

Microbial hydroxylation of steroids. Part 12. Hydroxylation of testosterone and related steroids by *Gnomonia fructicola* ATCC 11430

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The fungus *Gnomonia fructicola* ATCC 11430 hydroxylates testosterone at C-2 β in 40–60% yield. The enzyme involved has been shown to be an inducible CO sensitive, cyanide insensitive oxygenase, with a requirement for the Δ^4 -3-ketosteroid substrate functionality. The use of 2 α - and 2 β -deuterium labelled substrates has shown that hydrogen loss from C-2 during 2 β hydroxylation is isotope dependent and non-stereospecific. This is interpreted in terms of a $\Delta^{2,4}$ -dienolic enzyme-bound intermediate form of the substrate.

Key words: *Gnomonia fructicola*, hydroxylation, steroids, testosterone.

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Le champignon *Gnomonia fructicola* ATCC 11430 provoque une hydroxylation en position C-2 β de la testostérone, avec un rendement de 40–60%. On a démontré que l'enzyme implique une oxygénase qui peut être induite, qui est sensible au CO et insensible au cyanure et dont le substrat doit contenir une fonctionnalité Δ^4 -céto-3 stéroïde. Utilisant des substrats marqués au deutérium dans les positions 2 α et 2 β , on a pu démontrer que l'hydroxylation dépend de l'isotope et qu'elle n'est pas spécifique. On peut interpréter ce résultat, en suggérant que le substrat est lié à l'enzyme par le biais d'une forme intermédiaire $\Delta^{2,4}$ -diénolique.

Mots clés: *Gnomonia fructicola*, hydroxylation, stéroïdes, testostérone.

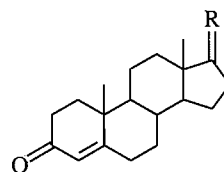
[Traduit par la revue]

The introduction of a hydroxyl functionality into a steroid molecule by fungal biotransformation, first discovered over 40 years ago (1), is still the most efficient route for the preparation of many hydroxysteroids, particularly in the corticosteroid family.

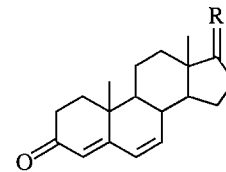
The enzymes responsible for this reaction were slow to yield their secrets, but their essential nature is now understood (2). Their characterization as cytochrome P-450 dependent mono-oxygenases preceded investigation of enzyme-substrate interactions: the latter have centred on the nature of enzyme-substrate binding through existing electronegative substituents such as substrate carbonyl or hydroxyl groups (2), and on the activation of ketosteroids towards electrophilic hydroxylation at positions adjacent or vinylogous to carbonyl by substrate enolization (3).

A report of the hydroxylation of testosterone (1) at C-2 β by *Gnomonia fructicola* ATCC 11430 in unspecified yield (4), and the continued question of the intermediacy of a 2 β -hydroxylation step in the conversion of testosterone to estradiol by the human aromatase enzyme (5, 6), suggesting a possible role for enzymic 2 β hydroxylation of steroids in the control of human breast tumour cell proliferation (7, 8), prompted us to examine both mechanistic and synthetic parameters of the fungal hydroxylation of testosterone at C-2 β by *G. fructicola*. In addition, hydroxylation of testosterone at C-2 β by mammalian liver enzymes represents one of the major routes of testosterone metabolism in the liver (9, 10), suggesting a further role for the fungal biotransformation as a possible model for mammalian metabolism.

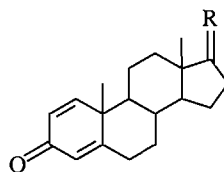
The original report of the steroidal 2 β hydroxylating capability of *G. fructicola* was concerned largely with the use of Δ^4 -3-ketopregnane substrates, reporting 2 β hydroxylation of cortexone in 40% isolated yield (4, 11), with a more complex pattern of metabolism for progesterone, including hydroxylation



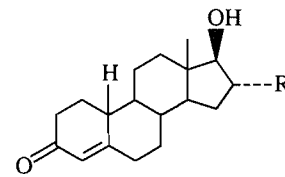
- 1 R = α H, β OH
 2 R = α H, β OAc
 3 R = O
 4 R = α H, β COCH₃



- 5 R = α H, β OH
 6 R = O



- 7 R = α H, β OH
 8 R = O



- 9 R = H
 10 R = OH

tion at the C-2 β , -6 β , -14 α , and -16 β positions (11). To establish the best substrate for mechanistic purposes, we first investigated the fungal metabolism of the simple Δ^4 -3-ketosteroids 1–4, using a simplified procedure for biotransformation that replaced the original conditions (incubation in a complex growth medium (11)) with growth in a simpler medium followed by replacement culture incubation with resting cells in distilled water (for details, see experimental section). Under our conditions, testosterone was routinely converted to the 2 β -hydroxy derivative 23 in isolated yields of 40–60%, accompanied by a trace of oxidation at C-17 to give the dione 4. The results of this and other incubations are summarized in Table 1. Testosterone

TABLE 1. Transformations of steroids by *G. fructicola*

Substrate	Product (%)
1	3 (1), 23 (40–60)
2	3 (1), 23 (30), 31 (2)
3	28 (4), 30 (5)
4	32 (5), 33 (8), 34 (12), 35 (21)
5	6 (16)
7	8 (7)
9	10 (6)
11	36 (3), 37 (20), 38 (3)
12	39 (7), 40 (19), 41 (17)
13	None
14	13 (28)
15	None
16	None
17	1 (2), 3 (1), 23 (17), 28 (2)
18	19 (5)
18*	19 (5), 42 (8)
20	21 (3), 22 (10)

