)), 201 (7.5), 174 (10), 173 (3.3), 147 (3.2), 146 (2.1), 145 (4.2), (1), 118 (1), 117 (1), 97 (17), 89 (5.5), 69 (4.2), 63 (3.2), 41 (7.3), 2.6) and 18 (8.4). IR  $\nu \frac{KBr}{max} cm^{-1}$ : 2920, 1755, 1710, 1615, 1590, ), 1400, 1325, 1290, 1210, 1180, 1150 and 1025. <sup>1</sup>H NMR ) MHz, CDCl<sub>3</sub>): Table 1. <sup>13</sup>C NMR [5] (50 MHz, CDCl<sub>3</sub>, pable assignments, OFR): singlets 173.82 (C-10''), 160.37 (C-48.65 (C-8a), 148.26 (C-7), 137.02 (C-3''), 131.45 (C-8), 130.18 3''), 125.95 (C-6), 116.58 (C-4a); doublets 146.70 (C-2'), 144.32 k, C-7''), 123.94 (C-2''), 114.76 (C-3), 113.54 (C-5), 106.83 (C-'9.54 (C-6''); triplets 69.64 (C-1''), 43.37 (C-5''); quartets 17.24  $\nu'$ ), 10.58 (C-4').

ydrolysis of wampetin. Wampetin (1) (30 mg) was dissolved in 1 HOAc to which 2 drops of conc  $H_2SO_4$  were added. The tion mixture was heated for 30 min, cooled, diluted with ice ) and extracted with EtOAc. The EtOAc layer was washed 1  $H_2O$ , dried (Na<sub>2</sub>SO<sub>4</sub>) and the solvent removed to afford le product (20 mg) which on purification by prep. TLC on a gel yielded xanthotoxol (6 mg) mp 240–243° (uncorr.), <sup>KBR</sup> cm<sup>-1</sup>: 3400 (–OH), 1730, 1700 (coumarin carbonyl). Acknowledgement—We thank Dr. Takashi Harayama, Kyoto University, Japan for <sup>1</sup>H NMR, <sup>13</sup>C NMR and mass spectral determinations and Prof. W. Rahman, Chairman, Department of Chemistry, Aligarh Muslim University, for providing the necessary facilities. S. W. I. N. and K. I. are grateful to U.G.C. and CSIR, New Delhi respectively for financial support.

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# DLUBOLIN, A 4-PHENYL-2H-1-BENZOPYRAN-2-ONE FROM DALBERGIA VOLUBILIS

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### (Revised received 15 April 1983)

Key Word Index—Dalbergia volubilis; Leguminosae; young branches; 7-hydroxy-4-methyl coumarin; dalbergin; biochanin-A; 7-hydroxy-4-(3-hydroxy-4-methoxyphenyl)-2H-1-benzopyran-2-one.

stract—From the ether soluble portion of a methanolic extract of young non-green branches of *Dalbergia volubilis*, isterol, 7-hydroxy-4-methyl-2H-1-benzopyran-2-one, dalbergin, p-hydroxy cinnamic acid, biochanin-A and a new 4inylcoumarin, volubolin, have been isolated. The structure of volubolin as 7-hydroxy-4-(3-hydroxy-4-methoxy inyl)-2H-1-benzopyran-2-one has been established on the basis of spectral and chemical evidence. Cooccurrence of iethyl- and 4-phenyl-coumarins with isoflavones is of biogenetic interest.

wers and stem bark of *Dalbergia volubilis* have been rked out earlier [1-5], but no work has been done on young non-green branches. We have now isolated five own compounds: dalbergin, biochanin-A, sitosterol, *p*-lroxycinnamic acid, 7-hydroxy-4-methyl-2H-1-benzoan-2-one and a new 4-phenyl-2H-1-benzopyran-2-one ich we name volubolin.

