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Photodissociation of $\text{Kr}_2\text{F}(4^2\Gamma)$ in the ultraviolet and near-infrared: Wavelength dependence of $\text{KrF}(B^2\Sigma)$ yield

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The photoabsorption spectrum of the $4^2\Gamma$ excited state of Kr_2F has been measured in the 280–850 nm region by fluorescence suppression spectroscopy. Both the $9^2\Gamma \leftarrow 4^2\Gamma$ and $6^2\Gamma \leftarrow 4^2\Gamma$ bands, peaking at 320 and ~ 710 nm, respectively, have been observed—the latter for the first time. Although the position of the ultraviolet band is consistent with both theory and previous experiments, its spectral width is $\sim 40\%$ of that measured by Greene and McCown [Appl. Phys. Lett. **54**, 1965 (1989)] in absorption experiments in a pulsed discharge. The relative yield of $\text{KrF}(B)$ produced in the photodissociation of $\text{Kr}_2\text{F}(4^2\Gamma)$ was found to be independent of wavelength over the 280–360 nm and 590–840 nm spectral intervals and the mechanism responsible for photodissociation appears to be predissociation of the $\text{Kr}_2\text{F } 9^2\Gamma$ and $6^2\Gamma$ states by potentials correlated with the $\text{KrF}(D) + \text{Kr}$ and $\text{KrF}(C) + \text{Kr}$ limits, respectively. © 1997 American Institute of Physics. [S0021-9606(97)00313-9]

I. INTRODUCTION

The lowest electronic excited state of the rare gas-halide triatomic molecules, denoted $4^2\Gamma$, is ionic in character (i.e., Rg_2^+X^- , where Rg and X represent rare gas and halogen atoms, respectively) and *ab initio* calculations^{1–3} suggest that the effect of the halogen anion on the structural and optical properties of the rare gas dimer cation is negligible. Consequently, the absorption spectra for the $9^2\Gamma \leftarrow 4^2\Gamma$ and $6^2\Gamma \leftarrow 4^2\Gamma$ transitions of Kr_2F , for example, are expected to be similar to those for the corresponding transitions of Kr_2^+ : $D^2\Sigma_g^+ \leftarrow A^2\Sigma_u^+$ and $B^2\Pi_g \leftarrow A^2\Sigma_u^+$, respectively.

Measurements of the photoabsorption cross section for the $4^2\Gamma$ state of Xe_2Cl at several wavelengths in the ultraviolet (UV) were described by McCown *et al.*⁴ in 1985 and, with the fluorescence suppression approach, McCown⁵ and Hakuta and co-workers⁶ subsequently determined the $\text{Kr}_2\text{F}(4^2\Gamma)$ absorption cross section at 248 nm. Similar measurements were made by Geohegan and Eden^{7,8} for Kr_2F at 35 wavelengths in the 248–572 nm region with a dye laser and were supplemented by continuous scans from 335 to ~ 600 nm. These results showed that the peak of $\text{Kr}_2\text{F}(4^2\Gamma)$ absorption in the UV lay between 310 and 335 nm. By producing Kr_2F in a transverse discharge, Greene and McCown⁹ were able to directly observe $\text{Kr}_2\text{F}(4^2\Gamma)$ in absorption between 225 and 455 nm and confirmed that the $9^2\Gamma \leftarrow 4^2\Gamma$ band peaks at 315 nm.

In 1988, McCown and co-workers¹⁰ were the first to demonstrate the direct production of the rare gas-fluoride diatomic species when a UV photon is absorbed by the triatomic in the $4^2\Gamma$ state. Specifically, these experiments revealed that the photodissociation of $\text{Kr}_2\text{F}(4^2\Gamma)$ in the

near-UV (308–351 nm) yields $\text{KrF}(B^2\Sigma)$ with a quantum efficiency approaching unity.¹¹ An analogous observation for $\text{Ar}_2\text{F}(4^2\Gamma)$ was later reported in Ref. 12 but early experiments on Xe_2Cl had searched unsuccessfully for this process.⁴

Experiments in which the $\text{Kr}_2\text{F}(4^2\Gamma)$ absorption spectrum was recorded from 280 to 850 nm are described here. The bands associated with both the $9^2\Gamma \leftarrow 4^2\Gamma$ and $6^2\Gamma \leftarrow 4^2\Gamma$ transitions and lying in the UV (280–390 nm) and near-infrared (550–850 nm), respectively, are observed in their entirety. In particular, the $6^2\Gamma \leftarrow 4^2\Gamma$ absorption band of the molecule which peaks at ~ 710 nm is observed for the first time and the position and breadth of the band profile are consistent with theoretical predictions. In addition to absorption spectra obtained by fluorescence suppression spectroscopy, the production of KrF excited states by photodissociating $\text{Kr}_2\text{F}(4^2\Gamma)$ was studied by recording the peak $\text{KrF}(B \rightarrow X)$ emission intensity as the excitation wavelength was scanned. A comparison of the absorption (fluorescence suppression) and excitation spectra reveal that the relative yield of $\text{KrF}(B)$ in the photodissociation of $\text{Kr}_2\text{F}(4^2\Gamma)$ is independent of wavelength from 280 nm to beyond 800 nm.

