

## 117. Short Syntheses of ( $\pm$ )-Grandisol and ( $\pm$ )-Lineatin *via* a Common Intermediate

by Ivana Aljancic-Solaja<sup>1)</sup>, Max Rey, and André S. Dreiding\*

Organisch-Chemisches Institut der Universität Zürich, Winterthurerstr. 190, CH-8057 Zürich

(16. III. 87)

A 6-step synthesis of ( $\pm$ )-grandisol (**1**) is presented, which involves dichloroketene addition to 3-methyl-3-butenyl acetate (**4**), reductive dechlorination of the adduct **6** to the ketone **7** and saponification to **8**, aldolization of **7** or **8** with acetone and cyclization to the bicyclic ketone **9**, *Wolff-Kishner* reduction to **14**, and finally ring opening to **1**. Since **9** is a known intermediate of the synthesis of ( $\pm$ )-lineatin (**2**), the latter can now be obtained in 6 steps.

We present convenient and stereoselective syntheses of ( $\pm$ )-grandisol (**1**)<sup>2)</sup> and ( $\pm$ )-lineatin (**2**)<sup>3)</sup> involving 6 steps each from commercially available 3-methyl-3-butenol (**3**). The first 4 (new) steps lead to the common intermediate **9**, which has already been transformed [5] to **2** and is now converted to **1**.



The acetate **4** (94% yield from **3**) was reacted with dichloroketene **5**, prepared *in situ* from trichloroacetyl chloride and Zn, to give the dichlorocyclobutanone **6** (*Scheme 1*)<sup>4)</sup>. Dechlorination of **6** with Zn afforded the  $C_3$ -symmetrical acetoxy ketone **7** (63% from **4**), which was saponified to the hydroxy ketone **8** (85%).

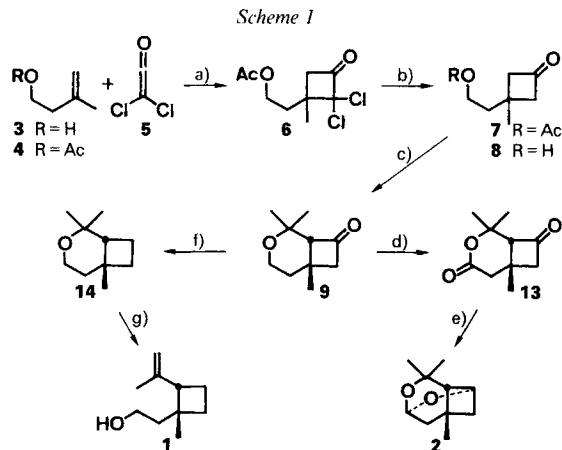
Reaction of **7** or **8** with acetone in the presence of NaOH/H<sub>2</sub>O and a phase-transfer catalyst (Et<sub>3</sub>(PhCH<sub>2</sub>)N<sup>+</sup>Cl<sup>-</sup>) introduced the remaining 3 C-atoms at one of the two enantiotopic CH<sub>2</sub> groups of the 4-membered ring to yield the bicyclic keto ether **9** (39% from **8** or 34% from **7**). This transformation involved an aldol condensation with dehydration (to **10**) and an intramolecular *Michael*-type addition of the OH group to the conjugated double bond. Obviously, the AcO group of **7** was hydrolyzed at some intermediate stage. The conditions for this aldol condensation (see *Exper. Part*) are essential for its success. As by-products, we also observed some of the hydroxy enone **10**, the keto

<sup>1)</sup> On leave of absence from Institute of Chemistry, Technology and Metallurgy, Belgrade, Yugoslavia.

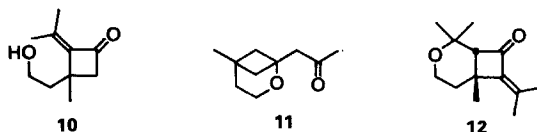
<sup>2)</sup> Aggregation pheromone component isolated from *Anthonomus grandis* (Curculionidae, Coleoptera) [1]. There are 21 syntheses of grandisol (**1**) [2].

