

SCIENCE ()DIRECT.

BIOORGANIC & MEDICINAL CHEMISTRY

Bioorganic & Medicinal Chemistry 11 (2003) 3193-3204

Parallel Liquid Synthesis of N,N'-Disubstituted 3-Amino Azepin-2-ones as Potent and Specific Farnesyl Transferase Inhibitors☆

Thierry Le Diguarher,^a Jean-Claude Ortuno,^a Gilbert Dorey,^a David Shanks,^a Nicolas Guilbaud,^{b,†} Alain Pierré,^b Jean-Luc Fauchère,^a John A. Hickman,^b Gordon C. Tucker^b and Patrick J. Casara^{a,*}

> ^aDepartment of Medicinal Chemistry, Institut de Recherches Servier, 125 chemin de Ronde, 78290 Croissy sur Seine, France ^bDepartment of Experimental Oncology, Institut de Recherches Servier, 125 chemin de Ronde, 78290 Croissy sur Seine, France

> > Received 31 January 2003; accepted 25 March 2003

Abstract—A rapid structure–activity study was performed by parallel liquid synthesis on N,N'-disubstitution of 3-amino azepin-2-one to afford potent and specific farnesyl transferase inhibitors with low nM enzymatic and cellular activities. The activities of the selected compounds were validated in vivo, and compounds **41a** and **44a** presented significant antitumour activity. \bigcirc 2003 Elsevier Science Ltd. All rights reserved.

Introduction

Protein farnesyl transferase (FTase) is a zinc-dependent enzyme that catalyses the attachment of a farnesyl lipid group to the sulfur atom of a cysteine residue of numerous proteins involved in cell signalling, including the oncogenic H-Ras protein.¹ This lipid side chain is critical for the cell membrane anchoring of these proteins,² and therefore FTase inhibition has been recognized as a valuable antitumour therapeutic approach,³ although the target proteins whose farnesylation is inhibited by FTase inhibitors are not all identified. Another more ubiquitous prenylation of proteins can occur via geranylgeranyl transferase (GGTase-1) processing.¹ Indeed the RhoB protein, another putative FTase target, may be geranylgeranylated to an inactive form after FTase inhibition.⁴ The issue of selectivity of the inhibition of FTase versus GGTase has been addressed and the optimal degree of selectivity is still

debatable. However, access to specific inhibitors would help to understand the relationship between the various enzymes and different physiological pathologies. The substrates of these enzymes present a consensus sequence of four amino acid residues (CAAX) at the carboxylic terminal of the protein, with the cysteine in fourth, two aliphatic residues and the terminal residue X, being preferably methionine for FTase and leucine for GGTase.⁵ In the design of FTase inhibitors, histidine was found to be a good bioisosteric replacement of the cysteine residue, and numerous non-thiol CVIM (1) peptidomimetics were developed around an imidazol function.⁶ Therefore, among the compounds in clinical development we can distinguish those containing an imidazole associated to a benzodiazepine (BMS-214662);⁷ a piperazinone (L-778,123),⁸ or a quinolinone (R115777)⁹ scaffold and the tricyclic SCH66336,10 containing a piperidine obtained from HTS. Crystal structures of the complex between FTase and 1 are in favour of an extended conformation of **1** in the active site cavity.¹¹ The caprolactam derivatives 2 are proposed as the core of constrained CVIM bioisosteres in an elongated conformation including an imidazole zinc-complexing function. This scaffold can be substituted by R_3 (and/or eventually R₂) on the amine function N-1 (and/or eventually N-4) to mimic part of the Ras protein R (and/or

^{*}Supplementary data associated with this article can be found at doi:10.1016/S0968-0896(03)00218-9 [Tables 7–9].

^{*}Corresponding author. Tel.: +33-1-5572-2361; fax: +33-1-5572-2470; e-mail: patrick.casara@fr.netgrs.com

[†]Present address: Onco*design*, Parc Technologique Toison d'Or, 28 rue Louis de Broglie, 21000 Dijon, France.



Figure 1. Interactions of compound 1 with FTase and structures of compounds 1 and 2.



Scheme 1. Synthesis of compounds **28–63**. (i): (Boc)₂O, NaOH, H₂O/tBuOH, rt, 12 h, (**4**, 88%); (ii₁) (a) BiPh₃, pyridine, Cu(OAc)₂, CH₂Cl₂, rt, 24 h (**5**, 77%); (ii₂) (a) NaH THF/DMF, 0 °C, 30 min; (b) PhCH₂Br, N(Bu)₄, I, rt, 24 h (**6**, 97%). (ii₃) (a) NaH, DMF, 0 °C, 30 min; (b) Ph(CH₂)₂OTs, 0 °C, 12 h (**7**, 75%); (ii₄) (a) NaH DMF, 0 °C, 30 min, rt, 2 h; (b) Ph(CH₂)₃OTs, rt, 12 h (**8**, 86%); (ii₅) (a) LiHMDS, 0 °C, 15 min; (b) PhSO₂Cl, 0 °C, rt, 2 h (**9**, 77%); (ii₆) (a) PhSCH₂Cl, KOH, N(Bu)₄I, THF, rt, 12 h; (b) dimethyldioxirane, acetone, 0 °C, rt, 2 h, (**10**, 40%); (iii) TFA, CH₂Cl₂, 0 °C, rt, 2 h or 4 N HCl/dioxane, rt (**11–16**); (iv) (a) dihydroxyacetone dimer, KSCN, MeCN/EtCO₂H, 60 °C, 2 h; (b) H₂O₂, H₂O, 40 °C, 2 h; (c) DMSO, SO₃/pyridine, NEt₃, rt, 2 h (**17–27**: Tables 4 and 6); (v) (a) **17–27**, NaBH(OAc)₃, CH₂Cl₂, 0 °C, rt, 12 h; (b) HCl, ether, or fumaric acid, EtOH (**28–33**: Tables 1; **34–53**: Tables 2 and 7; **54** to **63**: Tables 4 and 8).

the farnesyl co-substrate) and R_1 separated by a spacer Y on the lactam nitrogen atom N-7 to replace the lipophilic non polar side chains of the isoleucine/methionine residue. Moreover, a synthesis of **2** by a methodology applicable to liquid parallel synthesis was performed to allow an extensive and rapid investigation of the $Y/R_1/R_2/R_3$ parts of the molecule (Fig. 1).

Results and Discussion

Investigation on the N-lactam substitution R₁

A straightforward retro-synthetic analysis links 2 to 2-aminocaprolactam (3) by two selective N-alkylations. The primary amino function of **3** was protected as a Boc derivative to give the intermediate 5 ready for *N*-lactam substitution and further substitution after deprotection. In order to validate the caprolactam core hypothesis and investigate the N-lactam substitution, the proximal N-p-cyanobenzyl function (PCB) was used as the imidazole substituent R₃, as a well described cysteine bioisostere.¹² Then the replacement of the isoleucine/ methionine by a lipophilic side chain was first considered by using various spacer Y and a phenyl group as R_1 . When Y was a bond, the *N*-arylation was achieved by treatment of 4 with triphenylbismuth to give 5.¹³ The *N*-alkylation was obtained by lactam deprotonation and addition of the benzyl chloride to give 6. Based on the same strategy, various alkyl spacers were also considered to complete the study to give the corresponding compounds 7 and 8. A sulfonyl spacer was also introduced either by lactam deprotonation and addition of the corresponding sulfonyl chloride to give 9, or by sequential alkylation of the chloromethyl sulfide and oxidation to give 10. Then, they were deprotected quantitatively under acidic conditions to give the corresponding primary amines 11–16, and the (N-1-PCB) imidazol-5-yl-methyl substituent was introduced by reductive amination with the corresponding aldehyde 17 (R_3 =PCB), to afford the compounds 28 to 33 (Scheme 1, Table 1).

The inhibitory activities of these compounds were examined against FTase and against GGTase-1 and the results were compared to the various clinical references (Table 1). Their cellular activities for FTase inhibition were determined on an H-Ras based assay.

