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Botrylactone: new interest in an old molecule—review of its absolute configuration and related compounds

Javier Moraga, Cristina Pinedo, Rosa Durán-Patrón, Isidro G. Collado, Rosario Hernández-Galán*

Departamento de Química Orgánica, Facultad de Ciencias, UCA, Polígono Rio San Pedro s/n, 11510 Puerto Real, Cádiz, Spain

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ABSTRACT

The absolute configuration of botrylactone, a unique compound with an interesting polyketide lactone skeleton with two oxirane bridges previously isolated from *Botrytis cinerea* and described as a powerful antibiotic, has been reviewed on the basis of sign of the optical rotation, NOE experiments and NMR method. The isolation of 7-deoxybotrylactone and 5-hydroxy-7-(4-hydroxydec-2(3)-enoyl) botrylactone enables us to characterize an intriguing new family of compounds with this interesting polyketide skeleton. A common biosynthetic origin with botcinin derivatives is proposed.

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1. Introduction

Botrytis cinerea is a well-known pathogen affecting a number of commercial crops, which produces several structurally unique metabolites. *Botrytis* produces two series of phytotoxic metabolites: a family of characteristic sesquiterpene metabolites with the basic botryane skeleton, principally botrydial and dihydrobotrydial and a family of polyketide lactones. The first isolated polyketide was reported by Cutler in 1993 who called it botcinolide (**1**), proposing a highly hydroxylated nonanolactone structure.¹ Based on botcinolide structures and in addition to other botcinolide metabolites, we described² new metabolites and investigated the biosynthesis of the botcinolide skeleton.³

In 2005, Nakajima's group reported the isolation of a group of antifungal metabolites, which they designated as botcinins.⁴ The absolute configuration of botcinin A (**2**) was determined through the modified Mosher method. A careful reinvestigation of the spectroscopic data reported for botcinolide analogues allowed them to revise the structures of botcinolide derivatives to botcinic (**3**) and botcineric (**4**) acids and their cyclized derivatives, botcinins $A-F.^5$ Ultimately, the revised structures of this group of natural products were unequivocally determined by total synthesis⁶ and a revision comparing the botcinolides with their corresponding botcinin structures has been reported.⁷

Recently, a genetically modified strain of *B. cinerea*, $\Delta bcbot2$, which is unable to produce botryanes but significantly overproduces botcinic acid (**3**) and its derivatives was constructed.⁸ The higher production capacities of this strain prompted us to reinvestigate the

metabolites produced by it. Additionally, in the course of our investigation on biotransformation with a wild strain of *B. cinerea* (UCA 992), a new metabolite with a botrylactone skeleton was isolated. This paper focuses on the characterization and biosynthetic route proposal of an intriguing new family of compounds, botrylactones, which could be an intermediate in the biosynthesis to botcinins. Additionally, the absolute configuration of botrylactone (**6**) and the structure of 2-epihomobotcinolide (**5**) are revised.

2. Results and discussion

Botrylactone is a unique C-9 polyhydroxylated lactone described as a powerful antibiotically active compound reported by Welmar et al.⁹ Several unsuccessful attempts to isolate this intriguing compound have been made and some of them reported.¹⁰ Recently we isolated it from the *B. cinerea* cat 2 strain.³

Isolation of botrylactone together with botcinins,³ and the similarity of their spectroscopic data, led us to believe that they are closely related metabolites. However, comparison of the stereo-chemistry proposed for botrylactone (**6**) and that proposed for botcinin A (**2**) turned out be diametrically opposed. These data, together with the overproducer mutant $\Delta bcbot2$, prompted us to conduct a study of **6** focusing on botcinic acid derivatives.

The *Botrytis cinerea* mutant, $\Delta bcbot2$, was cultured on malt agar medium at 24–26 °C for 7 days. The fermentation broth, after filtration and extraction was purified following the methodology described in the Experimental section. In addition to botcinin A (**2**) and B (**7**), (+)-botrylactone (**11**) and 3-acetylbotcineric acid (**8**), a metabolite whose spectroscopic and physical constants were identical to those described for 2-epihomobotcinolide (**5**)³ and a new compound with a botrylactone skeleton, 5-hydroxy-7-(4-hydroxydec-2(3)-enoyl)botrylactone (**12**), were isolated.





