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# A diastereoselective synthesis of $7\alpha$ -nitromethyl steroid derivative and its use for an efficient synthesis of eplerenone

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### 1. Introduction

The reagent nitromethane or its derivatives have been used in steroid chemistry to introduce C-7 nitromethyl group by 1,6-conjugated addition to steroidal 4,6-dien-3-one. However, to our knowledge, only the corresponding 7 $\beta$ -nitromethyl steroid derivatives **1**, **2** [1,2] and nitromethylene **3** [3] were reported (Fig. 1). When compound **4** was treated with NaOMe/MeNO<sub>2</sub>/H<sub>2</sub>O in DMF at room temperature for 1 day [2,3], the conjugated addition gave a mixture of compound **5** $\alpha$  and  $\beta$  with poor stereoselectivity ( $\alpha$ / $\beta$  = 1.6:1) and very low yield (trace). This result intrigued us to optimize the reaction conditions and developed a novel method to stereoselectively introduce 7 $\alpha$ -nitromethyl group, for the 7 $\alpha$ -nitromethyl group on steroid **5** $\alpha$  was very easy to be converted into carboxy by Nef reaction without the corresponding configuration changing [4–6]. The idea to apply steroid **5** $\alpha$  in a synthesis of eplerenone also impelled us to optimize reaction conditions.

Eplerenone is cardiovascular drug, an aldosterone antagonist used as an adjunct in the management of chronic heart failure, marketed by Pfizer under the trade name Inspra [7]. In the total synthesis of eplerenone, the most principal challenge was the introduction of the carbomethoxy substituent. In 1984, Grob accomplished the first synthesis of eplerenone by employing Nagata hydrocyanation of  $\Delta^{9(11)}$ -canrenone as the key step [8], but with moderate stereoselectivity, tedious column chromatography and using toxic cyanide. After that, many companies and research

### ABSTRACT

A novel and efficient method of stereoselectively introducing  $\alpha$ -nitromethyl group to C-7 position of 11 $\alpha$ hydroxyl canrenone **4** was described. In addition, this method was successfully applied in a total synthesis of Eplerenone **8**. The route was characteristic of simple operation, moderate reaction conditions with 5 steps and 55% total yield, at the same time, without any expensive or toxic reagent in use.

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groups were eager to install carbomethoxy group at C-7 by a convenient method in the total synthesis of eplerenone. In 1997, John Ng used toxic acetone cyanohydrin to construct enamine bridge at C4 and C7, and then converted it to carbomethoxy [9]. 6 years later, Pearlman developed a new approach to the furan degradation [10] and used it in the synthesis of eplerenone with higher yield and better selectivity, but this approach needed as low as -50 °C and five steps to degrade furan, which was not atom efficient and problematic in industrial use. Recently, Wuts chose dehydroepiandrosterone as start material by allylic substitution at C7 with moderate stereoselectivity by using expensive stoichiometric amount transition metal salt and an inert environment [11,12]. In 2011, our group developed a new and practical method for the stereoselective construction of a steroid 5α,7β-oxymethylene derivative via 1,6-Michael conjugate addition of steroidal 4,6-dien-3-one at as low as  $-68 \,^{\circ}$ C and applied it in the synthesis of eplerenone [13].

We finally found, when **4** was treated with  $K_2CO_3/MeNO_2$  in DMF in the presence of a catalytic amount of octadecyl dimethyl benzyl ammonium chloride at 90 °C for two days, it was result in 1,6-conjugated addition and a stereochemical conversion at C-7 position in one step to afford the 7 $\alpha$ -nitromethyl steroid **5\alpha** with 90% yield and 96% *de* (Scheme 1). With compound **5\alpha** in hand, we succeeded in a total synthesis of Eplerenone **8** with 5 steps and 55% total yield.

### 2. Experimental

All melting points were determined on a Büchi 510 melting point apparatus and were uncorrected. <sup>1</sup>H and <sup>13</sup>C NMR spectra





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**Fig. 1**. 7β-Nitromethyl-4,5-dihydrocanrenone; 7β-nitromethyl-Δ<sup>9(11)</sup>-4,5-dihydrocanrenone and 6β,7β-methylene-(1'S)-nitro-3-oxo-17α-pregn-4-en-21-carboxylic acid, γ-lactone.