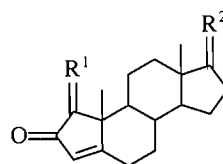
*Following enzyme induction by testosterone.

acetate is similarly transformed but in lower yield, presumably following enzymic hydrolysis of the ester function at C-17, but androst-4-ene-3,17-dione (3) underwent oxidation predominantly in the D-ring, yielding the 17 β -hydroxy-16-ketone 30. This compound may be formed by the rearrangement of an intermediate but not isolated 16-hydroxy-17-ketosteroid, a common phenomenon during fungal biotransformation (12). The biotransformation of progesterone gave, as originally reported (11), a mixture of products, from which the 2 β and 14 α alcohols and 2 β ,6 β and 6 β ,14 α diols were identified.

In view of the experiments outlined above, further work concentrated on steroids of the androstane skeleton, which carried the 17 β -alcohol functionality. Substrate specificity was investigated by using a series of substrates (5, 7, 9, and 11–17); in only one instance (11) was 2 β hydroxylation observed, and then in minor amount together with other reactions. Alternative modes of biotransformation that were noted included alcohol oxidation (for 5, 7, 12, and 17), carbonyl reduction (for 11, 12,

and 14), 16 α hydroxylation (for 9 and 11), and 5 β hydroxylation of 12.

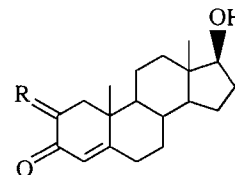
Product structure determination relied heavily on the interpretation of ¹³C nmr data (see Experimental), the following shifts being particularly characteristic: for 16 α hydroxylation (products 10, 37, and 38) the regiochemistry of hydroxylation is given by downfield shifts of signals assigned to C-17 and C-15 compared with their position in the corresponding 16-unsubstituted steroids (13), and stereochemistry is confirmed by a γ upfield shift at C-14 of 2.8–3.3 ppm; for 5 β hydroxylation (products 40 and 41), regiochemistry is reflected in downfield



20 R¹ = H₂, R² = α H, β OH

21 R¹ = H₂, R² = O

22 R¹ = R² = α H, β OH



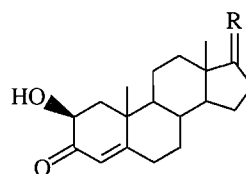
23 R = α H, β OH

24 R = α H, β D

25 R = α D, β H

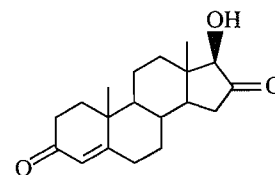
26 R = α D, β OH

27 R = α H, β ¹⁸OH

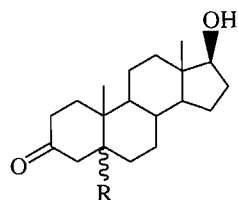


28 R = O

29 R = α H, β OAc

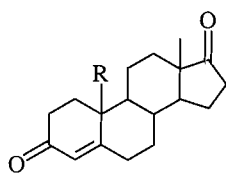


30



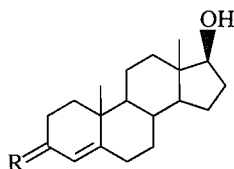
11 R = α H

12 R = β H



13 R = CH₂OH

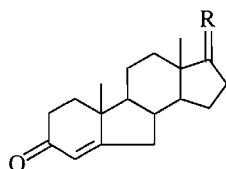
14 R = CHO



15 R = H₂

16 R = CH₂

17 R = α H, β OH



18 R = α H, β OH

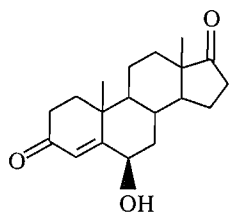
19 R = O

shifts at C-4 and C-6, together with a γ upfield shift at C-1, and stereochemistry by the latter and by the characteristic C-9 signal of 5 β steroids (13); for 2 β hydroxylation (products 23, 28, 34, and 38), regiochemistry is given by upfield shifts at C-1, and characteristic shifts at C-4, -5, -9, and -10, and stereochemistry by a downfield shift of the C-19 signal by 3–5 ppm (13).

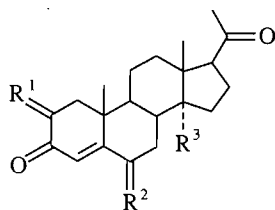
The α stereochemistry of reduction of the C-3 carbonyl group in 5 α steroids (product 38) is reflected in γ upfield shifts for C-1 and C-5, while the β stereochemistry of reduction (products 36 and 37) is apparent from direct comparison with literature values reported for C-1–C-5, C-10, and C-19 for 3 β -hydroxy-5 α -androstanes (13). The stereochemistry of reduction at C-3 in the 5 β -androstane series, giving the 3 α -hydroxy product 41, is confirmed by the absence of a γ -*gauche* upfield shift on the C-1 resonance in the spectrum of 41 compared with that of 40.

It is apparent from the above that the 2 β -hydroxylase enzyme requires a Δ^4 -3-keto steroid substrate with a C-19 methyl group for normal activity. It is noteworthy that neither 19-nortestosterone (9) nor the 19-functionalized steroids 13 and 14 are substrates for this enzyme. Complete removal of the C-3 oxygen function (compounds 15 and 16) produces steroids that are not hydroxylated, while reduction at C-3 (compound 17) is reversed by fungal oxidoreductases prior to hydroxylation (see Table 1).

