

Avicholic Acid: A Lead Compound from Birds on the Route to Potent TGR5 Modulators

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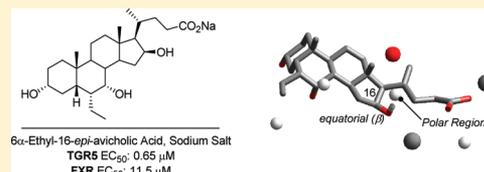
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Supporting Information

ABSTRACT: Grounding on our former 3D QSAR studies, a knowledge-based screen of natural bile acids from diverse animal species has led to the identification of avicholic acid as a selective but weak TGR5 agonist. Chemical modifications of this compound resulted in the disclosure of 6 α -ethyl-16-*epi*-avicholic acid that shows enhanced potency at TGR5 and FXR receptors. The synthesis, biological appraisals, and structure–activity relationships of this series of compounds are herein described. Moreover, a thorough physicochemical characterization of 6 α -ethyl-16-*epi*-avicholic acid as compared to naturally occurring bile acids is reported and discussed.



KEYWORDS: bile acid, TGR5, avicholic acid, 16-*epi*-avicholic acid, FXR, structure–activity relationship, CMC

Bile acid (BA)-activated receptors are attractive targets for drug discovery efforts.^{1–5} The chief members of this family, namely, farnesoid X receptor (FXR) and TGR5, have been implicated in a number of liver and metabolic diseases, including cholestasis, nonalcoholic steatohepatitis (NASH), obesity, and type II diabetes, to name a few. In the framework of the development of ligands for BA-activated receptors, starting from cholic acid (CA, 1) as a lead compound, we have recently reported the design, synthesis, and pharmacological effects of INT-777 (3, Figure 1) as a potent and selective TGR5

agonist in vivo.^{6,7} Via TGR5 activation, INT-777 (3) is able to stimulate type 2 iodothyronine deiodinase (D2) activity in brown adipose tissue (BAT) and muscle, as well as induce the release of glucagon-like protein 1 (GLP-1) in enteroendocrine cells. While these results have provided the first proof-of-concept that modulating TGR5 may represent a novel viable strategy for the treatment of metabolic disorders such as type 2 diabetes,⁷ further aspects of TGR5 activation still remain poorly understood. These include, in particular, the inhibition of hepatic inflammatory responses,^{8,9} the protection of liver against the production of reactive oxygen species in sinusoidal endothelial cells,¹⁰ and, more recently, the release of nitric oxide in enteric neurons leading to the control of intestinal motility.¹¹

As a continuation of our efforts aimed at providing potent and selective chemical tools with tailored pharmacokinetic properties to probe the functions of BA activated receptors in different tissues,^{6,12–17} we have been devising further chemical modifications of the BA side chain and nucleus. To this end, we hypothesized that the diverse BA biosynthetic pathways across animal species could be a potential source of novel lead compounds. Starting from our previous results of three-dimensional quantitative structure–activity relationship studies (3D QSAR) that evidenced how different regions around the BA scaffold affect TGR5 activity (Figure 2),¹⁸ we thus investigated naturally occurring BAs from 677 vertebrate

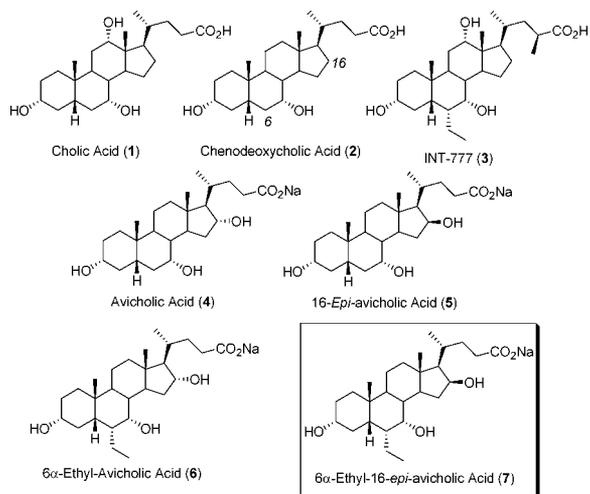


Figure 1. Natural and semisynthetic BA derivatives.