<sup>3)</sup> Aggregation pheromone component produced by *Trypodendron lineatis* (Scolytidae, Coleoptera) [3]. There are 10 syntheses of lineatin (**2**) [4–6].

<sup>4)</sup> A [2 + 2] addition of dichloroketene to construct the cyclobutane ring of **2** was also used in [5] and [6].



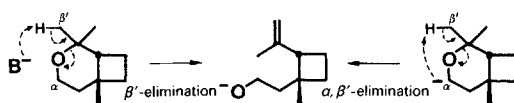
a)  $\text{Cl}_2\text{CCOCl}$ , Zn,  $\text{Et}_2\text{O}$ . b) Zn, AcOH, pyridine. c)  $\text{CH}_3\text{COCH}_3$ , 30% aq. NaOH soln.,  $\text{Et}_3(\text{PhCH}_2)\text{N}^+\text{Cl}^-$ .  
 f)  $\text{NH}_2\text{NH}_2$ ,  $\text{K}_2\text{CO}_3$ , triethylene glycol,  $180\text{--}220^\circ$ . d)  $\text{RuO}_4$  [5]. g)  $\text{LiN}(\text{i-Pr})_2$ , hexane. e) DIBAH,  $\text{H}_3\text{O}^+$  [5].



ether **11**, and the enone **12**. Keto ether **9** and hydroxy enone **10** were found to be interconvertible, a 72:28 equilibrium being reached from both sides under the aldol reaction conditions. Keto ether **11** resulted from an alternative aldol condensation, namely by attack of acetone at the C=O group of **8**, followed by an intramolecular *Michael* addition of the OH group to the enone system. The success of our synthetic approach is due to the fact that the cyclization **10**→**9** causes the desired stereoselectivity by thermodynamic preference of the *cis*-fusion of the six-membered to the four-membered ring.

The bicyclic keto ether **9** had previously been prepared by another procedure and been converted to lineatin (**2**) by  $\text{RuO}_4$  oxidation (→**13**) followed by double carbonyl reduction with diisobutylaluminium hydride (DIBAH) and acidic workup [5]. Thus, our synthesis of **9** represents a formal total synthesis of **2**.

Scheme 2



Our conversion of **9** to ( $\pm$ )-grandisol (**1**) starts with a *Wolff-Kishner* reduction to the bicyclic ether **14** (85%). Simple  $\text{LiN}(\text{i-Pr})_2$  treatment then gave **1** (95%). The ring opening **14**→**1** may have occurred by direct  $\beta'$ - and/or by indirect, intramolecular  $\alpha, \beta'$ -elimination (Scheme 2). Such reactions are known [7] to occur with acyclic dialkyl ethers upon treatment with alkyl lithium or alkyl sodium.

This work was supported by the *Swiss National Science Foundation* and by *Sandoz AG*, Basel.

### Experimental Part

1. *General*. Anal. GC: *SE-54 WCOT* column (25 × 0.3 mm), H<sub>2</sub> as carrier gas, FI detector, split injection. LC: *Merck LiChroprep Si 60* on silica gel (40–63 μ) at 2–6 bar. IR: *Perkin-Elmer 298*. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR: *Varian XL-200* (200 MHz) and *Bruker AM-400* (400 MHz); the <sup>13</sup>C-one-bond multiplicities were obtained from DEPT pulse spectra. MS: *Varian MAT 711* or *1129* (CI = chemical ionisation). GC/MS/IR: *SE-54* (25 m × 0.3 mm) or *OV-1701* (12 m × 0.3 mm) *WCOT* column coupled with a *Digilab-FTS-15-FT-IR* spectrometer with *Digilab GC/C* interface and a *Hewlett-Packard 5970B* mass selective detector, He as carrier gas.