In the series of compounds 28-33, the introduction on the caprolactam moiety of an aryl substituent to replace the two terminal amino acids of CVIM afforded very potent inhibitors of FTase when the junction was made either by a bond or a methylene group in compounds 28 and 29, respectively. Further chain elongation decreased the potency in the case of an ethyl group (30) and even abolished the activity in the case of the propyl spacer (31). The sulfone spacer, previously used in the case of diaryl as VIM surrogate,¹⁴ was detrimental to inhibition in our case, either alone (32) or with a methyl group (33). These results suggest a different binding mode for the caprolactam scaffold, possibly in a conformation closer to that of the peptide ligand. The stereospecificity of the enzyme inhibition was validated by using the enantiomers of **29**, respectively **29a** and **29b**, separated by chiral HPLC, 29a being significantly more active than

Table 1. Structures and biological data of compounds 28–33



Compd (conf.)	Y	FTase IC ₅₀ nM	GGTase IC ₅₀ nM	Ras Cell IC ₅₀ nM	Compd (conf.)	Y	FTase IC ₅₀ nM	GGTase IC ₅₀ nM	Ras Cell IC ₅₀ nM
BMS-214662 ⁷ found		1.3 16	2300 7600	25 29	29a (S) 29a (S) ^a	CH ₂	9 5	16,200	10
L-778,123 ⁸ found		1.6 8	13,200	70 31	29b (<i>R</i>) ^a	CH_2	2800	12,000	ND
SCH66336 ¹⁰		1.9	> 50,000	10	30 (<i>RS</i>)	$(CH_2)_2$	69	5200	288
R115777 ⁹ found		$\begin{array}{c} K_{\mathrm{i}} \ 0.5 \\ 4 \end{array}$	> 50,000	1.7 2	30a (S)	(CH ₂) ₂	39	5900	160
28 (RS)	Bond	28	11,200	32	31 (<i>RS</i>)	(CH ₂) ₃	1000	6200	4400
28a (S)	Bond	17	3300	17	32 (<i>RS</i>)	SO_2	293	10,600	260
29 (RS)	CH_2	8	10,600	ND	33 (<i>RS</i>)	CH_2SO_2	168	9700	3800

^aEnantiomer obtained from the racemate by separation on chiral HPLC.

29b. To assess the C-5 configuration, a stereospecific synthesis of the S enantiomer for the most active compounds was carried out from the S caprolactam **3a**. The retention of the configuration during the synthesis was confirmed by the synthesis of the S enantiomer of **28**, **29** and **30**, respectively **28a**, **29a** and **30a**, with an enantiomeric excess up to 85%. The conservation of activity was observed for each of these S enantiomers versus their racemates, which corroborated the stereospecificity of the enzymatic and the cellular inhibition, with activities already in the same range as those of the reference compounds (Table 1). The active configuration S of these inhibitors is also in agreement with the hypothesis of a close mimicry of the peptide ligand.

Further investigation on the primary amine substitution R₁

Since the best enzymatic and especially cellular activities were observed for compounds 28a and 29a, they could be considered for further lead optimization, with a slight advantage for the latter. Moreover, the benzyl derivative 28a was also the most attractive in terms of molecular diversity for R_1 optimization, with a large commercial access to benzylic reagents compared to triarylbismuth derivatives. Therefore, solution phase parallel synthesis, was performed by sequential alkylation of the S enantiomer 4a with a series of benzyl derivatives to give, after acidic deprotection, the amines. These intermediates were condensed, without purification, by reductive amination with 17, to afford compounds 34a to 53a. Purification of these compounds by simple automated SPE, led average yield, but complied with the high criteria purity necessary for the accuracy of the biological tests (Scheme 1, Table 2).

In this second selection, the majority of the random substitution of the aromatic moiety in **29a** maintained

or even improved the enzymatic activity with an IC₅₀ in the low nM range (Table 2). Some exceptions included its replacement by a 4-pyridyle or substitution by a para trifluoromethyl group (compounds 35a and 38a, respectively) with a 10-fold decrease in potency and the *para* substitution by an ester leading to an inactive derivative (compound 37a). However, the cellular assay was even more discriminating, with a 3- to 5-fold decrease in potency between enzymatic and cellular activities for a naphtyl, cyclohexyl and dimethyl substitution (compounds 34a, 36a, 40a, respectively) and more than a 10-fold decrease in potency for para and poly substitution by various halogen atoms (compounds 43a and 46a–53a). No simple explanation to this phenomenon could be provided, but differences in cellular potency versus the isolated enzyme may originate at the level of membrane crossing, protein binding or chemical stability. However, the ortho or meta mono and disubstitution by a methyle or an halogen maintained the cellular activity at the low nM range IC₅₀ (compounds **39a**, **41a**, **42a**, **44a**, **45a** and **51a**).

Investigation on the amine substitution R_{2xv}

In order to investigate the importance of the N-1 substituent on the imidazole, the unsubstituted derivative **65a** ($R_2 = R_3 = H$) was synthesized by reductive amination of the benzyl caprolactam **12a** with the 1-trityl-4carboxaldehyde imidazole. Since this compound was devoid of any enzyme inhibitory activity (Table 3), the bis PCB ($R_2 = R_3 = PCB$) **66a**, and the mono PCB derivatives ($R_2 = H$, $R_3 = PCB$) **67a** were synthesized by mono and dialkylation of **64a**. Surprisingly, these compounds were equipotent, but with an IC₅₀ one order of magnitude higher that than of **29a.** To assess the potential of this substitution R_2 , a series of N-4 substituted derivatives of **65a** was rapidly generated by the

Table 2. Structures and biological data of compounds 34a–53a



Compd	R ₁	FTase IC ₅₀ (nM)	GGTase IC ₅₀ (nM)	Ras cell IC ₅₀ (nM)	Compd	R ₁	FTase IC ₅₀ (nM)	GGTase IC ₅₀ (nM)	Ras cell IC ₅₀ (nM)
29a	E	9	16,200	10					
34a	H	43	2500	156	44a	Br	3	5600	2
35a	н N	87	20,500	10 ³	45a	F	5	19,800	4
36a		13	18,400	70	46a	F	2	5100	64
37a	ECO2Me	9000	3600	> 10 ⁴	47a	CI T	9	ND	160
38a	CF ₃	113	4200	350	48 a	CI	20	ND	370
39a	н	5	5500	4	49a	F F	2	ND	120
40a	80:20 u	7	24,500	46	50a	F	2	9300	44
41 a	CI	3	7400	7	51a	F	1	25,800	14
42a	Cl	6	10,500	2	52a	F F F	5	ND	210
43a	L. CI	19	ND	184	53a	F F F F F	41	ND	465
						Г Р Г			

Table 3.	Biological	data o	of compounds	65a-96a
----------	------------	--------	--------------	---------

Compd	FTase IC ₅₀ (nM)	Compd	FTase IC ₅₀ (nM)
65a	>104	67a	255
66a	298	68a-96a	>104

parallel liquid phase procedure by either aminoreduction with various aldehydes to give compounds **67a** to **82a**, or by *N*-acylation with various acylchlorides to give compounds **83a** to **96a** (Scheme 2, Table 3).

In this third selection, the random amino substitution of **65a** was produced, but did not allow recovery of any significant FTase inhibitory activity (compounds **68a**–**96a**, Table 3). These data also diverged from the results



Scheme 2. Synthesis of compounds 65a-96a. (i) NaBH(OAc)₃, CH₂Cl₂, 0 °C to rt 12 h, (64a, 88%); (ii) (a) TFA, Et₃SiH, CH₂Cl₂, rt, 12 h; (b) fumaric acid, EtOH (65a); (iii) (a) pCNPhCH₂Br, AcOEt, 60 °C, 12 h; (b) MeOH, 60 °C, 12 h; (c) fumaric acid, EtOH (66a); (iv₁) (a) NaBH(OAc)₃, R₂CHO, CH₂Cl₂, rt 12 h; (b) TFA, Et₃SiH, CH₂Cl₂, rt, 12 h (67a-82a: Fig. 2, Table 9); (iv₂) (a) R₂COCl polyvinylpyridine, CH₂Cl₂, rt 12 h; (b) TFA, Et₃SiH, CH₂Cl₂, rt 12 h; (c) HCl, ether, or fumaric acid, EtOH (83a-96a: Fig. 2, Table 9).



Figure 2. Substitutents of componud 67a–96a R₃=H R₂=

reported with the aryl scaffold where the N substitution of the imidazol-5-yl-methylamine function either on the imidazole ring or the amine function afforded equipotent compounds.¹² In this case, the authors proposed a nearby second binding pocket at this cysteine region. A plausible explanation, in absence of X-ray data, is a different recognition of the two scaffolds, the caprolactam following the peptide side chain and the aryl being perpendicular allowing the fit of substituents in the various amino acid pocket sites of the enzyme.

