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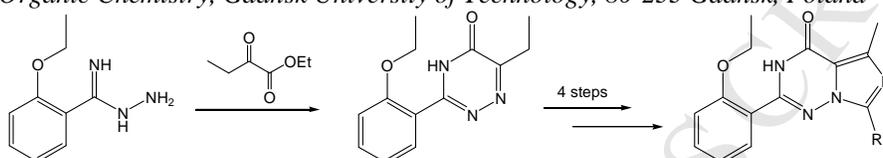
Graphical Abstract

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Alternative method for the synthesis of imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one – a substrate for the preparation of phosphodiesterase(5) inhibitors

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ABSTRACT

Imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-ones, as isosteres of purine, are of interest for pharmaceutical research as potential substrates for the synthesis of cGMP-PDE5 inhibitors. We present a novel, alternative method for the synthesis of imidazotriazinones, that differs from the previously reported ones with respect to the **method of construction of the triazinone ring in the molecule**. The key step in our approach is condensation of an appropriate α -keto-ester with amidrazones, leading to the triazinone heterocycle. Several different substituted imidazolotriazinones have been synthesized in this manner.

1. Introduction

Phosphodiesterases (PDEs) are a large family of enzymes which are responsible for breaking the phosphodiester bonds in biological molecules and some of them are involved in regulation of physiological functions.^{1,2} Some of the PDEs are drug targets for the treatment of various diseases, including: heart failure, depression, asthma, inflammation and erectile dysfunction.³⁻⁵ In particular, phosphodiesterase 5 (PDE5), which is involved in the hydrolysis of a secondary messenger, cyclic guanosine monophosphate (cGMP), present in the corpus cavernosum tissue, plays an important role in mediating the sexual response.⁶⁻⁸ Inhibition of PDE5 increases the cGMP level, triggering erection *via* relaxation of the penile arterioles.⁹ Selective inhibitors of PDE5 have a great clinical significance in treatments of the erectile dysfunction disease and their other therapeutic applications are being proposed and investigated.¹⁰⁻¹¹ There are three commercially available drugs acting as PDE5 inhibitors, namely: sildenafil citrate (the active ingredient in Viagra), vardenafil (Levitra) and tadalafil.¹² A core structure in vardenafil **2**, which has been approved by the FDA and launched in 2003, is imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one – general structure **1** (Fig. 1).

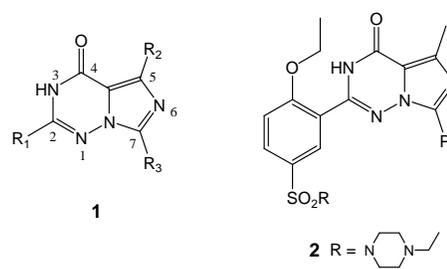
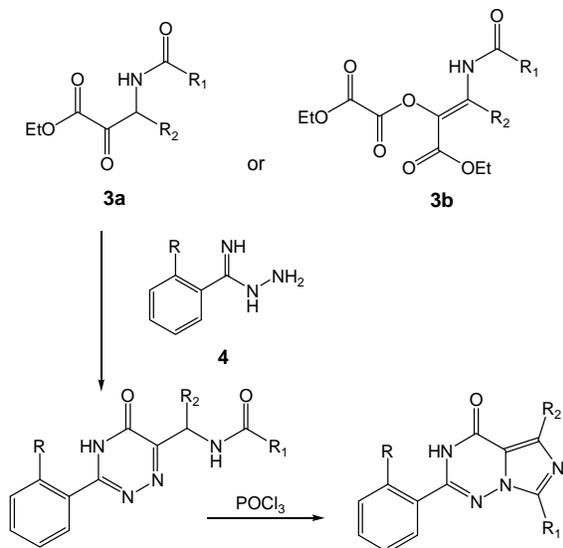


Figure 1. Chemical structure of imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one **1** and vardenafil **2**

Several method for the synthesis of imidazotriazinones have been reported so far.¹³⁻¹⁷ They can be divided into two major groups depending on the sequence of steps of ring construction. These in which the triazinone ring is built at the beginning are generally based on the method described by Charles *et al.*,¹⁷ where the ring is formed by condensation of an acyloamino- α -keto-ester **3a** or enol ester **3b** with an benzamidrazone **4** or generally amidrazone, as shown in Scheme 1