2. *3-Methyl-3-butenyl Acetate (4)*. Procedure given in [8] modified as follows: A mixture of 3-methyl-3-butenol (**3**; 21.5 g, 250 mmol), pyridine (21.8 g, 275 mmol), and Ac<sub>2</sub>O (28.1 g, 275 mmol) was left at r.t. for 18 h, poured into ice/H<sub>2</sub>O, acidified to pH 4 with 1N HCl and extracted twice with Et<sub>2</sub>O. The extract was washed with sat. NaHCO<sub>3</sub> soln. and brine and dried (MgSO<sub>4</sub>). Distillation through a short *Vigreux* column afforded 30.1 g (94%) of **4**, b.p. 143–145°/760 Torr ([8]: 143–145°/760 Torr), as a colourless oil, 99% pure by GC. IR (film): 3090w, 2980m, 1748s, 1657w, 1370m, 1245s, 1050s, 900m. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>): 4.81, 4.74 (2 split s, 2H–C(4)); 4.18 (t, *J* = 7, 2H–C(1)); 2.34 (t, *J* = 7, 2H–C(2)); 2.05 (s, CH<sub>3</sub>COO); 1.76 (t, *J* = 1, CH<sub>3</sub>–C(3)). MS (70 eV): no *M*<sup>+</sup>, 73 (2), 69 (3), 68 (35), 67 (20), 43 (100).

3. *(2,2-Dichloro-1-methyl-3-oxocyclobutyl)ethyl Acetate (6)*. To a stirred suspension of **4** (38.5 g, 300 mmol) and commercial (*Merck & Co.*) Zn(Cu) couple (58.5 g, 895 mmol) in dry Et<sub>2</sub>O (900 ml), a soln. of CCl<sub>3</sub>COCl (70.9 g, 390 mmol) in dry Et<sub>2</sub>O (210 ml) was added dropwise within 4 h at reflux. After stirring for an additional 6 h at reflux, the excess of metal was filtered off, and the filtrate was washed with H<sub>2</sub>O, sat. NaHCO<sub>3</sub> soln., and brine and dried (MgSO<sub>4</sub>). Evaporation left 72.3 g of crude **6** as a dark brown oil. A pure sample of **6** (1.86 g, 68%) was obtained as a colourless oil by bulb-to-bulb distillation at 105°/0.05 Torr of 2.50 g of crude **6** from another experiment performed in the same way on a 27-mmol scale (5.90 g of crude **6**). IR (film): 2970m, 2930m, 1812s, 1742s, 1365m, 1235s, 1135m, 1045m, 990m, 760m. <sup>1</sup>H-NMR (200 MHz; CDCl<sub>3</sub>): 4.25 (t, *J* = 6.7, CH<sub>2</sub>CH<sub>2</sub>O); 3.32, 2.86 (AB, *J* = 16.9, 2H–C(4')); 2.29, 2.12 (2 dt, *J* = 14.5, 6.7, CH<sub>2</sub>CH<sub>2</sub>O); 2.08 (s, CH<sub>3</sub>COO); 1.40 (s, CH<sub>3</sub>–C(1')). MS (70 eV): no *M*<sup>+</sup>, 200 (1), 198 (3), 196 (4), 161 (4), 140 (4), 138 (26), 136 (42), 101 (8), 43 (100). Anal. calc. for C<sub>9</sub>H<sub>12</sub>Cl<sub>2</sub>O<sub>3</sub> (239.10): C 45.21, H 5.06; found: C 44.99, H 5.01.