Investigation on the imidazol ring substitution R₃

In the fourth selection, the effect of the modification of the N-1 substitution of the imidazole side chain on **39a** was investigated. For this purpose, a series of N-1 substituted imidazol-5-ylcarboxaldehydes **18** to **27** were synthesized from various primary amines by a modification of the method described by Aulaskari.¹⁵ They were then condensed by reductive amination with the amine intermediate precursor of **39a**, to afford compounds **54a** to **63a** (Scheme 1, Table 4).

In this selection, the removal of the cyano substituent or the reduction of the aromatic ring (compounds **54a** and 55a, respectively) led to a 10-fold loss of enzymatic inhibition and a 100-fold loss in cellular potency compared to 39a. In addition, the spacer elongation by one or two carbons, led to inactive compounds despite further substitution (compounds 56a, 57a and 61a-63a). The replacement of the cyano substituent by various halogen (compounds 58a-60a) kept the enzymatic activity almost at the same level as those of 39a, but induced a one order of magnitude loss in cellular potency in spite of an increase in their CLogP. Therefore, no better substitution for the PCB group was found so far.

Pharmacological Evaluation

As a preliminary evaluation of antitumour activity, selected compounds, with low nM IC₅₀ in the enzymatic and cellular assays, were tested in an in vivo model. H-Ras transfected fibroblasts were grafted sc into nude mice, and compounds were administered at 100 and 200 mg/kg po for 5 days, starting from day 6 after tumour implantation. Among these compounds, **41a**, **42a** and **44a** showed significant activity with a marked reduction of the tumour volume (39, 85 and

Table 4. Structures of compounds 17-27 and 54a-63a and biological data of compounds 54a-63a



Compd	Compd	R ₃	FTase IC ₅₀ (nM)	Ras cell IC ₅₀ (nM)	Compd	Cpmpd	R ₃	FTase IC ₅₀ (nM)	Ras cell IC ₅₀ (nM)
17	39a	L CN	5	4					
18	54a	H	48	430	23	59a	" Cl	16	180
19	55a	ж. СN	81	730	24	60a	s Br	27	190
20	56a	#	3000	ND	25	61a	La Cl	3800	ND
21	57a	#	1700	ND	26	62a	[∎] ↓ 2 Br	5300	ND
22	58a	щ. F	19	210	27	63a	Me OMe	2700	ND

40%, respectively, at 200 mg/kg). However, these activities remained lower than those of the reference compound BMS-214662 which inhibited tumour growth by 50 and 80% at 100 and 200 mg/kg, respectively, in this model. Moreover, since signs of toxicity occurred with **42a** at the higher dose, only

Table 5. In vivo antitumour activity

No.	% Control 100 mg/kg po	% Control 200 mg/kg po		
BMS-214662	48	22		
29a	83	72		
41a	66	61		
42a	51	15 (Tox.)		
44a	99	60		
45a	103	71		
51a	126	76		

Table 6. Yields and analytical data of compounds 18-27

41a and **44a** were selected for further pharmacological evaluation (Table 5).

Conclusion

An efficient synthetic access was developed from N, N'disubstituted 2-amino caprolactam to generate a new class of inhibitors of FTase selective versus GGTase-1. A rapid lead optimization by liquid parallel synthesis allowed independent investigation of the substituents and afforded very potent compounds with low nM IC₅₀ in the enzymatic and cellular assays. It is interesting to note that a differently substituted caprolactam scaffold was previously used to design FTase inhibitors as reverse-turn mimics, but led to almost inactive compounds (IC₅₀ in the μ M range).¹⁶ The activities of the selected compounds were validated in vivo, and compounds **41a** and **44a** presented significant antitumour activity.

Table 0.	The of the stand and should be compounds to 27									
Compd	18	19	20	21	22	23	24	25	26	27
R ₃	a l	u CN	¤[]₂	[₩] () ₃	» F	u Cl	n Br	R Cl	r-Br	n 1 2 OMe
Yield%	33	27	60	37	51	46	52	18	32	26
Analysis	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR	RMN/IR

Experimental

Chemistry general techniques

All reactions were carried out under nitrogen or argon atmosphere under anhydrous conditions unless otherwise noted.

Parallel synthesis was performed on STEM Blocks, Workup was performed on Zymate. Yields refer to chromatographically and spectroscopically (¹H NMR) homogenous material unless otherwise stated. All reagents were purchased at highest commercial quality and used without further purification unless otherwise stated. All reactions were monitored by thin-layer chromatography carried out on 0.2 mm Merck silicagel plates using UV light and ethanol phosphomolybdic acid or *p*-anisaldehyde and heat as developing agent.

Preparative flash chromatography separations were carried out on Kiesegel 60 (0.04–0.063 mm) Merck silicagel. Normal-phase solid-phase extraction (SPE) were carried out on cartridges loaded with 8 g of Kiesegel 60 $(0.04-0.063 \times \text{mm})$ Merck silicagel. Reversed-phase chromatography were carried out on columns loaded with of Kiesegel C18 (LiChroprep RP-18; 25–40 µm) Merck. Mobile phase $H_2O/CH_3CN + 0.5\%$ TFA. Reverse-phase HPLC analysis was performed on an Agilent 1100 instrument using a Waters Xterra MS C18 column with detection at 210 nm using a $H_2O/$ CH₃CN+0.1% CH₃SO₃H gradient over 10 min. NMR spectra were recorded on a Brucker PPX 200 instrument as indicated, calibrated using TMS as an internal reference. The following abbreviations were used to explain multiplicities: s, singlet; d, doublet; t, triplet; q, quadruplet; m, multiplet; b, broad. IR spectra were recorded on a Brucker Vector 22 spectrophometer.

Electrospray mass spectra were recorded in a positive mode on a Finnigan TSQ 7000 spectrophometer, by infusion at $15 \,\mu$ L/min of a 0.1 mg/mL sample solution in a mixture of CH₃CN/H₂O (75:25).

Preparation of (*R*,*S*) **1-phenyl-2-oxo-3-(***tert***-butoxycarbonylamino)-azepine (5).** To a solution of **4** (1.6 g, 7 mmol), Cu(OAc)₂ (1.36 g, 7 mmol) and pyridine (0.6 mL, 7 mmol) in CH₂Cl₂ (20 mL) was added dropwise triphenylbismuth (3.4 g, 7.6 mmol), at room temperature. The reaction mixture was stirred 24 h and then filtrated over Celite and washed several times with AcOEt. The filtrate was concentrated in vacuo, and the residue was purified by flash chromatography (silica gel, petroleum ether/AcOEt; 60:40) to give **5** (0.84 g, 38%) as an oil. ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.73 (m, 5H, *Ph*), 3.85–3.65 (m, 3H, *CH*₂N, NC*H*CO), 2.2– 1.7 (m, 6H, (*CH*₂)₃), 1.5 (s, 9H, *tBu*). IR: *V*_{max} 3375– 3317, 1647 cm⁻¹

Preparation of (S) 1-phenyl-2-oxo-3-(*tert*-butoxycarbonylamino)-azepine (5a). Compound 5a (1.2 g, 43%) was prepared by the same procedure used for 5 with compound 4a (2 g, 8.7 mmol), Cu(OAc)₂ (1.7 g, 8.7 mmol) and pyridine (0.7 mL, 8.7 mmol) and triphenylbismuth (4.2 g, 9.6 mmol) in CH₂Cl₂ (22 mL). RMN; IR similar to 5. Preparation of (R,S) 1-benzyl-2-oxo-3-(tert-butoxycarbonylamino)-azepine (6). To a suspension of NaH (60% in oil, 1.9 g, 47 mmol) in THF (20 mL) was added a solution of compound 4 (9.8 g, 43 mmol) in THF (90 mL) at room temperature. Then this mixture was stirred for 1 h and a solution of benzyl bromide (7.6 mL, 64 mmol) in THF (20 mL) was added. After an additional stirring of 2h the reaction mixture was cooled down to 0 °C, treated with a satured solution of NH₄Cl and extracted with AcOEt. The organic phase was dried over Na₂SO₄, filtrated and concentrated in vacuo. The residue was purified by flash chromatography (silica gel, heptane/AcOEt; 70:30) to give compound 6 as a solid (12.7 g, 93%). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.25 (m, 5H, Ph), 4.85 (d, 1H, CHPh), 4.4 (m, 2H, CHPh, CHNH), 3.6–3.15 (2dd, 2H, CH₂N), 2.2–1.8 (m, 6H, (CH₂)₃) 1.5 (s, 9H, tBu). IR: V_{max} 3413, 1711, 1648 cm^{-1} . Anal. (C₁₈H₂₆ N₂O₃): C, H, N calcd: 67.90; 8.23; 8.80. Found: 67.82; 8.21; 8.81.