were run on Bruker AM-300, Bruker AM-400 spectrometer using tetramethyl silane as the internal standard ( $\delta = 0$ ). Splitting patterns were designated as "s, d, t, q, and m"; these symbols indicated "singlet, doublet, triplet, quartet, and multiplet", respectively. X-ray crystallographic data were obtained on a CAD-4 diffractometer. Optical rotations were recorded on a Jasco-Dip-181 polarimeter. Mass spectra and high-resolution mass spectra were measured on a Finnigan MAT-95 mass spectrometer. High performance liquid chromatograms (HPLC) were taken on Agilent 1100 instruments and Waters 2998 with column Venusii MP<sup>™</sup> C18, 5 µm, 100 Å,  $10 \times 250 \text{ mm}$  and SunFine<sup>TM</sup> C18, 5  $\mu$ m,4.6  $\times$  150 mm. Silica gel 60 H (200-300 mesh) manufactured by Qingdao Haiyang Chemical Co. (China) was generally used for chromatography. N,N-dimethyl formamide (DMF) and MeNO<sub>2</sub> were distilled and dried by 4 Å molecular sieves overnight. K<sub>2</sub>CO<sub>3</sub> was well ground into a powder, baked at 145 °C for 10 h and then cooled in desiccator with P<sub>2</sub>O<sub>5</sub>. TMG, TEA and DIPEA were short for trimethylglycine, triethyl amine and N,N-Diisopropylethylamine, respectively. The purity of each step product was >98%, which was determined by HPLC and H NMR.

### 2.1. 11 $\beta$ ,17 $\beta$ -Dihydroxy-7 $\alpha$ / $\beta$ -nitromethyl-pregna-4-en-3-one-21carboxylic acid, $\gamma$ -lactone ( $5\alpha$ and $\beta$ )

To a solution of compound **4** (1.07 g, 3 mmol) in MeNO<sub>2</sub> (10 ml), 1,8-Diazabicyclo [5.4.0] undec-7-ene (DBU) (0.8 ml, 5.3 mmol) was added at room temperature and stirred overnight. The resulting reaction was quenched by a.q. HCl (10% w/w) to adjust the pH to 6. Then solvent was removed under vacuum and the residue was purified by silica column chromatography eluted with EtOAc/ cyclohexane (1:4, v/v) to afford diastereomers **5** $\alpha$  (308 mg, 25% yield) and  $\beta$  (192 mg, 15% yield). The compound **5** $\alpha$  was readily recrystallized from EtOAc to afford colorless crystals.Compound **5** $\alpha$ : mp:281–282 °C; [ $\alpha$ ]20D: 51° (c 0.24, CHCl<sub>3</sub>); IR  $\nu_{max}$ (KBr) 3467, 2950, 1766, 1662, 1550, 1458, 1441, 1419, 1382, 1344, 1274, 1192; UV<sub>max</sub>: 242 nm; <sup>1</sup>H NMR (300 MHz,CDCl<sub>3</sub>) δ5.82 (s, 1H), 4.38–4.20 (m, 2H), 4.10 (tt, *J* = 10.29, 10.29, 4.95, 4.95 Hz, 1H), 2.81 (td, *J* = 13.99, 4.16, 4.16 Hz, 1H), 2.73–2.27 (m, 10H), 2.07–1.70 (m, 7H), 1.68–1.59 (m, 2H), 1.50 (ddd, *J* = 24.57, 12.19, 5.69 Hz, 1H), 1.38 (s, 3H), 1.04 (s, 3H); <sup>13</sup>C NMR(100 MHz,CDCl<sub>3</sub>) δ 199.092, 176.239, 164.980, 128.330, 94.707, 74.029, 68.710, 53.387, 46.162, 45.170, 43.219, 39.867, 37.704, 37.471, 36.477, 35.951, 35.328, 34.207, 31.127, 29.087, 22.276, 18.343, 15.679; EI MS(70 eV, m/z): 417(M<sup>+</sup>, 21%), 399(17%), 370(44%), 111(100%); HR-MS(EI) calcd for C<sub>23</sub>H<sub>31</sub>NO<sub>6</sub>(M<sup>+</sup>) 417.2151, Found 417.2153; *Anal.* Calc. for C<sub>23</sub>H<sub>31</sub>NO<sub>6</sub>: C, 66.17; H, 7.48; N, 3.35. Found: C, 66.35; H, 7.63; N, 3.31%.

