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C3-THIA AND C3-CARBA ISOSTERES OF A CARBOCYCLIC INFLUENZA NEURAMINIDASE INHIBITOR, (3*R*,4*R*,5*S*)-4-ACETAMIDO-5-AMINO-3-PROPOXY-1-CYCLOHEXENE-1-CARBOXYLIC ACID

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Abstract. The importance of the oxygen atom in the C_3 ether side chain of a carbocyclic influenza neuraminidase inhibitor 3 was investigated by replacement of the C_3 ether oxygen atom of 3 with either a sulfur atom (compound 4) or a carbon atom (compound 5). The regio- and stereospecific syntheses of both isoteres are described starting from (-)-quinic acid. © 1997 Elsevier Science Ltd.

Recently, we have described a new class of potent carbocyclic influenza neuraminidase (NA) inhibitors, which are transition-state analogues of the sialic acid cleavage reaction by NA.¹ In this series, a new hydrophobic pocket in the region corresponding to the glycerol subsite of sialic acid was identified. In order to understand the structure-activity relationships of this class of compounds systematic modifications of substituents attached to the cyclohexene ring were investigated.² Of these, the C₃-lipophilic side chains proved to play a significant role for NA inhibitory activity. From extensive structure-activity relationship studies of the C₃ lipophilic side chain and X-ray crystallographic analysis, the 3-pentyloxy side chain as shown in compound 1 appeared to be optimal for NA inhibition.¹ On the basis of in vitro and in vivo activity, compound 2 (designated as GS4104) was selected as a clinical candidate for the oral treatment and prophylaxis of influenza infection.³



The present investigation addresses the importance of the ether oxygen at the C_3 position for NA inhibition. It has been proposed that the catalytic mechanism for the cleavage of sialic acid from glycoconjugates involves the formation of a C_2 carbonium cation, which in turn is stabilized by the neighboring oxygen atom.⁴ We expected that inhibitory activity might be affected by replacement of the oxygen atom of the ether side chain at the C_3 position with either a sulfur or a carbon atom since they would have different electronic and binding properties.

The preparation of the sulfide analogue 4 was accomplished starting from the trityl protected aziridine 6, which is synthesized from (-)-quinic acid 7 (Scheme 1). Ring opening of 6 with 1-propanethiol under BF₃•Et₂O catalysis followed by acetylation with acetyl chloride gave acetamide 8. Catalytic hydrogenation of the azide group in 8 over Lindlars catalyst at atmospheric pressure gave the amine that was directly converted into the

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corresponding hydrochloride salt 9. Surprisingly, attempted saponification of 9 with aqueous KOH gave mostly decomposed material. At the present time it is not clear as to why 9 is unstable to basic hydrolysis. Consequently, 9, which is in essence the methyl ester prodrug of the amino acid 4, was used for the enzymatic assay.⁵



^aReagents: (a) 1-propanethiol, BF₃•Et₂O; 73% (b) AcCl, Et₃N, CH₂Cl₂; 98% (c) H₂, Lindlars catalyst, EtOAc then 1 N HCl; 67%.

The *n*-butyl side-chain analogue **5** was synthesized starting from the readily accessible protected quinic acid **10**⁶ (Scheme 2). Oxidation of **10** with PCC followed by Wittig olefination gave olefin **11** as a 2:1 mixture of double-bond isomers. Protection of the tertiary hydroxyl group followed by catalytic hydrogenation at atmospheric pressure gave exclusively the *n*-butyl derivative with the desired α stereochemistry.⁷ The hydrogenated product was then deprotected with TBAF to afford hydroxy ester **12** and then directly eliminated in a regiospecific manner with sulfuryl chloride to the α , β -unsaturated ester. Subsequent deprotection with 40% TFA affords diol **13**. Treatment of **13** with 1.1 equivalents of methanesulfonyl chloride gave a 6:1 mixture of C₅ and C₄ mesylates **14a** and **14b**, which was treated with DBU at ambient temperature to give the desired epoxide **15** with NaN₃/NH₄Cl, mesylation of the resultant alcohol and followed by aziridine formation with PPh₃. The aziridine **17** was then opened in a regio- and stereospecific manner with NaN₃/NH₄Cl and acetylated with acetyl chloride. Finally, reduction of the azide group with PPh₃ followed by saponification with aqueous KOH provides the amino acid **5**.