Preparation of (S) 1-benzyl-2-oxo-3-(*tert*-butoxycarbonylamino)-azepine (6a). Compound 6a (0.615 g, 94%) was prepared by the same procedure used for 6 with compound 4a (0.5 g, 2.2 mmol), NaH (60% in oil, 0.096 g, 2.4 mmol) and benzyl bromide (0.46 mL, 3.6 mmol) in THF (5 mL). RMN; IR similar to 6.

Preparation of (*R*,*S*) 1-(2-phenylethyl)-2-oxo-3-(*tert*butoxycarbonylamino)-azepine (7). Compound 7 (0.45 g, 41%) was prepared by the same procedure used for **6** with compound **4** (1 g, 4.3 mmol), NaH (60% in oil, 0.2 g, 4.8 mmol) and 2-phenylethyltosylate (1.3 g, 4.8 mmol) in DMF (5 mL). ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.3 (m, 5H, *Ph*), 4.25 (m, 1H, *CH*NH), 3.5–3.25 (m, 4H, *CH*₂N*CH*₂), 2.7 (d, 2H, *CH*₂Ph), 2.1–1.7 (m, 6H, (*CH*₂)₃) 1.4 (s, 9H, *tBu*). IR: V_{max} 3422, 1703, 1640 cm⁻¹

Preparation of (S) 1-(2-phenylethyl)-2-oxo-3-(*tert***-butoxy-carbonylamino)-azepine (7a).** Compound **7a** (5.1 g, 47%) was prepared by the same procedure used for **6** with compound **4a** (7.4 g, 32.3 mmol), NaH (60% in oil, 1.4 g, 35 mmol) and 2-phenylethyltosylate (10.7 g, 38.6 mmol) in DMF (130 mL). RMN; IR similar to **7**.

Preparation of (R,S) 1-(3-phenylpropyl)-2-oxo-3-(*tert*butoxycarbonylamino)-azepine (8). Compound 8 (2.6 g, 86%) was prepared by the same procedure used for 6 with compound 4 (2 g, 8.7 mmol), NaH (60% in oil, 0.384 g, 9.5 mmol) and 3-phenylpropyltosylate (3 g, 10.3 mmol) in DMF (43 mL). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.4 (m, 5H, *Ph*), 4.4 (dd, 1H, *CH*NH), 3.8–3.3 (m, 4H, *CH*₂N*CH*₂), 2.7 (d, 2H, *CH*₂Ph), 2.1– 1.7 (m, 6H, 2(*CH*₂)₂) 155 (s, 9H, *tBu*). IR: *V*_{max} 3411, 1711, 1642 cm⁻¹.

Preparation of (*R*,*S*) 1-(benzenesulfonyl)-2-oxo-3-(*tert*butoxycarbonylamino)-azepine (9). Compound 9 (6g, 77%) was prepared by a similar procedure used for 6 with compound 4 (5g, 21.8 mmol), LiHMDS (1.3 M in THF, 20 mL, 26 mmol) and benzenesulfonyl chloride (3.3 mL, 26 mmol) in THF (50 mL) at 0 °C. ¹H NMR (200 MHz, DMSO- d_6): δ (ppm) 7.95–7.55 (m, 5H, *Ph*), 4.5–4.3 (m, 2H, *CHa*N, NC*H*CO), 3.75–4.3 (dd, 1H, *CHa'*N), 2.0–1.4 (m, 6H, (*CH*₂)₃), 1.35 (s, 9H, *tBu*). IR: V_{max} 3426, 1702 cm⁻¹.

Preparation of (R,S) 1-(benzenesulfonylmethyl)-2-oxo-3-(tert-butoxycarbonylamino)-azepine (10). Preparation of (R,S)1-(benzenesulfinylmethyl)-2-oxo-3-(tert-butoxycarbonylamino)-azepine. To a mixture a solution of compound 4 (1 g, 4.36 mmol), phenylsulfinylmethyl chloride (1.7 mL, 13 mmol) and NBu₄I (0.24 g, 0.64 mmol) in THF (10 mL) was added KOH in powder (0.73 g, 13 mmol) at room temperature. Then this mixture was stirred overnight at room temperature and treated with a saturated solution of NH₄Cl and extracted with AcOEt. The organic phase was dried over Na₂SO₄, filtrated and concentrated in vacuo. The residue was purified by flash chromatography (silica gel, petroleum ether/AcOEt; 85:15) to give the title compound as an oil (0.745 g, 49%). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.8–7.4 (m, 5H, Ph), 5.1–4.8 (AB, 2H, SCH₂), 4.25 (m, 1H, CHNH), 3.55 (m, 2H, CH₂N), 2.7 (d, 2H, CH_2Ph), 1.8 (m, 6H, $(CH_2)_3$), 1.3 (s, 9H, tBu). IR: V_{max} 3415, 1711, 1654 cm⁻¹.

Preparation of (10). To a solution of compound described in step a (0.46 g, 1.3 mmol), in acetone (3 mL) was added dropwise a solution of dimethyldioxirane (0.05 M, in acetone, 34 mL, 1.7 mmol) at 0 °C. The reaction mixture was stirred 2 h at room temperature and then concentrated in vacuo to give **10** (0.32 g, 81%) as a white solid. ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.8–7.4 (m, 5H, *Ph*), 5.1–4.8 (AB, 2H, SCH₂), 4.25 (m, 1H, CHNH), 3.55 (m, 2H, CH₂N), 2.7 (d, 2H, CH₂Ph), 1.8 (m, 6H, (CH₂)₃), 1.3 (s, 9H, *tBu*). IR: *V*_{max} 3408, 1707, 1653, 1383, 1148 cm⁻¹. Anal. (C₁₈H₂₆ N₂O₅S): C, H, N, S calcd: 56.53; 6.85; 7.32, 8.38. Found: 56.53; 6.86; 7.04, 8.21.

Preparation of (R,S) 1-phenyl-2-oxo-3-amino-azepine (11). To a solution of 5 (0.69 g, 2.3 mmol) in CH₂Cl₂ (14 mL) was added dropwise trifluoroacetic acid (1.7 mL) at room temperature. The reaction mixture was stirred 2 h and then concentrated in vacuo. The residue was purified by flash chromatography (silica gel, CH₂Cl₂/MeOH; 95:5) to give 11 (0.32 g, 70%) as an oil. ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.3 (m, 5H, *Ph*), 3.85–3.65 (m, 3H, *CH*₂N, NCHCO), 2.2–1.7 (m, 6H, (CH₂)₃). IR: V_{max} 3375–3317, 1647 cm⁻¹.

Preparation of (S) 1-phenyl-2-oxo-3-amino-azepine (11a). Compound 11a (0.53 g, 69%) was prepared by the same procedure used for 11 with compound 5a (1.2 g, 3.9 mmol) and trifluoroacetic acid (3 mL). RMN; IR similar to 11.

Preparation of (R,S) **3-amino-2-oxo-1-benzyl-azepine** (12). To a solution of **6** (12.7 g, 40 mmol) in dioxane (20 mL) was added dropwise a solution of 4 N HCl in dioxane (100 mL) at room temperature. The reaction mixture was stirred 3 h and then concentrated in vacuo. The residue was triturated in ether to give 12 (10.6 g, 100%) which was used for the next step without further purification. ¹H NMR (200 MHz, DMSO- d_6): δ (ppm) 7.3 (m, 5H, *Ph*), 4.7–4.5 (2d, 2H, *CH*₂Ph), 4.35 (m, 1H, *CH*NH), 3.6–3.3 (m, 2H, *CH*₂N), 2.2–1.2 (m, 6H, (*CH*₂)₃). IR: *V*_{max} 3414, 2937, 1657 cm⁻¹. Anal. (C₁₃H₁₈ N₂O) HCl, 2H₂O: C, H, N calcd: 53.70; 9.63; 11.9. Found: 53.14; 7.94; 9.2.

Preparation of (S) 1-benzyl-2-oxo-3-amino-azepine (12a). Compound 12a (5.05 g, 95%) was prepared by the same procedure used for 11 with compound 6 (7.74 g, 24.3 mmol) and trifluoroacetic acid (19 mL). RMN; IR similar to 12.