4. *(1-Methyl-3-oxocyclobutyl)ethyl Acetate (7)*. Crude **6** (72.2 g; see *Exper. 3*) was added dropwise to a stirred suspension of Zn dust (105.3 g, 1.61 mol) in AcOH (330 ml) and pyridine (48 ml) during 45 min at 35–40°. After stirring for 2 h at 40° and 2 h at 70°, the mixture was cooled, diluted with Et<sub>2</sub>O (1200 ml), and the precipitated Zn salts were filtered off. H<sub>2</sub>O (150 ml) was added to the filtrate and the mixture neutralized with solid NaHCO<sub>3</sub>. The precipitated NaOAc was filtered off, the filtrate washed with brine and dried (MgSO<sub>4</sub>). Evaporation left 67.3 g of a pale yellow oil which, after distillation through a short *Vigreux* column, afforded 32.3 g of **7**, b.p. 127–130°/14 Torr, as a colourless oil (63% from **4**; 96% pure by GC). IR (film): 2960m, 2920m, 2875w, 1785s, 1740s, 1390m, 1370s, 1240s, 1145m, 1040m. <sup>1</sup>H-NMR (200 MHz; CDCl<sub>3</sub>): 4.19 (t, *J* = 6.9, CH<sub>2</sub>CH<sub>2</sub>O); 3.0–2.6 (m, 2H–C(2'), 2H–C(4')); 2.05 (s, CH<sub>3</sub>COO); 2.00 (t, *J* = 6.9, CH<sub>2</sub>CH<sub>2</sub>O); 1.35 (s, CH<sub>3</sub>–C(1')). MS (70 eV): no *M*<sup>+</sup>, 128 (13), 110 (4), 68 (77), 43 (100). Anal. calc. for C<sub>9</sub>H<sub>14</sub>O<sub>3</sub> (170.21): C 63.51, H 8.29; found: C 63.19, H 8.32.

5. *(2-Hydroxyethyl)-3-methylcyclobutan-1-one (8)*. To a stirred soln. of **7** (2.55 g, 15 mmol) in MeOH (50 ml), Ba(OH)<sub>2</sub> (1.28 g, 7.5 mmol) was added in small portions during 10 min at 0°. After 30 min, the mixture was warmed to r.t., the MeOH evaporated, and the residue stirred with dry Et<sub>2</sub>O (30 ml) for several min. The insoluble material was filtered off, washed with Et<sub>2</sub>O (10 ml), and the combined filtrates were dried (MgSO<sub>4</sub>). Evaporation and bulb-to-bulb distillation at 90–95°/0.1 Torr afforded 1.63 g (85%) of **8** as a colourless oil, 100% pure by GC. IR (film): 3430s, 2960s, 1780s, 1650w, 1380m, 1140m, 1060m, 1040m. <sup>1</sup>H-NMR (200 MHz; CDCl<sub>3</sub>): 3.79 (t, *J* = 6.7, 2H–C(2')); 3.1–2.7 (m, 2H–C(2), 2H–C(4)); 1.93 (t, *J* = 6.8, 2H–C(1')); 1.62 (s, OH, exchangeable with D<sub>2</sub>O); 1.35 (s, CH<sub>3</sub>–C(3)). MS (70 eV): no *M*<sup>+</sup>, 113 (1), 111 (1), 110 (1), 100 (36), 68 (76), 67 (68), 56 (100), 41 (88). Anal. calc. for C<sub>7</sub>H<sub>12</sub>O<sub>2</sub> (128.17): C 65.59, H 9.44; found: C 65.82, H 9.64.

6. *2,2,6-Trimethyl-3-oxabicyclo[4.2.0]octan-8-one (9)*. 6.1. *From 8*. To a stirred soln. of **8** (0.64 g, 5 mmol) in acetone (10 ml, 136 mmol), 30% aq. NaOH soln. (5 ml, 50 mmol) and triethylbenzylammonium chloride (0.25 ml of a 1M soln. in H<sub>2</sub>O) were added dropwise separately and simultaneously over 10 min at r.t. After stirring for 42 h, the mixture was extracted with Et<sub>2</sub>O (3 × 20 ml), the combined extracts dried (MgSO<sub>4</sub>), and the solvent removed at 100 Torr. The crude product consisted, according to GC and GC/MS/IR, of a mixture of 4-methyl-3-penten-2-one,