The enzymatic neuraminidase inhibitory activities for the three analogues are shown in Table 1.⁸ It can be seen that when the ether oxygen atom of 3 is replaced with a methylene group (compound 5) there is a small decrease in activity. This suggests that the electronic difference at the C_3 position has a minimal effect on NA inhibitory activity. Further confirming this, the neuraminidase inhibitory activity of the sulfide analogue 4 was also very comparable to that of 3. However, it cannot be ruled out that the thioalkyl chain of 4 is considerably longer than the ether side chain of 3, which may result in different hydrophobic interactions in the active site.



^aReagents: (a) PCC, pyridine; 74% (b) *n*-BuPh₃Br, *n*-BuLi, THF; 81% (c) Et₃SiOTf, 2,6-lutidine; 93% (d) H₂,10% Pd/C; 97% (e) TBAF, THF; 77% (f) i. SO₂Cl₂ ii. MeOH; 90% (g) 40% TFA, CH₂Cl₂; 75% (h) MsCl, Et₃N; 87% (i) DBU, THF; 71% (j) NaN₃, NH₄Cl, MeOH, H₂O; 95% (k) MsCl, Et₃N; 97% (l) i. PPh₃, THF ii. Et₃N/H₂O;71% (m) NaN₃, NH₄Cl, DMF; 86% (n) AcCl, pyridine; 68% (o) PPh₃, THF, H₂O; 95% (p) aq. KOH; 85%.

compound ^a	x	enzyme ^b IC ₅₀ (nM)	
3	0	130	
4	S	212 ^c	
5	CH ₂	220	

Table 1. Influenza Neuraminidase Inhibition

^aAll compounds gave satisfactory spectral and analytical data. ^bNA (H1N1)

^cCompound 9 was incubated in rat plasma at 37 [°]C prior to assay.

References and Notes:

- Kim, C. U.; Lew, W.; Williams, M. A.; Liu, H.; Zhang, L.; Swaminathan, S.; Bischofberger, N.; Chen, M. S.; Mendel, D. B.; Tai, C. Y.; Laver, W. G.; Stevens, R. C. J. Am. Chem. Soc. 1997, 119, 681.
- Williams, M. A.; Kim, C. U.; Lew, W.; Mendel, D. B.; Tai, C. Y.; Escarpe, P.; Laver, W. G.; Stevens, R. C. manuscript submitted.
- 3. Kim, C. U.; Lew, W.; Williams, M. A.; Zhang, L.; Swaminathan, S.; Bischofberger, N.; Chen, M. S.; Mendel, D. B.; Li, W.; Tai, L.; Escarpe, P.; Cundy, K. C.; Eisenberg, E. J.; Lacy, S.; Sidwell, R. W.; Stevens, R. C.; Laver, W. G. 1996. New Potent, Orally Active Neuraminidase Inhibitors as Anti-Influenza Agents: In Vitro and In Vivo Activity of GS4071 and Analogs, abstr. H44, P171. *In* Abstracts of the 36th Interscience Conference on Antimicrobial Agents and Chemotherapy, American Society for Microbiology, Washington, D.C.
- (a) Taylor, N. R.; von Itzstein, M. J. Med. Chem. 1994, 37, 616. (b) Chong, A. K. J.; Pegg, M.S.; von Itzstein, M. Biochem. Int. 1991, 24, 165. (c) Janakiraman, M. N.; White, C. L.; Laver, W. G.; Air, G. M.; Luo, M. Biochemistry, 1994, 33, 8172.
- Compound 9 was incubated in rat plasma for 30 min at 37 °C to convert the methyl ester to the parent amino acid 4. This was then used directly in the enzymatic assay.
- 6. Montchamp, J-L.; Tian, F.; Hart, M. E.; Frost, J. W. J. Org. Chem. 1996, 61, 3897.
- 7. The steric bulk of the triethylsilyl group is essential for the stereochemical outcome of the hydrogenation since direct hydrogenation of 11 under the same conditions yields a 2:1 mixture of $\alpha:\beta$ *n*-butyl derivatives.
- Potier, M.; Marneli, L.; Belisle, M.; Dallaire, L.; Melancon, S. B. Anal. Biochem. 1979, 94, 287. Also see ref 1.