II. EXPERIMENT

The experimental apparatus and data acquisition techniques have been discussed in detail previously⁸ and will only be reviewed here. A pump–probe approach to selectively examining the optical properties of Kr_2F was adopted in which the initial laser pulse serves to produce the triatomic species. The second, time-delayed (probe) pulse depletes the $\text{Kr}_2\text{F}(4^2\Gamma)$ population by absorption and the observables are the magnitude and wavelength dependence of the $\text{Kr}_2\text{F}(4^2\Gamma \rightarrow 1^2\Gamma)$ fluorescence suppression and the concomitant appearance of $\text{KrF}(B)$ emission. Briefly, $\text{Kr}_2\text{F}(4^2\Gamma)$ molecules were produced by the beam from an ArF laser (193 nm) which was focused into a quartz cell

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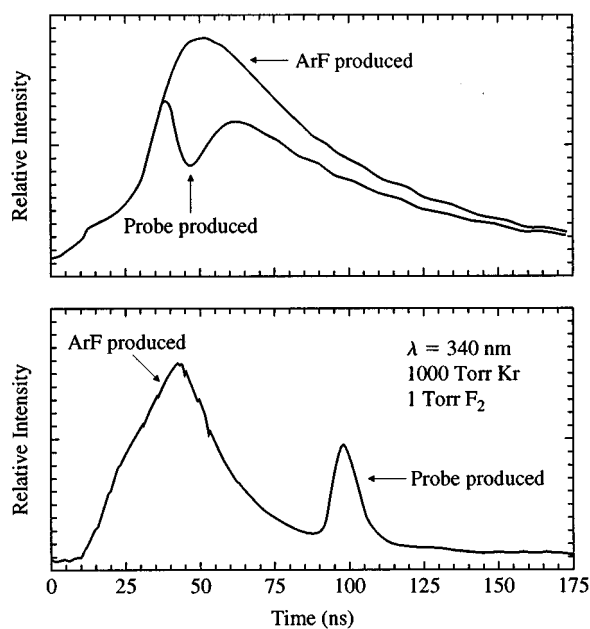


FIG. 1. Typical $\text{Kr}_2\text{F}(4^2\Gamma \rightarrow 1^2\Gamma)$ and $\text{KrF}(B \rightarrow X)$ emission wave forms: (top) Kr_2F fluorescence wave forms produced by ArF laser irradiation of a 1000 Torr Kr/1 Torr F_2 gas mixture and the suppression of the fluorescence by a dye laser pulse ($\lambda=340$ nm); (bottom) $\text{KrF}(B \rightarrow X)$ emission generated by the ArF laser pulse (early fluorescence peaking at ~ 40 ns) and the trailing dye laser (probe) pulse. Notice that, for illustrative purposes, Δt is quite different for the two sets of wave forms ($\Delta t \sim 35$ and 90 ns for top and bottom traces, respectively).

containing a room temperature mixture of Kr and F_2 . Production of Kr excited states and Kr^+ ions by nonresonant, two photon excitation and 2+1 REMPI, respectively, triggers a series of collisional processes that culminates in the formation of $\text{KrF}(B, C)$ and $\text{Kr}_2\text{F}(4^2\Gamma)$. An adjustable time delay (Δt) later, a pulse from an Nd:YAG-pumped dye laser (~ 10 ns, $\Delta\tilde{\nu} \approx 0.2$ cm^{-1} , ~ 1 mm^2 cross-sectional beam area), directed along the axis of the cell and coincident with the focus of the excimer, excites a fraction of the Kr_2F molecules. The absorption of dye laser radiation by $\text{Kr}_2\text{F}(4^2\Gamma)$ is monitored by observing the suppression of $\text{Kr}_2\text{F}(4^2\Gamma \rightarrow 1^2\Gamma)$ fluorescence in the violet with a photomultiplier and appropriate optical filters. As noted earlier, the ultimate fate of the Kr_2F species is also of interest; consequently, the relative peak intensity of the $\text{KrF}(B \rightarrow X)$ emission produced by the second (dye) laser pulse was also recorded as the laser wavelength was varied. Unless noted otherwise, the gas mixture can be assumed to be 1000 Torr Kr/1 Torr F_2 and Δt is typically set at 60 ns.