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2. Results and discussion

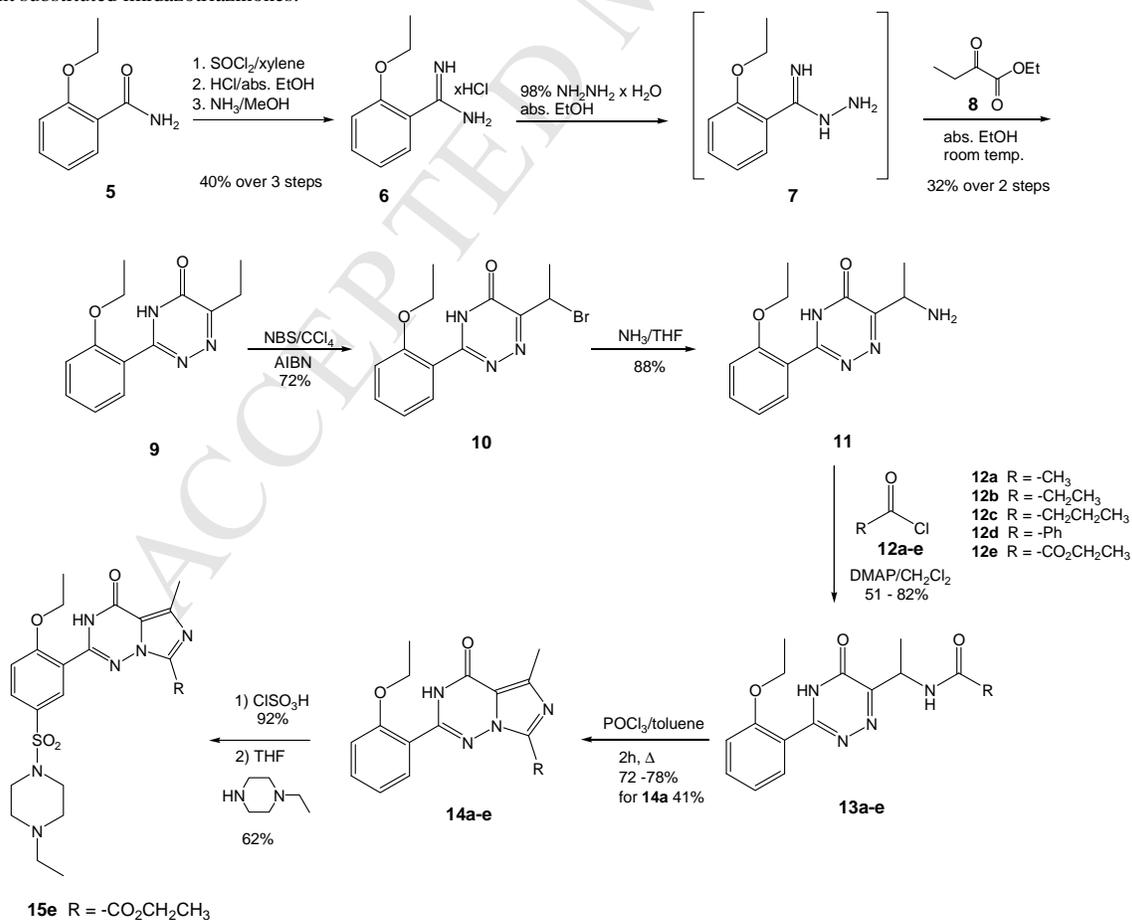


Scheme 1. Synthesis of imidazo[5,1-f][1,2,4]triazin-4(3H)-ones¹⁷

The main drawback in this approach is the availability of the active intermediate **3a** or **3b**, both of which are obtained from α -amino acids and ethyl oxalate *via* the Dakin-West reaction.^{18,19} It is well known that these compounds are very capricious and cannot be obtained with purity greater than 50%. This limitation led us to investigate a novel route for the construction of the imidazo[5,1-f][1,2,4]triazin-4(3H)-one core which does not require the reactive intermediate **3a** or **3b** and gives rise to the possibility of synthesis of different substituted imidazotriazinones.

In the proposed strategy, a triazinone ring is formed by condensation of benzamidrazone **7** with the stable 2-oxo-butyrac acid ethyl ester **8**. The sequence of reactions leading to the eventual formation of the imidazotriazinone is shown in Scheme 2.

The substituted benzamidrazone **7**, an essential intermediate for the construction of the triazinone ring, was prepared in four steps. The commercially available 2-ethoxybenzamide **5** was used as a starting substrate which, upon treatment with thionyl chloride in xylene was dehydrated to the 2-ethoxybenzimidrazone in 90% yield. This intermediate was then converted to the benzamidine hydrochloride **6** by subsequent reactions with gaseous dry hydrogen chloride and ammonia in absolute alcohols. Alternatively, compound **6** can be obtained *via* reaction between 2-ethoxybenzimidrazone and $\text{AlMe}_2\text{CNH}_2$,²⁰ prepared from AlMe_3 and NH_4Cl in one step or in reaction with lithium hexamethyldisilazane (LHMDS) in THF at room temperature, as described by Mao and co-workers.²¹ Hydrazinolysis of benzamidine hydrochloride **6** to the intermediate **7** was performed with hydrazine hydrate in ethanol at 0-5 °C. Benzamidrazone **7** is an unstable compound but it can be isolated as a picrate salt in 56% yield. The key step of the whole synthesis is the condensation of benzamidrazone **7** with 2-oxo-butyrac acid ethyl ester **8**, leading to the formation of a triazinone ring in compound **9**. Due to the high reactivity of benzamidrazone **7**, several products may be formed at this stage. The highest yield of **7** was achieved when a freshly prepared benzamidrazone was used without purification. Anhydrous ethanol was found to be the optimal solvent in this reaction; the yield of condensation was poorer when the reaction was run in methanol. The imidazole ring was then built up on the imidazotriazinone molecule *via* bromination at the benzylic position, followed by amination with ammonia, acylation, and a final dehydration-cyclisation in the presence of POCl_3 .



Scheme 2.

Bromination of the benzylic position in the compound **9** was performed using NBS as a source of bromine. Treatment of the resulting intermediate **10** with 14% ammonia in THF afforded **11** in an unequivocal way. When MeOH was used as a solvent in this reaction, a methoxy derivative was formed together with **11**. The amine derivative **11** was treated with acid chlorides **12a-e** in dichloromethane at 0 °C, in the presence of a catalytic amount of DMAP. The reaction afforded amides **13a-e** in a high yields, 80-90%. Cyclisation of compounds **13a-e** with phosphoryl chloride, either in methylene chloride or in toluene, gave final products, compounds **14a-e**, containing the required imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one core (compound **14c** is identical with the vardenafil core), of high purity and with overall good yield. Structure of all final products was confirmed by MS and NMR analysis. One of them, namely **14b** was crystallized as large crystals and its 3D structure was determined by X-ray diffraction. The structure is shown in Fig. 2. X-ray diffraction analysis of the **14b** revealed that this compound crystallizes in the *P*-1 space group. The molecule is essentially planar. Both the imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one ring and the phenyl ring lie in one plane with maximum deviations of 0.035 Å. In addition, a weak intramolecular N-H...O hydrogen bonding between the O1 atom of the ethoxy group in the phenyl ring and N3-H3 in the imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one moiety is observed.

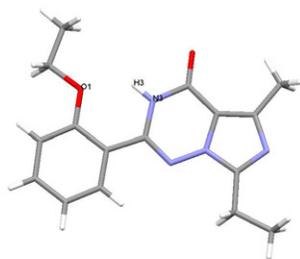
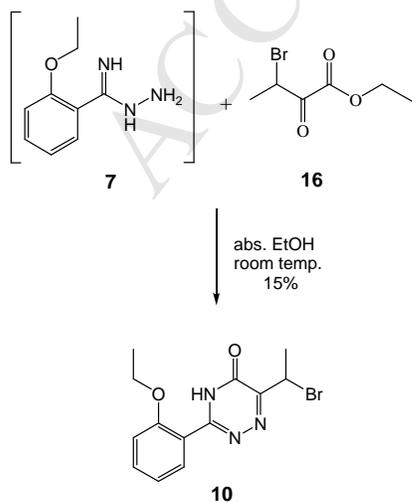


Figure 2. Licorice representation of the X-ray structure of **14b**.

