procter, *Angew. Chem. Int. Ed.*, 2005, **44**, 452; (b) L. A. McAllister, R. A. McCormick, K. M. James, S. Brand, N. Willetts and D. J. Procter, *Chem. Eur. J.*, 2007, **13**, 1032; (c) K. M. James, N. Willetts and D. J. Procter, *Org. Lett.*, 2008, **10**, 1203; (d) C. Ovens, J. C. Vogel, N. G. Martin and D. J. Procter, *Chem. Commun.*, 2009, 3101.
- 12 For cleavage of an analogous linker in solid-phase synthesis, see: (a) L. A. McAllister, S. Brand, R. de Gentile and D. J. Procter, *Chem. Commun.*, 2003, 2380; (b) K. L. Turner, T. M. Baker, S. Islam, D. J. Procter and M. Stefaniak, *Org. Lett.*, 2006, **8**, 329; (c) L. A. McAllister, K. L. Turner, S. Brand, M. Stefaniak and D. J. Procter, *J. Org. Chem.*, 2006, **71**, 6497. For a review of sulfur and selenium linkers in phase tag-assisted synthesis, see: L. A. McAllister, R. A. McCormick and D. J. Procter, *Tetrahedron*, 2005, **61**, 11527.
- 13 For a review of the chemistry of Sm(III) enolates, see: I. M. Rudkin, L. C. Miller and D. J. Procter, *Organometallic Chemistry*, 2008, **34**, 19.
- 14 (a) P. B. Alper, C. Meyers, A. Lerchner, D. R. Siegel and E. M. Carreira, *Angew. Chem. Int. Ed. Engl.*, 1999, **38**, 3186; (b) A. Lerchner and E. M. Carreira, *J. Am. Chem. Soc.*, 2002, **124**, 14826; (c) A. Lerchner and E. M. Carreira, *Chem. –Eur. J.*, 2006, **12**, 8208. See also ref. 6h and 6j.
- 15 See the Supporting Information for CCDC numbers and X-ray crystallographic data†.
- 16 The ratio of the product diastereoisomers **19a** and **19b** remains constant throughout the course of the reaction, suggesting epimerisation at the quaternary stereocentre in the products is not taking place. The epimerisation of pyrrolidinyl-spirooxindoles through a retro-Mannich/Mannich pathway is well known. See ref. 6j and references therein.
- 17 Carreira obtained several diastereoisomers from his stepwise cyclisation process in a ratio of 8 : 1 : 8 (major : minor : other diastereoisomers). Epimerisation gave improved ratios of 5 : 1 to 7 : 1. See ref. 6j.
- 18 See the Supporting Information for details of the recycling of the fluorous thiol†.
- 19 For an example of the use of the Bn^F protection on nitrogen, see: S. Werner and D. P. Curran, *Org. Lett.*, 2003, **5**, 3293.
- 20 MTT assays assessing cytotoxicity in MDA-468 cells. Spirotryprostatin A (110 μM ref. 6b), **36** (95 μM). The synthesis and biological evaluation of more active spirooxindoles is currently underway.