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Regio- and stereoselective anomeric esterification of glucopyranose 1,2-diols and a facile preparation of 2-O-acetylated glucopyranosyl trichloroacetimidates from the corresponding 1,2-diols

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Abstract—A highly regio- and stereoselective anomeric esterification of 3-O-allyl (or benzyl, or benzyl)-4,6-O-isopropylidene- α , β -D-glucopyranose with acetyl chloride, or allyl chloroformate, or ethyl chloroformate gave the corresponding 2-OH, 1- β -acetates or -carbonates in excellent yields. The 2-OH, 1- β -acetates were readily converted to the corresponding 2-O-acetylated glucopyranosyl trichloroacetimidates by reaction with trichloroacetonitrile via base promoted acetyl migration, while the 2-OH, 1- β -carbonates were good glycosyl acceptors for the synthesis of (1 \rightarrow 2)-linked oligosaccharides. © 2007 Elsevier Ltd. All rights reserved.

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1. Introduction

Regioselective introduction of protecting group is of crucial importance in carbohydrate chemistry.¹ Acyl groups, especially acetyl and benzoyl, are generally used as electron-withdrawing protecting groups to block hydroxyls, and also used as good neighboring participating groups at 2-position for anomeric stereocontrol in glycosylation reactions. Although some selective protection approaches such as the use of 1-(benzoyloxy)benzotriazole (BzOBT),^{2–5} the dibutyltin oxide mediated selective monoprotection strategy,^{6,7} the phase-transfer method,⁸ the silver(I) oxide promoted acylation technique,⁹ the Cu(II)-mediated acylation procedure,¹⁰ or the selective activation of hydroxyl groups through stannylene compounds,¹¹ as well as enzyme methods¹² are available, these methodologies are mostly focused

on the selective acylation of sugar 2,3- or 4,6-diols. Regioselective protection at the anomeric hydroxyl group is, however, a challenge for chemists.^{4,13}

Previously, we have revealed that selective acylation of allyl 4,6-O-isopropylidene- α -D-mannopyranoside¹⁴ or allyl 4-O-benzoyl- α -L-rhamnopyranoside¹⁵ with acetyl chloride or chloroacetyl chloride in pyridine gave highly selective 3-O-acylation products, and with this method, a series of complex rhamnose and mannose oligosaccharides were synthesized efficiently.^{14–18} As an extension to this method, we wish to report herewith the regio-and stereoselective C-1-O-acylation of glucopyranose 1,2-diols and their use in the preparation of glucose trichloroacetimidate donors and oligosaccharides.

The synthesis of 4,6-*O*-isopropylidene-3-*O*-allyl- (8), -3-*O*-benzyl- (9), and -3-*O*-benzyl- α , β -D-glucopyranose (10) is depicted in Scheme 1. Allylation, or benzylation, or benzoylation of 1,2:5,6-di-*O*-isopropylidene- α -D-glucofuranose (1) gave compounds 2,¹⁹ or 3, or 4,²⁰ respectively. Subsequent hydrolysis was carried out in an aqueous solution of sulfuric acid (4%) under heating at

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Scheme 1. Reagents and conditions: (a) AllBr, NaH, DMF for 2; BnBr, NaH, DMF for 3; BzCl-pyridine for 4; (b) 4% H₂SO₄, reflux, 3–4 h; (c) TsOH, DMF, 1.05 equiv 2-methoxypropene (overall yield from 1: 76%, 81%, and 75% for 8, 9, and 10, respectively).

reflux, and the reaction was accompanied by ring expansion.¹⁹ Selective 4,6-O-isopropylidenation of the resultant tetraols 5–7 with 2-methoxypropene in DMF in the presence of catalytic amounts of TsOH afforded the target diol compounds. The above three steps were processed in a consecutive manner without chromatographic separation of the intermediates, making a much simplified preparation of **8–10**.

The 1,2-diols 8-10 were then employed for the selective acylation studies. Benzoylation of 8 with an equivalent amount of benzovl chloride in pyridine at -15 °C led to a mixture of 1-O-Bz, 2-O-Bz, and 1,2-di-O-Bz products, indicating that the benzovlation was not selective. This phenomenon was also observed in stereoselective benzovlation of D-glucosamine derivatives.⁴ When diols 8-10 were treated with allyl chloroformate, or ethyl chloroformate, or acetyl chloride in dichloromethane at room temperature in the presence of 4 equiv of pyridine, their C-1 hydroxyl group was selectively blocked, and 1-allyloxyformates 11, 14, and 17, or 1-ethvloxyformates 12, 15, and 18, or 1-acetates 13, 16, and **19.** with predominant β anomer were obtained, respectively, after separation by silica gel column chromatography (Table 1, entries 1-9). Low temperature (-15 to -10 °C), slow addition of the dichloromethane diluted acetyl chloride or chloroformates were necessary to ensure the high regioselectivity of the reaction. The regioselectivity was indicated clearly by the ¹H NMR spectra of the 1-esters, which exhibited characteristic downfield signals for H-1. Since the proton of the 1hydroxy group is more acidic than the 2-hydroxy group, the high regioselectivity was not surprising.