III. RESULTS AND DISCUSSION

A. Wave forms, typical data

Fluorescence wave forms representative of those observed throughout these experiments are displayed in Fig. 1. In the upper portion of the figure are shown two oscillograms illustrating $\text{Kr}_2\text{F}(4^2\Gamma \rightarrow 1^2\Gamma)$ fluorescence suppression arising from absorption by $\text{Kr}_2\text{F}(4^2\Gamma)$ at 340 nm. In the absence of a probe pulse, the Kr_2F spontaneous emission peaks ~ 50

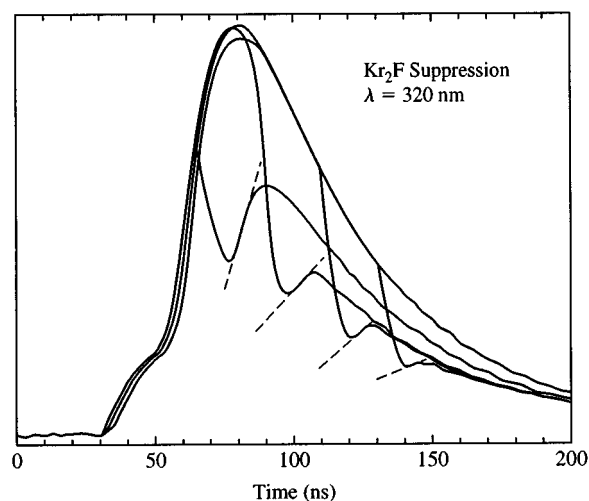


FIG. 2. Five superimposed Kr_2F fluorescence wave forms illustrating the dependence of the suppression profile on Δt for a fixed probe laser intensity and $\lambda=320$ nm. The dashed lines reflect the slowing of the recovery of the Kr_2F population that results from the rapidly declining $\text{KrF}(B, C)$ populations.

ns following the onset of the ArF laser pulse and exhibits a decay time which is consistent with the known spontaneous emission lifetime for the $4^2\Gamma$ state and the rate constants for quenching of $\text{Kr}_2\text{F}(4^2\Gamma)$ by Kr and F_2 (Refs. 8 and 13). When the time-delayed probe pulse ($\lambda=340$ nm) is introduced, strong suppression ($\sim 46\%$) of the Kr_2F fluorescence is observed but note that, following the termination of the probe pulse, the Kr_2F population partially recovers (to $>85\%$ of its pre-probe value). In previous experimental studies of Kr_2F absorption by fluorescence suppression, recovery of the Kr_2F population was slight⁵ or not detectable,⁶⁻⁸ an interesting effect that is closely tied to the temporal width of the probe pulse and three body rate constant for Kr_2F formation and which will be discussed in more detail later.

The trace in the lower portion of Fig. 1 shows the $\text{KrF}(B \rightarrow X)$ spontaneous emission that is produced by the initial (ArF pump) pulse and the probe radiation ($\lambda=340$ nm, $\Delta t \approx 90$ ns). For the remainder of this paper, we will focus on the latter. The temporal history of the KrF fluorescence produced by the probe (dye) laser pulse replicates the dye laser wave form itself; one concludes, therefore, that $\text{Kr}_2\text{F}(4^2\Gamma)$ is converted to $\text{KrF}(B)$ on a time scale of $\leq 1-2$ ns. A superposition of five Kr_2F fluorescence suppression wave forms, recorded for different values of Δt and a probe laser wavelength of 320 nm, is presented in Fig. 2. Not only do these traces illustrate the reproducibility of the experiment but they also demonstrate that the *rate and degree* of $\text{Kr}_2\text{F}(4^2\Gamma)$ population recovery following the probe pulse (which is reflected by the magnitude of the slope of the wave form during recovery—cf. Fig. 2) falls rapidly as Δt increases. This is precisely what one would expect since the re-formation of $\text{Kr}_2\text{F}(4^2\Gamma)$ depends upon the $\text{KrF}(B, C)$ populations, both of which have considerably shorter effective lifetimes than that for $\text{Kr}_2\text{F}(4^2\Gamma)$.

