

Visible-light-induced synthesis of a variety of trifluoromethylated alkenes from potassium vinyltrifluoroborates by photoredox catalysis†

Yusuke Yasu, Takashi Koike* and Munetaka Akita*

Cite this: *Chem. Commun.*, 2013, **49**, 2037

Received 28th December 2012,
Accepted 22nd January 2013

DOI: 10.1039/c3cc39235j

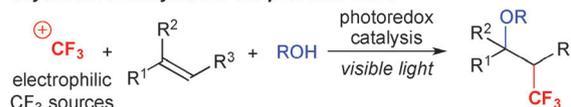
www.rsc.org/chemcomm

A facile synthesis of trifluoromethylated alkenes by the radical-mediated trifluoromethylation of vinyltrifluoroborates has been developed. Togni's reagent serves as a CF₃ radical precursor in the presence of the photoredox catalyst [Ru(bpy)₃](PF₆)₂ under visible light irradiation. This new photocatalytic protocol can be applicable to a wide variety of vinylborates containing electronically diverse substituents and hetero-aromatics.

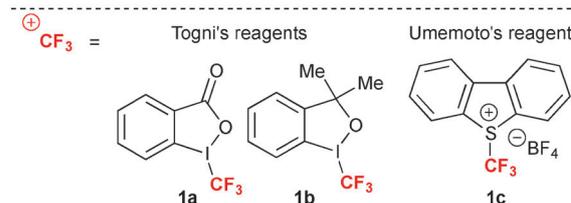
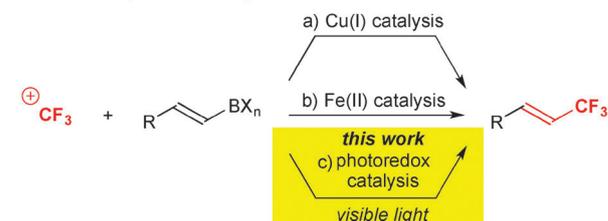
The trifluoromethyl (CF₃) group is a useful structural motif in many biologically active molecules as well as materials.^{1,2} Thus, the development of new methodologies for highly efficient and selective incorporation of a CF₃ group into diverse skeletons has attracted great interest of synthetic chemists.³ Over the past few years, a variety of transition-metal-catalyzed trifluoromethylations of aromatic molecules have been well developed.⁴ On the other hand, effective construction of C_{alkenyl}-CF₃ bonds is significantly more limited although trifluoromethylated alkenes are promising building blocks for pharmaceuticals, agrochemicals and functional materials.^{2a,3b,5-7}

The strategies for trifluoromethylation using electrophilic trifluoromethylating (⁺CF₃) reagents such as Togni's reagents **1a** (1-trifluoromethyl-1,2-benziodoxol-3-(1*H*)-one), **1b** (1-trifluoromethyl-1,3-dihydro-3,3-dimethyl-1,2-benziodoxole) and Umemoto's reagent **1c** (*S*-(trifluoromethyl)dibenzothiophenium tetrafluoroborate) have attracted great attention because the reagents are easy-to-handle chemicals in terms of being shelf-stable solid compounds at room temperature.⁸⁻¹⁰ Over the past few years, several catalytic synthetic methods for construction of C_{alkenyl}-CF₃ bonds with "⁺CF₃" reagents have been reported.¹¹ The groups of Shen and Liu have shown non-stereoselective trifluoromethylation of vinylboronic acids by Cu(I) catalysis (Scheme 1a).^{11a,b} Furthermore, Buchwald and co-workers reported an Fe(II)-catalyzed stereoselective trifluoromethylation

oxytrifluoromethylation: our previous work



trifluoromethylation of vinylboronic acid derivatives



Scheme 1 Photoredox-catalysed trifluoromethylation using electrophilic trifluoromethylating reagents (**1a–c**).

of vinyltrifluoroborates (Scheme 1b).^{11c} In addition, there are a few reports on copper-mediated trifluoromethylation of α,β -unsaturated carboxylic acids,^{11d} alkynes^{11e} and enamides.^{11f} These methods, however, exhibit limited substrate scopes with respect to hetero-aromatics and functional groups. Therefore, a new protocol for construction of C_{alkenyl}-CF₃ bonds with broad scope as well as in an efficient and stereoselective manner is highly desirable.