Compound **5**β: <sup>1</sup>H NMR (300 MHz,CDCl<sub>3</sub>) δ 5.73 (s, 1H), 4.73 (dd, J = 12.72, 3.72 Hz, 1H), 4.37 (dd, J = 12.51, 7.64 Hz, 1H), 4.15–3.99 (m, 1H), 2.67–1.37 (m, 21H), 1.35 (s, 3H), 1.03 (s, 3H); <sup>13</sup>C NMR(100 MHz,CDCl<sub>3</sub>) δ199.556, 176.137, 166.179, 125.006, 93.623, 80.058, 68.297, 58.043, 47.763, 47.027, 43.126, 41.537, 39.625, 38.306, 37.275, 37.004, 35.457, 34.181, 30.969, 29.052, 26.114, 18.600, 16.023; UV<sub>max</sub>: 243 nm; EI MS(70 eV, m/z):417(M<sup>+</sup>, 17%), 111(100%), 370(37%); HR-MS(EI) calcd for C<sub>23</sub>H<sub>31</sub>NO<sub>6</sub>(M<sup>+</sup>) 417.2151, found 417.2157.

### 2.2. 11 $\beta$ ,17 $\beta$ -dihydroxy-7 $\alpha$ -nitromethyl-pregna-4-en-3-one-21carboxylic acid, $\gamma$ -lactone ( $5\alpha$ )

To a solution of compound **4** (20 g, 56 mmol), octadecyl dimethyl benzyl ammonium chloride (2.32 g, 6 mmol) and MeNO<sub>2</sub> (60 ml, 1120 mmol) in DMF (190 ml) equipped with a CaCl<sub>2</sub> drying tube and a reflux condenser, K<sub>2</sub>CO<sub>3</sub> (78 g, 565 mmol) was added at 90 °C with stirring. The mixture was stirred at this temperature for 48 h. The resulting mixture was cooled to room temperature, and then filtered. The filter cake was washed with dichloride methane (DCM) for three times. The organic layers were combined with the filtrate, and adjusted pH to 6 with a.q. HCl (10%, w/w). The HPLC



Scheme 1. Synthesis of  $5\alpha$  and its application in synthesis Eplerenone 8.

analysis suggested the ratio of diastereomer  $5\alpha$  to  $\beta$  was 48:1. The combined organic layers were concentrated under vacuum to give crude product as brown solid. The crude product was washed with water three times and then filtered. The residue was diluted with DCM, and then filtered. The filtrate was washed with a.q. HCl (10%, w/w), brine and dried by Na<sub>2</sub>SO<sub>4</sub> to afford crude powder product  $5\alpha$  (23 g), which was used directly for next step without purification. The crude was readily recrystallized from EtOAc to afford pure product  $5\alpha$  (21 g, 90% yield).

## 2.3. 11 $\beta$ ,17 $\beta$ -dihydroxy-7 $\alpha$ - methoxycarbonyl -pregna-4-en-3-one-21-carboxylic acid, $\gamma$ -lactone (**6**)

To a solution of  $5\alpha$  (5 g, 12 mmol) in DMF (40 ml), AcOH (2.9 ml, 51 mmol) was added at 45 °C with stirring. 30mins later, NaNO<sub>2</sub> (2.484 g, 36 mmol) was slowly added in batches to the mixture. 1 h later, a solution of AcOH (2.9 ml, 51 mmol) in DMF (3 ml) was added dropwise to the mixture. After 4 h, the reaction was cooled to 0 °C and then quenched by sat. Na<sub>2</sub>SO<sub>3</sub> solution (54 ml, 89 mmol). It was allowed to warm to room temperature and stirred overnight. The solvent was removed completely under vacuum to give powder solid. The solid was diluted with acetone and filtered. The filter cake was washed with acetone for three times. The organic layers were combined with filtrate and then concentrated to 100 ml. To the result solution, MeI (1.5 ml, 24 mmol) and K<sub>2</sub>CO<sub>3</sub> (5 g, 36 mmol) were added at 0 °C. The mixture was allowed to warm to room temperature and stirred overnight. The mixture was filtered and the filter cake was washed with DCM for three times. The organic layers were combined with filtrate and then the solvent was completely removed. The residue was diluted with DCM and washed with a.q HCl (10%, w/w), sat. NaHCO<sub>3</sub>, brine and dried by Na<sub>2</sub>SO<sub>4</sub>, then filtered and the filtrate was concentrated to afford crude product 6, which could be purified by recrystallization from DCM/toluene to afford colorless crystals 6 (4 g, 80% yield) [14]: mp.:230–232 °C; <sup>1</sup>H NMR (300 MHz,CDCl<sub>3</sub>)  $\delta$  5.68 (s, 1H), 4.04 (dt, J = 10.23, 9.84, 3.98 Hz, 1H), 3.64 (s, 3H), 2.85-2.71 (m, 2H), 2.68-2.18 (m, 8H), 2.10-1.57 (m, 9H), 1.49-

#### Table 1

1,6-Conjugated addition of MeNO<sub>2</sub> to dienone 4.