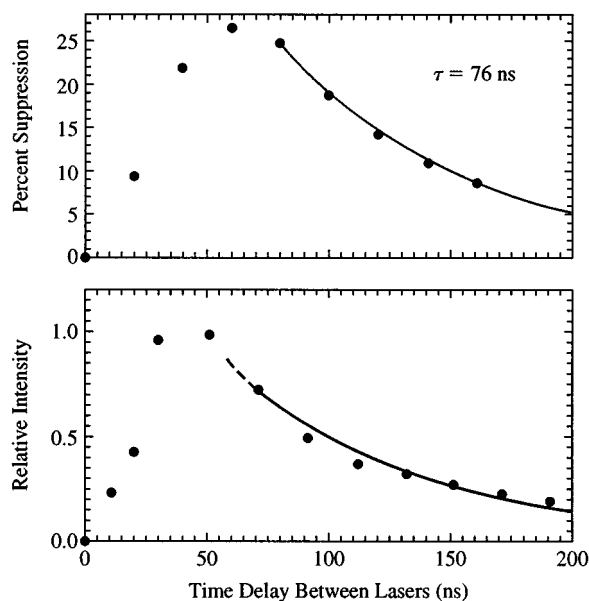


FIG. 3. Comparison of the temporal histories of: (a) fractional suppression of Kr₂F fluorescence in the violet, and (b) relative peak KrF(*B*→*X*) emission, as functions of the time delay, Δt . The decay of both sets of data is well described by a single exponential having a time constant of 76 ± 2 ns for (a) and 74 ± 2 ns for (b). All data were obtained for a probe laser intensity of ~ 20 MW·cm⁻².

Insight as to the mechanism responsible for the production of KrF(*B*) is provided by the data of Fig. 3 which compares the magnitude of the Kr₂F violet fluorescence suppression with the peak KrF(*B*→*X*) emission intensity as a function of Δt . The fluorescence suppression data were acquired with a probe laser intensity lower than that of Fig. 2 in order to avoid saturating the Kr₂F absorptive transition. Note that the decline in the Kr₂F fluorescence suppression data is well-described by a single exponential having a time constant of 76 ± 2 ns. Similarly, the KrF(*B*) emission decays exponentially with Δt and, as shown by the solid curve in Fig. 3(b), the time constant that best fits the data is $(74 \pm 2 \text{ ns})^{-1}$. The fact that the decay constants for both sets of data in Fig. 3 match (to within experimental uncertainty) each other **and** the known effective lifetime of Kr₂F(*4*² Γ) for this gas mixture provides strong evidence that, as suggested by McCown *et al.*,^{10,11} the KrF(*B*→*X*) fluorescence generated by the probe pulse is, indeed, the result of the photodissociation of Kr₂F(*4*² Γ). Also, results virtually identical to those of Fig. 3 were obtained when the probe laser wavelength was tuned to the expected position¹ of the Kr₂F(*6*² Γ ←*4*² Γ) transition in the near-infrared ($\lambda \sim 710$ nm)—namely, the suppression of Kr₂F(*4*² Γ →*1*² Γ) spontaneous emission corresponded to the production of KrF(*B*). Consequently, not only is the *9*² Γ state of Kr₂F unstable with respect to the KrF*+Kr limit (where the asterisk denotes an electronic excited state of the diatomic) but the *6*² Γ state is as well.

B. KrF(*B*) excitation and Kr₂F(*4*² Γ) absorption spectra

By monitoring the peak KrF(*B*→*X*) emission intensity while scanning the probe laser wavelength, the excitation

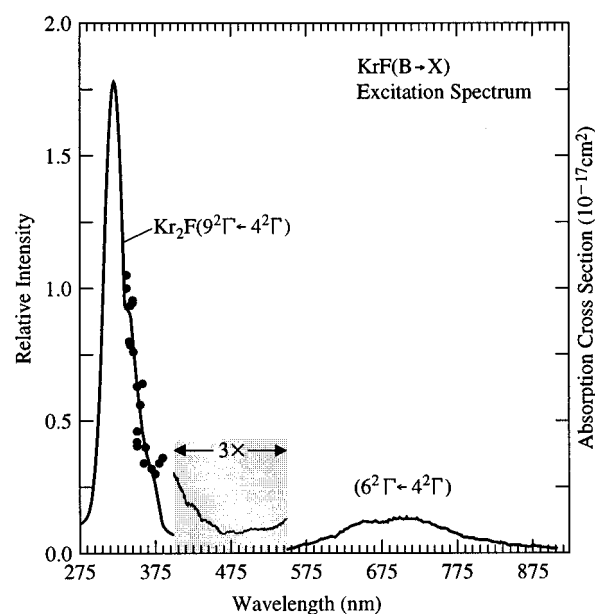


FIG. 4. Excitation spectrum for the production of KrF from Kr₂F(*4*² Γ) in the 280–900 nm region. In acquiring these data, the relative KrF(*B*→*X*) emission was monitored and the probe laser intensity was fixed at ~ 20 MW·cm⁻². For the sake of clarity, the 400–550 nm region has been magnified and the data for $\lambda < 390$ nm have been smoothed. A few of the Kr₂F(*4*² Γ) absorption cross sections in the 337–385 nm interval from Ref. 8 are also shown.

