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## A Novel Synthesis of α-Cyanoaziridines

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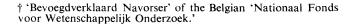
A new route to  $\alpha$ -cyanoaziridines has been developed by reaction of  $\alpha$ -chloroketimines with potassium cyanide in methanol, which involves nucleophilic addition and subsequent intramolecular nucleophilic substitution.

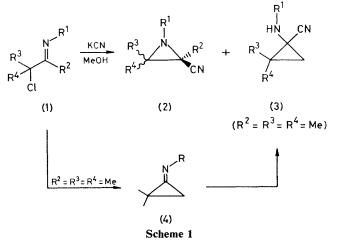
 $\alpha$ -Cyanoaziridines are a class of functionalized aziridines which have received considerable attention in view of their ability to undergo 1,3-cycloadditions to, for instance, olefins or alkynes *via* azomethine ylides.<sup>1,2</sup> The title compounds have already been prepared by reaction of  $\alpha,\beta$ -unsaturated nitriles with nitrenes,<sup>3</sup> *N*-unsubstituted oxaziridines<sup>4</sup> and organic azides,<sup>5</sup> the reaction of  $\alpha$ -halogeno- $\alpha,\beta$ -unsaturated nitriles with amines,<sup>6</sup> the addition of diazoacetonitrile to certain benzylideneanilines,<sup>7</sup> or substitution by cyanide of  $\alpha$ -chloroaziridines.<sup>8</sup> We report a novel and simple synthesis of  $\alpha$ -cyanoaziridines (2) by reaction of the readily available  $\alpha$ -chloroketimines (1)<sup>9a</sup> with potassium cyanide in methanol (Table 1, Scheme 1). All  $\alpha$ -cyanoaziridines reported in this paper are new compounds. In two cases they are obtained as mixtures of *cis*and *trans*-isomers (2;  $\mathbb{R}^3 = Me$ ;  $\mathbb{R}^4 = H$ ;  $\mathbb{R}^1 = \Pr^i$ , cyclohexyl), which can easily be separated by gas chromatographic analysis [*cis*/*trans* ratio 85:15 (or *vice versa*; tentatively assigned); *cis* refers to the substituent at C-3 with respect to the nitrile function]. Alkyl-substituted  $\alpha$ -cyanoaziridines (2)

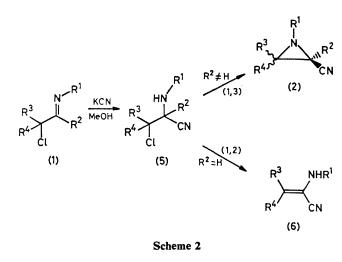
Table 1. S	Synthesis	of <i>α</i> -cyano	aziridines	(2). <sup>a</sup>
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R <sup>1</sup>	$\mathbb{R}^2$	R <sup>3</sup>	R4	Yield <sup>b</sup>	B.p./°C (mmHg)
Pri	Me	Me	н	88	58-66 (13)°
But	Me	Me	н	74	70-75 (11) <sup>d</sup>
Cyclohexyl	Me	Me	Н	73	123-130 (18)°
Pr <sup>i</sup>	Me	Me	Me	73	69—73 (19)
Cyclohexyl	Me	Me	Me	70	119-122 (14)
Pri	Me	Ph	н	82e	f

<sup>a</sup> Reaction conditions: 1.5–2 equiv. of KCN in methanol (reflux 1–2 h); all compounds gave spectral data (<sup>1</sup>H and <sup>13</sup>C n.m.r., i.r., m.s.) in agreement with the structure and gave satisfactory analytical data. <sup>b</sup> Isolated yield after distillation. <sup>c</sup> Mixture of *cis/trans* isomers (85:15). <sup>d</sup> One isomer (tentatively assigned as *trans*). <sup>e</sup> *cis*-Aziridines exclusively. <sup>t</sup> Isolated by preparative column chromatography (Sigel; CCl<sub>4</sub>: diethyl ether 4:1).







