

Asymmetric Synthesis

DOI: 10.1002/anie.200600451

Biaryl Axis as a Stereochemical Relay for the Enantioselective Synthesis of Antimicrotubule Agents***Agnès Joncour, Anne Décor, Sylviane Thoret, Angèle Chiaroni, and Olivier Baudoin***Dedicated to the memory of Pierre Potier*

Allocolchicine (**1**) and steganacin (**2**) are two naturally occurring chiral biaryls that inhibit the polymerization of tubulin into microtubules in a similar way to colchicine.^[1–3] Recently, colchicine-type antimicrotubule agents got a second wind with the discovery that a prodrug of *N*-acetylcolchinol (**3**; NAC) caused the selective destruction of tumor vasculature.^[4] Steganacin (**2**) contains a stereogenic biaryl axis with a stable *aR* configuration, with atropisomerization being prevented by the eight-membered bridging ring conformation.^[3] In contrast, the seven-membered ring of allocolchicinoids **1** and **3** allows atropisomerization, and these molecules occur as a mixture of equilibrating atropisomers.^[2] The biaryl-axis configuration of **1–3** and analogues was shown to be a crucial

[*] A. Joncour, Dr. A. Décor, S. Thoret, A. Chiaroni,^[†] Dr. O. Baudoin
Institut de Chimie des Substances Naturelles
CNRS, Avenue de la Terrasse
91198 Gif-sur-Yvette (France)
Fax: (+33) 1-690-77247
E-mail: baudoin@icsn.cnrs-gif.fr

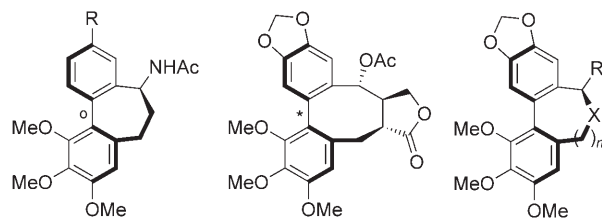
[†] X-ray crystal structure analysis.

[**] We thank M.-E. Tran Huu Dau for the AM1 calculations, F. Guéritte and D. Guénard for support, and E. Bacqué for a useful suggestion. This work was financially supported by the ICSN-CNRS.



Supporting information for this article is available on the WWW under <http://www.angewandte.org> or from the author.

parameter for their tubulin-binding properties, the activity being often restricted to *aR* atropisomers.^[1] We report herein a versatile enantioselective synthesis of bioactive biaryls **4**, simple new hybrid analogues of **1–3** containing a heterocyclic



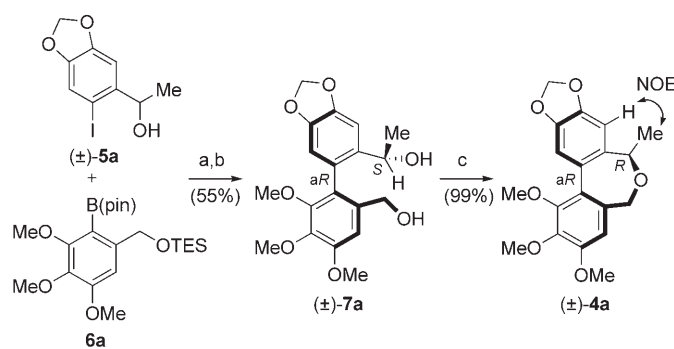
allocolchicine (**1**): R = CO₂Me
N-acetylcolchicinol (**3**): R = OH

steganacin (**2**)

4

bridge, by using the biaryl stereogenic axis as a stereochemical relay.^[5] First, the biaryl configuration is controlled by a benzylic stereocenter through an atropo-diastereoselective Suzuki coupling,^[6] then the biaryl axis relays its stereochemical information to the temporarily destroyed stereocenter in a *S_N1*-type dehydrative cyclization.