As for the high β anomeric stereoselectivity, it was attributed to the kinetic stereoelectronic effect or 1,3diaxial repulsion,²¹ enabling the C-1-O⁻ to take the equatorial orientation. Because of the poor solubility of **10** in dichloromethane, more pyridine (10 equiv) was used, and the ratio of α anomer rose considerably (Table 1, entries 7–9). The configuration of the products was readily assigned from their ¹H NMR spectra as indicated in Table 1, that is, the α anomers showed small coupling constants ($J_{1,2}$ 3.3–4.0 Hz), while the β anomers showed large coupling constants ($J_{1,2}$ 7.9–8.2 Hz). Trichloroacetimidate has widely been used as an excellent leaving group in glycosidic bond formation,²¹ and also as a temporary protective group for hydroxyl function.^{22–24} In our research, transformation of the 2-OH of the obtained glucosyl acetates and carbonates to the corresponding 2-trichloroacetimidate was studied. It was found that carbonates **14** and **15** were readily converted to the corresponding 2-trichloroacetonitrile in the presence of K₂CO₃ and a catalytic amount of DBU, affording **20** and **21** in good yields, and no α/β anomerization was observed (Scheme 2).

However, in the case of acetates 13 and 16, the reactions gave quite different results. Trichloroacetimidation of 13 and 16 with trichloroacetonitrile under the same conditions gave the corresponding 2-O-acetylated glucopyranosyl trichloroacetimidates 24 (84%) and 25 (79%), respectively. Compounds 24 and 25 showed identical NMR data with those of the authentic samples prepared from 8 and 9 via 1.2-di-O-acetvlation. 1-O-deacetvlation, and trichloroacetimidation. The facile migration of acvl groups in partially acvlated sugars under mildly alkaline conditions is well known and has been the subject of numerous reports in the literature.^{25,26} Apparently, migration of the anomeric acetate and anomerization occurred during the process of trichloroacetimidation as outlined in Scheme 3. Similar 1,2-Osilvl group migration^{27,28} has been noticed by Schmidt and co-workers. Trichloroacetimidates 24 and 25 are very valuable glucopyranosyl donors as they contain orthogonal protective groups, being capable of application in the synthesis of complex branched oligosaccharides. It was also found that transformation of 8 to 24 via 13 (or 9 to 25 via 16) can be conveniently performed in a one-pot manner, leading to a 70% overall yield for 24 (or 64% for 25), and thus the preparation of 24 or 25 was greatly simplified.

Our attention turned then toward the formation of $(1\rightarrow 2)$ -linked disaccharides from the obtained glucose 2-OH acetate or 2-OH carbonate acceptors since $(1\rightarrow 2)$ -linked oligosaccharides are abundant in nature.^{29–31} It was found that TMSOTf catalyzed condensation of 1-acetate **16** with 2,3,4,6-tetra-*O*-benzoyl- α -D-manno-

Table 1. Selective esterification reactions of the glucose 1,2-diols

| | Me | 0 0 0 0 0 0 | R'COCl-CH ₂ Cl ₂ ; Me ₂ C O CH ₂ Cl ₂ -pyridine | | | |
|---------------------------------------|-----------------|----------------------------|-----------------------------------------------------------------------------------------------------------|------------------------|-----------|---------------------|
| HO OH -15 °C; HO OH OH H' | | | | | | |
| Entry | Reactant number | R'COCl | Product (%) | β : α^{a} | Yield (%) | $J_{1,2}$ (Hz) |
| 1 | 8 | | Me ₂ C 11 AllO OH O OAll | 1:0 I | 82 | 8.2 |
| 2 | 8 | EtO CI | Me ₂ C 12 AllO OH OH OE | 1:0 t | 85 | 8.5 |
| 3 | 8 | Me CI | Me ₂ C 0 13 Allo OH O CH ₃ | 30:1 | 74 | 8.0 (β); 3.9 (α) |
| 4 | 9 | | Me ₂ C 14 BnO OH OAI | 1:0 I | 86 | 7.9 |
| 5 | 9 | EtO CI | Me ₂ C 15 BnO OH OH | 1:0 t | 91 | 7.9 |
| 6 | 9 | Me CI | Me ₂ C 16 O C C H ₃ | 45:1 | 79 | 8.8 (β); 3.8 (α) |
| 7 | 10 | | Me ₂ C 17 0 BzO OH OAI | 14:1 I | 83 | 7.9 (β); 3.5 (α) |
| 8 | 10 | EtO CI | Me ₂ C 18 BzO OH OH OE | 7:1 t | 81 | 8.0 (β); 3.8 (α) |
| 9 | 10 | | Me ₂ C 19 BzO OH OCH ₃ | 7.3:1 | 72 | 8.1 (β); 4.0 (α) |

^a The β : α ratio was estimated from the ¹H NMR spectra of the products.

pyranosyl trichloroacetimidate (26) was unsuccessful as a complex mixture was obtained. However, coupling reactions of carbonate 14 with trichloroacetimidates 26, 28, and 30 in the presence of catalytic TMSOTf under normal conditions³² produced disaccharides 27 (95%), **29** (88%), and **31** (91%), respectively (Scheme 4). The anomeric center of the sugar residue at the reducing end of the obtained disaccharides kept its β configuration as indicated in the NMR spectra of the disaccharides. Compared to the corresponding acetates,



Scheme 2.