Recently, there have been several reports on radical trifluoromethylation by photoredox catalysis, which is regarded as a useful redox tool for organic compounds *via* single electron transfer (SET) in the presence of well-defined ruthenium(II) polypyridine complexes (e.g. [Ru(bpy)₃]²⁺) and the relevant cyclometalated iridium(III) derivatives under visible light irradiation,¹²⁻¹⁶ and conventional trifluoromethyl radical ([•]CF₃) sources such as CF₃SO₂Cl **1d** and gaseous CF₃I were used. These "[•]CF₃" precursors, however, require operational carefulness and special apparatus. In contrast, we have previously

Chemical Resources Laboratory, Tokyo Institute of Technology, 4259 Nagatsuta, Midori-ku, Yokohama 226-8503, Japan.

E-mail: koike.t.ad@m.titech.ac.jp, makita@res.titech.ac.jp; Fax: +81-45-924-5230; Tel: +81-45-924-5230

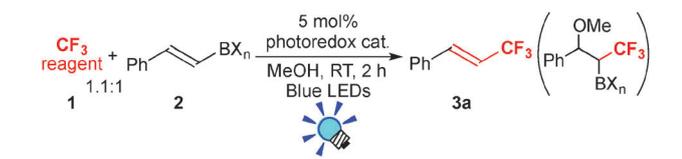
† Electronic supplementary information (ESI) available: Experimental details, spectral data, and crystallographic results. CCDC 900630. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c3cc39235j

reported the photoredox-catalyzed oxytrifluoromethylation of alkenes with “ ${}^+CF_3$ ” reagents based on usual laboratory techniques and equipment.^{14c} Importantly, easy-to-handle “ ${}^+CF_3$ ” reagents can serve as “ ${}^+CF_3$ ” precursors in the presence of photoredox catalysts under visible light irradiation.

Herein we disclose highly efficient and stereoselective deboronated trifluoromethylation of vinyltrifluoroborates with commercially available and air-stable Togni's reagent **1a** by visible-light-driven photoredox catalysis (Scheme 1c). In particular, this new photocatalytic protocol is effective in the selective and large-scale synthesis of (*E*)-trifluoromethylated alkenes bearing a π -electron-deficient hetero-aromatic moiety, which are difficult to access by previous catalytic systems.

We initially examined the photocatalytic trifluoromethylation of (*E*)-styrylboronic acid derivatives **2** with 1.1 equiv. of Togni's reagent **1a** as the CF_3 source in the presence of 5 mol% photoredox catalyst $Ir(ppy)_3$ ¹⁷ in MeOH under visible light irradiation (blue LEDs: $\lambda_{max} = 425$ nm) for 2 h. Every entry resulted in formation of β -trifluoromethylstyrene **3a** (Table 1, entries 1–3). As a result, the choice of the boron auxiliary turned out to be crucial for better yield and selectivity. The reaction of potassium (*E*)-styryltrifluoroborate **2a** afforded the product **3a** with an *E/Z* ratio of 94/6 in 85% yield (entry 3). In contrast, the substrates with $B(OH)_2$ and Bpin groups gave mixtures of **3a** and methoxytrifluoromethylated by-product. Next, examination of the CF_3 reagents revealed that **1a** is the best choice in terms of yield and selectivity. The use of **1b** and **1c** provided product **3a** in good yields, but did not improve the *E/Z* selectivity (entries 4 and 5). A typical CF_3 radical source **1d** gave a mixture of product **3a**, styrene and unidentified by-products (entry 6). Finally, the Ru photocatalyst $[Ru(bpy)_3](PF_6)_2$ also promoted the present reaction, providing the product **3a** with excellent *E/Z* selectivity (98/2) in 81% yield upon 5 h irradiation (entry 7). Notably, the product **3a** was not obtained at all either in the dark or in the absence of the photoredox catalyst (entries 8 and 9), strongly supporting that the photoexcited species of the photoredox catalyst play key roles in the reaction.