Because our goal was to access imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-one core from stable and easily accessible substrates, apart from using 2-oxo-butyric acid ethyl ester **8** as a starting material in our synthetic route (Scheme 2), we also tested the possibility of getting compound **10** by condensation of benzamidrazone **7** with 3-bromo-2-oxo-butyric acid ethyl ester **16** which was obtained in the reaction of ethyl 2-oxobutanoate **8** with CuBr₂ in yields as high as 87% after purification.

**Scheme 3.**

The cyclization carried out in ethanol led to the required product **10**. However, presumably due to the higher reactivity of 3-bromo-2-oxo-butyric acid ethyl ester **16** compared to 2-oxo-butyric acid ethyl ester **8**, the yield was moderately low.

The synthesized compounds **14a-d** can be easily transformed to vardenafil and its analogues by two subsequent reaction, namely chlorosulfonation and sulfonamide formation, conditions of which are described in the literature.²² Only derivative **14e**, which is to our knowledge a new compound, has not been transformed in this way. Even if this transformation was not aim of our work, we decided to check whether the reaction condition could affect the substituent ester function at the 7-position of imidazotriazinone **14e**. As shown in Scheme 2, chlorosulfonation of the compound **14e** in chlorosulfonic acid at 0 °C and following after amination with *N*-ethylpiperazine proceeded smoothly and selectively at the 5'-position of the phenyl ring to afford the target product **15e**. As a new analogue of vardenafil, compound **15e** is worth testing for its pharmacological properties.

3. Conclusion

In conclusion, we have developed a new, alternative method for the synthesis of imidazotriazinones – substrates for the synthesis of potential PDE5 inhibitors. The main advantage of the novel strategy is the use of the more stable substrate in the key condensation, which results in an excellent reproducibility of the reaction at this stage and higher yields. Following our procedure, five differently substituted imidazotriazinones were synthesized and this synthesis demonstrates the potential of the novel method for preparation of a wide range of compounds based on the imidazo[5,1-*f*][1,2,4]triazin-4(3*H*)-ones ring system. The method was applied for a formal synthesis of vardenafil substrate and could be alternative for scale-up preparation of vardenafil.

4. Experimental**4.1. General**

All solvents and reagents were used as obtained from commercial source. ¹H NMR and ¹³C NMR spectra were obtained at 500 and 125 MHz (Varian Unity Plus), respectively, and the deuterated solvents were used as internal lock. Infrared spectra were recorded on a FT-IR Bruker IFS 66 instrument. Band positions are reported in reciprocal centimeters (cm⁻¹). Melting points were determined on a melting point apparatus equipped with thermometer and were uncorrected. Column chromatography was carried out in silica gel 0.040-0.063 mm. The mass spectra analyses were carried out using the MALDI-TOF Bruker BiFlex III mass spectrometer. Elemental analyses were recorded on a Perkin Elmer 240C Elemental Analyzer.

4.2. 2-Ethoxybenzamidine hydrochloride 6

2-Ethoxybenzimidine was prepared from 2-ethoxybenzamide **5** according to the method reported by Nowakowski²³ in 89% yield (lit.²³ yield 92%); R_f (hexane/EtOAc 9:1) 0.45; v_{max}(crystal): 2231cm⁻¹ (C≡N). ¹H NMR (CDCl₃, δ ppm): 7.57-7.51 (m, 2H, 4-H and 5-H of C₆H₄), 7.01(d, *J* 7.3 Hz, 1H, 6-H of C₆H₄), 6.97 (d, *J* 8.8 Hz, 1H, 3-H of C₆H₄), 4.17(q, *J* 7 Hz, 2H, OCH₂), 1.49 (t, *J* 7.1 Hz, 3H, CH₃). Ethyl 2-ethoxybenzimidate hydrochloride was synthesized by passing dry HCl_(gas) through the solution of 2-ethoxybenzimidine in anhydrous ethyl alcohol. This compound was obtained as a white solid (46%); m.p. 49-54 °C; R_f (hexane/EtOAc 7:3) 0.36. ¹H NMR (CDCl₃, δ ppm): 13.01 (bs, 1H, NH), 10.14 (bs, 1H, NH), 8.04 (dd, *J*

1.7 and 8 Hz, 1H, 6-H of C₆H₄), 7.74-7.70 (m, 1H, 4-H of C₆H₄), 7.18-7.14 (m, 2H, 3-H and 5-H of C₆H₄), 5.04 (q, *J* 7 Hz, 2H, NCOCH₂), 4.39 (q, *J* 7 Hz, 2H, OCOCH₂), 1.63 (t, *J* 7.1 Hz, 3H, NCOCH₂CH₃), 1.57 (t, *J* 6.8 Hz, 3H, OCOCH₂CH₃). 2-Ethoxybenzimidine hydrochloride **6** was prepared from ethyl 2-ethoxybenzimidate hydrochloride and NH₃/methanol. Crude product was purified by crystallization (methanol/ethyl ether) to provide **6** as a white solid crystal (96%); m.p. 190-192 °C; (lit.²⁴ yield 91%, mp 195-196 °C); R_f (n-butanol/acetic acid/water 4:4:1) 0.31; ¹H NMR (CDCl₃+DMSO-*d*₆, δ ppm): 9.76 (bs, 2H, NH), 8.41 (bs, 2H, NH), 7.80 (d, *J* 7.8 Hz, 1H, 6-H of C₆H₄), 7.55-7.51 (m, 1H, 4-H of C₆H₄), 7.06 (t, *J* 7.6 Hz, 1H, 5-H of C₆H₄), 7.0 (d, *J* 8.3 Hz, 1H, 3-H of C₆H₄), 6.69 (bs, 2H, NH₂), 4.15 (q, *J* 7 Hz, 2H, OCH₂), 1.43 (t, *J* 7.1 Hz, 3H, CH₃). ν_{max} (crystal): 1662 cm⁻¹ (C=N). MS (ESI, matrix DHB): [M+H]⁺ found: 165.1. C₉H₁₂N₂O requires 165.095.