Scheme 3.



Scheme 4. Reagents and conditions: TMSOTf, CH_2Cl_2 , -10 °C to rt; 95% for 27: 88% for 29, 91% for 31.

carbonates generally were much more stable presumably because of the resonance effect of the second oxygen.

In conclusion, we have successfully developed a technique for the highly regioselective and stereoselective esterification of 3-O-allyl (or benzyl, or benzoyl)-4,6-Oisopropylidene-D-glucopyranose with acetyl chloride, or allyl chloroformate, or ethyl chloroformate to afford the corresponding 2-OH, 1- β -acetates or -carbonates in good yields. With the aid of base promoted acetyl migration, an efficient preparation of the valuable 2-acetate 1-trichloroacetimidate donors from the corresponding glucopyranosyl 1,2-diols via the 1- β -acetate intermediates was achieved. The 2-OH, 1- β -carbonates were proved to be good acceptors in the coupling reactions for the preparation of $(1\rightarrow 2)$ -linked disaccharides.

2. Experimental

2.1. General methods

Optical rotations were determined with a Perkin-Elmer model 241-MC automatic polarimeter for soln in a 1dm, jacketed cell. NMR spectra were recorded in deuterochloroform soln with a Bruker DPX300 spectrometer, using tetramethylsilane as an internal standard. Elemental analysis was performed on a Yanaco CHN Corder MF-3 automatic elemental analyzer. Thin-layer chromatography (TLC) was performed on Silica Gel HF with detection by charring with 30% (v/v) H₂SO₄ in MeOH or by UV detection. Column chromatography was conducted by elution of a column $(8 \times 100 \text{ mm}, 16 \times 100 \text{ mm})$ 240 mm, 18×300 mm, 35×400 mm) of Silica Gel (200–300 mesh) with EtOAc/petroleum ether (bp 60– 90 °C) as the eluent. Analytical LC was performed with a Gilson HPLC consisting of a pump (model 306), stainless steel column packed with Silica Gel (Spherisorb SiO_2 , 10×300 mm or 4.6×250 mm), differential refractometer (132-RI Detector), and UV-vis detector (model 118). EtOAc-petroleum ether (bp 60-90 °C) was used as the eluent at a flow rate of 1-4 mL/min. Solutions were concentrated at a temperature <60 °C under diminished pressure.

2.2. General procedure for the preparation of the glucopyranose 1,2-diols 8–10

To a soln of 1 (7.80 g, 30 mmol) in DMF (40 mL) was added 95% NaH (1.13 g, 45 mmol) in small portions at 0 °C. The reaction mixture was stirred for 0.5 h, at the end of which time allyl chloride or benzyl chloride (35 mmol) was added dropwise. (For the synthesis of compound 10, compound 1 was benzoylated with 1.5 equiv of benzoyl chloride in pyridine, and the work-up procedure was according to the standard method.) After stirring for 2 h at room temperature, TLC (4:1 petroleum ether–EtOAc) indicated that the reaction was complete. The reaction mixture was quenched with MeOH (2 mL), and diluted with CH₂Cl₂ (150 mL), washed with 1 N HCl, water and satd aq NaHCO₃. The organic layer was concentrated, and the residue was dissolved in 4% aq H₂SO₄ (200 mL) and then refluxed for 4 h. The resulting soln was cooled down to room temperature and extracted 2 times with EtOAc. The organic phase was discarded and the aq phase was then stirred with $CaCO_3$ (30 g) for 2 h. The reaction mixture was filtered and the residue was repeatedly washed with MeOH. The combined filtrate and washings were evaporated under diminished pressure. The residual syrupy material was treated with a minimum vol of water and filtered in order to remove some insoluble material. The residue was washed with MeOH and the combined filtrate and washings were concentrated to give the 3-O-protected glucopyranose derivatives 5–7 as syrups. These syrups were dried under high vacuum for 4 h and then were taken in anhyd DMF (80 mL), TsOH·H₂O (95 mg, 0.5 mmol) and 2-methoxypropene (3.6 mL, 36 mmol) were added successively under N₂ atmosphere. The mixture was stirred at room temperature for 2 h, and TLC (1:1 petroleum ether-EtOAc) indicated that the reaction was complete. NaHCO₃ (2.52 g, 30 mmol) was added to the reaction mixture, and the mixture was stirred for an additional 1 h. After filtration, the mixture was concentrated under diminished pressure to give a residue, which was subjected to silica gel column chromatography (2:1 petroleum ether-EtOAc) to give the desired diol compounds.

2.2.1. 3-*O*-Allyl-4,6-*O*-isopropylidene-α,β-D-glucopyranose (8). Yield 5.95 g, overall yield from 1: 76%, amorphous solid. $[\alpha]_D^{22}$ +33 (*c* 0.4, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 5.95–5.89 (m, 1H, CH₂=CH–CH₂O), 5.26 (d, 1H, *J* 3.1 Hz, H-1, α-anomer), 5.32–5.15 (m, 2H, CH₂=CH–CH₂O), 4.66 (d, 1H, *J* 7.9 Hz, H-1, β-anomer), 4.40–4.33 (m, 2H, CH₂=CH–CH₂O), 3.93–2.88 (m, 6H), 1.49, 1.41 (2s, 6H, *Me*₂C). Anal. Calcd for C₁₂H₂₀O₆: C, 55.37; H, 7.74. Found: C, 55.09; H, 8.02.