**Fig. 2.** Ionic bond and  $\pi$ - $\pi$  stacking interaction.

1.40 (m, 2H), 1.35 (s, 3H), 1.00 (s, 3H); HR-MS(EI) calcd for  $C_{24}H_{32}O_6$  (M<sup>+</sup>) 416.221 found 416.225; *Anal.* Calc. for  $C_{24}H_{32}O_6$ : C,69.21, H,7.74. Found: C, 68.32; H, 7.72%. UV<sub>max</sub>: 244 nm.

2.4.  $17\beta$ -Hydroxy- $7\alpha$ -carbomethoxy-3-oxo-pregna-4,9(11)-diene-21-carboxylic acid,  $\gamma$ -lactone (**7**)

Compound **7** was prepared from **6** according to the reference [12]. mp.: 204–2061 °C; <sup>1</sup>H NMR (300 MHz,CDCl<sub>3</sub>)  $\delta$  5.67 (s, 1H), 5.60 (d, *J* = 5.84 Hz, 1H), 3.54 (s, 3H), 2.93 (dd, *J* = 4.48, 2.00 Hz, 1H), 2.78 (ddd, *J* = 14.93, 5.15, 1.64 Hz, 1H), 2.61–2.36 (m, 6H), 2.33–2.05 (m, 5H), 1.99–1.75 (m, 4H), 1.51–1.38 (m, 2H), 1.35 (s, 3H), 0.90 (s, 3H); <sup>13</sup>C NMR(100 MHz,CDCl<sub>3</sub>)  $\delta$  198.805, 176.667, 172.828, 166.858, 142.504, 125.871, 119.134, 95.252, 51.605, 44.638, 43.935, 43.250, 40.719, 40.526, 35.902, 35.586, 34.359, 33.912, 33.068, 31.616, 29.378, 27.354, 23.448, 14.234.

### 2.5. Eplerenone (8)

Compound **8** was prepared from **7** according to the reference [10].  $[\alpha]$ 20D:+2.2°(c 0.415, CHCl<sub>3</sub>) (lit. [8]  $[\alpha]$  = +5° (*c* = 0.437, CHCl<sub>3</sub>)); mp. 242–244 °C (lit. [8] 240–242 °C); UV<sub>max</sub>: 240 nm; *Anal.* Calc. for C<sub>24</sub>H<sub>30</sub>O<sub>6</sub>: C,69.54; H,7.30. Found: C, 69.25; H, 7.30%. X-ray data of Eplerenone **8** was included in Supplementary data.



Entry	Base	Catal.	Sol.	Temp. (°C)	Time (h)	Yield (%) <sup>a</sup>	α:β <sup>b</sup>
1	NaOMe	-	DMF	RT	9	Trace	1.6:1
2	$K_3PO_4$	A <sup>c</sup>	DMF	RT	9	5	1.6:1
3	$K_3PO_4$	А	DMF	40	9	60	4:1
4	$K_3PO_4$	А	DMF	80	6	50	5:1
5	K <sub>2</sub> CO <sub>3</sub>	Α	DMF	90	9	80	5:1
6	K <sub>3</sub> CO <sub>3</sub>	B <sup>d</sup>	DMF	90	9	30	1:1
7	K <sub>2</sub> CO <sub>3</sub>	А	DMF	90	48	88	10:1
8	K <sub>2</sub> CO <sub>3</sub>	C <sup>e</sup>	DMF	90	48	90	48:1

<sup>a</sup> The yields were determined by column chromatography purification.

<sup>b</sup> The ratios of crude products were determined by HPLC and <sup>1</sup>H NMR.

<sup>c</sup> A was benzyl triethyl ammonium chloride (10% mol).

<sup>d</sup> B was tetraethyl ammonium chloride (10% mol).

<sup>e</sup> C was octadecyl dimethyl benzyl ammonium chloride (10% mol).