^{*} Corresponding author. Tel.: +34 956016371; fax: +34 956 016193; e-mail address: rosario.hernandez@uca.es (R. Hernández-Galán).

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A careful spectroscopic study of the physical constants of **5** and its comparison with those of botcinin derivatives described in the literature showed that the ¹H NMR and ¹³C NMR of **5** were very similar to those of botcinin E (**9**).⁵ The ¹³C NMR, including DEPT data, showed that **5** has two more methylenes than **9**. The fragment ion assignable to the fatty acyl portion was detected at *m*/*z* 169, which is characteristic of a C₁₀H₁₆O. Thus, **5** differ from **9** in the length of the acyl portion. The key NOE correlations were identical to those of **9** indicating that **5** and **9** share the same relative stereochemistry. Therefore, the structure of 2-epihomobotcinolide (**5**) should be revised to **10** and renamed as botcinin G.

Botrylactone was reported for the first time by Welmar et al.⁹ who initially proposed structure **6a** based on spectroscopic data. Although they used the term, 'absolute configuration' in the text of the paper, they did not determine the absolute configuration but its relative configuration by X-ray diffraction of its acetate derivative (**6b**).⁹ Later, Redlich et al. synthesized it and revised the originally published structure to 7-OH α -botrylactone (**6**).¹¹

The optical rotation of natural botrylactone had not been previously reported. Welmar et al. prepared acetyl botrylactone and reported its optical rotation as $+88 (c \ 1, CHCl_3)$,⁹ while Redlich et al. described it for synthetic botrylactone (**6**) as $-31 (c \ 0.14, MeOH)$.¹¹

The isolation of a sufficient amount of natural botrylactone from $\Delta bcbot2$ has enabled us to measure its optical rotation determined as +19 (c 2.6, CHCl₃), the opposite sign of that described for the synthetic compound. Acetylation of it led us to a product whose spectroscopic data were identical to those described for natural acetyl botrylactone whose optical rotation proved to be +73 (c 2.4 mg, $CHCl_3$), coinciding with the sign described by Welmar et al.⁹ As a result, it can be inferred that the structure synthesized by Redlich et al. should be the enantiomeric to the natural compound, (-)-botrylactone (6). In order to confirm this hypothesis we determined the absolute configuration of natural (+)-botrylactone using the NMR method. (+)-Botrylactone was treated with R(-) and *S*(+)-MPA acid yielding the corresponding *R* and *S*-MPA esters **11a** and **11b**, respectively. Comparison of the chemical shifts in ¹H NMR spectra of the two compounds showed a negative $\Delta \delta^{\text{RS}}$ value for C_8 -CH₃ and C_9 -CH₃, (-0.57 and -0.26 ppm, respectively), while the $\Delta \delta^{\text{RS}}$ for H-6 and C₆–CH₃ were positive (0.11 and 0.47 ppm). Application of the Mosher rule¹² showed a 7*R* configuration for esterificated products **11a** and hence for (+)- botrylactone (**11**).

The stereochemistry of the rest of the chiral carbons was confirmed on the basis of NOE data (Fig. 1). Irradiation of the H-7 signal of compound **11a** produced NOE enhancement of H-5 α , C₆–CH₃ and C₉–CH₃, establishing a boat conformation for this tetrahydropyran ring, which was confirmed by irradiation of H-5 β that enhanced the



Fig. 1. NOEs interactions in 11a.

signals of H-6, C₆—Me, C₄—Me and H-5 α . A quasi-boat conformation was established for pentanolide ring on the basis of NOE enhancement of C₂—Me, C₄—Me and H-3 observed when H-2 was irradiated. These data coincide with a configuration where H-2, C₄—Me and C₈—Me were on the β face and H-3, H-7, C₆—Me and C₉—CH₃ were on the α . The absolute stereochemistry of natural botrylactone, (+)- **11**, was definitively established as 2*R*, 3*S*, 4*S*, 6*S*, 7*R*, 8*R*, 9*R*.