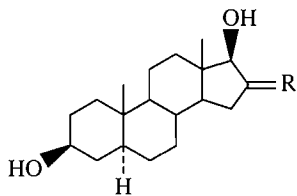
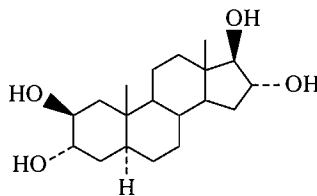
The occurrence of 16 α hydroxylation (substrates 9, 11, and possibly 3) may involve the 2 β -hydroxylase enzyme binding the substrate in an inverted manner, a common phenomenon for fungal steroid hydroxylases (2), and other steroid metabolizing enzymes (14) as suggested in Fig. 1, but in the absence of



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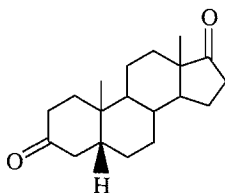


- 32 $R^1 = \alpha\text{H}, \beta\text{OH}, R^2 = \text{H}_2, R^3 = \text{H}$
 33 $R^1 = R^2 = \text{H}_2, R^3 = \text{OH}$
 34 $R^1 = R^2 = \alpha\text{H}, \beta\text{OH}, R^3 = \text{H}$
 35 $R^1 = \text{H}_2, R^2 = \alpha\text{H}, \beta\text{OH}, R^3 = \text{OH}$

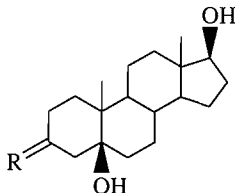
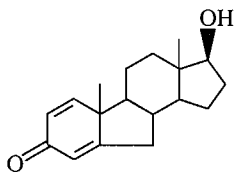
36 $R = \text{H}_2$ 37 $R = \alpha\text{H}, \beta\text{OH}$ 

38

isolated enzymes this proposal must remain speculative. As previously noted (4), the 2β hydroxylase is able to tolerate an A -nor- $\Delta^{3(5)}$ -2-one skeleton such as that of A -nortestosterone (20), giving a low yield of hydroxylation product 22, analogous in structure to that observed for the normal Δ^4 -3-keto skeleton. The B -norsteroid 18 was not hydroxylated under normal incubation conditions, the only biotransformation product (19) being the result of oxidation at C-17, but when testosterone was used as an enzyme inducer during the fungal growth period, as described in the experimental section, and 18 subsequently used as substrate, a low yield of the B -nor- $\Delta^{1,4}$ -diene 42 was



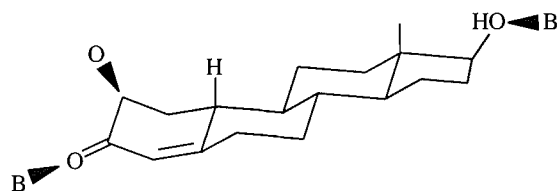
39

40 $R = \text{O}$ 41 $R = \alpha\text{H}, \beta\text{OH}$ 

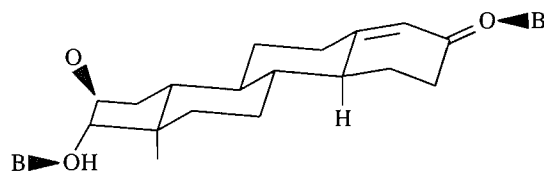
42

obtained. This may have arisen by acid-catalysed dehydration of a 2β -hydroxy intermediate in the growth medium (pH 4.6 at the end of the incubation), but in the absence of an isolated intermediate this proposal is speculative. The use of testosterone as an enzyme inducer was also investigated for incubations with substrates 7, 9, 11, and 20, but in each case the products were the same as those obtained from non-induced fungus.

We have addressed the question of enzymic activation of the



Normal binding



Inverted binding

B = binding site

O = oxidation site

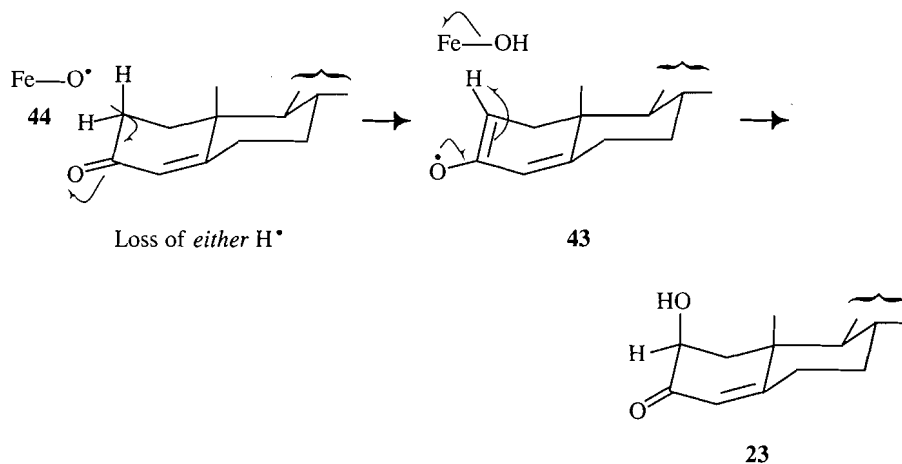
FIG. 1. Possible method for 16α hydroxylation by a 2β -hydroxylase.

substrate towards electrophilic oxidation via formation of the $\Delta^{2,4}$ -dienol 43 (Scheme 1) by the use of testosterone substrates labelled with deuterium at C- 2β and C- 2α (24 and 25, respectively), and the results of these experiments are summarized in Table 2. A meaningful interpretation depends upon a fuller knowledge of the nature of the 2β -hydroxylase enzyme, and to this end the experiments summarized in Table 3 were performed.