4.3. Ethyl 2-oxobutanoate **8**

Ethyl 2-oxobutanoate **8** was prepared from 2-ketobutyric acid (25 g, 0.245 mol), *p*-toluenesulfonic acid (0.5 g), absolute ethanol (140 mL) and toluene (80 mL) using a Dean and Stark apparatus. The reaction proceeded for 48 h at reflux. After the standard work-up purification of the product was performed by distillation under reduced pressure (water pump). The ethyl 2-oxobutanoate **8** was collected as a fraction boiling at 70-75 °C (20 mm Hg), yield 65% (lit.²⁵ bp₁₈ 65 °C). ¹H NMR (CDCl₃, δ ppm): 4.33 (q, *J* 7.2 Hz, 2H, OCOCH₂), 2.88 (q, *J* 7.2 Hz, 2H, OC-CH₂), 1.37 (t, *J* 7.1 Hz, 3H, OCOCH₂CH₃), 1.14 (t, *J* 7.3 Hz, 3H, OC-CH₂CH₃).

4.4. 3-(2-Ethoxyphenyl)-6-ethyl-4H-[1,2,4]triazin-5-one **9**

To an ice-cold, stirred suspension of amidine hydrochloride **6** (2.1 g, 0.01 mol) in absolute ethanol (10 mL), the 98% solution of hydrazine hydrate (0.51 mL, 0.01 mol) was added. The mixture was allowed to stand at 0 to 5 °C overnight and then the precipitated ammonium chloride was filtered off. Ethyl 2-oxobutanoate **8** (1.3 g, 0.01 mol) dissolved in anhydrous methanol (10 mL) was added to the filtrate, resulting in an immediate formation of a white solid. The resulting mixture was allowed to stand at room temperature for 8 h. The solid (inorganic salt) was then filtered off and the filtrate was concentrated under reduced pressure. Dissolution of the residue in chloroform (10 mL) led to the precipitation of another portion of inorganic salt which was filtered off. The residue was concentrated and purified by chromatography on a silica gel column eluted with 30-50% (v/v) ethyl acetate in petroleum ether to give the triazine **9** as a pale yellow oil. This oil was crystallised from ethyl acetate/hexane to give the *title product* **9** (0.85 g, 32%) as a white solid; m.p. 114-116 °C; [Found: C, 63.55; H, 6.19; N, 17.19. C₁₃H₁₅N₃O₂ requires C, 63.66; H, 6.16; N, 17.13%]; R_f (hexane/EtOAc, 1:1) 0.28; ¹H NMR (CDCl₃, δ ppm): 12.10 (brs, 1H, NH...O), 8.57 (dd, *J* 1.7 and 8.1 Hz, 1H, 6-H of C₆H₄), 7.53 (t, *J* 8.8 Hz, 1H, 4-H of C₆H₄), 7.13 (t, *J* 7.6 Hz, 1H, 5-H of C₆H₄), 7.04 (d, *J* 8.3 Hz, 1H, 3-H of C₆H₄), 4.34 (q, *J* 7 Hz, 2H, OCOCH₂), 2.80 (q, *J* 7.5 Hz, 2H, C-CH₂), 1.60 (t, *J* 7.1 Hz, 3H, CH₃CH₂O), 1.26 (t, *J* 7.3 Hz, 3H, CH₃CH₂C). ¹³C NMR (CDCl₃, δ ppm): 160.7, 158.2, 158.1, 135.3, 132.2, 122.4, 116.6, 112.8, 65.7, 19.3, 15.0, 9.3.

4.5. 6-(1-Bromoethyl)-3-(2-ethoxyphenyl)-4H-[1,2,4]triazin-5-one **10**

To a magnetically stirred solution of 3-(2-ethoxyphenyl)-6-ethyl-4H-[1,2,4]triazin-5-one (**9**) (3.36 g, 0.0137 mol) in anhydrous carbon tetrachloride (350 mL) under an argon atmosphere, *N*-bromosuccinimide (2.7 g, 0.015 mol) and 2,2'-azo-bis-isobutyronitrile (174 mg) were added. The reaction mixture was stirred under reflux for 14 h and then cooled to room temperature. The solvent was evaporated *in vacuo* and the residue was treated with chloroform (200 mL) and water (150 mL). The mixture was shaken vigorously

until complete dissolution of all solid. The chloroform layer was separated, washed with water, dried (MgSO₄) and filtered. The solvent was evaporated. The crude product was purified by column chromatography on silica gel (hexane:ethyl acetate 1:1 v/v). The final product **10** was crystallised from ethyl acetate as a light yellow solid (3.19 g, 72%); m.p. 116-121 °C; [Found: C, 48.31; H, 4.38; N, 12.91. C₁₃H₁₄BrN₃O₂ requires C, 48.17; H, 4.35; N, 12.96%]; R_f (hexane/EtOAc, 1:1) 0.5. ¹H NMR (CDCl₃, δ ppm): 8.56 (d, *J* 8.3 Hz, 1H, 6-H of C₆H₄), 7.55 (t, *J* 7.8 Hz, 1H, 4-H of C₆H₄), 7.15 (t, *J* 7.6 Hz, 1H, 5-H of C₆H₄), 7.06 (d, *J* 8.3 Hz, 1H, 3-H of C₆H₄), 5.47 (q, *J* 6.8 Hz, 2H, CH₂Br), 4.36 (q, *J* 7 Hz, 2H, CH₂O), 2.01 (d, *J* 6.8 Hz, 3H, CH₃CH), 1.61 (t, *J* 6.8 Hz, 3H, CH₃CH₂). ¹³C NMR (CDCl₃, δ ppm): 162.2, 158.1, 157.9, 154.6, 135.2, 132.1, 122.2, 116.6, 112.8, 65.7, 44.7, 17.1, 15.0.