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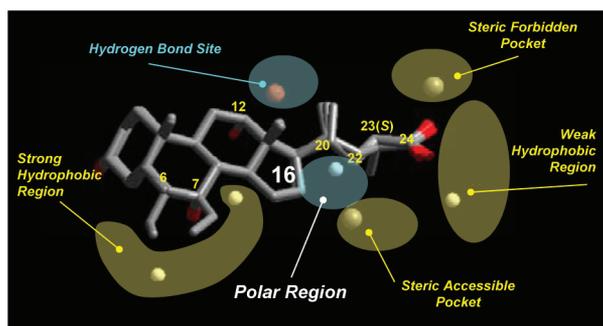


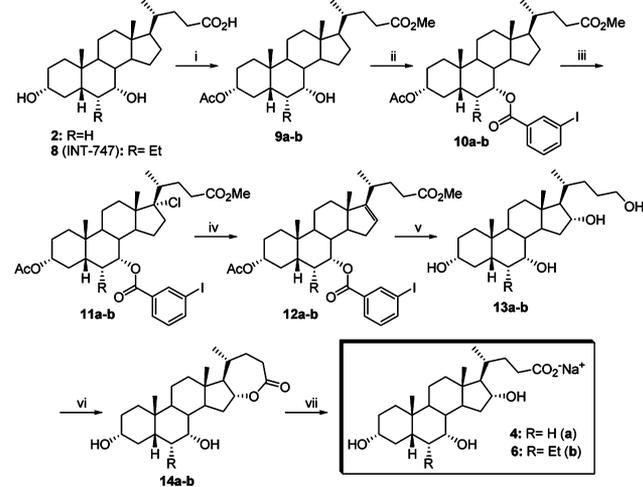
Figure 2. 3D QSAR scheme of TGR5 agonists.¹⁸

species¹⁹ to identify those that could be endowed with a suitable scaffold in terms of potential activity.

Our attention, in particular, was attracted by C₂₄ BAs bearing a hydroxy group at the C₁₆α-position, as determined in the biliary composition of birds and snakes. Polar groups at this position were indeed suggested by the QSAR model as favoring interaction with TGR5 (Figure 2).¹⁸ Accordingly, in this work, we report the synthesis and preliminary evaluation of 3α,7α,16α-trihydroxy-5β-cholan-24-oic acid (avicholic acid, **4**), a natural BA first isolated from avian species (Shoebill stork and herons) in 2002,²⁰ and its derivatives **5–7** as TGR5 ligands (Figure 1).

Avicholic acid derivatives **4** and **6** were prepared according to the synthetic approach reported previously by Mukhopadhyay (Scheme 1).²¹ Thus, CDCA (**2**) or 6-ECDCa (**8**)^{12,17} was

Scheme 1. Synthesis of Avicholic Acid Derivatives **4** and **6**^a



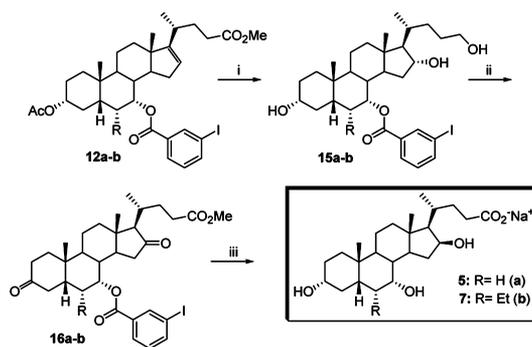
^aReagents and conditions: (i) (1) MeOH, *p*TSA; (2) Ac₂O, NaHCO₃, THF. (ii) 3-I-benzoylchloride, CaH₂, BnEt₃N⁺Cl⁻. (iii) PhICl₂, *t*-BuOH 0.3 M, *hν*. (iv) Py, reflux. (v) (1) BH₃·THF; (2) NaOH, H₂O₂; (3) KOH, MeOH. (vi) NCS, TEMPO, Aliquat, CH₂Cl₂/H₂O. (vii) NaOH, MeOH.

