

A Convergent Approach to Swinholide A. Stereoselective Construction of the C₃–C₁₇ Fragment of Swinholide A

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A strategy for a total synthesis of the cytotoxic dimeric macrolide swinholide A (**1**, Scheme 1) is outlined and a stereoselective construction of the suitably functionalized C₃–C₁₇ fragment **4**, starting with building blocks **6**, **7**, **9** and **10** is described.

Swinholide A (**1**, Scheme 1) is a highly functionalized, 44-membered cytotoxic macrolide with an interesting biological profile.¹ Isolated from the marine sponge *Theonella swinhoi*² and fully characterized by NMR and X-ray crystallographic analyses,³ this compound displays potent cytotoxic activity against L1210 (IC₅₀ = 0.03 µg ml⁻¹) and KB (IC₅₀ = 0.04 µg ml⁻¹) tumour cell lines. It has been proposed¹ that the high cytotoxicity may be related to swinholide's preference to assume a severely folded conformation (resembling a twisted saddle), with turns at the dihydropyran rings and the C₂₁ and C_{21'} atoms. Interestingly, the hydroxy acid monomer of swinholide A was recently isolated from marine sources as a natural product,⁴ suggesting a possible biosynthetic pathway to swinholide through its monomer. Furthermore, misakinolide, a related compound, lacking the two disubstituted double bonds adjacent to the carbonyls C₁ and C_{1'}, has been found in the same sponge, revealing additional information about the occurrence and biosynthesis of these intriguing compounds.⁵ In this communication, we outline a highly convergent strategy towards this target⁶ (Scheme 1), and describe a stereoselective construction of the fully functionalized C₃–C₁₇ fragment **4** (Scheme 2).

Scheme 1 depicts the strategic bond disconnections and retrosynthetic analysis of swinholide A **1**. Sequential disconnections at *a*, *a'*, *b* and *b'* (structures **1** and **2**) lead to the dimeric and monomeric structures **2** and **3**, respectively, with a ketophosphonate–aldehyde condensation⁷ and an esterification⁸ playing crucial roles in the synthetic strategy. Further disconnection at the C₁₇–C₁₈ bond of **3** as indicated by *c*, reveals fragments **4** and **5** as potential key intermediates for the construction of **3**.⁹ Fragment **4** can be further simplified by disconnections *d* and *e* yielding the lactol **8** and the silylenol ethers **6** and **7**. Finally, disassembly of the six-membered ring as shown (disconnection *f*),¹⁰ gives epoxide **9** and the sulfone orthoester **10** as potential starting points for the synthesis of **4**.

Scheme 2† summarizes the construction of **4** starting with (*S*)-dimethyl malate **11** which upon reduction with BH₃·Me₂S/NaBH₄ gave the corresponding 1,2-diol (92%).¹¹ Sequential silylation of this diol with TBDPSCI and TBSOTf (78% overall yield), followed by DIBAL reduction¹² (96%) furnished the bis-silyl ether aldehyde **12**. Asymmetric crotylboration of **12** using (+)-β-methoxydiisopinocampheylborane under Brown's conditions¹³ (90%), followed by methylation (91%), led to the formation of compound **13** (>20:1 diastereoselectivity by ¹H NMR analysis). Ozonolysis of **13**, followed by reductive work-up and protection with *p*-methoxybenzyl trichloroacetimidate,¹⁴ afforded the corresponding *p*-methoxybenzyl ether in 78% overall yield. Complete desilylation with excess TBAF and re-protection with TBDPSCI gave **14** (85%). Epoxide **9** was then obtained after mesylation of the secondary alcohol (97%), and exposure to anhydrous TBAF (91%).

The employment of Ghosez's methodology for the synthesis of α,β-unsaturated δ-lactones,¹⁰ was instrumental in establishing the desired lactone **15**. Thus, the epoxide **9** was treated with the lithio-derivative of methyl-3-phenylsulfonyl ortho-propionate **10** to afford, after acid hydrolysis and DBU-

induced elimination, lactone **15**‡ in 92% overall yield (one pot, three steps). Subsequent DIBAL reduction of **15** led to lactol **8** (95%) and C-glycosidation of **8**, using Bu^tMe₂SiOCH = CH₂ 7¹⁵ and ZnCl₂, furnished aldehyde **16** in 65% yield (*ca.* 4:1 α:β isomeric ratio chromatographically separated).