4.6. General procedure for the synthesis of the *N*-[1-[3-(2-ethoxyphenyl)-5-oxo-4,5-dihydro[1,2,4]triazin-6-yl]ethyl]-alkanoamide **13**

The bromoderivative **10** (0.6 g, 0.00185 mol) placed in a round-bottomed flask was treated with NH₃/THF solution (60 mL). The reaction flask was closed tightly and allowed to stand at room temperature with occasional shaking until all substrate **10** was consumed (TLC control). The solvent was removed by evaporation and the oily residue was dissolved in dichloromethane (30 mL), washed twice with water (10 mL), dried (MgSO₄) and evaporated to dryness under reduced pressure. The crude amino product **11** (0.42 g, 88%) was used without further purification in the next step. To the stirred, ice-cooled solution of the crude amino compound **11** (0.43 g, 0.00165 mol) in dry dichloromethane (20 mL), protected from moisture by a calcium chloride drying tube, triethylamine (0.46 mL, 0.0033 mol) was added. The mixture was stirred for 5 minutes and the acyl chloride **12a-e** (0.0018 mol) was added dropwise. After 10 minutes the cooling bath was removed and the reaction mixture was allowed to stand for 1 h at room temperature. The reaction was quenched by the addition of water (10 mL). The organic layer was washed with water, brine, and dried (MgSO₄). The solvent was removed *in vacuo* and the crude product **13** was purified by two consecutive silica gel column chromatography separations. In the first one, an ethyl acetate/petroleum ether mixture was used as an eluting solvent and the second column was developed with the ethyl acetate/methanol mixture 6:0.5 (v/v).

4.6.1 *N*-[1-[3-(2-Ethoxy-phenyl)-5-oxo-4,5-dihydro-[1,2,4]triazin-6-yl]ethyl]acetamide **13a.** The *title product* **13a** was obtained as a light yellow oil (0.29 g, 51%); [Found: C, 59.65; H, 6.02; N, 18.47. C₁₅H₁₈N₄O₃ requires C, 59.59; H, 6.0; N, 18.53%]; R_f (EtOAc/EtOH, 24:0.5) 0.25. ¹H NMR (CDCl₃, δ ppm): 12.12 (brs, 1H, NH...O), 8.57 (d, *J* 7.8 Hz, 1H, 6-H of C₆H₄), 7.56 (t, *J* 7.8 Hz, 1H, 4-H of C₆H₄), 7.18 (t, *J* 7.3 Hz, 1H, 5-H of C₆H₄), 7.08 (d, *J* 8.3 Hz, 1H, 3-H of C₆H₄), 6.89 (brd, *J* 8.8 Hz, 1H, NH), 5.25-5.22 (m, 1H, CHN), 4.35 (q, *J* 6.8 Hz, 2H, CH₂O), 2.0 (s, 3H, CH₃CO), 1.61 (t, *J* 7.1 Hz, 3H, CH₃CH₂O), 1.54 (d, *J* 7.3 Hz, 3H, CH₃CH). ¹³C NMR (CDCl₃, δ ppm): 169.7, 158.2, 158.1, 135.3, 132.2, 122.4, 116.5, 112.9, 65.7, 47.9, 23.7, 19.3, 15.0.

4.6.2 *N*-[1-[3-(2-Ethoxy-phenyl)-5-oxo-4,5-dihydro-[1,2,4]triazin-6-yl]ethyl]propioamide The *title product* **13b** was obtained as a light yellow solid (0.39 g, 68%); m.p. 72-75 °C; [Found: C, 60.59; H, 6.39; N, 17.79. C₁₆H₂₀N₄O₃ requires C, 60.75; H, 6.37; N, 17.71%]; R_f (EtOAc/EtOH, 24:0.5) 0.27. ¹H NMR (CDCl₃, δ ppm): 8.6 (d, *J* 7.8 Hz, 1H, 6-H of C₆H₄), 7.60 (t, *J* 7.8 Hz, 1H, 4-H of C₆H₄), 7.18 (t, *J* 7.6 Hz, 1H, 5-H of C₆H₄), 7.09 (d, *J* 8.3 Hz, 1H, 3-H of C₆H₄), 6.97 (d, *J* 7.3 Hz, 1H, NHCO), 5.29 (qv, *J* 7.3 Hz, 1H, CHN), 4.37 (q, *J* 6.8 Hz, 2H, CH₂O), 2.26 (q, *J* 7.6 Hz, 2H, CH₂CO), 1.62 (t, *J* 6.8 Hz, 3H, CH₃CH₂O), 1.56 (d, *J* 6.8 Hz, 3H, CH₃CH), 1.17 (t, *J* 7.3 Hz, 3H, CH₃CH₂C). ¹³C NMR (CDCl₃, δ ppm): 169.7,

158.2, 158.1, 135.3, 132.2, 122.4, 116.5, 112.9, 65.7, 47.9, 23.7, 19.3, 15.0, 14.8.

4.6.3. N-[1-[3-(2-Ethoxyphenyl)-5-oxo-4,5-dihydro-[1,2,4]triazin-6-yl]ethyl]butyramide 13c. The *title product 13c* was obtained as a light yellow oil yield (0.42 g, 69%); [Found: C, 61.67; H, 6.68; N, 16.92. $C_{17}H_{22}N_4O_3$ requires C, 61.80; H, 6.71; N, 16.96%]; R_f (EtOAc) 0.28. 1H NMR ($CDCl_3$, δ ppm): 12.42 (s, 1H, NH \cdots O), 8.56 (dd, J 1.5 and 7.8 Hz, 1H, 6-H of C_6H_4), 7.56 (td, J 8.5 and 1.5 Hz, 1H, 4-H of C_6H_4), 7.16 (t, J 7.8 Hz, 1H, 5-H of C_6H_4), 7.07 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 6.95 (d, J 8.8 Hz, 1H, NHCO), 5.24 (dd, J 9 and 6.8 Hz, 1H, CHN), 4.35 (q, J 6.8 Hz, 2H, CH_2O), 2.19 (t, J 7.6 Hz, 2H, CH_2CO), 1.66 (m, 2H, $CH_2CH_2CH_3$), 1.60 (t, J 6.8 Hz, 3H, CH_3CH_2O), 1.52 (d, J 7.3 Hz, 3H, CH_3CH), 0.93 (t, J 7.3 Hz, 3H, CH_3CH_2). ^{13}C NMR ($CDCl_3$, δ ppm): 172.7, 158.0, 135.2, 132.0, 122.3, 116.5, 112.9, 65.7, 47.4, 39.0, 29.9, 19.5, 19.3, 15.0, 14.0.