treated with *p*-toluensulfonic acid (*p*TSA) and MeOH at room temperature to get the corresponding methyl ester analogues, which were selectively protected at the C₃ position by reaction with acetic anhydride in the presence of NaHCO₃ in refluxing THF (quantitative yield). The 3-acetoxy intermediate **9a,b** thus formed was functionalized at the C₇ position with a 3-iodobenzoyl moiety by reaction with 3-iodo-benzoylchloride in the presence of CaH₂ and BnEt₃N⁺Cl⁻ in toluene to afford **10a,b** in good yield (80 and 74%, respectively).

Breslow's remote functionalization was achieved by photolysis of PhICl₂ in the presence of methyl 3α-acetoxy-7α-iodobenzoyloxy-5β-cholan-24-ate **10a,b** in 0.3 mM *t*-BuOH/CH₂Cl₂ solution to yield the 17-Cl derivative **11a,b**, in nearly quantitative yield. The corresponding C_{16–17} olefinic intermediate **12a,b** was obtained by refluxing **11a,b** in pyridine (81 and 72%). Hydroboration-oxidation of **12a,b** by borane–THF complex and alkaline hydrogen peroxide, followed by basic hydrolysis (NaOH/MeOH), furnished the tetrol **13a,b** in moderate yield (55 and 47%, respectively). Selective oxidation of the C₂₄ position was achieved by using (2,2,6,6-tetramethylpiperidin-1-yl)oxyl (TEMPO) in CH₂Cl₂/H₂O and Aliquat as a phase transfer catalyst to afford the lactone **14a,b** in 37 and 30% yield, respectively. Alkali hydrolysis (NaOH/MeOH) followed by a reverse phase RP-18 chromatography afforded avicholic acid sodium salt (**4**) and 6α-ethyl-avicholic acid sodium salt (**6**) in 10 and 6% overall yield, respectively.

The corresponding C₁₆-epimers **5** and **7** were prepared according to Scheme 2.²² Thus, the C_{16–17} olefinic BA

Scheme 2. Synthesis of 16-*epi*-Avicholic Acid Derivatives **5** and **7**^a



^aReagents and conditions: (i) (1) BH₃·THF; (2) NaOH, H₂O₂; (3) K₂CO₃, MeOH. (ii) (1) Jones reagent; (2) MeOH, *p*TSA. (iii) (1) *t*-BuNH₂BH₃, CH₂Cl₂, reflux; (2) NaOH, MeOH.

derivative **12a,b** was submitted to hydroboration-oxidation reaction using the previously reported protocol and selectively deprotected at C₃ position by using a solution of K₂CO₃ in MeOH at room temperature. The 7α-*m*-iodobenzoyloxy derivative **15a,b**, obtained in 56 and 50% yield, respectively, was reacted with Jones reagent and esterified at the terminal carboxylic group by treatment with *p*TSA in MeOH at room temperature to furnish **16a,b**, in nearly quantitative yield. Finally, reduction of the keto groups with *tert*-butylamine–borane complex in CH₂Cl₂ at reflux, followed by hydrolysis with NaOH in MeOH and purification by medium pressure liquid chromatography, afforded the desired 3α,7α,16β-trihydroxy-5β-cholan-24-oic acid sodium salt (**5**) and 3α,7α,16β-trihydroxy-6α-ethyl-5β-cholan-24-oic acid sodium salt (**7**) in 15 and 11% overall yield, respectively. Compounds **4–7** were unstable in the free acid form, spontaneously reacting to give the corresponding lactone derivatives. Therefore, the compounds were prepared and tested as sodium salts.