The enzyme uses molecular oxygen as the source of hydroxyl, giving 2β -hydroxytestosterone- $2\text{-}^{18}\text{O}$ (27, ^{18}O content 50%) when the reaction was performed under an $^{18}\text{O}_2$ enriched atmosphere (^{18}O content 62%); it is insensitive to the presence of cyanide but is inhibited by carbon monoxide at saturation concentration in water, and is substrate inducible, the reaction being completely suppressed by the presence of the protein synthesis inhibitor, cycloheximide, unless preinduction with testosterone was carried out. All of these properties are consistent with the 2β -hydroxylase enzyme being a cytochrome P-450 dependent mono-oxygenase (15a), although in the absence of an isolated, purified enzyme this cannot be definitively proven.

The active site species of cytochrome P-450 dependent mono-oxygenases has been formulated as the ironoxy radical complex 44 of Schemes 1 and 2, and the current consensus of opinion favours the radical abstraction/recombination mechanism for hydroxylation outlined in Scheme 2 (15b). The data presented in Table 2 can now be interpreted on the basis of this mechanistic outline.

The substrates whose hydroxylation is summarized in Table 2 were prepared as described elsewhere (16). The β -labelled isomer (62% d_1 , 85% of label in the β position (16)) was hydroxylated to give a product that contained 39% d_1 species, the location of label at C- 2α being apparent from ^1H nmr integration, and from ^2H nmr and ^{13}C nmr data. The α -labelled isomer (80% d_1 , 87% of label in the α position (16)) was hydroxylated to give 2β -hydroxytestosterone with 75% d_1 in the 2α position. It is therefore apparent that, within error, label originally present at C- 2α is maintained during the hydroxyla-



SCHEME 1. Possible route for 2β hydroxylation of testosterone.

TABLE 2. Transformations of deuterium labelled testosterone by *G. fructicola*

Substrate (% composition)*	2β-hydroxytestosterone (% composition)*	% ² H retention
1 (38) + 24 (53) + 25 (9)	23 (61), 26 (39)	63
1 (20) + 24 (10) + 25 (70)	23 (25), 26 (75)	94

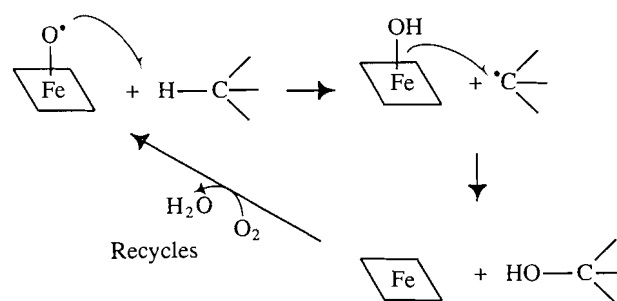
*±2.5%

TABLE 3. Transformations of testosterone by *G. fructicola*

Conditions of incubation	Product (%)
Normal	23 (40–60)
Under ¹⁸ O ₂ atmosphere	27 (58)
In 5 × 10 ⁻⁴ M KCN	23 (50)
Under 15% CO atmosphere	0
In 3 × 10 ⁻⁴ M cycloheximide	0
In 3 × 10 ⁻⁴ M cycloheximide, cells preinduced with testosterone	23 (10)

tion (94% retention). While this result alone would indicate hydroxylation with retention of configuration (the mode of cytochrome P-450 dependent hydroxylation most commonly observed (15b)), the data obtained from the β-labelled substrate, which retains 63% of the original label, show that this cannot be the case.

The analysis of the data of Table 2 depends on the relationship between stereoelectronic effects (SEE) (favouring axial (β) loss or addition of groups from C-2 (17)), and the intramolecular kinetic isotope effect (KIE) favouring loss of H over D, by a factor that may be as large as 11 in cytochrome P-450 hydroxylations (18). As outlined in Scheme 1, the loss of H from C-2 to the enzyme would be exclusively from the β face if the KIE and SEE are cooperative (for 2α-deuterium labelled substrate, **25**). However, these effects become counteractive for the C-2β labelled substrate, **24**, in which case the ratio of loss from the α and β sides is the result of an SEE/KIE competition. In both the cases reported in Table 2, recovered substrate shows no loss or epimerization of label, confirming that the phenomena reported are indeed a consequence of the hydroxylation process. The final stereochemistry of hydroxylation (conversion



SCHEME 2. Mechanism of C hydroxylation by cytochrome P-450 dependent mono-oxygenase enzymes.

of **43** to **23**) would be expected, proceeding under SEE control, to produce exclusive β product (17).

The magnitude of the SEE for hydrogen loss from C-2 during enolization of Δ⁴-3-ketosteroids has been estimated at 4:1 (β:α) (17). The data of Table 2 suggest that, of the 39% *d*₁ species obtained from the β-labelled isomer, 30% was formed by epimerization of label from the β to the α position during the reaction (the remaining 9% being obtained from the 9% of **25** originally present). In this event, the ratio of β:α loss from C-2 of **24** can be estimated as (53–30):30 or 1:1.3, requiring an intramolecular KIE of ca. 1.3 × 4 or 5.2, well within the observed range of such effects (18, 19). The above analysis is approximate, ignoring as it does the role of intermolecular KIE's, but as these are usually close to unity in cytochrome P-450 dependent hydroxylation (19, 20), their contribution is generally small.