2.2.2. 3-*O*-Benzyl-4,6-*O*-isopropylidene-α,β-D-glucopyranose (9). Yield 7.50 g, overall yield from 1: 81%, amorphous solid. $[\alpha]_D^{22}$ +30 (*c* 0.7, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 7.39–7.26 (m, 5H, Ar-*H*), 5.22 (d, 1H, *J* 2.6 Hz, H-1, α-anomer), 4.92–4.69 (m, 2H, CH₂Ph), 4.57 (d, 1H, *J*₁ 6.8 Hz, H-1, β-anomer), 3.92–3.24 (m, 6H), 1.47, 1.42 (2s, 6H, *Me*₂C, β-anomer), 1.45, 1.41 (2s, 6H, *Me*₂C, α-anomer). Anal. Calcd for C₁₆H₂₂O₆: C, 61.92; H, 7.15. Found: C, 61.76; H, 7.30.

2.2.3. 3-O-Benzoyl-4,6-O-isopropylidene-α,β-D-glucopyranose (10). Yield 7.30 g, overall yield from 1: 75%, foamy solid. $[\alpha]_D^{22}$ +26 (*c* 0.8, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.08–7.43 (m, 5H, Bz-*H*), 5.39 (dd, *J* 9.5, 9.7 Hz, H-3, α-anomer), 5.32 (d, 1H, *J* 3.7 Hz, H-1, α-anomer), 5.25 (dd, *J* 9.5, 9.7 Hz, H-3, β-anomer), 4.84 (d, 1H, *J* 7.6 Hz, H-1, β-anomer), 4.05–3.39 (m, 5H), 1.50, 1.43 (2s, 6H, *Me*₂C, β-anomer), 1.49, 1.42 (2s,

6H, Me_2C , α -anomer). Anal. Calcd for $C_{16}H_{20}O_7$: C, 59.25; H, 6.22. Found: C, 59.03; H, 6.30.

2.3. General procedure for the preparation of compounds 11–19

1.2-Diols (10.0 mmol) were dissolved in dry CH₂Cl₂ (40 mL) containing pyridine (40 mmol for 8 and 9; 100 mmol for 10), then under N_2 atmosphere and stirring, a soln of allyl chloroformate, or ethyl chloroformate, or acetyl chloride (11 mmol) in anhyd CH₂Cl₂ (10 mL) was added dropwise within 30 min at -15 to -20 °C. The reaction mixture was slowly warmed to room temperature and stirred for 2 h, at the end of which time TLC (3:1 petroleum ether-EtOAc) indicated that the reaction was complete. The reaction mixture was diluted with CH₂Cl₂ (100 mL), washed with 1 N HCl, satd aq NaHCO₃, and the organic layer was dried over Na₂SO₄. The soln was concentrated under diminished pressure, and the residue was purified by column chromatography on a silica gel column (5:1 petroleum ether-EtOAc). Yields and spectral data of 11-19 are as follows.

2.3.1. 3-*O*-Allyl-1-*O*-allyloxycarbonyl-4,6-*O*-isopropylidene- β -D-glucopyranose (11). Yield 2.82 g (82%), syrup. $[\alpha]_D^{22}$ +33 (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 5.93–5.78 (m, 2H, 2CH₂=C*H*–CH₂O), 5.38 (d, 1H, *J* 8.2 Hz, H-1), 5.34–5.09 (m, 4H, 2CH₂=CH–CH₂O), 4.63–4.56 (m, 2H, CH₂=CH– CH₂O), 4.40–4.10 (m, 2H, CH₂=CH–CH₂O), 3.88 (dd, 1H, *J* 5.2, 10.5 Hz, H-6a), 3.67 (dd, 1H, *J* 10.5, 10.1 Hz, H-6b), 3.61 (dd, 1H, *J* 8.2, 8.8 Hz, H-3), 3.50 (dd, 1H, *J* 8.2, 8.8 Hz, H-2), 3.39 (dd, 1H, *J* 8.8, 8.7 Hz, H-4), 3.37–3.20 (m, 1H, H-5), 3.12 (br s, 1H, OH), 1.42, 1.34 (2s, 6H, *Me*₂C). Anal. Calcd for C₁₆H₂₄O₈: C, 55.81; H, 7.02. Found: C, 55.53; H, 6.88.

2.3.2. 3-O-AllyI-1-O-ethyloxycarbonyI-4,6-O-isopropyl-idene-β-D-glucopyranose (12). Yield 2.82 g (85%), syrup. $[\alpha]_D^{22}$ +52 (c 0.9, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 6.00–5.87 (m, 1H, CH₂=CH=CH₂O), 5.45 (d, 1H, J 8.5 Hz, H-1), 5.33–5.19 (m, 2H, CH₂=CH-CH₂O), 4.42–4.35 (m, 1H, 1CH₂=CH-CHHO), 4.28–4.17 (m, 3H, 1CH₂=CH-CHHO, CH₃CH₂O), 3.94 (dd, 1H, J 5.3, 10.8 Hz, H-6a), 3.75 (dd, 1H, J 10.8, 10.8 Hz, H-6b), 3.65 (dd, 1H, J 8.8, 9.0 Hz, H-3), 3.50 (dd, 1H, J 8.5, 8.8 Hz, H-2), 3.45 (dd, 1H, J 9.0, 9.1 Hz, H-4), 3.34–3.35 (m, 1H, H-5), 2.91 (br s, 1H, OH), 1.49, 1.41 (2s, 6H, Me₂C), 1.29 (t, 3H, J 13.2 Hz, CH₃CH₂O). Anal. Calcd for C₁₅H₂₄O₈: C, 54.21; H, 7.28. Found: C, 54.52; H, 7.44.