Scheme 2. Compound 5 was used in synthesis of Eplerenone 8. Reagents and conditions: (a) (1) NaNO<sub>2</sub>/DMF/AcOH, 35–40 °C; (2) K<sub>2</sub>CO<sub>3</sub>/MeI/acetone, RT; (b) (1) MsCI/Et<sub>3</sub>N/ CH<sub>2</sub>Cl<sub>2</sub>, 0 °C; (2) HCOOK/HCOOH/Ac<sub>2</sub>O, 100 °C; (c) Cl<sub>3</sub>CCONH<sub>2</sub>/H<sub>2</sub>O<sub>2</sub> (30% w/w)/CH<sub>2</sub>Cl<sub>2</sub>/K<sub>2</sub>HPO<sub>4</sub>, 10–20 °C.

### 2.6. The convertion of compound $5\beta$ into $5\alpha$

To a solution of compound  $5\beta$  (1 g, 2.4 mmol), octadecyl dimethyl benzyl ammonium chloride (93 mg, 0.22 mmol) and MeNO<sub>2</sub> (2 ml, 37 mmol) in DMF (6 ml) equipped with a CaCl<sub>2</sub> drying tube and a reflux condenser, K<sub>2</sub>CO<sub>3</sub> (3.3 g, 24 mmol) was added at 90 °C with stirring. The mixture was stirred at this temperature for 48 h. The resulting mixture was cooled to room temperature, and then filtered. The filter cake was washed with dichloride methane (DCM) for three times. The organic layers were combined with the filtrate and all the solvent was removed under reduced pressure to afford crude product  $5\alpha$ . There was no compound  $5\beta$  in crude product detected by HPLC and H NMR. The crude product was placed under vacuum to afford brown powder. The crude powder product was washed with water in sonic washing machine and then filtered for three times. The residue was readily recrystallized from EtOAc to afford pure product  $5\alpha$  (980 mg, 98% yield).



### R=OH M<sup>+</sup>=quaternary ammonium cation B<sup>-</sup>=base anion

Fig. 3. The mechanism of stereochemical conversion of a steroid 7β-nitromethyl derivative into 7α.



Fig. 4. X-ray crystallographic structure of Eplerenone 8.

### 3. Results and discussions

Our synthetic efforts began with preparation of  $7\alpha$ -nitromethyl canrenone by using commercial  $11\alpha$ -hydroxy canrenone **4** as the starting material. The results of 1,6-conjugated addition to 11α-hydroxy canrenone **4** in different conditions are summarized in Table 1. Because NaOMe (5 eq.)/MeNO<sub>2</sub> (11 eq.)/H<sub>2</sub>O (11 eq.) in DMF at room temperature were found as the most frequently used conditions, it was initially employed (entry 1). Indeed, under these conditions, there were trace amounts of compound  $5\alpha$  and  $\beta$  found. We tried to increase the reaction temperature. But when it was as high as 50 °C, the reaction became complex. A lot of NaCH<sub>2</sub>NO<sub>2</sub> was precipitated from the solution when NaOMe was added, and HPLC analysis indicated there was no compound  $5\alpha$  or  $\beta$  detected even for 2 days if without water adding. We thought water helped make NaCH<sub>2</sub>NO<sub>2</sub> dissolve in DMF. Different solvents were screened, such as nonpolar and polar ones and different bases, including organic and inorganic bases, but all failed.

Quaternary ammonium salts were usually used in improving MeNO<sub>2</sub> conjugate addition to  $\alpha$ , $\beta$ -unsaturated ketone [15,16]. The pre-transition state of the enantioselective Michael addition of nitromethane to ketone catalyzed by a chiral quaternary ammonium salts was described by Corey [17]. We envisioned the benzyl group or ArCH<sub>2</sub>-substituent was necessary in quaternary ammonium salt to improve the nucleophilicity of nitromethane anion by ionic bond and a  $\pi$ - $\pi$  stacking interaction [18] between benzyl group and nitromethane anion (Fig. 2).