The new compound, volubolin, analysing for  ${}_{6}H_{12}O_{5}$ , showed a deep blue fluorescence under UV it and  $\lambda_{max}$  at 224, 284 and 340 nm. It gave a red colour h alcoholic ferric chloride and showed strong absorpn bands at 1715 and 3300 cm<sup>-1</sup> indicating the presence a lactone carbonyl and a hydroxylic function in it. The sence of strong absorption bands at 1624, 1607, 1580 1 1546 cm<sup>-1</sup> in its IR spectrum indicated that the

positions 5 and 8 of the coumarin nucleus were free [6]. Since it gave a pink colour with Mg–HCl which changed from blue to brown, the compound was considered to be a 4-phenylcoumarin [7]. In its <sup>1</sup>H NMR spectrum it showed a one proton singlet at  $\delta$  5.93 which could be assigned to a proton at position 3 of the 4-phenylcoumarin. Two one proton singlets at  $\delta$  9.85 and 10.3, exchangeable with deuterium, could be assigned to two phenolic protons while a three proton singlet at  $\delta$  3.8 could be attributed to three methoxyl protons. It showed a prominent molecular ion at m/z 284 followed by a loss of CO to give a strong  $[M - 28]^+$  peak at m/z 256 which further lost a molecule of CO to yield a peak at m/z 228  $[M - 56]^+$ . Other prominent peaks at m/z 269  $[M - 15]^+$  and m/z 241  $[M - 43]^+$  could be accounted for

by the loss of a methyl radical and CO from the molecular ion. This spectral fragmentation is characteristic of a 7hydroxycoumarin [6]. The strong peak at m/z 225 [M  $(-59]^+$  could be due to the combined loss of a methoxyl radical and CO while that at  $m/z 210 [M - 74]^+$  may be attributed to the loss of water from the  $[M - 56]^+$  peak at m/z 228 or to the loss of a methoxyl radical from the [M  $[-43]^+$  fragment. A peak at  $m/z 211 [M - 73]^+$  may be considered to be due to the loss of two CHO radicals while the peak at  $m/z 213 [M - 71]^+$  could be attributed to the loss of two CO molecules from the  $[M-15]^+$  (m/z 269) fragment. All the above evidence suggested this compound to be 7-hydroxy-4-(3-hydroxy-4-methoxyphenyl-)-2H-1-benzopyran-2-one and this was supported by two multiplets each integrating for three protons at  $\delta$  6.7 and 7.2 in its <sup>1</sup>H NMR spectrum which could be assigned to three aromatic protons due to rings A and B respectively. The structure was confirmed by methylation followed by oxidation with neutral potassium permanganate when 3,4-dimethoxybenzoic acid, oxalic acid and 2-hydroxy-3,4,4'-trimethoxybenzophenone were identified by comparison with authentic samples. That the hydroxyl and methoxyl groups were in positions 3 and 4 respectively and not vice versa was proved by mild permanganate oxidation of volubolin when an acid, mp 250°, was isolated which corresponds to 3-hydroxy-4-methoxybenzoic acid (lit. [8] mp 255°).

### EXPERIMENTAL

Air dried non-green young branches of *Dalbergia volubilis* were defatted with petrol (bp 60-80°) and extracted with MeOH which on concn yielded a dark yellow-black residue (4 g), which was extracted with  $Et_2O$ . The  $Et_2O$  extract on CC (silica gel) using  $C_6H_6$  and petrol for elution yielded compounds  $A_1, A_2, A_3, A_4$  and  $A_5$ .

Compound  $A_1$ . Yield 50 mg. Crystallized from CHCl<sub>3</sub>-MeOH as colourless silky needles, mp 139–140°,  $[\alpha]_D^{25} = 30.5^\circ$  (c 0.60, CHCl<sub>3</sub>). It was identified as sitosterol by comparison with an authentic sample (mp, mmp, Co-IR, Co-TLC).

Compound  $A_2$ . Yield 30 mg. Crystallized from EtOAc-petrol as light yellow shining needles, mp 181–182°. It gave a bright blue fluorescence (UV light).  $\lambda_{mac}^{MeOH}$  nm: 325. The acetate crystallized from EtOAc, with mp 150–151°. <sup>1</sup>H NMR ( $\delta$ , CDCl<sub>3</sub>) 2.32 (3H, s, OCOMe), 2.4 (3H, d, J = 1.5 Hz, Me), 6.18 (1H, q, H-3), 6.79 (2H, m, H-6, H-8), 7.54 (1H, d, J = 10 Hz, H-5). It was identified as 7hydroxy-4-methyl coumarin by comparison with an authentic synthetic sample (mp, mmp, Co-TLC and Co-IR).