2.3.3. 1-O-Acetyl-3-O-allyl-4,6-O-isopropylidene-\alpha,\beta-D-glucopyranose (13). Yield 2.23 g (74%), foamy solid. $[\alpha]_{D}^{22}$ +49 (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃, 300 MHz):

δ 6.24 (d, 0.03H, J 3.90 Hz, H-1, α-anomer), 6.02–5.80 (m, 1H, CH₂=CH–CH₂O, α,β-anomer), 5.58 (d, 0.9H, J 8.0 Hz, H-1, β-anomer), 5.36–5.11 (m, 2H, CH₂=CH–CH₂O, α,β-anomer), 4.42–4.14 (m, 2H, CH₂=CH–CH₂O, α,β-anomer), 3.92 (dd, 0.9H, J 5.5, 10.7 Hz, H-6a, β-anomer), 3.76–3.38 (m, 5H, H-6a, α-anomer; H-2–5, α,β-anomer; H-6b, α,β-anomer), 3.12 (br s, 1H, OH), 2.14 (s, 3H, β-CH₃CO), 2.09 (s, 3H, α-CH₃CO), 1.49, 1.41 (2s, 6H, α,β-Me₂C). Anal. Calcd for C₁₄H₂₂O₇: C, 55.62; H, 7.33. Found: C, 55.35; H, 7.20.

2.3.4. 1-O-Allyloxycarbonyl-3-O-benzyl-4,6-O-isopropylidene-β-D-glucopyranose (14). Yield 3.38 g (86%), syrup. $[\alpha]_D^{22}$ +41 (*c* 0.8, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 7.40–7.27 (m, 5H, Ar-*H*), 5.98–5.85 (m, 1H, CH₂=CH–CH₂O), 5.45 (d, 1H, *J* 7.9 Hz, H-1), 5.40–5.25 (m, 2H, CH₂=CH–CH₂O), 4.90 (d, 1H, *J* 11.6 Hz, CHHPh), 4.73 (d, 1H, *J* 11.6 Hz, CHHPh), 4.68–4.64 (m, 2H, CH₂=CH–CH₂O), 3.96 (dd, 1H, *J* 5.3, 10.7 Hz, H-6a), 3.79–3.42 (m, 4H, H-2, H-3, H-4, H-6b), 3.44–3.35 (m, 1H, H-5), 2.53 (br s, 1H, OH), 1.48, 1.42 (2s, 6H, *Me*₂C). Anal. Calcd for C₂₀H₂₆O₈: C, 60.90; H, 6.64. Found: C, 60.73; H, 6.78.

2.3.5. 3-O-Benzyl-1-O-ethyloxycarbonyl-4,6-O-isopropylidene-β-D-glucopyranose (15). Yield 3.48 g (91%), syrup. $[\alpha]_D^{22}$ +44 (*c* 1.2, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 7.36–7.25 (m, 5H, Ar-*H*), 5.44 (d, 1H, *J* 7.9 Hz, H-1), 4.93 (d, 1H, *J* 11.7 Hz, C*H*HPh), 4.73 (d, 1H, *J* 11.7 Hz, C*H*HPh), 4.26–4.19 (m, 2H, CH₃C*H*₂O), 3.94 (dd, 1H, *J* 5.2, 10.6 Hz, H-6a), 3.78–3.50 (m, 4H, H-2, H-3, H-4, H-6b), 3.43–3.35 (m, 1H, H-5), 2.61 (d, 1H, *J* 3.0 Hz, O*H*), 1.47, 1.42 (2s, 6H, *Me*₂C), 1.31 (t, 3H, *J* 7.2 Hz, C*H*₃CH₂O). Anal. Calcd for C₁₉H₂₆O₈: C, 59.68; H, 6.85. Found: C, 59.66; H, 6.57.

2.3.6. 1-*O*-Acetyl-3-*O*-benzyl-4,6-*O*-isopropylidene-α,β-D-glucopyranose (16). Yield 2.78 g (79%), syrup. $[\alpha]_D^{22}$ +42 (*c* 1.3, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 7.36–7.24 (m, 5H, Ar-*H*, α,β-anomer), 5.35 (d, 0.02H, *J* 3.8 Hz, H-1, α-anomer), 4.84 (d, 0.97H, *J* 8.8 Hz, H-1, β-anomer), 4.85–4.58 (m, 2H, CH₂Ph, α,β-anomer), 4.21 (dd, 0.02H, *J* 5.8, 10.5 Hz, H-6a, α-anomer), 3.95–3.55 (m, 6H, H-2–6, α,β-anomer), 2.05 (s, 2.9H, CH₃CO, β-anomer), 2.03 (s, 0.1H, CH₃CO, α-anomer), 1.49, 1.43 (2s, 6H, *Me*₂C, α,β-anomer). Anal. Calcd for C₁₈H₂₄O₇: C, 61.35; H, 6.86. Found: C, 61.30; H, 7.19.