Compound **12** was isolated as an oil whose molecular formula was established as $C_{24}H_{38}O_8$ by HRMS and ¹³C NMR data requiring 6° of unsaturation. Its ¹H NMR spectrum was very similar to that of botcinins however the presence of a signal in ¹³C NMR at δ 104.2 ppm, attributable to a ketal group plus an additional methyl group, should correspond to an additional C₂ unit resulting in a structure similar to that of botrylactone (11). However, the ¹H NMR showed signals at H 7.00, 6.07 and 4.33 ppm characteristic of the fatty acyl portion on C-7 of botcinins. The fragment ion detected at m/z169 was characteristic of a fatty acyl chain with molecular formula C₁₀H₁₆O. The HMBC experiment performed on 12 showed correlations between the quaternary ketalic carbon signal (δ_{C} 104.2 ppm), two methyl singlet groups ($\delta_{\rm H}$ 1.53 and 1.10 ppm), which were further correlated with a quaternary oxygenated carbon at $\delta_{\rm C}$ 79.2 ppm and two signals at $\delta_{\rm H}$ 4.93 and 3.52 ppm attributable to H-7 and H-3, respectively (Fig. 2). H-3 was further correlated with signal at δ_C 171.0 ppm corresponding to carbon C-1 while H-7 exhibited correlation with the carbon signal at δ_{C} 165.4 ppm corresponding to the C-1 of the fatty acyl portion. Additional correlations were observed between carbon C-3 (δ_{C} 80.9 ppm) and two methyl groups at δ_{H} 1.46 (d) and 1.14 (s) ppm. The COSY experiment showed correlations between the signal doublet at $\delta_{\rm H}$ 4.93 ppm (H-7) and a multiplet at $\delta_{\rm H}$ 1.86 ppm, which was further correlated with two doublets at $\delta_{\rm H}$ 3.67 ppm and 1.02 ppm, pointing to the presence of a fragment O-CH-CH(CH₃)-CH-O in the molecule. All these data are consistent with a structure of 5-hydroxybotrylactone bearing a fatty acyl chain on C-7 for this compound. The stereochemistry was determined by N.O.E experiments and was consistent with the stereochemistry assigned to compound 11 confirming the structure of 2R, 3S, 4S, 5S, 6S, 7R, 8R, 9R-5-hydroxy-7-(4-hydroxydec-2(3)-enoyl) botrylactone for compound 12.



Fig. 2. Selected HMBC-COSY correlations for 12 and 13.

Furthermore, during the course of our biotransformation experiments with *B. ciner*ea UCA 992¹³ we found a new compound **13** with an NMR pattern very similar to that of (+)-botrylactone (**11**), the principal difference being the absence of the characteristic H-7 signal in its ¹H NMR spectrum and one of the carbon bearing to oxygen and the presence of a signal corresponding to one more methylene in the ¹³C NMR spectrum. Therefore, **13** differ from **11** in the absence of the hydroxyl group on C-7. HRMS confirms this showing an ion assignable to $C_{13}H_{22}O_3$ [M–CO]⁺ at *m*/z 226.1573. The botrylactone skeleton was confirmed by the 2D NMR data where correlation in the HMBC experiment between the characteristic

quaternary carbon signal (δ_c 104.8 ppm) and signals corresponding to H-3, H₂-7 and two methyl singlet groups, which were further correlated with C-8 were observed. The key NOE correlations were identical to those of **11a**, indicating that **11** and **13** share the same relative stereochemistry. On the basis of these data the structure of **13** was suggested to be 2*R*, 3*S*, 4*S*, 6*S*, 8*R*, 9*R*-7dehydroxybotrylactone.

The occurrence of botrylactone derivatives and botcinins in the same strain, their structural homology, the resulting absolute configuration of natural botrylactone (**11**) therefore being identical to that assigned to botcinins by Nakajima⁴ except in C-8, seems to indicate that both compounds may be biogenetically related.

Incorporation studies with ¹³C and ²H-labelled precursors conducted by our group³ indicated that 3-O-acetylhomobotcinolide (**14**) was an acetate derived polyketide whose methyl groups originate from *C*-methyl methionine. This conclusion is applicable to its revised structure, **8**, but an important question, the presence of the methyl group derived from *C*-methyl methionine on the C-8 carbon atom, the alleged carbon two of the starter-acetate unit, has yet to be resolved.

