Synthesis of *N*-[2-(2,4-Difluorophenoxy)trifluoromethyl-3-pyridyl]sulfonamides and Their Inhibitory Activities against Secretory Phospholipase A₂

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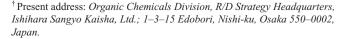
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N-[2-(2,4-Difluorophenoxy)trifluoromethyl-3-pyridyl]sulfonamide derivatives 3—6 were prepared by the reaction of 3-pyridylamines and sulfonyl chlorides. Inhibitory activities of these compounds toward secretory phospholipase A₂ (sPLA₂) were examined and N-[2-(2,4-difluorophenoxy)-5-trifluoromethyl-3-pyridyl]-2-naphthalenesulfonamide (5c) was found to be the most potent against sPLA₂ with an IC₅₀ value of 90 μ M.

Key words trifluoromethylpyridine; sulfonamide; secretory phospholipase A_2 inhibitor; 2,4-difluorophenoxy; naphthalene-sulfonamide

Fluoride functions involving trifluoromethyl (CF₂) group have a profound effect on the physical, chemical, and biological properties of organic molecules, including lipophilicity and chemical and metabolic stability.¹⁾ Due to its strong electron withdrawing property and relatively larger size as compared with a methyl group, the trifluoromethyl group is featured in many important pharmaceuticals²⁾ and pesticides.³⁾ To date, a great number of CF₃-substituted aromatic and aliphatic compounds have been synthesized.⁴⁻⁶ On the other hand, the corresponding CF₃-substituted pyridine derivatives have received relatively less attention among the CF₃-substituted aromatic compounds.⁷⁾ The strong electron withdrawing property of the CF₃ group and the electron deficient pyridine ring confer unique chemical and physical properties to heterocyclic compounds, and therefore, these compounds are quite attractive. In fact, CF₂-substituted pyridines have shown interesting profiles in some biological molecules including agrochemicals⁸⁾ and medicinal candidates.⁹⁾

CF₃-substituted pyridylsulfonamide 1 was reported to inhibit secretory phospholipase A2 (sPLA2).101 Recently, we have simplified the structure of 1 and synthesized compounds $2.^{11}$ The structurally more simple CF₃-substituted 2pyridylsulfonamides 2a and b possess almost the same level of sPLA₂ inhibitory activity (IC₅₀ values, 0.58, 0.68 mm, respectively) as that (IC₅₀ values, 0.52 to 2.1 mM) of 1. In order to extend these studies, we have designed N-(6- or 5-trifluoromethyl-3-pyridyl)sulfonamides 3-6 as a new sPLA₂ inhibitors (Fig. 1). They are positional isomers of compound 2. A nitrogen atom at the N-1 position in the structure of 2c was moved to the para (C-4) position in the structure of 3. In addition, we envisaged replacing a hydrogen atom at the C-2 position with a 2,4-difluorophenoxy group on the pyridine ring, and expected an enhancement of its activity, because highly lipophilic 2,4-difluorophenyl aryl or heteroaryl ether is a functional unit, which has often been used for improvement of the activity in drug design. In fact, many successful cases for enhancing an activity have been reported with this substituent.12-17)



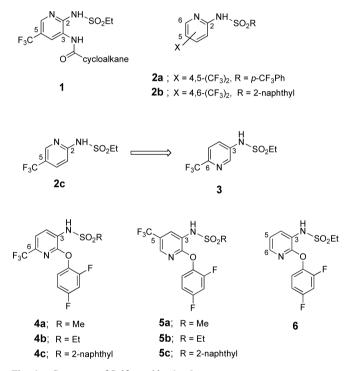


Fig. 1. Structures of Sulfonamides 1-6

In this paper, we report the synthesis of these derivatives 3-6 and describe their inhibitory activities against sPLA₂.

Results and Discussion

We first prepared a simple sulfonamide **3**. Sulfonylation of 3-amino-6-trifluoromethylpyridine with ethanesulfonyl chloride in pyridine at room temperature gave a mixture of *N*-ethanesulfonamide and *N*,*N*-bisethanesulfonamide, which was heated in aq. KOH and 1,4-dioxane at 90 °C for 30 min to afford **3** in 72% yield (Chart 1).

Inhibitory activity against porcine pancreatic sPLA₂ was examined for compounds 2c and 3 by the same method described in the preceding paper.¹¹ Compound 2c inhibited sPLA₂ only weakly by 4.6% at 0.5 mM concentration, while compound 3 inhibited the enzyme by 21.2% at the same con-

centration. Therefore, we would expect higher inhibitory activity with the N-sulfonamide analogues of 3-amino-6-(trifluoromehtvl)pyridine than with the analogues in the series of 2.