Our synthetic strategy was initially implemented with racemic dibenzoxepine (**4a**; Scheme 1), thus following on from our early investigations.^[7] The reoptimized Suzuki coupling of racemic iodide **5a** with boronate **6a** catalyzed by Pd(OAc)₂/**L1**^[8] followed by removal of the triethylsilyl (TES) group on the major diastereoisomer (d.r. = 87:13 for the Suzuki coupling) gave biphenyl diol **7a** in 55% yield. The *S,aR* relative configuration of **7a** was determined by X-ray diffraction analysis.^[9] As expected, no atropisomerization of **7a** was detected at temperatures below 160°C. We found that the dehydrative cyclization of **7a** occurred in the presence of CSA in acetone, probably through an intramolecular *S_N1* process, thus furnishing racemic **4a** in quantitative yield. The *R,aR* relative configuration of **4a** was deduced from NOESY

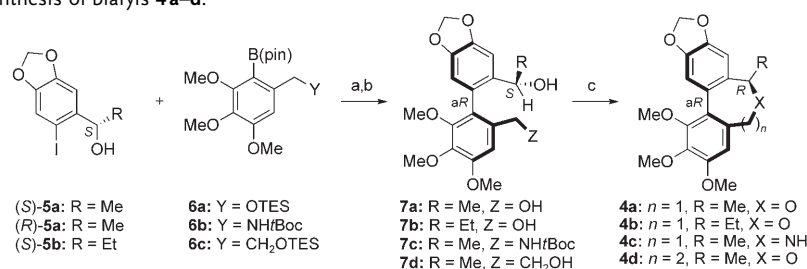


Scheme 1. Synthesis of racemic dibenzoxepine (**4a**). Reagents and conditions: a) (±)-**5a**, **6a** (1.5 equiv), Pd(OAc)₂ (5 mol%), **L1** (10 mol%), Ba(OH)₂·8 H₂O (1.1 equiv), dioxane/H₂O (9:1; *c* = 1 M), 100°C (d.r. = 87:13); b) *n*Bu₄NF, THF, 20°C; c) CSA (1.0 equiv), acetone, 20°C (99%). **L1** = 2-(dicyclohexylphosphino)-2'-(*N,N*-dimethylamino)biphenyl, CSA = camphorsulfonic acid, pin = pinacolato.

experiments (Scheme 1). Similar to other allocolchicinoids,^[2] **4a** occurred as a 96:4 mixture of interconverting *aR/aS* atropisomers in CDCl₃, as shown by the presence of exchange correlations on the NOESY spectrum.^[10] We were delighted to find that racemic **4a** significantly inhibited the assembly of microtubules in vitro, with an IC₅₀ value of 13.1(±2.9) μM versus 8.2(±1.6) μM for (–)-colchicine.

We next embarked on an asymmetric synthesis of (*R,aR*)-**4a** and other analogues, on the assumption that only this enantiomer was responsible for the antimicrotubule activity of (±)-**4a**. Our general strategy for the asymmetric synthesis of tricyclic biaryls **4a–d** with a seven or eight-membered bridging ring containing an oxygen or nitrogen atom is depicted in Table 1. The *S* enantiomer of **5a** was obtained in 72% yield and 97% *ee* from 3,4-methylenedioxyacetophenone by reduction with catalytic (*R*)-CBS-oxazaborolidine (CBS = Corey, Bakshi, Shibata), followed by electrophilic

Table 1: Enantioselective synthesis of biaryls **4a–d**.^[a]



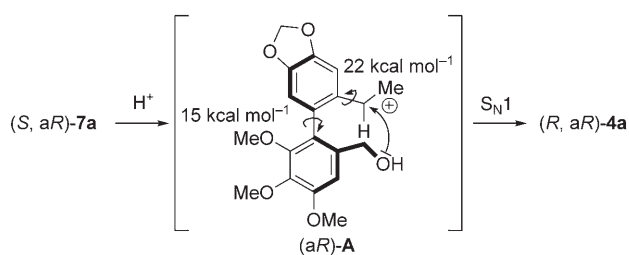
Entry	Iodide	<i>ee</i> [%] ^[b]	Suzuki coupling			Yield [%] ^[d]	d.r. ^[e]	Dehydrative cyclization			
			Boronate	Ligand	Product ^[c]			Product ^[c]	<i>T</i> [°C]	Yield [%] ^[g]	<i>ee</i> [%] ^[b]
1	(<i>S</i>)- 5a	97	6a	L1	(<i>S</i> , <i>aR</i>)- 7a	54	87:13	(<i>R</i> , <i>aR</i>)- 4a ^[f]	–50	86	96
2	(<i>R</i>)- 5a	96	6a	L1	(<i>R</i> , <i>aS</i>)- 7a	34	87:13	(<i>S</i> , <i>aS</i>)- 4a ^[f]	–50	86	94
3	(<i>S</i>)- 5b	98	6a	L1	(<i>S</i> , <i>aR</i>)- 7b	42	74:26	(<i>R</i> , <i>aR</i>)- 4b ^[f]	–78	77	95
4	(<i>S</i>)- 5a	97	6b	L1	(<i>S</i> , <i>aR</i>)- 7c	39	60:40	(<i>R</i> , <i>aR</i>)- 4c	–78	95	88
5	(<i>S</i>)- 5a	97	6c	L2	(<i>S</i> , <i>aR</i>)- 7d	57	81:19	(<i>R</i> , <i>aR</i>)- 4d	–50	84	96