Aldehyde **16** was then subjected to a Mukaiyama-type aldol reaction using vinylketene acetal **6**¹⁶ and BF₃·Et₂O to afford hydroxy ester **17** in 99.5% yield (*ca.* 1.4:1 diastereoisomeric ratio at C-7, by ¹H NMR). Chromatographic separation of the two isomers and silylation at the C-7 hydroxy group of the major isomer **17** using TBSOTf, led to silyl ether **18** (89% yield). The stereochemistry in compounds **17** and **18** was based on ¹H and ¹³C NMR comparisons with a similar intermediate reported by Paterson's group.^{6c} Liberation¹⁷ of the primary alcohol in **18**, with DDQ (76%), followed by Swern¹⁸ oxidation (92%) and dithiane formation (85%), afforded **19**. Finally, the targeted intermediate **4**‡ was reached in two steps by reduction of **19** with DIBAL and silylation in 75% overall yield.

The following communication¹⁹ describes the construction of the remaining requisite fragment **5** (C₁₈–C₃₂).

This work was supported by the National Institutes of Health, the University of California at San Diego, The Scripps Research Institute, and Merck Sharp & Dohme. M. J. T. is a recipient of the NSERC (Canada) and Merck Sharp & Dohme postdoctoral fellowships.

Received, 18th January 1994; Com. 4/00299G

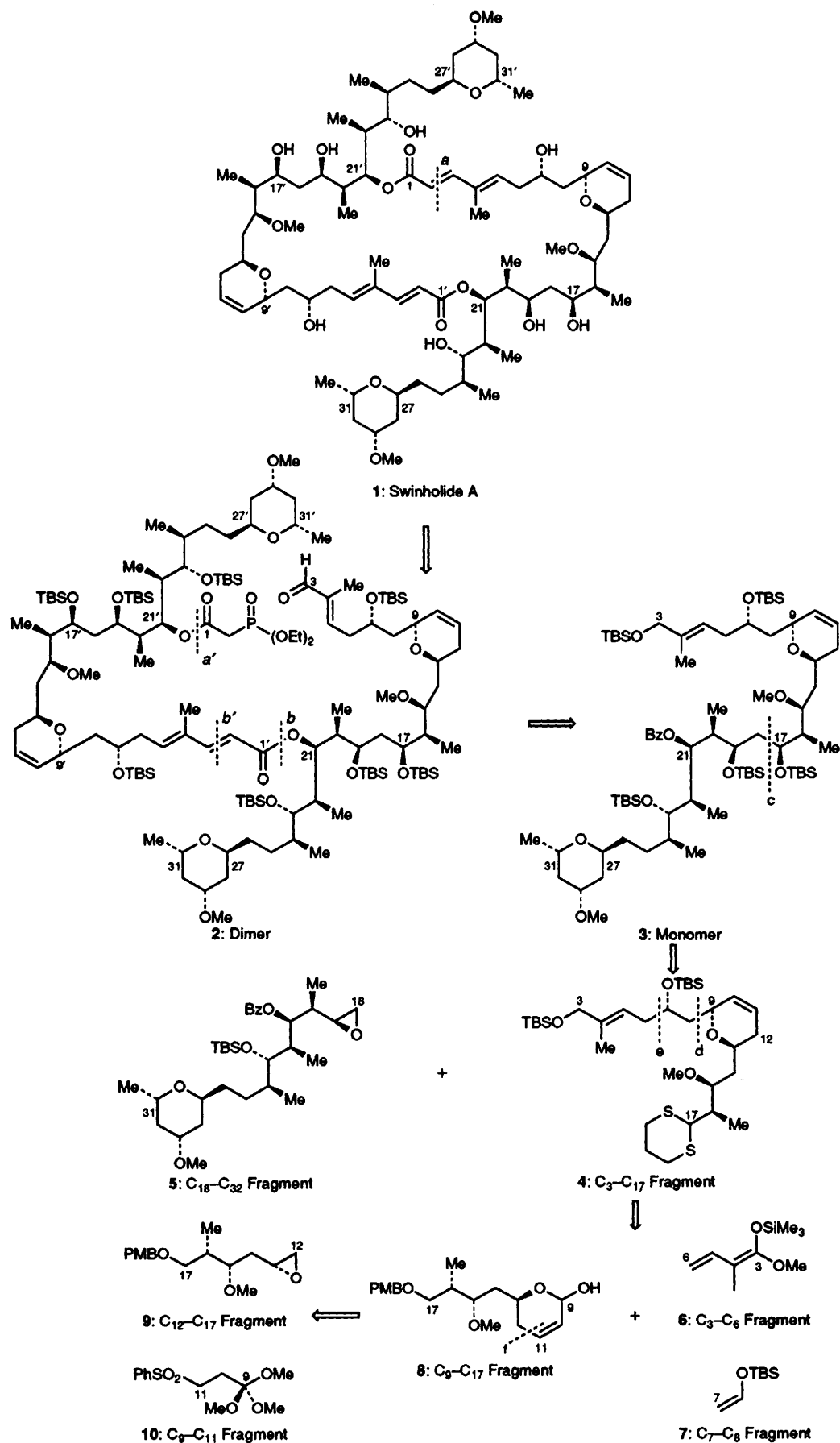
Footnotes

† All new compounds exhibited satisfactory spectral and analytical and/or exact mass data. Yields refer to chromatographically and spectroscopically homogeneous material.

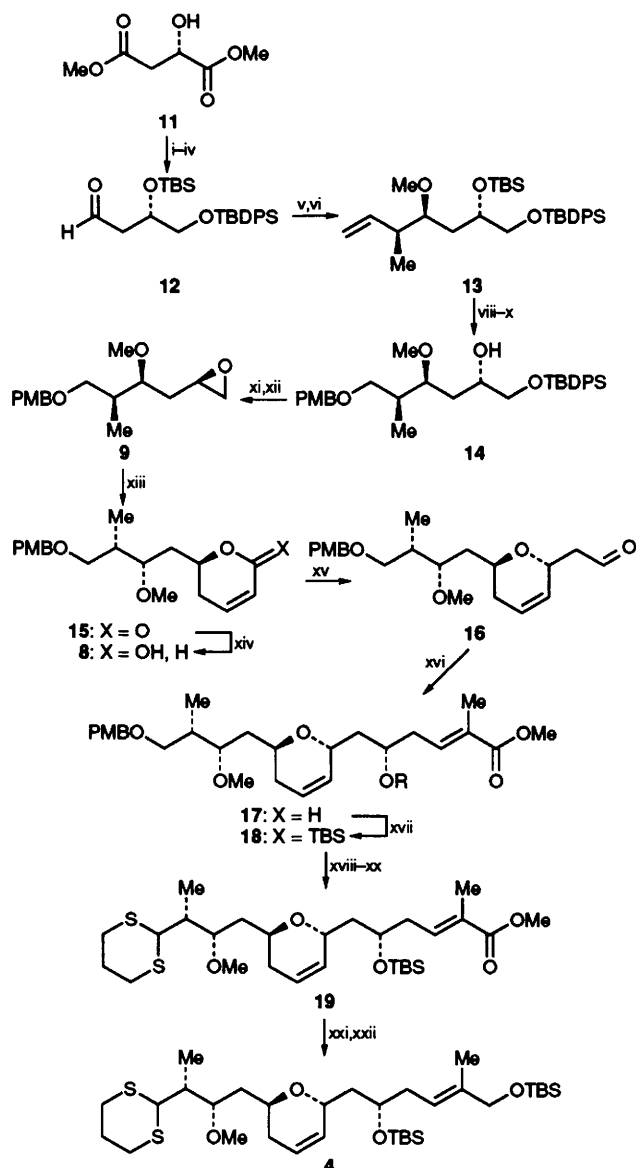
‡ Selected data for compounds:

15: Pale yellow oil; *R*_f 0.30 (silica, 70% Et₂O in light petroleum); IR (neat) ν_{\max} /cm⁻¹: 2932.6, 1710.0, 1611.7, 1585.6, 1512.7, 1462.2, 1387.8, 1301.7, 1246.7, 1173.8, 1088.7, 1035.9, 960.4, 817.6, 755.6 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 7.27–7.25 (d, *J* 8.5 Hz, 2 H, Ar-H), 6.88–6.87 (d, *J* 8.5 Hz, 2 H, Ar-H), 6.87–6.85 (m, 1 H, H-11), 6.04–6.02 (d, *J* 10 Hz, 1 H, H-10), 4.56–4.50 (ddt, *J* 5.5, 5.0, 1.3 Hz, 1 H, H-13), 4.43 (s, 2 H, ArCH₂O), 3.81 (s, 3 H, CH₃-OAr), 3.50–3.41 (m, 2 H, H-17), 3.34–3.29 (m, 1 H, H-15), 3.32 (s, 3H, C-15-OCH₃), 2.38–2.36 (m, 2 H, H-12), 2.07–2.00 (m, 1 H, H-16), 2.03–1.98 (m, 1 H, H-14), 1.80–1.76 (m, 1 H, H-14), 0.95–0.92 (d, *J* 6.5 Hz, 3 H, C-16-CH₃); HRMS (FAB) Calc. for C₁₉H₂₆O₅CS (M + Cs): 467.0835; found *m/z* 467.0842.

4: Pale yellow oil; *R*_f 0.21 (silica, 6% Et₂O in light petroleum); IR (neat) ν_{\max} /cm⁻¹: 2927.1, 2854.8, 1723.6, 1676.8, 1461.8, 1360.4, 1252.1, 1187.5, 1073.1, 1005.5, 938.6, 836.5, 775.2, 699.5 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ 5.77–5.73 (m, 1 H, 11-H), 5.65–5.53 (dt, *J* 10.5, 1.6 Hz, 1 H, H-10), 5.56–5.52 (dt, *J* 8.1, 1.2 Hz, 1 H, H-5), 4.35–4.32 (bd, *J* 10.5 Hz, 1 H, H-9), 4.21–4.19 (d, *J* 7.3 Hz, 1 H, H-17), 4.16–4.10 (m, 1 H, H-7), 4.03 (s, 2 H, H-3), 3.79–3.70 (dt, *J* 7.7, 5.2 Hz, 1 H, H-15), 3.56–3.50 (m, 1 H, H-13), 3.40 (s, 3 H, C-15-OCH₃), 2.92–2.79 (m, 4 H, CH₂S), 2.27–2.22 (m, 2 H, H-6), 2.11–2.06 (m, 1 H, SCH₂CH₂CH₂S), 2.04–1.95 (m, 1 H, H-12), 1.93–1.80 (m, 5 H, H-16, H-14, H-12, SCH₂CH₂CH₂S), 1.68–1.58 (m, 1 H, H-8), 1.60–1.59 (s, 3 H, C-4-CH₃), 1.43–1.33 (ddd, *J* 14.3, 10.5, 2.4 Hz, 1 H, H-8).



Scheme 1 Strategic bond disconnections and retrosynthetic analysis of swinholide A 1. Definition of requisite intermediates for a total synthesis of fragments 4 and 5. TBS = *tert*-butyldimethylsilyl; TMS = trimethylsilyl; PMB = *p*-methoxybenzyl.