The phenomenon of non-stereospecific hydrogen loss followed by stereospecific hydroxylation is one that we have previously reported for enzymic hydroxylation (19, 20), and the present analysis is analogous to that given by us for the steroid 6β-hydroxylase enzyme of *Rhizopus arrhizus* (3, 12), where the stereochemistry of hydrogen loss is also isotope-dependent. This phenomenon may be a general one for hydroxylations at sites subject to stereoelectronic effects due to their proximity to carbonyl groups or other centres of π-electron density.

The data reported herein for the 2β-hydroxylase enzyme of *G. fructicola* therefore suggest that this enzyme is a cytochrome P-450 dependent mono-oxygenase, operating via a stepwise oxidation mechanism, the stereochemical details of which are subject to substrate stereoelectronic and kinetic isotope effects.

Experimental

Apparatus, material, and methods

Melting points were determined on a Kofler heating stage and are uncorrected. Infrared spectra were recorded with an Analect 6260FX spectrometer. Proton nmr spectra were recorded at 80 MHz with a Bruker WP80CW or at 200 MHz with an AC200 spectrometer, using CDCl₃ as solvent (except where stated) and TMS as internal standard. Carbon spectra were recorded at 50.3 MHz (JMOD mode) using the latter instrument with chloroform as solvent and internal standard. Deuterium spectra were recorded at 30.72 MHz (AC200) or at 76.8 MHz (Bruker AM500, McMaster University), using chloroform as solvent and natural abundance CDCl₃ as internal standard. Mass spectra were obtained with an AEI MS30/Kratos DS55 system. Isotopic abundances were determined from the molecular ion or M - H₂O spectral regions following standard correction procedures for natural ¹³C abundances, and are accurate to ±2.5%. Thin-layer chromatography was performed on Merck silica gel 60F-254, and flash column chromatography on silica gel (230–400 mesh, Merck 9385).

Maintenance of microorganisms

Gnomonia fructicola ATCC 11430 was grown at 24–26°C on slopes of potato dextrose agar containing 0.5% strawberry petiole extract (21), then stored at 4°C and transferred at intervals not exceeding 8 weeks.

Preparation of substrates

With the exception of those listed below, the substrates used were commercial samples. Hitherto unreported spectral data are tabulated below for substrates not commercially available. 3-Deoxytestosterone (**15**): preparation, ref. 22; ¹³C nmr, δ: 11.0 (C-18), 19.3 (C-2), 19.5 (C-19), 21.0 (C-11), 23.4 (C-15), 25.8 (C-6), 30.5 (C-16), 32.0, 32.5 (C-3, C-7) 36.0 (C-8), 36.8 (C-12), 37.1 (C-10), 37.9 (C-1), 42.7 (C-13), 50.9 (C-14), 54.6 (C-9), 81.7 (C-17), 117.2 (C-4), and 144.0 (C-5) ppm. 3-Methyleneandrost-4-en-17β-ol (**16**): preparation, ref. 23; ¹³C nmr, δ: 11.1 (C-18), 18.6 (C-19), 21.1 (C-11), 23.5 (C-15), 27.3 (C-2), 30.6 (C-16), 32.4 (2C, C-6, C-7), 36.1 (C-8), 36.8 (C-12), 37.4 (C-1), 37.7 (C-10), 43.0 (C-13), 50.9 (C-14), 54.4 (C-9), 82.0 (C-17), 108.0 (=CH₂), 122.8 (C-4), 144.0 (C-3), and 149.0 (C-5) ppm. Androst-4-ene-3β,17β-diol (**17**): preparation, ref. 24; ¹³C nmr, δ: 11.0 (C-18), 19.0 (C-19), 20.6 (C-11), 23.4 (C-15), 29.5 (C-2), 30.5 (C-16), 32.1, 32.6 (C-6, C-7), 35.4 (C-1), 36.0 (C-8), 36.6 (C-12), 37.4 (C-10), 42.9 (C-13), 50.8 (C-14), 54.6 (C-9), 67.9 (C-3), 81.8 (C-17), 123.5 (C-4), and 147.4 (C-5) ppm. *B*-Norandrost-4-en-17β-ol-3-one (**18**): preparation, ref. 25. *A*-Norandrost-4-en-17β-ol-3-one (**20**): preparation, ref. 26. Androst-4-en-17β-ol-3-one-2β-d₁ (**24**): preparation, ref. 16. Androst-4-en-17β-ol-3-one-2α-d₁ (**25**): preparation, ref. 16.

Fungal biotransformations

General procedure

G. fructicola was grown in a medium of potato dextrose broth (200 mL) in a 1-L Erlenmeyer flask, maintained on a rotary shaker (180 rpm) at 24–26°C. After a growth period of 5 days, the fungus was collected by centrifugation and resuspended in distilled water (200 mL) in a 1-L Erlenmeyer flask. In a typical procedure, 15 such flasks were used to transform 1 g of substrate, added at the rate of 70 mg in 2 mL 95% ethanol per flask. Incubations with substrate were then performed under conditions identical with those described above (24–26°C, 180 rpm), for a 5-day period, following which the fungus was removed by filtration and products isolated by continuous extraction of the aqueous medium (methylene chloride), and purified by column chromatography. Extraction of the fungal biomass led to the recovery of starting material in most cases. This latter process was not routinely performed and was not quantified.