3. Conclusions

A plausible explanation for this, which has been previously reported for other natural products, such as aurovertin,¹⁴ is that *B. cinerea* biosynthesis involves a C_{10} -polyketide, which is methylated at activated methylene groups, followed by the loss of the starter-acetate unit through a retro-Claisen type C–C bond cleavage with inversion of configuration at C-8 (Scheme 1).



Scheme 1. Biosynthetic route proposal for botrylactones and botcinins.

This hypothesis is in agreement with the common biosynthetic origin of botrylactone (**11**) and botcinins, where a hypothetical bicyclic acid intermediate **15** could be the branching point to give botrylactone or botcinin derivatives as shown in Scheme 1. This proposal explains the presence of the methyl group on C-8 of botcinins and the different stereochemistry in this carbon.

4. Experimental section

4.1. General experimental procedures

¹H and ¹³C NMR measurements were recorded on Varian Unity 400 MHz and Varian Inova 600 MHz spectrometers with SiMe₄ as the internal reference. Chemical shifts were referenced to CDCl₃ ($\delta_{\rm H}$ 7.25, $\delta_{\rm C}$ 77.0). HPLC was performed with a Hitachi/Merck L-6270 apparatus equipped with a differential refractometer detector (RI-7490). A Lichrofer Si 60 (5 μm) LichoCart (250 mm×4 mm) column and a Lichrofer Si 60 (10 μm) LichoCart (250 mm×10 mm) were used in isolation experiments. Silica gel (Merck) was used for column chromatography. TLC was performed on Merck Kiesegel 60 F_{254} , 0.25 mm thick.

4.2. Microorganism

B. cinerea mutant strain, *bcbot2* Δ , was supplied by Dr. Muriel Viaud of the UMR BIOGER, INRA (Versailles, France). The strain was maintained viable on mycelia discs of 0.5 cm diameter submerged in 80% glycerol at -40 °C.

B. cinerea (UCA 992) was obtained from grapes from Domecq vineyard, Jerez de la Frontera, Cadiz, Spain. This culture is deposited at the Universidad de Cadiz, Facultad de Ciencias Mycological Herbarium Collection (UCA).

4.3. Culture conditions

bcbot2 Δ was grown on malt agar medium (20 g of D-glucose, 10 g of malt extract, 20 g of agar, pH 6.57 per liter of water) at 25 °C and used to inoculate Roux bottles or Erlenmeyer flasks. For surface culture, mycelium was grown in 1 L Roux bottles containing 150 mL of modified Czapek-Dox medium (50 mg of D-glucose, 1 g of yeast extract, 5 g of KH₂PO₄, 2 g of NaNO₃, 0.5 g of MgSO₄·7H₂O and 0.01 g of FeSO₄·7H₂O, pH 6.57.0 per liter of water) at room temperature. For shaken cultures, mycelium was grown in Erlenmeyer flasks containing 200 mL of the same medium agitated on an orbital shaker at 140 rpm at 25 °C. Each Roux bottle or Erlenmeyer flask was inoculated with mycelium on six small slices of agar (1 cm).

4.4. Extraction and isolation of metabolites

After 7 days of incubation under fluorescent light, the culture media were filtered, saturated with NaCl, extracted with ethyl acetate (3×0.5 vol) and washed with water (3×0.25 vol). The organic extracts were dried over Na₂SO₄ and concentrated to dryness.

Preliminary fractionation of the extracts was achieved by column chromatography eluting with petroleum ether/ethyl acetate mixtures containing increasing percentages of ethyl acetate (10–100%) to give 10 fractions F1–F10. Final purification of each fraction was carried out by means of semi-preparative or analytical HPLC. Botcinin A (**2**), botcinin B (**2a**), botcinin G (**10**), 3-O-acetylbotcineric acid (**8**), (+)-botrylactone (**11**) and 5-hydroxy-7-(4hydroxydec-2(3)-enoyl)botrylactone (**12**) were obtained.

4.4.1. Botcinin A (**2**) and B (**2a**)⁴. Compounds 2 and 2a were obtained from purification of fractions F4 and F5 as a colourless oil. Semi-preparative HPLC: hexane–ethyl acetate 70:30; flow 3 mL min⁻¹; t_R around 29 and 35 min, respectively.