Table 1 Optimization of photocatalytic trifluoromethylation of (*E*)-styrylboronic acid derivatives **2**

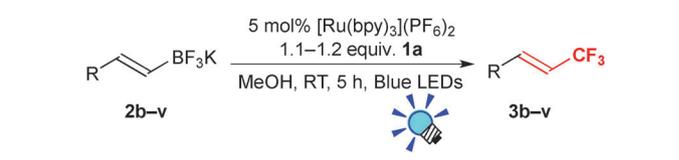


Entry	CF_3 reagent	BX_n	Photoredox catalyst	<i>E/Z</i> ^a	Yield ^a /%
1	1a	$B(OH)_2$	$Ir(ppy)_3$	57/43	53
2	1a	Bpin	$Ir(ppy)_3$	54/46	36
3	1a	$BF_3 \cdot K$ 2a	$Ir(ppy)_3$	94/6	85
4	1b	$BF_3 \cdot K$ 2a	$Ir(ppy)_3$	90/10	83
5	1c	$BF_3 \cdot K$ 2a	$Ir(ppy)_3$	85/15	79
6 ^b	CF_3SO_2Cl 1d	$BF_3 \cdot K$ 2a	$Ir(ppy)_3$	99/1	27
7 ^c	1a	$BF_3 \cdot K$ 2a	$[Ru(bpy)_3](PF_6)_2$	98/2	81
8	1a	$BF_3 \cdot K$ 2a	none	—	0
9 ^d	1a	$BF_3 \cdot K$ 2a	$[Ru(bpy)_3](PF_6)_2$	—	0

ppy = 2-phenylpyridine, bpy = 2,2'-bipyridine. For reaction conditions, see the ESI.^a Yields and *E/Z* ratios were determined by ¹H and ¹⁹F NMR spectroscopies. ^b CD_3CN was used as solvent. ^c Reaction time = 5 h. ^d In the dark.

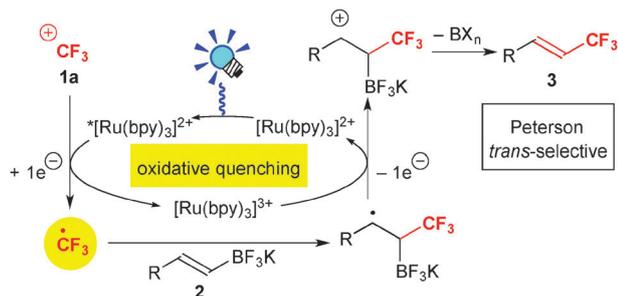
The scope and limitations of the present photocatalytic trifluoromethylation of vinyltrifluoroborates are summarized in Table 2. (*E*)-Styrylborates with electron-donating substituents, Me (**2b**, **1**) and MeO (**2c**), smoothly produced the corresponding β - CF_3 -styrene products with high *E/Z* selectivities in good yields (**3b**, **c**, **1**). In addition, this reaction can be applied to vinylborates bearing a variety of electron-withdrawing functional groups, F (**2i**), Cl (**2d**), Br (**2e**), CF_3 (**2f**, **j**), OCF_3 (**2g**), CN (**2h**), and NHBoc (**2k**). The corresponding trifluoromethylated alkene products (**3d–k**) were obtained without deterioration by the functional groups with excellent *E/Z* selectivities (>94/<6) in high yields (68–93%). This broad applicability of vinylborates is remarkable. The above-mentioned Buchwald's catalytic system is ineffective for electron-deficient substrates.^{11c}