Avicholic acid and its derivatives **4–7** were evaluated, along with CDCA (**2**) and LCA as reference controls, for their ability to activate TGR5 and FXR using homogeneous time-resolved fluorescence (cell-based) assay and ALPHAscreen assay, respectively (Table 1). The results indicate that avicholic acid

Table 1. TGR5 and FXR Activities of Natural and Semisynthetic BAs^{a,6,15}

BA	TGR5 ^b		FXR ^c	
	EC ₅₀	efficacy ^d	EC ₅₀	efficacy ^d
CA (1)	13.6 ± 0.7	101	>100	0
CDCA (2)	6.7 ± 0.3	105	13.0 ± 0.7	62
INT-777 (3)	0.82 ± 0.04	166	>100	18
4	160 ± 8	194	>100	0
5	25 ± 1	148	55 ± 3	132
6	87 ± 4	198	22 ± 1	121
7	0.65 ± 0.03	120	11.5 ± 0.6	210

^aData represent mean values ± SDs of at least three independent experiments. ^bUnits are μM for EC₅₀ and % of 10 μM LCA (EC₅₀: 5 μM) value for efficacy. ^cUnits are μM for EC₅₀ % of 10 μM CDCA (2) value for efficacy. ^dEfficacy values are here used as a measure of fluorescence change.

(4) is a weak TGR5 agonist, with no activity at FXR. Interestingly, its C₁₆-epimer 5 shows an increased potency by 1 order of magnitude toward TGR5 accompanied by FXR activity, suggesting the preference of the hydroxyl group to form hydrogen bond interactions when placed on the equatorial position of the steroidal nucleus. This finding is also in agreement with the results of the QSAR model,¹⁸ indicating that the equatorial position of C₁₆ is favored for polar interactions with TGR5 (Figure 3).

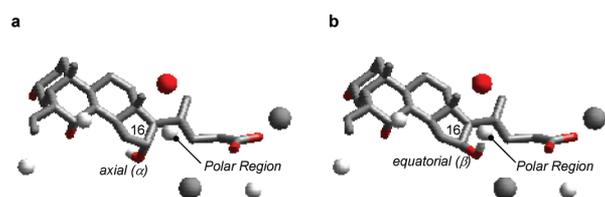


Figure 3. Alignment of 6 α -ethyl-avicholic acid (6, panel a) and 6 α -ethyl-16-*epi*-avicholic acid (7, panel b) according to the 3D QSAR scheme. The hydroxyl group of C₁₆ nicely fits the polar region of interaction when located on the equatorial position (β) of the steroidal nucleus rather than the axial (α) position.

The 6 α -ethyl derivatives 6 and 7 were synthesized to assess whether the insertion of this alkyl group at the 6 α position of the parent derivatives 4 and 5 would have improved the potency toward TGR5 and/or FXR, according to previously reported SARs.^{6,15} As shown in Table 1, in the case of 6 α -ethyl-16-*epi*-avicholic acid (7), the introduction of the alkyl group markedly increased the potency at TGR5 by about 2 orders of magnitude, while showing a slight improvement of FXR activity. In the case of 6 α -ethyl-avicholic acid (6), however, this effect is partially reversed, with enhanced agonistic activity toward FXR rather than TGR5.

Table 2 reports the physicochemical data of 6 α -ethyl-16-*epi*-avicholic acid (7) as compared to INT-777 (3) and natural BAs, including CA (1), CDCA (2), and avicholic acid (4). All of the studied BA presents the same acidity constant with a pK_a = 5 like all natural occurring BA,²³ and the hydroxyl group on the C₁₆ position does not affect the C₂₄ carboxylic ionization. In addition, it was found that the water solubility (Ws) of 6 α -ethyl-16-*epi*-avicholic acid (7) as a protonated acid is slightly higher than the trihydroxy BA INT-777 (3) and lower than the parent compound 4.

