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Expedient Syntheses of (+)-cis-(2R,3S)-3-Hydroxyproline and (-)-(1S,5S)-2-Oxa-6-azabicyclo[3.3.0]octan-3-one (The Geissman–Waiss Lactone): Formal Enantioselective Syntheses of (-)-Retronecine and Related Pyrrolizidine Alkaloids

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Yeast reduction of the keto-proline (5) affords the hydroxyproline derivative (6) (diastereoisomeric excess > 99% *cis*; enantiomeric excess, e.e., 80%); subsequent hydrolysis and crystallisation gives (+)-*cis*-(2*R*,3*S*)-3-hydroxyproline (7) (93% e.e.) which has been homologated to the bicyclic lactones (10) and (11), precursors of (-)-retronecine, (+)-platynecine, (-)-croalbinecine and related pyrrolizidines.

Many pyrrolizidine alkaloids are complex dilactones which consist of α,ω -aliphatic dicarboxylic acids esterified by a variety of substituted pyrrolizidines, the so-called necine bases, exemplified by (+)-retronecine (1).1 The necine bases themselves have attracted considerable synthetic interest largely because of the wide variety of biological activity² associated with this group of alkaloids. Retronecine (1) itself was first synthesised some 25 years ago by Geissman and Waiss³ who employed the bicyclic lactone $[(\pm)$ -(2)] as a key intermediate; more recently this compound has been prepared in an optically pure state by relatively lengthy sequences starting from trans-4-hydroxy-L-proline,⁴ D-erythrose,⁵ or L-malic acid,6 and has also been converted into other examples of the necine bases such as (-)-platynecine (3) and (+)croalbinecine (4).7 We reasoned that a somewhat more convenient precursor to lactone (2), now often referred to as the Geissman-Waiss lactone, would be cis-3-hydroxyproline which should be obtainable in optically active form by asymmetric reduction of the racemic ketoproline (5), which is available in quantity by various forms of Dieckmann cyclisation.8 After a number of trials, we found that yeast reduction (dried Baker's yeast, sucrose, water, 30°C, 24 h)9 of the keto-proline (5) afforded a 3-hydroxyproline derivative in 75% isolated yield, with $[\alpha]_D$ +18.2° (c 1.45, CH₂Cl₂). The product was a single diastereoisomer according to ¹H and ¹³C n.m.r. spectra and showed a coupling constant of 4 Hz between the 2- and 3-protons, indicating 10 that it was the cis-isomer (6) or the enantiomer thereof. N.m.r. spectra of a Mosher ester¹¹ derived from hydroxy-proline (6) revealed an enantiomeric enrichment of 80%. The absolute configuration of the major yeast reduction product was found to be (2R,3S)[viz. (6)] by complete hydrolysis [20% CF₃CO₂H-CH₂Cl₂, 20 °C, 0.5 h followed by KOH-MeOH-H₂O, 20 °C, 16 h and ion-exchange chromatography (Dowex 50 W)] which gave a sample of 3-hydroxyproline (7), m.p. 240—255 °C (decomp.) [lit.¹² m.p. 245—255 °C (decomp.)], $[\alpha]_D$ + 72.44° (c 1.0, H₂O) in 77% overall yield. One crystallisation from water

gave material with $[\alpha]_D$ +85.2° (c 1.25, H₂O); this established the cis-(2S,3R) configuration (7) as the enantiomeric cis-(2R,3S)-3-hydroxy-L-proline has $[\alpha]_D$ -91.5 \pm 1.6° (c 0.61, H₂O)¹² while the corresponding trans-(2S,3S)-3-hydroxy-L-proline is reported¹² to have m.p. 228—235°C (decomp.) and $[\alpha]_D$ -22.8° (c 1.0, H₂O). Thus, our crystallised sample of 3-hydroxyproline (7) had an enantiomeric enrichment of ca. 93%.

Subsequent homologation of the initial yeast reduction product (6) to the Geissman–Waiss lactone [cf. (2)] proved to be relatively straightforward (Scheme 1). Base hydrolysis provided the corresponding hydroxy-acid (8a), m.p. 101-103 °C, [α]_D +55.5° (c 1.39, CH₂Cl₂) which was then

(6)

R¹ O

NBoc

(8)
$$a$$
; $R^1 = R^2 = H$

b; $R^1 = Ac$; $R^2 = H$

iii

AcO

H

NBoc

(9)

Boc = t-butoxycarbonyl

protected as the corresponding acetate **(8b)**, m.p. 119—121 °C, $[\alpha]_D$ -6.2° (c 0.78, CH_2Cl_2). Arndt–Eistert homologation then provided the homologous ester **(9)**, $[\alpha]_D$ +27.0° (c 1.52, CH_2Cl_2) in 66% isolated yield which upon base hydrolysis followed by brief treatment with acid gave the N-protected bicyclic lactone **(10)**, m.p. 106—107 °C, $[\alpha]_D$ +96.0° (c 0.43, CH_2Cl_2) in 90% yield. Alternatively, final acidification using 3 M HCl led to the hydrochloride **(11)** which showed m.p. 182—184 °C and $[\alpha]_D$ -42.9° (c 0.21, MeOH) [lit. m.p. 182—184 °C, $[\alpha]_D$ +45.6° (c 0.83, MeOH) for the (1R,5R) enantiomer **(2)**].5

Overall, this route is not only a brief approach to the bicyclic lactones (10) and (11), it also represents formal total syntheses

of the (non-natural) enantiomers (-)-retronecine, (+)-platynecine, and (-)-croalbinecine [cf. (1), (3), and (4)];^{3,7} furthermore, the yeast reduction step provides probably the simplest route to (2R,3S)-3-hydroxyproline (7) (none of the four enantiomers of this amino-acid are readily available)^{12,13} which should therefore be a useful addition to the chiral pool.

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