Following the initial plan, the 2,4-difluorophenoxy group was introduced at the C-2 position by substitution reaction of 2-chloropyridine with 2,4-difluorophenol (Chart 2). The reaction of 2-chloro-3-nitro-6-(trifluoromethyl)pyridine with 2,4-difluorophenol in the presence of potassium carbonate in acetone gave 2-pyridyl ether 10 in 62% yield. Due to the anion stabilizing effect of the C-5 CF₃ group in cooperation with the C-3 nitro group under an SNAr reaction mechanism, the reaction of 8 gave 11 in excellent yield. The reaction of 3-nitro-2-pyridyl chloride 9 afforded 12 in 83% yield.¹⁸⁾ Reduction of the nitro group with sodium dithionite in a mix-

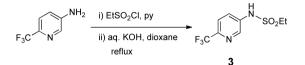


Chart 1. Synthesis of 3

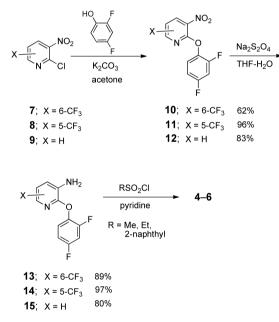


Chart 2. Synthesis of Sulfonamides 4-6

Table 1. Chemical Yields of Sulfonamides 3-6 and Their sPLA₂ Inhibitory Activities

ture of tetrahydrofuran (THF) and water gave amines 13, 14, and 15¹⁹ in 89, 97, and 80% yields, respectively.

Treatment of 3-pyridylamine 13 and 14 with methane- and 2-naphthalenesulfonyl chloride by the same manner described for the synthesis of 3 gave the corresponding sulfonamides, 4a, 4c, 5a and 5c, in moderate to excellent yields as shown in Table 1. The reaction of 15 with ethanesulfonyl chloride gave 6 in 97% yield. It is interesting that the electron deficient CF₃-substituted pyridylamine reacted with sulfonyl chloride. In contrast, the reaction with ethanesulfonyl chloride with 13 and 14 gave sulfonamides 4b and 5b in low to medium yields with recovery of starting material, probably due to the less reactive ethanesulfonyl chloride and poor nucleophilic reactivity of CF₃-substituted pyridylamine.

All new sulfonamides were evaluated for their inhibitory activity against sPLA₂.²⁰⁾ We determined IC₅₀ values for all the sulfonamides. The values were generated from at least three trials, and each trial consisted of a mean value generated from duplicate assays. These results are summarized in Table 1.

Compound 3 was a poor inhibitor, an IC_{50} value of >3 mm(entry 1). The introduction of a 2,4-difluorophenoxy group at the C-2 position of 3-amino-6-(trifluoromehtyl)pyridine greatly increased the inhibitory potency of the corresponding methanesulfonamide 4a, which inhibited sPLA₂ with an $IC_{50}=0.81$ mM. When the methanesulfonamide was replaced with ethanesulfonamide, the potency of compound 4b increased by 2 times (entries 2 and 3). In the previous investigation with compounds 2, 2-naphthalenesulfonamide indicated an excellent profile in sPLA₂ inhibition. Indeed, the 6-CF₃-susbutitued 2-naphthalenesulfonamide 4c exhibited the highest inhibition activity, with an IC_{50} value of 0.12 mM(entry 4). In the series of 5-CF₃ analogues, a similar trend of inhibition was observed (entries 5-7). 2-Naphthalenesulfonamide 5c exhibited an IC_{50} value of 0.09 mM, which was the highest value in this series and 6.4- and 7.5-fold more potent than 2a and b. On the other hand, the corresponding analogue lacking the CF₃ group was a poor inhibitor with an IC_{50} value of >3 mM (entry 8). Therefore, the CF₃ as well as 2,4-difluorophenoxy groups appear to perform an essentially important role in enhancing the inhibitory activity against sPLA₂.