The commercially available benzyl triethyl ammonium chloride was chose as the phase transfer catalyst at very beginning. Increasing reaction temperature could significantly improve the yield and slightly improve the diastereomeric ratio (entry 2 and 3). But when the temperature was increased to higher than 80 °C in the presence of K<sub>3</sub>PO<sub>4</sub>, the reaction became complex and the yield was obviously decreased even in 6 h (entry 4). So we turned to much weaker bases, such as K<sub>2</sub>HPO<sub>4</sub>, KHCO<sub>3</sub>, CsF etc., and found reaction proceed with better yield only in the presence of K<sub>2</sub>CO<sub>3</sub> at 90 °C (entry 5). The reaction time was another key factor effecting on the stereose-lectivity. A longer extension time could dramatically increase the diastereomeric ratio and slightly improve the yield (entry 7). The

structure of quaternary ammonium salts also had effect on the yield and the stereoselectivity. A quaternary ammonium salts containing benzyl group was required to ensure the better yield and the better stereoselectivity. There was screening of many commercial alkyl quaternary ammonium salts. For examples, tetrabutyl ammonium bromide, methyl trioctyl ammonium chloride or stearyl trimethyl ammonium chloride gave moderate yield and poor stereoselectivity similarly as tetraethylammonium chloride did (entry 6). We finally found the bulky commercial benzyl quaternary ammonium salts, such as octadecyl dimethyl benzyl ammonium chloride, was the best phase transfer catalyst. Compound **4** with MeNO<sub>2</sub> and K<sub>2</sub>CO<sub>3</sub> under catalysis of octadecyl dimethyl benzyl ammonium chloride (10% mol) at 90 °C afforded compound **5** $\alpha$  in satisfactory yield (90%) and good *de* (96%) (entry8).

We envisioned the mechanism of stereochemical inversion of 7 $\beta$ -nitromethyl group into more thermodynamically stable 7 $\alpha$ one was that the elimination of nitromethane from compound **5**B was much more easily than that from compound  $5\alpha$ . The formation of six-membered-ring pre-transition state accounted for preferred elimination of nitromethane from compound 5ß at high temperature (Fig. 2). The exclusive formation of compound  $5\alpha$  was quite likely the result of a thermodynamic control. It was supported by the fact when compound  $5\beta$  was treated with MeNO<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub> and octadecyl dimethyl benzyl ammonium chloride (10% mol) at 90 °C for 2 days, which were the same conditions as entry 8 showed, compound  $5\beta$  was almost completely converted into  $5\alpha$ , meanwhile there was a trace amount of compound **4** detected by HPLC. The ammonium salt catalyzed this procedure maybe by exposure of oxygen on nitromethyl group. The reason was that if without the octadecyl dimethyl benzyl ammonium chloride, the inversion of the configuration would be unable to happen even stirring for a week.

All the intermediate compounds were fully characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR and mass spectral data. The structural assignments of the 7 $\alpha$ - and 7-isomers were determined by <sup>1</sup>H NMR. It had been reported that the equatorial 7 $\beta$ -nitromethyl derivative **1** and **2** showed <sup>1</sup>H NMR signal for C-7  $\beta$  nitromethyl group (–CH<sub>2</sub>NO<sub>2</sub>) at 4.76 (dd, *J* = 12.16, 2.60 Hz, 1H), 4.35 (dd, *J* = 12.36, 8.70 Hz, 1H) [2] and 4.70 (dd, *J* = 12.38, 4.26 Hz, 1H), 4.37 (dd, *J* = 12.40, 9.06 Hz, 1H) [1], respectively. Compound **5** $\alpha$  and **5** $\beta$  showed <sup>1</sup>H NMR signal for C-7 nitromethyl group at 4.38–4.20 (m, 2H) and 4.73 (dd, *J* = 12.72, 3.72 Hz, 1H), 4.37 (dd, *J* = 12.51, 7.64 Hz, 1H), respectively.

With the important compound  $5\alpha$  in hand,  $7\alpha$ -carboxyl derivative from compound  $5\alpha$  was achieved by modified Henry reaction [4], without purification, followed by methylation of  $7\alpha$ -carboxy group to afford ester **6** with 80% yield, without detecting any C- $7\beta$  methyl carboxyl diastereomer of compound **6** by <sup>1</sup>H NMR or HPLC analysis. The ester **6** was converted into Eplerenone **8** by a sequence same to that was previously developed for the construction of 9(11)epoxide [10](Scheme 2). The structure assignment of Eplerenone **8** was confirmed by X-ray (Fig. 3).

In conclusion, we succeeded in an introduction of nitromethyl group at C-7 from commercial available  $11\alpha$ -hydroxy canrenone **4** and a stereochemical conversion of 7 $\beta$ -nitromethyl group into 7 $\alpha$  in one step with 90% yield and 96% *de*. An efficient route for the synthesis of Eplerenone **8** was established with easy operation, five steps and 55% total yield and without any expensive or toxic reagent in use Fig. 4.

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### Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.steroids. 2012.04.017.

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