4.6.4. N-[1-[3-(2-Ethoxy-phenyl)-5-oxo-4,5-dihydro-[1,2,4]triazin-6-yl]ethyl]benzamide 13d. The *title product 13d* was obtained as a light yellow oil yield (0.55 g, 82%); [Found: C, 65.85; H, 5.51; N, 15.32. $C_{20}H_{20}N_4O_3$ requires C, 65.92; H, 5.53; N, 15.38%]; R_f (EtOAc) 0.5. 1H NMR ($CDCl_3$, δ ppm): 12.50 (brs, 1H, NH \cdots O), 8.56 (dd, J 1.7 and 8.1 Hz, 1H, 6-H of C_6H_4), 7.835 (d, J 7.3 Hz, 3H, C_6H_5), 7.57 (td, J 1.7, 8.8 Hz, 1H, 4-H of C_6H_4), 7.48 (t, J 7.3 Hz, 1H, C_6H_5), 7.42 (t, J 7.3 Hz, 2H, C_6H_5), 7.15 (t, J 7.6 Hz, 1H, 5-H of C_6H_4), 7.07 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 5.48 (m, 1H, CHN), 4.34 (q, J 7 Hz, 2H, CH_2O), 1.65 (d, J 6.8 Hz, 3H, CH_3CH), 1.60 (t, J 6.8 Hz, 3H, CH_3CH_2O). ^{13}C NMR ($CDCl_3$, δ ppm): 166.8, 158.1, 135.3, 134.5, 132.1, 131.8, 128.7, 127.4, 122.3, 116.5, 112.9, 65.7, 48.2, 19.7, 15.0.

4.6.5. N-[1-[3-(2-Ethoxyphenyl)-5-oxo-4,5-dihydro-[1,2,4]triazin-6-yl]ethyl]oxalamic acid ethyl ester 13e. The *title product 13e* was obtained as a light yellow oil (0.38 g, 57%); [Found: C, 56.63; H, 5.62; N, 15.51. $C_{17}H_{20}N_4O_5$ requires C, 56.66; H, 5.59; N, 15.55%]; R_f (hexane/EtOAc, 3:7) 0.29. 1H NMR ($CDCl_3$, δ ppm): 12.03 (brs, 1H, NH \cdots O), 8.56 (dd, J 1.5, 7.8 Hz, 1H, 6-H of C_6H_4), 8.29 (d, J 8.8 Hz, 1H, NHCO), 7.57 (td, J 8.5, 1.5 Hz, 1H, 4-H of C_6H_4), 7.15 (t, J 7.8 Hz, 1H, 5-H of C_6H_4), 7.07 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 5.33-5.30 (m, 1H, CHN), 4.35-4.33 (m, 4H, CH_2O), 1.63-1.59 (m, 6H, $CH_3CH+CH_3CH_2O$), 1.37 (t, J 7.1 Hz, 3H, CH_3CH_2O). ^{13}C NMR ($CDCl_3$, δ ppm): 160.6, 158.1, 156.1, 135.3, 132.1, 122.3, 116.4, 112.9, 65.8, 63.4, 47.4, 19.2, 15.0

4.7. General procedure for the synthesis of the 2(2-ethoxyphenyl)-7-alkyl-5-methyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14

To a magnetically stirred solution of **13** (0.0010 mol) in toluene (35 mL), phosphorus oxychloride (0.17 mL, 0.0019 mol) was added at room temperature. The resulting mixture was heated under reflux for 2 h and then cooled to room temperature. The solvent and an excess of phosphorus oxychloride were evaporated *in vacuo* and the residue was treated with saturated aqueous sodium bicarbonate solution (15 mL) and chloroform (15 mL). The mixture was shaken vigorously until all solid had dissolved. The chloroform layer was separated and the aqueous phase was extracted with another portion of chloroform (2 \times 15 mL). The chloroform extracts were combined, dried ($MgSO_4$), filtered and the solvent was evaporated. The residue was purified by column chromatography on silica gel (chloroform) afforded a product which was then crystallised from an appropriate solvent.

4.7.1. 2-(2-Ethoxyphenyl)-5,7-dimethyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14a. The *title compound* was prepared from **13a** (0.12 g, 0.0004 mol). The final product was crystallised from ethyl acetate/ethyl ether to give the *title product 14a* (46 mg, 41%) as white crystals, m.p. 209-210 $^{\circ}C$; [Found: C, 63.61; H, 5.65; N, 19.65. $C_{15}H_{16}N_4O_2$ requires C, 63.37; H, 5.67; N, 19.71%]; R_f (petroleum ether/EtOAc, 1:1) 0.16. v_{max} (crystal): 3280, 2981, 2923, 1695, 1619,

1585 cm^{-1} . 1H NMR ($CDCl_3$, δ ppm): 10.0 (brs, 1H, NH \cdots O), 8.18 (dd, J 1.5, 7.8 Hz, 1H, 6-H of C_6H_4), 7.50 (t, J 8.5 Hz, 1H, 4-H of C_6H_4), 7.11 (t, J 7.6 Hz, 1H, 5-H of C_6H_4), 7.04 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 4.26 (q, J 7 Hz, 2H, CH_2O), 2.65 (s, 3H, CH_3C), 2.63 (s, 3H, CH_3C), 1.56 (t, J 7.1 Hz, 3H, CH_3). MS (ESI, matrix CCA): $[M+H]^+$ found 285.3. $C_{15}H_{16}N_4O_2$ requires 285.1273.