Table 2 The scope of the present trifluoromethylation of vinylborates^{a,b}



3b : 81% <i>E/Z</i> = 92/8	3c : 88% <i>E/Z</i> = 99/1	3d : 72% <i>E/Z</i> = 94/6
3e : 81% <i>E/Z</i> = 95/5	3f : 79% ^{c,d} <i>E/Z</i> = 98/2	3g : 69% <i>E/Z</i> = 95/5
3h : 78% <i>E/Z</i> = 96/4	3i : 68% <i>E/Z</i> = 96/4	3j : 81% ^{c,d} <i>E/Z</i> = 97/3
3k : 93% <i>E/Z</i> = 94/6	3l : 70% <i>E/Z</i> = 95/5	3m : 63% ^{e,f} <i>E/Z</i> = 96/4
3n : 68% <i>E/Z</i> = 97/3	3o : 56% <i>E/Z</i> = 98/2	3p : 66% <i>E/Z</i> = 98/2
3q : 70% <i>E/Z</i> = 98/2	3r : 69% <i>E/Z</i> = 97/3	3s : 71% <i>E/Z</i> = 98/2
3t : 82% <i>E/Z</i> = 97/3	3u : 60% <i>E/Z</i> = 63/37	3v : 79% ^g <i>E/Z</i> = 73:27

For reaction conditions, see the ESI. ^a Isolated yields. ^b *E/Z* ratios were determined by ¹H and ¹⁹F NMR spectroscopies of the crude product mixtures. ^c Photocatalyst $Ir(ppy)_3$ (0.5 mol%) was used. ^d Reaction time = 2 h. ^e 1.1 equiv. of vinylborate with respect to **1** was used. ^f Protodeboronated by-product was obtained in 2% yield. ^g NMR yield.



Scheme 2 A plausible reaction mechanism.

Next, vinylborates containing heteroaryl groups such as thiophene, pyridine, quinoline and benzothiazole were examined. Hetero-aromatic substrates are relevant to pharmaceuticals and agrochemicals because hetero-arenes are ubiquitous structures in biologically active compounds.¹⁸ Remarkably, the photocatalytic reactions of **2m–t** gave the corresponding CF₃-substituted alkenes **3m–t** with excellent *E/Z* selectivities (>96/<4) in good yields (56–82%). The catalytic formation of C_{alkenyl}-CF₃ bonds connected with π -electron-deficient hetero-aromatics has not been so far reported,¹¹ thus the present photocatalytic protocol might be attractive from the viewpoint of regio- and stereoselective introduction of a CF₃ group into these scaffolds. Furthermore, alkylvinylborates (**2u** and **v**) can also be applicable to this photocatalytic system with moderate selectivities and good yields (**3u** and **v**).

As a demonstration of scalability of this transformation, the trifluoromethylation of **2s** was carried out on a gram scale. As a result, the trifluoromethylated alkene with quinoline moiety **3s**, which is a herbicide-related molecule,⁷ was obtained in good yield (64%, 1.06 g) with excellent *E/Z* selectivity (>99/<1) (see the ESI[†]).

A plausible reaction mechanism based on a SET photoredox process (oxidative quenching) is shown in Scheme 2. According to our previous report,^{14c,19} •CF₃ can be generated from 1e-reduction of electrophilic Togni's reagent **1a**. •CF₃ reacts with a C=C bond of vinylborates **2** to give β -borato-stabilized radical intermediates²⁰ in a regioselective manner. Subsequent 1e-oxidation by the high-oxidation state photocatalyst [Ru(bpy)₃]³⁺ produces β -borato cation intermediates. Finally predominant Peterson elimination of the boron-based group with *trans*-selectivity provides (*E*)-trifluoromethylated alkenes **3**.²¹ However we could not rule out the possibility that a mechanism including radical chain propagation occurs.