Controlled atmosphere and inhibitor experiments

The experiments summarized in Table 3 were performed in closed 1-L vessels (¹⁸O₂ and 15% CO atmospheres), or by replacement of distilled water in the above procedure by the appropriate inhibitor

solution. Preinduction of enzymes by testosterone was carried out by the addition of testosterone (4 mg in 1 mL 95% ethanol per flask) to the growing fungus 4 days after inoculation, followed by a further 24-h growth period. Subsequent procedures were carried out as described above. In all cases, yields reported in Table 3 refer to isolated, purified products.

Substrates

Isolated yields of products are listed in Tables 1 and 2. The following paragraphs contain (in order) substrate; recovered substrate from aqueous medium (%); products (%), in order of elution from column chromatography, together with hitherto unpublished spectral and (or) analytical data.

Androst-4-en-17β-ol-3-one (testosterone, **1**), 17; androst-4-ene-3,17-dione (**3**), 1, identified by comparison with authentic material; androst-4-ene-2β,17β-diol-3-one (**23**), 40–60; ¹³C nmr, δ: 11.3 (C-18), 22.5 (C-11), 22.8 (C-19), 23.3 (C-15), 30.4 (C-16), 33.0 (C-7), 34.5 (C-6), 35.8 (C-8), 36.5 (C-12), 39.5 (C-1), 41.4 (C-10), 43.4 (C-13), 50.3, 50.5 (C-9, C-14), 68.6 (C-2), 81.4 (C-17), 118.8 (C-4), 175.2 (C-5), 199.9 (C-3) ppm, identified by comparison with authentic material (27).

Androst-4-en-17β-ol-3-one, 2β-d₁ (**24**), 9; ²H nmr, δ: 2.40 (C-2β²H) ppm; androst-4-ene-2β,17β-diol-3-one, 2α-d₁ (**23** + **26**), total 25, ¹H nmr included signals at δ: 3.63 (1H, t, C-17αH), 4.15 (0.61 H, dd, C-2αH), and 5.78 (1H, s, C-4H) ppm; ²H nmr δ: 4.16 (C-2α²H) ppm.

Androst-4-en-17β-ol-3-one, 2α-d₁ (**25**), 9; ²H nmr, δ: 2.32 (C-2α²H) ppm; androst-4-ene-2β,17β-diol-3-one, 2α-d₁, (**23** + **26**), total 20; ¹H nmr included signals at δ: 3.63 (1H, t, C-17αH), 4.15 (0.25 H, dd, C-2αH), and 5.79 (1H, s, C-4H) ppm; ²H nmr δ: 4.16 (C-2α²H) ppm.

Androst-4-en-17β-ol-3-one, 17 acetate (testosterone acetate, **2**), 0; androst-4-ene-3,17-dione, **1**; androst-4-ene-2β,17β-diol-3-one (**23**), 30; androst-4-en-6β-ol-3,17-dione (**31**), 2, identified by comparison with authentic material (27).

Androst-4-ene-3,17-dione (**3**), 5; androst-4-en-2β-ol-3,17-dione (**28**), 4, identified by comparison with an authentic sample (27); androst-4-ene-17β-ol-3,16-dione (**30**), 5; ¹H nmr included signals at δ: 0.78 (3H, s, C-18H), 1.2 (3H, s, C-19H), 3.75 (1H, s, C-17αH) and 5.7 (1H, s, C-4H) ppm, identified by spectral, tlc, and mp comparisons with data from authentic material (28).

Pregn-4-ene-3,20-dione (progesterone, **4**), 10; pregn-4-en-2β-ol-3,20-dione (**32**), 5, identified by comparison of spectral and physical data with those reported (11); pregn-4-en-14α-ol-3,20-dione (**33**), 8, identified by comparison with reported spectral data (29, 30); pregn-4-ene-2β,6β-diol-3,20-dione (**34**), 12; ¹H nmr including δ: 0.68 (3H, s, C-18H), 1.48 (3H, s, C-19H), 2.1 (3H, s, C-21H), 4.2–4.4 (2H, m, C-2 and -6 H's), and 5.82 ppm, obtained in admixture with pregn-4-ene-6β,14α-diol-2,20-dione (**35**), 21. The latter compound was purified by repeated chromatography and had physical and spectral data identical with those reported (29, 30).

Androsta-4,6-dien-17β-ol-3-one (**5**), 25; androsta-4,6-diene-3,17-dione (**6**), 16, identical with an authentic commercial sample.

Androsta-1,4-dien-17β-ol-3-one (**7**), 15; androsta-1,4-diene-3,17-dione (**8**), 7, identical with an authentic samples prepared by Jones' oxidation of **7**.

19-Norandrost-4-en-17β-ol-3-one (19-nortestosterone, **9**), 12; 19-norandrost-4-ene-16α,17β-diol-3-one (**10**), 6, mp 172–174°C (lit. (31) mp 185–186°C); ¹H nmr included peaks at δ: 0.78 (3H, s, C-18H), 3.49 (1H, t, C-16βH), 4.10 (1H, br s, C-17αH), and 5.81 (1H, s, C-4H) ppm; ¹³C nmr δ: 12.2 (C-18), 25.7 (C-11), 26.5 (C-1), 30.6 (C-7), 33.7 (C-15), 35.4 (C-6), 36.4 (2C, C-2 and C-12), 40.0 (C-8), 42.6 (C-10), 43.6 (C-13), 47.6 (C-14), 49.5 (C-9), 78.4 (C-16), 89.7 (C-17), 124.7 (C-4), 166.5 (C-5), and 200.0 (C-3) ppm; ms *m/z* (%): 290 (M⁺, 90), 272 (100), 257 (25), 254 (35).