2.3.7. 1-*O***-Allyloxycarbonyl-3-***O***-benzoyl-4,6-***O***-isopropylidene-\alpha,\beta-D**-glucopyranose (17). Yield 3.39 g (83%), foamy solid. For β isomer: $[\alpha]_D^{22} + 29$ (*c* 1.1, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.07–7.44 (m, 5H, Bz-*H*), 6.00–5.87 (m, 1H, CH₂=CH–CH₂O), 5.58 (d, 1H, *J* 7.9 Hz, H-1), 5.45–5.29 (m, 2H, CH₂=CH–CH₂O),

5.27 (dd, 1H, J 9.3, 9.2 Hz,) 4.70–4.65 (m, 2H, CH₂=CH–CH₂O), 4.03 (dd, 1H, J 5.4, 10.8 Hz, H-6a), 3.94–3.77 (m, 3H, H-2, H-4, H-6), 3.59–3.51 (m, 1H, H-5), 3.08 (br s, 1H, OH), 1.50, 1.37 (2s, 6H, Me_2 C). Anal. Calcd for C₂₀H₂₄O₉: C, 58.82; H, 5.92. Found: C, 59.03; H, 5.77.

2.3.8. 3-*O*-**Benzoyl-1**-*O*-ethyloxycarbonyl-4,6-*O*-isopropylidene- α , β -D-glucopyranose (18). Yield 3.21 g (81%), foamy solid. For β isomer: $[\alpha]_D^{22}$ +33 (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.08–7.44 (m, 5H, Bz-H), 5.58 (d, 1H, *J* 8.0 Hz, H-1), 5.26 (dd, 1H, *J* 9.2, 9.3 Hz, H-3), 4.30–4.23 (m, 2H, CH₃CH₂O), 4.04 (dd, 1H, *J* 5.0, 10.5 Hz, H-6a), 3.92–3.76 (m, 3H, H-2, H-4, H-6b), 3.59–3.50 (m, 1H, H-5), 3.00 (d, 1H, *J* 4.3 Hz, OH), 1.49, 1.37 (2s, 6H, *Me*₂C), 1.33 (t, 3H, *J* 9.4 Hz, CH₃CH₂O). Anal. Calcd for C₁₉H₂₄O₉: C, 57.57; H, 6.10. Found: C, 57.84; H, 6.02.

2.3.9. 1-*O*-Acetyl-3-*O*-benzoyl-4,6-*O*-isopropylidene-α,β-**D**-glucopyranose (19). Yield 2.64 g (72%), foamy solid. $[\alpha]_D^{22}$ +36 (*c* 1.1, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.09–7.43 (m, 5H, Bz-*H*, α,β-anomer), 6.21 (d, 0.11H, *J* 4.0 Hz, H-1, α-anomer), 5.73 (d, 0.88H, *J* 8.1 Hz, H-1, β-anomer), 5.40 (dd, 0.11H, *J* 9.2, 9.3 Hz, H-3, α-anomer), 5.26 (dd, 0.88H, *J* 9.2, 9.3 Hz, H-3, β-anomer), 4.36–3.50 (m, 5H, H-2–6, α,β-anomer), 2.97 (br s, 1H, OH), 2.20 (s, 0.33H, CH₃CO, α-anomer), 215 (s, 2.64H, CH₃CO, β-anomer), 1.49, 1.37 (2s, 6H, *Me*₂C, α,β-anomer). Anal. Calcd for C₁₈H₂₂O₈: C, 59.01; H, 6.05. Found: C, 59.29; H, 6.23.

2.4. General procedure for the preparation of compounds 20, 21, 24, and 25

Compound 14, or 15, or 13, or 16 (5 mmol) and K_2CO_3 (2.0 g) were dried under high vacuum for 4 h and then CH₂Cl₂ (20 mL) was added to the mixture. Then under N₂ atmosphere, CCl₃CN (1.0 mL, 10 mmol) was added, the mixture was stirred for 2 h, at the end of which time DBU (135 µL, 0.9 mmol) was added. The reaction mixture was stirred for 12 h, and TLC (2:1 petroleum ether– EtOAc) indicated that the reaction was complete. After filtration, the mixture was concentrated under diminished pressure to give a residue, purification of the crude product on a silica gel column with 5:1 petroleum ether– EtOAc as the eluent furnished the desired compounds.