4-hydroxy-4-methyl-2-pentanone, 3,5,5-trimethyl-2-cyclohexen-1-one, **8**, **11**, **9**, **10**, and **12** (order of GC elution) in the ratio 14:33:3:5:4:28:5:6. Using tridecane as internal standard in GC, the yield for **9** was determined to be 50%, and the yields of **10**, **11**, and **12** were 10, 7, and 9%, resp.; the starting material was still present to the extent of 14%. LC (hexane/Et<sub>2</sub>O 8:2) afforded a major fraction from which, after bulb-to-bulb distillation at 85°/2 Torr, 0.33 g (39%) of **9** were obtained as a colourless oil, 99% pure by GC. Spectral data: as reported in [5].

6.2. From 7. Acetate **7** (0.85 g, 5 mmol) was reacted and worked up as described in 6.1 (stirring for 21 h). The crude product was treated once more with the same amounts of acetone, NaOH, and Et<sub>3</sub>(PhCH<sub>2</sub>)N<sup>+</sup>Cl<sup>-</sup> (stirring for 20 h) and worked up as in 6.1. The crude product consisted, according to GC and GC/MS/IR, of a mixture of 4-methyl-3-penten-2-one, 4-hydroxy-4-methyl-2-pentanone, 3,5,5-trimethyl-2-cyclohexen-1-one, **8**, **11**, **9**, **10**, and **12** in the ratio of 14:33:2:8:4:25:4:4. Using tridecane as internal standard, the yields estimated for **8**, **11**, **9**, **10**, and **12** were 21, 7, 42, 9, and 5%, resp. LC (hexane/Et<sub>2</sub>O 8:2) gave, aside from 4-methyl-3-penten-2-one, two fractions. Repeated LC (hexane/Et<sub>2</sub>O 8:2) of the faster moving fraction yielded a small amount of pure **12** and, after bulb-to-bulb distillation at 85°/2 Torr, 0.29 g (34%) of **9**, 98% pure by GC. From the other fraction, after repeated LC (hexane/Et<sub>2</sub>O 1:1), small amounts of pure **11** and **10** were obtained as colourless oils.

1-(5-Methyl-2-oxabicyclo[3.1.1]hept-1-yl)propan-2-one (**11**): IR (CHCl<sub>3</sub>): 2960s, 2920s, 2860m, 2730w, 1710s, 1455m, 1425m, 1362m, 1312m, 1240m, 1208m, 1085m, 1055s. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>): 4.03 (t, J = 6.7, 2H-C(3')); 2.58 (s, 2H-C(1)); 2.15 (s, 3H-C(3)); 2.05-1.93, 1.69-1.64 (2 m, 2H-C(6), 2H-C(7')); 1.86 (t, J = 6.7, 2H-C(4')); 1.10 (s, CH<sub>3</sub>-C(5')). <sup>13</sup>C-NMR (50.3 MHz, CDCl<sub>3</sub>): 207.2 (s, C(2)); 77.5 (s, C(1')); 60.6 (t, C(3')); 52.5 (t, C(1)); 43.8 (2 t, C(6'), C(7')); 37.2 (t, C(4')); 36.4 (s, C(5')); 31.5 (q, CH<sub>3</sub>-C(5')); 26.6 (q, C(3)). MS (70 eV): 153 (4), 140 (97), 111 (11), 110 (10), 98 (20), 69 (68), 68 (39), 43 (100). CI-MS: 169. Anal. calc. for C<sub>10</sub>H<sub>16</sub>O<sub>2</sub> (168.24): C 71.39, H 9.59; found: C 71.42, H 9.68.