Preparation of (*R*,*S*) 1-(2-phenylethyl)-2-oxo-3-amino azepine (13). Compound 13 (1.4 g, 95%) was prepared by the same procedure used for 11 with compound 7 (2 g, 6.05 mmol) and trifluoroacetic acid (4.7 mL). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.3 (m, 5H, *Ph*), 4.15 (m, 1H, *CH*NH), 3.4–3.25 (m, 4H, *CH*₂N*CH*₂), 2.6 (d, 2H, *CH*₂Ph), 2–1.3 (m, 6H, (*CH*₂)₃). IR: *V*_{max} 3362, 3297, 1645 cm⁻¹.

Preparation of (S) 1-(2-phenylethyl)-2-oxo-3-amino-azepine (13a). Compound 13a (3.2 g, 91%) was prepared by the same procedure used for 11 with compound 7a (5.1 g, 15.2 mmol) and trifluoroacetic acid (12 mL). RMN; IR similar to 13. Anal. ($C_{14}H_{20}$ N₂O) 0.25H₂O: C, H, N calcd: 70.93; 8.65; 11.82. Found: 70.71; 8.53; 11.49.

Preparation of (*R*,*S*) 3-amino-2-oxo-1-(3-phenylpropyl)azepine (14). Compound 14 (1.2 g, 100%) was prepared by the same procedure used for 11 with compound 8 (1.3 g, 3.74 mmol) and trifluoroacetic acid (2.9 mL). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.25 (m, 5H, *Ph*), 4.1 (dd, 1H, *CH*NH), 3.7–3.3 (m, 4H, *CH*₂N*CH*₂), 2.55 (d, 2H, *CH*₂Ph), 1.9–1.25 (m, 6H, 2(*CH*₂)₂). IR: *V*_{max} 3500– 3000, 1668 cm⁻¹.

Preparation of (*R*,*S*) 3-amino-2-oxo-1-(benzenesulfonyl)azepine (15). Compound 15 (3.45 g, 79%) was prepared by the same procedure used for 11 with compound 9 (6 g, 16.2 mmol) and trifluoroacetic acid (12.5 mL). ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.95–7.6 (m, 5H, *Ph*), 4.4 (m, 1H, *CHa*N, NC*H*CO), 3.95 (d, 1H, NC*H*CO), 3.7 (dd, 1H, *CHa'*N), 2.1–1.8 (m, 6H, (*CH*₂)₃). IR: V_{max} 3600–2300, 1700 cm⁻¹.

Preparation of (*R***,***S***) 3-amino-2-oxo-1-(benzenesulfonylmethyl)-azepine (16).** Compound **16** (0.31 g, 100%) was obtained as a yellow gum by the same procedure used for **11** with compound **10** (0.4 g, 1.04 mmol) and trifluoroacetic acid (0.8 mL). ¹H NMR (200 MHz, CDCl₃): δ (ppm) 7.85 (d, 2H, o-*Ph*), 7.75–7.5 (m, 3H, m, p-*Ph*), 5.0 (AB, 2H, SCH₂), 3.7 (m, 1H, CHNH), 3.5–3.3 (m, 2H, CH₂N), 2.7 (d, 2H, CH₂Ph), 1.8–1.2 (m, 6H, (CH₂)₃). IR: V_{max} 3367–3100, 1670 cm⁻¹.

Preparation of [3-(4-cyanobenzyl)-3-H-imidazol-4-yl]carboxaldehyde (17). Step 1: Preparation of [3-(4-cyanobenzyl)-3-H-2-thioimidazol-4-yl]-methyl alcohol. To a mixture of 4-cyanobenzylamine (18 g, 100 mmol), and propionic acid (30 mL) in H₂O/CH₃CN (97:3, 300 mL) was added sequentially KSCN (11 g, 115 mmol) and dihydroxyacetone (dimer, 11.2 g, 115 mmol). This mixture was stirred for 2 h at 60 °C and overnight at room temperature. Then the precipitate was filtrated off and was washed with CH₃CN and then H₂O. The solid was dried in vacuo to give the title compound (24 g, 98%) which was used for the next step without further purification. ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 12.2 (s, 1H, *SH*), 7.8–7.35 (2d, 4H, *Ph*), 6.9 (s, 1H, *NCHN*), 5.4 (s, 2H, Ph*CH*₂N), 4.15 (s, 1H, *CH*₂O), 5.25 (s, 2H, *OH*). IR: *V*_{max} 3600–3200, 2300, 2222, 1025 cm⁻¹. MH⁺(246).

Step 2: Preparation of [3-(4-cyanobenzyl)-3-H-imidazol-4-yl]-methyl alcohol. To a solution of the thioalcohol obtained in step 1 (23 g, 100 mmol) in diluted acetic acid $(H_2O/AcOH; 80:20, 250 \text{ mL})$ was added dropwise an aqueous solution of H_2O_2 (50%, 20 mL, 700 mmol) during 10 min and the temperature was maintained below 40 °C. Then this mixture was stirred for 3 h at room temperature and treated with an aqueous solution of Na₂SO₃ (20%, 450 mL) and filtrated. The filtrate was neutralized with aqueous ammonia (5%) and extracted with AcOEt. The organic phase was dried over Na_2SO_4 , filtrated, concentrated in vacuo and dried to give the title compound (14 g, 74%) which was used for the next step without further purification. ¹H NMR (200 MHz, DMSO-d₆): δ (ppm) 7.75-7.3 (2d, 4H, Ph), 7.7 (s, 1H, NCH=N), 6.85 (s, 1H, NCH=C), 5.35 (s, 2H, PhCH₂N), 5.15 (s, 2H, OH), 4.3 (s, 1H, CH₂O). IR: $V_{\rm max}$ 3200, 2232, 1025 cm⁻¹.

Step 3: Preparation of 17. To a solution of the alcohol obtained in step 2 (10 g, 47 mmol) and TEA (26 mL, 200 mmol) in DMSO (230 mL) was added portion wise pyridine sulfur trioxide (50%, 18.6 g, 117 mmol) during 10 min at room temperature. Then this mixture was stirred for 3 h at room temperature, treated with a saturated solution of NH₄Cl and extracted with AcOEt. The organic phase was dried over Na₂SO₄, filtrated and concentrated in vacuo. The residue was triturated in diethyl ether to give 17 (9.6 g, 95%) as a white solid. ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 9.75 (s, 1H, *CHO*) 7.9–7.8 (2s, 2H, N*CH*=C, N*CH*=N), 7.65–7.25 (2d, 4H, *Ph*), 5.6 (s, 2H, Ph*CH*₂N). IR: *V*_{max} 2229 cm⁻¹.

Preparation of compounds [3-(R_3)-3-H-imidazol-4-yl]-carboxaldehyde 18–27. The aldehydes **18** to **27** listed in Table 6 were obtained by the same procedure used for **17** by using the corresponding amine in step 1.

Preparation of (R,S) 4-{5-[(1-phenyl-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (28). To a mixture of compounds 11 (0.28 g, 1.4 mmol) and 17 (0.29 g, 1.4 mmol) in DCE (5 mL), was added NaB-H(OAc)₃ (0.435 g, 2.1 mmol) at 0 °C. Then this mixture was stirred overnight at room temperature and concentrated in vacuo. The residue was diluted with CH₂Cl₂ washed with saturated NaHCO₃, brine, dried over Na₂SO₄ and concentrated in vacuo. Flash chromatography (silica gel, CH₂Cl₂/MeOH; 95:5) gave 3201

compound **28** (0.28 g, 50%) as an oil which was crystallized in diethylether as the fumarate. ¹H NMR (200 MHz, CDCl₃): δ (ppm) 9.3 (s, 1H, NCHN), 7.85–7.35 (m 9H, *Ph*), 3.85–3.65 (m, 3H, *CH*₂N, NCHCO), 2.2–1.7 (m, 6H, (CH₂)₃). IR: V_{max} 2800, 2229, 1702, 1655, 1561 cm⁻¹. Anal. (C₂₄H₂₅ N₅O) 1.4 C₄H₄ O₄, 1H₂O. C, H, N calcd: 61.72; 5.62; 12.06. Found: 61.40; 5.64; 11.74.

Preparation of (*S*) 4-{5-[(1-phenyl-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (28a). Compound 28a (0.22 g, 65%) was prepared by the same procedure used for 28 with compound 11a (0.2 g, 0.98 mmol), 17 (0.21 g, 0.98 mmol) and NaBH(OAc)₃ (0.31 g, 1.5 mmol). Mp: 75 °C (decomposed); RMN; IR similar to 28. Anal. ($C_{24}H_{25}$ N₅O) 1.2 C_4H_4 O₄, 1H₂O: C, H, N calcd: 61.82; 5.69; 12.5. Found: 64.15; 6.05; 11.80. The enantiomeric excess was determined by HPLC using a Chiralcel OD column (eluant/heptane/ isopropanol:DEA; 500:500:0.5) and a UV detector at 230 nm (ee 89%).