5α-Androstan-17β-ol-3-one (**11**), 12; 5α-androstane-3β,17β-diol (**36**), 3, analytical and spectral data identical with those reported (13); 5α-androstane-3β,17β,16α-triol (**37**), 20, mp 238–240°C (lit. (32) mp 256–260°C corr.); ¹H nmr included signals at δ: 0.66 (3H, s, C-18H), 0.73 (3H, s, C-19H), 3.26 (d, *J* = 5.6 Hz, C-17αH), 3.30–3.45 (1H,

m, C-3 α H), and 3.88 (1H, d of t, $J = 2.1, 5.6$ Hz, C-16 β H) ppm; ^{13}C nmr δ : 11.5, 11.7 (C-18, C-19), 19.6 (C-11), 27.8 (C-6), 30.8, 30.9 (C-2, C-7), 33.1 (C-15), 34.3 (C-8), 34.8 (C-10), 36.2 (2C, C-1 and C-12), 37.5 (C-4), 42.7 (C-13), 44.1 (C-5), 48.0 (C-14), 53.8 (C-9), 69.6 (C-3), 77.3 (C-16), and 88.9 (C-17) ppm; converted to the triacetate, mp 172–175°C (lit. (32) mp 174–176°C); ^1H nmr included signals at δ : 0.75, 0.77 (each 3H, s, C-18H and C-19H), 2.0, 2.01, 2.04 (each 3H, s, COCH₃), 4.5–4.75 (1H, m, C-3 α H), 4.89 (1H, d, $J = 5.6$ Hz, C-16 β H) ppm; ^{13}C nmr δ : 12.2 (C-19), 13.0 (C-18), 20.4 (C-11), 21.0, 21.1, 21.4 (acetyl CH₃) 27.5 (C-2), 28.4 (C-6), 31.4 (C-7), 32.3 (C-15), 34.0 (C-4), 34.9 (C-8), 35.6 (C-10), 36.7, 36.8 (C-1, C-12), 43.6 (C-13), 44.7 (C-5), 49.1 (C-14), 54.1 (C-9), 73.6 (C-3), 78.1 (C-16), 86.3 (C-17), 170.6, 170.6, 170.7 (acetyl CO) ppm; 5 α -androstane-2 β ,3 α ,16 α ,17 β -tetrol (38), 3, mp 247–250°C; ^1H nmr included signals at δ : 0.63 (3H, s, C-18H), 0.88 (3H, s, C-19H), 3.28 (1H, d, $J = 5.4$ Hz, C-17 α H), 3.35–3.7 (2H, complex m, C-2 α and C-3 β H's), and 3.90 (1H, d of t, $J = 2, 5.4$ Hz, C-16 β H) ppm; ^{13}C nmr (DMSO-*d*₆) δ : 11.9 (C-18), 13.7 (C-19), 19.4 (C-11), 27.6 (C-6), 31.1 (C-4), 30.9 (C-7), 33.1 (C-15), 33.9 (C-8), 35.2 (C-10), 36.4 (C-12), 38.3 (C-1), 43.0 (C-13), 48.2 (C-14), 54.8 (C-9), 69.5, 70.6 (C-2, C-3), 77.2 (C-16), and 89.2 (C-17) ppm; the signal from C-5 was obscured by solvent peaks; ms, m/z (%): 324 (M⁺), (25), 306 (86), 291 (40), 288 (45), 273 (40), 262 (43), 249 (75), 231 (100). 38 converted to the tetraacetate, mp 82–84°C from methanol; ^1H nmr included signals at δ : 0.76 (3H, s, C-18H), 0.87 (3H, s, C-19), 2.00, 2.03, 2.035, 2.04 (total 12H, each s, COCH₃), 4.87 (2H, br s, C-17 α and C-2 α or C-3 β H), 4.75–5.0 (1H, m, C-2 α or C-3 β H), and 5.10 (1H, d of t, $J = 2, 5.5$ Hz, C-16 β H) ppm; ^{13}C nmr δ : 13.0 (C-18), 13.7 (C-19), 20.0 (C-11), 21.1, 21.2, 21.3 (2C, acetyl CH₃), 27.7 (C-6), 29.3 (C-4), 31.2 (C-7), 32.3 (C-15), 34.4 (C-8), 35.1 (C-10), 36.7 (C-12), 37.7 (C-1), 39.6 (C-5), 43.6 (C-13), 49.0 (C-14), 54.7 (C-9), 69.8, 70.2 (C-2, C-3), 78.1 (C-16), 86.2 (C-17), 169.6, 169.8, 170.7, 170.9 (acetyl CO) ppm.