[a] Reaction conditions: a) iodide (1 equiv), boronate (1.5 equiv), Pd(OAc)₂ (5 mol%), **L1** or **L2** (10 mol%), Ba(OH)₂·8 H₂O (1.1 equiv), dioxane/H₂O (9:1; *c* = 1 M), 100°C (**L2** = 2-dicyclohexylphosphino-1,1'-dimethoxy-2,2'-biphenyl); b) for **7a–b** and **7d**: *n*Bu₄NF, THF, 20°C; c) TFA (5 equiv), CH₂Cl₂.

[b] Measured by chiral HPLC, using the racemic mixture as a reference. [c] Relative configuration determined by NOESY experiments, absolute configuration confirmed by superimposition of the CD spectrum on an authentic sample of (–)-NAC (**3**; see the Supporting Information). [d] Yield of the isolated major diastereoisomer from steps (a) and (b). [e] Measured by ¹H NMR spectroscopic analysis of the crude mixture obtained in step (a). [f] Configuration of the major atropisomer (the compound occurs as a mixture of interconverting atropisomers). [g] Yield of the isolated product.

iodination. Atropo-diastereoselective Suzuki coupling with boronate **6a** followed by removal of the TES group on the major diastereoisomer provided (*S,aR*)-**7a** in 54% yield (entry 1). The stereochemically crucial dehydration of this compound was first attempted under the same conditions as the racemic mixture at 20 °C. This step gave **4a** with 74% *ee* in favor of the putative *R,aR* enantiomer. Gratifyingly, carrying out the cyclization at -50 °C with trifluoroacetic acid (TFA) in CH₂Cl₂ allowed almost complete conservation of the optical purity (96% *ee*, 86% yield). The *R,aR* absolute configuration of the product was confirmed by the superimposition of its CD spectrum on that of an authentic sample of (-)-NAC (**3**). Repeating the same reaction sequence from enantiomeric (*R*)-**5a** (synthesized in 96% *ee*) furnished (*S,aS*)-**4a** in 94% *ee* (entry 2). Introduction of another alkyl group on the oxepine ring proved feasible, as illustrated by the synthesis of the ethyl analogue (*R,aR*)-**4b** (entry 3). This analogue was obtained with 95% *ee* from (*S*)-**5b** (98% *ee*).^[11] The dibenzazepine analogue (*R,aR*)-**4c** could be obtained accordingly, starting from (*S*)-**5a** and boronate **6b** (entry 4). In this case, a small loss of optical purity was observed (88% *ee*), although the dehydration occurred at -78 °C. Cleavage of the *tert*-butyloxycarbonyl (*t*Boc) group was observed upon warming the reaction mixture to room temperature. Finally, dibenzoxocine (*R,aR*)-**4d** (eight-membered median ring) was synthesized with 96% *ee* from (*S*)-**5a** and boronate **6c** containing a homologated side chain. In this case, **L2** (S-Phos)^[12] gave a better yield than **L1** in the Suzuki coupling. Compound **4d** occurred as a single atropisomer in solution, contrary to **4a,b**, because of the presence of the larger bridging ring, similar to stegane-type molecules.^[3]

The stereoselectivity of the dehydrative cyclization of diol (*S,aR*)-**7a** can be rationalized by the formation of chiral benzylic cation (*aR*)-**A**,^[13] in which the C⁺-H bond eclipses the biaryl axis to minimize *A*^{1,3} allylic strain (Scheme 2). At