Preparation of (R,S) 4-{5-[(1-benzyl-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (29). Compound 29 was prepared by the same procedure used for 28 with compound 12 (10.6 g, 40 mmol), 17 (11.05 g, 51.6 mmol), and $NaBH(OAc)_3$ (8.9 g, 43 mmol). Flash chromatography (silica gel, $CH_2Cl_2/$ MeOH; 95:5) gave compound 29 as an oil which was crystallized in diethylether as the dihydrochloride (10.2 g, 52%). ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 9.25 (s, 1H, NCHN), 7.85-7.3 (m 9H, PhPh), 5.85 (s, 2H, CH₂Ph), 4.8–4.4 (m, 3H, CH₂Ph, CHNH), 3.7–3.25 (m, 2H, CH_2N), 2.35–1.2 (m, 6H, $(CH_2)_3$). IR: V_{max} 3317, 2226, 1651, 1538 cm⁻¹. Anal. (C₂₅H₂₇ N₅O) 2HCl, 0.5H₂O: C, H, N calcd: 60.60; 6.26; 14.14. Found: 61.06; 5.93; 14.18.

Preparation of (S) 4-{5-[(1-benzyl-2-oxo-azepan-3-yl-amino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (29a). Compound 29a was prepared as the dihydrochloride (0.135 g, 26%) by the same procedure used for 28 with compound 12a (0.25 g, 1.14 mmol), 17 (0.24 g, 1.14 mmol) and NaBH(OAc)₃ (0.34 g, 1.6 mmol). RMN; IR similar to 29. Anal. ($C_{25}H_{27}$ N₅O) 2HCl, 1H₂O: C, H, N calcd: 59.46; 6.14; 13.87. Found: 59.58; 6.17; 13.45. The enantiomeric excess was determined by HPLC using a Chiralcel OD column (eluant: methanol/DEA; 1000:1) and a UV detector at 230 nm. (ee 90%).

Preparation of the enantiomers of 2[2-(4'-chlorobiphenylthio)]-5-[2-(4-oxo-4H-benzo[d] [1,2,3]triazin-3-ylmethyl)]cyclopentanecarboxylic acid (29a/29b). The enantiomeric mixture 29 (1.6 g) was separated by preparative HPLC using a Chiralpack AD 250*4.6 mm column (eluant: methanol/DEA; 1000:1) and a UV detector at 230 nm to give: Pic 1: 29a (0.6 g, elution time 7.75 min), ee >98%, $[\alpha]_D^{20} = +6.69$ (c 0.99; MeOH); ¹H NMR RMN; IR similar to 29. Anal. (C₂₅H₂₇ N₅O) 2HCl, 1H₂O: C, H, N calcd: 59.46; 6.14; 13.87. Found: 59.68; 6.43; 13.90. Pic 2: 29b (0.6 g, elution time 11.5 min), ee >99%, $[\alpha]_D^{20} = -6.52$ (c 0.89; MeOH); ¹H NMR RMN; IR similar to 29. Anal. (C₂₅H₂₇ N₅O) 2HCl, 1H₂O: C, H, N calcd: 59.46; 6.14; 13.87. Found: 59.67; 6.49; 13.88.

Preparation of (*R*,*S*) 4-{5-[(1-(2-phenylethyl)-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (30). Compound 30 was prepared as the dihydrochloride (0.155 g, 31%) by the same procedure used for 28 with compound 13 (0.25 g, 1.1 mmol), 17 (0.23 g, 1.1 mmol) and NaBH(OAc)₃ (0.34 g, 1.6 mmol). Mp: 227 °C; ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 9.2 (s, 1H, NCHN), 7.95–7.25 (m, 10H, NCH, PhPh), 5.8 (AB, 2H, CH₂Ph), 4.55 (m, 1H, CHNH), 4.25–3.85 (AB, 2H, CH₂N), 3.8–3.3 (m, 4H, CH₂N, CH₂N), 2.75 (t, 2H, CH₂Ph), 2.3–1.2 (m, 6H, (CH₂)₃). IR: V_{max} 3600–2400, 2236, 1659, 1606 cm⁻¹. Anal. (C₂₅H₂₇N₅O) 2HCl, 1.5H₂O: C, H, N calcd: 59.14; 5.87; 13.27. Found: 58.86; 6.28; 13.11.

Preparation of (*S*) 4-{5-[(1-(2-phenylethyl)-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (30a). Compound 30a was prepared as the dihydrochloride (0.215 g, 22%) by the same procedure used for 28 with compound 13a (0.5 g, 2.15 mmol), 17 (0.45 g, 2.15 mmol) and NaBH(OAc)₃ (0.64 g, 3 mmol). Mp: 167 °C; RMN; IR similar to 30. Anal. ($C_{25}H_{27}$ N₅O) 2HCl, 2H₂O: C, H, N calcd: 58.14; 5.96; 13.04. Found: 57.76; 6.33; 12.76. The enantiomeric excess was determined by HPLC using a Chiralcel OD column (eluant: heptane/isopropanol/DEA; 500:500:0.5) and a UV detector at 210 nm. (ee 85%).

Preparation of (*R*,*S*) 4-{5-[1-(3-phenylpropyl)-2-oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (31). Compound 31 was prepared as the fumarate (0.12 g, 19%) by the same procedure used for 28 with compound 14 (0.3 g, 1.2 mmol), 17 (0.26 g, 1.2 mmol) and NaBH(OAc)₃ (0.52 g, 2.45 mmol). Mp: 180 °C; ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.75–7.1 (m, 10H, N*CHN PhPh*), 6.8 (s, 1H, N*CH*), 6.6 (s, 2H, *CH=CH*), 5.45 (s, 2H, Ph*CH*₂NH), 3.5 (AB, 2H, Im*CH*₂N), 3.1–2.55 (m, 7H, *CH*NH, 2 *CH*₂N, *CH*₂Ph), 1.8–1.1 (m, 8H, (*CH*₂), (*CH*₂)₃). IR: *V*_{max} 2800–1800, 2229, 1655, 1577 cm⁻¹. Anal. (C₂₇H₃₁ N₅O) C₄H₄O₄: C, H, N calcd: 66.77; 6.33; 12.56. Found: 66.67; 6.38; 12.54. M⁻H⁺(440)

Preparation of (*R*,*S*) 4-{5-[(1-(benzenesulfonyl)-2-oxoazepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}-benzonitrile (32). Compound 32 was prepared as the free base (0.193 g, 37%) by the same procedure used for 28 with compound 15 (0.3 g, 1.11 mmol), 17 (0.236 g, 1.11 mmol) and NaBH(OAc)₃ (0.33 g, 1.56 mmol). ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 7.95–7.25 (m, 10H, NCHN PhPh), 6.75 (s, 1H, NCH=), 6.5 (s, 2H, CH=CH), 5.35 (s, 2H, PhCH₂N), 4.35 (m, 1H, CHNH), 3.8–3.6 (2dd, 2H, CH₂NH), 3.4 (m, 2H, CH₂NSO₂), 2.0– 1.1 (m, 6H, (CH₂)₃). IR: V_{max} 3308, 2227, 1693, 1348, 1164 cm⁻¹. Anal. (C₂₄H₂₅ N₅O₃S) C₄H₄O₄: C, H, N calcd: 62.19; 5.44; 15.11. Found: 62.15; 5.44; 14.80. MH⁺ (464).

Preparation of (R,S) 4-{5-[(1-(benzenesulfonylmethyl)-2oxo-azepan-3-ylamino)-methyl]-imidazol-1-ylmethyl}benzonitrile (33). Compound 33 was prepared as the dihydrochloride (0.156 g, 27%) by the same procedure used for **28** with compound **16** (0.27 g, 1.04 mmol), **17** (0.2 g, 1.04 mmol) and NaBH(OAc)₃ (0.44 g, 2.1 mmol). Mp: decomposed. ¹H NMR (200 MHz, DMSO-*d*₆): δ (ppm) 9.2 (s, 1H, NCHN), 7.95–7.4 (m, 9H, *PhPh*), 7.75 (s, 1H, NCH=), 5.75 (AB, 2H, PhCH₂N), 5.05 (AB, 2H, SO₂CH₂N), 4.35 (AB, 2H, CH₂NH), 3.8–3.6 (2dd, 2H, CH₂NH), 3.4 (m, 2H, CH₂NSO₂), 2.35–1.3 (m, 6H, (CH₂)₃). IR: V_{max} 3407, 2800–2300, 2231, 1643, 1321, 1141 cm⁻¹. Anal. (C₂₅H₂₇N₅O₃S) 2HCl, H₂O: C, H, N, S calcd: 52.77; 5.45; 12.31; 5.62. Found: 52.72; 5.55; 11.75; 5.45. MH⁺(478).