Compound  $A_3$ . Yield 40 mg. Crystallized from a  $C_6H_6$ -MeOH mixture as light yellow needles, mp 211–212°. It exhibited  $\lambda_{max}^{MeOH}$  at 235, 260 and 380 nm and yielded a methyl ether which crystallized from MeOH to yield prisms, mp 146–147°, (lit. [10] mp of methyldalbergin 145–146°). Compound  $A_3$  was therefore identified as dalbergin by comparison with an authentic sample (mp, Co-IR).

Compound  $A_4$ . Yield 40 mg. Crystallized from EtOAc as colourless needles, mp 214–216°. It analysed for  $C_{16}H_{12}O_5$ . (Found: C, 67.7, H, 4.25;  $C_{16}H_{12}O_5$  requires C, 67.6; H, 4.23  $\frac{1}{\sqrt{6}}$ .) It was identified as 5,7-dihydroxy-4'-methoxy isoflavone (biochanin-A) by spectral measurements and by comparison with an authentic sample [10] (Co-TLC, Co-IR).

Compound  $A_5$ . Yield 25 mg. Crystallized from MeOH as colourless needles, mp 210–212° and was identified as *p*-hydroxycinnamic acid (mp, mmp, Co-TLC).

Compound  $A_6$ . Yield 90 mg. Obtained from the petrol- $C_6H_6$ (7:3) eluate and was further purified by prep. TLC (silica gel, petrol- $C_6H_6$ , 1:1), when a yellow semisolid was obtained. It analysed for  $C_{16}H_{12}O_5$ . (Found: C, 67.4; H, 4.4;  $C_{16}H_{12}O_5$  requires C, 67.6; H, 4.3 %.)  $\lambda_{max}^{MeOH}$  nm: 224 ( $\epsilon$  4.19), 284 ( $\epsilon$  4.40), 340 ( $\epsilon$  3.98).  $\nu_{max}^{RBT}$  cm<sup>-1</sup>: 3300, 1715, 1480, 1345. MS (m/z % abundance) 284 (21.9), 272 (40.00), 270 (2.12), 263 (26.8), 268 (89.40), 267 (13.6), 256 (25.6), 255 (14.4), 141 (14.1), 240 (14.6), 239 (12.1), 230 (17.8), 229 (17.3), 228 (45.6), 227 (18.1), 226 (18.4), 225 (54.0), 213 (33.3), 212 (41.0), 211 (38.4), 210 (90.2), 209 (24.7), 208 (16.0), 207 (30.5), 205 (13.6), 201 (26.0).

KMnO<sub>4</sub> oxidation of volubolin. A soln of volubolin in Me<sub>2</sub>CO (10 mg, 5 ml) and a soln of KMnO<sub>4</sub> in Me<sub>2</sub>CO (50 mg, 10 ml) were warmed on a water bath till the colour of reaction mixture persisted. The solvent was distilled off, decolourized by adding NaHSO<sub>3</sub> (0.5 g) and acidified with HCl. It was extracted with EtOAc and the residue obtained after evaporation of solvent subjected to prep. TLC when an acid, mp 250° was obtained (lit. [8] mp of 3-hydroxy-4-methoxybenzoic acid, 255°).

Oxidation of volubolin methyl ether. A soln of volubolin in  $Me_2CO$  (50 mg, 60 ml) was refluxed with  $Me_2SO_4$  (2 ml) and  $Na_2CO_3$  (2 g) for 20 hr. The reaction mixture was filtered and the filtrate concd under red. pres. to 20 ml.  $KMnO_4$  (1 g) was added to it and the mixture warmed till the colour persisted.  $MnO_2$  was decomposed with NaHSO<sub>3</sub> soln (5 ml, 10%) and the soln was extracted with EtOAc. The EtOAc extract, after concn, was again extracted with 10% KHCO<sub>3</sub> which on acidification with HCl yielded oxalic acid and 3.4-dimethoxybenzoic acid (confirmed by their TLC comparison with authentic synthetic samples). The non-aqueous organic extract on evaporation of solvent yielded a pale yellow compound (10 mg), mp 135-137°. The compound was identified to be 2-hydroxy-3'.4.4'-trimethoxybenzophenone (lit.[11] mp 140°) by comparison with an authentic synthetic sample(mp, mmp, Co-IR).

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