do not undergo interconversion of cis- and trans-isomers on heating, but the aryl-substituted derivative cis-2-cyano-1isopropyl-2-methyl-3-phenylaziridine (2) (obtained as the sole product after reaction with KCN; Ph at C-3 cis with respect to the nitrile group) was completely isomerized to the transisomer by gas chromatographic analysis. This isomerism can best be explained in terms of conrotatory opening of the cisisomer and isomerism of the azomethine thus formed.<sup>1</sup> The formation of  $\alpha$ -cyanoaziridines can be interpreted as originating from a nucleophilic addition of cyanide anion to the imino-function, followed by intramolecular nucleophilic substitution (1,3-dehydrochlorination), whereas in the case of  $\alpha$ -chloroaldimines (1;  $R^2 = H$ ) a 1,2-dehydrochlorination of the resulting cyanohydrin, leading to  $\alpha$ -cyanoenamines (6), was observed<sup>9</sup> (Scheme 2). In this respect,  $\alpha$ -chloroketimines parallel, in their reactive behaviour,  $\alpha$ -halogenoketones which yield  $\alpha$ -cyanoepoxides on treatment with cyanide.<sup>10</sup> Tertiary  $\alpha$ -chloroketimines (2; R<sup>3</sup>, R<sup>4</sup>  $\neq$  H), with potassium cyanide in methanol, gave rise to a side reaction, namely the formation of cyclopropane derivatives (3). The isolation of these compounds in ca. 10% yield indicates the occurrence of a competitive Favorskii-type reaction, leading to intermediate cyclopropylideneamines (4), which are trapped as the cyanohydrin adducts (3). Formal 1,3-dehydrohalogenations to afford isolable or intermediate cyclopropylideneamines have recently been described.<sup>11,12</sup> The trapping of cyclopropylideneamines by hydrogen cyanide reported in this paper provides more proof for the intermediacy of the nitrogen analogues of the well known cyclopropanones during Favorskii-type reactions. On the other hand, the high yield synthesis of the title compounds provides new possibilities in the medicinal field because it was recently found that  $\alpha$ -cyanoaziridines possess cancerostatic and immunostimulating properties.<sup>13</sup>

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## References

- 1 F. Texier, J. Guenzet, and B. Merah, C. R. Acad. Sci., Ser. C, 1973, 277, 1371.
- 2 T. Uchida, J. Heterocycl. Chem., 1978, 15, 241.
- 3 H. Person, A. Foucaud, K. Luanglath, and C. Fayat, J. Org. Chem., 1976, 41, 2141.
- 4 K. Jähnisch, E. Schmitz, and E. Gründemann, J. Prakt. Chem., 1979, 321, 712.
- 5 G. Szeimies, Chem. Ber., 1976, 106, 3695.
- 6 T. Wagner-Jauregg, *Helv. Chim. Acta*, 1961, 44, 1237; K.-D. Gundermann, G. Holtmann, H.-J. Rose, and H. Schulze, *Chem. Ber.*, 1960, 93, 1632.
- 7 F. Roelants and A. Bruylants, Tetrahedron, 1978, 34, 2229.
- 8 J. Deyrup and R. B. Greenwald, J. Am. Chem. Soc., 1965, 87, 4538.
- 9 (a) N. De Kimpe, R. Verhé, L. De Buyck, L. Moëns, and N. Schamp, *Synthesis*, in the press; (b) N. De Kimpe, R. Verhé, L. De Buyck, H. Hasma, and N. Schamp, *Tetrahedron*, 1976, 32, 3063.
- 10 H. Galons, H. H. Ibrahim, C. Combet Farnoux, and M. Miocque, C. R. Acad. Sci., 1981, 292, 1517, and references cited therein.
- 11 H. Quast, E. Schmitt, and R. Frank, Angew. Chem., 1971, 83, 728; H. Quast and R. Frank, Liebigs Ann. Chem., 1980, 1939.
- 12 N. De Kimpe and N. Schamp, *Tetrahedron Lett.*, 1974, 3779; *J. Org. Chem.*, 1975, 40, 3749; N. De Kimpe, R. Verhé, L. De Buyck, L. Moëns, and N. Schamp, *Tetrahedron Lett.*, 1981, 1837.
- W. Kampe, M. Thiel, U. Bicker, and G. Hebold, Ger. Offen. 2,632,517 (Appl. 20 Jul. 1976), Chem. Abs., 1978, 88, 136445;
  H. Berger, R. Gall, W. Kampe, U. Bicker, and G. Hebold, Ger. Offen. 2,727,550 (Appl. 18 Jun. 1977), Chem. Abs., 1979, 90, 121395 and Ger. Offen. 2,644,820 (Appl. 05 Oct. 1976), Chem. Abs., 1979, 90, 38772; H. Berger, R. Gall, W. Kampe, U. Bicker, and R. Kuhn, Ger. Offen. 2,731,264 (Appl. 11 Jul. 1977), Chem. Abs., 1979, 90, 151963; W. Schaumann, H. Determann, U. Bicker, W. Rothe, G. Lettenbauer, and W. Kampe, Ger. Offen. 2,736,296 (Appl. 12 Aug. 1977), Chem. Abs., 1979, 90, 197861; E. Bosies, H. Berger, W. Kampe, U. Bicker, A. Grafe, Brit, UK Pat. Appl. 2,026,863 (13 Feb. 1980), Chem. Abs., 1981, 94, 208691.