4.4.2. Botcinin G (10). Compound 10 was obtained from purification of fractions F7 as a colourless oil. Semi-preparative HPLC: hexane–ethyl acetate 75:25; flow 3 mL min⁻¹; $t_R = 7.0$ min $[\alpha]_D^{35}$ -34° (c 2.0, CHCl₃); IR v_{max} (film) 3452, 2933, 2870, 1723, 1652, 1456, 1166 cm⁻¹; ¹H NMR (400 MHz, CD₃OD) δ 7.02 (dd, 1H, J=15.5, 4.9 Hz, H-3'), 6.05 (dd, 1H, J=15.5, 1.7 Hz, H-2'), 4.50 (dd, 1H, J=10.5, 9.7 Hz, H-7), 4.25 (m, 1H, H-4'), 4.08 (d, 1H, J=9.4 Hz, H-3), 3.94 (d, 1H, *J*=10.9 Hz, H-5), 3.77 (dq, 1H, *J*=9.7, 5.9 Hz, H-8), 3.20 (dq, 1H, J=9.4, 7.3 Hz, H-2), 2.21 (m, 1H, J=10.5, 6.4 Hz, H-6), 1.49 (m, 2H, H-5'), 1.31 (m, 2H, H-5'), 1.17 (m, 6H, H-7', H-8', H-9"), 1.17 (s, 3H, C₄-CH₃), 1.11 (d, 3H, J=5.9 Hz, C₈-CH₃), 1.13 (d, 3H, J=7.3 Hz, C2-CH3), 1.07 (d, 3H, J=6.4 Hz, C6-CH3), 0.89 (t, 3H, J=5.0 Hz, H-10'); ¹³C NMR (100 MHz, CD₃OD) 177.4 (s, C-1), 167.6 (s, C-1'), 154.0 (d, C-3'), 119.9 (d, C-2'), 79.7 (d, C-5), 77.9 (d, C-7), 78.1 (s, C-4), 75.0 (d, C-3), 71.6 (d, C-12), 69.6 (d, C-8), 39.6 (d, C-2), 37.5 (t, C-5'), 36.8 (d, C-6), 32.9 (t, C-8'), 30.2 (t, C-7'), 26.4 (t, C-6'), 23.6 (t, C-9'), 18.6 (q, C₈-CH₃), 14.3 (q, C-10'), 13.9 (q, C₆-CH₃), 11.5 (q, C₄-CH₃), 10.4 (q, C₂-CH₃); EIMS *m*/*z* (rel int.) 394 [M-H₂O]⁺ (9), 226 (26), 169 (9), 140(23), 124 (83), 109 (100); HREIMS calcd for $C_{22}H_{34}O_6\,[M-H_2O]^+$ 394.2355, found 394.2353.

4.4.3. 3-O-Acetylbotcineric acid (**8**). Compound **8** obtained from fractions F9 and F10. Semi-preparative HPLC: hexane–ethyl acetate 95:5; flow 3 mL min⁻¹; $t_{\rm R}$ =13.2 min. Colourless oil; $[\alpha]_{\rm D}^{20}$ –5.9° (*c* 1.4, ethyl acetate).¹⁵

4.4.4. (+)-*Botrylactone* (**11**). Compound **11** obtained from fractions F5. Analytical HPLC: hexane—ethyl acetate 65:35; flow 1 mL min⁻¹; $t_{\rm R}$ =28.98 min. Colourless oil; [α]_D²⁵ +19° (*c* 2.3, CHCl₃).^{9,11}