4.7.2. 2-(2-Ethoxyphenyl)-5-methyl-7-ethyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14b. The *title compound* was prepared from **13b** (1.028 g, 0.0032 mol). The final product was crystallised from ethyl acetate to afford the *title product 14b* (0.7 g, 72%) as a light yellow solid, m.p. 176-178 $^{\circ}C$; [Found: C, 64.53; H, 6.10; N, 18.83. $C_{16}H_{18}N_4O_2$ requires C, 64.41; H, 6.08; N, 18.78%]; R_f (petroleum ether/EtOAc, 1:1) 0.23. v_{max} (crystal): 3307, 2979, 2941, 2916, 1697, 1618, 1597 cm^{-1} . 1H NMR ($CDCl_3$, δ ppm): 10.03 (bs, 1H, NH \cdots O), 8.19 (dd, J 1.5, 7.8 Hz, 1H, 6-H of C_6H_4), 7.53 (dt, J 1.7, 7.8 Hz, 1H, 4-H of C_6H_4), 7.15 (dt, J 0.9, 8.3 Hz, 1H, 5-H of C_6H_4), 7.07 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 4.29 (q, J 7 Hz, 2H, CH_2O), 3.10 (q, J 7.65 Hz, 2H, CH_2C), 2.68 (s, 3H, CH_3C), 1.58 (t, J 7.1 Hz, 3H, CH_3CH_2O), 1.45 (t, J 7.6 Hz, 3H, CH_3CH_2C). MS (ESI, matrix CCA): m/z $[M+H]^+$ found 299.4. $C_{16}H_{18}N_4O_2$ requires 299.1430.

4.7.3. 2-(2-Ethoxyphenyl)-5-methyl-7-propyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14c. The *title compound* was prepared from **13c** (0.26 g, 0.0008 mol). The final product was crystallised from ethyl acetate to afford the *title product 14c* (0.19 g, 78%) as a light yellow solid, m.p. 142-144 $^{\circ}C$; [Found: C, 65.45; H, 6.43; N, 17.88. $C_{17}H_{20}N_4O_2$ requires C, 65.37; H, 6.45; N, 17.94%]. R_f (petroleum ether/EtOAc, 1:1) 0.31. v_{max} (crystal): 3284, 2965, 2954, 2929, 1683, 1612, 1598 cm^{-1} . 1H NMR ($CDCl_3$, δ ppm): 9.98 (brs, 1H, NH \cdots O), 8.16 (dd, J 1.5, 7.8 Hz, 1H, 6-H of C_6H_4), 7.49 (t, J 8.5 Hz, 1H, 4-H of C_6H_4), 7.13 (t, J 7.6 Hz, 1H, 5-H of C_6H_4), 7.05 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 4.26 (q, J 7 Hz, 2H, CH_2O), 3.0 (t, J 7.6 Hz, 2H, CH_2C), 2.64 (s, 3H, CH_3C), 1.90-1.86 (m, J 7.5 Hz, 2H, $CH_2CH_2CH_3$), 1.56 (t, J 7.1 Hz, 3H, CH_3CH_2O), 1.02 (t, J 7.3 Hz, 3H, CH_3CH_2C). ^{13}C NMR ($CDCl_3$, δ ppm): 157.2, 155.2, 146.3, 146.1, 139.9, 133.3, 130.3, 121.9, 117.8, 114.1, 113.3, 65.5, 28.2, 21.2, 14.9, 14.7, 14.2. MS (ESI, matrix CCA): $[M+H]^+$ found 313.3. $C_{17}H_{20}N_4O_2$ requires 313.1586.

4.7.4. 2-(2-Ethoxyphenyl)-5-methyl-7-phenyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14d. The *title compound* was prepared from **13d** (0.37 g, 0.001 mol). The final product was crystallised from ethyl acetate to afford the *title product 14d* (0.27 g, 77%) as a light yellow solid, m.p. 181-182 $^{\circ}C$; [Found: C, 69.55; H, 5.26; N, 16.21. $C_{20}H_{18}N_4O_2$ requires C, 69.35; H, 5.24; N, 16.17%]; R_f (petroleum ether/EtOAc, 1:1) 0.48. v_{max} (crystal): 3266, 2981, 1697, 1617, 1597 cm^{-1} . 1H NMR ($CDCl_3$, δ ppm): 10.18 (brs, 1H, NH \cdots O), 8.39 (d, J 7.8 Hz, 2H, C_6H_5), 8.18 (d, J 8.3 Hz, 1H, 6-H of C_6H_4), 7.51 (q, J 7.2 Hz, 3H, C_6H_5), 7.45 (t, J 7.1 Hz, 2H, 4-H of C_6H_4), 7.13 (t, J 7.6 Hz, 1H, 5-H of C_6H_4), 7.04 (d, J 8.3 Hz, 1H, 3-H of C_6H_4), 4.26 (q, J 6.8 Hz, 2H, CH_2O), 2.74 (s, 3H, CH_3C), 1.57 (t, J 7.1 Hz, 3H, CH_3CH_2O). ^{13}C NMR ($CDCl_3$, δ ppm): 157.3, 155.0, 146.9, 142.4, 141.0, 133.5, 130.3, 129.7, 129.0, 128.7, 122.0, 117.4, 115.5, 113.2, 65.5, 14.9, 14.8. MS (ESI, matrix CCA): $[M+H]^+$ found 347.2. $C_{20}H_{18}N_4O_2$ requires 347.1430.

4.7.5. 2-(2-Ethoxyphenyl)-5-methyl-7-carboethoxy-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 14e. The *title compound* was prepared from **13e** (1.0 g, 0.0028 mol). Crystallisation from ethyl acetate afforded the *title product 14e* (0.69 g, 73%) as a light yellow solid, m.p. 214-216 $^{\circ}C$; [Found: C, 59.65; H, 5.32; N, 16.31. $C_{17}H_{18}N_4O_4$ requires C, 59.64; H, 5.30; N, 16.37%]; R_f (petroleum ether/EtOAc, 1:1) 0.21. v_{max} (crystal): 3278, 2971, 2919, 1720, 1698, 1614, 1592 cm^{-1} . 1H NMR ($CDCl_3$, δ ppm): 10.56 (brs, 1H, NH \cdots O), 8.41 (d, J 7.8 Hz, 1H, 6-H of C_6H_4), 7.56 (t, J 7.8 Hz, 1H, 4-H of C_6H_4), 7.18 (t, J 7.6 Hz, 1H, 5-H of C_6H_4), 7.08 (d, J 8.8 Hz, 1H, 3-H of C_6H_4), 4.56 (q, J 7 Hz, 2H, CH_2O), 4.323 (q, J 7 Hz, 2H, CH_2O), 2.74 (s,

3H, CH₃C), 1.62 (t, *J* 7 Hz, 3H, CH₂CH₂O), 1.52 (t, *J* 7.1 Hz, 3H, CH₂CH₂O). ¹³C NMR (CDCl₃, δ ppm): 157.54, 157.48, 154.4, 148.4, 141.2, 134.0, 132.8, 130.8, 122.3, 117.8, 116.7, 113.2, 65.7, 62.2, 14.9, 14.8, 14.6. MS (ESI, matrix CCA): [M+H]⁺ found 343.2. C₁₇H₁₈N₄O₄ requires 343.1328.