spectrum for the production of KrF* in the photodissociation of Kr₂F(*4*² Γ) over the 280–900 nm region was obtained and is shown in Fig. 4. Throughout the data acquisition process, the peak intensity of the probe pulse was maintained at 20 MW·cm⁻² (2 mJ, ~ 10 ns). Two bands are clearly present in this spectral region, the strongest of which lies in the UV and peaks at ~ 320 nm. As was discussed earlier, virtually all previous studies of Kr₂F absorption have focused on this band that arises from *9*² Γ ←*4*² Γ transitions of the molecule. Aside from the data of Ref. 8 that spanned the 335–600 nm region, the only other *4*² Γ absorption cross sections in the literature were measured at 248 nm (Refs. 5–8), 308 nm (Ref. 8), and 358 nm (Ref. 14). Although the location of the *9*² Γ ←*4*² Γ band peak in Fig. 4 is in agreement with the value reported by Greene and McCown⁹ (315 nm), our present measurements (as well as the results of Ref. 8) suggest that the bandwidth is only ~ 33 nm FWHM, or at least a factor two smaller than that given in Ref. 9 (85 nm). The difference may be attributable to the considerably different environments in which the two spectra were recorded. In Ref. 9, Kr₂F was produced in a commercial, transverse discharge device and absorption measurements were made in a 20 ns window during the active discharge or in the early afterglow. It is likely, therefore, that excited vibronic levels of the Kr₂F(*4*² Γ) state populated by electron impact have the effect of significantly broadening the *9*² Γ ←*4*² Γ spectrum.

A much weaker band, extending from below 650 nm to beyond 850 nm, is also observed in Fig. 4 and is assigned to the *6*² Γ ←*4*² Γ transition of Kr₂F. The existence of this band, which corresponds to the $1(1/2)_g \leftarrow 1(1/2)_u$ transition

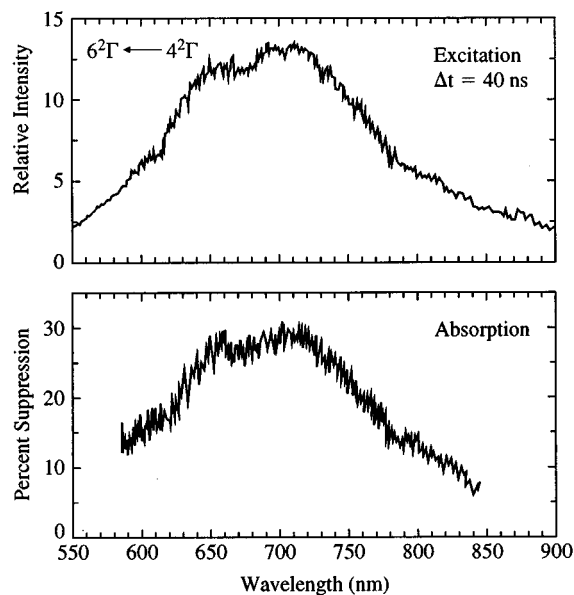


FIG. 5. KrF(*B*) excitation spectrum (top) and Kr₂F(*4 2*Γ) fluorescence suppression (absorption) spectrum in the red and near-IR showing maxima at 655 and 710 nm attributed to the *6 2*Γ←*4 2*Γ transition of Kr₂F.

(*B 2*Π_{*g*}←*A 2*Σ_{*u*}⁺) of the parent Kr₂⁺ ion,^{1,8} was predicted by Wadt and Hay¹ but, to our knowledge, has not been reported previously. Also, the peak intensity for this band relative to that for the UV band is consistent with the ratio of oscillator strengths calculated in Ref. 1 for the corresponding transitions.

Aside from these two obvious bands, other faint but reproducible structure appears in the 420–490 nm region and in the vicinity of 600 nm. For convenience, the 400–550 nm interval in Fig. 4 has been magnified and, although the signal is too weak to make definitive statements, it is quite possible that these features are associated with the *8 2*Γ←*4 2*Γ and *7 2*Γ←*4 2*Γ transitions of Kr₂F that were predicted by Wadt and Hay¹ to peak at 478 and 603 nm, respectively. The latter band appears to be partially blended with the blue tail of the stronger *6 2*Γ←*4 2*Γ band discussed earlier.

Figure 5 gives a comparison of the KrF(*B*) excitation spectrum in the 550–900 nm region with the Kr₂F(*4 2*Γ) absorption spectrum measured by fluorescence suppression over approximately the same wavelength interval. Although the *S/N* ratio for the fluorescence suppression data is roughly one half of that for the excitation spectrum, it is nevertheless clear that the two spectra exhibit the same structure. Local maxima at ~655 and 710 nm appear in both spectra and their overall profiles are virtually indistinguishable. Recalling that the excitation spectrum is a measure of the production of KrF* as a function of wavelength, then normalizing the excitation spectrum to the fluorescence suppression (absorption) spectrum provides a measure of the wavelength dependence of the relative yield of KrF* in the photodissociation of Kr₂F(*4 2*Γ). Consequently, it is reasonable to conclude that the KrF(*B*) yield is independent of wavelength in the 590–840 nm region. The same result was obtained for the *9 2*Γ←*4 2*Γ spectra—the profiles for both the excitation and

fluorescence suppression spectra in the 280–360 nm region are essentially identical, indicating that the yield of KrF(*B*) in this wavelength range is also constant.