5 β -Androstan-17 β -ol-3-one (12), 7; 5 β -androstane-5,17 β -diol-3-one (40), 19, mp 154–165°C (dec.); ^1H nmr included signals at δ : 0.75 (3H, s, C-18H), 1.0 (3H, s, C-19H), 3.0 (1H, half of ABq, $J = 15$ Hz, C-4H), and 3.65 (1H, t, $J = 8$ Hz, C-17 α H) ppm; ^{13}C nmr δ : 11.1 (C-18), 16.2 (C-19), 21.4 (C-11), 23.4 (C-15), 28.7 (C-17), 30.6 (C-16), 31.2 (C-1), 35.0 (C-8), 36.6 (C-12), 37.0, 37.3 (C-2, C-6), 40.4 (C-10), 43.1 (C-13), 44.1 (C-9), 49.4 (C-4), 51.3 (C-14), 78.6 (C-5), 81.7 (C-17), and 210.8 (C-3) ppm; ms, m/z (%): 288 (M – H₂O, 10), 274 (10), 246 (10), 236 (18) relative to 71 (100); converted to the 17-monoacetate, mp 174–176°C (lit. (33) mp 176–178°C); ^1H nmr included signals at δ : 0.78 (3H, s, C-18H), 0.98 (3H, s, C-19H), 2.02 (3H, s, COCH₃), 3.0 (1H, half of ABq, $J = 15$ Hz, C-4H), and 4.57 (1H, t, $J = 8$ Hz, C-17 α H) ppm; ^{13}C nmr δ : 12.1 (C-18), 16.2 (C-19), 21.1 (acetyl CH₃), 21.3 (C-11), 23.6 (C-15), 27.6 (C-16), 28.7 (C-7), 31.2 (C-1), 34.7 (C-8), 36.3, 36.9 (C-6, C-12), 37.3 (C-2), 40.8 (C-10), 42.7 (C-13), 44.0 (C-9), 49.5 (C-4), 51.0 (C-14), 78.5 (C-5), 82.6 (C-17), 170.8 (acetyl CO), and 210.6 (C-3) ppm; 5 β -androstane-3 α ,5,17 β -triol (41), 17; mp 218–221°C (lit. (34) mp 220–223°C); ^1H nmr included signals at δ : 0.73 (3H, s, C-18H), 0.90 (3H, s, C-19H), 3.4–3.7 (1H, m, C-3 β H), and 3.64 (1H, t, $J = 8$ Hz, C-17 α H) ppm; ^{13}C nmr δ : 11.1 (C-18), 16.4 (C-19), 20.8 (C-11), 23.4 (C-15), 28.2 (C-7), 29.8, 30.1 (2C, C-1, C-2, C-16), 35.1 (C-8), 36.7, 36.9 (C-6, C-12), 39.8 (C-10), 42.0 (C-9), 43.0 (C-13), 43.5 (C-4), 51.2 (C-14), 68.1 (C-3), 75.7 (C-5), and 81.8 (C-17) ppm; ms, m/z (%): 308 (2), 290 (26), 275 (10), 273 (10), 272 (30), 257 (15), 236 (100).

Androst-4-en-19-ol-3,17-dione (13) was recovered quantitatively and unchanged from incubation with *G. fructicola*.

Androst-4-en-19-al-3,17-dione (14) was converted to the alcohol 13 in 85% isolated yield by *G. fructicola*.

Androst-4-en-17 β -ol (15) and 3-methyleneandrost-4-en-17 β -ol (16) were recovered unchanged in yields of 28 and 36% respectively from the aqueous medium following incubation with *G. fructicola*. Extraction of the fungal biomass yielded the balance of the substrates.

Androst-4-ene-3 β ,17 β -diol (17), 0; androst-4-en-17 β -ol-3-one (1), 2; androst-4-ene-3,17-dione (3), 1; androst-4-ene-2 β ,17 β -diol-3-one

(23), 17; and androst-4-en-2 β -ol-3,17-dione (28), 2; all of which were identified by comparison with authentic samples (see above).

B-Norandrost-4-en-17 β -ol-3-one (18), 24; B-norandrost-4-ene-3,17-dione (19), 5; spectral data identical with those reported (25).

B-Norandrost-4-en-17 β -ol-3-one (18), with induced enzymes, 45; B-norandrost-4-ene-3,17-dione (19), 5, identified as reported above; B-norandrost-1,4-dien-17 β -ol-3-one (42), 8, mp 156–159°C (lit. (35) mp 157–159°C); ^1H nmr included signals at δ : 0.80 (3H, s, C-18H), 1.10 (3H, s, C-19H), 3.68 (1H, t, $J = 8$ Hz, C-17 α H), 6.0 (1H, s, C-4H), 6.15 and 7.02 (2H, ABq, C-2 and C-1 H's) ppm; ^{13}C nmr δ : 11.5 (C-18), 20.5 (C-19), 20.8 (C-11), 23.9 (C-15), 30.6 (C-16), 32.7 (C-6), 36.4 (C-12), 38.5 (C-8), 44.9 (C-13), 48.2 (C-10), 50.6 (C-14), 53.4 (C-9), 81.3 (C-17), 123.7 (C-4), 128.4 (C-2), 152.6 (C-1), 174.1 (C-5), and 186.7 (C-3) ppm.

A-Norandrost-3(5)-en-17 β -ol-2-one (A-nortestosterone, (20), 60; A-norandrost-3(5)-ene-2,17-dione (21), 3, mp 163–166°C (lit. (36) mp 166–166.5°C), identified by comparison with an authentic sample produced by Jones' oxidation of A-nortestosterone; A-norandrost-3(5)-ene-1 β ,17 β -diol-2-one (22), 10, mp 236–238°C (lit. (37) mp 240–243°C); ^1H nmr included peaks at δ : 0.80 (3H, s, C-18H), 1.10 (3H, s, C-19H), 3.62 (1H, t, $J = 8$ Hz, C-17 α H), 3.75 (1H, s, C-1H), and 5.80 (1H, s, C-4H) ppm; ^{13}C nmr δ : 11.1 (C-18), 15.5 (C-19), 23.0 (C-11), 23.3 (C-15), 28.1 (C-6), 30.0 (C-16), 31.5 (C-7), 35.7 (C-8), 36.4 (C-12), 43.4 (C-13), 50.0 (C-10), 50.4 (C-14), 53.7 (C-9), 81.6 (C-17), 83.0 (C-1), 122.1 (C-3), 186.3 (C-5), and 207.4 (C-2) ppm; ms, m/z (%): 290 (78), 275 (52), 259 (20), relative to 57 (100).

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