2.4.1. 3-*O*-Benzyl-1-*O*-allyloxycarbonyl-4,6-*O*-isopropylidene-2-*O*-trichloroacetimidoyl-β-D-glucopyranose (20). Yield 2.39 g (89%), syrup. $[\alpha]_D^{22}$ +19 (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.65 (s, 1H, CN*H*CCl₃), 7.31–7.23 (m, 5H, Ar-*H*), 5.94–5.68 (m, 1H, CH₂=C*H*–CH₂O), 5.71 (d, 1H, *J* 8.0 Hz, H-1), 5.41 (dd, 1H, *J* 8.0 8.1 Hz, H-2), 5.38–5.23 (m, 2H, CH₂=CH–CH₂O), 4.83–4.71 (m, 2H, CH₂Ph), 4.67– 4.55 (m, 2H, CH₂=CH–C H_2 O), 4.01–3.75 (m, 4H, H-3, H-4, H-6), 3.51-3.43 (m, 1H, H-5), 1.50, 1.43 (2s, 6H, Me_2 C). Anal. Calcd for C₂₂H₂₆Cl₃NO₈: C, 49.04; H, 4.86; N, 2.60. Found: C, 49.31; H, 5.02; N, 2.98.

2.4.2. 3-*O*-Benzyl-1-*O*-ethyloxycarbonyl-4,6-*O*-isopropylidene-2-*O*-trichloroacetimidoyl-β-D-glucopyranose (21). Yield 2.45 g (93%), syrup. $[\alpha]_D^{22}$ +27 (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.65 (s, 1H, CN*H*CCl₃), 7.32–7.24 (m, 5H, Ar-*H*), 5.72 (d, 1H, *J* 8.0, H-1), 5.41 (dd, 1H, *J* 8.0, 8.1 Hz, H-2), 4.83–4.71 (m, 2H, CH₂Ph), 4.21 (q, 2H, *J* 7.1 Hz, OCH₂CH₃), 4.01–3.75 (m, 4H, H-3, H-4, H-6), 3.51–3.45 (m, 1H, H-5), 1.49, 1.43 (2s, 6H, *Me*₂C), 1.26 (t, 3H, *J* 7.1 Hz, OCH₂CH₃). Anal. Calcd for C₂₁H₂₆Cl₃NO₈: C, 47.88; H, 4.97; N, 2.66. Found: C, 47.75; H, 4.71; N, 2.29.

2.4.3. 2-*O*-Acetyl-3-*O*-allyl-4,6-*O*-isopropylidene-α,β-D-glucopyranosyl trichloroacetimidate (24). Yield 1.87 g (84%), syrup. $[\alpha]_D^{22} + 20$ (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.66 (s, 1H, CNHCCl₃, β-anomer), 8.60 (s, 1H, CNHCCl₃, α-anomer), 6.45 (d, 1H, *J* 3.9 Hz, H-1, α-anomer), 5.83 (d, 1H, *J* 8.2 Hz, H-1, β-anomer), 5.93–5.76 (m, 1H, CH₂=CH–CH₂O, α,β-anomer), 5.21 (dd, 1H, *J* 8.1, 8.2 Hz, H-2, β-anomer), 4.99 (dd, 1H, *J* 3.9, 8.2 Hz, H-2, α-anomer), 4.31–3.43 (m, 5H, H-3, H-4, H-5, H-6), 2.05 (s, 3H, CH₃CO, α-anomer), 2.04 (s, 3H, CH₃CO, β-anomer), 1.51, 1.42 (2s, 6H, *Me*₂C, β-anomer). Anal. Calcd for C₁₆H₂₂Cl₃NO₇: C, 43.02; H, 4. 96; N, 3.14. Found: C, 42.83; H, 4.70; N, 3.30.

2.4.4. 2-O-Acetyl-3-O-benzyl-4,6-O-isopropylidene-α,β-**D**-glucopyranosyl trichloroacetimidate (25). Yield 1.96 g (79%), syrup. $[\alpha]_{D}^{22}$ +12 (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.65 (s, 1H, CNHCCl₃, β-anomer), 8.58 (s, 1H, CNHCCl₃, α-anomer), 7.35-7.24 (m, 5H, Ar-H, α,β-anomer), 6.46 (d, 1H, J 3.8 Hz, H-1, αanomer), 5.83 (d, 1H, J 7.9 Hz, H-1, β-anomer), 5.24 (dd, 1H, J 7.8, 7.9 Hz, H-2, β-anomer), 5.00 (dd, 1H, J 3.8, 7.9 Hz, H-2, α-anomer), 4.88–4.64 (m, 2H, CH₂Ph, α,β-anomer), 4.02–3.46 (m, 5H, H-3, H-4, H-5, H-6), 1.97 (s, 3H, CH₃CO, α-anomer), 1.96 (s, 3H, CH₃CO, β-anomer), 1.51, 1.45 (2s, 6H, Me₂C, α-anomer), 1.50, 1.44 (2s, 6H, Me₂C, β-anomer). Anal. Calcd for C₂₀H₂₄Cl₃NO₇: C, 48.36; H, 4.87; N, 2.82. Found: C, 48.00; H, 4.59; N, 3.05.

2.5. General procedure for the coupling reaction: synthesis of compounds 27, 29, and 31

Compound **12** (120 mg, 0.30 mmol) and sugar trichloroacetimidates **26**, or **28**, or **30** (0.36 mmol) were dried together under high vacuum for 4 h, then dissolved in anhyd CH₂Cl₂ (10 mL). TMSOTf (1.8 μ L, 0.010 mmol) was added at -10 °C under N₂ atmosphere. The reaction mixture was stirred for 2 h, during which time the temperature was gradually warmed to ambient temperature. Then the mixture was neutralized with Et₃N and concentrated to dryness under diminished pressure and the residue was purified by flash chromatography (4:1 petroleum ether–EtOAc). The physical data of the obtained disaccharides **27**, **29**, and **31** are as follows.