In conclusion, we have developed a new and facile visible-light-induced synthesis of a variety of trifluoromethylated alkenes using a Ru photoredox catalyst. Togni's reagent **1a** plays an important role as a CF₃ radical source. This photocatalytic system allows easy access to (*E*)-trifluoromethylated alkenes bearing electron-deficient moieties such as π -electron-deficient hetero-aromatics. We expect this protocol to be of broad utility in the synthesis of biologically active organofluorine molecules.

The financial support from the Japanese government (Grants-in-Aid for Scientific Research: No. 23750174) is gratefully acknowledged.

Notes and references

1 T. Hiyama, *Organofluorine Compounds: Chemistry and Applications*, Springer, Berlin, 2000.

- (a) Special issue on "Fluorine in the Life Sciences", *ChemBioChem*, 2004, **5**, 570; (b) K. Müller, C. Faeh and F. Diederich, *Science*, 2007, **317**, 1881; (c) *Fluorine in Medicinal Chemistry and Chemical Biology*, ed. I. Ojima, Wiley-Blackwell, Chichester, 2009.
- For recent reviews on synthesis of CF₃-containing molecules, see: (a) J.-A. Ma and D. Cahard, *Chem. Rev.*, 2004, **104**, 6119; (b) M. Shimizu and T. Hiyama, *Angew. Chem., Int. Ed.*, 2005, **44**, 214; (c) J.-A. Ma and D. Cahard, *J. Fluorine Chem.*, 2007, **128**, 975; (d) N. Shibata, S. Mizuta and T. Toru, *J. Synth. Org. Chem. Jpn.*, 2008, **66**, 215; (e) J.-A. Ma and D. Cahard, *Chem. Rev.*, 2008, **108**, PR1; (f) T. Furuya, A. S. Kamlet and T. Ritter, *Nature*, 2011, **473**, 470.
- For recent reviews on synthesis of CF₃-containing aromatic compounds, see: (a) I. Katsuyama, *J. Synth. Org. Chem. Jpn.*, 2009, **67**, 992; (b) H. Amii, *J. Synth. Org. Chem. Jpn.*, 2011, **69**, 752; (c) O. A. Tomashenko and V. V. Grushin, *Chem. Rev.*, 2011, **111**, 4475; (d) T. Besset, C. Schneider and D. Cahard, *Angew. Chem., Int. Ed.*, 2012, **51**, 5048.
- A. Rivkin, T.-C. Chou and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2005, **44**, 2838.
- (a) M. Shimizu, Y. Takeda, M. Higashi and T. Hiyama, *Angew. Chem., Int. Ed.*, 2009, **48**, 3653; (b) M. Shimizu, Y. Takeda, M. Higashi and T. Hiyama, *Chem.-Asian J.*, 2011, **6**, 2536; (c) Z. Shi, J. Davies, S.-H. Jang, W. Kaminsky and A. K.-Y. Jen, *Chem. Commun.*, 2012, **48**, 7880.
- T. P. Selby, *Preparation of substituted fused heterocyclic herbicides*, PCT International Patent Application US5389600 (A), 1995.
- (a) P. Eisenberger, S. Gischig and A. Togni, *Chem.-Eur. J.*, 2006, **12**, 2579; (b) I. Kietlsch, P. Eisenberger and A. Togni, *Angew. Chem., Int. Ed.*, 2007, **46**, 754.
- T. Umemoto, *Chem. Rev.*, 1996, **96**, 1757, and references therein.
- (a) Y. Macé and E. Magnier, *Eur. J. Org. Chem.*, 2012, 2479; (b) N. Shibata, A. Matsnev and D. Cahard, *Beilstein J. Org. Chem.*, DOI: 10.3762/bjoc.6.65.