In conclusion, we have synthesized N-[2-(2,4-difluoro-

X-FN-SO ₂ R							
Entry	Compound No.	Х	R	Y	$\operatorname{Yield}^{a)}(\%)$	mp (°C)	IC ₅₀ (тм)
1	3	6-CF ₃	Et	Н	72	101—102	>3
2	4a	6-CF ₃	Me	$OC_6H_3(2,4-F_2)$	58	104-105	0.81
3	4b	6-CF ₃	Et	$OC_6H_3(2,4-F_2)$	39	89—90	0.44
4	4c	6-CF ₃	2-Naphthyl	$OC_6H_3(2,4-F_2)$	79	90—92	0.12
5	5a	5-CF ₃	Me	$OC_6H_3(2,4-F_2)$	72	88—90	1.03
6	5b	5-CF ₃	Et	$OC_6H_3(2,4-F_2)$	26	75—77	0.41
7	5c	5-CF ₃	2-Naphthyl	$OC_6H_3(2,4-F_2)$	91	44—49	0.09
8	6	Н	Ēt	$OC_6H_3(2,4-F_2)$	97	109-110	>3
9	2a	$4,5-(CF_3)_2$	b)	NHSO ₂ -4-CF ₃ C ₆ H ₄	_	_	0.58
10	2b	$4,6-(CF_3)_2$	b)	NHSO ₂ -2-naphthyl			0.68

a) Isolated yields. b) C-3 substituent is H instead of NHSO₂R.

phenoxy)-6- and 5-trifluoromethyl-3-pyridyl]sulfonamides 4 and 5, and revealed their inhibitory activity against sPLA₂. The potency of 4c and 5c increased by 5—7 fold as compared to the previous series of compounds 2a and 2b.

Experimental

General Melting points were determined on a Yanagimoto micro-melting point apparatus and were not corrected. ¹H-NMR spectra were recorded on a JEOL JNM-GSX 400 (400 MHz) or JNM-AL 300 (300 MHz) spectrometers in CDCl₃ with tetramethylsilane as an internal standard. IR spectra were recorded on JASCO FT/IR-410 instrument. Column chromatography was carried out using Merck silica gel 60 (70–230 mesh). Mass spectra were recorded on a JEOL JMS-GCmate spectrometer.

Preparation of 2-(2,4-Diffuorophenoxy)pyridime To a solution of 2chloropyridine (3.0 mmol) and 2,4-diffuorophenol (3.0 mmol) in anhydrous acetone (20 ml) was added powdered K_2CO_3 (480 mg, 3.45 mmol). The mixture was stirred for 1—2 h at room temperature and refluxed for 3 h. After cooling, ice water (50 ml) was added to the reaction mixture and the mixture was extracted with ether (100 ml). The organic layer was dried over Na₂SO₄, and condensed. The crude product was purified by flash chromatography on silica gel eluted with 10 to 20% EtOAc in hexane to give the aryl ether.

Compound **10** was obtained from **7** in 62% yield. Oil, Rf=0.31 (10% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 6.91—7.02 (2H, m), 7.22—7.30 (1H, m), 7.57 (1H, d, *J*=8.1 Hz), 8.54 (1H, dd, *J*=8.1, 0.73 Hz). Electron ionization (EI)-MS m/z: 320 (M⁺), 129, 101, 63. EI-high resolution (HR)-MS m/z: 320.0212 (Calcd for C₁₂H₅F₈N₂O₃: 320.0220).

Compound **11** was obtained from **8** in 96% yield. Yellow prisms, mp 66— 68 °C. Rf=0.39 (10% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 6.92—7.03 (2H, m), 7.22 (1H, m), 7.44 (1H, s), 8.57 (1H, s). *Anal.* Calcd for C₁₂H₅F₅N₂O₃: C, 45.02; H, 1.57; N, 8.75. Found: C, 45.22; H, 1.60; N, 8.62.

Compound **12** was obtained from **9** in 83% yield.¹⁸) Pale yellow solid, mp 53—54 °C, Rf=0.32 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 6.92—7.01 (2H, m), 7.18—7.29 (2H, m), 8.30—8.43 (2H, m).

Reduction of Nitro Group to Amine To a solution of $Na_2S_2O_4$ (6.3 g, 36 mmol) in water (60 ml) was added a solution of 2-(2,4-difluorophenoxy)-3-nitropyridine (2.5 mmol) in THF (15 ml) with dropwise at room temperature. The reaction mixture was stirred for 2.5 h at the same temperature. Then, THF (50 ml) was added to the mixture. THF layer was separated and it was washed with brine and dried over MgSO₄. The solvent was removed to give solid product, which was purified by recrystalization from a mixture of EtOAc and hexane to give pure amine.

Compound **13** was obtained from **10** in 89% yield. White needles, mp 78—81 °C. Rf=0.42 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 4.20 (2H, bs), 6.90—7.01 (2H, m), 7.18 (1H, dd, J=2.1, 0.3 Hz), 7.24 (1H, m), 7.74 (1H, dd, J=2.1, 1.2 Hz). *Anal.* Calcd for C₁₂H₇F₅N₂O: C, 49.67; H, 2.43; N, 9.65. Found: C, 49.89; H, 2.44; N, 9.67. EI-MS *m*/*z*: 290 (M⁺), 271, 193, 141. EI-HR-MS *m*/*z*: 290.0484 (Calcd for C₁₂H₇F₅N₂O: 290.0478).