3-(2-Hydroxyethyl)-2-isopropylidene-3-methylcyclobutan-1-one (**10**): IR (film): 3440m, 2930m, 2870m, 1738s, 1665s, 1440m, 1370m, 1172m, 1078m, 1020m. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>): 3.9-3.6 (m, 2H-C(2')); 2.9, 2.58 (AB, J = 17, 2H-C(4)); 2.08, 1.80 (2 s, (CH<sub>3</sub>)<sub>2</sub>C=C(2)); 2.1-1.8 (m, 2H-C(1')); 1.43 (s, CH<sub>3</sub>-C(3)); 1.35 (s, OH, exchangeable with D<sub>2</sub>O). MS (70 eV): 168 (2, M<sup>+</sup>), 153 (7), 137 (9), 125 (26), 107 (14), 82 (64), 67 (100), 55 (37). Anal. calc. for C<sub>10</sub>H<sub>16</sub>O<sub>2</sub> (168.24): C 71.39, H 9.59; found: C 71.53, H 9.65.

7-Isopropylidene-2,2,6-trimethyl-3-oxabicyclo[4.2.0]octan-8-one (**12**): IR (film): 2930m, 2870m, 1740s, 1665s, 1440m, 1065m, 740m. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>): 3.9-3.6 (m, 2H-C(4)); 2.52 (s, H-C(1)); 2.08, 1.80 (2 s, (CH<sub>3</sub>)<sub>2</sub>C=C(7)); 2.1-1.9 (m, 2H-C(5)); 1.53, 1.40 (2 s, (CH<sub>3</sub>)<sub>2</sub>C(2)); 1.25 (s, CH<sub>3</sub>-C(6)). MS (70 eV): 208 (7, M<sup>+</sup>), 193 (7), 153 (7), 111 (100), 107 (13), 43 (23). Anal. calc. for C<sub>13</sub>H<sub>20</sub>O<sub>2</sub> (208.30): C 74.96, H 9.67; found: C 74.81, H 9.85.

7. Equilibration of **9** and **10**. A sample of pure **9** and a sample of pure **10** (each 3 mg, 0.018 mmol) were stirred separately, each with 30% aq. NaOH soln. (0.1 ml, 1 mmol) and Et<sub>3</sub>(PhCH<sub>2</sub>)N<sup>+</sup>Cl<sup>-</sup> (5 μl of a 1M soln. in H<sub>2</sub>O). After 30 min, the mixtures were extracted with Et<sub>2</sub>O. GC showed the soln. in both samples to contain **9** and **10** in the ratio of 72:28.

8. 2,2,6-Trimethyl-3-oxabicyclo[4.2.0]octan (**14**). To a soln. of **9** (0.97 g, 5.8 mmol) in triethyleneglycol (13 ml), hydrazine hydrate (5.6 ml, 116 mmol) was added and the mixture heated at 90° for 1 h, i.e. until GC showed the absence of **9**. After cooling to r.t., anh. K<sub>2</sub>CO<sub>3</sub> (2.75 g, 19.9 mmol) was added and the mixture heated in a bulb-to-bulb distillation apparatus to 180° for 30 min and then to 220° for 30 min, while the product was allowed to distill into the receiver bulb. The distillate was diluted with 80 ml H<sub>2</sub>O, extracted with pentane (2 × 30 ml), and the extract was washed with 1% HCl soln. (20 ml), H<sub>2</sub>O (2 × 30 ml), and brine (30 ml) and dried (MgSO<sub>4</sub>). The solvent was distilled off over a short Vigreux column at atmospheric pressure, and the residue was purified by bulb-to-bulb distillation at 95°/45 Torr to give 0.76 g (85%) of **14** [9] as a colourless oil, 96% pure by GC. IR (film): 2980s, 2950s, 2870s, 1465m, 1378m, 1365m, 1220m, 1095m, 1078s, 812m. <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>): 3.6-3.4 (m, 2H-C(4)); 2.0-1.1 (m, H-C(1), 2H-C(5), 2H-C(7), 2H-C(8)); 1.09, 1.06 (2 s, (CH<sub>3</sub>)<sub>2</sub>C(2)); 0.94 (s, CH<sub>3</sub>-C(6)). <sup>13</sup>C-NMR (50.3 MHz, CDCl<sub>3</sub>): 70.7 (s, C(2)); 57.7 (t, C(4)); 47.7 (d, C(1)); 35.5 (t, C(6)); 34.0, 32.7 (2 d, C(5), C(7)); 28.0, 26.3, 24.8 (3 q, (CH<sub>3</sub>)<sub>2</sub>C(2), CH<sub>3</sub>-C(6)); 18.2 (t, C(8)). MS (70 eV): no M<sup>+</sup>, 88 (4), 74 (6), 70 (9), 61 (14), 45 (24), 43 (100). Anal. calc. for C<sub>10</sub>H<sub>18</sub>O (154.25): C 77.86, H 11.76; found: C 77.80, H 11.76.