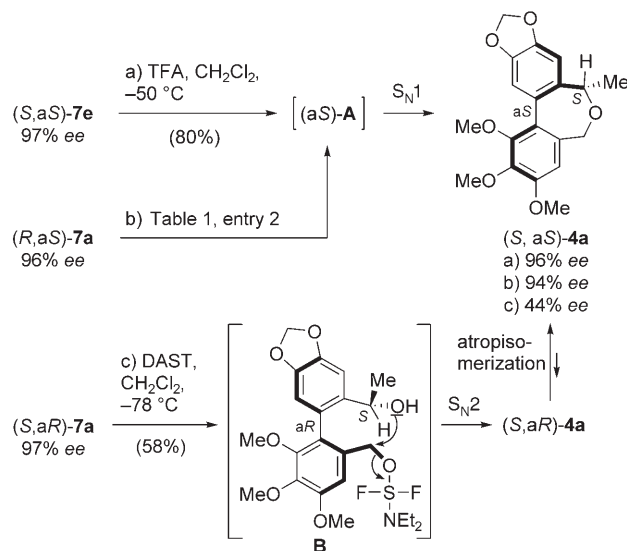


Scheme 2. Proposed cationic cyclization intermediate.

low temperature, this intermediate is configurationally stable and trapped by the internal nucleophile, thus giving (*R,aR*)-**4a** with inversion of configuration at the benzylic stereocenter. An atropisomerization barrier of 15 kcal mol⁻¹ was calculated for **A** (AM1 method), whereas the rotation barrier of the C(Ar)-C⁺ bond was significantly higher (22 kcal mol⁻¹), as expected from conjugation with the aromatic ring. This behavior indicates that the observed racemization of (*R,aR*)-**4a** at higher temperatures might occur preferably by atropisomerization. Overall, the biaryl axis, therefore,

functions as a stereochemical relay for the benzylic stereocenter that is temporarily destroyed in intermediate **A**.

Additional evidence of a chiral carbocationic intermediate in the dehydrative cyclization was provided by the reaction of the minor diastereoisomer (*S,aS*)-**7e** obtained in a small amount after Suzuki coupling of (*S*)-**5a** with **6a** and deprotection (Scheme 3, path a). This reaction furnished



Scheme 3. Stereoconvergent syntheses of (*S,aS*)-**4a**.

(*S,aS*)-**4a** with 96% *ee*, most likely through the same carbocationic intermediate (*aS*)-**A** as that formed from (*R,aS*)-**7a** (path b). A third stereoconvergent pathway could be devised for the synthesis of (*S,aS*)-**4a** (path c). When diol (*S,aR*)-**7a**, which was previously converted into (*R,aR*)-**4a** with TFA (Table 1, entry 1), was treated with (diethylamino)sulfur trifluoride (DAST) in CH₂Cl₂ at -78 °C, (*S,aS*)-**4a** was obtained as the major enantiomer in 44% *ee*. This result can be best rationalized by the regioselective reaction of the primary alcohol of **7a** with DAST to give intermediate **B**, followed by intramolecular S_N2.^[14] This reaction would produce (*S,aR*)-**4a**, which interconverts into the more stable atropisomer (*S,aS*)-**4a**. The loss of optical purity could be ascribed either to incomplete regioselectivity in the reaction of the diol with DAST or to a mixed S_N2/S_N1 mechanism.

The antimicrotubule activity of biaryls **4a-d** was examined and compared to that of (-)-colchicine and (-)-NAC (**3**). First, no activity was found for (*S,aS*)-**4a**, as expected. The IC₅₀ values for the inhibition of the microtubule assembly for the target compounds and the reference compounds were: 2.9(±0.7) μM for NAC (**3**); 8.2(±1.6) μM for colchicine; 12.3(±2.5) μM for (*R,aR*)-**4a**; 4.9(±0.4) μM for (*R,aR*)-**4b**; 11.1(±2.0) μM for (*R,aR*)-**4d**. Dibenzazepine (*R,aR*)-**4c** was found to be inactive. Thus, all oxygen-containing analogues were strong inhibitors of tubulin polymerization, with (*R,aR*)-**4b** being the most active (1.7 × more active than colchicine).^[15]

In conclusion, we have reported a general and efficient enantioselective synthesis of potent antimicrotubule biaryls

by using a novel type of asymmetry relay by a biaryl stereogenic axis. These molecules could represent promising new leads for the development of vascular-targeting agents.