Compound **14** was obtained from **11** in 97% yield. White prisms, mp 68—70 °C. Rf=0.28 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 4.32 (2H, bs), 6.90—7.04 (2H, m), 7.02 (1H, d, J=7.8 Hz), 7.22 (1H, d, J=7.8 Hz), 7.28 (1H, m). *Anal.* Calcd for C₁₂H₇F₅N₂O: C, 49.67; H, 2.43; N, 9.65. Found: C, 49.91; H, 2.21; N, 9.59. EI-MS *m*/*z*: 290 (M⁺). EI-HR-MS *m*/*z*: 290.0473 (Calcd for C₁₂H₇F₅N₂O: 290.0478).

Compound **15** was obtained from **12** in 80% yield.¹⁹⁾ Pale yellow powder, mp 87—89 °C. Rf=0.21 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 1.97 (2H, bs), 6.84 (1H, dd, J=7.5, 4.8 Hz), 6.87—6.97 (2H, m), 7.30 (1H, dd, J=7.5, 1.5 Hz), 7.23 (1H, m), 7.49 (1H, dd, J=4.8, 1.5 Hz).

General Synthesis of Sulfonamide To a stirred solution of 3-aminopyridine 13, 14, or 15 (3.0 mmol) in pyridine (3 ml) were added dropwise methane-, ethane-, or 2-naphthalenesulfonyl chloride (9 mmol) at room temperature, and the mixture was stirred for 2—6 h at the same temperature. Then, the mixture was quenched with cold $2 \times$ hydrochloric acid (50 ml), and extracted with EtOAc (50 ml). The organic extract was washed with brine and the solvent was removed. The residual oil was dissolved in a mixture of $3 \times$ aqueous KOH (5 ml) and 1,4-dioxane (13 ml) and heated at 90 °C for 30 min. After cooling, ice water (60 ml) was added to the reaction mixture was extracted with EtOAc (50 ml×2) and the combined extract was washed with brine, and dried over MgSO₄. Evaporation of solvent and purification of the crude product by recrystalization or column chromatography on silica gel provided sulfonamides.

N-(6-Trifluoromethyl-3-pyridyl)ethanesulfonamide (**3**): White needles, mp 101—102 °C. Recrystallization from a mixture of Et₂O and hexane. ¹H-NMR (300 MHz, CDCl₃) δ: 1.42 (3H, t, *J*=7.5 Hz), 3.24 (2H, q, *J*=7.5 Hz), 5.99 (1H, bs), 7.67 (1H, d, *J*=8.4 Hz), 7.85 (1H, dd, *J*=8.4, 2.1 Hz), 8.53 (1H, d, *J*=2.1 Hz). IR (KBr) cm⁻¹: 1321, 1146. *Anal.* Calcd for $C_8H_9F_3N_2O_2S$: C, 37.79; H, 3.57; N, 11.02. Found: C, 37.73; H, 3.42; N, 10 92

N-[2-(2,4-Difluorophenoxy)-6-trifluoromethyl-3-pyridyl]methanesulfonamide (**4a**): White powder, mp 104—105 °C. Purification by flash chromatography on silica gel eluted with 20% EtOAc in hexane. *Rf*=0.37 (33% EtOAc in hexane). ¹H-NMR(CDCl₃) δ : 3.16 (3H, s), 6.91—7.01 (2H, m), 7.13 (1H, bs), 7.29 (1H, m), 7.44 (1H, d, *J*=8.0 Hz), 8.04 (1H, d, *J*=8.0 Hz). IR (KBr) cm⁻¹: 1331, 1164. *Anal.* Calcd for C₁₃H₉F₅N₂O₃S: C, 42.40; H, 2.46; N, 7.61. Found: C, 42.64; H, 2.45; N, 7.58.

N-[2-(2,4-Difluorophenoxy)-5-trifluoromethyl-3-pyridyl]methanesulfonamide (**5a**): Pale yellow powder, mp 88—90 °C. Purification by flash chromatography on silica gel eluted with 20% EtOAc in hexane. *Rf*=0.17 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 3.16 (3H, s), 6.94—7.03 (3H, m), 7.22 (1H, m), 8.13 (2H, s). IR (KBr) cm⁻¹: 1330, 1139. *Anal.* Calcd for C₁₃H₉F₅N₂O₃S: C, 42.40; H, 2.46; N, 7.61. Found: C, 42.57; H, 2.48; N, 7.63.