The uniform efficiency with which Kr₂F is photodissociated over such a broad spectral range is interesting, particularly in view of the fact that spontaneous emission from Kr₂F excited states higher than *4 2*Γ has not been reported in the literature. Diatomics-in-molecules (DIM) calculations carried out by Huestis and Schlotter² for the lowest nine electronic excited states of Kr₂F show that all are strongly bound (~2 eV) with respect to the Kr₂⁺+F⁻ limits. However, only one—*4 2*Γ—is stable with respect to the KrF(*B,C,D*)+Kr limits. Specifically, unpublished calculations by Huestis predict that the potentials correlated with Kr⁺F⁻(*C,D*)+Kr are dissociative. Since: 1) the symmetry species¹ for the KrF(*C 2*Π_{3/2})+Kr and KrF(*D 2*Σ_{1/2}⁺)+Kr limits are ²B₂ and ²A₁, respectively, and 2) once spin-orbit effects are considered, the Kr₂F(*6 2*Γ) state has both ²A₂ and ²B₂ character, then it is likely that the potential associated with KrF(*C*)+Kr is responsible for predissociation of the Kr₂F(*6 2*Γ) state. Similar considerations suggest that Kr₂F(*9 2*Γ) (which is almost entirely ²A₁ in structure)¹ is predissociated by the repulsive potential arising from KrF(*D*)+Kr.

It is, therefore, understandable that photoexciting the *9 2*Γ state of Kr₂F would rapidly culminate in the production of KrF(*B*→*X*) emission. The *D* and *B* states of KrF are separated in energy by approximately the spin-orbit splitting of Kr⁺ (²P_{3/2}–²P_{1/2}≈0.67 eV). Consequently, the *D 2*Σ_{1/2}⁺ state is “embedded” within the *B* state potential and a number of low *D 2*Σ_{1/2}⁺ vibrational levels are nearly degenerate with *v*'<30 levels of the *B 2*Σ_{1/2}⁺ state. Furthermore, at the gas pressures typical of these experiments, the *B* and *C* states of KrF are closely coupled by collisions—hence, predissociation of the *6 2*Γ state of Kr₂F would also quickly produce the KrF(*B*) species.

C. Modeling of Kr₂F fluorescence recovery: Estimate of KrF yield

In the discussion of Sec. III A, the partial recovery of the Kr₂F(*4 2*Γ) fluorescence suppression wave forms following the termination of the probe pulse was noted. One would expect the rate and degree of recovery to depend critically on several kinetic constants; in fact, the experimental wave forms serve as a sensitive and convenient test of the rate constant for the formation of Kr₂F(*4 2*Γ) from KrF* by three body collisions and, moreover, allow for a lower limit on the yield of KrF* in the photodissociation of Kr₂F to be determined.

In order to assess the impact of various experimental parameters and rate constants on the fluorescence suppression wave forms, a simple kinetics model of the KrF–Kr₂F system was constructed in which, for convenience, the photodissociation of Kr₂F (*4 2*Γ) was assumed to produce the KrF(*C*) species. Photoexcitation of Kr₂F can be described by the following set of three coupled rate equations:

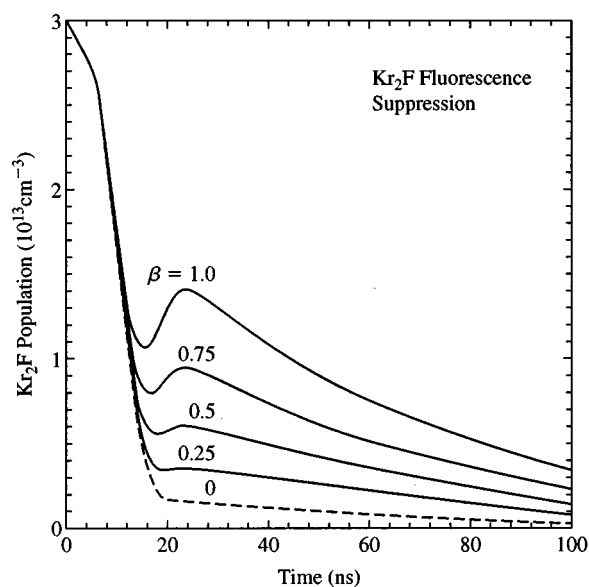


FIG. 6. Simulated Kr₂F(4 ²Γ) fluorescence suppression wave forms illustrating the dependence of the recovery on the KrF(C) quantum yield, β . The actual dye (probe) laser wave form was incorporated into the computations and the peak probe laser intensity was assumed to be 20 MW-cm⁻². The initial Kr₂F(4 ²Γ) and KrF(B,C) number densities were set at 3×10^{13} cm⁻³ and zero, respectively, and the probe laser wavelength was 337 nm (9 ²Γ ← 4 ²Γ).