Scheme 2 Synthesis of C₃–C₁₇ fragment **4**. **Reagents and conditions:** i, BH₃·SMe₂ (1.04 equiv.), –78 °C, 30 min then NaBH₄ (0.05 equiv.), THF (0.5 mol dm⁻³), 0 °C, 4 h, 92%; ii, TBDPSCI (1.1 equiv.), Et₃N (2.0 equiv.), DMAP (0.2 equiv.), CH₂Cl₂ (0.2 mol dm⁻³), 25 °C, 14 h, 91%; iii, TBSOTf (1.2 equiv.), 2,6-lutidine (1.75 equiv.), CH₂Cl₂ (0.25 mol dm⁻³), 25 °C, 2 h, 86%; iv, DIBAL (1.1 equiv.), CH₂Cl₂ (0.1 mol dm⁻³), quench MeOH (10.0 equiv.), –78 °C, 30 min, 96%; v, KOBu^t (1.0 equiv.), (Z)-but-2-ene (2.0 equiv.), BuⁿLi (1.0 equiv.), THF (0.5 mol dm⁻³), –78 to –55 °C then (+)-β-methoxydiisopinocampheylborane (1.2 equiv.), BF₃·OEt₂ (1.34 equiv.), aldehyde **12**, then –78 °C, 12 h, NaOH (1.8 equiv.), H₂O₂ (1.0 equiv.), –78 to 67 °C, 1 h reflux, 90%; vi, NaH (4.0 equiv.), MeI (10.0 equiv.), THF (0.15 mol dm⁻³), 0 °C, 14 h, 91%; vii, O₃, till blue, then NaBH₄ (4.0 equiv.), CH₂Cl₂ (0.025 mol dm⁻³), MeOH (0.025 mol dm⁻³), –78 °C, 95%; viii, *p*-methoxybenzyl 2,2,2-trichloroacetimidate (4.0 equiv.), CSA (0.15 equiv.), CH₂Cl₂ (0.25 mol dm⁻³), 25 °C, 14 h, 83%; ix, TBAF (2.2 equiv.), THF (0.1 mol dm⁻³), 0 °C, 2 h, 100%; x, TBDPSCI (1.1 equiv.), Et₃N (2.0 equiv.), DMAP (0.2 equiv.), CH₂Cl₂ (0.2 mol dm⁻³), 14 h, 85%; xi, MsCl (1.2 equiv.), Et₃N (2.0 equiv.), CH₂Cl₂ (0.2 mol dm⁻³), 0 °C, 1.5 h, 97%; xii, TBAF (3.0 equiv.), anhydrous, THF (0.1 mol dm⁻³), 25 °C, 10 h, 91%; xiii, 3-phenylsulfonyl-orthopropionate **10** (4.0 equiv.), DMPU (16.0 equiv.), THF (0.33 mol dm⁻³), BuⁿLi (4.0 equiv.), epoxide **9** (1.0 equiv.), –78 to –20 °C then 0 °C, H₂SO₄ (30.0 equiv.), 30 min then workup, next resuspend CH₂Cl₂ (0.33 mol dm⁻³), TsOH (0.38 equiv.), 48 h, then cool –10 °C, Et₃N (1.5 equiv.), DBU (4.0 equiv.), 2 h conc., 92%; xiv, DIBAL (1.25 equiv.), CH₂Cl₂ (0.05 mol dm⁻³), –78 °C, 30 min, 95%; xv, ZnCl₂ (1.0 equiv.), CH₂=CH₂OTBS (2.0 equiv.), CH₂Cl₂ (0.1 mol dm⁻³), –20 °C, 15 min, 65%; xvi, 1-methoxy-1-trimethylsilyloxy-2-methyl-1,3-butadiene **6** (2.06 equiv.),

Footnotes (continued)

1.09–1.07 (d, *J* 6.8 Hz, 3 H, C-16-CH₃), 0.89 (s, 9 H, BuⁿSi), 0.88 (s, 9 H, BuⁿSi), 0.15 (s, 6 H, Me₂Si), 0.07 (s, 6 H, Me₂Si); HRMS (FAB) Calcd. for C₃₃H₆₄O₄Si₂S₂Na (M + Na⁺): 667.3682; found *m/z* 667.3680.

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BF₃·OEt₂ (1.1 equiv.), CH₂Cl₂/Et₂O, 9:1 (0.09 mol dm⁻³), –78 °C, 1.4:1, 99.5%; xvii, 2,6-lutidine (4.0 equiv.), TBSOTf (2.0 equiv.), CH₂Cl₂ (0.01 mol dm⁻³), –78 °C, 1 h, 89%; xviii, DDC (2.0 equiv.), CH₂Cl₂/H₂O, 20:1 (0.018 mol dm⁻³), 25 °C, 45 min, 76%; xix, (COCl)₂ (5.0 equiv.), Me₂SO (7.0 equiv.), CH₂Cl₂ (0.15 mol dm⁻³), –78 °C, 20 min, then Et₃N (15.0 equiv.), –78 to 25 °C, 92%; xx, 1,3-propane-dithiol (5.0 equiv.), TiCl₄ (2.0 equiv.), CH₂Cl₂ (0.03 mol dm⁻³), –78 °C, 1 h 85%; xxi, DIBAL (3.0 equiv.), THF (0.03 mol dm⁻³), –78 °C, 2.5 h; xxii, 2,6-lutidine (4.0 equiv.), TBSOTf (2.0 equiv.), CH₂Cl₂ (0.033 mol dm⁻³), –78 °C, 1 h, 75% two steps. Tf = CF₃SO₂; Ms = MeSO₂; CSA = camphorsulfonic acid; DMPU = 1,3-dimethyl-3,4,5-tetrahydro-2(1H)-pyrimidinone; DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene; DIBAL = diisobutylaluminium hydride.

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