N-[2-(2,4-Difluorophenoxy)-6-trifluoromethyl-3-pyridyl]ethanesulfonamide (**4b**): White powder, mp 89—90 °C. Purification by flash chromatography on silica gel eluted with 20% EtOAc in hexane. *Rf*=0.48 (33% EtOAc in hexane). ¹H-NMR (CDCl₃) δ: 1.45 (3H, t, *J*=7.3 Hz), 3.24 (2H, q, *J*=7.3 Hz), 6.92—7.00 (2H, m), 7.01 (1H, bs), 7.27 (1H, m), 7.41 (1H, d, *J*=8.3 Hz), 8.06 (1H, d, *J*=8.3 Hz). IR (KBr) cm⁻¹: 1328, 1162. *Anal.* Calcd for C₁₄H₁₁F₅N₂O₃S: C, 43.98; H, 2.90; N, 7.33. Found: C, 44.00; H, 2.96; N, 7.31.

N-[2-(2,4-Difluorophenoxy)-5-trifluoromethyl-3-pyridyl]ethanesulfonamide (**5b**): White powder, mp 75—77 °C. Purification by flash chromatography on silica gel eluted with 20% EtOAc in hexane.*Rf*=0.34 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ : 1.46 (3H, t, *J*=7.3 Hz), 3.24 (2H, q, *J*=7.3 Hz), 6.95—7.19 (3H, m), 7.24 (1H, m), 8.10 (1H, s), 8.15 (1H, s). IR (KBr) cm⁻¹: 1345, 1138. *Anal*. Calcd for C₁₄H₁₁F₃N₂O₃S: C, 43.98; H, 2.90; N, 7.33. Found: C, 44.07; H, 3.00; N, 7.27.

N-[2-(2,4-Difluorophenoxy)-6-trifluoromethyl-3-pyridyl]-2-naphthalenesulfonamide (**4c**): White powder, mp 90—92 °C. Purification by flash chromatography on silica gel eluted with 10% EtOAc in hexane. *Rf*=0.40 (20% EtOAc in hexane). ¹H-NMR (CDCl₃) δ : 6.83—6.95 (2H, m), 7.01 (1H, m), 7.38 (1H, d, *J*=8.0 Hz), 7.62 (1H, bs), 7.66—7.75 (2H, m), 7.90 (1H, dd, *J*=8.6, 1.8 Hz), 7.95 (1H, d, *J*=8.0 Hz), 8.00 (1H, s), 8.02 (1H, m), 8.11 (1H, d, *J*=8.0 Hz), 8.54 (1H, s). *Anal.* Calcd for C₂₂H₁₃F₅N₂O₃S: C, 55.00; H, 2.73; N, 5.83. Found: C, 55.26; H, 2.66; N, 5.76.

N-[2-(2,4-Difluorophenoxy)-5-trifluoromethyl-3-pyridyl]-2-naphthalenesulfonamide (**5c**): White powder, mp 44—49 °C. Purification by flash chromatography on silica gel eluted with 20% EtOAc in hexane. *Rf*=0.27 (20% EtOAc in hexane). ¹H-NMR (400 MHz, CDCl₃) δ : 6.77—6.98 (3H, m), 7.37 (1H, bs), 7.61—7.70 (2H, m), 7.82 (1H, dd, *J*=8.4, 1.8 Hz), 7.90—8.00 (4H, m), 8.22 (1H, d, *J*=1.8 Hz), 8.45 (1H, s). *Anal.* Calcd for C₂₂H₁₃F₅N₂O₃S: C, 55.00; H, 2.73; N, 5.83. Found: C, 54.75; H, 2.45; N, 5.76.

N-[2-(2,4-Difluorophenoxy)-3-pyridy]ethanesulfonamide (6): White powder, mp 109—110 °C. Purification by flash chromatography on silica gel eluted with 5% Et₂O in hexane. *Rf*=0.18 (20% EtOAc in hexane). ¹H-NMR (300 MHz, CDCl₃) δ: 1.43 (3H, t, *J*=7.5 Hz), 3.18 (2H, q, *J*=7.5 Hz), 6.86 (1H, bs), 6.94—6.98 (2H, m), 7.04 (1H, m), 7.24 (1H, m), 7.86 (1H, m), 7.96 (1H, m). *Anal.* Calcd for C₁₃H₁₂F₂N₂O₃S: C, 49.68; H, 3.85; N, 8.91. Found: C, 49.83; H, 4.14; N, 8.90.

Acknowledgement The authors acknowledge to Drs. T. Haga, T. Koyanagi, S. Mizukoshi, F. Kato, H. Kimura and Mr. S. Yotsuya for their kind suggestions and support on this work.

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