- (a) J. Xu, D.-F. Luo, B. Xiao, Z.-J. Liu, T.-J. Gong, Y. Fu and L. Liu, *Chem. Commun.*, 2011, **47**, 4300; (b) T. Liu and Q. Shen, *Org. Lett.*, 2011, **13**, 2342; (c) A. T. Parsons, T. D. Senecal and S. L. Buchwald, *Angew. Chem., Int. Ed.*, 2012, **51**, 2947; (d) Z. He, T. Luo, M. Hu, Y. Cao and J. Hu, *Angew. Chem., Int. Ed.*, 2012, **51**, 3944; (e) P. G. Janson, I. Ghoneim, N. O. Iichenko and K. J. Szabó, *Org. Lett.*, 2012, **14**, 2882; (f) C. Feng and T.-P. Lou, *Chem. Sci.*, 2012, **3**, 3458.
- D. A. Nicewicz and D. W. C. MacMillan, *Science*, 2008, **322**, 77.
- For recent reviews on photoredox catalysis, see: (a) T. P. Yoon, M. A. Ischay and J. Du, *Nat. Chem.*, 2010, **2**, 527; (b) J. M. R. Narayanan and C. R. J. Stephenson, *Chem. Soc. Rev.*, 2011, **40**, 102; (c) F. Teplý, *Collect. Czech. Chem. Commun.*, 2011, **76**, 859; (d) J. W. Tucker and C. R. J. Stephenson, *J. Org. Chem.*, 2012, **77**, 1617; (e) J. Xuan and W.-J. Xiao, *Angew. Chem., Int. Ed.*, 2012, **51**, 6828; (f) S. Maity and N. Zheng, *Synlett*, 2012, 1851.
- (a) T. Koike and M. Akita, *Chem. Lett.*, 2009, 166; (b) Y. Yasu, T. Koike and M. Akita, *Chem. Commun.*, 2012, **48**, 5355; (c) Y. Yasu, T. Koike and M. Akita, *Angew. Chem., Int. Ed.*, 2012, **51**, 9567; (d) T. Koike, Y. Yasu and M. Akita, *Chem. Lett.*, 2012, 999; (e) Y. Yasu, T. Koike and M. Akita, *Adv. Synth. Catal.*, 2012, **354**, 3414.
- (a) D. A. Nagib and D. W. C. MacMillan, *Nature*, 2011, **480**, 224; (b) J. D. Nguyen, J. W. Tucker, M. D. Konieczynska and C. R. J. Stephenson, *J. Am. Chem. Soc.*, 2011, **133**, 4160; (c) N. Iqbal, S. Choi, E. Ko and E. J. Cho, *Tetrahedron Lett.*, 2012, **53**, 2005; (d) Y. Ye and M. S. Sanford, *J. Am. Chem. Soc.*, 2012, **134**, 9034; (e) C.-J. Wallentin, J. D. Nguyen, P. Finkbeiner and C. R. J. Stephenson, *J. Am. Chem. Soc.*, 2012, **134**, 8875; (f) N. Iqbal, S. Choi, E. Kim and E. J. Cho, *J. Org. Chem.*, 2012, **77**, 11383.
- A. Studer, *Angew. Chem., Int. Ed.*, 2012, **51**, 8950, and references therein.
- The photoexcited state of Ir(ppy)₃ is a strong reductant (*E*_{1/2} = -1.32 V vs. Cp₂Fe), see: L. Flamigni, A. Barbieri, C. Sabatini, B. Ventura and F. Barigelletti, *Top. Curr. Chem.*, 2007, **281**, 143.
- <http://www.drugs.com/top200.html>.
- Cyclic voltammograms of vinylborate **2a** showed a broad irreversible oxidation wave around *E*_{1/2} = +1.01 V vs. Cp₂Fe. We excluded generation of radicals from oxidation of vinylborates by the photoexcited state of [Ru(bpy)₃]²⁺ (*E*_{1/2} = +0.43 V vs. Cp₂Fe).
- H. Kim and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2008, **130**, 398.
- Elimination of the boron group via a β -borato-radical intermediate is a highly endothermic pathway and thereby presumed unlikely; see: J. C. Walton, A. J. McCarroll, Q. Chen, B. Carboni and R. Nziengui, *J. Am. Chem. Soc.*, 2000, **122**, 5455.