Table 2. Physicochemical Properties of BAs

BA	Ws ^a (μM)	CMC ^b (mM)	ST _{CMC} ^c (dyn/cm)	Log P _A ^d	albumin binding (%)
CA (1) ⁶	273	9.2	48	1.1	50
CDCA (2)	32	3.0	46	2.2	93
INT-777 (3)	99	2.0	50	1.4	62
4	188	10	50	1.1	58
7	120	5.9	53	1.6	83

^aWs: water solubility as protonated species in 0.1 M HCl water solution. ^bCMC: critical micellar concentration determined in 0.15 M NaCl water solution. ^cST_{CMC}: surface tension at CMC in 0.15 M NaCl water solution. ^dLog P_A (octanol/water partition coefficient of the ionized species): lipophilicity.

The detergent power of BA, that is, the tendency to form micelles, is the main concern for this detergent-like molecule.²⁴ The critical micellar concentration (CMC) and the surface tension at CMC (ST_{CMC}) values are indeed correlated with potential (cyto)toxicity of BAs. Molecules with high CMC and ST_{CMC} values are poor detergents and therefore poorly toxic to biological membranes. Furthermore, the lipophilicity (Log P_A) describes the passive diffusion of BAs across biological membranes, whereas the albumin binding reflects the amount of compound that is protein-bound in the blood. The CMC of 7 is higher than CDCA (2) and 3, while being lower than CA (1) and avicholic acid (4). This relatively high CMC value combines with a high ST_{CMC} value, suggesting a low micellar aggregation number and poor detergency for 7 that may favor low toxicity.

Among trihydroxy BAs, compound 7 shows the highest lipophilicity value (Log P_A). This is likely ascribed to the presence of the ethyl moiety in the C₆ α position that combines with the shift of the hydroxyl group from C₁₂ α of INT-777 (3) to C₁₆ β of 7. Interestingly, the shift of the hydroxyl from C₁₂ α of CA (1) to C₁₆ α of 4 does not play a role in affecting lipophilicity, as evidenced by the same Log P_A values reported in Table 2. The percent albumin binding of 7 is higher than CA (1), INT-777 (3), and avicholic acid (4). Again, this may be due to the presence of the 6 α -ethyl moiety and the C₁₆ β hydroxyl group. Interestingly, the value of albumin binding of 7 is compatible with a relatively fast hepatic uptake, similar to naturally occurring BAs such as CA (1).²⁵ The physicochemical properties of compound 7, including poor detergency and adequate hydrophilicity, are in agreement with previous structure–property relationship studies carried out on a large number of natural and synthetic BAs.²⁶ The 16 β orientation of the hydroxyl group, in particular, confers to the molecule a low detergency that, combined with the relatively high lipophilicity due to the presence of the 6 α ethyl group, may further facilitate the intestinal absorption by passive mechanism.

Compound 7 is fully conjugated by the mouse liver with taurine (unpublished data), in contrast with INT-777 (3) that is only partially conjugated with taurine after the C₂₃-methyl isomerization. Akin to naturally occurring taurine-conjugated bile salts,²⁷ the taurine-conjugated compound 7 may be additionally absorbed by the terminal ileum with an active mechanism. This would likely increase its bioavailability and modify its pharmacokinetics and biodistribution in comparison with INT-777 (3).

In conclusion, this work suggests that natural BAs can further provide additional lead compounds on the route to unravel physiological and therapeutic implications of TGR5 modu-

lation. Among these, herein, we have reported avicholic acid (**4**) as a selective TGR5 modulator, although showing weak potency. Chemical manipulations of **4** have led to the discovery of 6 α -ethyl-16-*epi*-avicholic acid (**7**) as a potent TGR5 agonist and moderately active FXR ligand, endowed with an interesting physicochemical profile. Further modifications of both the steroidal nucleus and side chain of **7** are currently being investigated with the aim of bestowing TGR5 selectivity over FXR and additional potency to this compound.

■ ASSOCIATED CONTENT

■ Supporting Information

Description of the synthetic procedures, biological methods, and analytical procedures of all target compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS

BA, bile acid; FXR, farnesoid X receptor; SAR, structure–activity relationship

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