4.4.5. 2R, 3S, 4S, 6S, 8R, 9R-5-Hydroxy-7-(4-hydroxydec-2(3)-enoyl) botrylactone (12). Compound 12 obtained from fractions F9 and F10. Semi-preparative HPLC: hexane-ethyl acetate 95:5; flow 3 mL min⁻¹; $t_{\rm R}$ =15.1 min. Colourless oil; $[\alpha]_{\rm D}^{20}$ +10.6° (c 0.9, ethyl acetate); IR v_{max} (film) 3445, 2931, 2859, 1727, 1652, 1456, 1116 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.00 (dd, 1H, J=15.6, 4.5 Hz, H-3'), 6.07 (dd, 1H, J=15.6, 1.6 Hz, H-2'), 4.93 (d, 1H, J=10.9 Hz, H-7), 4.33 (m, 1H, H-4'), 3.67 (d, 1H, J=11.1 Hz, H-5), 3.52 (s, 1H, H-3), 2.75 (q, 1H, J=7.4 Hz, H-2), 1.99 (m, 1H, H-6), 1.60 (m, 2H, H-5'), 1.53 (s, 3H, C₉-CH₃), 1.46 (d, 3H, J=7.4 Hz, C₂-CH₃), 1.28 (m, 8H, H-6', H-7', H-8', H-9'), 1.14 (s, 3H, C₄-CH₃), 1.10 (s, 3H, C₈-CH₃), 1.02 (d, 3H, J=6.4 Hz, C₆-CH₃), 0.87 (t, 3H, J=6.8 Hz, H-10'); ¹³C NMR (100 MHz, CDCl₃) 171.0 (s, C-1), 165.4 (s, C-1'), 151.9 (d, C-3'), 119.0 (d, C-2'), 104.2 (s, C-9), 80.9 (d, C-3), 79.2 (s, C-8), 76.2 (s, C-4), 75.1 (d, C-5), 74.2 (d, C-7), 71.1 (d, C-4'), 36.7 (d, C-5'), 36.5 (t, C-6'), 34.5 (d, C-2), 31.6 (t, C-8'), 29.1 (t, C-7'), 25.2 (t, C-6'), 22.5 (t, C-9'), 21.7 (q, C₉-CH₃), 18.5 (q, C₂-CH₃), 18.2 (q, C₄-CH₃), 18.1 (q, C₈-CH₃), 14.0 (q, C-10'), 13.8 (q, C₆-CH₃); EIMS m/z (rel int.) 454 [M]⁺ (1), 426 [M-CO]⁺ (7), 410 [M-CO₂]⁺ (6); HREIMS calcd for C₂₃ H₃₈O₇ [M-CO]⁺ 426.2618, found 426.2641.

4.5. Acetylation of botrylactone

Botrylactone (**11**, 10 mg) was dissolved in dry pyridine (1 mL, 0.001 mmol) and acetic anhydride (2.4 mL, 24,6 mmol) was added dropwise. The reaction mixture was stirred for 24 h. Then the solvent was removed and the crude reaction product chromatographed to give botrylactone acetate (**11c**). Colourless oil; $[\alpha]_D^{25}$ +73, (*c* 2.4 mg, CHCl₃).⁹

4.6. α-Methoxyphenylacetyl ester of botrylactone

A solution of the botrylactone (**11**, 10 mg, 0.037 mmol) in dry dichloromethane CH_2Cl_2 (1.5 mL) was treated with DMAP (9.05 mg, 2.0 equiv) and (+)-(2S)- or (-)-(2*R*)-2-methoxy-2-phenylacetic acid MPA (13.85 mg, 2.25 equiv). After 15 min stirring at room temperature, EDC (14.91 mg, 2.1 equiv) was added. Stirring was maintained for 24 h. The solvent was stirred under reduced pressure. Residue purification was achieved by flash column chromatography on silica gel (elution with 60:40 hexane/ethyl acetate).

4.6.1. (*R*)- α -Methoxyphenylacetyl ester of botrylactone (**11a**). ¹H NMR (400 MHz, CDCl₃) δ 7.40 and 7.33 (m, 2H and 3H,C₂'–Ph), 4.87 (d, 1H, *J*=10.8 Hz, H-7), 4.75(s, 1H, H-2'), 3.39 (s, 3H, C₂'–OMe), 3.27 (br s, 1H, H-3), 2.63 (q, 1H, *J*=7.3 Hz, H-2), 1.90 (m, 1H, H-6), 1.71 (dd, *J*=4.7, 13.5 Hz, H-5 β), 1.55 (dt, *J*=2.7, 13.2 Hz, H-5 α), 1.39(d, 3H, *J*=7.3 Hz, C₂–CH₃), 1.26 (s, 3H, C₉–CH₃), 1.12 (s, 3H, C₈–CH₃), 0.91 (d, 3H, *J*=6.4 Hz, C₆–CH₃), 0.86 (t, 3H, *J*=7.2 Hz, H-10').