Received: February 2, 2006
Published online: May 10, 2006

Keywords: antimicrotubule agents · asymmetric synthesis · atropisomerism · carbocations · Suzuki coupling

-
- [1] O. Baudoin, F. Guéritte in *Studies in Natural Product Chemistry, Vol. 29* (Ed.: Atta-ur-Rahman), Elsevier Science, Amsterdam, **2003**, pp. 355–417.
- [2] a) O. Boyé, A. Brossi in *The Alkaloids, Vol. 41* (Eds.: A. Brossi, G. A. Cordell), Academic Press, New York, **1992**, pp. 125–176; b) Q. Shi, K. Chen, S. L. Morris-Natschke, K.-H. Lee, *Curr. Pharm. Des.* **1998**, *4*, 219–248.
- [3] a) D. L. Sackett, *Pharmacol. Ther.* **1993**, *59*, 163–228; b) J. Chang, J. Reiner, J. Xie, *Chem. Rev.* **2005**, *105*, 4581–4609.
- [4] Reviews: a) M. A. Jordan, L. Wilson, *Nat. Rev. Cancer* **2004**, *4*, 253–265; b) P. E. Thorpe, *Clin. Cancer Res.* **2004**, *10*, 415–427; c) F. Donate, *Drugs Future* **2005**, *30*, 695–706.
- [5] Recent examples with axially chiral amides: a) J. Clayden, A. Lund, L. Vallverdú, M. Helliwell, *Nature* **2004**, *431*, 966–971; b) M. Petit, A. J. B. Lapierre, D. P. Curran, *J. Am. Chem. Soc.* **2005**, *127*, 14994–14995.
- [6] Reviews: a) O. Baudoin, *Eur. J. Org. Chem.* **2005**, 4223–4229; b) G. Bringmann, A. J. Price Mortimer, P. A. Keller, M. J. Gresser, J. Garner, M. Breuning, *Angew. Chem.* **2005**, *117*, 5518–5563; *Angew. Chem. Int. Ed.* **2005**, *44*, 5384–5427.
- [7] O. Baudoin, A. Décor, M. Cesario, F. Guéritte, *Synlett* **2003**, 2009–2012.
- [8] a) J. P. Wolfe, S. L. Buchwald, *Angew. Chem.* **1999**, *111*, 2570–2573; *Angew. Chem. Int. Ed.* **1999**, *38*, 2413–2416; b) J. P. Wolfe, R. A. Singer, B. H. Yang, S. L. Buchwald, *J. Am. Chem. Soc.* **1999**, *121*, 9550–9561.
- [9] CCDC-296412 ((±)-(S,aR)-**7a**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [10] The structures of both atropisomers were computed using random search calculations. The NOE interactions observed for the major aR atropisomer (Scheme 1) corresponds to a H(Ar)–CH₃ distance of 2.2 Å, whereas this distance is 3.6 Å in the aS atropisomer.
- [11] All target compounds **4a–d** were first synthesized in racemic form to serve as references for chiral HPLC analysis.
- [12] a) S. D. Walker, T. E. Barder, J. R. Martinelli, S. L. Buchwald, *Angew. Chem.* **2004**, *116*, 1907–1912; *Angew. Chem. Int. Ed.* **2004**, *43*, 1871–1876; b) T. E. Barder, S. D. Walker, J. R. Martinelli, S. L. Buchwald, *J. Am. Chem. Soc.* **2005**, *127*, 4685–4696.
- [13] F. Mühlthau, O. Schuster, T. Bach, *J. Am. Chem. Soc.* **2005**, *127*, 9348–9349.
- [14] D. F. Shellhamer, D. T. Anstine, K. M. Gallego, B. R. Ganesh, A. A. Hanson, K. A. Hanson, R. D. Henderson, J. M. Prince, V. L. Heasley, *J. Chem. Soc. Perkin Trans. 2* **1995**, 861–866.
- [15] By comparison (±)-steganacin (**2**) is 1.4 × less active than colchicine: F. Zavala, D. Guénard, J.-P. Robin, E. Brown, *J. Med. Chem.* **1980**, *23*, 546–549.
-