Preparation of compounds 34a-53a: general procedure

To a solution of compound 4a (0.23 g; 1 mmol) in THF (5 mL) was added NaH 80% in oil (32 mg) at 0 °C. After 5 min agitation, nBu₄NI (37 mg; 1% mol) and the corresponding benzyl bromide (1.05 mmol) were added. After 1h agitation at room temperature the reaction was guenched by adding of saturated NH₄Cl solution (1 mL) and H₂O (4 mL) then extracted twice with AcOEt (5 mL). The organic layer was dried over MgSO₄ filtrated and evaporated to dryness. The crude compound was treated with HCl (4 M) solution in dioxane (10 mL) and agitated at room temperature until complete Boc deprotection. The solvent was removed and the residue was dissolved in CH₂Cl₂ (10 mL) and washed with saturated NaHCO3 (5mL) dried over MgSO₄ filtrated and evaporated to dryness. The residue was dissolved in CH₂Cl₂ (5 mL) and aldehyde 17 (1 equiv) and NaBH(OAc)₃ (1.5 equiv) were added successively. After one night at room temperature was added saturated NaHCO₃ (5 mL) and the compound was extracted twice with CH_2Cl_2 (5 mL). The organic layer was dried over MgSO₄ filtrated and evaporated to dryness. The final compound was purified by reversed phase SPE affording the pure corresponding compound as a TFA salt. This salt was dissolved in CH₂Cl₂ (10 mL) and washed with saturated NaHCO₃ solution (5 mL). The organic layer was dried over MgSO₄ filtrated and evaporated to dryness to afford the free base of compounds 34a to 53a which was transformed into hydrochloride or fumarate salt to form an amorphous solid (%, Table 7).

Preparation of compounds 54a–63a

Compounds **54a–63a** were prepared as dihydrochloride or fumarate by the same procedure used for **29a** by reductive amination of the various aldehydes **18–27** with the intermediate amine of **39a** and NaBH(OAc)₃ (%, Table 8).

Preparation of (S) 1-benzyl-3-[1-trityl-imidazol-4-ylmethyl)aminol-azepan-2-one (64a). To a mixture of compounds 12a (7g, 24.2 mmol) and 1-trityl-4-carboxaldehyde imidazol (8.2g, 24.2 mmol) in DCE (85 mL), was added NaBH(OAc)₃ (10.2g, 48.5 mmol) at 0 °C. Then this mixture was stirred overnight at room temperature and concentrated in vacuo. The residue was diluted with CH_2Cl_2 washed with saturated NaHCO₃, brine, dried over Na₂SO₄ and concentrated in vacuo. Flash chromatography (silica gel, $CH_2Cl_2/MeOH$; 95:5) gave compound **64a** (10.5 g, 81%) as a foam. ¹H NMR (200 MHz, DMSO- d_6): δ (ppm) 7.45–7.1 (m, 20H, 4 *Ph*), 7.35 (s, 1H, N*CH*N), 6.75 (s, 1H, N*CH*=), 4.55 (AB, 2H, *CH*₂Ph), 3.8–3.5 (m, 3H, *CH*NH, *CH*₂N), 1.9–1.1 (m, 6H, (*CH*₂)₃). IR: V_{max} 3300, 1644 cm⁻¹.

Preparation of (S) 1-benzyl-3-[3H-imidazol-4-ylmethyl)amino]-azepan-2-one(65a). A solution of 64a (0.5 g, 0.925 mol) in HCO₂H (5 mL) was stirred 3 h at room temperature and then concentrated in vacuo. The residue was diluted in ether and extracted with 1 N aqueous HCl. The aqueous phase was treated with 10 N aqueous NaOH (to pH > 10) and extracted with CH_2Cl_2 . The organic phase was dried over Na₂SO₄ and concentrated in vacuo to give 65a which was crystallised as the fumarate (0.26 g, 67%). Mp: 60 °C decomposed; ¹H NMR (200 MHz, DMSO-d₆): δ (ppm) 7.65 (s, 1H, NCHN), 7.3 (m, 5H, Ph), 7.0 (s, 1H, NCH=), 6.5 (s, 2H, *CH=CH*), 4.55 (AB, 2H, *CH*₂NH), 3.85 (AB, 2H, *CH*₂NH), 3.55–3.2 (m, 2H, *CH*₂N), 1.95–1.3 (m, 6H, $(CH_2)_3$). IR: V_{max} 3600, 2200, 1705–1659, 1620, 1567 cm⁻¹. Anal. $(C_{17}H_{22}N_4O)$ 1.25 $C_4H_4O_4$, 1 H_2O : C, H, N calcd: 57.28; 6.72; 12.14; 5.62. Found: 57.56; 6.70; 11.90.

Preparation of (S) 4-{[3-(4-cyanobenzyl-3H-imidazol-4ylmethyl)-(1-benzyl-2-oxo-azepan-3-yl)-amino]-methyl}benzonitrile (66a). Preparation of (S) 4-{[1-trityl-1H-imidazol-4-ylmethyl)-(1-benzyl-2-oxo-azepan-3-yl)-amino]methyl}-benzonitrile. To a suspension of NaH (60% in oil, 0.11 g, 2.75 mmol) in DMF (6 mL) was added a solution of compound 64a (1g, 1.85 mmol) in DMF (5 mL) at 0 °C. Then this mixture was stirred 1 h and a solution of *p*-cyanobenzyl bromide (0.725 g, 2.7 mmol) and NBu₄I (0.07 g, 0.2 mmol) in DMF (2 mL) was added. After an additional stirring of 2h, the reaction mixture was cooled down to 0°C, treated with a saturated solution of NH₄Cl and extracted with AcOEt. The organic phase was dried over Na₂SO₄, filtrated and concentrated in vacuo. The residue was purified by flash chromatography (silica gel, heptane/AcOEt; 50:50) to give the title compound as a yellow foam (0.585 g, 48%)which was used directly for the next step.

Preparation of (66a). To a solution of the compound obtained in step a (0.25 g, 0.38 mmol) in AcOEt (1 mL) was added a solution of *p*-cyanobenzyl bromide (0.075 g, 0.38 mmol) in AcOEt (1 mL) at room temperature. Then the reaction mixture was heated at 60°C overnight and concentrated in vacuo. The residue was treated with MeOH (2mL), heated at 60 °C for 2h and concentrated in vacuo. The residue was purified by flash chromatography (silica gel, AcOEt/EtOH, 90:10) to give **66a** which was crystallized as a fumarate (0.036 g, 18%). Mp: $>100 \,^{\circ}$ C decomposed; ¹H NMR (200 MHz, DMSO d_6): δ (ppm) 7.8–7.0 (m, 15H, NCH=, NCHN, 3Ph), 7.75 (s, 1H, NCH=), 5.21 (AB, 2H, PhCH₂N), 4.55 (AB, 2H, PhCH₂N), 4.2-3.5 (2AB, 4H, CH₂NCH₂), 3.6 (m, 1H, CHN), 3.15 (m, 2H, CH₂N), 2.35–1.3 (m, 6H, $(CH_2)_3$). IR: V_{max} 3300–2300, 2228, 1700, 1645 cm⁻¹. Anal. (C₃₃H₃₂N₆O) 1C₄H₄O₄, 0.5H₂O: C, H, N calcd: 67.89; 5.66; 12.84. Found: 67.87; 5.91; 12.32.

Preparation of compounds 67a-96a. Preparation of compounds 67a-82a. General procedure. To a solution of compound 64a (200 mg, 0.37 mmol) in CH₂Cl₂ (5 mL) is added the corresponding aldehyde (0.37 mmol) at room temperature followed by NaBH(OAc)₃ (118 mg, 0.55 mmol). After one night at room temperature is added saturated NaHCO₃ (5mL). After 10min the organic layer was recovered and the aqueous solution was extracted twice with CH₂Cl₂ (5 mL). The organic layers were pooled and dried over magnesium sulphate, filtrated and evaporated to dryness. The recovered compound was dissolved in CH₂Cl₂ (2 mL), treated by Et₃SiH (0.1 mL) and TFA (2 mL). After overnight agitation at room temperature, the solvent was removed under vacuum and the final compound was purified by reversed-phase chromatography affording the pure corresponding compound as a TFA salt (%, Table 9).