2.5.1. 2,3,4,6-Tetra-O-benzoyl-a-D-mannopyranosyl-(1→2)-1-O-allyloxycarbonyl-3-O-benzyl-4,6-O-isopropylidene-β-D-glucopyranose (27). Yield 277 mg (95%), foamy solid. $[\alpha]_D^{22}$ +14 (*c* 1.2, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.10–7.17 (m, 25H, 5Ar-H), 6.08 (dd, 1H, J 10.2 Hz, H-4'), 5.86 (dd, 1H, J 3.3, 10.2 Hz, H-3'), 5.95-5.82 (m, 1H, OCH₂CH=CH₂), 5.67 (d, 1H, J 8.0 Hz, H-1), 5.59 (dd, 1H, J 1.4, 3.3 Hz, H-2'), 5.41 (d, 1H, J 1.4 Hz, H-1'), 5.34–5.17 (m, 2H, OCH₂CH=CH₂), 5.09-4.76 (m, 2H, CH₂Ph), 4.74-4.46 (m, 3H), 4.30 (dd, 1H, J 2.1, 12.5 Hz, H-6'a), 4.04–3.73 (m, 6H), 3.15–3.44 (m, 1H), 1.52, 1.45 (2s, 6H, Me_2C); ¹³C NMR: δ 166.0, 165.5, 165.4, 165.1, 153.2, 137.9, 133.3, 133.2, 132.8, 131.1, 130.2, 129.8, 129.7, 129.3, 129.2, 129.1, 128.5, 128.4, 128.3, 128.2, 128.0, 119.0, 99.5, 98.0, 96.9, 78.9, 77.2, 75.7, 74.8, 70.3, 69.8, 69.3, 68.9, 67.7, 66.6, 62.0, 29.0, 19.8. Anal. Calcd for C₅₄H₅₂O₁₇: C, 66.66; H, 5.39. Found: C, 66.84; H, 5.70.

2.5.2. 2,3,4,6-Tetra-O-benzoyl- β -D-glucopyranosyl- $(1 \rightarrow 2)$ -1-O-allyloxycarbonyl-3-O-benzyl-4,6-O-isopropylidene-β-**D-glucopyranose** (29). Yield 257 mg (88%), foamy solid. $[\alpha]_{D}^{22}$ +8.0 (c 1.1, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): & 8.02-7.19 (m, 25H, 5Ar-H), 5.85 (dd, 1H, J 9.8 Hz, H-4'), 5.90–5.77 (m, 1H, OCH₂CH=CH₂), 5. 68 (dd, 1H, J 9.8 Hz, H-3'), 5.67 (d, 1H, J 8.9 Hz, H-1), 5.55 (dd, 1H, J 9.0, 9.8 Hz, H-2'), 5.33-5.19 (m, 2H, OCH₂CH=C H_2), 522 (d, 1H, J 9.0 Hz, H-1'), 4.65-4.32 (m, 6H), 4.08-3.84 (m, 3H), 3.72-3.53 (m, 3H), 3.36-3.21 (m, 1H), 1.38, 1.25 (2s, 6H, Me_2C); ^{13}C NMR: δ 166.1, 165.8, 165.2, 165.1, 153.3, 138.59, 133.3, 133.2, 133.1, 132.9, 131.1, 129.8, 129.7, 129.6, 129.5, 129.1, 128.9, 128.8, 128.3, 128.2, 128.2, 127.6, 127.5, 119.1, 101.1, 99.3, 96.4, 81.4, 79.1, 77.2, 74.5, 73.8, 73.1, 72.1, 72.0, 69.8, 68.9, 67.0, 63.3, 61.9, 28.9, 18.9. Anal. Calcd for C₅₄H₅₂O₁₇: C, 66.66; H, 5.39. Found: C, 66.48; H, 5.24.

2.5.3. 2,3,4-Tri-*O*-benzoyl-β-D-xylopyranosyl-(1 \rightarrow **2**)-1-*O*-allyloxycarbonyl-3-*O*-benzyl-4,6-*O*-isopropylidene-β-D-glucopyranose (**31**). Yield 229 mg (91%), foamy, solid. [α]_D²²+22 (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃, 300 MHz): δ 8.01–7.13 (m, 20H, 4Ar-*H*), 5.93–5.80 (m, 1H, OCH₂C*H*=CH₂), 5.85 (dd, 1H, *J* 5.4 Hz, H-3'), 5.56 (d, 1H, *J* 7.9 Hz, H-1), 5.42 (d, 1H, *J* 4.4 Hz, H-1'), 5.36–5.21 (m, 3H, H-2', OCH₂CH=CH₂), 5.18 (m, 1H, H-4'), 4.75–4.41 (m, 5H), 4.00–3.69 (m, 6H), 3.44– 3.35 (m, 1H), 1.42, 1.33 (2s, 6H, Me_2C). Anal. Calcd for C₄₆H₄₆O₁₅: C, 65.86; H, 5.53. Found: C, 65.59; H, 5.44.

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