$$\frac{\partial[\text{KrF}(B)]}{\partial t} = -k_3[\text{Kr}]^2[\text{KrF}(B)] - \frac{[\text{KrF}(B)]}{\tau_b} + k_m\{[\text{KrF}(C)] - [\text{KrF}(B)]\}, \quad (1)$$

$$\frac{\partial[\text{KrF}(C)]}{\partial t} = k_m\{[\text{KrF}(B)] - [\text{KrF}(C)]\} + \frac{\sigma I(t)}{\hbar \omega} \beta[\text{Kr}_2\text{F}] - \frac{[\text{KrF}(C)]}{\tau_c}, \quad (2)$$

$$\frac{\partial[\text{Kr}_2\text{F}]}{\partial t} = -k_q[\text{F}_2][\text{Kr}_2\text{F}] - \frac{[\text{Kr}_2\text{F}]}{\tau_3} + k_3[\text{Kr}]^2[\text{KrF}(B)] - \frac{\sigma I(t)\beta}{\hbar \omega} [\text{Kr}_2\text{F}], \quad (3)$$

where brackets denote number densities, $[\text{Kr}_2\text{F}]$ represents the population of the 4 ²Γ state, k_3 is the three body formation rate constant for Kr₂F (6.7×10^{-31} cm⁶ s⁻¹), τ_b is the B state spontaneous emission lifetime (6.8 ns), k_m is the rate constant for the mixing of the B and C states by two body collisions with Kr (5×10^{-10} cm³ s⁻¹), τ_c is the C state radiative lifetime (75 ns), k_q is the rate constant for the quenching of Kr₂F(4 ²Γ) by F₂ (4.3×10^{-10} cm³ s⁻¹), τ_3 is the Kr₂F(4 ²Γ) spontaneous emission lifetime (178 ns), and β is the quantum yield of KrF(C) in the photodissociation of Kr₂F(4 ²Γ) (Refs. 11, 13, 15, and 16).

A series of five simulated Kr₂F fluorescence suppression wave forms are shown in Fig. 6 for β varied from 0 to 1. In these computations, it was assumed that the probe laser wavelength and peak intensity are 337 nm and 20 MW cm⁻²,

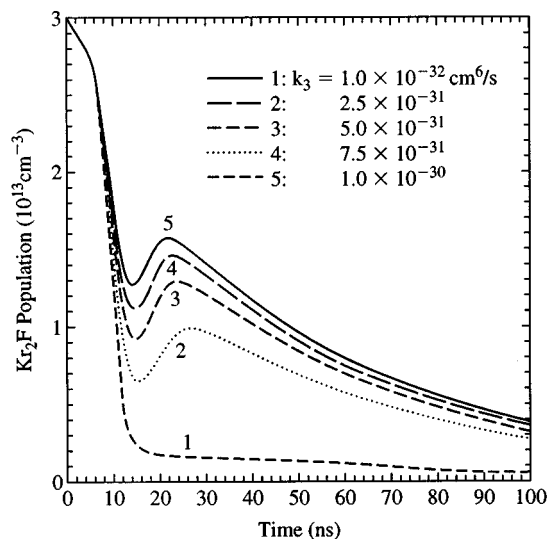


FIG. 7. Kr₂F(4 ²Γ) fluorescence suppression wave forms calculated for five values of the three body Kr₂F formation rate constant k_3 , ranging from 10^{-32} cm⁶ s⁻¹ to 10^{-30} cm⁶ s⁻¹.

respectively. Also, the initial Kr₂F(4 ²Γ) number density is taken to be 3×10^{13} cm⁻³ while the initial KrF(B,C) populations are set to zero. When $\beta=0$, re-feeding of the Kr₂F(4 ²Γ) state does not occur and, on the basis of a comparison with experimental wave forms, it is clear that β must be ≥ 0.75 . This result is consistent with the conclusions reached by McCown¹¹ who measured β for the photodissociation of both Kr₂F and Xe₂Cl in the 193–351 nm spectral interval. For Kr₂F, the branching ratio was determined to 0.9 ± 0.1 between 308 and 351 nm but vanished at 248 and 193 nm. Consequently, the result reported here—that β is constant in the 280–360 nm and 590–840 nm regions—is in agreement with the data of Ref. 11.