Preparation of compounds 83a–96a. General procedure. To a solution of compound **10** (170 mg, 0.31 mmol) in THF (5 mL) is added polyvinyl pyridine (500 mg) and the corresponding acylchloride or chloroformiate (0.4 mmol) at room temperature. After overnight agitation, the reaction mixture is filtrated and the resin is washed twice with ethyl acetate (5 mL). The organic layer was concentrated and the compound purified by quick filtration on normal phase SPE (AcOEt/CH₂Cl₂). The recovered compound was dissolved in CH₂Cl₂ (2 mL), treated by Et₃SiH (0.1 mL) and TFA (2 mL). After overnight agitation at room temperature, the solvent was removed under vacuum and the final compound was purified by reversed phase chromatography affording the pure corresponding compound as a TFA salt (%, Table 9).

Enzyme assays

FTase and GGTase-1 were purified from rat brains and enzymatic activities were evaluated using the Amersham scintillation proximity assay with the C-terminal sequence of lamin B as a substrate for FTase (biotin-YRASNRSCAIM) and the biotin-TKCVIL sequence for GGTase; [³H]-farnesyl- and [³H]-geranylgeranylpyrophosphate were used as donors. IC₅₀ of each product on each enzyme was calculated by EXCEL software using three points in the central linear range of fluorescence inhibition.

Cellular assays

The rat cell line Ras#1 obtained after transfection of fibroblastic RAT2 cells with *v*-*H*-*Ras*¹⁷ reverts to the fibroblastic phenotype after treatment with FTase inhibitors. This effect is accompanied by a reduction in growth kinetics and cell viability. When quantifying cell number as a function of dose, this translates in the formation of a plateau at doses effective to inhibit FTase eventually followed by a cytotoxic decrease at higher doses. A serum batch was chosen to ensure that this plateau occurred at less than 50% of the cell number so that IC_{50s} indicated in the tables refer to cell inhibition of FTase when appropriate and not to unrelated cytotoxicity of

the compounds. Cells were treated for four days in 96well tissue culture plates and cell number estimated indirectly through a tetrazolium viability assay.

Acknowledgements

The authors wish to thank Sabine Plantier, Sophie Sciberras, Christel Daumas, Stephanie Dupas and Annie Genton for their skilful technical assistance and Solange Huet for the preparation of the manuscript. We would also like to acknowledge the analytical department at the IdRS for performing all the spectral analyses.

References and Notes

1. Huang, C. C.; Casey, P. J.; Fierke, C. A. J. Biol. Chem. 1997, 272, 20.

2. Lerner, E. C.; Hamilton, A. D.; Sebti, S. M. Anti-Cancer Drug Design 1997, 12, 229.

3. (a) Prendergast, G. C.; Rane, N. *Exp. Opin. Investig. Drugs* **2001**, *10* (12), 2105. (b) Cox, A. D.; Der, C. J. Current. Opin. Pharm. **2002**, *2*, 388.

4. (a) Haluska, P.; Dye, G. K.; Adjei, A. A. *Eur. J. Cancer* **2002**, *38*, 13 1685. (b) Du, W.; Lebowitz, P. F.; Prendergast, G. C. *Mol.Cell. Biol.* **1999**, *19* (3), 1831 and references cited herein.

5. Moores, S. L.; Schaber, M. D.; Mosser, S. D.; Rands, E.; O'Hara, M. B.; Garsky, V. M.; Marshall, M. S.; Pompliano, D. L.; Gibbs, J. B. J. Biol. Chem. **1991**, 266 (22), 14603.

6. Hunt, J. T.; Lee, V. G.; Leftheris, K.; Seizinger, B.; Carboni, J.; Mabus, J.; Ricca, C.; Ning, Y.; Manne, V. J. Med. Chem. **1996**, *39* (2), 353 and references cited therein.

7. (a) Hunt, J. T.; Ding, C. Z.; Batorsky, R.; Bednarz, M.; Bhide, R.; Cho, Y.; Chong, S.; Chao, S.; Gullo-Brown, J.; Guo, P.; Kim, S. H.; Lee, F. Y. F.; Leftheris, K.; Miller, A.; Mitt, T.; Patel, M.; Penhallow, B. A.; Ricca, C.; Rose, W. C.; Schmidt, R.; Slusarchyk, W. A.; Vite, G.; Manne, V. J. Med. Chem. 2000, 43 (20), 3587. (b) Rose, W. C.; Lee, F. Y. F.; Fairchild, C. R.; Lynch, M.; Monticello, T.; Kramer, R. A.; Manne, V. *Cancer Res.* **2001**, *61*, 7507.

8. Britten, C. D.; Rowinsky, E. K.; Soignet, S.; Patnaik, A.; Yao, S.-L.; Deutsch, P.; Lee, Y.; Lobell, R. B.; Mazina, K. E.; McCreery, H.; Pezzuli, S.; Spriggs, D. *Clin. Cancer Res.* **2001**, 7, 3894.

9. (a) Kelland, L. R.; Smith, V.; Valenti, M.; Patterson, L.; Clarke, P. A.; Detre, S.; End, L.; Howes, A. J.; Dowsett, M.; Workman, P.; Johnston, S. R. D. *Clin. Cancer Res.* 2001, 7, 3544. (b) End, D. W.; Smets, G.; Todd, A. V.; Applegate, T. L.; Fuery, C. J.; Angibaud, P.; Venet, M.; Sanz, G.; Poignet, H.; Skrzat, S.; Devine, A.; Wouters, W.; Bowden, C. *Cancer Res.* 2001, *61*, 131.

 (a) Liu, M.; Bryant, M. S.; Chen, J.; Lee, S.; Yaremko, B.; Lipari, P.; Malkowski, M.; Ferrari, E.; Nielsen, L.; Prioli, N.; Dell, J.; Sinha, D.; Syed, J.; Korfmacher, W. A.; Nomeir, A. A.; Lin, C.-C.; Wang, L.; Taveras, A. G.; Doll, R. J.; Njoroge, F. G.; Mallams, A. K.; Remiszewski, S.; Catino, J. J.; Girijavallabhan, V. M.; Kirschmeier, P.; Bishop, W. R. *Cancer Res.* **1998**, *58*, 4947. (b) Reichert, A.; Heisterkamp, N.; Daley, G. Q.; Groffen, J. *Blood* **2001**, *97*, 1399. (c) Peters, D. G.; Hoover, R. R.; Gerlach, M. J.; Koh, E. Y.; Zhang, H.; Choe, K.; Kirschmeier, P.; Bishop, W. R.; Daley, G. Q. *Blood* **2001**, *97*, 1404.

11. (a) Long, S. B.; Hancock, P. J.; Kral, A. M.; Hellinga, H. W.; Beese, L. S. *P.NA.S.* **2001**, *98*, 12948. (b) Strickland, C. L.; Windsor, W. T.; Syto, R.; Wang, L.; Bond, R.; Wu, Z.; Schwartz, J.; Le, H. V.; Beese, L. S.; Weber, P. C. *Biochemistry* **1998**, *37*, 16601.

12. Ciccarrone, T. M.; MacTough, S. C.; Williams, T. M.; Dinsmore, C. J.; O'Neil, T. J.; Shah, D.; Culberson, J. C.; Koblan, K. S.; Kohl, N. E.; Gibbs, J. B.; Oliff, A. I.; Graham, S. L.; Hartman, G. D.. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 1991. 13. Chan, D. M. *Tetrahedron Lett.* **1996**, *37*, 9013.

14. Dinsmore, C. J.; Williams, T. M.; O'Neil, T. J.; Liu, D.; Rands, E.; Culberson, J. C.; Lobell, R. B.; Koblan, K. S.; Kohl, N. E.; Gibbs, J. B.. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 3301.

15. Aulaskari, P.; Ahlgrén, M.; Rouvinen, J.; Vainiotalo, P. J. *Heterocyl.* **1996**, *33*, 1345.

16. Liu, R.; Dong, D.; Sherlock, R.; Nestler, P.. Bioorg. Med. Chem. Lett. 1999, 9, 847.

17. Martin, A.; Gomez-Muñoz, A.; Waggoner, D. W.; Stone, J. C.; Brindley, D. J. Biol. Chem. **1993**, 268, 23924.