4.8. 2-[(2-Ethoxy-5-(4-ethylpiperazine-1-sulphonyl)phenyl)-7-etoxy carbonyl-5-methyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one 15e

2-(2-Ethoxyphenyl)-7-ethyl-5-methyl-imidazo[5,1-f][1,2,4]triazin-4(3H)-one **14e** (200 mg, 5.84×10⁻⁴ mol) was slowly added to chlorosulfonic acid (0.5 mL, 874.5 mg; 7.5×10⁻⁴ mol). The reaction mixture was stirred for 1.5 h at room temperature. The product was poured into ice-water (5 mL) and extracted with dichloromethane (3×15 mL). The chloroform extracts were combined, dried (MgSO₄), filtered and the solvent was evaporated to give a benzenesulfonyl chloride (234 mg, 92%) as white powder; R_f (EtOAc/EtOH 3:1) 0.34; R_f (EtOAc) 0.7. Benzenesulfonyl chloride (234 mg, 5.32×10⁻⁴ mol) was dissolved in THF (3 mL) and cooled to 0 °C. *N*-Ethylpiperazine (133.7 mg, 0.15 ml; 1.17×10⁻³ mol) was added to this solution. The reaction mixture was stirred at room temperature for 18 h. The solvent was evaporated *in vacuo* and the residue was dissolved in dichloromethane (5 mL). The organic solution was washed twice with water, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography on silica gel (hexane/EtOAc, 1:1 v/v) afforded the product as a light yellow solid which was crystallised from abs. ethanol to give 170 mg (62%), m.p. 215–7 °C; [Found: C, 53.45; H, 5.85; N, 16.25; S, 6.20. C₂₃H₃₀N₆O₆S requires C, 53.27; H, 5.83; N, 16.21; S, 6.18%]; R_f (EtOAc/EtOH, 6:1) 0.23; R_f (EtOAc) 0.1. *v*_{max}(crystal): 3289, 2981, 2935, 1710, 1614, 1590 cm⁻¹. ¹H NMR (CDCl₃, δ ppm): 10.18 (bs, 1H, NH...O), 8.69 (d, *J* 2.4 Hz, 1H, 6-H of C₆H₄), 7.92 (dd, *J* 2.4 and 8.8 Hz, 1H, 4-H of C₆H₄), 7.20 (d, *J* 8.8 Hz, 1H, 3-H of C₆H₄), 4.54 (q, *J* 7.2 Hz, 2H, CH₂O), 4.40 (q, *J* 7 Hz, 2H, CH₂O), 3.21 (bs, 4H, 2×CH₂N), 2.73 (s, 3H, CH₃C), 2.67 (bs, 4H, 2×CH₂N), 2.53 (bs, 2H, NCH₂CH₃), 1.64 (t, *J* 6.8 Hz, 3H, CH₃), 1.52 (t, *J* 7.1 Hz, 3H, CH₃), 1.12 (bs, 3H, NCH₂CH₃). ¹³C NMR: (CDCl₃, δ ppm): 157.5, 157.5, 154.4, 148.4, 141.2, 134.0, 132.8, 130.8, 128.5, 117.8, 116.7, 113.2, 65.7, 62.2, 54.2, 48.4, 45.8, 14.9, 14.6, 13.3. MS (ESI, matrix CCA): [M+H]⁺ found 519.3. C₂₃H₃₀N₆O₆S requires 518.20.

4.9. Ethyl 3-bromo-2-oxo-butylate **16**

This compound was prepared from CuBr₂ (20 g, 0.0895 mol) and ethyl 2-oxobutanoate (7.15 g, 0.055 mol) according to the method reported by Okonya²⁶ in 87% yield (lit.²⁶ yield 80%); ¹H NMR (CDCl₃, δ ppm): 5.18 (q, *J* 6.8 Hz, 1H, CHBr), 4.39 (m, 2H, CH₂O), 1.81 (d, *J* 6.8 Hz, 3H, CH₃CH), 1.40 (t, *J* 7.1 Hz, 3H, CH₃CH₂).

4.10. X-ray structure analysis of **14b**

Diffraction data were recorded on a KUMA KM4 diffractometer with graphite-monochromated Mo-K_α radiation, using a Sapphire-2 CCD detector (Oxford Diffraction Ltd). The apparatus was equipped with an open flow thermostat (Oxford Cryosystems), which enabled experiments at 120 K. The structures were solved with direct methods and refined with the SHELX98 program package, with the full-matrix least-squares refinement based on *F*².²⁷ The structural drawing was prepared using the Mercury software.²⁸ Crystals of **14b** suitable for X-ray diffraction were obtained by allowing a refluxed solution of the product in ethyl acetate to cool slowly at room temperature (without temperature control) and allowing the solvent to evaporate for 20h.

Crystal data for C₁₆H₁₈N₄O₂, *M* = 298.34, triclinic, space group *P* -1, *a* = 7.9963(8), *b* = 9.0334(11), *c* = 11.3995(12) Å, *V* = 724.36(14) Å³, *T* = 120 K, *Z* = 2, ρ_c = 1.368 g cm⁻³; μ(Mo K_α) =

0.093 mm⁻¹, λ = 0.71073 Å, 3756 reflection measured, 2504 unique (*R*_{int} = 0.0854). Final residuals for 202 parameters were *R*₁ = 0.1005, *wR*₂ = 0.2678 for 2504 reflection with *I* > 2σ(*I*), and *R*₁ = 0.1283, *wR*₂ = 0.299 for all data.

Crystallographic data (excluding structure factors) for the structure in this paper have been deposited with the Cambridge Crystallographic Data Center as supplementary publication number CCDC 905439. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge, CB2 1EZ, UK [fax: +44(0)-1223-336033 or e-mail: deposit@ccdc.cam.ac.uk].

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Supplementary Material

IR, MS, NMR spectra (^1H and ^{13}C) for all products and crystal data for **14b**. This material is available free of charge.

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