9. 2-Isopropenyl-1-methylcyclobutaneethanol (= Grandisol; **1**). To a soln. of Li(i-Pr)<sub>2</sub>N, prepared by dropwise addition of 1.4M BuLi in hexane (6.9 ml, 9.7 mmol) to (i-Pr)<sub>2</sub>NH (1.4 ml, 10 mmol) at -78° with stirring, was added **14** (135 mg, 0.88 mmol). The mixture was kept 1 h at -70°, allowed to warm to r.t. and then heated at 75-80° (oil bath) for 36 h. The mixture was poured into cold sat. NH<sub>4</sub>Cl, extracted with Et<sub>2</sub>O (2 × 20 ml), washed with 1% HCl soln. (20 ml) and brine (2 × 20 ml) and dried (MgSO<sub>4</sub>). After removing the solvent at atmospheric pressure, the residual yellow oil was bulb-to-bulb distilled at 120-130°/14 Torr to afford 128 mg (95%) of **1** as a colourless oil, 96% pure by GC. IR, <sup>1</sup>H-NMR, MS: as reported in [9].

## REFERENCES

- [1] J. H. Tumlinson, D. D. Hardee, R. C. Gueldner, A. C. Thompson, P. A. Hedin, J. P. Ninyard, *Science* **1969**, 166, 1969.
- [2] F. X. Webster, R. M. Silverstein, *J. Org. Chem.* **1986**, 51, 5226; M. Demuth, A. Palomer, H. Sluma, A. K. Dey, C. Krüger, Y. Tsay, *Angew. Chem.* **1986**, 98, 1093; A. I. Meyers, S. A. Fleming, *J. Am. Chem. Soc.* **1986**, 108, 306; G. Rosini, E. Marotta, M. Petrini, R. Ballini, *Tetrahedron* **1985**, 41, 4633 and ref. cited therein; Review: K. Mori, in 'The Total Synthesis of Natural Products', Ed. J. ApSimon, Wiley-Interscience, New York, 1981, Vol. 4, p. 80.
- [3] J. G. MacConnel, J. H. Borden, R. M. Silverstein, E. Stokkink, *J. Chem. Ecol.* **1977**, 3, 549; V. Schurig, R. Weber, D. Klimetzek, U. Kohnle, K. Mori, *Naturwissenschaften* **1982**, 69, 602.
- [4] L. Skattebøl, Y. Stenstrøm, *Acta Chem. Scand., Ser. B* **1985**, 39, 291 and ref. cited therein.
- [5] B. D. Johnston, N. K. Slessor, A. C. Oehlschlager, *J. Org. Chem.* **1985**, 50, 114.
- [6] K. Mori, T. Uematsu, M. Minobe, K. Yanagi, *Tetrahedron* **1983**, 39, 1735.
- [7] J. March, 'Advanced Organic Chemistry', 3rd edn., J. Wiley & Sons, New York, 1985, p. 903; A. Maercker, W. Demuth, *Liebigs Ann. Chem.* **1977**, 1909.
- [8] L. Maguet, M. Lerer, *Bull. Soc. Chim. Fr.* **1965**, 3262.
- [9] R. Zurfluh, L. L. Dunham, V. L. Spain, J. B. Siddall, *J. Am. Chem. Soc.* **1970**, 92, 425.