4.6.2. (*S*)- α -Methoxyphenylacetyl ester of botrylactone (**11b**). ¹H NMR (400 MHz, CDCl₃) δ 7.44 and 7.38 (m, 2H and 3H,C₂'–Ph), 4.81 (d, 1H, *J*=10.8 Hz, H-7), 4.76(s, 1H, H-2'), 3.40 (s, 3H, C₂'–OMe), 3.27 (br s, 1H, H-3), 2.67 (q, 1H, *J*=7.3 Hz, H-2), 1.79 (m, 1H, H-6), 1.62 (dd, *J*=5.0, 13.5 Hz, H-5 β), 1.55 (t, *J*=13.2 Hz, H-5 α), 1.52 (s, 3H, C₉–CH₃),

1.42(d, 3H, J=7.3 Hz, C_2-CH_3), 1.12 (s, 3H, C_4-CH_3), 1.04 (s, 3H, C_8-CH_3), 0.44 (d, 3H, J=6.4 Hz, C_6-CH_3).

4.7. Biotransformation of 2-benzylideneindan-1-one

Botrytis cinerea UCA 992 was grown at 25 °C on a Czapeck-Dox medium (200 mL per flask). The shaken culture was incubated in an orbital shaker at 140 rpm under fluorescent light. 2-Benzylideneindan-1-one was dissolved in ethanol and then distributed over 12 flasks (150 ppm per flask) and the fermentation continued for 5 days in six flasks and 10 days in the others. The mycelium was then filtered and the broth was extracted as described below. The solvent was then evaporated and the residue was purified first on a silica gel column and then with HPLC with an increasing gradient of ethyl acetate to petroleum ether.

Chromatography of the extract fermented for 5 days produced 2-benzylideneindan-1-one (4 mg), *O*-methyldihydrobotrydial (3.2 mg), botrydial (4 mg), dihydrobotrydial (3 mg), 7-deoxybo-trylactone (**13**,2.8 mg) and 2-(*p*-hydroxyphenylmethyl)indan-1-one (50 mg) ($[\alpha]_D^{20}$ +6° (*c* 0.1, MeOH), 36% ee).

Chromatography of the extract fermented for 10 days produced 2-benzylideneindan-1-one (14 mg), botrydial (2 mg), dihydrobo-trydial (3 mg), 7-deoxybotrylactone (**13**, 15.3 mg), botcinic acid (**3**, 60 mg) and 2-(*p*-hydroxyphenylmethyl)-7-hydroxyindan-1-one (17 mg) $[\alpha]_D^{20}$ –3.7° (*c* 0.1, MeOH, 7% ee).

4.7.1. 7-Deoxybotrylactone (**13**). Compound **13** obtained from fraction F3. Column chromatography: hexane—ethyl acetate 80:20. Amorphous solid; $[\alpha]_D^{25} + 2.45^{\circ}$ (c 0.3 CHCl₃); IR v_{max} (film) 2925, 1735, 1459, 1103, 950 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 3.28 (s, 1H, H-3), 2.70 (q, 1H, *J*=7.3 Hz, H-2), 1.75 (m, 1H, H-6), 1.63 (m, 1H, H-5), 1.57 (m, 2H, H-7) 1.44 (s, 3H, C₉–CH₃), 1.40 (d, 3H, *J*=7.3 Hz, C₂–CH₃), 1.37 (m, 1H, H-5'), 1.26 (m, 1H, H-7'), 1.14 (s, 3H, C₄–CH₃), 1.18 (s, 3H, C₈–CH₃), 0.95 (d, 3H, *J*=6.7 Hz, C₆–CH₃); ¹³C NMR (100 MHz, CDCl₃) 171.7 (s, C-1), 104.8 (s, C-9), 81.7 (d, C-3), 75.5 (s, C-8), 71.6 (s, C-4), 75.6 (d, C-5), 71.6 (d, C-12), 41.9 (t, C-5), 39.9 (t, C-7), 34.5 (d, C-2), 26.17 (q, C₄–CH₃), 25.0 (q, C₈–Me), 23.4 (d, C-6), 21.6 (q, C₉–CH₃), 20.5 (q, C₆–CH₃), 18.6 (q, C₂–CH₃); EIMS *m*/*z* (rel int.) 226 [M–CO]⁺ (26), 166 (36),123 (83), 109 (100); HREIMS calcd for C₁₃H₂₂O₃ [M–CO]⁺ 226.1569, found 226.1573.

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