The model-generated fluorescence suppression wave forms are also sensitive to the temporal width of the probe laser pulse and to the assumed value for the rate constant k_3 . Most of the previous measurements of Kr₂F(4 ²Γ) absorption cross sections by fluorescence suppression have involved an excimer laser or excimer laser-pumped dye laser having pulsewidths in excess of 20 ns. Since this is large compared to the effective KrF(B) lifetime for the gas mixtures of interest here, the simulations confirm prior observations^{5–7} that little recovery of the Kr₂F fluorescence following the termination of the probe laser pulse is to be expected. Only as the probe pulsewidth falls below 10 ns [i.e., becoming comparable to the KrF(B) lifetime] does recovery of the wave form become observable. The results of Fig. 7 illustrate the sensitivity of the degree and rate of fluorescence recovery to k_3 . Values in the literature for this rate constant differ by more than an order of magnitude but it is clear from the simulated wave forms that accounting for the experimental data requires that k_3 be $\geq 5 \times 10^{-31}$ cm⁶ s⁻¹. Finally, the influence of re-feeding of the Kr₂F state is clearly illustrated by Fig. 8 in which the variation of the *measured* fluorescence suppression

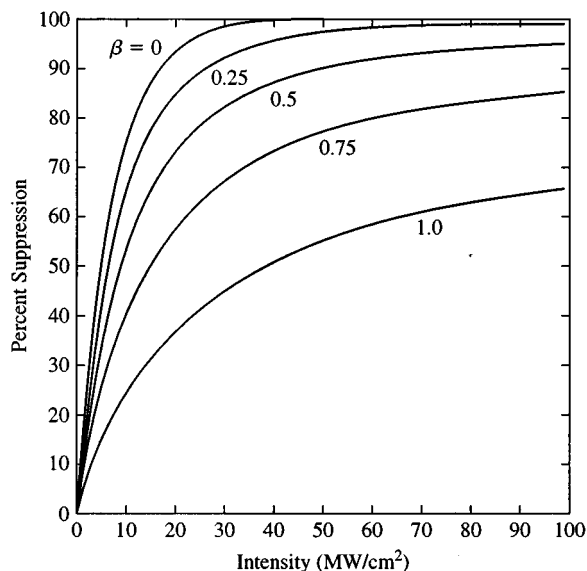


FIG. 8. Calculated dependence of the percentage fluorescence suppression on the probe laser intensity. The results are shown for five values of β . The actual dye laser wave form was incorporated into the simulations.

sion on probe laser intensity is shown for five values of β . Not surprisingly, as the efficiency for replenishing the $\text{Kr}_2\text{F}(4^2\Gamma)$ population is increased ($\beta \rightarrow 1$), the apparent (measured) absorption cross section is smaller than the true value and the effect is most pronounced at low probe laser intensities. However, it should be mentioned that the magnitude of the correction for this effect is reduced dramatically for longer probe laser pulses.

IV. SUMMARY AND CONCLUSIONS

Photodissociation of the $4^2\Gamma$ state of Kr_2F has been studied in the 280–850 nm wavelength region by fluorescence suppression and laser excitation spectroscopy. Bands attributed to the $9^2\Gamma \leftarrow 4^2\Gamma$ ($\lambda_{\text{max}} \sim 320$ nm) and $6^2\Gamma \leftarrow 4^2\Gamma$ ($\lambda_{\text{max}} \sim 710$ nm) transitions of the molecule have been observed, the latter for the first time. Weaker features lying in the blue (420–490 nm) and orange (~ 600 nm) regions of the spectrum were also detected and tentatively assigned to transitions terminating on the $8^2\Gamma$ and $7^2\Gamma$ states, respectively, of the triatomic.

Photoexcitation of $\text{Kr}_2\text{F}(4^2\Gamma)$ in the UV or near-IR results in dissociation of the molecule and the production of a KrF fragment in the C or D excited states. Comparing the Kr_2F absorption (fluorescence suppression) and $\text{KrF}(B)$ excitation spectra demonstrate that the quantum yield for KrF^* production is independent of wavelength in the 280–360 nm and 590–840 nm intervals. Modeling of the experimental fluorescence suppression wave forms indicates that the quantum yield for the process: $\text{Kr}_2\text{F}(4^2\Gamma) + \hbar\omega$ (3.4–4.4 eV) $\rightarrow \text{Kr}_2\text{F}(9^2\Gamma) \rightarrow \text{KrF}(D) + \text{Kr}$ is ≥ 0.75 . These results largely confirm the predictions of *ab initio* calculations concerning the positions and relative strengths of Kr_2F absorption bands as well as predissociation of the $6^2\Gamma$ and $9^2\Gamma$ excited states.

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